

FILM STUDY GROUP

SUBJECT INDEX AND REPORT

T.O.M. REEL NO. 67

Prepared by

GULF RESEARCH & DEVELOPMENT CO.

GULF RESEARCH & DEVELOPMENT COMPANY

Pittsburgh, Pennsylvania

ABSTRACT AND INDEX OF TECHNICAL OIL MISSION

MICROFILM

REEL NO. 67

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SCANNING OF REEL NO. 67

BAG 3445 TARGET 30/5.01

RUHRCHEMIE A. G. - STERKRADE-MOLTEN

Frame No.

ITEM 32: Iron Catalyst for Low Pressure

1

A short memorandum discussing an iron catalyst of special preparation suitable for a temperature of 200 to 225°C. and pressure of 15 ats. Nothing is said about the preparation of the catalyst. The experiments are reported in Reel 66, Item 25.

ITEM 33: Preparation of Oil of High Olefin Content

5

This is a very short report of one page on experiments described in Reel 66, Item 25.

ITEM 34: Report of 1941 of Recent K.W.I. Work (Outline)

7

This is a short report on polymerization of definite mono-olefins to viscous oils and city gas production in connection with the Kogasin synthesis.

ITEM 35: Micro-Iodine Number

16

The Bromine number is not reliable especially for determination of olefins for the OXO synthesis. The micro-iodine number is said to be reliable.

ITEM 36: Research Program for Medium Pressure Experimental Plant

18

These are only short program outlines for the experiments which are described in Reel 66, Item 25.

ITEM 37: Conference with Krupp on Recirculation of Water Gas

39

This is a short report on a conference with Krupp concerning hydrocarbon synthesis directly from water gas describing the operation, gas analysis and yields (see Reel 66, Item 25).

ITEM 38: Petroleum Industry of Russia, 1941

43

This is a printed publication from the high command of the Army about occurrence of petroleum in Russia, the composition of petroleums from the various fields are shown and location indicated by maps. There is a further chapter on refineries and the products from petroleum, the difficulties of the Russian petroleum industry and pipe lines and other means of transportation.

ITEM 39: Effect of Additions to the Properties of Fuels.

66

A printed pamphlet from V.D.I. Berlin for 1941 discussing the drawbacks of gasolines to which were mixed ethyl and methyl alcohol as is customary in Europe. The main drawbacks are sensitivity against water which is miscible with alcohols and brings about stratification into two layers and limited miscibility of methyl alcohol in gasoline. Such gasolines are especially sensitive to low temperatures because they become cloudy and stratify more easily into two layers. The paper deals with additions to such gasolines to counteract these drawbacks. Addition of dimethyl acetal is mentioned. Many tables and graphs for ternary mixtures are given.

Another drawback of the addition of methyl and ethyl alcohol to gasoline is the formation of minimum azeotropes which bring about a lowering of the temperature at which the gasoline begins to distill and may cause vapor lock. The addition of higher alcohols as well as addition of other compounds which are already used for addition to gasolines such as benzol, toluol, xylol, acetone, and isopropyl ether counteract this satisfactorily. Many graphs and tables and 15 references are given.

ITEM 40: Apparatus for Determining Absorption Isotherms

79

This is a short discussion of the dynamic and static method for determining absorption isotherms and a description of the apparatus for the static method which is considered superior for gases of low molecular weight.

These pages give sketches of the apparatus for the static method for determination of the absorption isotherm, and a graph of the ethylene isotherm.

ITEM 41 Parts of Summary of Licensees Results 1938-40

84

This is a graphical and tabular representation for different works showing percentage CO conversion, percentage contraction and other data, such as space velocities, H_2/CO , etc.

ITEM 42: Overhauling Oven 51

92

This is a lamella oven which was difficult to empty and therefore the oven was inspected and overhauled. The item is almost illegible.

ITEM 43: Time required for Emptying the Oven

95

These are details showing time and working hours required for emptying an oven.

ITEM 44: Detailed program for synthesis, Normal Pressure 1938-1940

98

There are many graphs showing the desired program of days of operation as abscissa and degrees C. as ordinate. Figures for space velocities, contraction, etc. are shown on the graphs.

