THE TEXAS COMPANY

REFINING DEPARTMENT TECHNICAL & RESEARCH DIVISION



REPORT ON

CARTHAGE HYDROCOL, INC.

BROWNSVILLE PLANT OPERATIONS

ANALYSIS OF SYNTHESIS REACTOR DATA

Engr'g Dev. Group (NY Office)
Report No. 25

Date March 29, 1952

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Introduction

The conversions and yields on the Synthesis Reactors at Brownsville have been consistently very much lower than those predicted in design and also much lower than those obtained with the same catalyst on pilot and laboratory units.

For over a year this has been the subject of much concern to many individuals and almost every conceivable reason to account for this discrepancy has been expressed at some time or another.

The latest widely accepted opinion is that poor catalyst contacting efficiency in the large commercial reactors relative to that obtained in the small pilot units is responsible for much if not most of the trouble and steps are now being taken to compartmentalize one of the reactors at Brownsville to simulate operation with a number of smaller reactors in parallel.

In addition, on the assumption that this may not be the answer, engineering studies are being made to determine what changes would be required to operate the two reactors in two stages instead of in parallel. So far, the opinion seems to be divided on whether two stage operation will result in better conversion and yields than parallel operation.

On the off-chance that an additional independent review of the situation, and particularly a detailed study of the Browns-ville data itself, might disclose some factors which had been overlooked, or help to establish which opinions are correct, arrangements were made for the writer to make such a survey at Brownsville. This study which was made during the period of February 12 to March 25, 1952 is the subject of the present report.

Scope

The first step in this survey was to trace all pertinent lines in the field to become familiar with the location of flow meters, sampling connections etc. and to look for by-passing and leakage possibilities. None of the latter were found. In the course of this step flow diagrams were made of the important lines and these diagrams are included in the appendix for future reference.

Next, a detailed review was made of the methods used by the plant to calculate and report run data. Checks were made on the reliability of the data, as discussed in the report where pertinent, and then after innunerable false starts the correlations presented here were developed.

Correlations - H2 Conversion vs %CO in FF to C3+

The following Figs. 1, 1A and 1B, which are a plot of $\rm H_2$ converted vs % of CO in fresh feed which went to $\rm C_3^+$, are a graphical comparison of all the following data:

- 1. All available Brownsville data.
- 2. Stanolind Filot unit data on Allan Wood and Brownsville Mill Scale catalysts.
- 3. Montebello Pilot unit data on Allan Wood, Brownsville Mill Scale and Spent CM&S catalysts.

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- 4. Beacon data on Brownsville Mill Scale and Spent CM& S catalysts.
- 5. HRI Olean Runs H-24 & H-25 on Spent CM&S catalyst, which runs formed the basis for the Brownsville Plant Design.
- 6. The Brownsville Design.

On these graphs each point which is joined by lines, represents about 24 hrs. operation. The points are joined to-gether in chronological order. A ring has been placed around the first day of each run. The unjoined points which are all laboratory or pilot unit data, represent averages of several periods or of a whole run.

The Brownsville data were divided into three plots, Figs. 1, 1A & 1B simply to avoid cluttering. The Beacon, Montebello and Stanolind data obtained with Brownsville Mill Scale catalyst, the design point and the Montebello data on CM&S catalyst have been repeated on Fig. 1A.

The solid lines on all three of these plots, one for Fresh Feed having an $\rm H_2/CO$ ratio of 2.0 and the other for an $\rm H_2/CO$ ratio of 1.0 are those which were developed in EDG Report No. 1 dated May 20, 1947. These lines represent an extremely large number of runs made at various conditions in HRI, Beacon, Stanolind and Jersey laboratories.

The data for Figs. 1, 1A & 1B and for others which will be described later are included in Tables I & II of the appendix.

We shall discuss these figures after we have developed sufficient background to understand what they mean.

Sources of Data

The sources of these data to-gether with a brief description of the reactors on which they were obtained are as follows:

Montebello Data

The Montebello Data for Run 49 with Allan Wood catalyst were obtained from the detailed yield calculation sheets included in the appendix of partial report TDC 802 - 37 P.

The data for Run 63 on Brownsville's Mill Scale catalyst and for Run 59 on spent CM&S catalyst were obtained from the daily yield calculation sheets which were sent to us from Montebello by mail.

The data for the single points representing parts of Runs 57, 64, 65 & 66 with Mill Scale catalyst were obtained from the calculation sheets included in the undated report which Mr. duBois Eastman presented at the recent Tulsa Meeting with Stanolind.

actor which is described in detail in P.R. 37 P referred to above. This reactor is a straight pipe 30 ft. long with 3-2" sched. 80 pipes about 18 ft. long inside the reactor for cooling. Steaming water is used as the cooling medium. The 11 ft. above the cooling tubes is disengaging space. An external cyclone is used to separate the carried over fines from the reactor effluent but no fines are returned to the unit. Preheat of the total feed is varied automatically to control reactor bed temperature.

Stanolind Data

The Stanolind Data for Run D-201-29 on Allan Wood Catalyst were obtained from the calculation sheets included in the appendix of partial report TDC 802 - 37 P. and the data for the other runs on Mill Scale Catalyst from the Tulsa Meeting Report referred to above.

The Stanolind D-201 reactor is 8" I.D. about 20 ft. long and is cooled by an external jacket. About 6 to 8 ft. of the upper part of the reactor is expanded to provide catalyst disengaging space. An external cyclone is used and feed preheat is varied automatically as at Montebello.

Beacon Data

The Beacon data for old runs 7024, 7027 and 8001 on Spent CM&S Catalyst were taken from the 1947 Report (EDG #1). The first two were made on a 1 5/8" I.D. stirred reactor and the other on a 3/4" I.D. 9 ft. long baffled reactor similar to that still in use and on which the other data shown on the graphs were obtained. This reactor is topped with a catalyst disengaging space 4 to 6" in diam. and about 2 ft. long and filters are used to knock back the catalyst. These laboratory reactors are jacket cooled and the total feed is normally regularly preheated to about 600°F.

HRI Data

The HRI data on spent CM&S catalyst were reproduced from the 1947 correlation referred to above. They were obtained on an 11 1/2" I.D. reactor 18 1/2 ft. long with 19 - 1" ex. hvy. tubes provided for cooling with Dowtherm. An expanded section about 2 ft. long was provided for catalyst disengaging and filters were used for final catalyst separation. In the particular runs shown here the total feed was preheated to 600 - 650°F.

Brownsville Design Conditions

The following yields and operating conditions were predicted by HRI in Case VI Process Specs. Sect. 350 and are the goal we are shooting for:

	Per Reactor	<u>Total</u>
Syn. Gas Feed Rate (MMSCFH) Recycle/F.F. ratio Catalyst Holdup Tons Steam Prod'n. M#/Hr. Cat. bed Density #/C.F. Reactor Press. Reactor Effluent Temp. °F. % CO Conv. (on F.F.) % H2 Conv. (on F.F.) % H2+CO Conv. (on F.F.) Fresh Feed Comp: Mol % CO H2 CO2 N2+ A CH4 H2O	5.0 1.0 99 215.5 100 425 650 98.2 86.3 90.4 33.14 61.12 1.30 2.95 1.07 0.43	10.0 1.0 198 431

In the design, the water-gas shift reaction was assumed to be in equilibrium $(H_2)(CO_2)$ = 22 at 650°F) at the reactor outlet. $(CO)(H_2O)$

The predicted yields were as follows:

	BPOD	$rac{ ext{BPOD}}{ ext{Ex} ext{ Casinghead}} \ 10_n^{\prime\prime} ext{ RVP Gaso.*}$
Casinghead C4's Casinghead Gaso. Synthetic C4's Synthetic Gaso. (DB) Poly Gaso. Total Gaso.	150 179 164 4781 <u>915</u> 6189	6079
Gas Oil Waxy Btms. Poly Tar Total Oil	946 103 <u>99</u> 7337	947 103 <u>95</u> 7224
Water Sol. Chemicals 213,168% Water Sol. Simple Solution Solution Simple Solution S	/day 631 9144	631
Total Synthetic Oil plus WSC Bbls/MMSCF of Syn. Gas	31.8	32.6

^{*} From PR TDC 802 - 37P page 88

The total C_3^+ yield, including water soluble chemicals can be calculated from the breakdown of the reactor effluent streams reported in the HRI Process Specs. for Sect. 350 (Case 6 design) as follows:

Table A

Total C₃+ Design Yields For Two Reactors
And 10,000,000 SCFH of Fresh Feed

	#/Hr. Flash Vapor	#/Hr. Stripper Feed	Net Reactor Effluent	#/Gal.	#/Bbl.	ВРН
N C C C C C C C C C C C C C C C C C C C	21,865 4,412 69,529 7,447 11,554 6,062 5,649 5,871 4,435 388 6,251 2,466 408 121 60 17	73 20 2,666 10 179 232 306 710 1,249 5,565 6,183 4,7,719 7,924 2,580 2,719 7,924 2,970 2,580 2,230 4,540 64,309	21,938 4,432 72,195 4,457 11,733 6,955 6,581 5,684 5,684 11,8648 7,840 7,940 7,940 2,930 2,930 4,540 209,347	8.33 8.33 8.33 8.33 8.33 8.33 8.33 8.33	349.9 349.9 349.9 349.9 130.5 110.5 181.5 211.6 2207.2 237.2 2486.2 276.9 296.2 296.	36.18 36.40 26.85 26.56 53.59 36.38 31.43 29.69 25.68 8.84 10.03 7.48 15.01 335.92
Water	Sol. Chem	S.	8,882		336.0	26.30
Water Total Out		104,045 322,274			362.22	
Total In (F.F.)		322,571		870	O BPD	

The total C_3+ and WSC in the reactor effluent amounts to 362.2 BPH or 8700 BPD. The difference between this figure and the lower corresponding figure reported above in the predicted yield table is, of course, due to the poly contractions and the C_3 , C_4 and other losses.

The total C_3 + Incl. WSC leaving the reactors amounts to 91,225%/Hr. which divided by 14 = 6,516 CH $_2$ radicals per hr. which is 74.05% of the 8,800 Mols of CO fed to the reactors in the fresh feed.

The other design factors used in these correlations were similarly calculated as follows:

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% CO in F.F. to WSC = 7.2
% CO in F.F. to CO_2 Made = 14.73
% CO in F.F. to CH_L Made = 5.1
% H_2 in F.F. to H_2O Made = 34.9
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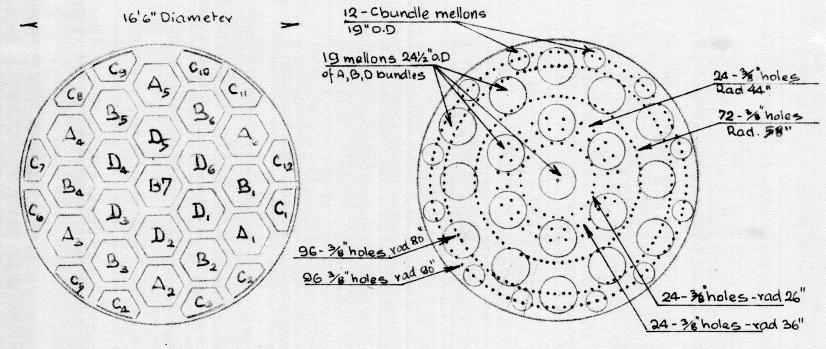
Brownsville Reactors

An assembly drawing of the Brownsville Reactors (Combustion Engineering Drg. No. E-152-572-7) is included in the appendix for ready reference. These reactors, of which there are two alike, are 16'-6" I.D. and 21'-9" high from bend line to bend line with hemispherical heads. Each reactor is filled with steam tubes which are arranged in bundles as shown in plan on the following Sketch A.

There are 1420 - 1 7/8" 0.D. tubes in all, arranged in 31 bundles. All but 12 of the bundles have 61 tubes each with 24" melons at top and bottom. The other 12 bundles have 21 or 22 tubes with 19" melons. The height of the bundles from ctr. line of bottom melons to ctr. line of top melons is 16'-6" for most bundles and 18'-6" for a few to stagger the top melons. The bottom melons are all on the same plane with their center line 21 1/4" above the grid.

SKETCH A

Arrangement of water tube bundles in reactor K 351B Arrangement of reactor grid holes and location of mellons.



Total of 1% 0.0 water tubes - 1420 in 19 bundles A,-A6, B, to B7, D, to D6, each Gltubes Total number of \$1.0 holes - 336 and 12 bundles C, to C, each 21/22 tubes. Mellons installed 24" above grid (to center)

Distance between mellon centers vertically: Bundles A.B. - 16/2A, Bundles C&D: 18/2ft

Grid consist of 24 pie sections, center plate and 48 sectors at periphery, each with Aholes

The free area between tubes is 187 sq. ft. and the distance from the grid to the center line of the lower row of top melons is $18^{\circ}-6^{\circ}$. This is probably the maximum height of catalyst bed that can be used and still have enough submerged cooling surface to take away the exothermic heat of reaction.

The grid is made up of pie shaped castings and contains 336 openings as shown in the plan on the above Sketch A. These openings were originally $3/4^n$ diameter but in May 1951, after Run 7/8, these were reduced to $3/8^n$ diameter by means of sleeve inserts. This was done on both reactors.

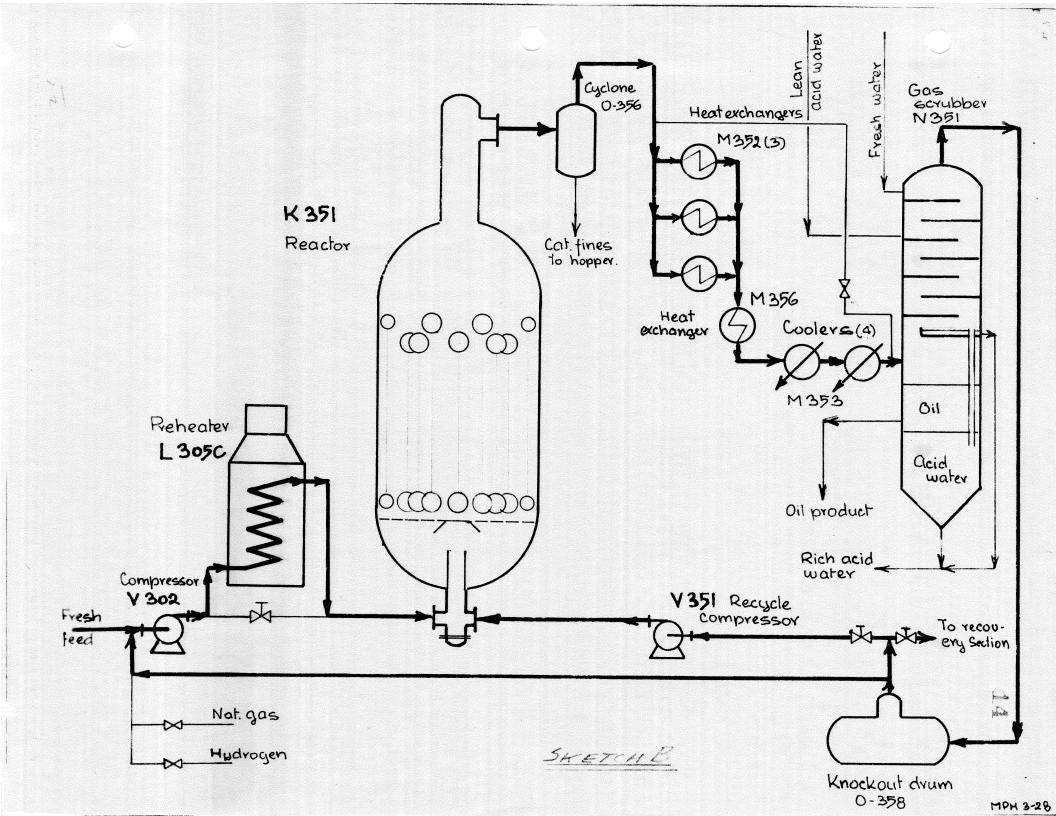
In the original reactors there were three horizontal rows of louvre type baffles installed between the cooling tubes near the tops of the bundles. These caused localized erosion of some of the tubes and were removed from both reactors in Feb. 1951 after Run #5. Reactor Operations

Before attempting to analyze the Brownsville data it is well to briefly review the operating procedures used on the reactor system.

The following Sketch B is a simplified flow diagram of the Reactor System. Su pose we start with the reactor in operation and about to come down.

A. Catalyst Stripping After Shutdown - Circulation

First the synthesis gas feed is shut off at the fresh feed compressor (V-302) section. The compressor V-302 will then be on recycle gas alone, supplemented by the Nat. Gas used for Aeration, about 55,000 SCFH and an additional 45,000 SCFH or so of Nat. Gas admitted to the V-302 suction when the Syn. Gas is cut off. The recycling rate is maintained to hold a lineal velocity thru the bed of 0.6 ft/sec with a minimum of 0.4 ft/sec to insure maintaining



fluidization. The flow will be from the V-302 compressor thru the steam superheater L-305C (now used all the time as gas htr.) thru the reactor, the M-352 & M-356 exchangers, the M-353 cooler and then thru the Gas Scrubber N-351 and thus back to the V-302 suction. The V-351 recycle compressors will be shut down.

During stripping, the system pressure is reduced to about 100% and the reactor effluent temperature is held at $650\text{--}700^{\circ}\text{F}$. without exceeding 800° on the L-305C outlet. To maintain the bed temperature at this level the steam bundles have to be cut out of service. When the mass spec. analysis shows 90 to 94% CH₄ in the tail gas the stripping is considered complete.

The reactor temperature is then reduced to 270 to 300°F. by lowering the L-305C heater outlet. The acid water is then cut out of the gas scrubber (N-351) and city water only is used. After displacing the water in the acid water line to Stanolind for one hour the water from the gas scrubber is discharged to the pond.

If the catalyst is to be continuously circulated between runs, the conditions described in the previous paragraph are the ones used except that the reactor pressure may be lowered to about 80%. The rate required to maintain a lineal velocity of 0.6 ft/sec at 80% press. and 300° F. is about 1,800,000 SCFH.

Starting From Scratch

When coming up from a dead stop the same method of circulation is used except that the bed temperature is held at 500 to 700°F. while adding the catalyst to the reactor.

Unloading Reactor

The same circulation procedure (at 300°F.) is also used when catalyst is unloaded from the reactor. After all possible catalyst has been removed in this manner (thru bottom center outlet)

the circulation is stopped and the reactor washed with hot water before unheading. When the reactors are opened several tons of catalyst are always found packed between the tubes. This has to be removed by hand prodding.

Reduction Before Startup

With the reactor circulating Natural Gas as described above, the reactor temperature is raised to 700-725°F. and the pressure increased to 200% press. while raising the recycle rate and Nat. Gas input as necessary to maintain a lineal velocity of 0.6 ft/sec. This requires a rate of about 2,800,000 SCFH.

Hydrogen is then cut into the V-302 suction at the maximum rate of production, about 20,000 SCFH and the Nat. Gas is cut out. Aeration is maintained by recirculation in the separate (reciprocating) compressors (V-352). After about 48 hours the hydrogen concentration in the system will have reached about 40% and reduction is considered completed. In some runs where in a hurry to get started this time has been cut to 12 hrs. with only 30% H₂ concentration attained.

Nothing but city water is used in the gas scrubber N-351 during reduction.

Carbiding

Prior to the last three runs #15, #16 and #17 a carbiding step has been added after reduction. This is accomplished by simply shutting off the hydrogen and substituting synthesis gas from the generator with reactor at 650 to 700°F. and 250# pressure, synthesis gas is added to the V-302 compressor section at a rate of about 50,000 SCFH and is increased by that amount every 15 minutes until a total of about 700,000 SCFH is being added. This condition is maintained for about 20 hrs.

Shortly after the start of syn. gas admission acid water washing is started in the gas scrubber N-351.

Start Up

The normal starting up procedure is essentially the same as that described above for carbiding except that the increase in syn. gas admission rate is not interrupted until the full desired rate, usually 3,000,000 SCFH has been attained. Also as the syn. gas rate is increased the reactor pressure is raised to 350% and steam bundles are placed in service as necessary to maintain and equalize reactor bed temperatures.

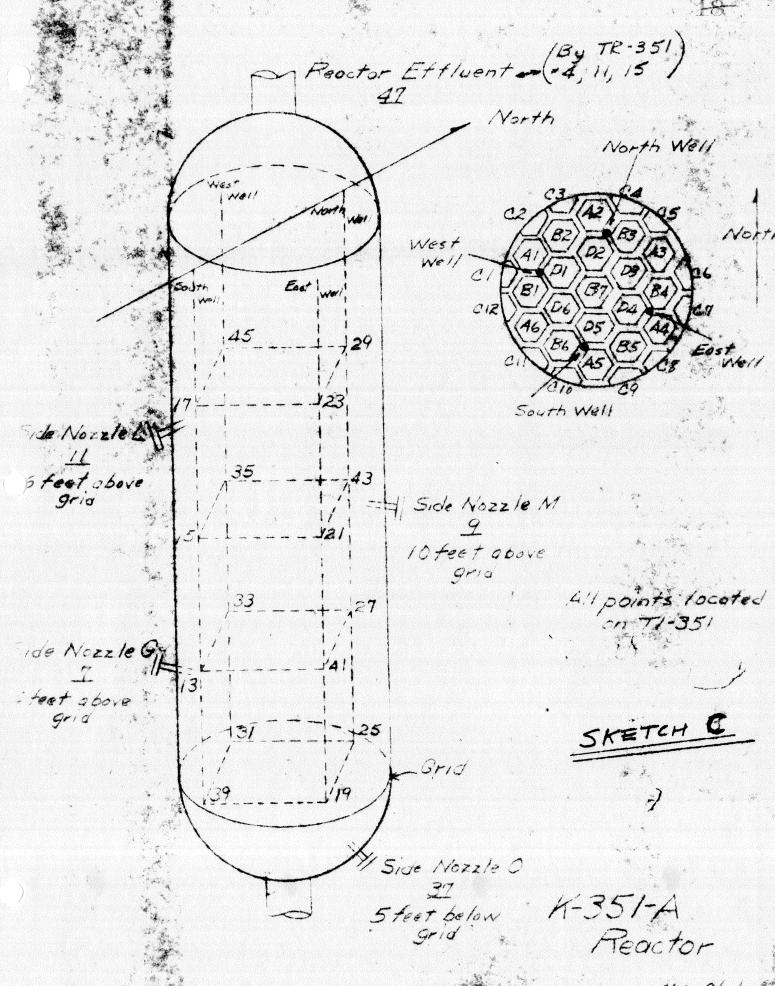
In the earlier runs some of the bundles were left out of service entirely but in recent runs it has been the practice to place them all in service pinching off each bundle circuit as necessary to equalize bed temperatures. No record is kept of the amount of pinching off of the individual steam circuits.

The following Sketch C shows the location of the thermowells in Reactor A with respect to the tube bundles. Those in Reactor B are opposite hand.

Whether carbiding or just starting up directly, acid water is admitted to the gas scrubber as soon as or shortly after synthesis gas is added to the reactor.

The recycle compressor V-351 is started as soon as pressure has been built up on the system.

The plant is always run to load the fresh feed compressor V-302, with fresh feed plus part of the recycle and this entire stream, usually about 4,500,000 SCFH is routed thru the L-305C heater and there preheated to 650°F. The remainder of the recycle is handled by the V-35l compressor and is not preheated. Thus when running with a fresh feed rate of about 3,000,000 SCFH, the



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output when only one 02 plant is running, and with a recycle to fresh feed ratio (R/FF) of one, approximately 1,500,000 SCFH of the recycle is preheated along with the fresh feed and an equal amount of the recycle is not. Therefore when the fresh feed rate is increased with the R/FF remaining the same, less of the recycle goes thru the V-302 compressor heater and therefore the combined steam reactor inlet temperature drops. In other words when running with a F.F. rate of 3,000,000 SCFH and R/FF of 1.0, 75% of the total feed to the reactor is preheated to 650°F. resulting in a total feed temperature of about 530°F. whereas when the F.F. rate is increased to 4,500,000 SCFH and the R/FF rate is still maintained at one, then only 50% of the total feed is preheated and the combined feed temperature drops to about 415°F.

Brownsville Data

Seventeen runs have been made at Brownsville. The first of any consequence so far as the reactor is concerned was Run \$\frac{1}{2}\$. No data were worked up for Run \$\frac{1}{2}\$. A general description of each run with causes of shutdown etc., is periodically made up by Carthage in N.Y. and is called "Summary of Operations". Our copy of this summary with supplementary notes is included in the appendix.

The following Table B is a list of Runs 5 to 17 inclusive showing the dates of the runs, their duration and the catalysts used. It will be noted that all runs except #8 were less than 2 weeks long and all but four were of 10 days or less. Run #10 was the first run made with well reduced catalysts and Run #11 was the first made with the fine grind catalyst now being used.

TABLE B
BROWNSVILLE REACTOR RUNS

Run	n		Reac	Reactor	Catalyst			
No.	Start	End	Days	Used	Tons		% Fe	Red'n. Temp.
5	1/16/51	1/25/51	9		140 80	Used AW & MS* Fresh MS Added	72.3 90	770
6	3/11/51	3/20/51	9		170 100	Fresh MS Fresh MS Added	88 90	750
7	4/8/51	4/13/51	6		Run Not Used. Data unavailable			
ප්	4/21/51	5/15/51	24		160 194	Used in Run 7 Fresh MS Added	73 • 7 90	
9	6/5/51 8 PM.	6/9/51 3:15 PM	4	В	126 57	Used in Run #8 Fresh Added	7 4 88	
10	7/20/51 11 AM	8/1/51 12:35 AM		В	240 43	Fresh Fresh Add e d	95+ 97	
general grant of	11/11/51 6 AM	11/20/51 10:15 AM	10	Å	248 61	Fresh Fresh Add e d	96 9 7	725
12	11/28/51 11 PM	12/7/51 12 N	9	A	214 82	Used in Run ll Fresh Added	79.3 96.6	725
7	12/21/51 5 PM	1/2/51 9:13 AM	13	В	130 127 60 49	Used Fresh Fresh Added Fresh Added**	80.9 97.3 97 96.2	725
14	1/9/52 5 PM	1/22/52 11 AM	14	В	207 55	Us ed Fresh Added	80.7 95.3	775
15	2/2/52 9:30 AM	2/11/52 4:30 PM	10	A	79 121 13 42	Us e d Fresh Used Added Fresh Added**	77.7 94.8 77.7 95.5	810
16	2/19/52 12:30 PM	2/22/52 9:30 AM	4	А	104 24 71	Used Fresh** Fresh	78.2 96 95	825 810
17	3/4/52 9:13 AM	3/13/52 2 PM	10	A	130 6 55 31	Used Fresh Fresh Added Fresh Added	80.3 95 80.3 87	

^{*} AW = Allan Wood used only in Run #5 and before all other runs used Mill Scale MS. Where so noted the catalyst was added during the run.

^{**} These batches were reduced and precarbided before adding to reactor. In addition in situ carbiding was practiced in the reactor before the beginning of Runs 15, 16 & 17.

The stock room at Brownsville works up a daily (6 AM to 6 AM) statement of reactor operations using 24 hr. averages of meter readings, temperature and pressure measurements with averages of three mass spectrometer analyses on the fresh feed and two on the recycle.

In this calculation, because there is no direct measure of the fresh feed to the reactor, the fresh feed is found by balance in which the output is made up of:

- 1. Absorber tail gas metered and analyzed.
- 2. Raw Primary Oil gaged when treating unit not running, otherwise metered.
- 3. Water Soluble Chemicals obtained from ctanolind.
- 4. Water make obtained from Stanolind.

The absorber tail gas is adjusted to remove metered aeration gas (Nat. Gas) and the total output weight is then used together with fresh feed analysis to calculate the fresh feed rate and input of each fresh feed component.

A check against this calculated rate can be obtained by subtracting from the metered rate of the combined fresh feed and part of the recycle on the fresh feed compressor discharge (preheater outlet), the rate of recycle to the fresh feed compressor which is measured by a pitot venturi not considered reliable enough to use for the daily calculations. The remainder of the recycle, that going thru the recycle compressor, is metered separately.

An additional check on the calculated reactor fresh feed rate is obtained by comparing with the syn. gas generator output which in turn is calculated by carbon balance around the generator.

22

The accuracy of the reactor calculation is also noted daily by comparing the calculated argon input with that leaving the system which is calculated by difference as outlined above. We have made spot checks on all these calculated rates and found them reasonably reliable.

The above method of calculation applies to all runs which were made since March 1951 after Run #5. Run #5 was calculated separately but in a similar manner.

These stock room data do not agree exactly with those reported in the teletypes because the teletypes are based on spot 6 AM readings using midnight sample analyses and short-cut calculations.

At the end of each run the daily stock room data are tabulated to form what is called the run "Operating and Yield Summary" which is sent to New York and copies of which are included in the appendix.*

The data from the run summaries, supplemented by a few additional data, were tabulated and recalculated for correlation purposes and are included in Tables I & II in the appendix as referred to above.

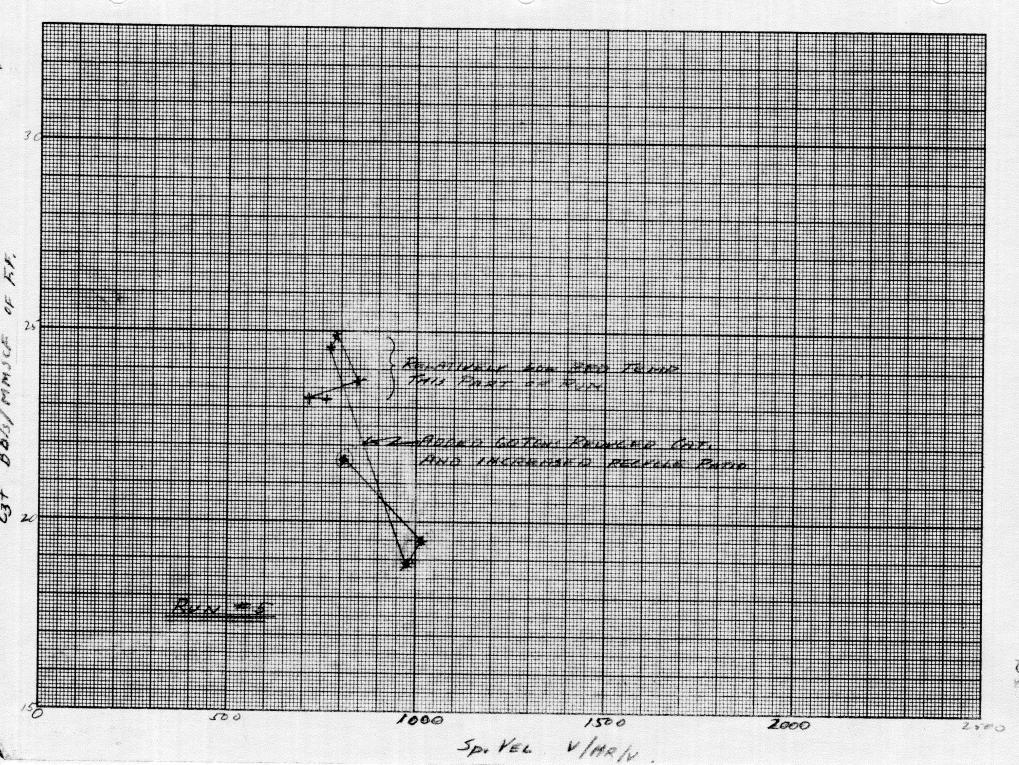
Reliability of Brownsville Data

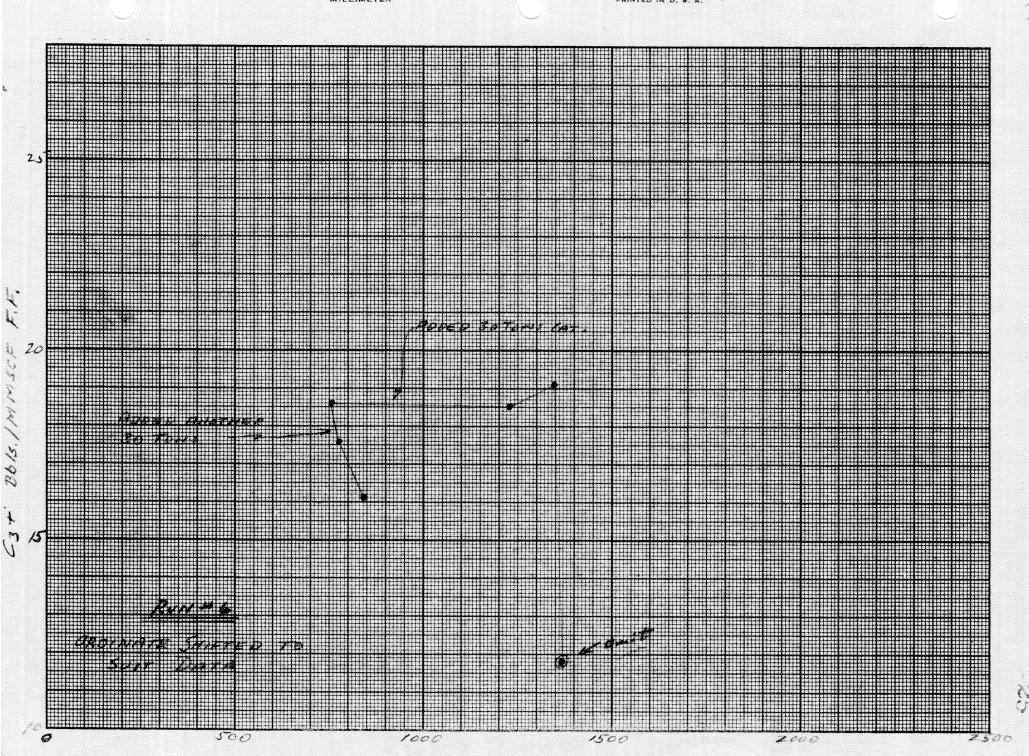
It will be evident from the above description of the method of daily data calculations and from consideration of the many
factors involved, that the possibility of inconsistencies in the
Brownsville data from day to day and run to run is great. Therefore,

^{*} The weight percentages in the Run Summaries do not add up to 100% because the CO₂ and CH₄ in the fresh feed have been subtracted from the CO₂ and CH₄ respectively in the reactor effluent. The total products shown are the 'make' products they are all expressed as wgt.% of the fresh feed synthesis gas and not of the pure H₂ plus CO input.

