

SINCLAIR REFINING COMPANY

Spherical Chromium-Alumina Catalyst

(a) Polymerization of n-butane

The following values have been obtained in long-time experiments with n-butane, using a spherical catalyst prepared from a chromium-alumina catalyst.

Charge 1:500 Total time in operation (without regeneration) 390 hrs

In operation for 80 separate periods, averaging 5 1/2 hrs
initial values: 40% conversion rate with 60% output in the course of 7 1/2 hrs
terminal values: 35% " " " 30%

At the end of the catalyst was broken into pieces. Initially we tried to manufacture a catalyst by adding binding material and adjusting the composition of the catalyst. The following table shows the conversion rate and the amount of the catalyst are affected by surrounding agents.

Its motion is conspicuously more even, even though not yet quite smooth. The thermocouple case, mounted in the axis of the pipe, may have a disturbing effect. The conversion rates obtained with the spherical catalyst are as high as those given by catalyst lumps. Since wear and tear was excessive on using a screw conveyor we may install a conveyor like in Lemma. When the catalyst enters the conveyor or the drum, it will frequently get stuck.

The comparison of the results obtained at Lemma and Agnes shows clearly that the Agnes catalyst is by far superior to the "old" Lemma catalyst and about equal to the "new" Lemma catalyst with respect to conversion rate and output. But it is not yet possible to manufacture the new Lemma catalyst in major charges, whereas the large-scale manufacture of the Agnes catalyst offers no difficulty.

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