ITEM 45: Specifications for Treib Gas 105

These are specifications for mixtures of C₂ to C₄ hydrocarbons with respect to heating value, vapor pressure and impurities.

ITEM 46: Extraction - Oil 106

This is a table giving the distillation curve for extraction oil for Ovens 31 and 32 of March 28, 1941.

ITEM 47: Results of the Recirculation with an Injector 108

This is a short plant report concerning preliminary results in the water gas recycling synthesis to produce more olefins by using an injector for the recycling of the off gas without removing the light naphtha.

ITEM 48: CO/H₂ ratio for medium pressure synthesis 114

This is a memorandum concerning the composition of the synthesis gas, the gasoline and the CO/H₂ ratio in the off gas in stages I, II, and III. It was found that gasification increases in the second and third stage if the CO/H₂ ratio increases about to 1.8, therefore the ratio should be reduced in the second stage.

ITEM 49: Water Gas Conversion 116

These are two flow sheets showing the catalyst ovens, saturators, coolers, etc. and a graph of data during the starting of the pressure oven in November, 1938 showing the hours of operation as abscissa and space velocity, temperature and contraction as ordinates.

ITEM 50: Medium Pressure Synthesis, 1938. 119

This is the same graph shown in Item 49.

ITEM 51: Scheme for Working Up Products 121

This gives the flow sheet for processing C₃ & C₄ hydrocarbons and light and heavy naphtha from the activated carbon. A flow sheet is given for the C₃ & C₄ hydrocarbon washer, extracting the CO₂ with K₂CO₃.

ITEM 52: Work In Block 9, Stage 1 (High Methane Formation) 125

Tables and a graph of results when starting Oven 92 at a space velocity of 500 m³/hr. at 161°C. are given. The CH₄ content of the off gas in the first 35 hours was up to 50 per cent and then decreased within a few hours to 3 to 5%.

ITEM 53: Details of Ovens 129

The measurements of the catalyst chambers are given as follows:

Low pressure ovens:

Lamellas 5000 x 1525 x 2500 mm. high.
555 plates, 1.6 mm. thick
7.4 mm. apart

Aggregates 1 to 5 and 8 to 12

126 tubes, 42.5 mm. diameter on top.
504 tubes 35 mm. diameter at the bottom
Catalyst volume 12.95 m³

Aggregates 6 and 7

630 tubes 35 mm. diameter
Catalyst volume 13 m³.

Medium Pressure Ovens:

2046 double tubes 4500 x 44 i.d./24 mm. o.d.
Catalyst volume 10 m³
Water volume 12.25 m³
Water cooled catalyst surface 2200 m²

- ITEM 54: Work Instructions for Solvent and Hydrogen Treatment 130
Instructions for the hydrogenation and extraction of low pressure ovens.
- ITEM 55: Alteration in Method of Calculating Yields 134
Methods for measuring the products of medium pressure; paraffin, low pressure oil, condensate, C₃ & C₄, etc. are given.
- ITEM 56: Yields from Gas Analyses, April, 1940. 138
This is a table giving gas analyses and balance for April, 1940.
- ITEM 57: Stepwise Degassing of the Pressure Wash Water 139
A short table of procedure.
- ITEM 58: Formation of Vapor Locks in Gasoline Lines 140
This is a long report (partly illegible), from the Institute for Automobile Internal Combustion Machines of Dresden, regarding formation of vapor lock by gasolines showing the laboratory apparatus used, in sketches and photographs, and giving many curves of vapor pressure.
- ITEM 59: Loss of Cobalt at the Catalyst Factory at Helten 220
Several reports (partly illegible) describing recovery of cobalt and giving tables concerning losses.
- ITEM 60: Two Works Laboratory Notebooks 243
These are handwritten (partly very illegible) notebooks dealing with:

1. Influence of the purification of synthesis gas with activated carbon and FRN on catalyst activity.
2. Carbon determination in used catalysts.
3. CO₂ determination in life tests using different sealing liquids.
4. Absorption of CO₂ on catalysts.
5. Synthesis gas samples. This is a test to determine whether a dry gas sample changes on long standing similar to the changes on storing in contact with water.
6. Poisoning tests with condensate of Chemische Werke Essener Steinkohle.
7. Testing of catalysts under pressure with consideration of conditions of the plant.
8. Tests on synthesis gas I over a period of 6 to 10 hours.
9. Determination of the activity of catalysts by testing the oxygen absorptivity.
10. Continuation of Number 7.
11. Clay catalysts. These are catalysts prepared from clay with cobalt and thorium oxide. Treatment of clay with HCl is described.
12. Testing of catalysts from the plant.
13. Comparative reduction experiments.
14. The influence of adding water vapor to synthesis gas especially with respect to conversion under conditions of the synthesis.
15. Sulphide decomposition of spent Feinreinigen Mass.
16. Corrosion testing of several paints for resistance to gasoline, oil, heating oil and cracked gasoline.
17. Testing of the activity of catalysts, poisoning experiment with Rheinpreussen gasoline.
18. Testing of the activity of catalysts, addition of acetylene.
19. Experiments on desulphurization of waste water by absorption of the sulphur set free.
20. Concerning pyrophoric spent cobalt thorium catalyst.

21. Olefin determination on products. (Table).
22. Carbon monoxide conversion on cobalt thorium magnesium catalysts. (Short table).
23. Cobalt in paraffin.
24. Effect of sulphur purification in synthesis (table and graph).
25. Laboratory experiment concerning H_2S from pressure cooling water.
26. Sampling of C_3 and C_4 hydrocarbon fraction.
27. Airing of Hoesch catalyst (cobalt thorium magnesium).
28. Aging experiments.
29. Impregnating experiments.
30. Testing of catalyst activity in Oven 2 (Impregnating experiment).
31. Experiments on reduction of catalyst.
32. Investigation of water gas before fine purification for C, H, and carbonyl.
33. Investigation of chlorinated lime for effective chlorine.
34. Continuation of experiments on working out a method to test catalysts at medium pressure.
35. Same as No. 32.
36. Testing of the activity of a cobalt catalyst with a mixed gas containing ammonia.
37. Absorption of oxygen on cobalt thorium catalysts.
38. Short newspaper clipping concerning separation of oil emulsions from Rundschau deutscher Technik No. 41, Oct. 1940.
39. Dismulgan (this is the compound described in the newspaper clipping).
40. A. S. O., a crude purification mass (Japanese).
41. Interferometric experiments on mixtures of nitrogen and hydrocarbon.
42. Experiments concerning spontaneous inflammability of paraffin in insulating materials.

43. Action of ammonia on alkazid liquor.

ITEM 61: Stability of Paraffin Waxes

635

A graph showing the chemical stability (Oct. 19, 1942) of normal cake paraffin (liquefying point/52), RB hard wax (liquefying point 100°C), and catalyst paraffin (liquefying point 100°C.). Days at 130°C. shown as abscissa and the neutralization number as ordinate.

ITEM 62: Process for Removing Calcium from Cobalt Solutions

636

German Patent No. 683691 of November 13, 1939. This patent describes the removal of calcium as calcium fluoride by adding to the solution a water soluble fluoride.

ITEM 63: Catalyst Factory at Holten

640

These are 6 flow sheets showing the plant at Holten.

ITEM 64: Synthetic Soaps.

641

These are several reports on synthetic soaps especially from the fatty acids formed in CO hydrogenation and includes the separation of the higher fatty acids. The OXO synthesis is mentioned. This item also includes a report concerning the separation of alcohol from the waste water of the activated carbon plant which contains an average of 0.4 per cent alcohols, aldehydes, and ketones.

ITEM 65: aromatization

659

This is a report on the aromatization of straight chain aliphatic hydrocarbons from the Fischer Tropsch synthesis using chromium oxide and alumina-chromium oxide. There is a description of the work done and plans for large scale operations.

ITEM 66: Preparation of Isopetrol

673

A report on the preparation of iso-C₄ hydrocarbons and aviation gasoline from synthesis products. Polymerization of C₃ and C₄ hydrocarbons with phosphoric acid catalysts and the cracking of Fischer Tropsch products for aviation gasolines is described (Houdry)

ITEM 67: Preparation of Isopropyl Alcohol and Methyl-Ethyl Carbinol.

695

Description of a process for producing alcohols from olefins using sulphuric acid, describing the difficulties of the process and giving data with regards to acid concentration, ratio of acid to olefin and the temperature during olefin absorption.