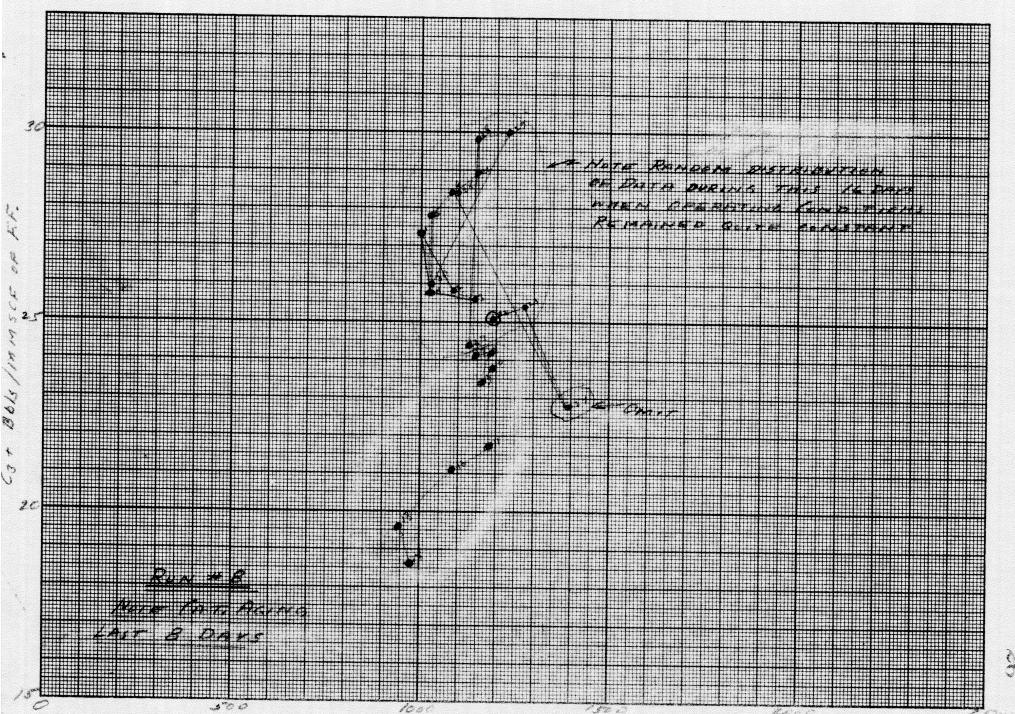
in order to explore the relative reliability of the point to point data and to examine the effect, if any, of changes in operations made during the runs a plot was made of all the Brownsville data run by run in the following separate graphs. These graphs are plots of space velocity (SCF of F.F./Hr/Cu.Ft. of fluidized catalyst) against yield of C₃+ expressed in Bbls/MMSCF of fresh feed. They are numbered Run 5, Run 6, Run 8 etc. On each plot the points have been joined in chronological order with a ring around the first day's operation. Examination of these individual graphs and Table II in the appendix, which lists the plotted data as well as the operating conditions used, discloses the following:

- 1. Some points, but not too many, are obviously way out of line and can legitimately be discarded if necessary to avoid confusion. (Note: A great many points which were out of line in the original rough graphs were corrected when on checking back to the original stock room calculations, it was found that errors in arithmetic or errors in transcription or errors of ommission were responsible).
- 2. In the first part of Runs 8, 13 & 14 all of Runs 11, 12 and the last part of Run 17 there is obviously a wide random distribution of the data during periods when operating conditions were deliberately being maintained as constant as possible in the plant. This random distribution shows that the plant data cannot be depended upon too much, from point to point at least, to show anything but large effects of changes in operating variables. We must look instead to the mass of the data to show trends
- 3. The effect of catalyst deactivation is quite pronounced in many runs. Note especially the sharp drop after the first or second day in Runs 5, 11, 12, 14 & 15. Note also the

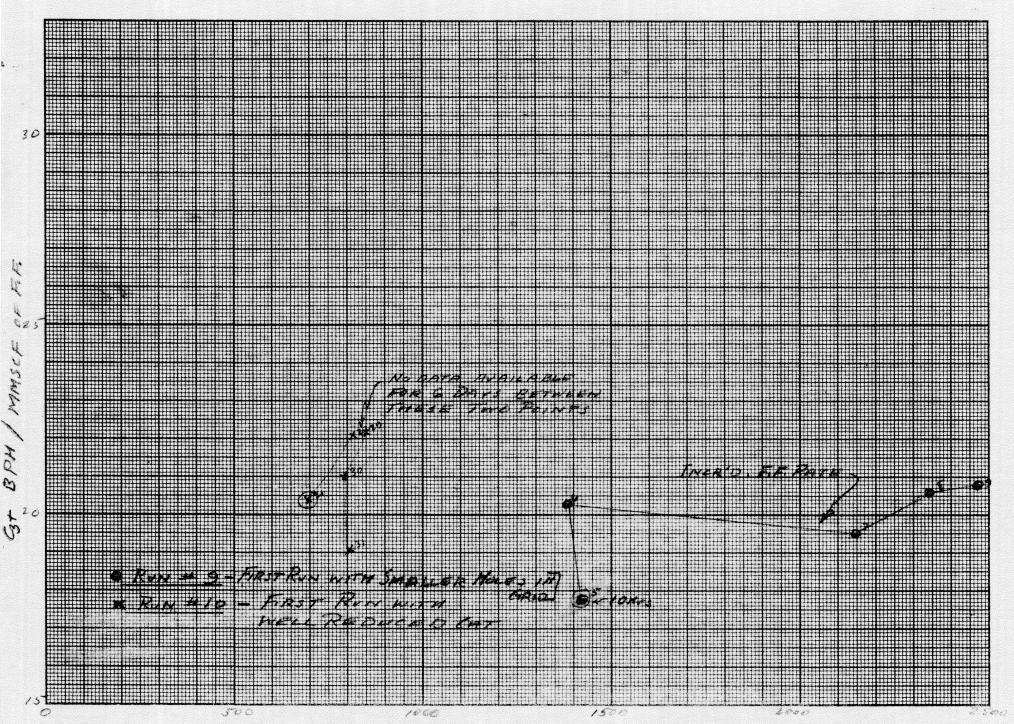




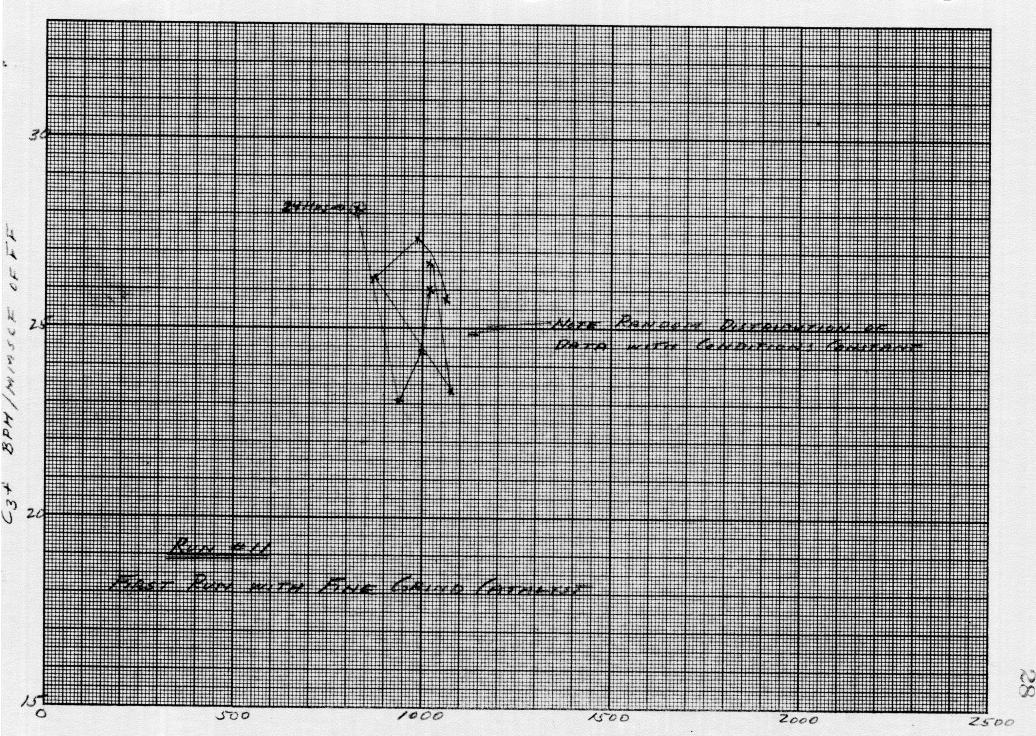
SP. VEL. V/MR/V



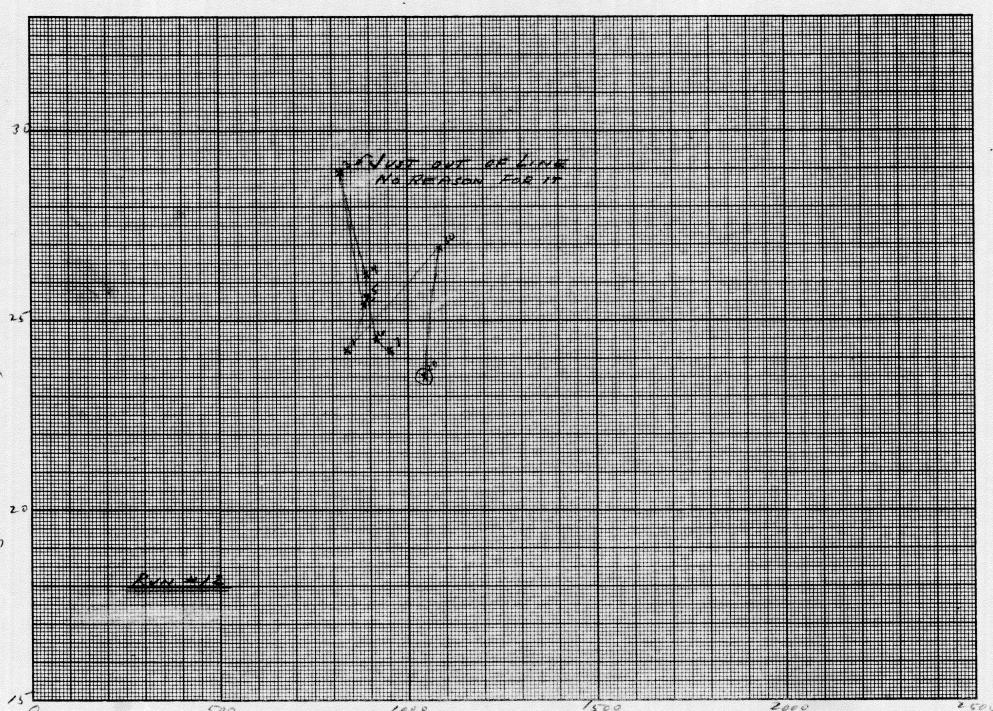
SPACE DIEL. VINAL



Sp. Vec. V/MR/V

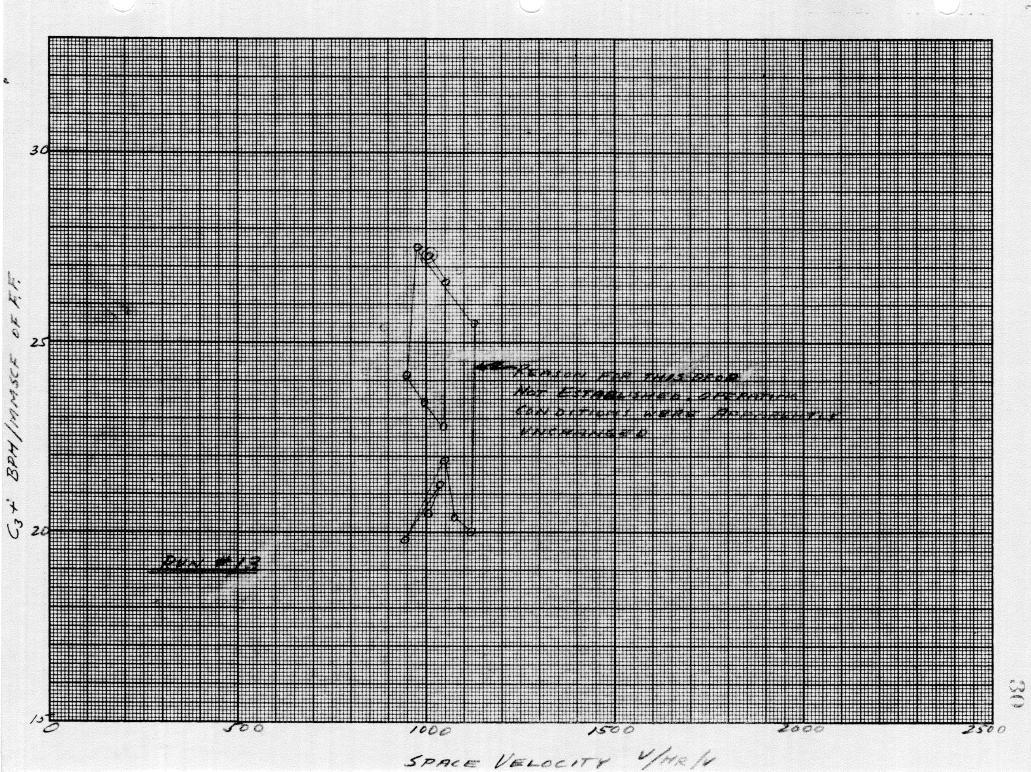


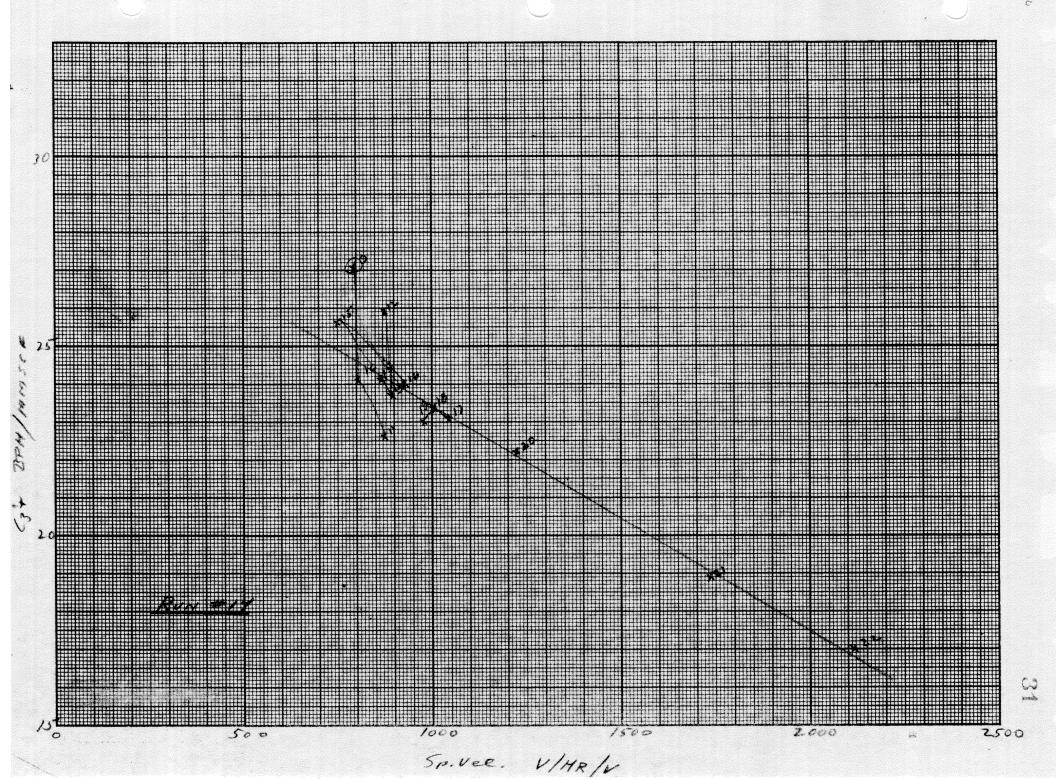
SPACE VELOCITY V/HR/V

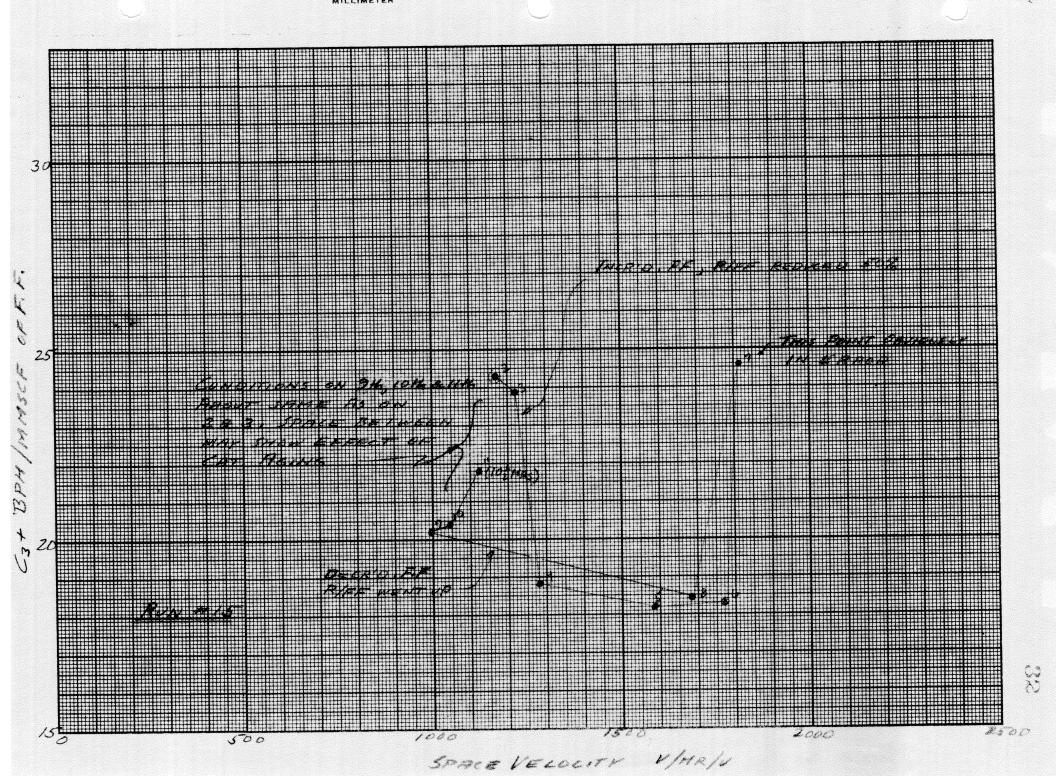


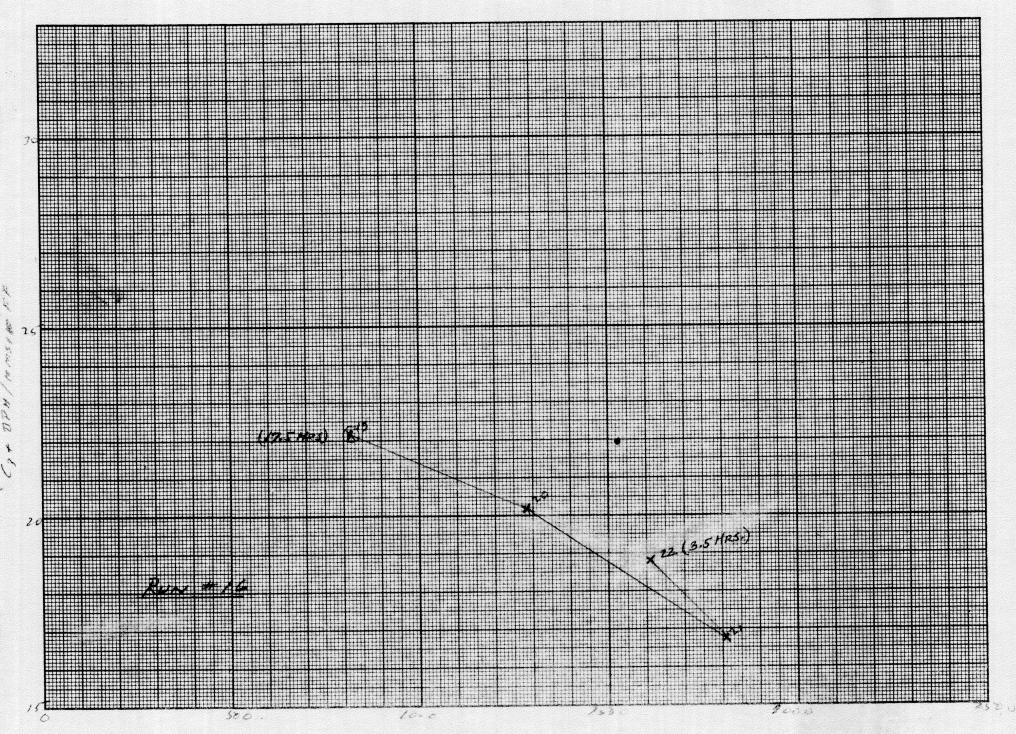
professor.

SPACE VEL. YMR /V



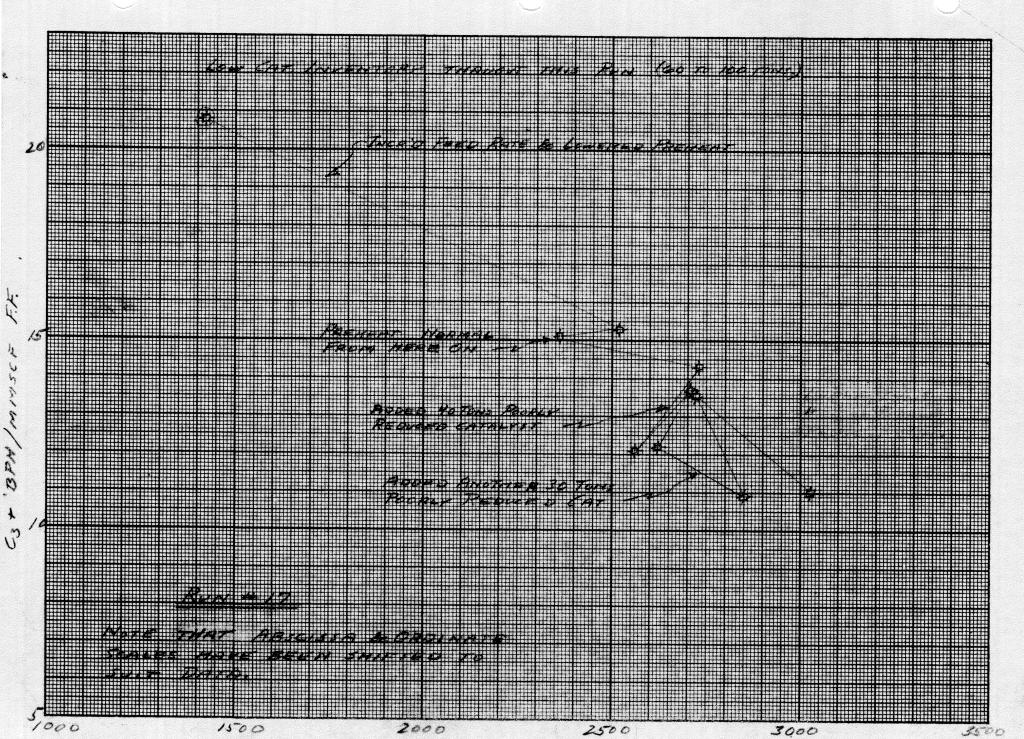






4.5

Sp. Vec



SPACE VEL. V/HR/V

sharp drop which occurred during the later part of Runs 13 & 8. The drop in Run 13 may have been caused by a sharp change in mass spectrometer operations (could not be pinned down as such however) but that in Run #8 it is too consistent, in spite of maintaining constant operating conditions to be anything but catalyst deactivation.

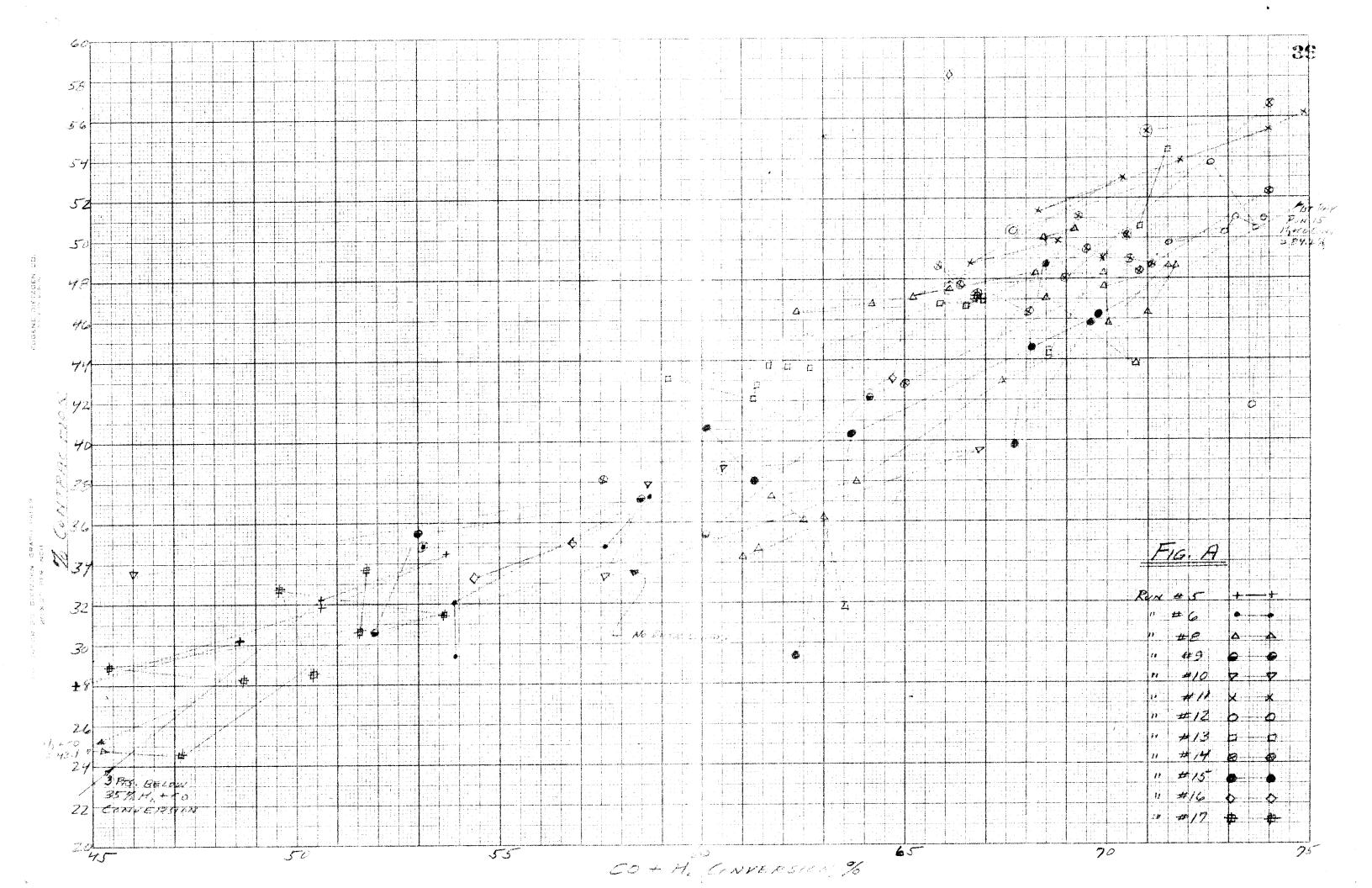
4. In runs 14, 16 and 17 there is an apparent pronounced effect of space velocity. This is not solely the effect of space velocity however because space velocity was always increased as the run progressed and therefore the effect of catalyst aging is superimposed.

In summary, these individual run graphs indicate that some rapid deactivation takes place the first day. The graphs also indicate that if the runs were longer, some further deactivation would occur as it did in run 8 where the catalyst apparently went to pot completely. Space velocity has an apparent effect but the magnitude of its effect is clouded by the superimposed. catalyst aging and possibly the effect of automatically reducing the combined feed inlet temperature when total feed rate is increased as discussed above.

The point to point data of each individual run are not accurate enough by themselves to show the effect, if any, of other operating variables.

Fig. A H_2+CO Conv. vs % Contraction

In order to get some idea of the consistency of the runs with each other, we have plotted <u>all</u> of the data on the following Fig. A where $\rm H_2$ + CO conversion is plotted against % contraction. This relationship is almost mathematical and if the methods of measurement and calculations were correct, the points should all



fall on a single line.

It will be noted that altho the deviation is fairly great, all but a few points fall within a band which should be good enough for our purpose. The points that are outside that band are all unreliable and not real effects of operating conditions. These points cannot be discarded yet however because altho they may be in error so far as % contr. or H_2+CO conv. is concerned they may be correct when other factors are correlated.

Fig. B $\rm H_2+CO$ Conversion vs $\rm C_3+$ Yields/MMSCF of Feed

Another plot used for the purpose of establishing the consistency of the data with each other is Fig. B where we have shown $\rm H_2+CO$ conversion vs the yield of $\rm C_3+$ in Bbls/MMSCF of F.F. for all points. Here the scattering is even greater than in Fig. A because the $\rm C_3+$ yield is numerically dependent on the accuracy of many more values than the % contraction.

All of this merely reiterates that we must draw conclusions from the mass of the data and not try to deduce too much from a point to point comparison.

It will be noted from this Fig. B and also from Fig.1A that the C_3 + yields reported for Run #5 are way out of line. As a matter of fact these yields are stoichoimetrically impossible for the H_2 and CO conversions reported. This particular run was not worked up by the usual stock department procedure and the complete recalculation of the original data might have disclosed the cause of the discrepancy. However, this expenditure of time was not warranted and we have elected to ignore Run $\frac{1}{10}$ 5 instead.

All the data in Run #8 are suspect. This is unfortunate since this is the longest run. An examination of the data discloses that the mass spectrometer analyses on the reactor fresh feed

(V-301 effluent) was in error. The $\rm H_2/CO$ ratio averaged about 1.7 instead of the 1.8 in other runs before and after Run %8. This has such a great effect on the correlations which use $\rm H_2$ conversion or $\rm H_2+CO$ conversion as a base that Run #8 must be ignored for this purpose. The error however was consistent throughout the run and therefore the run may be used to evaluate the effect of operating variables & catalyst age.

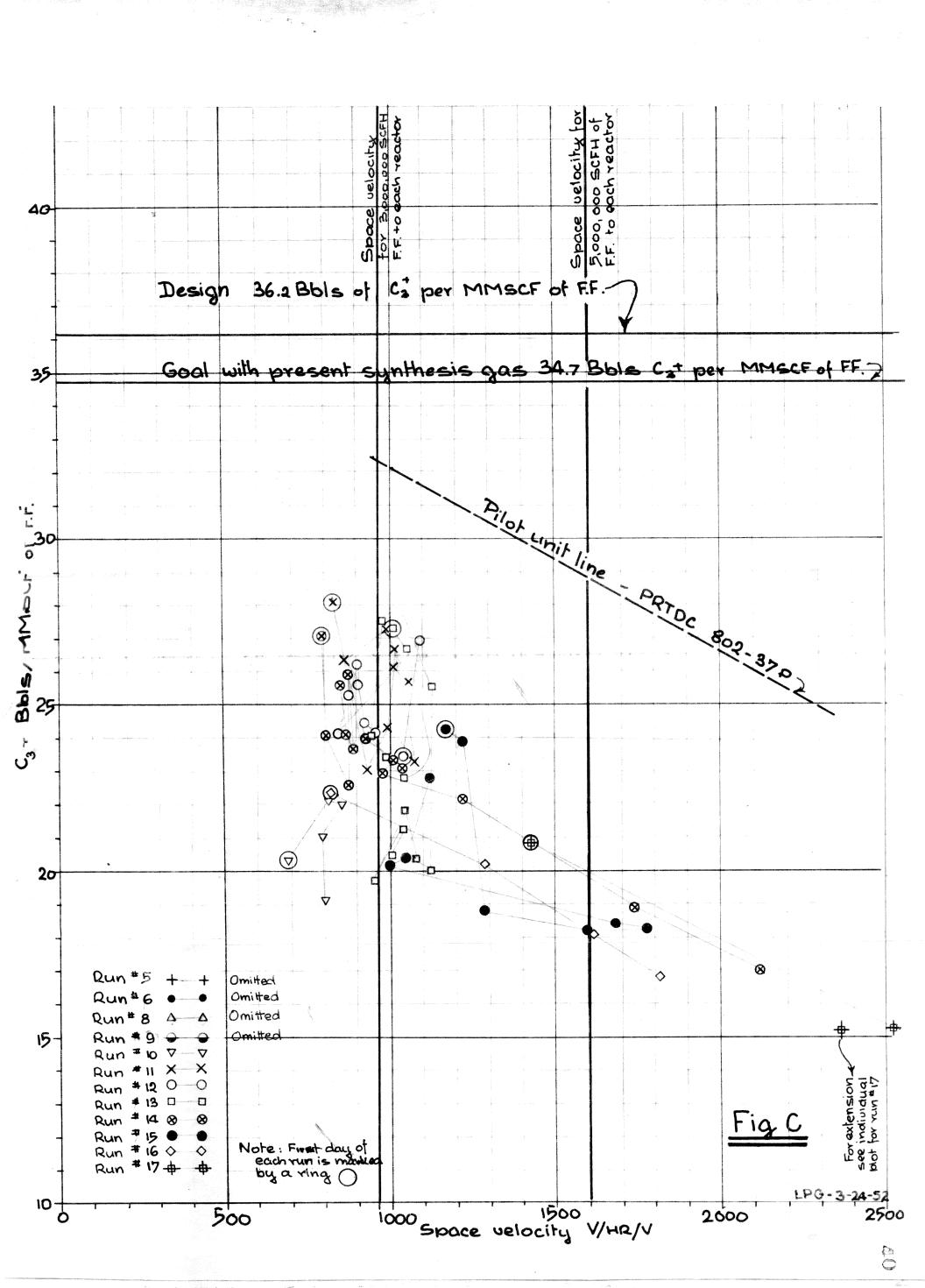
We also note from Fig. B that the following points should be treated with suspicion.

- 1. The last three points in Run #9
- 2. The last two points in Run #10.

Fig. C Space Velocity vs C_3 + Yields

The following Fig. C is a plot of space velocity vs C₃+ yields in Bbls. per million SCF of synthesis gas feed for all Brownsville Runs from Run #10 on. This plot was made up to see what these recent and more reliable runs, shown on the individual plots above, would look like if superimposed on each other. In general it will be noted that they check each other reasonably well. From examination of this plot and Table II of the appendix there is no clear cut evidence that difference in operating conditions from run to run made any consistent change in results. We were particularly surprised to find no large consistent effect of differences in degree of catalyst reduction. In general it can be concluded that catalyst aging and space velocity, as modified by catalyst aging and perhaps preheat, are the only two factors that make large differences in results.

One might argue that the fine grind catalyst, used in Run #10 and also in Runs #6 & 9 (see Fig. 1) were not as good as those ob-



tained in the succeeding runs. (The Run #8 data are unreliable as discussed above)

On this same Fig. C we have shown for comparison the effect of space velocity in pilot unit operations as reported in Table #5, p. 21 of P.R. TDC 802 - 37 P. We have also shown for comparison horizontal lines representing the design goal of C_3 + yields both as originally proposed and as adjusted for the quality of the present synthesis gas as follows:

Effect of Synthesis Gas Quality on Yields

From the Case 6 design breakdown of the reactor effluent previously submitted, it was found that the total C_3 + yield (incluing WSC) per MM SCFH of syn. gas feed predicted was 36.22 Bbls. This was for a synthesis gas that was quite different from that now being produced at Brownsville.

Shown in the following Table C are radomly selected analyses of the synthesis gas in various runs compared with that predicted in design.

On Nov. 29, 1951, the end of Run 13, the practice of adding about 15% steam to the natural gas feed to the generator in order to reduce soot formation was started. It will be noted that this made a perceptible difference in the syn. gas composition, H_2/CO ratio and H_2+CO purity as shown below.

		Average Runs %13 & %16 Incl.	Average Runs $\frac{7}{16}$ & $\frac{4}{12}$ Incl.*
	Design	with Steam	No Steam
Syn. Gas Rate MMSCFH	5.0	3.18	3.10
Mol % Dry: CO	33.23	31.73	33.27
H ₂	61.35	59.78	60.15
co ₂	1.31	3.90	2.97
N ₂ +A CH,	3.01	2.12	1.94
$CH_{I_{\mathbf{b}}}$	1.10	2.47	1.80
н ₂ 7со	1.845	1.88	1.81
(Ĥ ₂ +CO)%	94.2	91.5	93.42

^{*} Run %8 Excluded

TABLE C

COMPARISON OF FRESH FEED COMPOSITIONS

Run # Date Syn Gas Rate	Design	2/22/	#16 /52 2/29/52	$\frac{2}{2/3}$	1 <u>5</u> 52 2/10/52	1/11/5	#14 52 1/21/5	2 1/22/52	#1 12/23/5	.3 51 12/30/51	
H2 61 CO2 N2 CH4 H2O	3.1 33.2 1.1 61.3 1.3 1.3 3.0 3.0	559.70 1 4.63 1 1.34 0 2.76	3.30 .63 32.34 58.98 3.87 1.60 2.58	2.856 .69 32.20 59.65 3.71 1.31 2.44	6 2.663 .69 31.43 60.14 4.01 1.11 2.62	2.646 .73 31.74 59.58 3.90 1.45 2.60	3.354 .68 31.62 60.23 3.63 1.68 2.16	3.771 .65 31.69 60.08 3.77 1.38 2.43	2.999 .73 32.08 59.82 3.71 1.43 2.23	2.990 .70 31.56 59.87 3.84 1.61 2.42	
H ₂ /CO 1.8 (H ₂ +CO)%94.2	B45 2	1.93 90.62	1.81 91.32	1.85 91.85	1.91 91.57	1.88 91.32	1.90 91.85	1.90 91.77	1.86 91.90	1.90 91.43	
Run # Date Syn. Gas Rate A CO H2 CO2 N2 CH4	#1 11/30 3.03 .69 33.48 59.77 3.06 .93 2.07	2 12/5 2.81 .68 32.14 59.86 3.88 1.41 2.03	3.00 3 .65 32.88 34 60.89 59 2.87 2 1.04 1	.05 2 .71 .03 32 .97 60 .69 2	#10 /21 7/29 2.51 2.67 .66 .61 2.66 34.42 0.93 60.56 2.65 2.49 1.08 .88 2.02 1.04	#9 6/6 6 2.81 .67 33.313 60.936 2.74 1.07 1.28	7/8 4.77 3.93 3.93 3.93 3.07 2. 1.18 1.	07 3.46 74 .79 70 34.95 40 58.05 94 2.58	.74 33.30 32. 58.63 59. 3.12 3. 1.88 1.	06 73 51	
H2/CO H2+CO)%	1.79 93.25	1.86 92.00			1.87 1.76 3.59 49.80	1.83 94.249		68 1.66 10 93.00	1.76 1. 91.93 92.		

If it is assumed that steam will continue to be added to the generator to control soot formation the syn. gas will contain only 90.32 of $\rm H_2+CO$ having an $\rm H_2/CO$ ratio of 1.845, assuming the extra hydrogen is not beneficial, thus

% CO = 31.73 x 1.845 = 58.6 = % Effective
$$H_2$$
 31.73 + 58.6 = 90.3% H_2 +CO

It follows therefore that the anticipated yield of $C_3+/MMSCF$ of synthesis gas must be reduced to 90.3/94.2 = 95.9% of that originally anticipated due to inferior synthesis gas composition. This amounts to 34.73 Bbls/MMSCF of syn. gas.

Similarly if steam is not added the syn. gas will contain 92.1% of $\rm H_2+CO$ of 1.845 $\rm H_2/CO$ ratio and the anticipated yield must be reduced to $\frac{92.1}{94.2}$ = 97.82 of that predicted (35.42 Bbls/MMSCF FF). Roughly therefore some 3 or 4 percentage points of the discrepancy between anticipated and actual yields of $\rm C_3+/MMSCF$ of syn. gas at Brownsville can be accounted for by differences in the synthesis gas compositions.

Incidentally, it is noted that the Nitrogen plus Argon in the Brownsville gas is consistently less than that predicted. This is because it was assumed in design that the nat. gas would contain 5% N₂ whereas this has been running consistently at 0.3 to 0.5%.

