COMMERCIAL-SCALE DEMONSTRATION OF THE LIQUID PHASE METHANOL (LPMEOH™) PROCESS

FINAL REPORT (Volume 2: Project Performance and Economics) for the period 16 October 1992 – 30 June 2003

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ABSTRACT

This project, which was sponsored by the U.S. Department of Energy (DOE) under the Clean Coal Technology Program to demonstrate the production of methanol from coal-derived synthesis gas (syngas), has completed the 69-month operating phase of the program. The purpose of this Final Report for the "Commercial-Scale Demonstration of the Liquid Phase Methanol (LPMEOH[™]) Process" is to provide the public with details on the performance and economics of the technology. The LPMEOH[™] Demonstration Project was a \$213.7 million cooperative agreement between the DOE and Air Products Liquid Phase Conversion Company, L.P. (the Partnership). The DOE's cost share was \$92,708,370 with the remaining funds coming from the Partnership. The LPMEOH[™] demonstration unit is located at the Eastman Chemical Company (Eastman) chemicals-from-coal complex in Kingsport, Tennessee.

The technology was the product of a cooperative development effort by Air Products and Chemicals, Inc. (Air Products) and DOE in a program that started in 1981. Developed to enhance electric power generation using integrated gasification combined cycle (IGCC) technology, the LPMEOHTM Process is ideally suited for directly processing gases produced by modern coal gasifiers. Originally tested at the Alternative Fuels Development Unit (AFDU), a small, DOE-owned process development facility in LaPorte, Texas, the technology provides several improvements essential for the economic coproduction of methanol and electricity directly from gasified coal. This liquid phase process suspends fine catalyst particles in an inert liquid, forming a slurry. The slurry dissipates the heat of the chemical reaction away from the catalyst surface, protecting the catalyst, and allowing the methanol synthesis reaction to proceed at higher rates.

The LPMEOH[™] Demonstration Project accomplished the objectives set out in the Cooperative Agreement with DOE for this Clean Coal Technology project. Overall plant availability (defined as the percentage of time that the LPMEOH[™] demonstration unit was able to operate, with the exclusion of scheduled outages) was 97.5%, and the longest operating run without interruption of any kind was 94 days. Over 103.9 million gallons of methanol was produced; Eastman accepted all of the available methanol for use in the production of methyl acetate, and ultimately cellulose acetate and acetic acid.

The slurry bubble column reactor was successfully scaled from the 10 short tons-per-day (sT/D) LaPorte AFDU to the 260 sT/D or 80,000 gallons-per-day LPMEOH[™] demonstration unit. Over the first 4 years of operation, the catalyst deactivation rate averaged 0.6 to 0.7% per day, which was higher than the basis for the design of the LPMEOH[™] demonstration unit obtained from the LaPorte AFDU (0.4% per day). Engineering and laboratory efforts focused on the removal of trace contaminants in coal-derived synthesis gas (syngas), and the development of a procedure to activate a full charge of catalyst in the LPMEOH[™] reactor (in-situ activation) followed by temperature programming (the increase of reactor temperature as necessary to maintain the desired methanol production rate) could reduce the capital cost by as much as 10 percent. Over the last 6 months of operation, these improvements demonstrated the potential to reduce the design basis for catalyst deactivation rate by 50%. Because the impact of catalyst cost on the economics of methanol production is significant (reducing the rate of catalyst deactivation to 0.2% per day can realize a 25 percent reduction in variable cost), this

performance demonstrated the potential of the LPMEOHTM process to meet the targets for catalyst consumption utilizing coal-derived syngas. Over the demonstration period, tests were successfully performed at a variety of conditions anticipated in IGCC/ coproduction applications (catalyst withdrawal and addition to maintain the desired methanol production rate, operation on a carbon monoxide (CO)-rich syngas feed, ramping, and on-off operation).

Methanol from the LPMEOHTM Demonstration Project was also used to conduct product-use testing to demonstrate how a centrally located clean coal electric power plant with methanol coproduct could provide energy services to local communities. Product-use testing proved that coal-derived methanol, free of sulfur and other impurities, could be used one day as a replacement for petroleum in transportation, a peaking fuel in combustion turbines IGCC power plants, a hydrogen (H₂) source for small fuel cells, or as a chemical feedstock.