ITEM 68: Inhibitor for Resin Formation.

704

This is a description of the inhibitor "Stadisol" which inhibits gum formation in cracked gasolines. The practical use of Stadisol is described but nothing is mentioned concerning its composition. It appears to be hydroquinone or a similar compound, it is stated that alkalinity should be avoided.

ITEM 69: Detailed Summaries of Yields, Etc. 1943-1944

710

These are tables showing yields in the Fischer-Tropsch synthesis and data on CO/H₂ ratio, end gas composition, contraction, CO + H₂ conversion, new formation of CH₄ and CO₂, synthesis gas m³/hr. and age of oven.

ITEM 70: Production Summaries, 1943-1944.

726

This is a collection of 10 day interval summaries for the synthesis which gives figures for oil and paraffin, synthesis gas, water gas, ratio synthesis gas to water gas, etc. (This is partly illegible).

ITEM 71: Miscellaneous Analytical Methods

806

1. A determination of the tar and coke number of fresh oil according to Ehlers. 807
2. Directions for testing aviation gasolines for use in diesel motors. (Changes and explanations). 809
3. Determination of the distillation curve for product boiling in the gasoline range region, for condensate oil and diesel oil, for solid products and for the total product. 812
4. Determination of content of olefins in aromatics. 815
5. Determination of the specific weight. 817
6. Determination of the iodine number. 821
7. Determination of the aniline point. 823
8. Determination of the vapor pressure. 824
9. Determination of neutralization and saponification numbers. 825
10. Determination of octane number. 827
11. Determination of the softening and the drop point. 828
12. Determination of moisture in gasolines. 830
13. Determination of the flash point and ignition point. 832
14. Determination of the solidifying point and melting point. 837
15. Determination of the evaporation tests and the bomb test, with figures and illustrations. 842
16. Determination of the absorptive power of wax for oil. 852
17. Octane number according to the motor method, including determination of the content of aromatics in gasolines. This is hardly legible. 853

18. Rapid determination of cobalt.	873
19. Improved calorimetric determination of the heating value of liquid fuels according to Angewandte Chemie, p. 146 (1938).	875
20. Separation of propylene and ethylene in the presence of CO and ethane.	878
21. Testing of diesel oils for miscibility.	895
22. Investigation of alkaline mercury cyanide solution for absorptive power for ethylene and CO.	896
23. Determination of gaseous paraffin hydrocarbons.	899
24. Determination of the diene number of hydrocarbons.	908
25. Testing of diesel fuels.	910
26. Behavior of diesel fuels to zinc.	912
27. Filterability of diesel gas oils.	914
28. Determination of molecular weight according to Rast.	926
29. Determination of chlorine in dechlorinated paraffins.	927
30. Determination of resin in gases Brennstoffchemie p. 24(1940).	928
31. Determination of the residue on evaporation (resin content).	929
32. Change of method of determining the sulphur content (with illustrations).	934
33. Testing of the products made from fats.	940
34. Determination of the fatty acid content of fatty acid liquors.	942
35. Detection of benzene in gasolines (Rung method).	943
36. Determination of the CO and Rhodan number (thiocyanate number).	944
37. Same as No. 35.	953
38. Determination of olefins with phosphorus pentoxide-sulphuric acid.	954

39. Modified method to determine the neutralization number and saponification number of gasolines, diesel oils and lubricating oils.	957
40. Behavior of light gasolines to copper in the corrosion bomb Heinze-Marder.	986
41. Dimethyl sulphate number.	987
42. Total impurities.	988
<u>ITEM 72: Decade Conferences 1943-1944.</u>	989

These are 10 day interval reports on operation of the Fischer-Tropsch Synthesis plant. They deal mainly with the difficulties in operation, especially with emptying the ovens, overhauling the generators, quality of water gas delivered, repairs made during shut downs, different coke qualities, etc.

<u>ITEM 73: Oxidation of Waxes</u>	1071
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This is a short report of January, 1942, on the raw materials used, yields and difficulties due to power shutdown for an experimental plant. A flow sheet for the semi-commercial plant is included.