On the above Fig. C we have also shown vertical lines representing the space velocity required per reactor for the present production of syn. gas (abt. 3,000,000 SCFH/Reactor) and that required ultimately for the design rate of 5,000,000 SCFH for each reactor. These values were developed as follows:

Space Velocity:

The design space velocity calculated from the data submitted in the Case 6 design specifications is 2520 V/HR/V. These specs. show a horizontal reactor however and later, if my memory is correct, as changes were made in the design 2200 was consistently referred to as the design space velocity.

In actual practice the bed density has been found to run much higher than the 100 %/c.f. predicted in design. It actually is nearer 150 at Brownsville and about 140 at Montebello.

The present reactors can probably accomodate a bed beight of about 17 ft. which, with an average free area between the tubes of 187 sq. ft. amounts to a total catalyst volume of 3180 C.F. or a space velocity with the design rate of 5,000,000 SCFI of Syn. gas of 1570 V/HR/V. At 150 // c.f. catalyst density, this bed height requires about 240 tons of catalyst which is the amount now normally used for a full load.

Although the ultimate goal is to operate at 1600 space velocity the present synthesis gas output with one oxygen plant in service is only 3 million SCF of syn. gas instead of the design value of 5 million. At this actual rate therefore the space velocity required per reactor is about 960 SCF of Gas/Hr/CF of catalyst bed.

Discussion of Fig.C.

It will be noted from Fig. C that there is a long way to go before Brownsville results will equal those of the pilot units and, what is more important, the pilot units have almost as far again to go to reach the desired goal at the ultimately desired space velocity of 1600.

It will be noted also from Fig. C that if a line is drawn thru the best of the Brownsville high space velocity data and extended to zero space velocity the yields would still not be as good as those obtained on the pilot units at 960 space velocity. Fig. D Space Velocity vs C_2 + Yield for Averaged Data

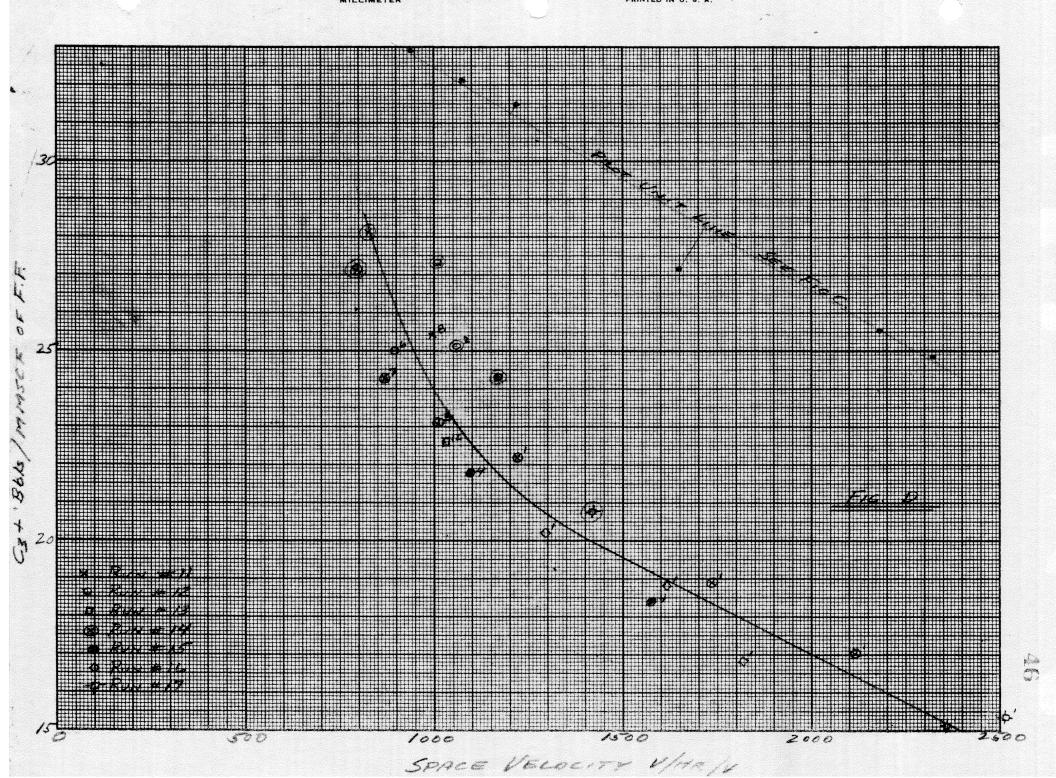
In plotting the points on Fig. C it was observed that in many runs the yields in the first day's operation were substantially higher than those for succeeding days at the same operating conditions. This effect was obscured in Fig. C because of the random distribution of the points representing succeeding days. Furthermore in nearly all runs the space velocity was increased only after several day's operation (time required to get the second oxygen plant in service). We therefore averaged the data for periods where operating conditions were kept constant leaving the first day's operation separate and obtained the following plot Fig. D. On this plot we have shown besides each point the number of days represented by the point and we have put a ring around each pt. representing the first day.

On this graph the abscissa should have been made longer to permit showing all the points at very high space velocities in Run 17. These however are simply an extension of the graph as can be seen thru reference to the individual plot for Run 17.

This is one of the most important graphs in this report. It correlates well all the recent run data, Runs ll thru 17, excepting only the first day of Run 16 which covered a 17 1/2 Hr. period but was obviously out of line.

The line thru the Brownsville data shows the combined effect of space velocity and catalyst deactivation. *

*To a certain extent the lower end of the curve, beyond 1000 to 1200 $\mathrm{Sp}_{\bullet}\mathrm{vel}_{\bullet}$ also may be influenced by total feed inlet temperature but as shown before this temperature actually doesn't change very much (530° F to 415° F) and therefore its effect on this graph should not be



The pilot unit line on the other hand since it was based on the first few days operation of several different runs (see Table V p. 21 PR 37P) shows the effect predominantly of space velocity alone.

From the relative position of the first day points the Brownsville catalyst deactivates very rapidly at the very beginning of each run. It will be noted also that if the steep portion of the curve were to be extrapolated to, say, the first hour's operation, the yields would very likely be right up there with those obtained at the same space velocity on the pilot units.

Moving down The Brownsville line to the points representing the largest number of days (at about 1000 space velocity) it will be noted that the slope is still very steep until we get to space velocity of about 1200 beyond which the line parallels the true effect of space velocity on the pilot units. This steep intermediate portion of the curve, where most of the data fall, indicates that poisons are entering the system with the gases. A small increase in space velocity increases the rate of poison entering the system so that catalyst activity and yields drop off rapidly. Beyond that, the catalyst is virtually dead anyway and the effect is almost purely one of space velocity alone.

This graph and the previous Fig. C also show that if the initial degree of reduction of the catalyst and catalyst carbiding have any effect at all, the effects are well within the accuracy of the data.

Also since Run 17 made predominantly at very high space velocity because of very low catalyst bed, falls on the curve with other runs of deeper bed where the space velocity was high only

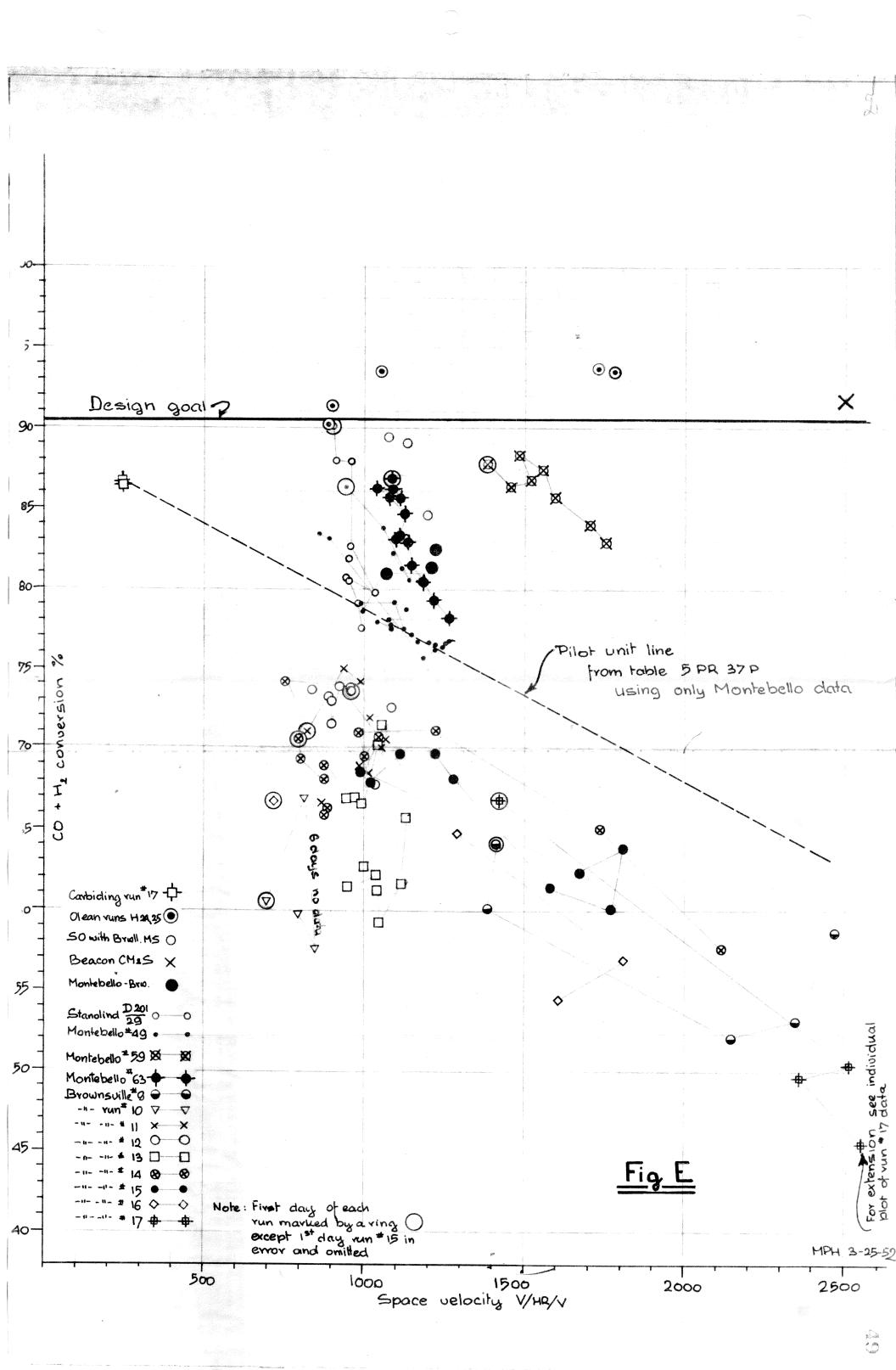
because of high thruput instead, Fig. D indicates that at this low level of activity, at least, bed height and therefore probably catalyst contacting efficiency have little to do with the poor conversions, at least not at these low levels of activity.

Fig. E. Space Velocity vs H_2 + CO Conversion:

Fig. E is a plot similar to Fig. C except that H_2+CO conversion is the ordinate instead of the C_3+ yields. This graph like Fig. 1, permits a direct comparison of Brownsville and Pilot Unit data and also shows the H_2+CO conversions obtained at Brownsville at very low space velocity during carbiding operations.

The solid line at 90.4% H₂+CO conversion represents the design goal. It will be noted that the HRI H series runs H-24 & H-25 (0) which it will be remembered were made on an ll.5" i.d. reactor with spent CM&S catalyst all fall on or above the goal. The Beacon data (x) with the same catalyst and with Brownsville Mill Scale catalyst also all fall along this line but only one run (7027) was available at low enough space velocity to be shown on this plot. The others were all made at space velocities of 5,000 to 15,000 V/HR/V.

Referring to the Stanolind and Pilot unit data it will be noted that the deactivation with time when using either mill scale of Allan Wood Catalyst at a given space velocity is quite pronounced though much slower than which takes place at Brownsville as shown in Fig. D above. The deactivation in the Stanolind Run #25 with Allan Wood catalyst (0-0) was somewhat more rapid and more consistent than that obtained with the same catalyst at Montebello.



The Montebello Run #59 (\boxtimes) with spent CM&S catalyst, on the other hand shows a much slower rate of deactivation, if any. A line thru the points for this run is almost parallel to the dotted line showing the effect of space velocity in the Pilot units. This line is the same as that shown on Fig. D above.

It is quite interesting to note that the Brownsville carbiding runs fall exactly on an extrapolation of this space velocity line. This indicates that during carbiding operations when operating with a fresh catalyst and relatively high hydrogen partial pressure The Brownsville Reactor acts just about the same as the Pilot unit reactor. On the other hand it is evident, of course, that this space velocity line does not go thru the best pilot unit results but perhaps represents instead average results with a catalyst that has become somewhat stabilized.

Except for Run #10 and the latter part of Run #13, the plant scale data, at a lower conversion level show about the same effect of space velocity on H_2+CO conversion as does the pilot unit line. As we have seen from Fig. D however this is not the whole story.

Discussion of Figs. 1, 1A & 1B

Figs. 1, 1A & 1B were included at the beginning of the report because they show quite strikingly the differences between Brownsville, pilot unit and laboratory results. They were among the first plots made and all the Brownsville data were included even though we now know that some of these data, notably Runs 5 & 8, are in error. These form the first of a series of graphs where

Brownsville and Pilot unit results are compared with base lines established in the 1947 correlations (EDG Report No.1) from an extremely large number of data obtained at several laboratories.

Since the points are joined in chronological order with rings around the first day of each run, the drift downward to lower $\rm H_2$ conversion with time is quite evident. It is also clear that the drift on Pilot units is of the same order though slower than that experienced at Brownsville.

With reference to the Pilot unit results it will be noted that the Montebello data fall about where they should with respect to the base lines because the Montebello gas has an $\rm H_2/CO$ ratio of only 1.6 compared to 1.8 for Brownsville and 2.0 for the lab. and the Stanolind Pilot unit runs. This, incidentally, may explain the relatively high selectives for a given H2+CO conversion consistently reported by Montebello because conversion is always expressed as % of $\rm H_2+CO$ and Montebello has less $\rm H_2$ to start with.

On the other hand, The Allan Wood catalyst at Montebello apparently results in better selectivity than that obtained with Mill Scale.

It would appear from Figs. 1, 1A & 1B that the problem of raising Brownsville conversions is the same <u>kind</u> of problem as that of raising Pilot unit results to those of the smaller units.

Although a little difficult to see on Fig. 1 & 1A, it will be noted that the drift in activity with time of the spent CM&S catalyst on the Montebello unit was not as great as with Mill Scale.

It was still there however and even with this apparently more stable catalyst periodic reactivations would be required to maintain activity. Such periodic reductions were used on the "1000 Hr. demonstration Run 19-6 with <u>fresh</u> CM&S catalyst at Trenton.

In general it will be observed from Figs. 1, 1A & 1B that all the data, comm'l., pilot unit and lab. fall pretty close to the base lines established in 1947. This is pertinent to succeeding discussions.

$\rm H_2$ vs CO Conversion - Fig. 2

The following Fig. 2 is a plot of H_2 vs CO Conversion for all the data used in Figs. 1, lA & lB except for a few points and Runs 5 & 8 which were discarded above. On this plot the CO Conversions for Run 10 are obviously in error and should be ignored. The nomenclature on this and succeeding plots is the same as in Fig. 1.

The solid lines on this graph are those that were established in the 1947 correlation (EDG Report #1) and the dotted line is an interpolation for Brownsville feed showing where Brownsville data ought to fall.

I strongly believe that this graph is the key to the entire problem. If we can explain why the Brownsville and pilot unit data on Allan Wood and Mill Scale fall below the lines where they should, that is, why the CO disappearance for a given H₂ disappearance is less than it should be, we shall probably have a clear understanding not only of the poor Brownsville yields but also of the reason why the pilot units at Stanolind and Montebello

Do over

BROWNSPALLE 1. 545 CLEAN RYNG H-24 & N-25 1.95 BEST KUNS . CHES S CHEMENT * BEGGEN PLANT ON CLOSE SCATEL PS 2.00 MONTERED RON 45 1.63 Stance in Rene-201-22 1193 MONTEBELLO RUNS ON MALL SCALE 1.62 STANGERMO ROMO ON MILL SCALE
MONTERALLO RUNGON GO ON MILL SCALE
MONTERALLO RUNGO ON MILL SCALE 1.00 2.00 1.62. BROWNSVIALE RUN # 60 1.80 BROWNSVILLE RUN S 1.70 (MASS STAKE BROWNSVILLE RUN 49 OPominos 72/20 BUN 410 1.82 RUN # 11 1.82 1.85 RUN # 3 1.87 RUN # 14 1.87 RUN WKG 1.84 RIA #/ 1.84 MONTE BELLO RUN #50 SPENT CMAS SONVERSION S 0 V TO HE CONVERSION

do not check the results obtained on the smaller Beacon units or in the H series runs on CM&S catalyst in the $11 1/2^n$ reactor at Olean. Incidentally, we shall also probably explain the as yet unexplained reason why the HRI 14 and 15 series runs in the $4 1/2^n$ reactor at Olean and the Trenton 19-6 Run also fall below the base lines which were established predominantly by Beacon and HRI H series data in 1947 (see EDG Report #1).

It will be noted that the design point falls on the line. This, plus the fact that Beacon has exceeded the desired H₂ and CO conversions at very high space velocities (up to 15,000) indicates that the goal can be attained.

Before attempting to explain the difficulty, we shall introduce the next plot.

H_2 Conv. vs CO to CO_2 and H_2 to H_2O - Fig. 3

The following Fig. 3 is a plot of all the data, except that previously discarded, showing the relationships between H_2 Conversion and the amount of CO. in the fresh feed that went to CO_2 and also the amount of H_2 in the fresh feed that went to H_2O_4 .

The solid lines in the ${\rm CO}_2$ plot are, as before, those which were established in our correlation of 1947. The corresponding lines in the ${\rm H}_2{\rm O}$ plot are shown dotted because they not only correlated the data well in 1947 but they are actually calculated by oxygen balance assuming that the solid lines in the ${\rm CO}_2$ plot on Fig. 3 and the solid lines in ${\rm H}_2$ vs ${\rm CO}$ plot on Fig. 2 are correct. The fact that the 1947 data fell on these calculated lines is therefore a check on the validity of the lines of ${\rm CO}$ conv. and ${\rm CO}$ to ${\rm CO}_2$ which were established by the data

themselves.

The fact that the present data falls on this same $\rm H_2O$ line while it falls below the $\rm CO_2$ line in Fig. 3 and the CO line in Fig. 2 shows that the speed of the water gas shift reaction with respect to that of the Fischer-Tropsch reaction is less at Brownsville, Montebello and Stanolind on Mill Scale than it was in the HRI H-Series reactor and in Beacon's runs on CM&S catalyst which runs formed the basis predominantly for the 1947 correlation.

Mill Scale may have exactly the same effect on the Beacon reactors but there, where conversions are so high, the effect is too small to detect.

Discussion

As discussed extensively in the correlation report of 1947, the water gas shift and the Fischer-Tropsch reaction go on simultaneously both using up CO as the reactants proceed up the reactor thus:

- 1) $2H_2 + CO \longrightarrow (=CH_2) + H_2O$ Fischer-Tropsch
- 2) $CO + H_2O \longrightarrow H_2 + CO_2$ Water-Gas Shift

Since the feed to the reactor, total feed in this case, is relatively dry it is far removed from equilibrium and should proceed quite rapidly as soon as water is made by the Fischer-Tropsch reaction. Normally as these two reactions proceed simultaneously the CO disappears rapidly, part of it going quickly to $\rm CO_2$ until the concentration of $\rm CO_2$ has been built up and the CO concentration reduced to the point where the water gas shift reverses thereby converting the $\rm CO_2$ back to CO by reaction with the remain-

ing H₂ thus:

$$CO + H_2O \leftarrow H_2 + CO_2$$

This reversal of the water gas shift is of course what accounts for the humped shape of the $\rm CO_2$ base line in Fig. 3. When the shift reaction is occurring at normal speed this reversal must take place if one wishes to exceed an $\rm H_2$ conversion of 50 to 60% (for $\rm H_2/CO$ ratio feed of 2) otherwise the reactions would stop simply because practically all the CO has been used up. If the water gas shift is slow however and doesn't use up the CO sc rapidly, the $\rm H_2$ conversion can exceed this figure of 50 to 60% materially but never can it get up to the level of conversion we want, of the water gas shift is occurring at all, because the water gas shift inevitable uses up part of the CO and if it is too slow to reach a point of reversal it never gives the CO back.

The 1947 Correlation shows that about 30% of the CO is initially consumed by the water gas shift before it's reversal occurs. At that time most of the rest of the CO has been consumed simultaneously by the F-T reaction.

Consider for example the case for $50\%~{\rm H_2}$ conversion. From the base lines for 2/1 gas we get the following:

H_2	Conv.	50%
CO	Conv.	92.5%
CO	to CH ₂	42%
CO	to CO_2	30%
H ₂	to H ₂ O	16%

The equations can be written as follows:

W-G Shift 30 CO + 30
$$H_2$$
0 ____ 30 H_2 + 30 CO_2
F-T 70 CO + 200 H_2 ____ 42 CH_2 + 8.5 CO + 70 H_2
+ 62 H_2 0 + 19.5 CH_2 .66 *

Overall 100 CO + 200
$$H_2$$
 ___ 42 CH_2 + 70 CO_2 + 32 $H_2O+100H_2$ + 8.5 CO + 19.5 $CH_{2.66}$

The water gas shift ratio at this point is

 $\frac{(H_2) (CO_2)}{(H_2O) (CO)}$ = 11 and shortly thereafter the water gas

shift starts reversing to convert ${\rm CO_2}$ back to ${\rm CO_{ullet}}$

If on the other hand an average line is drawn thru the Brownsville data on Figs. 1, 2 & 3 however the same approach shows that at 50% H₂ conversion the amount of CO used up by the water gas shift is only 22.5% of the total, some 25% less than normal, thus:

For Brownsville data H_2/CO in FF = 1.84

 H_2 Conv. = 50%

CO Conv. = 77.5%

CO to $CO_2 = 22.5$

CO to $CH_2 = 42$

 H_2 to $H_20 = 17.5$

 $184 \text{ H}_2 + 100 \text{ CO} = 92 \text{ H}_2 + 42 \text{ CH}_2 + 22.5 \text{ CO} + 22.5 \text{ CO}_2 + 32.2 \text{ H}_2\text{O} + 13 \text{ CH}_3 \frac{*}{.5}$

Water Gas Shift ratio = $\frac{(92)(22.5)}{(22.5)(72.2)} = 2.9$

*Found by C & H Balance

Precisely the same effect is evident at 80% H₂ conversion level from a similar comparison of the 1947 correlation base lines and an average line thru the pilot unit data on Mill Scale. It is suspected that the same <u>kind</u> of thing is true also in the Beacon Runs with Mill Scale but in this case, at the high levels of conversion attained, the differences are too small to notice.

This characteristic of Mill Scale catalyst (slow water-gas shift) and it's relationship to the fact that lower conversions are obtained at Brownsville than on Pilot units could stand some further thought and development. It is felt however that the facts are already sufficiently developed to permit drawing conclusions and therefore, in the interest of getting this report out, indulgence in further discussion will be left until later.

Conclusions

Tropsch catalyst. It changes activity \underline{up} or \underline{down} very rapidly depending upon the composition of the gases surrounding it and in addition it does not promote the water-gas shift as much as it should. A freshly reduced Mill Scale catalyst gives excellent results in the Beacon reactors where \underline{pure} \underline{H}_2 and \underline{CO} and stable conditions are used, and altho there is even there a slight deactivation with time aggravated by the fact that the composition of the recycle gas is gradually changing, the catalyst appears quite stable.

On the pilot units, on the other hand, the synthesis gas is not so pure (it contains ${\rm CO_2}$, ${\rm CH_4}$ and some ${\rm H_2O}$) the bed conditions are not as stable and as time goes on the catalyst continuously readjusts itself to the surrounding gases chemically and in effectiveness and deactivates much more rapidly than on the purer laboratory units. This again is aggravated by recycling and constantly changing recycle composition. This change occurs even though in the very beginning, say in the first hour, of the pilot unit runs the catalyst activity and results obtained are fully as good as those obtained under more ideal conditions on the laboratory units.

The same thing happens in the plant but to a greater extent. There operating conditions are still more unstable and the situation is apparently further aggravated by the fact that the gases contain poisons as well as the ${\rm CO_2}$, ${\rm CH_4}$ and ${\rm H_2O}$ also found in the pilot unit.*

The stability of the operating conditions is quite important. In the F-T reaction there tries to be a dynamic

^{*} Poisons may come from the salt water used to wash the fresh feed, acid water used to wash the recycle or highly chlorinated city water used as trim on both.

equilibrium between the solid and gaseous phases. In recycling operations any momentary change in operating conditions and therefore conversions causes a change in recycle composition which in turn affects the solid phase and its activity thereby upsetting the equilibrium to create a vicious circle.

It appears therefore that the low conversions at Browns-ville are caused by exactly the same factors that prevent the pilot units from maintaining continuously the same results as obtained on the laboratory units which are operating under more nearly perfect conditions.

Since Mill Scale catalyst deactivates rapidly when certain changes occur in its surrounding atmosphere it, conversely, reactivates rapidly when the atmosphere is favorable. That explains why a catalyst sample drawn from the Brownsville reactor at say 12 or 96 hrs. operation and which was giving poor results at Brownsville gave almost immediately good results under Beacon's condition.

The difference in results obtained at Beacon on the samples drawn from the Brownsville reactor at the 0, 12 & 96th hr. of Run #11 are quite significant in this respect. In spite of widely different chemical compositions the results obtained at Beacon on the three samples were all very high. The small differences obtained at Beacon were probably due to permanent poisoning which had occurred in even that very short period.

The X-Ray analyses of these three samples together with the relative yields were as follows. They are compared with the Brownsville, Pilot unit and other laboratory data on Fig. 1.

Age of Catalyst Hrs.	0	12	96
Beacon Run $\frac{\#}{\pi}$	11143	11146	11144
Analysis			
% Fe	100	25	-
% Carbide	-	60	75
% Oxide	-	15	25
C ₃ + Yields gms/cm	152.4	142.0	141.0

The solution to the problem is to use a catalyst which resist changes effected by the surrounding atmosphere. The fused catalysts such as spent CM&S catalyst are apparently much more stable in this respect. Even these however will show a drift in activity with time especially at Brownsville and a method must be developed to periodically reactivate the catalyst in place. Permanent poisons, whatever they may be, will have to be identified and removed from the fresh feed and recycle gases.

It might be argued that the Mill Scale activity could be maintained by periodic reactivation. This is probably true but with this catalyst reactivations would probably be required every 4 hrs. or so.

Another hypothetical solution to the problem is to build 600 - 11.5" I.D. units like Montebello but even then Mill Scale catalyst would probably have to be reactivated every 2 days or so. Effect of Operating Variables

The effect of changes in operating conditions and physical features of the reactor will be insignificant compared to the effect of the catalyst discussed above. This is true at present low conversion levels but when higher levels are being attained the choice of operating conditions may become quite important. It seems quite reasonable to expect that best results will be obtained with a full bed of well reduced catalyst. There is also a little evidence that fine catalyst is better than coarse. There is also some evidence that both the fresh feed and recycle streams should be preheated

to full reaction temperature. Whether this and an additional reactor, to permit operating at lower space velocity, will be required when a good intermittently reactivated catalyst is used remains to be seen.

Recommendations

The following recommendations are made:

- 1. After Beacon, Montebello and others involved have reviewed this report we should get together and agree upon a single concerted position and be ready to work toward a common goal.
- 2. Perhaps we should consult with experts on Fischer-Tropsch mechanism and catalysts such as Craxford and Herrington.
- 3. Once having agreed upon a goal a series of experiments should be layed out for Beacon, Montebello and Brownsville to:
 - A. Confirm these conclusions.
 - B. Learn how to manufacture a good catalyst.
 - C. Learn how to keep it active by in situ reactivation.

The program of experimentation should be layed out in conference but the following are a few suggestions for consideration.

- 1. Beacon should learn to operate its reactor with diluted catalyst or at higher space velocities so that it can deliberately produce a lower level of conversion where differences in catalysts should show up.
- 2. Montebello should continue working, first with Mill Scale and then with spent CM&S catalyst and should try to develop a procedure to periodically reactivate the catalyst. Montebello should also set up fused catalyst manufacturing conditions and with Beacon's small unit assistance learn how to manufacture a good stable catalyst commercially.
- 3. Montebello should try poisoning its gas with salt water, acid water and highly chlorinated drinking water washes as at Brownsville to see if these, and which of them are really responsible for permanent poisons if any.

In the meantime Brownsville should also be conducting a series of experiments, to confirm the conclusions reached here, as follows:

- l. A small reactor should be set up at once to operate in parallel with the commercial unit. Runs should be made first with their catalyst, fresh feed and recycle gas. The results of this run should be compared with another made with the same catalyst but with synthesis gas from the small hydrogen generator (K-303) which has been washed with only fresh water. Runs should also be made with fresh & spent Brownsville catalysts, methods of in situ reactivation should be tried etc.
- 2. It is also suggested that Brownsville try running once thru (no recycling) with a fresh catalyst to see by difference if the acid water used in the effluent gas scrubber (recycle) is responsible for poisons.
- 3. They should also make a similar test run (test run meaning changing only one thing at a time) using once the operation with a closed water cooling system on the synthesis gas wash tower.

Acknowledgement

I wish to thank the plant for the excellent cooperation I received in this study and particularly do I gratefully acknowledge the help of Mr. M. P. Hnilicka who was assigned to work with me throut my stay in Brownsville.

LPG-(MF-MI-ML-JR)

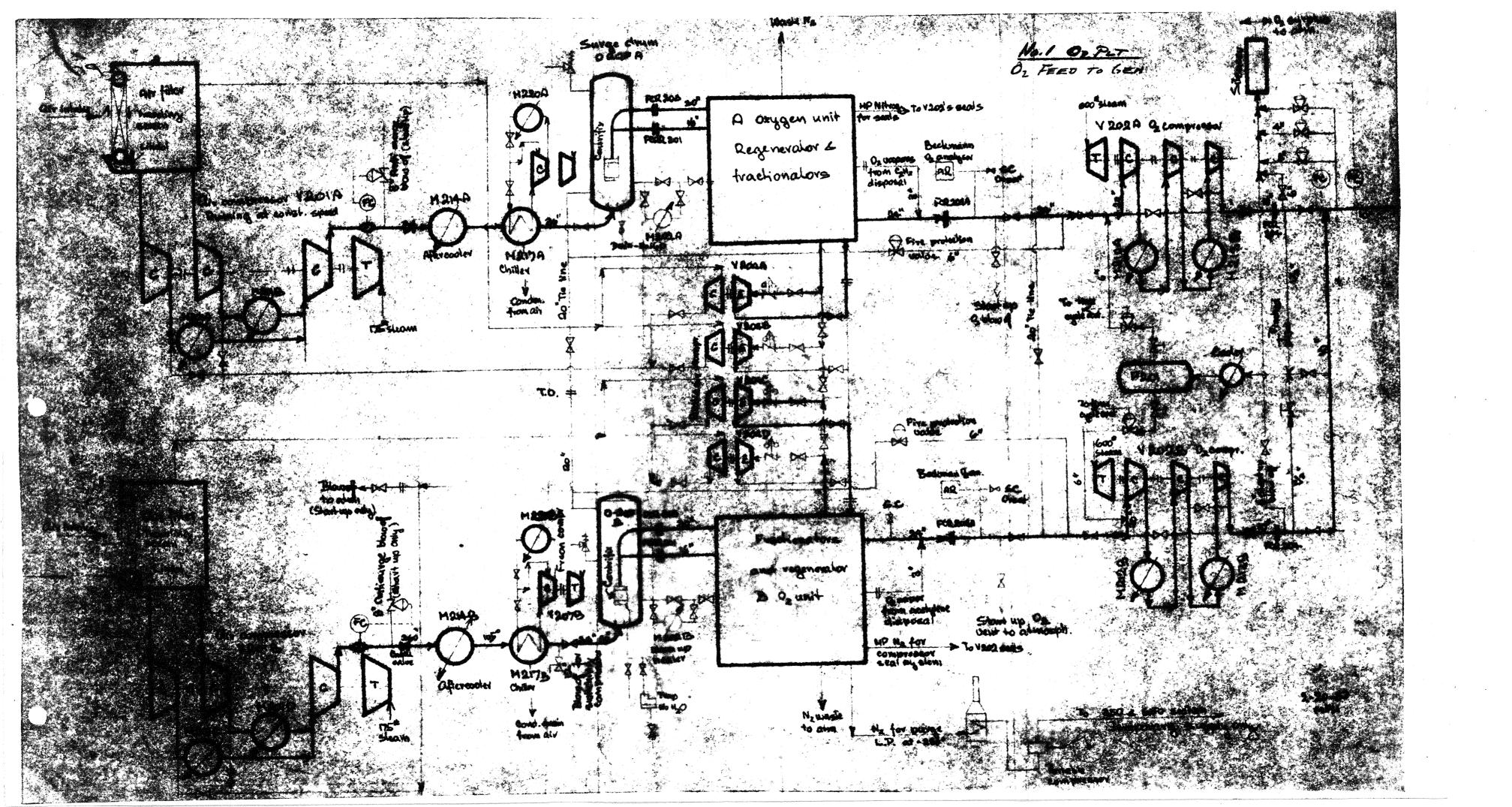
Report by: Thurches

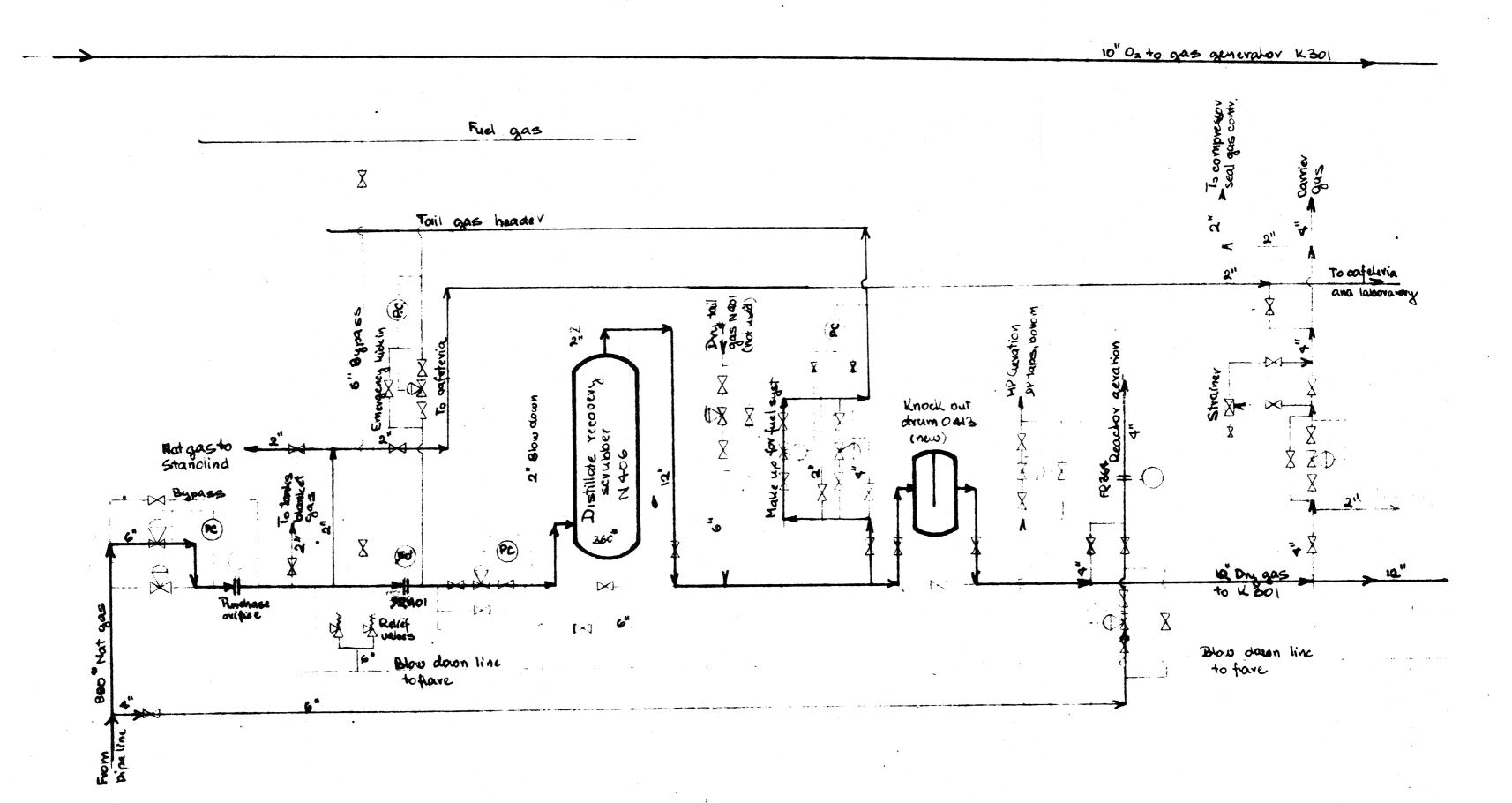
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WMS-KGM

