

SINCLAIR REFINING COMPANY

July 26, 1940

1783

Translation Book #170

Drawings

Reel 123

Frame 139

Two Drawings

S-102

Report - L-T Distillation of Oil Shale by the process of Dr. L. Schweitzer June 6, 1945

The Process : Oil shale is charged into an iron shaft kiln, and ignited at the top. Air is sucked from the top to the bottom, causing the combustion zone to travel downwards in the charge. Ahead of it travels a temperature drop, in which the shale carbonizes, or under goes a thermal decomposition. The so-called fixed carbon remains in the shale layers. This fixed carbon is burnt and keeps up the exotherm process in the combustion zone. The hot exit gases of the combustion zone provide the heat required for the endotherm carbonization process and ensure that there is no oxygen in the air. When the carbonization and combustion zones have reached the bottom of the kiln, operation is discontinued; the kiln is discharged, re-charged and re-ignited.

The Distillation products of the thermal shale decomposition flow with the combustion gases of the combustion zone downward. A certain part will condense in the cold shale. The major part of the hydrocarbons evolves as an oil mist. (The particles of which the mist consists are precipitated by an electrofilter (electrostatic gas purifier). In the exit gases are the low molecular hydrocarbons (benzines), which are absorbed by washing oil. The enriched washing oil is then freed from benzine by distillation, and re-used. One of the distillation products is also the carbonization water. Part of it will already condense in the kiln, that is, as long as the temperature of the shale layers subsequent to the carbonization zone has not yet risen. This will be the case until the combustion zone has approximately reached the bottom of the shaft. Otherwise, the carbonization water is contained in the oil sprays and is precipitated together with them, mainly in the electro-gas-purifier. When the condensation is terminated, nitrogen which had been introduced with the combustion air will be the main constituent of the exit gases. Moreover, these gases will contain residues of non-converted oxygen, CO and CO₂ from the combustion zone, slight amounts of combustible gases and gaseous hydrocarbons, so that the exit gases are not of a substantial heating value.

The solid carbonization products consists of inconsiderable amounts of fixed carbon which is burnt, and of the anorganic slag.

Dr. Schweitzer developed the process in the following steps.

- 3.) Lias - plant Frommern: 28 Kilns Ø 3000mm
- 2.) Pilot plant NETZINGEN: 1 kiln Ø 1500mm and 1 kiln Ø 3000mm

Pilot plant NETZINGEN

FEED: shale, grain size 0-300mm, bulk density 1.2
Carbonization advance: 0.2-0.25m/hr.
Fisher analysis of charges: 4.5%
Yields of oil obtained: ca. 77% of Fischer content
" of benzine " : ca. 10% " " " "
Carbonization (exit) gas: ca 500m³/to of shale/ca. 500cal.

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Lias -- plant Proment;

The carbonization and condensation processes differ from METZ WINGEN scheme only in so far that ahead and behind the blower two more coolers have been installed.

The 28 carbonization kilns are installed within the reach of a crane which moves on tracks. A three - motor crane brings the carbonized kilns to the dumping grounds and takes re-charged kilns to the carbonization place.