<u>ITEM 74: Miscellaneous Patents of General Interest</u>	1074
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1. Patent 746,767, 8-23-44, Herman Pemsel. Process to increase the activity of Fullers' Earth. This is a continuation of patent 711,454. Claimed is an improvement in drying by soaking the purified Fullers' Earth with a neutral liquid of a boiling point higher than water and drying by bubbling through warm air. 1075
2. Patent 746,573, 8-12-44, Gehr Herrmann, (Inventor Max Stauber). Process for production of colloidal silica solutions. This is a continuation of Patent 739,751 of 12-4-42. It describes a 2 stage method of neutralizing sodium silicate solutions with ion exchange resins. 1077
3. Patent 746,512 of 6-19-44, Christian Stiasni. Device for feeding liquid gas into internal combustion engines. 1079
4. Patent 746,303, 7-21-44. Siemens-Lurgi Electrofilter Co. Inventor Friedrich Wilhelm Hoss. Device for supplying high tension current into gas filled chambers such as electrofilters. 1081
5. Patent 746,074, 6-17-44. I.G. Farbenindustrie. Inventor Heinrich Medick. Continuation of Patent 738,521 process for preparation of ethylene derivatives. This is a pharmaceutical patent for producing estrogenic compounds. 1084
6. Patent 740,772, 4-25-44, Karl Schmid. Process for continuously reacting mixtures of liquids which form separate phases of different specific weight. 1086

7. Patent 744,686, 6-21-44. Auer Gesellschaft A. G., inventor Walter Stiller. Protective filter against inorganic hydrides of multivalent metalloids. The patent describes the use of silver salts or finely divided silver or silver oxide on activated carbons for H_2S , etc. 1088
8. Patent 740,013, 7-21-44. Aug. Schnakenberg & Co. Device for evaporating liquids which flow in thin layers over inclined planes with heating canals. 1090
9. Patent 746,636, 8-16-44. Julius Pintsch, Inventor Hellmut Stock. Still for continuous heating of light oils from low temperature carbonization for vapor phase refining over solid catalysts. 1093
10. Patent 746,541, 8-12-44. Felix Engelhardt. Process for production of citric acid soluble phosphate fertilizers. Heating of all phosphates with sodium nitrate from 500 to 700°C. under oxidizing conditions. 1096
11. Patent 745,338, 5-27-44. W. Cuypers & Stalling, Inventor. Paul Cuypers. Lubricating oils for watches and similar instruments. It is claimed that the addition of small amounts (approximately 1 to 2 per cent) of metal soaps especially aluminum stearate gave improvement in oiliness, retention of the oil in the bearing and makes unnecessary the addition of larger amounts of fatty oils as currently practiced. 1098
12. Patent 746,787, 8-25-44. I. G. Farbenindustrie Inventor Fritz Fried. Cast waxes especially for dentistry. The patent claims an improvement by using synthetic resins instead of the natural resins currently used in compounding the waxes. The waxes show better properties. More synthetic resin can be added than natural resin. as an example cyclohexanone resins are used according to German patents 337,993; 357,091; 511,092; and 711,888. 1100
13. Patent 745,684, 5-15-44. I. G. Farbenindustrie Process for making highly condensed polyamides from diamines and carbonic acid. These are super-polyamides similar to nylon using carbonic acid instead of long chain dibasic acid as in nylon. 1102
14. Patent 744,318, 6-17-44. I. G. Farbenindustrie Process for making concentrated stable emulsions. The patent describes polymerization in emulsion for synthetic plastics. By polymerizing the mixture with small amounts of vinyl sulphonic acid or salts and an emulsifying agent of soapy character. 1105
15. Patent 746,670, 8-18-44. Aktiengesellschaft der Kohlenwertstoff-Verbände Gruppe Benzin-Benzol-Verband. Inventors Rudolf Weller and Josef Gohm. 1107

Process for improving diesel fuels. The invention relates to an improvement of the filterability of diesel fuels containing amorphous paraffin to produce diesel oils of low cloud point by adding 0.05 to about 0.5 per cent of crystalline art paraffin.

16. Patent 746,861, 8-28-44. I. G. Farbenindustrie, inventor 1109
Rudolf Halder and George Weber. Device for determination of
the lubricating property of oils. The device measures the
wearing of a test piece which is pressed against a moving steel
band. Two drawings are included.