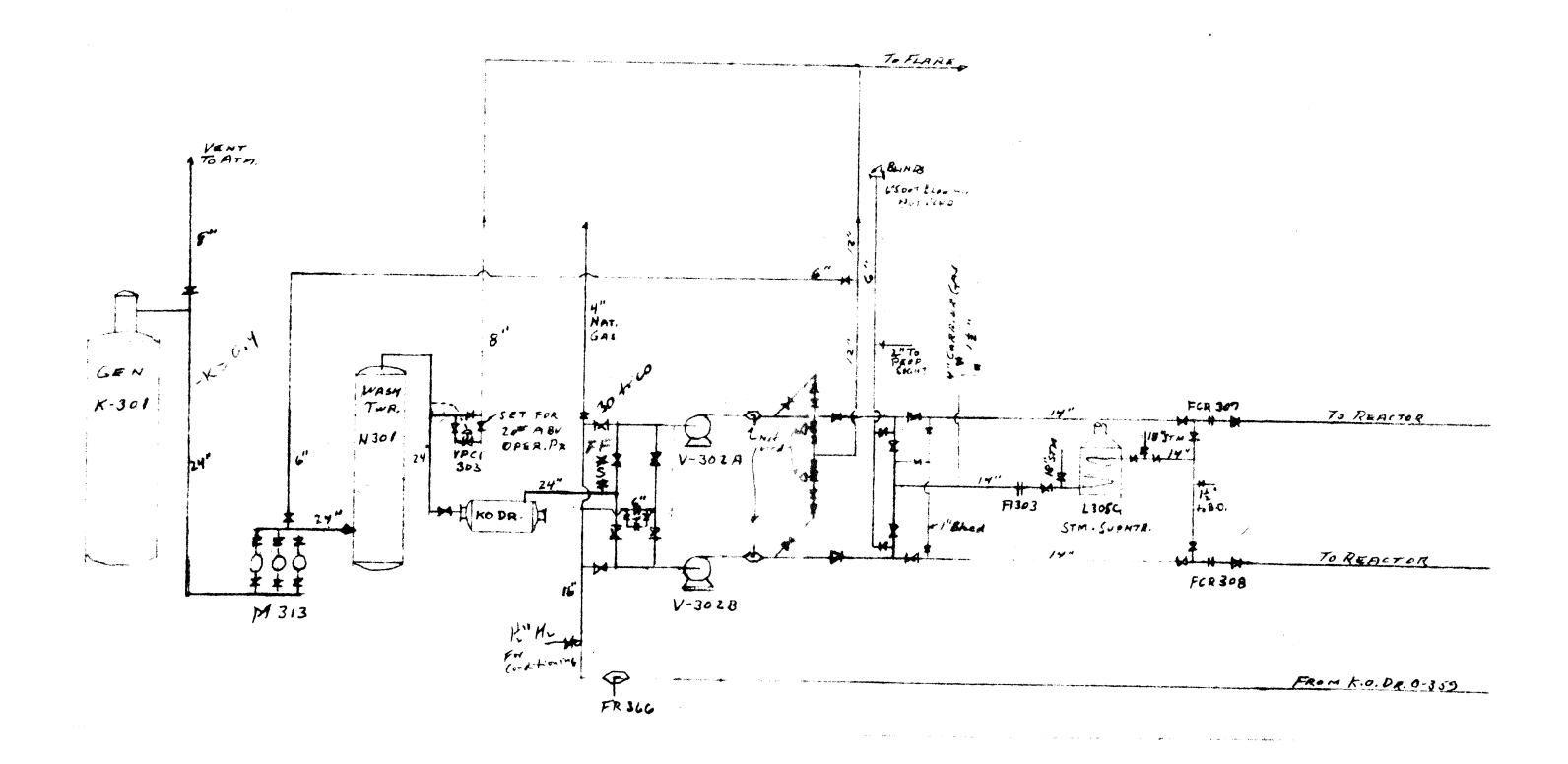
FMDawson(2), RHAitken(2) from LCKjr

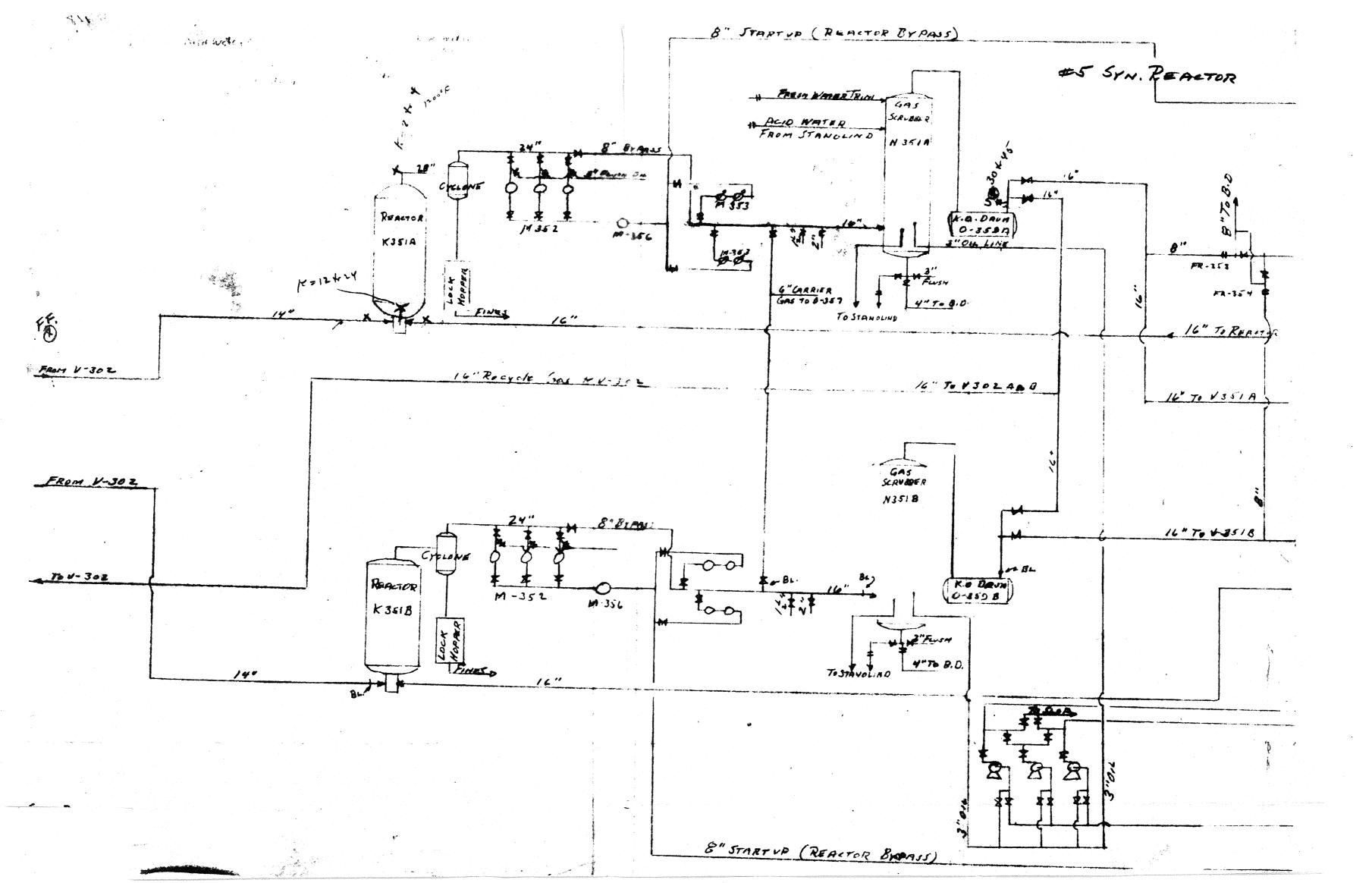
APPENDIX





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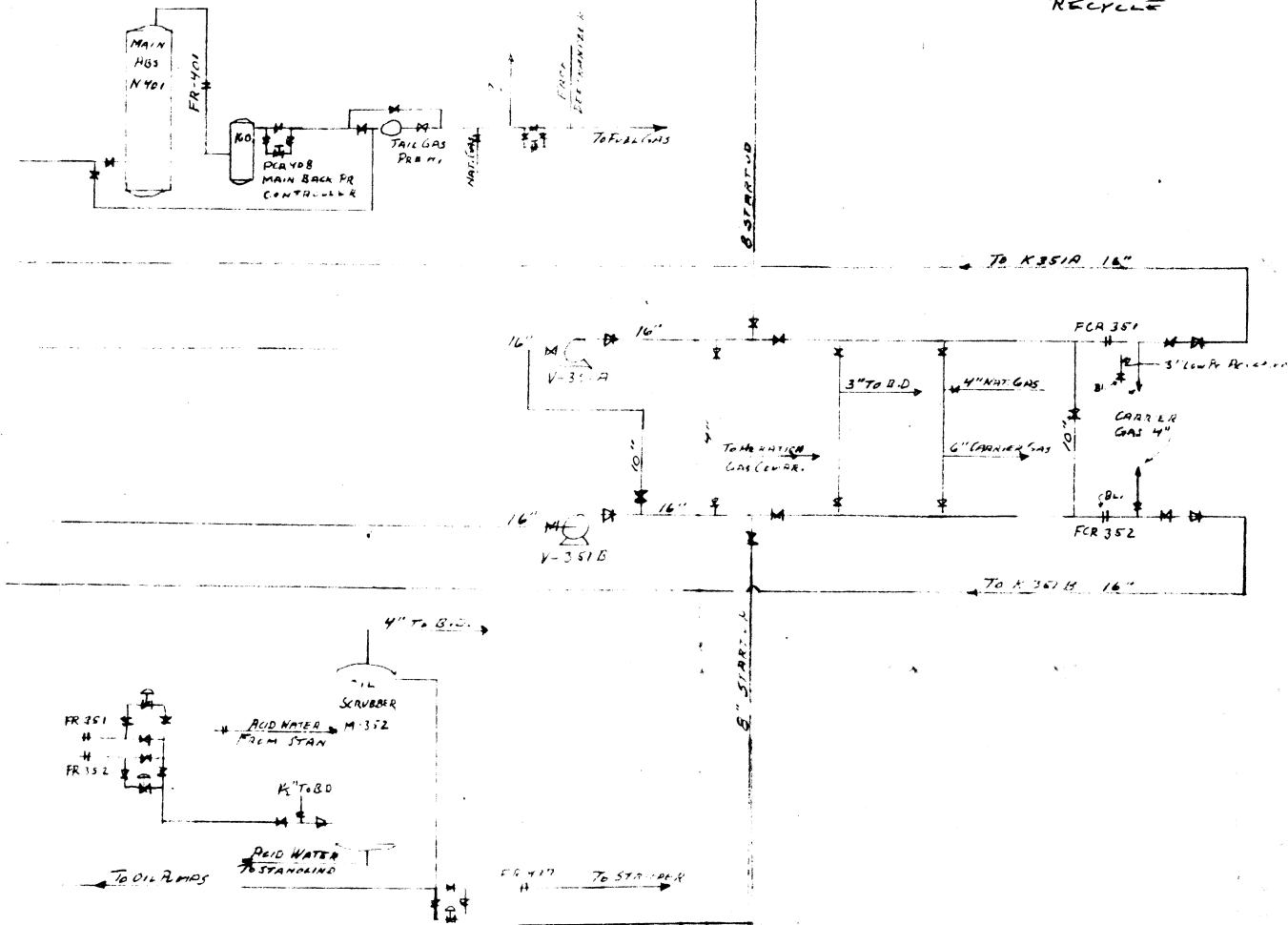


TABLE I

TABULATION OF DATA USED IN H2 CONVERSION CORRELATIONS

Montebello Run #63 Brownsville Mill Scale Cat.	FF H2/CO	CO Conv. %	H ₂ Conv. %	Mols CO ₂	CO to CO2	(H ₂)(CO ₂) (CO)(H ₂ O)	Tot. ^C 3+ lbs./hr.	CH ₂ Mols/Hr.	Mols CO Fed	co to c ₃₊	R/FF	CO +H2	Mols CH4 Make	co to CH4	Net ^H 2 ^O Make MPH	H ₂ MPH in Feed	H ₂	Space Velocity	Reactor inlet °F.
A B C D E F G H I J K L M N O	1.56 1.63 1.60 1.63 1.62 1.62 1.62 1.62 1.68 1.63 1.64 1.61	34.36 94.15 94.56 94.47 94.13 93.71 93.33 93.53 93.52 92.42 91.07 90.81 89.46	80.95 80.30 80.85 81.11 82.28 80.87 80.76 80.27 79.14 77.16 77.33 75.28 74.03 71.22	3.220 3.250 3.238 3.074 3.075 3.369 3.324 3.485 3.485 3.413 3.519 3.620 3.637	21.2 21.1 19.8 20.1 21.5 21.9 22.8 22.1 22.3 22.9 22.8	12.84 12.65 13.11 12.25 11.38 11.58 10.88 11.58 11.98 11.78 11.60 11.05 11.93 10.76 11.41	130.89 131.92 136.50 137.02 135.37 131.58 134.65 129.66 133.96 122.98 123.85 124.92 122.06 120.64 122.20	9135 9.42 9.75 9.67 9.67 9.62 9.57 8.89 8.92 8.62 8.73	15.213 15.343 15.338 15.511 15.284 15.639 15.777 15.684 15.891 15.878 15.432 15.763 15.813 15.856	61.5 61.4 63.1 63.1 61.0 59.2 55.3 57.6 55.1 55.1	0.99 0.98 0.96 0.97 0.96 0.95 0.98 0.99 0.99	86.19 85.55 86.13 86.20 86.73 85.76 85.55 85.33 84.51 83.07 82.86 81.28 80.39 79.11 78.13	1.114 1.275 0.944 1.437 1.177 1.249 1.265 1.310 1.259 1.328 1.315 1.326 1.259 1.301	7.32 8.30 6.15 9.25 7.70 7.98 8.05 7.92 8.35 8.41 7.96 8.18	7.471 7.325 7.316 7.506 7.410 7.248 7.255 7.283 6.883 6.883 6.887 6.448 6.229 6.080	23.751 25.065 24.512 25.224 25.397 25.413 25.621 25.621 25.943 25.915 25.915 25.656 26.018	31.45 29.22 29.84 29.75 29.17 28.52 28.31 28.63 27.42 26.23 25.25 24.88 24.27 23.36	1039 1079 1085 1096 1084 1095 1114 1102 1126 1112 1133 1149 1218 1218	415 414 412 415 414 413 416 415 412 409 409 408 406 404 402
MISC. LAB. & PILOT U DATA FROM EASTMAN'S TULSA REPORT	NIT																		and the second
Beacon 11140 Stanolind	2.0	98.60	90.66	0.897	10.22	23.31	87.29	6.23	8.775	71.0	2.0	93.31	0.838	9.54	6.895	17.575	39.23	9120	•
D-201-42 3/5 D-201-45 3/5 D-201-46 3/5	1.97 1.85 1.84	97.94 93.85 97.23	84.97 79.21 84.45	0.652 0.753 0.655	14.61 16.41 14.49	20.33 9.07 15.55	44.68 39.37 43.98	3.19 2.81 3.14	4.462 4.586 4.520	71.5 61.3 69.5	1.0 1.0 1.0	89.34 84.35 88.94	0.446 0.496 0.440	10.0 10.8 9.73	2.850 2.564 2.796	8.791 8.471 8.332	32.41 30.26 33.55	1072 1195 1137	• •
Montebello 57 F/O 64 F/P 65 F/N 66 F/O	1.57 1.62 1.62 1.63	93.75 90.26 90.87 91.15	78.57 75.44 77.00 75.03	2.850 3.403 3.114 3.430	22.47 22.32 20.49 22.88	13.37 10.85 9.18 11.82	102.87 119.32 120.97 117.28	7.35 8.52 8.64 8.38	12.678 15.240 15.194 14.985	58.0 55.9 56.9 55.9	0.98 1.06 1.05 1.00	84.47 81.10 82.31 81.17	1.023 1.166 1.216 1.262	8.06 7.65 8.00 8.42	5.712 6.402 7.108 6.294	19.925 24.645 24.537 24.358	28.66 25.97 28.96 25.83	1119 1138 1218 1210	••
MONTEBELLO RUN #59 (SPENT CM&S)																			
A B C D E F G H	1.65 1.65 1.66 1.65 1.72 1.77 1.77	93.71 92.74 94.53 95.29 95.58 94.57 93.19 92.59	84.10 82.32 81.90 83.81 82.38 80.48 78.60 77.31	3.175 3.271 3.587 3.264 3.504 3.343 3.562 3.423	19.51 19.95 22.01 19.99 21.73 20.89 22.26 21.83	7.72 12.80 12.19 15.46 13.63 12.77 13.16	150.6 142.21 143.97 155.63 145.05 145.10 134.67 138.35	10.76 10.16 10.28 11.12 10.36 10.36 9.62 9.88	16.267 16.394 16.293 16.327 16.125 15.996 15.999	66.14 61.97 63.09 68.11 64.25 64.76 60.12 63.02	0.87 0.88 0.85 0.84 0.85 0.87 0.87	87.73 86.25 86.65 88.14 87.23 85.56 83.86 82.74	1.444 1.137 1.258 1.168 1.229 1.210 1.201 1.238	8.87 6.93 7.72 7.15 7.62 7.56 7.50 7.89	7.893 8.470 7.558 8.079 7.940 7.841 7.491 7.005	26.810 27.128 27.070 26.975 27.748 28.385 28.384 28.459	29.44 31.22 27.92 29.94 28.61 27.62 26.39 24.61	1383 1463 1517 1488 1565 1596 1705 1761	140 146 153 155 155 152 152 156

MONTEBELLO RUN 49 ALLAN WOOD CAT.	FF H _{2/CO}	CO Conv. %	Space Vel.	H ₂ Conv. %	CO to C ^O 2	(H ₂)(CO ₂) (CO)(H ₂ O)	Tot.C ₃ + lbs./Hr.	(Ch ₂) Mols/Hr.	Mols CO Fed	CO to C3+	R/FF	Mols CH ₄	CO to	Net MPH H ₂ O <u>Make</u>	H ₂ MPH in Feed	H ₂ to	CO +H ₂ Conv.
A B C D E F G H I	1.59 1.60 1.65 1.65 1.615 1.664 1.645	91.63 91.09 88.25 87.41 89.67 87.98 87.52 86.34	945 1057 1140 1120 1088 997 994 1094 1122	82.74 79.12 75.48 77.26 77.46 72.57 73.80 74.41	16.0 15.4 13.1 18.9 19.6 19.2 18.9 18.9	6.2 6.5 7.5 7.2 7.8 7.7 7.2 6.7	134.67 138.62 123.01 132.94 121.91 127.14 127.87 124.88 126.24	9.62 9.91 8.78 9.49 8.70 9.08 9.13 8.92 9.02	14.506 14.705 14.890 14.565 15.053 14.858 15.149 15.212	66.3 67.4 58.7 59.7 61.4 58.9	0.96 1.03 1.02 .99 1.00 .99 .99 .99	1.135 0.889 0.845 0.859 0.907 0.816 0.785 0.804 0.973	7.8 6.0 5.8 6.2 5.3 6.4	7.845 8.377 6.959 6.296 7.473 7.092 6.985 7.129 6.610	23.065 23.499 23.870 24.623 24.024 24.310 24.724 24.760 24.188	34.0 35.6 29.2 25.6 31.1 29.2 28.3 28.6 26.7	86.17 83.73 80.40 81.08 82.06 78.46 78.95 79.15 77.43
K L M N O P Q R S T U V W X Y	1.59 1.63 1.62 1.62 1.59 1.61 1.59 1.64 1.52 1.70 1.62 1.55	N 0 85.998 85.66.85 85.66 85.66 85.85 84.50 84.98 82.86 88.89 88.89	1037 1033 1085 1127 1083 1163 1142 1244 1256 1203 1252 1185 1219 894 856	d a t 73.14 73.03 72.81 75.53 72.51 71.30 71.99 71.13 72.18 71.64 72.80 71.99 79.16	20.0 17.3 18.5 18.6 18.6 18.5 18.0 17.7 16.6 16.4 14.4	717074899486206	117.23 127.17 123.30 126.43 120.95 119.94 125.97 125.59 126.66 125.15 123.68 125.46 123.56 99.53 98.18	5.37 9.09 8.81 9.03 8.64 5.00 5.97 9.095 8.83 6.83 7.11 7.01	14.828 14.775 14.731 15.064 14.904 14.927 15.503 15.556 15.556 15.536 15.539 10.558 10.626	561°8904795577930 561°99879795576766	1.03 1.03 1.01 1.02 1.04 1.49 1.55 1.47 1.46 1.46 1.06 0.99	0.938 0.939 0.8863 0.856 0.856 0.876 0.773 0.837 0.837 0.837 0.837 0.539 0.539	340774871468211 5.11	6.218 6.792 6.534 6.948 6.747 6.909 7.110 6.913 6.949 7.198 7.584 5.962 6.362	23.561 24.083 23.548 24.329 24.169 23.756 24.588 24.588 24.588 24.5922 23.603 25.360 25.360 25.157 16.220 16.441	26.4 28.7 28.7 28.9 28.9 28.9 29.4 29.1 27.9 28.7 29.1 29.1 29.3 29.3 29.3 29.3 29.3 29.3 29.3 29.3	77.80 77.92 77.72 78.61 77.52 76.53 77.11 76.30 76.71 76.53 76.48 75.57 76.11 83.0 83.11
MONTEBELLO 46 & STANOLIN (FROM TDC802-	RUN D RUN 2																
46 F/1 46 J/P 46 Q/W Stand. 29-1 Stan2 -3 -4 -5 -6 -7 -8 -9 -10 Stan. Ave.	1.68 1.69 1.92 1.90 2.04 1.89 1.92 1.96 1.97 1.97	68.60 75.29 77.44 97.89 95.08 98.28 92.29 91.29 91.75 92.58 91.18 90.15	856 856 902 912 956 957 1033 945 952 982 991 954	54.31 58.20 61.60 85.94 82.19 82.80 77.24 73.69 74.14 72.53 70.79 76.26	18.1 18.9 18.5 11.0 10.4 14.2 16.4 18.8 15.7 16.4 17.5	5.8 5.9 5.7 16.98 7.60 31.2 8.73 8.81 9.27 10.1 9.1 8.9	91.75 104.03 104.85 33.92 32.93 30.45 29.96 30.04 28.06 29.11 29.03 27.75 28.01	6.55 7.49 2.42 2.35 2.17 2.15 2.00 2.08 2.07 1.98 2.00	15.680 15.964 15.510 3.078 3.112 3.025 3.204 3.192 3.054 3.075 3.176 3.219 3.098	41.8 46.5 48.6 75.7 66.4 67.6 67.6 67.6 64.6	1.12 1.04 1.05 1.03 1.01 1.00 .99 1.00 1.01 1.01	0.664 0.721 0.754 0.334 0.104 0.439 0.273 0.286 0.263 0.261 0.292	4.5 4.9 10.8 3.5 5.9 8.6 2.6 14.5 98.6 98.6 98.6 98.6	5.268 6.000 6.353 2.062 2.210 1.738 1.774 1.689 1.715 1.689 1.731	26.353 26.531 26.155 5.909 5.902 6.185 6.045 6.135 5.977 5.970 6.029 6.111	20.0 22.6 34.9 37.4 28.3 28.3 28.3 27.8 27.8 27.8 27.8	90.8 87.5 87.6 87.6 80.5 81.8

Run_#	11.138	11, 143	11,144	11,146	11,004	11.096
Catalyst	Brownsville Mill Scale reduced at Beacon Best run with Mill Scale Cat. Rec'd. 10/10/51	Brownsville 0 Hr. Sample Run #11 Run as Received	Brownsville 96 Hr. Sample Run #11 Run as Received	Brownsville 12 Hr. Sample Run #11 Run as Received	Spent CM&S Run At 200 #	Bethlehem Mill Scale
Lgt. of Run (Hrs.) R/FF Pressure, psig Temp.	2.0 400 675	84 2.0 400 675	84 2.0 400 675	72 2.0 400 67 5	84 2.0 200 650	84 2.0 200 650
Total ^C 3 ⁺ Yields (gms/CM) V/Hr/V	155 80	152.4 80.7	141.0 81.7	142.0 79.1	149.6 39.6 5.000	157.4 39.6
V/Hr/V (x125) H_/CO FF % Contr. H2 Conv. CO Conv.	2.02 87.1 94.4 99.2	2.0 86.7 93.8 99.0	1.89 78.3 86.0 97.5	2.05 82.2 89.3 98.1	1.99 87.0 94.6 99.4	1.96 85.3 92.1 99.3
CO to CO ₂ % CO to CH ₄ CO to C ₂ s CO to C ₃ + CO to Oxygenates CO to C ₃ + Total H ₂ to H ₂ O Chem.Fre	8.0 9.8 7.7 61.8 12.7 74.5	8.1 10.0 8.1 60.9 13.0 73.9 40.9	12.9 10.9 10.1 53.3 14.5 67.8 34.8	11.4 10.0 8.8 56.5 13.4 69.9 36.4	7.3 10.9 7.4 67.8 4.3 72.1 43.2	8.0 9.4 9.1 64.5 10.5 75.0 41.3
Wet Gas % Unsats. C2 C3 C1 (H2) (C02)	29.0 85.2 86.2	50.5 86.7 88.0	55.7 85.5 90.5	52.4 86.2 86.3	36.8 81.1 83.8	57•7 88•2 90•6
(CO) (H ₂ O)	23.0	18.9	21.1	21.3	23.5	21.9

BROWNSVILLE RUN #5 (Not used except in Fig. 1, A & B)

Date	1-16-51 (24 Hrs.)	1-17-51	1-18-51	1 == 20 == 51	1-21-51	1-22-51	1-23-51	1-24-51
Mols CO CO Conv. CO Unconv. Mols H ₂ H ₂ % Conv. H ₂ Unconv. % Unconv. Mols Unconv. Lbs.	100	100	100	100	100	100	100	100
	50.71	49.21	52.31	71.74	73.69	71.24	65.62	69.38
	49.29	50.79	47.69	28.26	26.31	28.76	34.38	30.62
	179	185	182	183	183	183	183	182
	21.79	26.08	24.00	39.18	43.21	39.24	32.92	36.86
	78.21	73.92	76.00	60.82	56.79	60.76	67.08	63.14
	140.0	136.75	138.32	111.30	103.9	111.2	122.8	114.9
	280	273.5	276.6	222.6	207.8	222.4	245.6	229.8
In Product Wt. % H ₂								
Total Prod. Lbs. Unconv. CO Lbs. Unconv. CO Wt. % Calc. Unconv. CO Wt. % Actual	4.72	6.33	6.69	4.99	4.77	4.85	5.32	5.17
	5932	4321	4135	4461	4356	4586	4617	4445
	1380	1422	1335	791	737	805	963	857
	23.3	32.9	32.28	17.73	16.91	17.6	20.9	19.3
	22.18	32.90	30.76	17.50	16.21	17.47	20.75	19.74
CO ₂ Wt. %	24.70	31.46	32.76	38.09	38.63	36.43	35.08	37.06
Lbs.	1465	1359	1355	1699	1683	1671	1620	1647
Mols (CO to CO ₂)	33.3	30.9	30.8	38.6	38.3	38.0	36.8	37.4
Water Wt. % Lbs. Mols H ₂ to H ₂ O %	32.20	8.96	9.22	12.84	13.21	15.44	14.94	10.63
	1910	387	381	573	575	708	690	473
	106	21.5	21.2	31.8	31.9	39.3	38.3	26.3
	59.2	11.6	11.6	17.4	17.4	21.5	20.9	14.5
C_3 + (Incl. Oxygenates) Wt. % Lbs. Mols CH_2 (CO to C_3 +)	10.94	13.24	13.22	15.62	15.81	14.75	14.35	15.02
	649	572	547	697	689	676	663	668
	46.4	40.9	39.1	49.8	49.1	48.3	47.4	47.7
CH ₄ Wt. %	0.91	0.98	0.95	3.87	4.53	4.53	3.73	4.12
Lbs.	54	42	39	173	197	208	172	183
Mols (CO to CH ₄)	3.4	2.6	2.4	8.1	12.3	13.0	10.8	14.4
C ₂ s Wt. % Lbs. Mols (CO to C ₂ s)	1.14 67.6 4.7	1.39 60.0 4.1	1.64 67.8 4.7	2.68 120 8.3	2.13 93 6.4	89 6.1	1.57 72 5.0	1.92 85 5.9
Oxy. Comps. Wt. %	0.58	1.52	1.57	1.96	2.02	2.10	2°21	2.88
Lbs.	34.4	65.7	64.9	8	88	96	102	128
CO to Oxygenates	2.5	4.7	4.6	6.2	6.3	6.9	7°3	9.1
R/FF	1.23	1.10	1.09	1.33	1.34	1.20	1.30	1.31

	3/13/51	3/14/51	3/15/51	3/16/51	3/17/51	3/18/51	3/19/51
CO Unconv. Mols. Mols. H ₂ H ₂ % Conv. H ₃ Unconv. %	100 70.78 29.22 185 43.54 56.46 104.45 208.9	100 74.71 25.29 180 48.14 51.86 93.35 186.7	100 75.28 24.72 176 49.24 50.76 89.34 178.6	100 176	100 76.51 23.49 174 48.53 51.47 89.56 179.1	100 73.15 26.85 183 43.36 56.64 103.65 207.3	100 73.76 26.24 183 42.49 57.51 105.24 210.5
Wt. % H ₂ Total Prod. Lbs. Unconv. CO Lbs. Unconv. CO Calc. Wt.% Actual Wt.% CO ₂ Wt.% Lbs. Mols CO to CO ₂	5.36 3897 818.2 21.0 20.8 25.46 992.2 22.6	4.79 3898 708 18.2 17.82 25.29 985.8 22.4	4.59 3891 692.2 17.8 17.63 21.94 853.7 19.4		4.60 3896 657.7 16.9 16.89 23.62 920.2 20.9	5.28 3926 751.8 19.1 19.01 28.55 1120.9 25.5	5.21 4040 734.7 18.2 18.89 28.38 1146.6 26.1
Water Wt.% Lbs. Mols. H2 to H ₂ O Mol.%	13.11 510.9 28.4 15.4	14.50 565.2 31.4 17.5	16.91 658.0 36.6 20.8		10.74 418.4 23.2 13.3	10.77 422.8 23.5 12.8	12.08 488.0 27.1 14.8
C3 Incl. Oxy. Wt.% Lbs. Mols CH2(Mols. CO to C3)	451.3	13.32 519.2 37.1	13.01 506.2 36.2		13.49 525.6 37.5	12.59 494.3 35.3	10.88 439.6 31.4
CH ₄ Wt. % Lbs. Mols.(CO to CH ₄)	198.7	5.93 231.2 17.5	6.97 271.2 17.0		10.72 417.7 26.1	4.40 172.7 10.8	4.83 195.1 12.2
C ₂ Wt. % Lbs. Mols. CO to C2 ⁸	2.21 86.1 5.9	2.20 85.8 5.9	1.87 72.8 5.0		2.41 93.9 6.5	2.41 94.6 6.5	2.37 95.7 6.6
Oxy. Comps. Wt. % Lbs. Mols.	2.20 85.7 6.1	2.24 87.3 6.2	2.23 86.8 6.2		2.55 99.3 7.1	2.69 105.6 7.5	2.74 110.7 7.9
R/FF	0.58	•51	•42		•53	•60	.61

	APRIL	APRIL 23	APRIL 24	APRIL	APRIL 26	APRIL	APRIL 28	APRIL 29	APRIL 30	MAY	MAY 2	MAY 3	MAY	MAY 5	MAY 6	MAY	MAY 8	MAY9	MAY 10	MAY	MAY 12	MAY	MAY 14
OWNSVILLE RUN #8 Ols CO Conv. Ounconv. Ols H ₂ Ols Conv. Conv. Cols H ₂ Ols Conv. Cols Conv. Cols H ₂ Ols Conv. Cols Conv. Conv. Cols Conv. Conv. Cols Conv. Conv. Cols	100% 79.5 30.5 172. 58.3 41.7 71.7	100% 77.09 22.91 174. 56.80 43.20 75.2 150.4	100% 75.57 24.43 173. 54.62 45.38 78.5 157.	100% 82.22 17.78 168. 60.26 39.74 66.76 133.5	100% 83.11 16.89 170. 63.42 36.58 62.19 124.4	100% 81.27 18.73 169. 60.40 39.60 66.9 133.2	100% 81.89 18.91 167. 61.66 38.34 64.0 128.0	100% 79.51 20.49 168. 56.57 43.43 72.96 145.9	100% 82.35 17.65 170. 61.49 38.53 65.5 131.0	100% 83.22 16.78 164. 63.45 36.55 59.9 119.8	100% 83.72 16.28 161. 63.34 36.66 59.0 118.0	100% 83.04 16.96 166. 62.12 37.88 62.9 125.8	100% 82.96 17.04 172. 62.31 37.69 64.8 129.6	100% 84.30 15.70 172 64.29 35.71 61.44 122.9	100% 83.8 16.2 168. 63.37 36.63 61.54 123.0	100% 78.76 21.24 161. 60.30 39.70 63.9 127.8	100% 79.47 20.53 166. 54.95 45.05 74.8 149.6	100% 77.24 22.76 172. 52.71 47.29 81.3 162.6	100% 77.95 22.05 175. 53.63 46.37 81.1 162.2	100% 79.19 20.81 166. 53.99 46.01 76.4 152.8	100% 77.94 22.06 175. 54.47 45.53 79.7 159.4	100% 79.49 20.51 175. ,50.46 49.54 86.7 173.4	100% 76.82 23.18 179. 52.76 47.24 84.6 169.2
Wgt. % H ₂ Total Prod. Lbs. Unconv. CO Lbs. Unconv. CO. Wgt.% Unconv. CO Actual	3.82 3754 854.0 22.75 15.16	4.01 3751 641 17.08 16.96	4.18 3756 684 18.21 18.11	3.57 3739 499 13.35 13.17	3.30 3770 473 12.55 12.44	3.57 3731 524 14.04 13.53	3.44 3721 529 14.21 13.48	3.89 3751 574 15.30 15.18	3.47 3775 494 13.09 12.97	3.20 3744 470 12.55 12.42	3.20 3688 456 12.36 12.46	3.36 3744 475 12.69 12.64	3.43 3778 477 12.62 12.51	3.27 3758 440 11.71 11.59	3.27 3761 454 12.07 11.96	3.44 3715 595 16.02 15.90	4.00 3741 575 15.37 15.27	4.33 3755 637 16.96 16.77	4.30 3772 617 16.35 16.27	4.06 3764 583 15.49 15.37	4.15 3841 617 16.06 16.01	4.54 3819 574 15.03 14.92	4.37 3643 649 17.81 16.66
O ₂ Wgt. % Lbs. Mols.	33.16 1244.80 28.29	32.98 1237.1 28.12	31.73 1191.7 29.08	31.79 1188.6 27.01	31.40 1183.7 26.90	32.22 1202. 27.32	32.19 1197.8 27.22	36.84 1382. 31.41	31.31 1182.0 26.86	31.61 1183.5 26.90	36.69 1353.1 30.85	31.48 1178.6 26.79	32.33 1221.4 27.76	31.71 1191.7 24.81	32.10 1207.3 27.44	33.62 1248.9 28.38	34.83 1303.0 29.61	34.82 1307.5 29.72	33.12 1249.3 28.37	34.93 1314.8 29.88	35.75 1373.2 31.21	35.40 1351.9 30.73	33.72 1228.4 27.92
ater Wgt. % Lbs. Mols. 2 to H ₂ O Mol. %	16.14 605.90 33.66	15.44 579.2 32.18 .18	15.27 573.5 31.86 .18	16.75 676.3 34.79	693。3	17.63 657.8 36.54	17.11 636.7 35.37	15.84 594.2 33.01	18.41 695.0 38.61 .23	24.53 918.4 51.02	19.92 734.6 40.81	21.04 78.77 43.76 .26	12.79 483.2 26.84 .16	19.10 717.8 39.88	18.51 696.2 38.68	16.87 626.7 34.82	14.51 542.8 30.16 .18	13.17 494.5 27.47	16.01 603.9 33.55	15.33 577.0 32.06 .19	14.37 552.0 30.67	14.97 571.7 31.76	14.93 543.9 30.21
3+ (Incl. water sol. chem.) Wgt. % Lbs. Tols CH2	17.60 660.7 47.19	18.18 686. 48.9	16.02 601.7 42.98	19.87 742.9 53.06	928.2	19.41 724.2 51.73	20.11 748.3 53.45	20:12 754.7 53.91	18.81 710.1 50.72	23.28 871.6 62.26	22.62 834.2 59.59	22.71 850.3 60.74	16.50 623.4 44.53	17.96 674.9 48.20	17.62 662.7 47.34	14.71 546.5 39.04	15.14 566.4 40.46	14.43 541.8 38.70	14.58 549.9 39.28	13.99 526.6 37.61	12.35 474.4 33.89	11.26 430.0 30.71	11.68 425.5 30.39
H ₄ Wgt. % Lbs. Mols.	5.60 210.2 13.1	5.24 196.6 12.28	7.38 277.2 17.33	7.57 283.0 17.69	5.93 223.6 13.98	5.49 204.8 12.80	5.55 206.5 12.91	5.34 200.3 12.52	6.73 254.1 15.88	5.50 205.9 12.87	5.93 219。 13.69	5.55 207.8 12.99	5.65 213.5 13.34	5.72 214.9 13.43	5.71 214.8 13.43	5.23 194.3 12.14	5.89 220.3 13.77	5.87 220.4 13.78	6.17 232.7 14.54	6.29 236.8 14.80	6.22 238.9 14.93	6.82 260.5 16.28	5.69 207.3 12.96
2 Wgt. % Lbs. Mols.	2.06 77.3 5.26	89.65	2.13 80.0 5.44	2.61 97.6 6.64	2.20 82.9 5.64	2.38 88.8 6.04	2.49 92.7 6.31	2.34 87.8 5.97	2.42 91.4 6.22	1.80 67.4 4.59	2.37 87.4 5.95	3.37 126.7 8.62	3.28 123.9 8.43	3.08 115.7 7.87	8,80	3.17 117.8 8.01	2.77 103.6 7.05	3.03 113.8 7.74	8.16		8.15	3.50 133.7 9.10	2.27 82.7 5.63
xy. Comps. Wgt. % Lbs. Mols.	3.14 117.87 8.42 1.50	3.46 129.8 9.27 1.41	3.31 124.3 8.88 1.24	3.59 134.2 9.59 1.35	3.55 133.8 9.56 1.35	3.59 133.9 9.56 1.32	3.88 144.4 10.31 1.22	3.60 135.0 9.64 1.32	3.76 141.9 10.14 1.32	3.54 132.5 9.46 1.19	3.80 140.1 10.01 1.21	3.76 140.8 10.05 1.23	4.24 160.2 11.44 1.28	4.62 173.6 12.40 1.21	4.68 176.1 12.58 1.27	4.31 160.1 11.44 .91	3.89 145.5 10.39 .71	4.05 152.1 10.86 1.0	4.18 157.7 11.26 .91	3.55 133.6 9.54 1.08	4.04 155.2 11.09 1,48	4.22 161.2 11.51 1.69	3.78 137.7 9.84 1.63