M. Beth

M. Beth

June 18, 1948

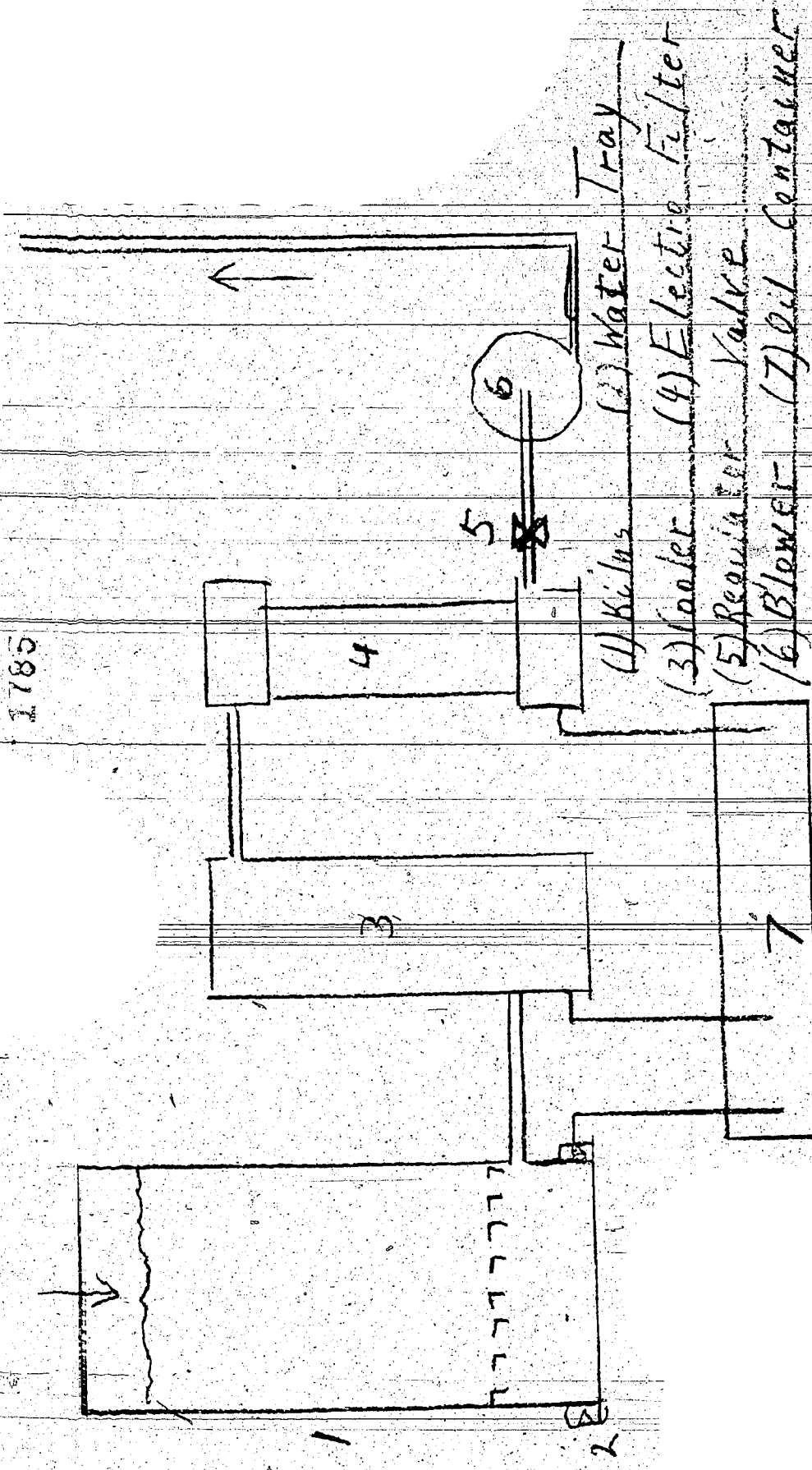
July 22, 1948

ME: BMC

Distribution: All Divisions

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Experimental Plant Metzenen

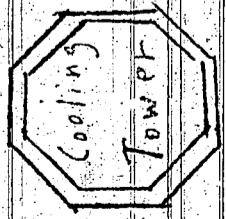
LIAS PLANT FROMMERN

M 1500

Tank

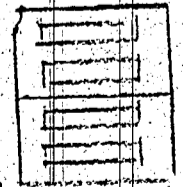
Tank

Tank



Exit gas

Distill. Tanks



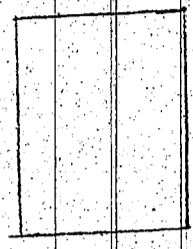
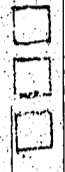
Washer



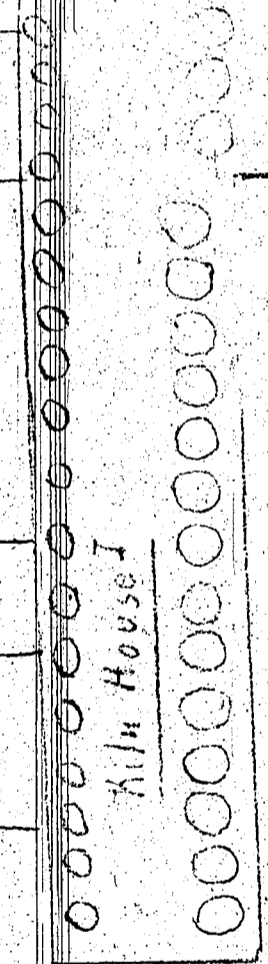
Oil Container



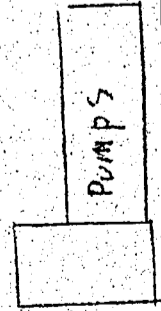
Labor



Kiln House I



Kiln House II

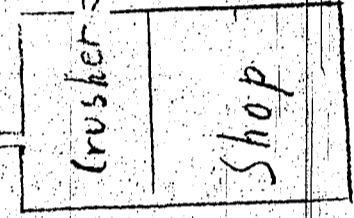


Pumps

Slag



Shake



Silo