17. Patent 746,304, 7-21-44. I. G. Farbenindustrie, inventor 1111
Hans Hauber and Josef Hirschbeck. Process for obtaining
hydrocarbon oils suitable as insulating oils. The invention
describes the production of synthetic lubricants made by
polymerization of higher olefins such as propylene and butylene
with Friedl-Craft catalysts such as aluminum chloride, zinc
chloride, ferric chloride and especially boron fluoride. In
contrast to synthetic lubricants made from ethylene the lubricants
described are, with regard to viscosity, etc., especially suitable
as insulating oils. The olefins may be obtained from Fischer-
Tropsch synthesis or by cracking of petroleum. The gas used for
polymerization should contain at least 30 per cent, preferably
40 per cent olefins. The following gas composition is given as
an example: CO₂ - 12.00 per cent, CO - 1.25 per cent, H₂ - 1.00
per cent, ethylene - 1.70 per cent, propylene - 20.80 per cent,
butylene - 19.60 per cent, amylene higher unsaturated hydro-
carbons - 16.10 per cent, methane and ethane - 1.75 per cent,
propane - 11.30 per cent, butane and higher saturated hydro-
carbons - 14.50 per cent. This gas is moistened with water and
polymerized with 3.3 volume per cent boron fluoride. After the
condensation had proceeded far enough the catalyst is destroyed
and the resulting oil is hydrogenated with hydrogen over a
catalyst (for instance nickel catalyst) and then treated with
superheated steam in the presence of a little caustic. The
treatment of the superheated steam may precede the hydrogenation.
An insulating oil obtained by this method had the following
properties: Flash point - about 170°C. specific weight at 20°C.
.810, acid number - 0, saponification number - 0, iodine number - 0,
viscosity at 20°C. - 4.91° Engler, viscosity at 38°C. - 2.09° Engler.
In considering this patent application the patent office cited the
following referenced: German patents Nos. 402,990; 505,265; 507,919;
524,891; I. & E.C. 23 606 (1931). French patents Nos. 677,973;
800,956; 44,501 (addition). Dutch patent 36,210. British patents
323,805; 421,118. U.S. patent 1,756,154, J.S.C.I. (Japan) 39 21B,
J.I.P.T. 21 952, Chemisches Zentralblatt (1936) I page 1760.

18. Patent 746,572, 8-16-44. Aktivkohle-Union Verwaltungs-G. 1113
s.m.b.H. Process for drying and cooling of absorbants. A
special arrangement for drying and cooling absorbers which
are connected in series.

19. Patent 746,888, 6-15-44. N. V. DeBataafsche Petroleum 1116
Maatschapij in Den Haag, Netherland - inventors William E.
Vaughan and Frederick Farlow Rust. Process for chlorinating
hydrocarbons. This is a priority application from U.S.A. of
Sept. 2 and 30, 1939, and the corresponding U.S. patent will be
available. The patent claims chlorination in the presence of very
small amounts (about 0.005 mol per cent) of metal organic or azo
compounds which under conditions of the process give 3 radicals
such as tetraethyl lead. German patent office gave the following
reference: Chemisches Zentralblatt (1939) vol. II page 3047.

ITEM 75: Preparation and purification of synthesis gas.

1119

This item consists of the following patents:

1. Patent No. 744,224, 7-21-44. Julius Pintsch, inventor 1120
Wermer Lehrisch. Process for production of synthesis gas.
The invention describes the production of synthesis gas by
reacting methane containing gases with CO₂ and steam which are
obtained from the combustion gases on heating the methane
reactor. Part of the initial gas is used to heat the reactor,
the exit gases are scrubbed, the CO₂ liberated from the wash
water and in mixture with steam is added to the main flow of
the gas entering the reactor. A sketch is shown.
2. Patent 746,818, 8-25-44. Wintershall A. G. and 1122
Hans Schmalfeldt - inventor Hans Schmalfeldt. Process for
gasification of pulverized fuels according to patent 686,761.
This is a process of gasification using fuels which are
comparatively dry and rich in bitumen, to produce a gas poor
in methane by taking off the gas between heater and gasifier.
A sketch is included.
3. Patent 746,747, 8-21-44. Heinrich Koppers inventor 1124
Adolph Weissendorn. Water gas generator. The grate and
distributor of the water gas generator are shown in a drawing.
4. Patent 747,426, 8-28-44. F. J. Collin A. G. Inventor 1127
Paul Stoller. Process for selectively washing out H₂S from
CO₂ containing gases. The gases are washed with ammonia water
and the selectivity for H₂S is claimed to be obtained by adding
small amounts, approximately 1 per cent of phenol or homologues.
A variation is to add, with the phenol, an equal amount of alkali
carbonate. A sketch is shown.
5. Patent 746,886, 8-28-44. Karl Weiss. Dust Filter for Air 1129
and Gases. The filter employs several layers of perforated
oil soaked paper. The papers can be easily removed for changing.
A drawing is included.