BROWNSVILLE RUNS #9 & #10	SESSECTION CONTRACT SECURISHMENT SECURISHMEN	R U	N # 9	and approximate the state of th	untida vasilar er eldalallangin, maksarı allalla valilalandı ülçün k	SERVICE OF THE PROPERTY OF THE		RUN	#10	tern. Elling to an algument to an exercision to the state of the state	and the second seco
Date	6-5-51 (10 Hrs.)	6-6-51	6-7-51	6-8-51	6-9-51 (9 Hrs.)	$\frac{7-21-51}{(24 \text{ Hrs.})}$	7-22-51	7-23 to $7-28-51$	7-29-51	<u>7-30-51</u>	7-31-51 (18 Hrs.)
Mols CO CO Conv. CO Unconv. Mols H ₂ H ₂ % Conv.	100 81.47 18.53 184.5 54.71	100 80.44 19.56 182:9 48.97	100 71.15 28.85 186.7 41.54	100 71.91 28.09 178.0 45.50	100 78.49 21.51 174.7 47.09	100 76.77 23.23 186.5 51.84	100 77.04 22.96 186. 61.87	No Data	100 68.63 31.37 175.9 51.33	100 72.55 27.45 178.4 52.42	100 54.96 45.04 182.2 41.08
H ₂ Unconv. % H ₂ Unconv. Mols H ₂ Unconv. Lbs.	45,29 83,6 167,2	51.03 93.3 186.6	58.46 109.1 218.2	54.50 97.0 194.0	52.91 92.4 184.8	48.16 89.8 179.6	38.13 70.9 141.8		48.67 85.6 171.2	47.58 84.9 169.8	58.92 107.4 214.8
In Product:	CASE OFFICE ASSESSMENT OF THE PROPERTY OF THE										
Wt. % H ₂ Total Prod. Lbs. Unconv. CO Lbs. Unconv. CO Wt. % Calc. Unconv. CO Wt. % Actual	3765.7 518.8 13.77 13.67	5.00 3732 547.7 14.67 14.55	5.80 3762 807.8 21.47 21.30	5.19 3738 786.5 21.04 20.36	4.86 3802 602.3 15.84 15.73	4.76 3773 650.4 17.23 19.09	4.14 3425 642.9 18.77 16.9		4.71 3635 878.4 24.16 23.99	4.61 3683 768.6 20.86 20.68	5.76 3729 1261.1 33.81 33.55
CO ₂ Wt. % Lbs. Mols CO to CO ₂ Water Wt. % Lbs. Mols H ₂ to H ₂ O Mol %	25.67 966.7 22.0 19.41 730.9 40.6 22.0	32.39 1208.8 27.5 8.37 310.9 17.3 9.5	28.84 1085.0 24.7 9.30 349.9 19.4 10.4	26.52 991.3 22.5 11.28 421.6 23.4 13.1	29.16 1108.7 25.2 12.43 472.6 26.3 15.1	23.69 893.8 20.3	21.89 749.7 17.0		20.07 729.5 16.6	25.56 941.4 21.4	11.76 438.5 10.0
C ₃ + incl. Oxyg. Wt. % Lbs. Mols CH ₂ (Mols CO to C ₃ +)	13.52 509.1 36.4	15.25 569.1 40.6	14.57 548.1 39.1	14.67 548.4 39.2	14.76 561.2 40.1	14.73 555.8 39.7	17.51 599.7 42.8		16.03 582.7 41.6	14.71 541.8 38.7	14.31 533.6 38.1
CH ₄ Wt. %	4.30	5.15	2.71	2.90	3.41	3.09	5.00		2.79	2.82	0.62
Lbs. Mols (CO to CH ₄)	161.9 10.1	192.2 12.0	102.0 6.4	108.4 6.8	129.6 8.1	116.6 7.3	171.3 7.1		101.4	103.9 6.5	23.1 1.4
C ₂ s Wt. % Lbs. Mols CO to C ₂ s	2.01 75.7 5.2	2.54 94.8 6.5	1.03 38.7 2.7	2.12 79.2 5.5	2.17 82.5 5.7	1.35 50.9 3.5	2.17 74.3 5.1		2.21 80.3 5.5	4.23 155.8 10.7	1.14 42.5 2.9
Oxy. Comps. Wt. % Lbs. CO to Oxygenates	1,80 67,8 4,8	2.94 109.7 7.8	2.25 84.6 6.0	2.84 106.2 7.6	3.07 116.7 8.3	2.03 76.6 5.5	3.16 108.2 7.7		3.31 120.3 8.6	3.28 120.8 8.6	3.73 139.1 9.9
R/FF	1 33	1.50	1.55	0.81	0.63	1.69	1.46		1,60	1.87	1.75

<u>Date</u>	11-11-51 (24 Hrs.)	11-12-51	11-13-51	11-14-51	11-15-51	11-16-51	11-17-51	11-18-51	11-19-51 (24 Hrs.)
Mols CO Mols CO Conv. Mols CO Unconv. Mols H ₂ % H ₂ Conv. Mols H ₂ Unconv.	100 85.3 14.7 183 63.3 67.2	100 88.1 11.9 185 67.7 59.8	100 86.6 13.4 184 67.2 60.4	100 84.4 15.6 185 65.0 64.8	100 81.4 18.6 186 61.3 72.0	100 82.7 17.3 178 63.4 65.1	100 80.5 19.5 176 58.7 72.7	100 80.8 19.2 176 62.0 66.9	100 81.5 18.5 188 63.8 63.7
Lb. H ₂ Unconv.	135.4	119.6	120.8	129.6	144.0		145.4	133.8	127.4
Product									•
Wt. % H ₂ Total Prod. Lb. Unconv. CO. Lb. Unconv. CO Wt. % Calc. Unconv. CO Wt. % Actual	3.57 3800 412 10.82 10.82	3.16 3780 333 8.81 8.73	3.24 3730 375 10.05 9.95	3.43 3780 437 11.56 11.50	3.86 3730 521 13.97 13.82	3.50 3720 484 13.01 12.85	3.92 3710 546 14.72 14.58	3.61 3710 538 14.50 14.30	3.56 3580 518 14.47 13.44
CO_2 Wt. % CO_2 Lb. CO_2 Mols CO_3 Mol %	24.98 949 21.56 21.6	23.22 878 19.95 20.0	23.08 861 19.56 19.6	21.71 821 18,65 18.7	23.97 894 20.31 20.3	22.76 847 19.24 19.2	22.14 821 18.65 18.7	22.68 841 19.11	20.60 737 16.74
Water Wt. % Lb. Mols Mol %	19.25 732 40.7 22.2	23.78 899 49.9 27.0	22.74 848 47.1 25.6	22.20 839 46.6 25.2	16.83 628 34.9 18.8	20.36 759 42.2 23.7		19.1 19.62 728 40.4 23.0	16.7 20.11 720 40.0 21.3
C ₃ + Wt. %	17.69 672	16.74 633	18.28 682	18.88 714	19.94	17.24 641	18.39 682	17.92 665	18.56 664
Mols CH ₂ Mol %(CO to C ₃ +)	48.0	45.2	48.7	51.0	53.1	45.8	48.7	47.5	47.4
CH ₄ Wt. % Lb. Mols Mol %	3.31 126 7.88	4.01 152 9.50	3.37 126 7.88	2.94 111 6.94	2.95 110 6.88	3.50 130 8.13	3.40 126 7.88	3.18 118 7.38	3.21 115 7.19
^C 2 Wt. % Lb. Mols Mol %	2.65 101 7.0	2,98 113 7.8	2.5 8 96 6.6	2.41 91 6.2	2.44 91 6.2	2.44 91 6.2	2.57 95 6.6	2.65 98 6.8	2.74 98 6.8
Oxyg。Wt. % Lb. Mols Mol %	4.06 154 11.0	3.99 151 10.8	4.24 158 11.3	4.36 165 11.8	4.22 157 11.2	4.45 166 11.9	4.08 151 10.8	4.33 161 11.5	4.14 148 10.6
R/FF	1.49	1.37	1.31	1.42	1.53	1.43	1.13	1.36	1.16

BROWNS	VILLE	RUN	#12

<u>Date</u>	$\frac{11-29-51}{(24 \text{ Hrs.})}$	11-30-51	12-1-51	12-2-51	12 3 51	12-4-51	12-5-51	12-6-51	12-7-51 (6 Hrs.)
Mols CO	100	100	100	100	100	100	100	100	100
Mols CO Conv.	82.41	85.28	85,87	85.11	85.22	84.48	84.34	86.13	85.90
Mols CO Unconv.	17.59	14.72	14.13	14.89	14.78	15.52	1 5. 66	13.87	14.10
Mols H ₂	180	178	183	184	185	184	186	192	190
% H ₂ Conv.	59.53	65.36	66.92	66.71	65.21	66.56	64.61	67.45	67.29
Mols H ₂ Unconv.	72.8	61.7	60.5	61.3	64.4	61.5	65.5	62.5	62.1
Lb. H ₂ Unconv.	145.6	123.4	121.0	122.6	128.8	123.0	131.6	125.0	124.2
Product									
Wt. % H ₂ Total Prod. Lb. Unconv. CO Lb. Unconv. CO Wt. % Calc. Unconv. CO Wt. % Actual	3.88	3.26	3.11	3.10	3.24	3.08	3.31	3.10	3.18
	3750	3785	3891	3955	3975	3994	3976	4032	3906
	493	412	396	417	414	435	438	388	395
	13.15	10.88	10.17	10.56	10.44	10.89	11.01	9.62	10.12
	12.99	10.79	10.07	10.45	10.31	10.74	10.92	9.55	9.60
CO ₂ Wt. %	23.90	21.47	19.65	17.90	20.09	17.47	19.16	17.76	18,22
CO ₂ Lb.	896	813	765	708	799	698	762	716	912
Mols Mol % Water Wt. % Lb. Mols Mol %	20.4	18.5	17.4	16.1	18.2	15.9	17.3	16.3	16.2
	19.17	22.04	24.11	24.14	18.48	22.73	20.40	21.56	22.79
	719	834	938	955	735	908	811	869	890
	39.9	46.3	52.1	53.0	40.8	50.4	45.0	48.3	49.4
	22.2	26.0	28.5	28.8	22.0	27.4	24.2	25.2	26.0
Mol % C3+ Wt. % Lb. Mols Mol %	17.13	19.00	17.27	18.10	20.18	18.26	18.03	17.48	17.19
	642	719	672	716	802	729	717	705	671
	4 5. 9	51.4	48.0	51.1	57.3	52.1	51.2	50.4	47.9
CH ₄ Wt. %	2.85	2.86	2.78	2.67	3.18	2.91	3.39	6.21	4.09
Lb.	107	108	108	106	126	116	135	250	160
Mols Mol %	6.7	6.8	6.8	6.6	7.9	7.3	8.4	15.6	10.0
C ₂ Wt. % Lb. Mols Mol %	2.49 93	2.21 84	2.45 95.3	2.46 97.3	2.67 106	2.54 101	2.64 105	2.65 107	2.78 109
Oxyg. Wt. % Lb. Mols Mol %	4.06	4.30	4.57	4.58	4.34	4.08	4.21	4.22	3.74
	152	163	178	181	173	163	167	170	146
	10.8	11.3	12.8	12.9	12.7	11.3	12.0	12.2	10.7
R/FF	1.60	1.41	1.31	1.41	1.55	1.40	1.52	1.50	1.42

<u>Date</u>	$\frac{12-21-51}{(13 \text{ Hrs.})}$	12-22-51	12-23-51	12-24-51	12-25-51	12-26-51	12-27-51	12-28-51	12-29-51	12-30-51	12-31-51	1-1-52	$\frac{1-2-52}{(3 \text{ Hrs.})}$
Mols CO Mols CO Conv. Mols CO Unconv. Mols H ₂ % H ₂ Conv. Mols H ₂ Unconv. Lbs. H ₂ Unconv.	100 81.25 18.75 181 61.55 69.6 139.2	100 84.28 15.72 187 64.65 66.1 132.2	100 82.84 17.16 186 63.24 68.4 136.8	100 79.98 20.02 185 59.22 75.4 150.8	100 80.81 19.19 183 59.37 74.4 148.8	100 79.50 20.50 189 60.21 75.2 150.4	100 79.99 20.01 186 58.06 78.0	100 77.27 22.73 189 53.38 88.1 176.2	100 74.10 25.90 189 51.27 92.1 184.2	100 75.66 24.34 189 53.52 87.8 175.6	100 76.03 23.97 190 53.37 88.6 177.2	100 76.53 23.47 189 54.45 86.1 172.2	100 76.85 23.15 189 55.09 84.9 169.8
In Product													
Wt. % H ₂ Total Prod. Lb. Unconv. CO Lb. Unconv. CO Wt. % Calc. Unconv. CO Wt. % Actual	3.56 3910 525 13.43 13.31	3.36 3930 440 11.20 11.11	3.44 3980 480 12.06 11.97	3.80 3970 561 14.13 13.99	3.79 3930 537 13.66 13.51	3.74 4020 574 14.28 14.15	3.89 4010 560 13.97 13.87	4.44 3970 636 16.02 15.86	4.53 4066 725 17.83 17.64	4.36 4030 682 16.97 16.62	4.32 4100 671 16.37 16.32	4.24 4060 657 16.18 16.03	4.18 4060 648 15.96 15.81
CO ₂ Wt. % CO ₂ Lb. CO ₂ Mols - Mol %	17.71 692 15.7	20.25 796 18.1	17.98 716 16.3	18.07 717 16.3	18.93 744 16.9	17.09 687 15.6	18.42 739 16.8	21.51 854 19.4	19.49 792 18.0	19.64 791 18.0	19.43 797 18.1	19.81 804 18.3	19.34 785 17.8
Water Wt. % Lb. Mols Mol % C ₃ + Wt. %	20.13 787 43.7 24.1 20.06	19.50 766 42.6 22.8 19.54	34.39 1369 76.0 40.9 16.65	21.81 866 48.1 26.0 16.67	21.55 847 47.1 25.7 17.22	21.01 845 47.0 24.9 18.20	19.78 793 44.1 23.7 17.62	18.04 716 39.8 21.1 14.67	16.66 677 37.6 19.9 15.08	16.73 674 37.4 19.8 16.83	18.51 759 42.2 22.2 14.27	17.85 725 40.3 21.3 15.29	19.43 789 43.8 23.2 14.66
Lb. Mols Mol,%	784 56.0	768 54.9	662 47.3	662 47.3	677 48.4	732 52.3	707 50°5	582 41.6	613 43.8	678 48.4	585 41.8	621 44.4	595 42.3
CH ₄ Wt. % Lb. Mols Mol %	3.00 117 7.3	3.26 128 8.0	2.20 88 5.5	2.31 92 5.8	2.62 103 6.4	1.85 74	2.07 83 5.2	2.29 91	1.34	2.18 88	1.64	1.56	1.49 61 3.8
C ₂ Wt. %	1.78 70	2.45	2.35	2,20	2.25	4.6 2.26	2.34	5.7 2.21	3.4 2.28	5.5 2.15	4.2 2.15	3.9 2.54	2.50
Mols Mol % Oxy. Wt. % Lb. Mols Mol %	4.8 3.04 119 8.5	96 6.6 4.33 170 12.1	94 6.5 4.21 168 12.0	87 6.0 4.04 160 11.4	88 6.1 4.05 159 11.4	91 6.3 4.12 166 11.9	94 6.5 3.72 149 10.6	88 6.1 3.97 158 11.3	93 6.4 4.57 186 13.3	87 6.0 4.10 165 11.8	88 6.1 3.71 152 10.9	103 7.1 4.02 163 11.6	102 7.0 4.33 176 12.6
R/FF	1.55	1.66	1.38	1.38	1.49	1.65	1.32	1.42	1.47	1.41	1.47	1.41	1.28

DDOWNOUTLIE DUN #31	1 0 50	1 10 50	2 22 50	1 10 50	1 12 70	3 34 50	3 3 5 50	2 2/ 50	3 35 50			* **		
BROWNSVILLE RUN #14	$\frac{1-9-52}{(13 \text{ Hrs.})}$) 1-10-52	1-11-52	1-12-52	1-13-52	1-14-52	1-15-52	1-16-52	1-17-52	1-18-52	1-19-52	1-20-52	1-21-52	$\frac{1-2}{(5)}$
Mols CO CO Conv. CO Unconv. Mols Mols H ₂	100.00 83.53 16.47 187	100.00 83.27 16.73	100.00 83.00 17.00	100.00 82.09 17.91 185	100.00 81.92 18.08	100.00 83.25 16.75	100.00 86.78 13.22 187	100.00 86.54 13.46 185	100.00 85.92 14.08 186	100.00 83.88 16.12 179	100.00 84.94 15.06 186	100.00 85.53 14.47 186	100.00 80.41 19.59 190	1
H ₂ % Conv. H ₂ % Unconv.	63.51 36.49	61.74	60,10 39,90	57.05 42.95	57.85 42.15	61.46 38.54	67.25 32.75	67.21 32.79	62.66	61.78	62.83	63.42	56.91 43.09	
H ₂ Unconv. Mols	68.24	70.40	75.01	79.46	77.98	73.23	61.24	60.66	69.45	68.41	69.14	68.04	81.87	
H ₂ Unconv. Lbs.	136.48	140.80	150.02	158.92	155.96	146.46	122.48	121.32	138.90	136.82	138.28	136.08	163.74	1
IN PRODUCT														
Wt. % H ₂ Total Product Lbs. Unconv. CO Lbs. Unconv. CO Wt. % Actual	3.32 4111 461.2 11.22 11.14	3.48 4046 468.4 11.58 11.49	3.71 4044 476.0 11.77 11.70	3.94 4034 501.5 12.43 12.36	3.95 3948 506.2 12.82 12.74	3.61 4057 469 11.56 11.48	3.02 4056 370.2 9.13 9.06	3.07 3952 376.9 9.54 9.48	3.51 3957 394.2 9.96 9.90	3.51 3898 451.4 11.58 11.03	3.44 4020 421.7 10.49 10.40	3.40 4002 405.2 10.12 10.02	4.10 3994 548.2 13.73 13.60	4(
CO ₂ Wt. % Lbs. Mols.	16.44 675.8 15.4	20.51 829.8 18.9	20.87 844.0 19.2	23.30 940 21.4	24.16 953.8 21.7	21.41 868.6 19.7	17.66 716.3 16.3	17.30 683.7 15.5	21.24 840.5 19.1	17.09 666.2 17.1	19.19 771.4 17.5	16.80 672.3 15.3	20.29 810.4 18.4	8
Water Wt. % Lbs. Mols H ₂ to H ₂ O Mol %	21.78 895.4 49.7 26.5	18.28 739.6 41.1 22.3	19.08 771.6 42.9 22.8	14.32 577.7 32.1 17.4	15.43 609.2 33.8 18.3	19.11 775.3 43.1 22.7	24.6 997.8 55.4 29.6	26.92 1063.9 59.1 31.9	22.93 907.3 50.4 27.1	23.65 921.9 51.2 28.6	22.44 902.1 50.1 26.2	27.81 1113.0 61.8 33.2	22.94 916.2 50.9 26.8	8
C ₃ + Wt. % Lbs. Mols CH ₂ (Mol % CO to C ₃ +)	19.49 801.2 57.2	18.30 740.4 52.9	16.95 685.5 49.0	19.15 772.5 55.2	16.97 670.0 47.9	17.05 691.7 49.4	18.15 736.2 52.6	17.32 684.5 48.9	16.70 660.8 47.2	17.03 663.8 47.4	16.42 660.1 47.2	16.47 659.1 47.1	14.02 560.0 40.0	4
CH ₄ Wt. % Lbs. Mols - Mol %	2,40 98.7 6.2	3.36 135.9 8.5	2.72 110.0 6.9	2.55 102.9 6.4	2.89 114.1 7.1	2.54 103 6.4	2.79 113.2 7.1	2.15 85.0 5.3	2.74 108.4 6.8	2.58 100.6 6.3	2.94 118.2 7.4	1.83 73.2 4.6	2.25 89.9 5.6	
C ₂ Wt. % Lbs. Mols - Mol %	2.19 90.0 6.2	2.46 99.5 6.9	2.61 105.5 7.3	2.55 102.9 7.1	2.47 97.5 6.7	2.76 112 7,7	2.47 100.2 6.9	2.50 98.8 6.8	2.48 98.1 6.8	2.45 95.5 6.6	2.37 95.3 6.6	2.12 84.8 5.8	2.17 86.7 6.0	
Oxy。Comps. Wt. % Lbs. Mols	3.39 139.4 10.0	4.16 168.3 12.0	4.46 180.4 12.9	3.82 154.1 11.0	3.35 132.3 9.5 1.64	4.21 170.8 12.2	4.65 188.6 13.5	4.19 165.6 11.8 1.95	4.26 168.6 12.0 1.45	4.13 161.0 11.5	4.36 175.3 12.5	4.53 181.3 13.0	3,86 154.2 11.0]

Date	2-2-52 (21-1/2 Hours)	2-3-52	2-4-52	2-5-52	2-6-52	2-7-52	2-8-52	2-9-52	2-10-52	2-11-52 (10-1/2
Mols CO CO Conv. CO Unconv. Mols Mols H ₂ H ₂ % Conv. H ₂ Unconv. % H ₂ Unconv. Mols H ₂ Unconv. Lbs.	100 91 19 8 81 182 80 35 19 65 35 76 71 52	100 83.70 16.30 185 62.08 37.92 70.15 140.30	100 82.89 17.11 185 60.15 39.85 73.72 147.44	100 79.81 20.19 186 51.31 48.69 73.73 147.46	100 78.04 21.96 188 50.58 49.42 92.91 185.82	100 81.68 18.32 183 53.87 46.13 84.42 168.84	100 78.41 21.59 189 53.68 46.32 87.54 175.08	100 83.30 16.70 187 60.52 39.48 73.83 147.66	100 82.62 17.38 191 59.92 40.08 76.55 153.10	Hours) 100 84.40 15.60 186 61.57 38.43 71.48 142.96
In Product Wt. % H ₂ Total Prod. Lbs. Unconv. CO Lbs. Unconv. CO Wt. % Calc. Actual	1.42	3.54	3.66	4 56	4.60	4.24	4.06	3.64	3.80	3.50
	5037	3963	4028	3234	4040	3982	4312	4057	4029	4085
	246.7	456.4	479.1	565.3	614.9	513.0	604.5	467.6	486,6	436.8
	24.90	11.52	11.8	17.4	15.22	12.88	14.01	11.53	12.08	10.69
	4.82	11.41	11.78	14.07	15.15	12.90	14.99	11.49	11.97	10.59
CO ₂ Wt. %	9.94	20.79	20.76	25,77	24.66	24.08	23.31	18.04	19.63	19.42
Lbs.	500.7	823.9	836.2	833,4	996.3	956.9	1005.1	731.9	790.9	793.3
CO to CO ₂ Mol %	11.4	18.7	19.0	18,9	22.6	21.7	22.8	16.6	18.0	18.0
Water Wt. % Lbs. Mols H_2 to H_2 0 Mol %	21.04	20.22	18.49	14.73	14.95	17.01	15.49	24.98	22.39	20.89
	106.0	801.3	744.8	476.4	604.0	677.3	667.9	1013.4	902.1	853.4
	58.8	44.5	41.4	26.5	33.6	37.6	37.1	56.3	50.1	47.4
	32.3	24.1	22.4	14.2	17.9	20.5	19.6	30.1	26.2	25.5
C ₃ + (incl. water chems.) Wt.9	613.65	17.11	17.95	13.17	13.08	17.01	13.12	14.43	14.62	16.22
Lbs.	687.6	678.1	723.0	425.19	528.4	677.3	565.7	585.4	589.0	662.6
Mols CH ₄ (Mol % CO to C ₃ +)	49.1	48.4	51.6	30.4	37.7	48.4	40.4	41.8	42.1	47.3
CH ₄ Wt. %	8.65	3.57	3.00	3,92	3.08	2.69	4°21	2.39	3.14	2.91
Lbs.	435.7	141.5	120.8	126,7	124.4	107.1	181.5	97.0	126.5	118.9
Mols Mol %	7.2	8.8	7.6	7,9	7.9	6.7	11.3	6.1	7.9	7.4
C ₂ Wt. %	1.63	2.11	2.36	2,25	2. 27	2.40	2.56	2.29	2.37	2.60
Lbs.	82.1	83.6	95.1	72,8	91.7	95.6	110.4	92.9	95.5	106.2
Mols Mol %	5.6	5.7	6.5	5,0	6.2	6.5	7.5	6.3	6.5	7.2
Oxy. Comps. Wt. %	2.49	3.23	3.95	2,97	2,62	2.58	2.69	4.03	3.61	4.25
Lbs.	125.4	128.0	159.1	96.0	105.8	102.7	116.0	163.5	145.4	173.6
Mols	9.0	9.1	11.4	6.9	7.6	7.3	8.2	11.7	12.6	12.4
R/FF	1.23	1.34	1.32	0.93	0.76	0.83	0.64	1.53	1.65	1.78

2					
	<u>Date</u>	2-19-52 (17-1/2 Hours)	2-20-52	2-21-52	2-22-52 (3-1/2 Hrs.
	Mols CO CO Conv. CO Unconv. Mols Mols H ₂ H ₂ % Conv. H ₂ % Unconv. H ₂ Unconv. Mols H ₂ Unconv. Lbs.	100.00 82.44 17.56 178 56.98 43.02 76.58 153.16	100.00 81.38 18.62 182 55.58 44.42 79.07 158.14	100.00 76.81 23.19 184 45.89 54.11 96.32 192.64	100.00 75.19 24.81 193 43.54 56.46 100.52 201.04
	In Product Wt. % H ₂ Total Prod. Lbs. Unconv. CO Lbs. Unconv. CO Wt. % Actual	3.71 4128 491.7 11.91 11.83	4.04 3914 521.4 13.32 12.91	4.94 3900 649.3 16.65 16.03	5.24 3837 694.7 18.11 16.55
	CO ₂ Wt. %	22.97	23.45	24.96	23.22
	Lbs.	948.2	917.8	973.4	891.0
	Mols	21.6	20.9	22.1	20.3
	Water Wt. %	25.18	16.42	14.69	11.13
	Lbs.	1039.4	642.7	572.9	427.1
	Mols	57.7	35.7	31.8	23.7
	H ₂ to H ₂ O Mol %	32.4	19.6	17.3	12.3
	C ₃ + Wt. %	15.00	14.24	11.87	13.47
	Lbs.	619.2	557.4	462.9	516.8
	Mols CH ₂ (Mol % CO to C ₃ +)	ԿԿ.2	39.8	33.1	36.9
	CHi ₊ Wt. %	2.75	4.04	2.84	2.97
	Lbs.	113.5	158.1	110.8	114.0
	Mols Mol %	7.1	9.9	6.9	7.1
	C ₂ Wt. %	2.90	2.61	2.29	2.3 ¹ 4
	Lbs.	119.7	102.2	89.3	89.8
	Mols Mol %	8.3	7.0	6.2	6.2
	Oxy. Comps. Wt. %	3.76	3.56	2.88	3.57
	Lbs.	155.2	139.3	112.3	137.0
	Mols	11.1	10.0	8.0	9.8
	r/ff	1.74	1.05	0.59	0.73

BROWNSVILLE RUN #17 Date	21 Hours 3-4-52		3-6-52	<u>3-7-52</u>	3-8-52	3-9-52	3-10-52	<u>3-11-52</u>	3-12-52	Shut Down 2 PM 3/13 8 Hours 3-13-52
Mols CO Mols CO Conv. Mols CO Unconv. Mols H ₂ from H ₂ /CO ratio % H ₂ Conv. H ₂ Unconv. % H ₂ Unconv. Mols H ₂ Unconv. Lbs.	100	100	100	100	100	100	100	100	100	100
	84.13	67.17	66.46	70.94	63.51	66.30	56.55	63.43	67.78	65.57
	15.87	32.83	33.54	29.06	36.49	33.70	43.45	36.57	32.22	34.43
	180.9	189.4	183.1	188.3	183.6	185.1	184.7	183.1	180.2	183.7
	57.23	41.57	40.26	44.39	35.52	39.19	35.78	38.25	42.5	44.14
	42.77	58.43	59.74	55.61	64.48	60.81	64.22	61.75	57.50	55.86
	77.37	110.67	109.38	104.71	118.39	112.56	118.61	113.06	103.62	102.61
	154.74	221.34	218.76	209.42	236.78	225.12	237.22	226.12	207.24	205.22
In Product										article de descriptor
Wt. % H ₂	3.86	5.52	5.51	5.28	5.95	5.73	5.98	5.77	5.23	5.19
Total Prod. Lbs.	4008.8	4009.7	3970.2	3966.2	3979.4	3928.7	3966.8	3918.8	3962.5	3954.1
Unconv. CO Lbs.	444.4	919.24	939.1	813.7	1021.7	943.6	1216.6	1024.0	902.2	964.0
Unconv. CO Wt. % Calc.	11.08	23.92	23.65	20.51	25.67	24.0	30.66	26.13	22.76	24.37
Wt. % Actual	10.99	22.75	23.49	20.35	25.48	23.84	30.44	25.91	22.58	24.18
CO_2 Wt. % CO_2 Lbs. CO_2 Mols (CO to CO_2)	22.32	23.38	22.49	24.91	21.7	23.61	17.06	21.70	20.86	17.77
	894.8	937.5	892.9	988.0	863.5	927.6	676.7	850.4	826.6	702.6
	20.3	21.3	20.3	22.5	19.6	21.1	15.4	19.3	18.8	16.0
Water Wt. % Lbs. Mols. H ₂ to H ₂ O Mol % C ₃ + Wt. % (incl. wsc) Lbs. Mols CH ₂ (CO to C ₃ +)	19.87	11.97	12.10	12.53	12.23	12.25	13.73	13.57	16.98	20.88
	796.5	480.0	480.4	497.0	486.7	481.3	544.6	531.8	672.4	825.6
	44.2	26.7	26.7	27.6	27.0	26.7	30.3	29.5	37.4	45.9
	24.4	14.1	14.6	14.7	14.7	14.4	16.4	16.1	20.8	25.0
	14.50	10.97	10.73	10.18	8.54	9.66	8.34	7.64	9.49	7.76
	581.3	439.9	426.0	403.8	339.8	379.5	330.8	299.4	376.0	306.8
	41.5	31.4	30.4	28.8	24.3	27.1	23.6	21.4	26.9	21.9
CH ₄ Wt. %	3.16	2.05	3.33	3.38	2.68	2.76	2.01	3.04	2.40	2.21
Lbs.	126.7	82.2	132.2	134.0	106.6	108.4	79.7	119.1	95.1	87.4
Mols (CO to CH ₄)	7.9	5.1	8.3	8.4	6.7	6.8	5.0	7.4	5.9	5.5
C ₂ Wt. % Lbs. (CO to C ₂)	2.41 96.6 6.7	1.94 77.8 5.4	2.04 81.0 5.6	2.00 79.3 5.5	1.76 70.3 4.8	1.83 71.9 5.0	1.39 55.1 3.8	1.73 67.8 4.7	1.88 74.5 5.1	64.5 4.4
Oxy. Comps. Wt. %	2.87	2.60	2.57	2.95	2.24	2.30	2.30	1.93	2.33	2.42
Lbs.	115.1	104.3	102.0	117.0	89.1	90.4	91.2	75.6	92.3	95.7
Mols	8.2	7.4	7.3	8.4	6.4	6.5	6.5	5.4	6.6	6.8
R/FF	1.00	0.51	0.66	0.59	0.56	0.67	0.69	0.67	0.62	0.63

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TABLE II

BROWNSVILLE REACTOR DATA

)	Run #5	Sp. Vel.	C3*Yield*	C ₃ +Yield* BPH/MMSCF FF	Reactor Bed Temp.	Reactor Effluent T.	Reactor Btm. T. Feed	R/FF	Pressure Reactor Top	Cat. Holdup Tons	Cat. Dens.	Bed** Htg.	F.F. to Reactor	Tot. Feed	% Contr.	% H ₂ + CO.	% Fe.	C3+/C1+ Selectivity	(B2) (CO2) (CO) (H2O)
D	1/16/51 17 18	801 1012 972	39.2 42.0 40.3	21.6 19.5 18.9	665 630 647		574 610 639 -No data-	1.23 1.10 1.08	260 305 310	131 137 183	134 137 150	10.6 10.6 10.9	1.81 2.16 2.13	4.04 4.54 4.44	24.71 23.66 22.93	31.9 34.2 33.8	74.2 71.8 74.0		
Proc. Engr. Calcs. Water Sol Chems. 305#/Bbl.	19 20 21 1. 22 23 24	770 782 840 707 761	48.8 50.1 51.2 49.5 49.8	24.6 24.9 23.7 23.2 23.2	659 659 646 612 610	No data	615 610 583 579 554	1.33 1.34 1.20 1.30 1.31	318 318 320 320 320	177 178 180 203 210	149 150 152 145 154	12.8 12.8 12.8 14.9 14.0	1.98 2.01 2.16 2.13 2.14	4.63 4.70 4.75 4.90 4.95	32.21 34.47 31.87 28.15 30.22	50.6 53.7 50.6 44.5 48.6	75.0 74.3 74.0 72.7 76.8	No data	No data
Ave.	2 days 17 & 18	992	41.2	19.2	639		625	1.09	318	160	144	10.8	2.15	4.49	23.30	34.0	72.9	No da t a	No data
Ave. 1st day S	5 days 20 to 25 Sep.	772	49.9	23.9	637		588	1.30	319	190	150	13.5	2.08	4.79	31.38	49.6	74.6	uubu	ua da
	Run #6 3/13/51 14 15 16 17 18 19	1365 1347 1222 753 772 840	35.6 61.0 59.8 55.7 54.0 49.1	11.8 19.1 18.5 18.6 17.6 16.1	660 693 653 645 664 656 659	660 675 670 645 666 655 660	624 650 630 610 575 515 505	0.59 0.51 0.42 0.53 0.60 0.61	310 305 305 305 305 305 290	131 118 121 153 189 185 183	128 104 99 105 104 102	11.0 12.2 13.1 15.6 19.5 19.5	3.03 3.20 3.24 2.30 3.06 3.05	4.80 4.60 4.80 4.60 4.90 4.90	34.85 34.65 37.30 12.18 32.10 29.40	53.1 57.6 58.7 58.8 53.9 53.9	75.0 75.2 75.8 73.3	No data	No data
Ave.	2 days 14 & 15 3 days 17,18,19	1285 788	60.4 52.9	18.8 17.4	673 660	673 660	640 531	0.47 0.58	305 300	120 186	102 103	12.7 19.4	3.22 2.80	4.75 4.80	35.98 24.6	58•2 55•5	75.0 74.6	No data	No data

Run #7	
4/8/51 9 10 11 12	Data not worked up.

^{*} Including water sol. chemicals. Actg. Dept. WSC=318 #/Bbl.

^{**} Effective reactor area = 187 sq. ft.