As a secondary objective to the LPMEOHTM Demonstration Project, design verification testing was completed to evaluate capability of the slurry bubble column reactor to produce dimethyl ether (DME) as a coproduct with methanol using the Liquid Phase Dimethyl Ether (LPDMETM) Process. The LPDMETM Process uses a physical mixture of a commercial methanol catalyst and a commercial dehydration catalyst in a single slurry reactor. During a 25-day design verification test at the LaPorte AFDU in 1999, the rate of catalyst deactivation for both the methanol synthesis and dehydration catalysts were slightly higher than the results for the LPMEOHTM Process from the first 3 weeks of the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89. Follow-up economic analysis showed that attractive economics for DME can likely be achieved at plant sizes approaching 1,100 sT/D. The program participants agreed that additional work on DME should be deferred so that all resources could be directed toward improving the catalyst performance for the LPMEOHTM Process during the remaining time within the operating program.

In evaluating the economics for the technology, studies showed that methanol from the LPMEOHTM Process can be coproduced with IGCC power at less than \$0.50 per gallon. If the improvement in catalyst performance from the last 6 months of operation can be extrapolated, the variable cost of production of methanol can be reduced by 25% over the current basis for future plants. Together, IGCC and LPMEOHTM technologies will be able to fill local needs for electric power, transportation fuels, and manufactured chemical products.

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TABLE OF CONTENTS

	Abstract	t i	ii
	Acknow	reagments	V
	Point of		V
			X.
		1gures XI	11
	List of A	AbbreviationsXiv	V
	List of U	Units	'1
	Glossary	y of 1 erms xvi	1
	Executiv	ve Summary XI	х
1.0	Introduc	tion 1-	1
	1.1	Purpose of the Project Performance and Economics Report 1-	1
	1.2	Overview of the Project	1
		1.2.1 Background and History of the Project 1-	1
		1.2.2 Project Organization 1-2	3
		1.2.3 Project Description 1-4	4
		1.2.4 Site 1-4	4
		1.2.5 Project Schedule	6
	1.3	Objectives of the Project 1-8	8
		1.3.1 Primary Objective	8
		1.3.2 Secondary Objective 1-8	8
	1.4	Significance of the Project 1-10	0
	1.5	DOE's Role in the Project	1
		1.5.1 Clean Coal Technology Program	1
		1.5.2 Management Plan 1-1	1
		1.5.2.1 DOE	2
		1.5.2.2 The Partnership 1-12	2
2.0	Technol	ogy Description 2-	1
	2.1	Description of the Demonstrated Technology	1
	2.2	Description of the Demonstration Facility	3
	2.3	Proprietary Information	3
	2.4	Simplified Process Flow Diagram.	4
	2.5	Stream Data 2-4	4
	2.6	Process and Instrumentation Diagrams 2-4	4
3.0	Undate	of the Public Design Report 3-	1
2.0	31	Design and Equipment Changes 3-	1
	5.1	3.1.1 Condensed Oil (29G-01) and Oil Makeun (29G-03) Pumps 3-	2
		3.1.2 Shurry Return Pump (29G-02) 3-	3
		3 1 3 Availability of Hydrogen Gas (H ₂ Gas) 3-	3
		3.14 Gas Snarger 3.4	4
	32	Demonstration Plant Equipment Cost Undate 3-2	4
	33	Demonstration Plant Operating Cost Undate 2-	5
4.0	Demons	tration Program 4	1
т.0	1 1	Test Plans 4	1 1
	4.1 1	Operating Procedures	1 2
	7.2	4-2.1 Instrumentation and Data Acquisition	2 Q
		4-6-4-6 4-6 4-6 4-6 4-6 4-6 4-6 4-6 4-6	0

TABLE OF CONTENTS (cont'd)

		4.2.2 Test Methods	4-8
	4.3	Analysis of Feedstocks, Products and Reagents	4-8
	4.4	Data Analysis Methodology	4-9
	4.5	Data Summary	4-9
	4.6	Operability and Reliability	4-10
	4.7	Section 4 Bibliography.	4-12
5.0	Techni	cal Performance	5-1
	5.1	Catalyst Performance	5-1
		5.1.1 Catalyst Performance: Catalyst Campaign 1	5-2
		5.1.2 Catalyst Performance: Catalyst Campaign 2	5-4
		5.1.3 Catalyst Performance: Catalyst Campaign 3	5-8
		5.1.4 Catalyst Performance: Catalyst Campaign 4	5-10
		5.1.5 Summary	5-11
	5.2	Identification and Removal of Trace Contaminants in Coal-Derived	5-13
		Syngas	
		5.2.1 Kingsport Catalyst Campaign 1	5-15
		5.2.2 Kingsport Catalyst Campaign 2	5-15
		5.2.3 Kingsport Catalyst Campaign 3	5-20
		5.2.4 Kingsport Catalyst Campaign 4	5-23
		5.2.5 Summary	5-24
	5.3	Gas Sparger Performance	5-25
		5.3.1 Initial Sparger Design (April – November 1997)	5-25
		5.3.2 Second Sparger Design (December 1997 – February 1999)	5-28
		5.3.3 Current Sparger Design (March 1999 – Present)	5-29
		5.3.4 Summary	5-32
	5.4	Process Variable Tests: H ₂ :CO Ratio	5-33
	5.5	Ramping and On/Off Testing	5-35
	5.6	In-Situ Activation and Temperature Programming	5-37
		5.6.1 Preparatory Engineering and Laboratory Work	5-37
		5.6.2 In-situ Activation Procedure and Results - August 2001	5-38
		5.6.3 Temperature Programming (August 2001 – March 2002)	5-40
		5.6.4 Second In-situ Activation Procedure and Results (June 2002)	5-40
		5.6.5 Summary	5-42
	5.7	Section 5 Bibliography	5-42
6.0	Enviro	nmental Performance	6-1
	6.1	Introduction	6-1
	6.2	Waste Streams and Their Disposal	6-1
		6.2.1 Spent Catalyst Slurry	6-1
		6.2.2 10C-30 Catalyst Guard Bed	6-3
		6.2.3 29C-40 Carbonyl Guard Bed	6-3
		6.2.4 Waste Oil	6-4
		6.2.5 Recovered Distillation Liquids	6-5
		6.2.6 Waste Water	6-6
	6.3	Potential Environmental Concerns	6-6
	6.4	Section 6 Bibliography	6-6