ITEM 76: A Metalgesellschaft Synthesis Patent

1131

Patent 746,887, 8-30-44. Metalgesellschaft, Inventors
Helmut Weittenhiller and Wilhelm Herbert.

Process for production of hydrocarbons from CO and hydrogen.
This is a patent in which the only change in procedure seems to be that
fresh cobalt catalysts are used in the 2nd or 3rd stage with high space
velocities for about 1 to 2 months until methane formation increases and
then the catalysts are used in the first stage at a lower space velocity
whereby they operate more satisfactorily. It is claimed that this process
gives an increase of 10 g liquid products per normal m³ of gas.

ITEM 77: Filtration of Crude Paraffin

1134

Difficulties in the plant were encountered when using Fullers
Earth. Laboratory experiments were carried out and satisfactory filtration

was obtained by the addition of 10 per cent and in another case 15 to 20 per cent "tensil". The original wax was dark brown but after filtration at 150° C. with tensil the color was good, one sample being pure white.

ITEM 78: Cracking of Residues from Wax Oxidation

1136

This is a report of Oct. 17, 1939, on the cracking of residues of synthetic fatty acids. It was found that cracking is possible but not suitable for the cracking plant because the residue has corrosive properties due to its high acid number and the coke residue of 4-1/2 per cent is too high. The distillate yield was 93.4 weight per cent.

ITEM 79: Synthesis of Paraffin and Benzine at Medium Pressure

1139

This is a report of 9-19-39 regarding paraffin and gasoline from the pressure experimental plant. A considerable number of tables of boiling ranges and yields are given. The boiling range of paraffin samples and their suitability for oxidation to fatty acids was investigated.

ITEM 80: Determination of Fraction of Wax Boiling Below 400°C.

1163

This is an addition to the previous item and describes the distillation of the hard paraffin boiling below 450°C. under a vacuum of 1 mm pressure.

ITEM 81: Catalyst Wax

1165

This is a report on the analyses of oven paraffin from stages 1 and 2, hard paraffin in flakes and distillation analyses and investigation of catalyst paraffin.

ITEM 82: Refining hard wax.

1173

This is a laboratory report on the refining of paraffin wax with "tensil" and a report on the production of an almost odorless cake paraffin obtained by passing steam which had passed first a tube filled with coarse "granosil" into refined cake paraffin at 120°C. A report is also included on the extraction of used tensil from the paraffin plant with different solvents.

ITEM 83: Removing ash from wax, March 16, 1940.

1189

A report of several experiments on filtering paraffin to remove the ash, (removing of colloidal cobalt particles). Good results were obtained by acid treatment for 2 hours at 120 to 130°C. with 8.4 per cent of 75 per cent sulphuric acid, with 11 per cent of tartaric acid and with 9.2 per cent of oxalic acid. The process using the sulphuric acid is preferable.

ITEM 84: Washing paraffin wax with alkali October 1938.

1197

Laboratory experiments on mixing paraffin with 10 per cent caustic solution to determine whether an emulsion would form. It was found that no emulsion was formed.

This is a report of experiments on cracking paraffin from the pressure synthesis. Liquid phase conditions (440°C., 100 atm.) for 20 minutes gave a conversion of 17% per pass to 8% gas, 12% gasoline and 80% Diesel oil (200-320°C.). Lower pressure and higher temperature or conversion gave more gas and gasoline, less Diesel oil. The cracking experiments are described and many tables and graphs are shown.