<u>Run #8</u>		% H ₂	C3+ Yield BP#	C3+ Yield BP H / MMSCF	Reactor Bed Temp.	Reactor Effl. Temp.	Reactor Btm. T.	R/FF	Pressure Reactor Top	Cat. Holdup Tons	Reactor Vel.	Cat. Dens.	Bed Htg.	FF to Reactor MMSCF	Tot. Feed	% Contr.	% H ₂ + CO % Conv. Fe	C ₃ †/ C ₁ + Select.	(H2) (CO2) (CO) (H20)
4/22/51 23 25 26 27 28 29 30	1197 1272 1093 1141 1125 1148 1237 1030	. •	74.9 79.15 87.39 93.36 77.1 96.11 95.07 83.94	25.1 25.4 28.4 28.9 24.4 29.8 30.0 26.0	675 675 680 670 675 670 685 678	680 678 670 684 668 670 675	270 285 281 360 375 380 380 380	1.50 1.40 1.36 1.29 1.32 1.21 1.31	365 365 358 355 358 359 360 360	132 122 146 147 142 138 126 154	0.50 •54 •50 •52 •50 •50 •50	112 107 112 114 108 105 105	12.6 12.2 14.0 13.8 14.1 14.0 12.8 15.2	2.981 3.116 3.074 3.231 3.157 3.222 3.168 3.224	7.463 7.507 7.240 7.406 7.348 7.148 7.342 7.295	47.5 46.8 47.1 43.8 50.1 50.5 47.0 48.3	66.1 75.6 64.2 74.3 68.5 70.7 75.4 68.4 76.7 69.2 75.6 65.2 74.9 69.2	· · · · · · · · · · · · · · · · · · ·	
5/1/51 2 3 4 5 6 7 8 9 10 11 12 13 14	1001 1085 1142 1029 1027 1087 1146 1186 1188 1183 1081 941 979		87.56 85.82 86.53 82.83 91.98 94.95 83.89 79.22 82.46 68.86 64.49 57.21 55.19	27.3 25.8 25.8 27.8 27.8 24.1 24.2 23.8 21.6 19.6	675 680 671 668 673 672 677 660 672 669 678	670 696 676 682 663 675 671 676 669 685 674	382 370 380 380 370 425 420 395	1.19 1.21 1.22 1.29 1.23 1.24 0.91 0.87 1.02 0.93 .51 .68	358 350 355 350 350 350 355 365 350 350 350	162 151 141 151 166 158 149 146 141 145 156 160	•50 •51 •54 •51 •51 •51	110 106 104 106 112 110 110 108 106 107 123 120 114	15.7 15.3 14.5 15.8 15.4 14.5 14.5 14.5 13.9 15.2	3.205 3.326 3.382 3.216 3.307 3.399 3.460 3.465 3.159 3.062 2.924 2.973	7.010 7.353 7.528 7.374 7.394 7.477 6.514 6.467 6.850 6.715 7.100 4.628 4.906 4.931	48.3 48.8 47.8 47.4 48.4 47.3 37.3 36.1 31.8 334.9	79.9 74.3 71.5 70.0 74.7 69.9 73.9 71.6 74.0 71.0 70.7 67.4 73.2 63.8 73.9 61.7 74.0 62.5 73.2 63.5 73.7 63.0 73.2 61.0 74.9 61.4 73.8	76.3 76.9 78.0 79.3 79.17	2.402 2.336 2.265 2.666 2.666 2.697 3.582 3.878 3.110 3.462 3.669 4.092 3.374
15 days Ave 22 50 7 8 days	1115		86.91	27.05	676	677	352	1.29	357	145		109	14.3	3.211	7.342	47•7	69.0 74.6	78.05	2•508
8 days Ave 7 to 15 5 days	1108		71.7	22.1	673	673	489	0.91	352	152		113	14.4	3.229	6.014	36.5	63.0 73.7	72.94	3.631
7 to 11 3 days	1167		78.49	23.46	671	671	402	•93	353	146		111	12.3	3.373	6.729	37.2	63.7 73.6	74.46	3.582
12 to 14	1000		58.96	19.76	676	676	605	•62	350	160		116	14.8	2.986	4.821	35.7	61.8 74.0	70.39	3.711
Run #9 6/5/51 10 Hrs. 6 7 8 9 2 days	1411 1384 2146 2347 2471	57.71 48.97 41.54 45.50 47.09	57.1 97.4 98.3	17.8 20.3 19.5 20.6 20.8	675 660 650 650 650	648 643 630 641 644	675 630 325 350 385	0.80	290 350 350 350 350	152 156 147 139 139	0.58 .52 .88 .98 .92	160 160 141 139 156	10.2 10.4 11.2 10.7 9.6	2.825 2.814 5.003 4.769 4.806	3.716 4.212 7.779 8.609 8.196	42.2 35.4 30.6 35.5 37.1	64.1 81.6 60.1 79.6 51.9 78.4 53.0 77.5 58.5 77.1	70.16 67.78 81.23 75.65 73.43	
Ave 5 and 6	1398	b	53.7	19.1	668	6 46	652		320	154		160	10.3	2.820	3.964	38.8	62.1 80.6	68.97	3.267
3 days Ave 7,8 and 9	2321		98.5	20.3	650	638	353	0.82	350	142		145	10.5	4.857	8.195	34•4	54.5 77.7	76.57	5.63

	* * * ;	1 200																				
i e																						
<u>Run #10</u>	CO Conv	Sp. Vel.	H ₂ Conv.	C 3 ⁺ BP H	C3+ BPH/ MMSCF	Reactor Bed Temp.	Reactor Effl. Temp.	Reactor Btm. T. Nozzle O	R/FF	Pressure Reactor Top	Cat. Holdup Tons	Reactor Vel.	Cat. Dens.	Bed Htg.	FF to Reactor MMSCFH	Tot. Feed	% Contr.	% H2+CO C	Conv.	% Fe	C3/ C1+ Select.	(H ₂) (CO ₂) (CO) (H ₂ O)
24 Hrs. 7/21/51 22 23 24 25 26 27 28 29 30 31	76.7° 77.0	7 699 812 761 769 791 736 827	51.84 61.87	51.3 62.2 No data	20.4 22.1	600 620 600 600 610 600 590 600	600 630 630 614 636 639 626 627 591 628 644	1.55	1.69 1.46 1.39 1.47 1.38 1.63 1.63 1.60 1.84	350 350 350 350 350 350 350 350 350	254 252 248 245 233 230 223 218 213 209 208	.86 .77 .78 .81 .81 .82 .84 .81	150 158 158 160 156 150 144 144 144 142	18.2 17.1 16.8 16.4 16.0 16.4 16.2 15.8 15.7	2.51 2.82 2.61 2.57 2.68 2.53 2.70 2.96 2.67 2.51 2.48	6.76 6.48 6.25 6.37 6.40 7.04 7.10 6.88 7.14 7.06	38.7 39.5 38.0 40.0 36.1 35.5		60.54 66.85	88.6 84.7 84.1 83.5 83.5 83.0 83.7	76.81 70.96	2.66 2.40
27 28 29 30 31	68.6 72.5 54.9	919	51.33 52.42 41.08	58.9 52.8 47.2	22.0 21.0 19.1	600 620 650 640	627 591 628 644	455 460 470 475	1.32 1.60 1.84 1.85	350 350 350 350	218 213 209 208	.81 .81 .85 .67	144 144 142 143	16.2 15.8 15.7 15.6	2.96 2.67 2.51 2.48	6.88 6.85 7.14 7.06	36.1 35.5 36.9 32.4 33.3 37.9		57.60 59.65 46.	83.7 81.4 81.4 81.5	76.24 74.57 89.04	2.26 3.39 1.222
Ave 4 days (1 day run lst day Sep.	L1 n)	816		55.3	21.1	633	624	465	1.69	350	221		147	16.1	2.62	6.91	36.1		57.5	82.3	77•52	2.39
Run #11 6 AM 11/11/ 12 13 14 15 16 17 18 19	51 85. 88. 86. 81. 82. 80. 80.	4 101 4 101 7 107 5 86 8 98	63.3 67.7 67.2 65.0 61.3 63.4 58.7 62.0 63.8	75.2 69.4 75.2 80.5 82.1 69.3 82.4 83.0	28.1 23.1 24.4 26.0 26.7 23.3 26.3 27.3 25.7	647 660 655 645 630 650 620 640 655	660 655 655 660 630 630 590 650	495 560 430 431 475 412 443 465 410	1.49 1.37 1.31 1.42 1.53 1.43 1.13 1.36 1.16	350 350 350 350 350 345 350 350	231 226 220 213 210 203 262 255 248	.83 .84 .88 .89 .91 .84 .87	152 154 152 150 148 154 158 168	16.3 15.7 15.5 15.2 15.2 14.1 17.8 16.3 15.8	2.68 3.00 3.08 3.10 3.07 2.99 3.05 3.09 3.23	Gradual decrease	55.3 56.2 55.4 53.4 53.8 49.0 49.0	6.68 7.12 7.13 7.51 7.16 7.25 6.50 7.11 6.96	71.0 74.9 74.0 71.8 68.3 70.4 66.6 68.8 69.9	85.1 83.5 82.7 82.6 81.8 81.0 80.1	74.35 75.51 75.56	3.42 2.89 2.63 2.43 3.10 2.44 2.62 2.30 2.35
8 days Ave 12 20 lst day Se	& p.	99		77•9	25.4	644	641	453	1.34	350	230		157	15.7	3.08		52.2	7.09	70.6	81.9	74.55	2.68
	>																					

CO Run #12 Conv.	Sp.	H ₂ Conv.	C3+ BPH	C ₂ + BPH/ MMSCF	Reactor Bedl. Temp.	Reactor Effl. Temp.	Reactor Btm. Temp.	R/FF	Pressure Reaction Top	Cat. Holdup Tons	Reactor Vel.	Cat. Dens.	Bed Htg.	FF to Reactor MMSCFH	Tot. Feed	% Contr.	% H ₂ + CO Conv.	% Fe	C3 [†] /C1 [†] Select.	(H ₂) (CO ₂) (CO) (H ₂ O)
11/29/51 82.41 30 85.28 12/1/51 85.87 2 85.11 4 84.48 5 84.34 6 86.13	1040 1081 832 882 891 891	59.53 65.36 66.92 66.71 66.56 64.61 67.45	68.3 81.6 73.1 76.1 76.1 71.8 68.5	23.5 26.9 24.2 25.4 26.2 25.6 24.5	650 650 660 650 630 630	618 648 628 638 645 657 670	455 468 478 473 470 455 447	1.60 1.41 1.31 1.41 1.40 1.52 1.50	350 350 350 350 350 350 350	215 210 210 281 252 246 243	•97 •94 •89 •91	162 160 168 168 166 166 166	14.2 14.1 13.4 17.9 16.3 15.9	2.91 3.03 3.02 3.00 2.90 2.81 2.80	7.575 7.298 6975 7.222 6950 7.067 6999	50.41 53.75 50.46 51.07 50.21 49.76	67.69 72.51 73.62 73.19 72.85 71.50	85.9 86.1 85.4 82.6 81.6	76.22 78.92 76.76 77.93 76.99	3.00 2.49 2.31 2.17 2.24 2.74 2.70
7 85.90	918 956 820	67.29 65.21	66.1	24.2	650 650	650 648	460 470	1.42	350 350	236 266	89	168 168	13.4	2.73	6598 6516	50.92 41.60 51.60	73.84	80.9 80.5	66.37 71.40	2.70 2.69 3.22
Ave 29 & 30 - 2 days 6 days Ave 1 to 8	1061		75.0	25.2	650	633	462	1.51	350	/2 13		161	14.2	2.97	7436	52.1	70.1	86.0	77.60	2.75
Ave I to8	895		72.0	25.0	645	648	464	1.43	350	245		168	15.4	2.88	6969	49.0	73.1	82.1	75.59	2.48
Run #13 12/21/51 81.25 22 84.28 23 82.84 24 79.98 25 80.81 26 79.50 27 79.99 28 77.27 29 74.10 30 75.66 31 76.03 1/1/52 76.53 2 76.85 6 days Ave 22 to	1004 1050 1043 993 944 971 1128 1115 1071 1044 947 1039 1000	61.55 64.65 63.24 59.22 59.37 60.26 53.38 51.27 53.52 53.57 54.45 55.09	77.6 73.5 68.4 71.9 76.5 81.0 63.3 65.7 58.4 61.6	27.3 26.6 22.8 23.4 24.1 27.5 25.5 20.0 20.4 21.9 19.8 21.3 20.5	640 620 625 635 635 620 615 612 620 610 660 654	615 606 617 640 640 630	545 455 477 425 450 380 450 480 480 480	1.55 1.65 1.38 1.48 1.65 1.42 1.47 1.41 1.47 1.40 1.28	330 350 350 350 350 350 350 350 350 350	195 188 200 204 211 210 200 192 183 190 203 200 215	1.0 .92 .90 .88 .95 .93 .98 .94 .92 .91 .84	150 148 150 150 148 148 146 144 144 144	13.6 14.6 15.3 15.3 14.6 14.8 14.9 14.9	2.84 2.76 3.00 2.93 2.98 2.79 3.17 3.03 2.99 2.92 2.98 3.02	7247 7345 7144 6984 7355 7400 7359 7678 7483 7221 7205 7165 6888	44.3 54.4 50.6 46.8 47.0 46.8 43.7 43.1 42.8 43.7 43.7	68.56 71.49 70.08 66.50 66.92 66.91 65.73 61.63 59.16 61.23 62.61	84.9 85.4 80.4 80.9 80.3 80.3 80.3 80.3	80.76 77.40 78.52 78.72 77.94 81.59 79.98 76.52 80.92 78.66 79.02 78.88 78.58	2.30 2.87 2.06 2.10 2.29 2.19 2.57 3.03 2.91 2.93 2.70 2.88 2.61
Ave 22 to 28 6 days	1022		73.3	25.0	626	625	452	1.48	350	202		148	14.6	2.94	7260	48.8	67.9	81.6	79.00	2.36
Ave 28 to	1036 1029		62.3	20.1	629	625	454	1.41	350	197		147	14.3	3.02	7273	43.2	61.3	80.4	78.76	2.86

CO Sp. Run #14 Conv. Vel.	H ₂ C ₃ + C ₃ Conv. BPH M		Reactor Bed Temp.	Reactor Effl. Temp.	Btm.	R/FF	Pressure Reactor Top	Cat. Holdup Tons	Reactor Vel. Ft./Sec.	Cat. Dens.		FF to Reactor MMSCFH	Tot. Feed	% Contr.	% H2 + C0 Con v .	% Fe	C3 [†] /C1 [†] Select. %	(H ₂) (CO ₂) (CO) (H ₂ O)
1/9/52 83.53 791 10 83.27 801	63.51 72.0 61.74 61.4	24.1	639 671	605 620	630 480	1.80	350 350	225 210 200	•95 •96 •92	146 144 144	16.5 15.6 14.9	2.65 2.55 2.65	7.44 6.99 7.04	50.15 51.11 46.34	70.49 69.32 68.06	82.9 79.7 79.8	81.0 74.9 76.1	2.375 3.204 3.227
11 83.00 876 12 82.09 872 13 81.92 894	60.10 60.3 57.05 65.9 57.85 59.5	25.9 23.7	676 676 682	660 660 635 660	500 525 490	1.65 1.68 1.64	350 350 350	190 181	.90 .87	142 140	14.3 13.8	2.54 2.50	6.80 6.61	48.68 47.64	65.85	79.2 79.4	78.4 76.0	4.614 4.10
14 83.25 867 15 86.78 850 16 86.54 924	61.46 57.9 67.25 59.2	24.1 25.6 24.0	669 667 671	660 650 650	475 495 460	1.87 2.00 1.95	350 350 350	172 164 156	.86 .90	136 134 130	13.5 13.1 12.8	2.40 2.31 2.38	6.90 6.93 7.03	48.02 56.76 52.26	68.98 74.06 74.00	79.9 80.0 79.7	76.9 77.5 78.8	3.207 2.396 2.066
17 85.92 1044 18 83.88 1003	67.21 57.0 62.66 59.9 61.78 58.5 62.83 59.9	23.1 23.3 23.0	674 667 669	660 645 635	470 445 470	1.45 1.75 1.80	350 350 350	152 143 115		132 124 130	12.3 12.3 11.5	2.59 2.51 2.60	6.35 6.91 7.28	48.31 49.48 48.99	70.80 69.50 70.55	79.4 79.5 80.2	76.2 77.3 75.9	2.967 2.453 2.738
20 85.53 1220 21 80.41 1734	63.42 60.9 56.91 63.5	22.2 18.9	679 669	645 645 595	470 450 445	1.60 1.13 0.99	350 350 350	122 106 110	2.8	120 116 134	11.0 9.5 8.8	2.74 3.35 3.77	7.02 7.13 7.43	48.64 42.86 38.11	71.14 65.00 57.56	79.0 79.8 80.6	80.7 76.1 78.0	2.738 2.079 2.45 2.42
22 74.34 2115 7 days	48.71 64.4	17.0	653	777	447			110	2.00	-)+	0.0	2011	1 • 42)0 4.11	71.00		, , ,	
Ave 10 to 17th 869 3 days	60.2	24.3	673	665	489	1.79	350	182		139		2.48	6.90	50.1	69.5	79 .7	76.9	
Ave 17,18 19 1008	59•4	23.1 7 8	670	647	462	1.67	350	137		129		2.57	6.85	48.9	70.3	79•7	76.5	
Others Sep. Run #15		eg. pro																
20½ Hrs.		Aw		4	.				0.40	7	10.0	2 02	4 50	E E 07	d1 2	70 A	56.67	
2/2/52 91.19 1168 3 83.70 1220 4 82.89 1280	80.35 71.0 62.05 68.5 60.15 74.5	24.3 650 23.9 690 18.8	650 645 640	645 635	695 490 50 0	1.57 1.35 0.73	350 350 350	141 137 133	0.82 .84 .88	128 126 122	12.3 11.5 11.3	2.92 2.86 3.95	6.52 6.70 6.83	55.07 46.19 44.61	84.2 69.7 68.1	78.0 80.3 79.4	75.15 77.02	
5 79.81 1587 6 78.04 1769	51.31 67.9 50.58 81.4	18.2 675 18.3 700	650 690	655 650	495 470	0.93	3 <i>55</i> 380	131	.90 0.98	120 114 112	11.5	3.73 4.45	7.21 7.82	38.03 40.72 40.37	61.3	79•9 77•5	67 .99 70 . 96	
8 78.41 1676	53.68 75.5 60.52 47.7	24.6 700 18.4 700 20.2 675	650	650 625	540 165 675	0.65 1.53 1.65	250 250 242 350	140 124	1.18 1.0	124 116	12.0 11.5 12.5	4.10 2.36	6.76 5.98	29.42 48.75 39.80	62.3 68.5 67.7	78.3 81.9	65.99 75.54	
9 83.30 998 10 82.62 1043 10½ Hrs.	59.92 54.4	20.2 675 20.4 675	600	640				136	0.86	116		2.66	7.06			78.5	72.64	
11 84.40 1114	61.59 57.4	22.8 675	630	640	470	1.78	350	132	0.87	128	11.0	2.52	7.02	45.97	69.6	78.2	74.60	
4 days Ave 3,9,		•								3.00		0.60	4 40	15.0	44 O	70 7	71 6	
10 & 11 1094 4 days	57.0	21.8	594		450	1.58	323	132		122		2.60	6.69	45.2	68.9	79.7	74.6	
4,5,6 & 8 1578 5 day period	74.8	18.4	658		501	0.77	334	133		120		4.06	7.16	38.2	63.0	78.8	70.5	
1st day Sep.																		

H2CO/ in FF	Run #16	CO Sp. Conv. Vel.	H ₂ Conv.	C3+ Yield BPH	C3 Yield BPH/ MMSCF FF	Reactor Bed Temp.	FF Preh e at	R/FF	Pressure Reactor Top	Cat. Holdup Tons	Reactor Vel. Ft./Sec.	Cat. Dens.	Bed Hgt.	FF to Reactor MMSCFH	Tot. Feed	% Contr.	% H ₂ + CO Conv.	% Fe % C	C ₃ +/ C ₁ + Select.
1.780 1.824 1.836	17½ Hrs. 2/19/52 20 21 3½ Hrs. 22	82.44 818 81.38 1283 *76.81 1808 75.19 1606	56.98 55.58 45.89 43.54	51.9 66.9 75.9	22.2 20.2 16.8 18.8	650 680 650	470	1.74 1.05 0.59	350 350 350 350	153 142 134 137	0.78 0.81 0.79	120 120 114 134	13.6 12.7 12.7	3.30 4.51	6.40 6.75 7.16 7.04	58.1 43.1 35.0	66.1 64.7 56.8 54.3	80.9 79.3 77.6 80.5	72.7 67.8 69.8 71.1
H2CO/ in FF	Run#17																		
1.809 1.894 1.831 1.883 1.836 1.851 1.847 1.831 1.802	21 Hrs. 3/4/52 5 6 7 8 9 10 11 12 8 Hrs. 13	84.13 1414 67.17 2516 66.46 2357 70.34 2726 63.51 2552 66.30 2696 56.55 2614 63.43 2846 67.78 2708	57.23 41.57 40.26 44.39 35.52 39.19 35.78 38.25 42.5	63.4 68.8 63.5 63.3 53.7 60.1 55.1 50.4 62.6	20.8 15.3 15.1 14.3 12.1 13.7 12.2 10.9 13.7	685 500 650 660 675 690 665 650 645	640 420 580 635 640 635 635 635 635	1.00 0.51 0.66 0.59 0.56 0.67 0.69 0.62	350 350 375 375 375 375 375 370 375	91 73 68 51 83 67 50 72 55		94 92 86 64 104 88 58 98 72 60	10.5 8.75 8.25 8.00 8.50 8.00 8.25 7.25	4.20 4.44 4.42 4.38 4.52 4.62 4.56	6.08 6.76 6.98 7.06 6.92 7.31 7.65 7.70 7.40	47.06 28.52 32.77 31.48 28.90 28.26 24.08 24.55 30.68	66.80 50.42 49.52 53.60 45.39 48.70 43.08 47.14 51.52	78.9 9.0 no samples	72.25 78.23 66.64 65.42 65.82 67.81 71.08 61.60 68.73

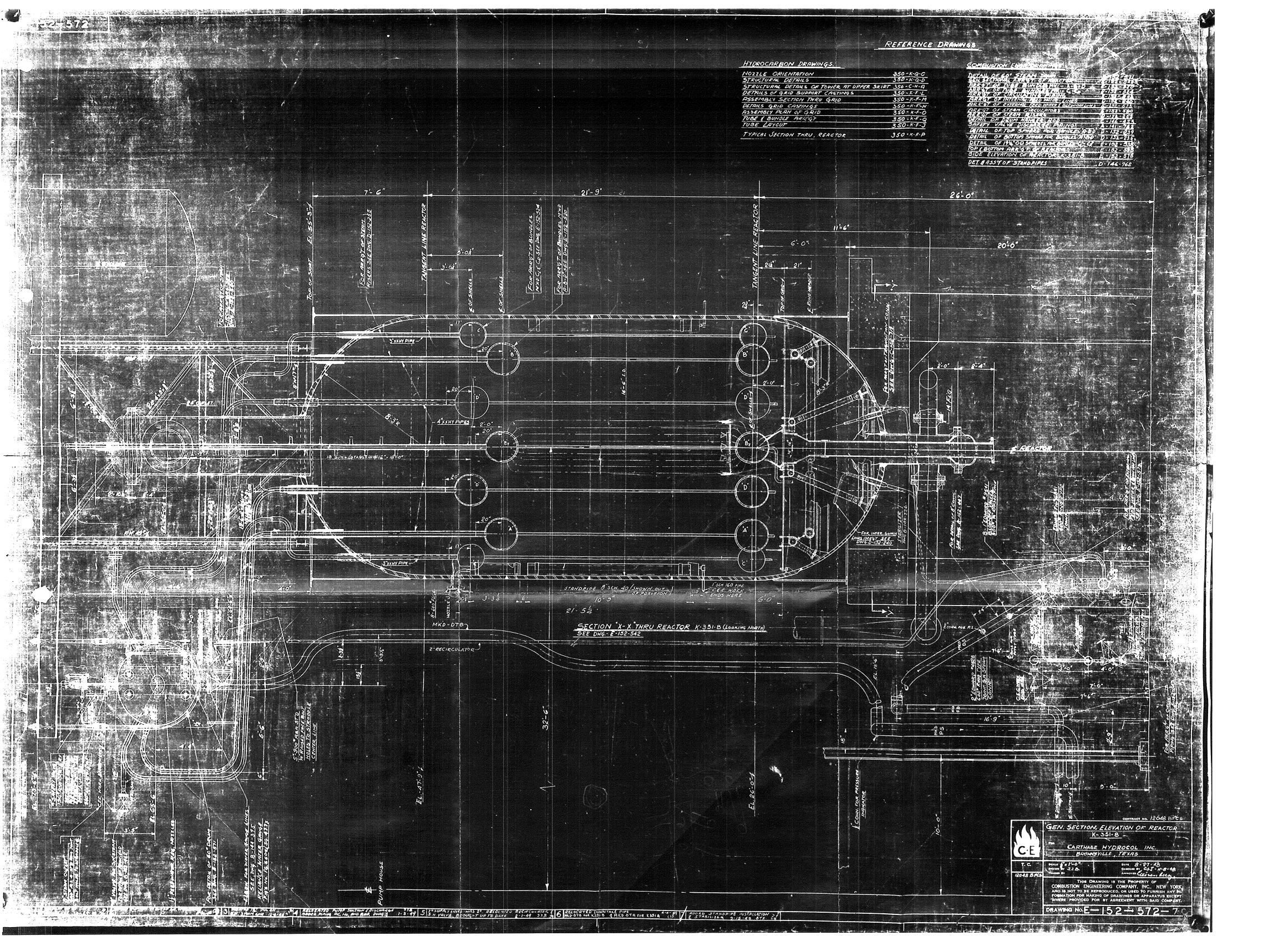
^{*} Started reducing preheat temp. stepwise at 3 PM.

PR TDC802 37 P

TABLE V

Run No.	Sp. Vel.	C ₃ +#/ MMSCF Y H ₂ +CO Fed.	% H ₂ †CO Present <u>In Gas</u>	C3 #/MMSCF	C ₃ +Bbls./	H2 [†] GO Conv.%
46-1	2825	5960	91.51	5454	22.82	58.4
46-2	2314	6460	n	5912	24.84	64.6
46-3	2178	6640	11	6076	25.52	67.5
45-1	1646	7070	H.	6469	27.18	71.4
49-2	1215	8210	n	7513	31.56	76.4
49-1	1072	8370	n	7659	32.18	78.0
29-3/6	939	8570	11	7842	32.95	85.1

^{*} Use ave #/Bbl. of $C_3^+ = 238 \#/Bbl.$



SUMMARY OF OPERATIONS

					charge Rate gn Capacity		Production Barrels Raw Chemicals to	Type of Gas	• •	Catalyst arge		
Rus No a		Shutdown Date	Days Duration	Plant (for 1 Unit)	Generator	Reactor (for 1 Reactor)	Primary Oil Stanolind (Total Run) Lbs. per Day	Generator		d Catalyst	Cause of Shutdown	Corrective Steps Taken Prior to Succeeding Run
]	1 8-24-50	8 –26– 50	2	52 (58)	19 (23)	-	Nil	First Slot Type	0	0	Blade failure Steam Turbine Oxygen Plant V-201 B Air Compressor	 Replacement of zirconia brick lining of generator combustion space with Alundum. Oxygen slots of burners were filled by welding, and replacing by drilling smaller (5/16*) holes.
	2 9-16-5 0	9-17-50	1	63 (70)	21 (31)	-	N11	Modified First Slot Type Field Drilled Oxygen Holes	0	0	Fire, resultant from split in lh* synthesis gas discharge line to B Reactor.	1. Extensive steps taken in Section 350 to avoid trapping of catalyst and to protect metal walls where it occurs, such as elimination of dead ends, removal of insulation from feed lines, installation of check and control valves, and new purge connections and lining inside bottom of reactors with insulating cement. 2. Repair of fire damage. 3. Changes in pipe work and instrumentation in Section 300 to avoid spentaneous fires in synthesis gas lines. 4. Installed gas oil flushing connections to eliminate plugging in M-352 Exchangers.
3	10-28-50	11-3-50	6	67 (72)	29 (31)	30 (-)	690	Same	0	0	Failure of Gas Generator Burner cooling water supply.	 Installed emergency boiler feed supply to burner cooling water circuit and continuous use of 2 pumps in this service. Replaced corroded 25-20 soot blow with chrome alloy and patched baffles and initiated engineering of water cooled soot blowers and refractory protected baffles. Installed new slot type burners with 3 rows of oxygen holes and placed castable refractory on burners.

tor to permit reduction of catalyst between runs.

5. Resurfaced gasket area of M-352 Exchangers.

				Charge Rate - 1% of Design Capacity Reached			Production			Type of Catalyst		
Run No.	Starting Date		Days Duration	Oxygen Plant (for 1 Unit)	Generator	Reactor (for 1 Reactor)	Barrels Raw Primary Oil	Chemicals to Stanolind Lbs. per Day	Type of Gas Generator Burner		Cause of Shutdown	Corrective Steps Taken Prior to Succeeding Run
.	12-28-50	1-1-51	ħ	71 (73)	30 (33)	30 (-)	1461	-	Second Slot Type 3 Rows Oxygen Holes	50 Tons A.W. used in Run#3 142 Tons Raw A.W. 30 Tons 76% Fe A.W. 75 Tons 91% Fe A.W.	Failure of Thrust Bear- ing V-202 B Oxygen Compress or	 Enlarged Balance Piston and Thrust Bearing on Oxygen Compressor V=202 B and made certain changes to oil system. Installed cross-over piping so that the A & B Oxygen Compressors could be used on opposite units. Retubed V=201 A Surface Condenser. Installed Centrifix in low pressure Tower A Oxygen Unit. Patched generator baffles and connected steam purge to oxygen header.
5	1-16-51	1-25-51	9	72 (76)	33 (35)	47 (63)	4167	36,840 (57,000)	during	24 44 140 Tons used in #4 (40 Tons MS Red. (to 88%`Fe g(40 Tons MS Red. (to 90.5%	Failure of Baffles and Soot Blowers due to corrosion.	 Retubed V-201 B Surface Condenser and made miscellaneous changes to attempt increased capacity of oxygen unit. Installed new water cooled soot blowers and refractory protected baffles in Gas Generator. Installed new type slot burners with 2 rows oxygen holes. Removed all steam bundles from both reactors, removed baffles from bundles, walded up holes in bundles and reassembled reactors without baffles.
6	3-11-51	3-20-51	9	80 (83)	38 (39)	58 (66)	6287	62,200 (68,800)	Holes	100 80 115 Tons Freshly reduced Mill Scale + 55 Tons red. cat. 100 Tons Mill Scale of red. cat. added during run.	Failure by burn-out of Gas Generator #5 Burner.	 Plugging end oxygen holes to decrease exposure of slot cooling chamber. Repaired castable refractory on #3 Baffle.

^{*} Figures not in parenthesis are average for run.

^{*} Figures in parenthesis are maximum attained during run.

IN S ERT

0 2 PJS.

SUMMARY OF OPERATIONS

					Rate - % capacity Res		Production			Type of Catalyst					
Run No.	Starting Date	Shutdown Date	Days du- ration	Oxygen Plant (for 1 Unit)	Generator	Reactor (for 1 Reactor)	Primary 011	Chemicals to Stanolind Lbs. per day	Type of Gas Generator Burner	ed Catal	% Reduc- yst. Finish	Cause of Shutdown	Corrective Steps Taken Prior to Succeeding Run.		
7	4-8-51	4-13-51	6	* 76 (83)	(†0) *	56 (62)	4,450	79,300 (1) (85,000)	Third Slot Type - plug- ged end holes	80 .	86	Sticking of air revers- ing valve Oxygen Plant - Oxygen Compressor kick- ed out.	 Installed Perlite packing in Regenerator "A" Oxygen Unit. Installed ferrules in holes of grid of "A" Reactor. 		
													4. Modified seal system of V-202-B Oxygen Compressor.		
8	4-21-51	5 -15-5 1	5/1	83 (86)	цо (58)	62 (65)	22,391	103,000 (1) (134,000)	Same		+ 57 of	Failure of Gas Generator Burner.	 Installed fourth slot-type burners (V-slot). Replaced castable refractory on center and water cooled baffles of generator. Repaired finger baffles. Installed ferrules in holes of grid of "B" Reactor. 		
9	6-5-51	6-9-51	Ţ	61 72 (67) (74)	52 (62)	80 (100)	3,059	74,000 (1) (109,500)	Fourth Slot Type - V Slot	100(2)	100	Bad start-up. Exces- sive soot blowing caused by by-passing of generator baffles.	 Installed Enco finger baffles. Reverted to use of Third type slot burners. Installed new rotors in Recycle Compressors for higher compression ratio. Modified seal system V-202-A Oxygen Compressor. 		
10	7-20-51	8-1-51	11	70 (72)	25 (26)	50 (52)	11,692(a)	66,630 (1) (75,200)	Third Slot Type - Same as Runs 7&8	100(3)	100	Hot spot in Boiler section of Generator.	 Completely dismantled generator internals. Installed new brick piers and incomel shields at water tube wall joints. Installed new incomel baffles in steam bundles. 		
11	11-10-51	11-20-51	10	82 (87)	39 (40)	60 (63)	12,517(b)	98,250 (1) (106,800)	Third Slot Type - Inconel	100	100	Power failure shutting down Oxygen Plant. (Burners found with carbon toadstools and badly burned in Oxygen slots.)	 Installed emergency power generator for Oxygen Plant. Decided use 5% Steam in gas to generator and up to 500°F gas preheat and to raise temperature burner cooling water - all in effort reduce carbon forming tendency. Replaced burners with new third slot type burners of same design. 		

FAM

^{*} Figures in parenthesis are maximum for run, those not in parenthesis are average.

(1) Total Chemicals production: Run 7 - 396,300 Run 10 - 737,300

Run 8 - 2,460,159 Run 11 - 982,500

Run 9 - 355,461

⁽²⁾ Theoretical only - batch was not started fresh.
(3) Reduced to 95% Fe.
(a) Includes 3,399 Poly Gaso.
(b) Includes 4,256 Poly Gaso.

¹¹⁻²⁶⁻⁵¹

138

					Charge Rate - 9		Pro	duction		Type of Catalyst		
Run No.		Shutdown Date	Days <u>Duration</u>	Oxygen Plant		Reactor (for 1 Reactor)	Barrels Raw Primary Oil	Chemicals to	Type of Gas Generator Burner	Charge % Reduced Catalyst Start Finish	Cause of Shutdown	Corrective Steps Taken Prior to Succeeding Run
12	11-28-51	12-7-51	9	79 (83)	* 36 (38)	* 58 (60)	9330 (1)	94,000 (2) (112,400)	Third Slot Type - Inco- nel	100 (3) 100	M-352 A&B Exchangers caused loss of Boiler Feed Water to K-301	l. Installed automatic valve to tie generator boiler feed water line into 600 area. To provide boiler feed in emergency.
											Generator steam drum.	2. Installed gunite liners in channel and floating heads of M-352 Exchangers on K-351 B reactor and are Monel lining similar parts of M-352 Exchangers on K-351 A Reactor.
												3. Planning installation of catalyst scrubbers on Reactor Effluent and attempting locate six field boilers to increase 175# steam availability.
13	12-20-51	1-2-52	13	83 (87)	(37)	62 (67)	10,246 (1)	89,500 (2) (113,000)	Third Slot Type - Inco- nel	100 (4) 100	Thrust bearing failure of third stage of V-202 A oxygen compressor.	 V-202 B compressor with 12.5% over sized thrust bearing placed in ser- vice while V-202 A undergoing re- pairs.
												2.V-202 A compressor third stage thrust bearing rebuilt with 50% larger bearing surface.
14	1-8-52	1-22-52	14	$7\frac{A}{1}$ $5\frac{B}{2}$ (78) (58)	36 (48)	(76)	11,205 (1)	88,200 (2) (104,600)	Third Slot Type - Stain- less Steel	100 (5) 100	Failure of "D" cooling water circuit on #3 burner.	<pre>1.Installed improved instrumentation for measuring generator burner pressure drop.</pre>
							The second secon					2.Finished installation acetylene absorbers on both oxygen plants.
												3. Switched from K-351 B to K-351 A reactor in order stainless steel line channel and floating heads on reactor exchanger system (B Unit), since gunnite lining not giving desired corrosion protection.
15	2-1-52	2-11-52	10	70 75 (72) (82)	35 (48)	67 (91)	11,445 (1) 1313 BPD	90,800 (96,000) 16.6% of design	Third Slot Type - Stain- less Steel without capil lary tubes.	duced & 96% Cat.	Leak at upstream flg. of valve on inlet & M-352 A. Leak in #5 Burners. Niggerheads found on all burners. Installed 3 new burners	
16	2-18-52	2-22-52	4	50 75 (68) (78)	38 (49)	71 (91)	3355 (1)	103,000 (106,900) 34.3% of design	Third Slot Type - Inco- nel. No cap- illaries.	vious run and 24	Hot Spot around generator nozzle due to running with two Φ_2 plts. All three burner badly burned at ends of slots.	rs f
17	Started Gen 3-3-52	erator 9 Al 3-13-52 2 PM Burner f Leak in 1	ailure							Entire run with low cat. level. Added 40 Tons 80% Fe poorly reduced cat. & a little more later didn't do any good.		

*Figures in parenthesis are maximum for run, those not in parenthesis are average.

⁽¹⁾ Includes poly gasoline - Run 12 - 3070 Barrels Run 14 - 2817 Barrels Run 13 - 5093 Barrels

⁽²⁾ Total chemicals production Run 12 - 800,000 Run 14 - 1,146,000 Run 13 - 1,165,000

⁽³⁾ Reduced to 96% Fe. - Same catalyst as used in Run 11.

⁽⁴⁾ Comprised 130 tons used catalyst from Run 12 and 127 tons fresh reduced catalyst. 60 tons reduced mill scale and 49 tons reduced and carbided mill scale added during run.

⁽⁵⁾ Comprised used catalyst from Run 13 plus 55 tons fresh reduced (245°F.) mill scale catalyst. Entire batch conditioned before run for 24 hrs. with fresh feed rate of 500 MSCFH in circulating natural gas stream at 620°F. bed temperature.

(3) Stanolind Data

a BLE 1 OPERATING & YIELD SUBLARY RUN #6 K-251-R REACTOR 7.00 PERIOD 13 MAR 14 MAR 15 MAR 16 MAR 17 MAR 18 KAR 19 KAR Operating Conditions Point No. Total Reactor Feed MNSCFH 4.80 4.90 4.60 4.80 4.60 4.90 4.90 FAR Synthesis Gas (1) MMSCFH 3.03 3.20 3.24 3.06 3.05 Recycle Gas MMSCFH 1.77 1.62 Life 1.36 1.85 1.84 Reactor Top Pressure PSI 310 305 305 305 305 305 290 Reactor Gas Feed OT **#20** 516 535 540 440 540 560 OF. Reactor Bed 660 693 653 645 664 656 659 Catalyst Holdup Tons 131 118 121 153 189 185 183 #/CF Catalyst Density 128 104 99 105 104 102 102 % Fe in Catalyst no sample no sample 75.01 75.78 75.17 no sample 73.27 % Reduced Catalyst Charged to Reactor 100 100 100 79.85 88.38 79.85 79.85 Spare Velocity - V/V/Hr. 1365 1347 1222 753 772 840 RESULTS Conversion CO 70.78 74.71 76.51 75.28 73.15 73.76 H2 43.54 48.138 49.24 48.53 43.36 42.49 H27 00 53.08 57.62 58.67 53.89 53.86 58.79 Contrastion 34.85 34.65 37.30 1218 92.10 29.40 Production - Gals 03 +/MSCF 00+H2 FF 0.629 0.737 0.736 0.787 0.674 0.600 Yields - Output Basis Wt. S BPH Wt. BPH Wt. S BPH WIN BPH WT. S BPH WT. S BPH CO 20.80.2 17.82 17.63 16.89 19.01 18.89 H₂ 4.79 5.36 4.59 4.60 5.28 5.21 36-75-25.46 COa 36.75 25,29 32.43 21.94 33.49 23.67 39.1226.44 30.11 2636 ¥2 4.05 3.31 4.33 4.74 5.87 7-51 2.93 9-12 6.97 CHA 6.14 5.10 12.75/1.17 6.03 H. 40 6.57 4.83 C2HL 1.41 1.39 1.21 1.30 1.67 1.57 CoH 0.81 0.80 0.66 0.74 1.11 0.80 C3H6 2.07 11.6 1.72 10.61 1.84 11.29 1.70 9.74 2.06 11.80 1.73 9.98 Calle 0.59 336 0.97 6.10 0.59 368 0.77 4.52 0.42 2.45 0.75 4.41 1.64 7.95 1.68 8.97 1.52 808 Calla 1.40 6.94 1.77 8.76 1.77 8.85 Cultin 0.18 0.92 0.29 1.61 0.59 2.84 0.29 1.50 0.46 2.36 0.34 1.77 C5 + PPO 4.90 4.70 6.42 25.78 6.24 26 14 6.78 24.69 5.19 19.83 3.55 15.04 Water Soluble Chemicals 2.2022417#2.2425167#2.23 2479 2 55 2652 2 69 2790 2 74 2867 13.11 7.10 14.50 7.9, 16.91 7.80 Process Water 10.74 8.34 10.77 8.77 1208 9.02 Tut C3+ Intl 35.63 60.05 52.83 55.73 53.97 42,07 (1) Calculated by Output Basis Dominion 11.76 18.58 18.47 12.00 1764 16.09 (2) Methane Bleed Gas Subtracted

Note; Periods are 6AM To. 6AM, except till 4:50AM March 20 for March 19 Period.

3 77 5

CHI 5-60-4, CS 5-14 3.70 8-38 7.98 767 5.51 5-63 4.05 5-14 3.75 6-66 3.82 6-34 3.52 6-	
PENIOD Point No. Point N	
Operating Conditions	2000
Total Associor Read P1 333 ASSOCIA 3, 922 2, 427 2, 920 5887 4, 900 3.077 3.797 4.051 3.0 4.0 4.3 5.797 Symthesis Gas [1] MSD77H 9.997 4.051 3.197 3.797 4.051 9.99 3.257 1.000 1.00	
Synthesis is in [1] MSDOTH 2.78	3274
Reactor Top Pressure 951 \$45 \$35 \$35 \$35 \$35 \$35 \$35 \$35 \$35 \$35 \$3	3.3 ,
Resolve Top Pressure Fold 1860	3 3 7
Reactor See Feed OF 615 4450 4450 560 405 646 420 625 625 Reactor Beau OF 615 457 477 580 670 675 676 676 675 170 675 Reactor Beau OF 615 477 570 580 670 675 676 675 676 675 170 675 Reactor Beau OF 615 472 172 175 185 172 183 172 185 172 Getelyst beheity Beneity 9/68 112 107 105 112 1/4 108 105 105 105 105 Results Deneity 9/68 112 107 105 112 1/4 108 105 105 105 105 Results Catalyst Charged to neactor 885 313 Share Velocity - V/I, dr. 1/17 1272 1399 1095 114/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 1237 1090 1091 RESULTS Scalingly 764 77.2 15.8 72.9 12.4 641/ 1/25 1144 12.5 12.5 12.5 12.5 12.5 12.5 12.5 12.5	135
Searchar Fac. OP 115 675 670 680 670 675 670 685 171 675 Gatalyst Holdup Tons 732 722 715 744 747 742 728 72 728 72 727 Stellyst Bonalty \$/07 772 775 743 743 743 743 743 747 747 727 746 75 76 72 Figured Catalyst Charged to Reacter 885 913 729 743	
Gatalyst Holdup Tone 72	335
Catalyst Denwity 9/08 1/2 197 105 112 114 108 105 105 105 112 % Fe in Satalyst 7.56 743 743 M maps 75.36 7.71 70.0 74.94 is many 75.36 7.71 70.0 74.94 is many 75.26	-
\$ Reduced Carlyst Charged to heacter 68.5 \$ Reduced Carlyst Charged to heacter 68.5 \$ Spare Velocity - V/Virt.	155
Spare Velocity - V/V, dr	150
Spare Velocity - V/V, dr.	An second
RESULTS Selection 19 76.4 77.2 75.8 72.9 12.4 6.12 78.0 71.7 78.4 71.3 Conversion CO 79.5 77.09 75.57 12.22 82.11 11.71 11.87 77.51 82.35 81.2	
Conversion CO 79.5 77.09 75.57 72.22 83.11 91.77 91.87 77.57 82.25 73.24	1000
Conversion CD 79.5 77.09 75.57 72.22 83.11 81.77 81.87 77.57 82.25 73.24 Ho	The second secon
H2 55.28 51.20 54.42 60.34 60.40 61.65 51.57 61.41 61.55 H2 + 00 66.64 64.20 62.30 63.45 70.72 63.42 69.22 55.55 61.2 70.74 Contraction R	76.3
H_2 + CO	1372
Contraction \$ \$4.8 \$4.1 \$42.1 \$3.8 \$50.7 \$50.5 \$47 \$43 \$41.3 \$8 \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$	13.36
Respect Ratio (total recycle: fresh	7,4
Production-Gala C3+/MSGF C04H3 \$7	48 0
Production—Gals C3+/MSGF COttin PT 0.912 0.905 0.88 0.94 1.150 1.255 1.105 1.132 1.309 1.387 Tields - Output Basis WH BPH WH	1 11
1201cs - Dutput Basis	1.500
A 252 243 229 244 257 240 236 232 246 242 246 242 246 242 246 242 246 242 246 242 246 242 246 242 246 242 246 242 246 246	
CO	
H2 CO2 382 401 410 3.57 3.30 3.57 3.44 501 5.47 5.47 5.47 5.47 5.47 5.47 5.47 5.47	_1 :
CO2 33.16-23.5 31.18 23.41 31.70 22.15 31.40 21.92 32.22 3.01 32.14 22.70 31.04 22.6 3.01 32.40 N2 3.94 3.94 3.10 3.95 3.10	<u> </u>
No. 3.94 3.19 2.07 2.07 3.20 3.20 3.21 3.10 3.51 3.22 CH. 5.60-4.05 5.44 3.70 438 2.98 3.57 5.63 4.05 5.44 3.75 6.65 3.82 5.43 3.70 438 2.89 2.59 4.05 5.44 3.75 6.65 3.82 5.43 3.70 5.45 3.59 6.28 6.28 6.28 6.28 6.28 6.28 6.28 6.28	322
CH, 560=4.05 5.24 3.70 8.38 7.98 7.67 5.57 5.43 4.05 5.47 5.75 6.56 3.82 5.43 5.75 5.43 4.63 5.49 3.59 6.24 6.2 6.24 6.25 6.25 6.25 6.25 6.25 6.25 6.25 6.25	
C2H6 1.41 1.71 1.37 1.93 1.46 1.62 1.70 660 1.59 1.68 C2H6 0.65 0.62 9.76 068 1.76 0.78 0.79 0.79 0.74 0.75 2.72 C3H6 1.52 1.328 1.96 11.58 2.89 13.21 236 14.18 2.30 14.11 2.31 14.12 2.45 15.15 2.33 16.41 2.31 14.12 2.45 15.15 2.33 16.41 2.31 14.12 2.45 15.15 2.33 16.41 2.31 14.12 2.45 15.15 2.33 16.41 2.31 14.12 2.45 15.15 2.33 16.41 2.31 14.12 0.73 4.12 0.13 4	الكائم المعالم
C2H6 C3H6	59 38
C3H6 235 13.28 1.96 11.58 2.09 13.21 236 14.18 2.30 14.11 231 14.12 245 15.52 23 16.07 2.37 C3H8 9.63 3.70 0.64 3.85 1.04 6.69 0.59 3.64 0.62 3.86 0.64 4.24 0.78 4.25 0.73 4.22 0.00 42 0.77 4.15 C1H8 1.91 9.52 1.94 9.94 1.64 8.99 1.73 2.90 2.59 15.74 1.77 237 204 11.07 1.89 12.07 1.88 2.16 2.16	<u> </u>
G3H8 0.63 3.70 OLA 3.85 LON 6.67 0.57 3.66 0.62 3.86 0.61 4.24 0.78 4.20 0.73 4.22 0.23 42: 0.73 4.15 C4H8 1.91 9.52 1.94 9.94 1.64 8.99 1.73 8.98 2.57 15.74 1.77 2.37 200 11.07 1.89 12.07 1.88 2.1 1.35	3.74
1.91 9.52 1.94 9.94 1.64 8.99 1.75 2.90 2.59 15.741.77 2.37 2.00 11.07 1.89 12.07 1.88 2 1.30	15 15
	27 44
CH : 100 A20 140 A20 141 A20 A20 141 A20	8 41
C. H. O. SZ 1.45 0.34 1.00 0.85 1.96 0.73 817 0.92 1.76 0.33 1.86 0.34 1.57 0.75 1.46 0.46	4.43
C. FRPO 9.30 3(2) 33.54/00 42.00 455/01 4451/1/2 50.39 1 30 50.00 43.00	
Water Soluble Chemicals (2) 305 /05/ 314 314 314 315 3717 331 374 357 301 315 402 316 402 316 402 316 402 316	
Process Water (2) N.A. 10.5154 1220 1521 1775 1625 17 1 1839 18.01 118 13.13 17.11 14.52 1534 17.27 1.14 13.02 24.53 13.03	
Statelyst Bei Age - Days 29.7 807 241 261 261 271 281 20.1 21	
(1) Calculated by Gutput Basis	and the same of the same of
	an e e e e e e e e e e e e e e e e e e e
TAN DPH 25.7 25.7 27.8 28.4 28.9 24.4 29.8 30.0 20.0 27.3	25.52

K-351-A REACTOR

TABLE 1
RUN #7 OPERATING & YIELD SUBMARY

4.472

		W-))1-W	REACTOR	RUN 71	ALBITA I TR				·		
PERIOD 1995	3May51	4 May 51	5 May 51	6May 51	7 May 51	8 May 51	9 May 51	10May 51	11 11/251	1211-51	13My 51
Operating Conditions Point No.								The state of the s			
Total Reactor Feed FI 303 MISCFI		4.034	3.932	3959	3 480	3.952	3.985	3,997	4.437	4.628	4.906
Synthesis Gas (1) MMSCFI	3.382	3.216	3,307	3.339	3,402	3.460	3.390	3,465	3.159	3.062	2.9.4
		3.340	3.4.2	9:374	3,402	2315	2:865	2,718	2,669	1. 100	1.382
Recycle Gas from V-351 MMSCFI Reactor Top Pressure PSI	355	350	3.4.2	9:378	345-2	3350	355	3365	1958	33632	1950
heactor Gas Feed OF	1.00	640	640	640	640	640	440	640	630	650	640
heactor Sed OF	680	671	468	683	673	672	677	660	672	669	678
Catalyst Holdup Tona	14	15/	166	158	149	146	141	145	154	15	163
Catalyst Density #/CF	104	106	1/2	110	110	108	106	107	123	120	: 114
	74.7	73.9	74.0	70.7	73.2	73.9	740	73.2	73.7	73,2	74.9
h Fe in Catalyst	* ···	/ 4 1-	95.5	Like Land			1		96.8		
& Reduced Catalyst Charged to Acad	1142	1029	+ O27	1087	1146	11.01	1162	1188	1183	1081	941
Spare Valuetty - V/V/Hr.	1	1029	, FUX			11.86	17.6	1	1		
	76.9	75.0	79.3	79.2.	77.67	75.81	75.56	74.35	71.02	720	67.33
MISULTE Selectivity		82.94	14.30			79.47	77.24	77,95	79.19	77.94	79 49
Conversion CO	83.04		1	83.0	78.76	Ì		T	53,99	54.47	50.46
	62.12	62.31	64.29	63.37	60.30	54.95	52.7/	53.63	63.46	63.01	61.22
H2 + CO	70.0	69.91	71.64	70.99	67.38	63.81	61.71	62.48	***************************************		7
Contraction 5	45.8	47.7	+8.8	47.4	43	38	97.3	36.05	31.81	36.3	343
Recycle Ratio (total recyclesfresh	1.23	1.29	1,23	1243	1 092	0.81	1.02	a94	1 25	0.19	0.61
Production-Gale Col/waCF Colly FF	1.039	1,084	1.139	1.176	A 999	1.007	0.958	0,967	2.915	0.854	0.785
Yields - Output basis	UHTO BPH	WHY BPH	Luth BPh	with Bin	NHE BPH	WITH BAH	WH6 BA	WHO BAY	witte BPH	WHE BLE	100 BA
<u> </u>	2.31	2.31	2.36	2.32	2.4	2.44	2.41	2.29	2.++	2,30	1237
GO .	1264	12.51	11,57	11.96	15590	15.27	Ke. 77	14.27	15.37	16.01	14.32
Кo	3.36	313	327	3.27	3.44	1,00	1:33	4.30	406	4.15	454
CO2	3448 22,69	323 23 45	3171 12.79	32.10 -3.17	19362 25,23	34413 2621	34.87 25.75	1222486	3413 2584	35.75 26.64	135 HO 26.9
	3.39	5.03	3.59	3,69	3.18	3.57	3.52	2.32	3.89	4.28	: : 1 1
Cit.	+5,5\$ 3.60	565 3.58	572 3104	5.71 371	15,23 3.26	5.89 3.83	587 3,84	6.17 4.11	6.29 4.42	6.22 4.31	61. 422
	1.08	1.51	1.64	1.58	1.47	1.61	1.67	4.76	1.69	1.47	1.46
	G 78	a #	0.78	0.73	081	0.66	0.64	0.69	0.78	0.70	1.00
C.H.	254 117	47 74	42.30	12.71 19.32	2.36 14.59	2// /3.9	4 2.39 15.2	62.49 N.M	1,40 1.51	212 1401	1250 /3.89
3 6	10.27 A 2	A (2 2 9)	1 0.12 A +2	3.0.76 44	AA 279	A5B 204	40.71 44	0 4.53 8.52	0.55 3.39	A57 3.94	AST 3.25
	I WILL IT!	1 140 40	alout Tida	A 12 BUR	143 620	1 19 10 1/	9.11 146	2 1.57 871	167 976	2.02 1010	168 781
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Spare Velocity - V/V/dr.	979		· · · · · · · · · · · · · · · · · · ·			erenden in it dente minder, media i spisi media				The Special Conference of the	
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CHB-118240, (5) 1302 32.31 32.521 32 58 AUT 2 SOCIALING & TIEST SUMMERTY Started 8 Day 2-352-1 1 0700 PERIOD 315 PM Operating Conditions Point No. 3447 3034 Total Reactor Feed F1 303 MASCFH 7.779 3.766 4212 5.441 4.447 Synthesis Cas (1) MNI3CF H 2.825 2.814 5,003 \$ 226 hecycle Gas from V-351 MASCPH 3,3 1 54 3,3100.43 13.50 1.35 PUTAFF Reactor Top Pressure 290 350 350 Reactor Les Ford 50 705 350 490 640 675 450 650 desator bes OF 660 250 147 Catalyst Holdup Tons 152 139 Catalyst Density 4/08 160 111 156 17.5 78.4 17.1 % Fe in Untalyst 81.55 79.6 % hedwood Catalyst Charged to Reactor 1411 Spere Velocity - V/V/Hr. 1384 2146 247/ Selectivity MISULIC 7514 70.16 67.78 81.23 73.43 Conversion 81.47 THE 30.44 7442 78.19 41.54 47.09 547/ 48,97 45.50 53404 H. + CG 6411 60.09 51,86 58.52 Contraction 42.24 35,35 30.56 37.14 Production-Gels Sys/MSSF COSHS FF 2.705 0.317 0.496 01555 0,803 0. 885 01858 0.747 0.843 · Tieles - Output Basis WHO BEH WIS BEN WITE BON WITE WITE BITT 24 231 2.30 240 60 21.30 2081 15.73 1207 14.55 H2 4.44 5,00 5.80 3/1055 5.67 Anot 32.39 3889 2884 37 A D Lato 1 4407 29.16 502 2.91 3 313 3.34 615 4.30 43+ 5.15 200 2.11 3.41 2.90 144 3.41 1.76 0.78 0.25 Care 151 7.72/16 951/189 MARCH 1872 214 MA 0.61 \$20 a.62 322 aug 5.70 0.51 450+0.59 5.50 9.26 1.43 6.33 639 10901.94 14.85 654 12.10 63/ 1.43 055 2.60 0.23 LEB OVA 507012 3,71 8.03-2 21/3 785 21/5 BM 50 90 43 40.25 7.0 Ester Soluble Themiesta (2) 180 5.16294 84 225. 1A66 2 44 1435 307 19.44 1.37 9.30 14.30 11.43 Process Water 3/14 Cot Ame (T) Maleulated by Output Basis 121 Stanoline Oate

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AV	CHB-118	ro
Form	CHB-118	F M'

118 = M_	32.66 60.93 K-351-B	CL TL REACTOR	RUN 19	- characteristic	100 A 11EL	59 FY SUMBARY	33.56 60.80	73.56	37.42	60.59	32.43 60.34 19 Ars +
PERIOD	7/21/5	1/22/51	7/23/51	7/24/51	7/25/51	7/26/51	7/21/51	7/28/51	7/29/51	7/30/51	7/31/57
Operating Conditions Paint No.	684	7	7 - 7 - 7 - 7	70.11	7	/	/ - / - / -	11-11-11-11	1 / 3	7 9 5 7 5 7	7, 57, 57
Total Resetor Feed FF-901 MISOFH	56.76	6.48	6.25	6.37	6.40	7.04	7.10	6.88	6.95	7,14	7.06
	FF14 2.51	2820	261	257	7 / 0	1 (13		2.96*	2.67.	2 5/4	2,48
Banton a Clar Comm W-263 Margrams	2,4	235	2.22	2.29.4	3.19	2 45	3.47	7.28	3 38	0 2:47	1 2.61
Reacter Top Pressure PSI	350	350	3 50	350	350	350	350	350	350	3 50	350
Reactor Ges Feed OF	625	625	640	640	640	660	660	660	660	660	646
Resotor Bed Or	600	620	600	680	610	600	590	600	620	650	640
Catalyst Holemp Tons	254	252	248	345	233	230	223	318	213	209	208
Catalyst Density #/CR	150	158	158	160	156	150	146	144	144	142	143
% Fe in Catalyst	8860	84.7	84.1	835	83.5	83.0	83.0	837	81.4	81.4	81.5
\$ Reduced Catalyst Charged to Reno		No	900.4		talust	added	- a a v	- 4		7/3	9//5
Spare Velocity - V/V/hr.	699	812	761	769	79/	736	827	919	849	794	807
Reazele natio	1.69	1.76	1,39	1.47	1.38	1.78	1.63	1.32	1.40	1.84	1.85
Defectivity	76.81	70.96		-	_			_	76,24	74.57	89,04
Conversion CO	76,77	77.04	_	-					68.63	7255	54.96
¥,	51.84	61.87	-	-		_			1		41.08
H ₂ + CO	60.54	66.85	-	-		· · ·			57.60	52.42	
Contraction	38,72	39 47	3796				3694		33.27	59.65	46.00
Recycle Ratio (total recycle:free)		1.30		4000	36.10 1.39	3545 1.78	1.63	32,43 (32		37.87	33.56
De Anna de la Arage do I	1.70	0 920	1.40	1.40	7.27	- (-/0	7 6 3		1,60	1.84	1,83
Production-Gals G-4/MSGF GO + H2	T 0.832								0,886	0.852	0.781
	i	HWT% BPH							WIY BPH		Wt % BPH
CO CO	22								2.0	20	1.58
	1709	16.9							23,97	20.68	33.55
H ₂	476	4.14		and the contract of the contra					4.71	461	5.76
	2369	2189				<i>*</i>		:	30.07	25.56	11.76
	1.43	275			- 100 1	INALY	5E5 -		4.0	3.7	2.61
CH.	209	50							2.79	2.82	0,62
04 ^H	0 58	1.41							.1,23	1.41	0.52
C_2H4	277	0.76							0.98	0.92	0.62
03 2 6		92.11 10.0	1						1		1.49 6.7
Colle		8078 3.8							ì	1	0.63 2.9
GAH8		6198 81	1						1.77 7.5	2.26 9.0	1.76 6.9
CANTO	221 0	80.59 2.5							0.43 1.9	0.60 2.4	0.36 1.5
G4 + RPO	7.5% 23	7 889 289									6.34 19.2
	Kr 203 166	9 316 2721	2758	3363	3/23	3658	3021	2742	3.31 2944	31.28 3050	3.73 3057
Process Vater (2)	12,124	15,625	14,000	13,712	13.021	10,029	14,604	11,646	12,646	110,783	14,233
	51.3			*		,	to a summaria de la		6.82	52,8	47.2
(1) Calculated by Output Basis	20,4	22,1							22,/	21,0	١, ١
(2) Stanolind Data							*			1/	

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Run #11

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P.O. BOX 1913

Brownsville, Texas

November 28, 1951

Mr. L. C. Kemp, Jr. The Texas Company 135 East 42nd Street New York 17, N. Y.

Dear Mr. Kemp:

The following letter presents history and identity of catalyst samples shipped to Beacon from the recent Run 10 on our K-351A Reactor. As noted below, a copy of this letter and its attachments has been sent to Dr. C. E. Moser at the Beacon Laboratories for his information.

Table 1 presents results of tests obtained locally on catalyst samples submitted, and Table 2 is a summary of results obtained on the reactor. The catalyst charge to the subject Run consisted of catalyst which had been reduced to 96% plus iron content prior to charging to the reactor. The catalyst was fluidised in the reactor for several days with a methane hydrogen mixture prior to introduction of synthesis gas at 12:30 P.M. on November 10th. Please advise if you desire any additional information on the catalyst samples covered by this letter.

Following termination of the above synthesis run on November 20th, the catalyst was retained in the reactor and kept in a fluidized state by circulation of methane and hydrogen. It is our intention to submit additional catalyst samples from the next reactor run for evaluation at Beacon

Very truly yours,

R. H. AITKEN Superintendent

WRS:mr Encl. Cc: Mr. F. M. Dawson

Dr. C. E. Moser

TABLE 1
DESCRIPTION OF SAMPLES SENT TO BEACON

Sample No.	1	2	3	4	5	6
Lab. No.	B-30725	B-30740	B-30854	B-30940	B-31190	B-31464
Date	11-10-51	11-10-51	11-14-51	11-15-51	11-17-51	11-19-51
Time	12:00 N.	12:00 M.	10:00 AM	12:00 M.	12:00 M.	12:00 N.
Reactor HI	·s. 0	12	94	132	180	216
≸ Pe	95.93	85.86	83.6	82.73	81.6	80.1
% K ₂ 0	0.97	0.63	0.62	0.64	0.67	***
% C		5.2	6.7	7.0	7.3	6.9
Dry Sieve On 40	3.4	5.6	2.8	2.8	3.2	2.0
60	20.6	16.8	16.2	16.2	19.4	15.0
80	16.4	13.6	14.6	13.2	16.4	15.4
100	17.0	12.6	15.8	11.2	14.4	14.8
120	14.0	18.2	18.8	20.0	21.8	24.2
200	21.4	26.0	27.0	29.4	22.6	25.6
Pan		8.2	5.8	6.8	2.4	3.0

h-451A REMUTOR RUN OPERAT & Y			11-13-51	11-11-51	11-75-61	11-16-51	11 11 61		11-19-51	· ·
OPERATING CONDITIONS					144=43=31	11-10-31	11-17-51	11-14-51	11-19-52	
SIN MESIS CAS - MASCEH 3)	2.00	3.00	3.00	3.10	3.07	2.99	3.05	3.09	3.23	
JUNGYCLE GAS - MMSCFN	4.00	4.10	4.05	4.42	4.69	4.26	3.45	4.02	3.73	
REUYCLE RAILO	1,49	1.37	1.31	1.42	1.53	1.43	1.13	1.36	1.16	
REACTOR TOP PRESSURE-PSI	350	350	350	350	350	350	345	350	350	
REAUTOR BLD TEMP OP	0k7	660	055	1 045	630	650	620	640	655	
CATALYST FICLIOUP - TONS	231	226	220	213	210	203	565	255	3,12	
MUSITY - #/JF	152	154	152	1,60	148	154	156	168	168	
0/ა Pe დ ⁰ /ი C	65.1 -	83.5 -	82.7/7.3	32.8/7.2	81.8/6.9	81.0/7.3	80.1/7.5		81.4/7.00	
SPACE VELOCITY - 7/V/HR	826	939	999	1019	1014	1073	365	988	1054	
H2/CO RATIO OR FRESH FEED	1.33	1.35	1.84	1.35	1.86	1.78	1.76	1.76	1.38	
00. VERSIONS 00 b/o	85.3	83.1	86.6	84.4	81.4	82.7	80.5	80.3	81.5	
H ₂ °/°	63.3	07.7	07.2	65.0	61.3	63.4	58.7	6.20	63.8	
CO/H2~U/S	71.0	74.9	74.0	71.8	68.3	70.4	66.6	63.8	69.9	
CUMTRACTION %	55.3	50.2	55,4	53.9	51.4	53.0	46.8	49.9	49.0	
SELECTIVITY C3/07/ 0/0	74.8	770.5	75.5	11.9	70.7	74.3	75.5	78.6	78.0	
							*			
PRODUCTION - OUTPUT*BASIS	} 			And the second	21 -					
GALS C3 /M CF H2/CO				- Annual Control of the Control of t					2)	*
SUMPONENT YIELDS 1)	Wty BPH	With BPR	W+ 2/0 HPH	MIO/a NOM	W+Q/- DBU	740/ 77		0.	Wt°/o BPE	
A	2.23	2.11	2.07	2.04	2.13	2.20	2.16	2.20	2.14	
ÜŲ	10.62	8.73	9.95	11.50	13.82.	12.85		14.30	13.44	
H ₂	3.57	3.16	3.24	3.43	3.36	3.50		3.61	3.56	
CUp	24.90	23.22	23.00	21.71	23.97	22:76	3.92	22.68		
li a	3.23	3.02	3.54	3.11	3.13		22.14		20,60	
OIN.	3.31	4.01	3.37	2.94	2.95	3.8 5	3.32 3.40	3.16 3.18	2.90 3.21	
Opli,	1.67	1.85								
Comb			1.50	1.51	1.39	1.44	1.46	1.54	1.72	
Calin	2.76.32	1.13		0.90	1.05	1.00	1.09	1.11	1.02	
C ₃ H _d	מגניין וביים	0 03 6 9	0 02 5 0	7 80 5 6	0.00 10.0	2.17. 12.0	7.T. 14.0	2,43 14.1		
on Ma	2 3/ 2 3	2 06 0 7	217 38 0	2103.9	2 20 35 5	0.67 3.7	1.20 0,9	0.17 9.5	.92 5.5	
ુենյ ₀									2.75 14.0	
USF RPU	7 50 25 4	0.31 1.5	0.31 1.5	044.2.5	0.49 2.4	0.69 3.4	0.40 2.4	0.51 2.0	.24 1.3	
NATER SOLUBLE CHLMICALS	h.06 11.h	3.44.92.4	1.21.12.5	4.0	9.07 33.9 1. 22 13 3	7.26 27.3	0. [/ 20, /	7.40 29.9 5 22 31 3	0.12 34.2	
									4.14 13.9	
PROCESS WATER	19.25	23.78	22.74	22.20	16.02	20.36	18,34	19.62	20.11	
								4		
				4		4. a				

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January 3, 1951

Mr. L. C. Kemp, Jr. The Texas Company 135 East 42nd Street New York 17, N. Y.

blind: The Texas Company K-351-A Catalyst Samples

Dear Sir,

Please refer to our letter of Movember 28, 1951 wherein we gave you operating data on Run No. 10 K-351-A Reactor, and tests on the samples of catalyst of different bed-life which were sent to Beacon Laboratory for possible evaluation. This letter gives similar data and tests for Run No. 11, November 26, 1951 through December 7, 1951.

Table I is a tabulation of results of tests on the catalyst samples and is self explanatory. Table II is a summary of results obtained on the reactor. The catalyst charge for this run was essentially the same as at the end of Run No. 10, previously covered, but the catalyst was kept fluidized with a mixture of me theme and hydrogen prior to cutting in synthesis gas on November 25, 1951. Additional catalyst was added during the run.

Samples of catalyst are being sent to Beacon during the current Run No. 12 which started December 20. 1951 and similar data will be furnished in the near future.

Yours very truly,

Signed: V. A. L.

V. K. BRANDENBURG Superintendent

WHR: ma Encl. ce: FMD

Dr. C.E. Moser, TTCO, Beacon

11m 6 3 2 2 125

TABLE I
DESCRIPTION OF SAMPLES SENT TO REACON

Date Caught	11-28-51	11-29-51	12-1-51	12-3-51	12-5-51	12-7-51
Time Caught	4:30PM	6:00 AM	8:00 AM	6:00AM	6:00 AM	6:00AM
Bed Life Hrs	Zero	12	62	108	156	204
Lab No. B =	32087	32127	32426	32701	33068	33430
Sample No.	12-1	12-2	12=3	12-4	12-5	12-6
% Pe	90-71	85,52	84.16	81,93	81,36	81,73
% K20	0.52	0.51	0,40	0.61	0.70	0.71
% C	5.88	de n	7.30	7.02	7.34	7.24
Dry Dieve						
on 40	2.4	2.8	2.2	1,8	3.0	2.0
60	15.4	16.6	16.8	15.6	25.2	15.8
80	14.8	15,0	14.8	15.2	16.6	16.0
100	9.6	9.0	8.6	10.4	5,6	9.8
120	23.2	24.2	25.4	25.8	29 . 2	22.6
200	30,3	28 ,6	31.2	2 9 。0	19.2	31,4
Pan	3,6	3.6	2.8	3 . 0	1.0	2.2

	· · · · · - · · · · · · · · · · ·	SUMMARY		The second contract of the second	The second secon			• • •	The second community
DATE	11-29-51	11-30-51	12-1-51	12-2-51	12-3-51	12-4-51	12-5-51	12-6-51	12-7-51
OPERATING CONDITIONS	e and a single and a single si	- * · · · · · · · · · · · · · · · · · ·	and the second	l		n was q	! •	*	
SYNTHESIS GAS - MMSCPH 2)	2.91		3.02	3.90	2.71	2.90	2.81	2,80	2.73
RECYCLE GAS - MMSCFH	4.66	4.26	3.95	4.22	4.21	4.05	4 _{0.} 26	4.20	3.87
REACTOR TOP PRESSURE - PSI	350	350.	350	350	350	350	350	350	350
REACTOR BED TEMP OF	650	650	660	650	650	630	630	650	650
CATALYST HOLDUP - TONS	215	210	210	281	266	252	246	243	236
" DENSITY - #/CF	162	160	168	168	168	166	166	168	188
% Pe & %C	85.9	⇒ 86.1	85.4	82,6	82.4	81.6	81.5	80.9	80.5
SPACE VELOCITY - V/V/HR	1040	1081	832	882	820	891		918	956
H2/CO RATIO OR PRESH FEED	1.80	1.78	1.83	1.84	1.85	1.84	1.86	1.92	1.90
CONVERSIONS - CO %	82.11	85.2 8	85.87	85,11	85.22	84.48	84.34	86.13	85.90
H ₂ \$	59.53	65.36	6 6 °92	66.71	65.21	66.56		67.45	67.29
CO/H ₂ %	67.69	72.51	73.62	73,19	72,23	72.85	71.50	73.84	73.53
CONTRACTION X	50.41	53.75	50.46	51.07	51.66	50.21	49.76	50.92	41.60
SELECTIVITY C3/C1/ %	76.22	78.92	76.76	77. 93	77.53	76.99	•	66,37	71.40
PRODUCTION - OUTPUT BASIS GALS C3//M CF H2/CO	1								
\$1.00\text{CM}\$ \$600_1 \ \text{cond}\$ \$600_1 \ \text{cond}\$ \$100_1	Wt% BP	H Wt% BPH	Wt% BPH	Wt% BPH	₩t % BPH	Wt% BPH	₩t% BPH	Wt% BPH	Wt% BPH
GALS C3/ /M CF H2/CO	Wt% BP	H Wt% BPH 2.20	Wt% BPH 2.05	Wt% BPH 1∘93	Wt% BPH 2.13	Wt% BPH 1.94	₩ t% BPH 2.34	wt% BPH 2.05	Wt% BPH 2.10
GALS C3/ /M CF H2/CO		2.20			2.13	1.94			•
GALS C ₃ / M CF H ₂ / CO COMPONENT YIELDS 1)	1.99	2.20	2.05	1.93	2.13	1.94	2,34	2.05	2.10
GALS C ₃ / /M CF H ₂ /CO COMPONENT YIELDS 1) A CO	1.99 12.99	2.20 10.79 3.26	2.05 10.07	1.93 10.45	2.13 10.31 3.24	1.94	2,34 10,92	2.05 9.55	2.10 9.60
GALS C ₃ / /M CF H ₂ /CO COMPONENT YIELDS 1) A CO H ₂	1.99 12.99 3.88	2.20 10.79 3.26	2.05 10.07 3.11	1.93 10.45 3.10	2.13 10.31 3.24	1.94 10.74 3.08	2.34 10.92 3.31	2.05 9.55 3.10	2.10 9.60 3.18
GALS C ₃ / M CF H ₂ / CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂	1.99 12.99 3.88 23.90	2.20 10.79 3.26 21.47	2.05 10.07 3.11 19.65	1.93 10.45 3.10 17.90	2.13 10.31 3.24 20.09	1.94 10.74 3.08 17.47	2,34 10,92 3,31 19,16	2.05 9.55 3.10 17.76	2.10 9.60 3.18 18.22
GALS C ₃ / M CF H ₂ / CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂ CH ₁	1.99 12.99 3.88 23.90 3.31	2.20 10.79 3.26 21.47 3.04	2.05 10.07 3.11 19.65 3.88	1.93 10.45 3.10 17.90 3.68	2.13 10.31 3.24 20.09 4.10	1.94 10.74 3.08 17.47 4.20	2.34 10.92 3.31 19.16 4.05	2.05 9.55 3.10 17.76 3.58	2.10 9.60 3.18 18.22 3.13 4.09
GALS C3 / /M CF H2/CO COMPONENT YIELDS 1) A CO H2 CO2 N2 CHL C2HL	1.99 12.99 3.88 23.90 3.31 2.85	2.20 10.79 3.26 21.47 3.04 2.86	2.05 10.07 3.11 19.65 3.88 2.78	1.93 10.45 3.10 17.90 3.68 2.67	2.13 10.31 3.24 20.09 4.10 3.18	1.94 10.74 3.08 17.47 4.20 2.91	2.34 10.92 3.31 19.16 4.05 3.39	2.05 9.55 3.10 17.76 3.58 6.21	2.10 9.60 3.18 18.22 3.13 4.09 1.47
GALS C ₃ / M CF H ₂ /CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂ CH ₁ C ₂ H ₁ C ₂ H ₆	1.99 12.99 3.88 23.90 3.31 2.85 1.54	2.20 10.79 3.26 21.47 3.04 2.86 1.23	2.05 10.07 3.11 19.65 3.88 2.78 1.40	1.93 10.45 3.10 17.90 3.68 2.67	2.13 10.31 3.24 20.09 4.10 3.18 1.58	1.94 10.74 3.08 17.47 4.20 2.91 1.50	2.34 10.92 3.31 19.16 4.05 3.39 1.60	2.05 9.55 3.10 17.76 3.58 6.21 1.52	2.10 9.60 3.18 18.22 3.13 4.09 1.47 1.31
GALS C ₃ / M CF H ₂ / CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂ CH ₁ C ₂ H ₆ C ₃ H ₆	1.99 12.99 3.88 23.90 3.31 2.85 1.54 .95	2.20 10.79 3.26 21.47 3.04 2.86 1.23	2.05 10.07 3.11 19.65 3.88 2.78 1.40 1.05	1.93 10.45 3.10 17.90 3.58 2.67 1.42 1.04	2.13 10.31 3.24 20.09 4.10 3.18 1.58 1.09	1.94 10.74 3.08 17.47 4.20 2.91 1.50 1.04	2.34 10.92 3.31 19.16 4.05 3.39 1.60 1.04	2.05 9.55 3.10 17.76 3.58 6.21 1.52 1.13	2.10 9.60 3.18 18.22 3.13 4.09 1.47
GALS C ₃ / M CF H ₂ /CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂ CH ₁ C ₂ H ₄ C ₂ H ₆ C ₃ H ₆ C ₃ H ₈	1.99 12.99 3.88 23.90 3.31 2.85 1.54 .95 2.74	2.20 10.79 3.26 21.47 3.04 2.86 1.23 .98 2.62	2.05 10.07 3.11 19.65 3.88 2.78 1.40 1.05 2.75	1.93 10.45 3.10 17.90 3.68 2.67 1.42 1.04 2.74	2.13 10.31 3.24 20.09 4.10 3.18 1.58 1.09 2.84	1.94 10.74 3.08 17.47 4.20 2.91 1.50 1.04 2.66	2.34 10.92 3.31 19.16 4.05 3.39 1.60 1.04 2.95 1.04	2.05 9.55 3.10 17.76 3.58 6.21 1.52 1.13 2.76	2.10 9.60 3.18 18.22 3.13 4.09 1.47 1.31 3.09
GALS C ₃ / M CF H ₂ / CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂ CH ₁ C ₂ H ₄ C ₂ H ₆ C ₃ H ₆ C ₃ H ₈ C ₄ H ₈	1.99 12.99 3.88 23.90 3.31 2.85 1.54 .95 2.74	2.20 10.79 3.26 21.47 3.04 2.86 1.23 .98 2.62 1.69	2.05 10.07 3.11 19.65 3.88 2.78 1.40 1.05 2.75	1.93 10.45 3.10 17.90 3.58 2.67 1.42 1.04 2.74 .83 2.21	2.13 10.31 3.24 20.09 4.10 3.18 1.58 1.09 2.84 .96 2.95	1.94 10.74 3.08 17.47 4.20 2.91 1.50 1.04 2.66 .97 2.65	2.34 10.92 3.31 19.16 4.05 3.39 1.60 1.04 2.95 1.04 2.31	2.05 9.55 3.10 17.76 3.58 6.21 1.52 1.13 2.76 .84	2.10 9.60 3.18 18.22 3.13 4.09 1.47 1.31 3.09 .94 2.36
GALS C ₃ / M CF H ₂ /CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂ CH ₁ C ₂ H ₄ C ₂ H ₆ C ₃ H ₆ C ₃ H ₈ C ₄ H ₈ C ₄ H ₈ C ₄ H ₁₀	1.99 12.99 3.88 23.90 3.31 2.85 1.54 .95 2.74 .82 2.04 .26	2.20 10.79 3.26 21.47 3.04 2.86 1.23 .98 2.62 1.69 2.32	2.05 10.07 3.11 19.65 3.88 2.78 1.40 1.05 2.75 .86 2.11 .28	1.93 10.45 3.10 17.90 3.58 2.67 1.42 1.04 2.74 .83 2.21	2.13 10.31 3.24 20.09 4.10 3.18 1.58 1.09 2.84 .96 2.95	1.94 10.74 3.08 17.47 4.20 2.91 1.50 1.04 2.66 .97 2.65	2.34 10.92 3.31 19.16 4.05 3.39 1.60 1.04 2.95 1.04 2.31	2.05 9.55 3.10 17.76 3.58 6.21 1.52 1.13 2.76 .84 2.30 .37	2.10 9.60 3.18 18.22 3.13 4.09 1.47 1.31 3.09 .94 2.36
GALS C ₃ //M CF H ₂ /CO COMPONENT YIELDS A CO H ₂ CO ₂ N ₂ CH ₁ C ₂ H ₆ C ₃ H ₆ C ₃ H ₆ C ₄ H ₈ C ₁ H ₈	1.99 12.99 3.88 23.90 3.31 2.85 1.54 .95 2.74 .82 2.04 .26 7.21 26	2.20 10.79 3.26 21.47 3.04 2.86 1.23 .98 2.62 1.69 2.32	2.05 10.07 3.11 19.65 3.88 2.78 1.40 1.05 2.75 .86 2.11 .28 6.70 26.2	1.93 10.45 3.10 17.90 3.58 2.67 1.42 1.01 2.74 .83 2.21 .2t 7.48 291	2.13 10.31 3.24 20.09 4.10 3.18 1.58 1.09 2.84 .96 2.95 .41 8.68 31.2	1.94 10.74 3.08 17.47 4.20 2.91 1.50 1.04 2.66 .97 2.65 .30 7.60 29.2	2.34 10.92 3.31 19.16 4.05 3.39 1.60 1.04 2.95 1.04 2.31 .30 7.22 26.2	2.05 9.55 3.10 17.76 3.58 6.21 1.52 1.13 2.76 .84 2.30 .37 6.99 25.2	2.10 9.60 3.18 18.22 3.13 4.09 1.47 1.31 3.09 .94 2.36 .33 6.73 23.4