ITEM 86: Use of Synthetic Hard Wax in the Candle Industry.

1216

This is a report on discussions with the candle manufacturers concerning usability of synthetic hard wax. The candle industry did not like this hard wax because the oil content made it unsuitable and the price was too high. The report mentions that 2 to 5% of the wax could be added with good results in the production of drawn candles. The wax in this case being used for hardening and opacity.

ITEM 87: Reactions of industry to the result of Koch and Billig, Oct. 1940

1221

This is a memo on the reaction of Ruhrbenzin personnel to an article of Koch and Billig in Brennstoffchemie 1940 No. 14 on investigations on solid paraffin hydrocarbons by determination of crystal form and reaction with chlorosulphonic acid to determine straight and branch chain molecules.

ITEM 88: Mixtures of synthetic Waxes, July 1940.

1224

This is a report on experiments determining the solidifying points of wax mixtures. Tables and graphs are given.

ITEM 89: Emulsions of Paraffin Wax, July 1940.

1232

This is a draft for a patent application on the production of wax emulsions suitable for polish and similar uses. The process consists of mixing oxidized wax (from paraffin of C₁₈ or higher) with hard paraffin and emulsifying the mixture in water. As emulsifying agents may be used primary fatty acids from CO hydrogenation which are neutralized in the mixture with alkali hydroxide or alkali carbonate.

ITEM 90: Reducing the Lower Melting Fraction in Paraffin Wax.

1234

An investigation to determine whether specification hard wax could be obtained without forming a cake paraffin. Distillation was carried out up to 440°C. and the residue was hard wax which contained 10 per cent boiling between 440 and 450°C. (10% cake paraffin). Calculations are given as to how much cake paraffin may be added to soft paraffin.

Hydrogenation experiments were carried out on crude to determine whether the amount of cake paraffin could be changed. No change was observed.

A report on determination of cake paraffin in hydrogenated crude paraffin indicates that the alcohols and esters in the water gas recycle paraffin do not influence appreciably the crystallization of the cake paraffin.

A short report describes how cake paraffin determinations are carried out.

ITEM 91: Sample Analysis of Paraffin Wax From Ruhrchemie and Licensees. 1239

Analysis of paraffin samples are given (distillation range and solidifying point). Numerous graphs are included, many of which are illegible.

ITEM 92: Distillation of Cold-Press Oil 1275

Distillation analysis of cold press oil is given.

ITEM 93: Suitability of Synthetic Wax for Oxidation to Fatty acids. 1278

This is a draft and final paper by Professor Martin concerning large scale production of fatty acids by the splitting of natural fats and oxidation of paraffins. It is an historical outline of the subject which describes generally which fatty acids are suitable for soaps and gives distillation analyses of synthetic paraffins to show how much of the paraffin is suitable for oxidation.

ITEM 94: Solubility of Paraffin Wax in Various Solvents 1318

The solubility of wax was determined in naphthas from the activated carbon plant (100-200°C.), from fractional distillation (150-200°C.), from diesel oil (200-300°C.) and from cracked gasoline (40-200°C.). Tables and graphs are given. It was found that the solubility does not depend on the kind of solvent as much as on the melting point of the paraffin and the temperature. A drawing of the apparatus used to determine solubility of solid materials in various solvents is shown.

ITEM 95: The Benzene-alcohol Method of Extracting Natural Wax 1327

This is a report of an inspection of a candle factory which obtains paraffin by distillation of low temperature tar. The paraffin contained approximately 10 to 15% of phenolic and asphaltic oil which was washed out at 0° C. by 5 to 7 volumes of a benzene alcohol mixture (4 parts of 90° benzene and 6 parts absolute alcohol). The temperature during operation should never exceed 5°C. in order to avoid paraffin losses. The flow sheet of the plant is included.

ITEM 96: Oven 15 Filling 8. 1330

These are similar tables to those in Reel 66 Item 9 for Oven 11. Oven 15 is filled with a low temperature iron catalyst.

ITEM 97: Oven 11 Filling 14 (Incomplete) 1337

These are numerous tables of yields and distillation analyses from operation with the low temperature iron catalyst at 9 atm. Much of the data is shown graphically.

END OF REEL