¹⁾ AERATION GAS SUBSTRACTED FROM OUTPUT FIGURES ALSO CHI & CO2 DATA ARE NET EXCLUSIVE OF GAS GENERATOR 2) CALCULATED ON AN OUTPUT BASIS

CARTHAGE HYDROCOL.INC P. O. Box 1913

BROWNSVILLE, TEXAS

January 25, 1952

& PDS DIV FEB 4 1952 KEY TCH HF WLD MPD IFG JHG EUR WRH 140 Hdl

Mr. L. C. Kemp, Jr. The Texas Company 135 East 42nd Street New York 17. N. Y.

Dear Sir:

GUY GEORGE GABRIEL SON Providen FRANK M DAWSON Executive Vice Pren Herri

R H AITKEN //oe President ALBERT L WOLFE Secretor

ALLEN K BREHM Vice Pres sent and mest in

Please refer to the file of correspondence which covers operating data on previous Synthesis Reactor runs and tests on special catalyst samples submitted. This letter gives similar data for the Reactor Run No. 12-A, Plant Run No. 13, December 21, 1951 to January 2, 1952 inclusive.

Table I is a tabulation of the results of tests covering six samples obtained as shown during the run. The catalyst in this reactor consisted of catalyst transferred from K-351-A used on the previous run (Reactor Run No. 11, Plant Run No. 12) and about 100 tons fresh catalyst. The catalyst was subjected to a "pretreating cycle then to a 24-hour carbiding cycle immediately prior to starting the synthesis reaction and approximately 50 tons of fresh catalyst were added near the end of this run.

Samples of catalyst from Reactor Run 12-B, Plant Run 14 have been submitted to Beacon and operating data will be furnished in the near future.

Yours very truly,

R. H. AITKEN.

Vice President

WHR: ma Ronel-2

cc: Dr. C.B.Moser, Beacon

Mr. du B. Eastman, Montebello

WMS-KGM

ON LCAY- (W) 3/7/53

TABLE I

DESCRIPTION OF CATALYST SAMPLES SUBMITTED

PLANT RUN NO. 13 REACTOR RUN NO. 12-A

Date Caught	12-20-51	12-21-51	12-23	12-26	12-29	12-31
Bed Life Hrs.	Note 1	Note 2	72 Hrs.	120 hrs.	188 hrs.	
Lab No. B -	34271	34297	34584	34905	35274	35646
Sample No.	13-1	13-2	13-3	13-4	13-5	13-6
%Pe	90.4	83.5	81.6	80.5	80.2	80.9
%K 20	0.61	o.58	0.69	0.69	•	0.58
≴c	3.4	4.2	6.0	7.3	7.50	6.5
Dry Sieve						
On 40 60 80 100 120 200 Pan	2.0 19.6 15.6 0.2 37.8 20.8 4.8	1.6 14.8 15.8 0.2 37.0 25.8	1.8 17.4 17.8 1.0 29.6 26.2 6.2	2.4 16.0 17.4 8.6 21.4 26.4 8.6	•	1.2 14.6 17.0 27.0 9.4 28.4 2.6

Note 1 - Sample obtained after "pretreating cycle" and before "carbiding cycle".

Note 2 - Sample obtained after "carbiding cycle" and prior to start of Synthesis.

		A	1	\$	1	3	S	1	ì
DATE	12-21	12-22	12-23	12-24	12-25	12-26	12-27	12-28	12-29
OPERATING CONDITIONS SYNTHESIS GAS - MASCIFE									
	2.84	2.76	3.00	2.93	2.98	2.79	3.17	3.17	3.03
RECYCLE GAS - MMSCPH	4.40	4.58	4.24	4.05	4.43	4.61	4.19	4.50	4.46
REACTOR TOP PRESSURE - PSI	330	350	350	350	350	350	350	350	350
REACTOR BED TEMPERATURE - CT			625 (17	635	635	620,44	620 62	615 47	612
CATALYST HOLDUF - TOHS	195	188	200	501	211	210	200	192	183
#" DENSITY - #/CF	150	148	150	150	148	3/18	148	146	144
% Fo & % C	84.9 - 7	.9 85.4 - 6.	82.11 - 4.	8 80.4 - 6.0	80.9 - 5.9	80.5 - 4.2	80.3 - 5.3	80.2 - 6.6	79.9 - 7.
SPACE VELOCITY V/V/Hr	1004	1050	1043	993	944	971	1128	1115	1071
H2/CO RATIO OF PRESE PEED	1.81	1.87	1.86	1.85	1.83	1.89	1.86	1.89	1.89
CONVERSIONS - CO \$	81,25	84.28	82.84	79.98	80.81	79.50	79.99	77.27	74.10
H ₂ %	61.55	64.65	63.24	59.22	59.37	60.21	58.06	53.38	51.27
CO/H2 %	68.56	71.49	70.08	66.50	66.92	66.91	65.73	61.63	59.16
CONTRACTION %	44.3	54.4	50.6	46.8	47.1	47.0	46.8	43.7	43.1
SELECTIVITY C3/C1/%	80.76	77.4	78.52	78.72	77.94	81.59	79.98	76.52	80.92
Production OUTPUT BASIS COMPONENT YIELDS	343 0155		v17			425	14,0	3 40	45.0
A	The state of the s	<u> </u>		Wt% HPH	Wt% BPH	Wt% 5PH	Wt% BPH	%t% BPH	WEX B
CO	2.03		2.13	2.08	2.00	1.98	2.08	2.23	2.17
H ₂	13,31	11.11	11.97		13.51	14.15	13.87	15.86	17.64
CO ₂	3.56	3.36	3.44	3.80	3.79	3.74	3.89	4-44	4.53
32	17.71	20.25	17.98	<u> </u>	18.93	17.09	18,42	21.51	19.49
	2.95	3.62	3.44	3.28	3.21	3.19	3.43	2.76	3.49
CHL	3.00	3.26	2.20	2.31	2.62	1.85	2.07	2.29	1.34
c ^S H [†]	1.21	1.51	1.54	1,36	1.45	1.46	1.51	1.46	1.43
с эн 6	.57	.94	. 81	° 81 [†]	.50	.80	.83	. 15	1.65
C3H6	2.79 15.	q 2.69 13.8	2.08 11.7	2.56 14.1	2,86 15.7		_	2.40 14.1	1
с ₃ н ₈	.98 5.	3 .93 4.8	.74 4.2	.97 5.4	.86 4.8		1.33 8.1	<u> </u>	1
СПив	2.10 9.7	2.03 9.0	1.84 8.9	2.08 9.9	2.71 12.9			1.65 8.4	
C/H10	.27 1.3	.32 1.5	.50 2.5		.28 1.4			•	
C5≠ RPO	10.88 37.0	9.24 31.8	7.28 27.6		6.46 24.1				6,18 22
WATER SOLUBLE CHEMICALS	3.04 9.3	4.33 12.6	4.21 13.5		4.05 12.			3.97 13.3	4.57 14.
PROCESS WATER	20.13 -	19.50	34.39 -	21.81 -	21.55 -	21.01 -			16.66 -
							7 • • •		
Tot. C3+	77.6	73.5	68.4						1

OPERATING & YIELD SUMMARY OF

K-351-B REACTOR - RUN NO. 12-B

PLANT RUN NO.14

WRS

HHM

HT J

TABLE I
DESCRIPTION OF CATALYST SAMPLES SUBMITTED

REACTOR RUN 12-B

PLANT RUN NO 14

Date Caught	1-8-52	1-9-52	1-10-52	1-12-52	1-14-52	1-16-52	1-18-52	1-20-52
Bed Life Hrs	Note 1	Note 2	12	60	108	156	204	252
Lab No. B -	35998	36068	36153	36374	36657	37112	37220	37386
Sample No.	14-1	14-2	14-3	14-4	14-5	14-6	14-7	14-8
% Fe	84.8	81.3	79.9	80.1	79.9	79.2	79.5	79.2
% K ₂ O	0.55	0.49	0.31	0.29	0.39	0.15	9.29	0 30
% C	5.2	7.2	6.9	8,1	୍ଦ	8.5	8.1	8.5
Dry Sieve Ana.								
on 40	1.2	1.6	1,2	2.2	2.8	2.0	1.6	1-4
60	10.2	11.4	11.4	15.6	18,0	16.4	14.8	12 2
80	13.6	13.0	14.0	15.0	18 _° 2	17.2	17.2	16.6
100	23.8	20.6	24.6	29.2	28,8	27.4	29.2	35.6
120	8.2	10.8	9.8	4.2	6.6	8.4	9.0	6.4
200	31.2	31.4	29 .0	28.6	24.4	27.6	27 . 8	26,2
Pan	12.6	12,0	9.4	5.8	1.6	1.4	1,2	1,2

Note 1 - This sample obtained at end of H2 "Pretreat" and prior to "Carbiding cycle".

Note 2 ~ This sample obtained at end of "carbiding cycle" and just prior to start of Synthesis reaction.

The here on tom "operation, Som " Processing and form."

Fred being and from "peration, Som " Processing and form."

K-351-B REACTOR RUN 1	2-B OPERATING	& YIELD SUKMA	RY	AbLE II		from Operating	Feed		Sheet No.
ATE	1-9-52	1-10-52	1-11-52	1-12-52	1-13-52	1-14-52	1-15-52	1-16-52	1-17-52
PERATING CONDITIONS -									
SYNTEHSIS DAS - MUSCEH	2.65	2.55	2,65	2.54	2.50	2.40	2.31	2.38	2,59
RECYCLE GAS - MMSCFH	4.79 1.74	4.44 r. 99	4.39 214	4.26 640	4.11 ""	4.50 4.90	4.62 4.93	4.657.03	3.76 ⁰
REACTOR TOP PRESSURE - PSI	325	350	350	350	350	350	350	350	3 50
REACTOR BED TEMPERATURE - O F	639	671	676	676	682	669	667	671	674 660
CATALYST HOLDUP - TONS	225 116	210 222	200 213	190 198	181 / (1	172 152	164 179	650 156 /65	
DENSITY - #/CF	146	144	: 144	142	140	136	134	130	132
%Fe & %C	82.9 = 6.0	79.7 - 7.8	79.8 - 8,3	79.2 - 7.5	79.4 - 7.9	79.9 - 8.1	80.0 - 8.4	79 .7 9.7	79.4 -
SPACE VELOCITY - V/V/Hr	791	801	876	872	894	867	850	924	1044
H2/CO RATIO OF FRESH FEED	1.87	1.84	1.88	1,85	1,85	1,90	1.87	1.85	1.80
CONVERSIONS - CO %	83.53	83.27	83.00	82.09	81,92	83.25	86.78	86.54	85.9
H ₂ %	63.51	61.74	60.30	57.05	57.85	61.46	7.25	67,21	62,60
CO/H2 %	70,49	69.32	68.06	65, 85	66.31	68,98	74.06	74.00	70.8
CONTRACTION %	50.15	51.11	46.34	48.68	49.64 47.64	48.02	56.76	52,26	48.3
SELECTIVITY C3 / C1/ %	80.95	74.93	76.06	78.41	76.02	76.89	77.51	78.81	76.2
PRODUCTION _ OUTPUT BASIS	Wt≸ BPH	Wt% BPH	wt% BPH	wt% BPH	Wt% BPH	ut% Bra	Wt% BPII	wt% BM	11 1%
COMPONENT YIELDS	with met								
A	2.05 ~	2,25	2.36	2.29 ~	2. 2 4	2.13	1.97	1.98	2.11
CO	11.14~/	11.49	11.7	12.36	12.74	11.48	9.06	9.48	9.90
H ₂	3.32 2.32	3.48	3.71	3.94	3.95	3.61	3.02	3.07	3,51
CO2	16.44 -	20.51/	20.87	23.30 ~	24.16	21.41	17.66	17.30	21,24
N ₂	3.61	4.38~	3.50	4.15.	4.57	5.10	4.26	4.12	3.23
CHI	2.40	3.36/	2.72	2.55~	2.89	2.54	2.79	2.15	2.74
C ⁵ H ^I	1.40 ~	1.56	2.29	1.69~	1,66	1,57	1.66	1,66	1.68
с ⁵ н ⁶	.79 ₽	.90~	.32	ه86 ه	.81	.79	.81	. 84	.80
С3Н6	2.51 12.6	2.47~ 12.0	1.59 7.9	2.74 13.1	2.43 11.7	2,39 10.7	2.76 12.0	2.07 9.3	2.46 1
C3H8	1.01 5.2			.93 4.6	.89 4.2	.85 3.9	.98 4.3	.85 3.9	, 89
	1.98 8.6	1.36- 5.7	1.72 7.4	1.91 7.9	1.88 7.6	1.57 6.1	2.02 7.6	1.83 7.1	1.68
с ^{ин} 10	.25~ 1.1			.25 1.0	.40 1.7	THE RESERVE THE PROPERTY OF THE PERSON NAMED IN COLUMN TWO IS NOT THE PERSON NAMED IN COLUMN TWO IS NAMED IN COLUMN TWO I	.25 1.0		
C5/ RPO 1949	1.22 4.8		and the contract of the contra	1.60 3.8	1.32 4.9				1.07
NATER SOLUBLE CHEMICALS	√°3°.39× 9.7	4.16~ 11.5	12.6 12.6	13.82 10.5	3.35 8.9	M4.21 10.8	4,65 11.6	14.19 10.7	4.26
		1	1	14.32	15.43	19.11	24.60	26.92	22.93

DATE	1=18	- 52	1-19-	52	1-20-	52	1-21-5	52	1-22	-52		<u> </u>		:		
PERATING CONDITIONS																-
SYNTHESIS GAS - MMSCFH	2,5.		2.60		2.7		3 235	No Constitution of the last		77						Name and Pos
RECYCLE GAS == MMSCFH	4,40	6.51	4.68	7.24	4.38	7.02	3.78	713	3.66	7,43	· · · · · · · · · · · · · · · · · · ·		wind and and and and and and and and and a			
REACTOR TOP PRESSURE - PSI	350		350		350		350		35	60						
REACTOR BED TEMPERATURE - O F	667		669	1	679		669		65	53		1				
REACTOR BED TEMPERATURE = C F CATALYST HOLDUP = TONS	143	153	115,	140	دون 122	13 L	106	13/		10 10	>		gerengger, en			
n Density - #/cf	124	124			120		116		13	34						
% Fe & % C	79.5	8.9	80.2	4.9	79.0	11,2	79.8	9.1	80.6	9.5		<u> </u>	-			
SPACE VELOCITY - V/V/Hr	1003		978		1220		1734		211			<u> </u>	المعقدية حصيرية ومايية	<u> </u>		-
H2/CO RATIO OR FRESH FEED 589	1:7	9	1.86		1.8	6	1.9	0	1.93-	59.1	12					
conversions - co %	83.88		84.94		85.53		80.41		74.34		,					
H2. %	61.78		62.83		63.42	- Appelling on the second of the second of the	56.91		48.71		Charles and the Control of the Contr					Mary Mary State of the Local Control of the Local C
co/H2 ≯	69.50		70.55		71.14	the research control of the second	65.00		57.56							
CONTRACTION %	49.48		48.99		48.64	manage () of a deleter of	42.86		38°11							-
SELECTIVITY C3 C1 A	77.28		75.92		80.66	The Hall of Average	76.07		77.97							
PRODUCTION - OUTPUT BASIS	Wt%	BPH	WtX	BPH	WES	BP H J	wt%	BPH	Wt%	ВРН						
COMPONENT YIELES																Day of Fac.
A	1.97		2.10		1,89		1.92		1.89					<u> </u>		
CO	11.03		10.40		10.02		13.60	angger or assert or desirable	17.82							****
E ₂	3.51		3.44		3.40		4.10		4.86				The second s	-		
C0 ^S	17.09		19.19		16.80		20.29		20.40				megicaphone 4 college.			.
N ₂	3.63		4.25		3.76		3.47	· Salarana	3.06							
CHL	2.58	and the second	2.94		1.83		2 .25		1.25					-		
c ₂ H ₄	1.63		1.61		1,48		1.55		1.48							
с ₂ н6	.72		, 76		۰,64		°65		.47						_	Printing and the Control of the Cont
С ₃ Н ₆	2.hc	11.4	2,22	12.3	1.89	9.7	2.10	13.0	1.62	11.4						M-70-1-22-
С3Н8	81	3.9	1.15	5.7	.73	3.8	,65	4.1	.56	4.0		<u> </u>				
cl ¹ H8	1.78	7.3	1.29	5. 5	1.44	6,4	1.32	7.1	.91	5. 5						
cμ ^H 10	,31	3.3	,.24		<u> </u>	1.8	.24	1.3	.11	7.7						
C ₅ ≠ RPO	1.30	19.5		27:3	1.08	4.4	5.0%	13.8	<u> </u>		المرابع والتنافي والمنافي والم		£2.~~	-		رى ، سمورو
WATER SOLUBLE CHEMICALS	4.13	11.2	14.36	12.2	10.4.53	13.3	3.86	13.7	1.33, 37	13.5	المستعدد الم		usi . To ra ng sampa ak oo sa	and the forest the second section of		tor (25%
PROCESS WATER	23.65	e e	22.44	5 0 co	27.81	~ > C:	22.94	& 57	21.44					‡		
		39.0		37.9 59.9		39.		43.0		46.4						

TABLE II

				<u> </u>				<u> </u>
DATE	2-2-52	2-3-12	2-4.12	2-5-52	2-6.52	2.7.12	7-4-12	1
OPERATING CONDITIONS	``							
SYNTHERIS GAS - MMSCFH	2.92	2.86	2,95	3.73	4,45	4,53	4.11	
RECYCLE GAS - MMSCFH	3,60	3.84	3.88	3,48	3.18	3,75	2.65	
** REACTOR TOP PRESSURE - PSI	350	350	350	355	380	380	250	
REACTOR BED TEMPERATURE - O F	650	645	640	650	690	670	650	
CATALYST HOLDUP - TONS	141	137	133	13/	129	129	140	
Density # #/cp	128	126	122	120	114	112	124	N. Committee
% Fo & % C	1295- 9.4	80.31-11.3	7943- 130	79.85- 11.2	17.47-13.6	79.6 -13.2	78:33- 10.4	
SPACE VELOCITY - V/V/Hr	1168	1220	1280	1587	1769	1809	1676	
H2/CO RATIO OF FRESH FEED	1.82	1.85	1.85	1.86	1.88	1.83	1.89	
conversions - co %	91.19	83.70	82.89	7981	78.04	81.68	78.41	
H ₂ %	80.35	62-08	60.15	5/3/	5018	53.87	53.68	
CO≰H ₂ %	84,20	69.66	68.1Z	6/26	40.13	63,70	62.25	à
CONTRACTION %	55.07	46.19	44.61	38.03	40.72	40,37	29.42	
SELECTIVITY C3/C1/2	56.67	75.15	1702	67.97	70.96	76.99	65.99	
PRODUCTION - OUTPUT BASIS								
ALC Cy/ As on Up/oo							2.7	
COMPONENT YIELDS	MAYS - BPH	wth ern	Auth BPH	NT% BP4	WITH BPH	WT EPH	with pro	
A	1.73	1.99	2.06	2.15	•	2.00	2.07	
CO	482	11.41	11.78	1407	15-15		14:99	
H2	1.42	3.77					4.66	
CO2	9.94	20.79				2463	23.31	
12	3.61	3.17	3.14	3.74		2.94	3.01	
CH _l	8.65	357	3.00	3.92		2.69	4.21	
C2HI	.54	1.44	1.63	1.57	1.64	1:72		
C2H6	-1.09	0.67	173	.68	.63		1:78 :78	
C3H6	and the second s	171- 91		238 - 16.0		1.68	2,54 19.4	
с ₃ н ₈	.90 64		.8 4.6			80 7.0	.83 6:4	
СЦВВ			255 123					
С <u>Ц</u> Н ₁₀						2.75 20.3	.92 6:1	
CGA RPO					.32 24	.45 3.4		
WATER SOLUBLE CHEMICALS		3.23 - 9.8	3.95 12.6		1.06 34.7			
PROCESS WATER	7104 -						269 1107	
							15.49 -	
	71.0	V1.5	74.5	67.9	91.4	11.7	75.5	
Dire				0 (=				
R/FF	1.23	134	1.32	0.93	0.76	0183	0.64	

NI

TABLE II

		1		1		
DATE	2-9652	2-10-12	2-11-52			
OPERATING CONDITIONS		1				
SYNTHESIS GAS - MNSCFH	2,36	2,66	2.52			
RECYCLE GAS - MMSCFH	3.62	440	4,49			
REACTOR TOP PRESSURE - PSI	242	350	350			
REACTOR BED TEMPERATURE - P	500	600	630		·	
CATALYST HOLDUP - TONS	124	136	132			
Density - #/cf	1/6	116	128			
% Fe & % C						
SPACE VELOCITY - V/V/Hr	1998	1043 -	1114			
H2/CO RATIO OR PRESH PEED	187	1.91	1.86			
conversions - co %	3.30	82.62	84.40			
н ₂ ≰	DC.52	59.92	61.57			
COÆH₂ %	68.47	67.71	69.55			
CONTRACTION %	48.05	39.80	45.97			
SELECTIVITY C3/C1/%	125-4	72.64	7460			
PRODUCTION - OUTPUT BASIS GALS 63//M OF 4/00						
COMPONENT YIELDS		•	WI'M EPH			
Å	1.95	202	2.7/			
CO H	1/43	1197	Muig			
H ₂	3.44	340	320			
^{CO} 2	1804	19.63	19.42			
42	2 PA	2.97	3.68			
CHL	289	3.14	2.91			
C ^S ⊞f [†]	1.52	1.58	1.80			
C5H6		.79	.80			
C3H6		7	2.77 10.9			
^C 3 ^H 8		81 41	.78 3.8			
сџвв	1.76 6.7	1.84 7.9	1.61 6.7			
с _Ц н ₁₀	1.30 12	30 /03	.27 1.2			
C5≠ RPO	# * · · · · · · · · · · · · · · · · · ·		2.04 23,2			
WATER SOLUBLE CHEMICALS		3.61 10.2	4.25-11.6			
PROCESS WATER	21.48 -		20,89 -			
	. 47.7					
RIFF	1,53	1,65	1.78			

TABLE I

DESCRIPTION OF CATALYST SAMPLES SUBMITTED

PLANT RUL 10.16

Date Caught	2-17-52	2-18-52	2-19-52	2-20-52	2-22-52
Bed Life Hrs	Note 1	Note 2	Note 3	24	72
Lab No. B∞	39415	39477	39545	39642	39799
Sample No.	16-1	16-2	16-3	16-4	16-5
% Fe	86.07	87.57	83.5	79.71	80,48
% K ₂ 0	1.45	0.78	0.69	0.48	0.52
% C	3.58	6.57	7.56	10.01	10 26
Dry Sieve Analysis					
on 40	0.8	0 . 2	1,00	0 8	1.2
60	7.8	11.0	12.0	9 . 8	10,6
80	11.2	16.0	17.6	14.6	16,4
. 100	11.2	13.0	14=4	12 2	13.4
120	7.6	9.2	11,0	8 - 6	9.2
200	31.0	31.6	28.4	35.8	38.8
Pan	30.6	19.0	158	17.6	11.4

Note 1 - This sample obtained at end of loading.

Note 2 - This sample obtained at end of H2 "Pretreat" and prior to "Carbiding Cycle".

Note 3 - This sample obtained at end of "Carbiding Cycle".

DATE	2-19-52	2∞20=52	2-21-52	-Note 1- 2-22-52			·		
OPERATING CONDITIONS									
SYNTHESIS GAS - MMSCPH	2.34	3.30	4.51	4.06					·
RECYCLE GAS - MMSCPH	4.06	3.4516	2.65 216	2.98 2.04	·				
REACTOR TOP PRESSURE - PSI	350	350	350 350	350					
REAUTOR BED TEMPERATURE _ Op	650	680	650	ú20					
CATALYST HOLDUP - TONS	153	142	134	137					
" DENSITY - #/CF	120 y.C	120	114	114					
% Po & %C	80.86 11.5	79.33 12.3	77.60 11.3	BO.48 10.3					
SPACE VELOCITY - V/V/Hr	818	1288	1808	1606				·	
H2/CO RATIO OF FRESH FEED	1.780	1.824	1.836	1.931		!			
CONVERSIONS - CO %	44، 82	81.38	76.81	75.19					
H ₂ %	56.98	55.58	45.89	43.54					
coxH2 %	66.14	64.71	56.79	54.34					
CONTRACTION %	58.08	43.09	35. OL	33.26					
SELECTIVITY C3 &C1 &%	72.67	67.83	69.80	71.07		-			
PRODUCTION - OUTPUT BASIS	wt% BPH	Wt% BPH	Wt% BPH	Wt% BPR					
COMPONENT YIELDS	w170 April	ut% BOH	wts Boy	mt22 BOH					
A	2.32	2.12	2.10	2.16					
CO	11.83	12.91	16.03	16.55					
н2	3.71	t-0t	4.94	5.24					
co2	22.97	33.45	24.96	23.22					
N ₂	5.52	4.15	3.51	3.80					
CH ₁	2.75	it. Oft	2.84	2.97	·				
c ^S H ^f	1.82	1.70	1,61	1.68					
c ⁵ H6	1.08	.91	.68	.66					
с3н6	2.26 10.3	2.57 16.1	1.93 16.4	1.83 14.0					
с ₃ н ₈	.67 3.1	.78 5.0	.57 5.0	.53 4.1					
с _{ГГ} н ⁸	2.28 9.0	1.94 10.5	1.58 11.6	1.78 11.8	***				
CLH ₁₀	.59 2.4	.28 1.5	.28 2.1	.35 2.4					
C ₅ ≠ RPO	5.44 17.30	5.11 21.8	4.63 26.8	5.41 28.5					
water soluble chemicals		at 3.56 12.0		the state of the same of the s					
PROCESS WATER .	5 25.18 545	16.42 66.9	14.69 75.3	11.13 76.14					

Note 1 - Unit Shut-down 9:30AK 2/22/52 after 32 hours on this operating day.

TABLE I

DESCRIPTION OF CATALYST SAMPLES SUBMITTED

PLANT RUN NO. 17

Date Caught	3-3-52	3-4-52	3~5 -5 2	3-14-52
Bed Life Ers	Note 1	Note 2	24	Note 3
Lab No. B =	40286	40328	prodoft	41399
Sample No.	17-1	17-2	2.7=3	17-4
% Fe	84.5	79.3	75.9	75.6
% K ₂ 0	0.49	0.30	0.54	0,38
% C	9.25	13.0	11.3	12.3
Dry Sieve Analysis				
on 40	0.6	2,0	12.0	1.4
60	10,2	10,0	15.2	12.6
80	19.4	18,0	25.6	59.4
100	11.4	8.4	7.2	0.2
120	7.4	7.0	2.6	0.4
200	37.0	32° li	29.6	24.2
Pan	14.2	21.0	18.0	1.4

Note 1 = This sample caught at end of "Pretreat" cycle. See attached letter.

Note 2 - This sample caught at end of "Carbiding" cycle. See attached letter.

Note 3 - This sample caught when emptying Feaster at end of Run 17.

No intermediate samples obtained due to low bed level in Reactor.

K-351-A	REACTOR	RUN	17	OPERATING	åc	YIELD	SUMMARY
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			The second secon				enterna jaroja paraturo (* epojaji). Po degas katalas karaturo (maja ili													
DATE	3-4-52		3 - 5·	-52	3-6-	-52	3-7-	-52	3-8-	-52	3-9-52		3-10-	-52	3-11-5	52	3-12-	-52	3-13-	52
L ERATING CONDITIONS					namponor de ligite por l'Alliana les		han dankan cahan sand													
SYNTHESIS GAS - KESCPH	3.0h		4.	.49	4.	.20	4.1	4	4.	.42	4.38		4.	.52	4.62	2	4.5	56	4.47	
RECYCLE GAS - MESCPH	3.04		2.	.27	2.	.78	2,6	52	2.	.50	2.93		3.	.13	3.07	7	2.84		2.83	
REACTOR TOP PRESSURE - FSI	350	`		350	375		375		375		375		375		370		375		375	
REACTOR BED TEMPERATURE - OF	685		9	500		650	66	60	(675	690		(565	650		61	1 5	645	
CATALYST HOLDUP - TONS	91		·	73	68		51		83		67		50		72		55		42	
n DENSITY - #/CF	94		·	92		86		64		104		88		58	98		72		60	
% Fe & % C	78.9	9.0		#		#	0.0	- 4		- #	a n 1	p.		#		#	-	#	96	*
SPACE VELOCITY - V/V Hr	1414		2	516	2	357	27	26	25	552	2690	6	26:	14	284	6	27	08	302	7
H2/CO RATIO OF FRESH FEED	1.809			1.894	1	.831	1.	883	1.	. 836	1.8	51	1.	847	1.8	31	1.	802	1.8	37
CONVERSIONS - CO %	84.13		6	7.17	6	6.46	70.	. 94	6	3.51	66.	30	56	.55	63.1	43	57	. 78	65.	57
H ₂ %	57.23			41.57		40.26	111	.39	3!	5.52	39.	19	35	.78	38.	25	42	.50	44.	24
CO/H ₂ %	66.80			50.42		49.52	53	.60	4.	5.39	48.	70	43	.08	47.	14	51	.52	51.	70
CONTRACTION %	47.06			28.52	32.77		31.48		28.90 28.26		26	24.08		24.55		30.68		33.71		
SELECTIVITY C3/C1/ %	72.25			78.23		66.64		.42	65.82		67.81		71.08		61.60		68.73		66.91	
PRODUCTION - OUTPUT BASIS	Wt%	врн	Wt%	BPH	Wt%	врн	Wt%	ВРН	n't%	BPH	Wt%	врн	Wt%	ВРН	wt%	BPH	WtX	врн	Wt%	BPH
COMPONENT YILLDS																				
A	2.13		2.13		2.12		2.12		2.18		2.05	***************************************	1.93		2.10		2.05		1.88	
CO	10.99		22.75		23.49	-	20.35		25.48	·	23.84		30.44		25.91		22.58		24.18	
H	3.86		5.52		5.51	•	5.28		5.95		5.73		5.98		5.77		5.23		5,19	
co ₂	22.32		23.38		22.49		24.91		21.69		23.61		17.06		21.70		20.86		17.77	
NS	3.89		2.83		2.93		3.20		3.24		3.36	-	3.17		3.27		2.84		2.95	
СНĮ	3.16		2.05		3.33		3.38	-	2,68		2.76		2.01		3.04		2.40		2.21	
c ^{SHf}	1.74		1.56		1.55		1.54		1.35		1.26		1.11		1.28		1.34		1.32	
C2H6	.67		.38		-49		.46		.41		.57		.28		.45		.54		.31	
С3Н6	2.81 1	6.3	1.37	11.4	2.02	15.9	2,14	17.6	1.33	11.1	2.11	17.2	1.75	14.9	1.58	13.6	1.35	11.7	1.66	13.9
C3H8	.88	5.2	.41	3.5	.47	3.8	.48	4.1	.57	4.9	.60	5.0	.82	7.1	.38	3.3	.46	4.0	.47	4.0
C14H8	1.60	8.0	1.77	12.7	1.34	9,2	1.34	9.5	1.17	8.4	1.28	9.0	1.11	8.1	1.02	7.6	1.49	11.2	.98	7.1
C4H10	.25	1.3	.23	1.7	.23	1.6	. 34,	2.5	.30	2.2	.17	1.2	.49	3.7	.36	2.8	.33	5.6	.10	0.7
C5/ RPO	6.09 2	3.1	4.59	27.10	4.10	21.4	2.93	15.8	2.93	16.5	3.20	17.0	1.87	10.3	2.37	13.6	3.53	21.6		11.7
WATER SOLUBLE CHEMICALS	2.87	9.5	2.60	12.4	2.57	11.6	2.95	13.8	2.24	10.6	2.30	10.7	2.30	11.0	1.93	9.5	 	11.5	 	11.6
PROCESS WATER	19.87	-	11.97		12.10	-	12.53		12.23	-	12.25	•	13.73	•	13.57	-	16.98	*	20.88	

^{*} No Sample obtained due to low bed level in Reactor.