TABLE OF CONTENTS (cont'd)

7.0	Mathanal D	Product Use Testing	71
7.0	7 1 Inte	roduction	7 1
	7.1 IIII 7.2 On	Site Testing of Methanol	7 2
	7.2 Off	f Site Testing of Methanol	7-2 7 /
	7.5 01	-Site Testing of Methanol	75
	1.5	7.2.1.1 Elorida Instituta of Tachnology Pus & Light Vahialas	7-5
		7.3.1.1 Florida Institute of Technology Bus & Light Vehicles	7-5
		7.3.1.2 ARCADIS Gelagily & Miller Flexible Fuel Venicle	7-3
		7.5.1.5 West Virginia University Study of Stabilized Methanol	77
	7.2	In Transit Bus	/-/
	1.3	7.2.2.1 West Viscinia University Stationers Can Tasking	7-8
		7.3.2.1 West Virginia University Stationary Gas Turbine	/-8
		7.3.2.2 Aircraft Ground Support Equipment	/-10
		7.3.2.3 Low-NO _x Stationary Microturbine Combustor	/-12
	- 4	7.3.2.4 University of Florida Fuel Cell	/-14
	7.4 Su	mmary	7-15
0.0	7.5 Sec	ction / Bibliography	7-16
8.0	Liquid Pha	se Dimethyl Ether (LPDME TM) Design Verification Testing (DVT)	8-1
	8.1 Int	roduction	8-1
	8.2 DN	AE Market Studies	8-2
	8.3 DV	/T Economic Studies	8-2
	8.4 DV	/T Catalyst Studies	8-3
	8.5 199	99 LPDME [™] Process Design Verification Test Run	8-4
	8.6 DN	ME DVT Conclusions	8-8
	8.7 Sec	ction 8 Bibliography	8-10
9.0	Economics		9-1
	9.1 Int	roduction	9-1
	9.2 Est	timated Process Capital Costs	9-3
	9.3 Pro	bjected Operating and Maintenance Costs	9-5
	9.4 Su	mmary of Performance and Economics	9-6
	9.5 Eff	fect of Variables on Economics	9-6
	9.5	5.1 Plant Size	9-6
	9.5	5.2 Sulfur Content	9-7
	9.5	5.3 Inert Content	9-8
	9.5	5.4 Capacity Factor	9-9
	9.6 Sec	ction 9 Bibliography	9-10
10.0	Commercia	alization Potential and Plans	10-1
	10.1 Ma	arket Analysis	10-1
	10.	.1.1 Applicability of the Technology	10-1
	10.	1.2 Market Size and Barriers	10-3
	10.	1.3 Economics Comparison with Competing Technologies	10-4
	10.2 Co	mmercialization Plans.	10-5
	10.3 Sec	ction 10 Bibliography	10-6
11.0	Conclusion	ns and Recommendations	11-1
12.0	Other Proje	ect Reports and Publications	12-1

APPENDICES

- Appendix A
- Analytical Measurements for LPMEOH[™] Demonstration Unit Sample Material Balances from LPMEOH[™] Demonstration Unit Appendix B
- Appendix C Details on Data Analysis Methodology
- Appendix D Data Summary Tables

LIST OF TABLES

<u>Table</u>	<u>Title</u>	Page
2 5-1	Heat and Material Balance Summary	2-6
3.2-1	Kingsport LPMEOH TM Demonstration Unit - Update of	3-5
	Equipment Cost Summary	
3.3-1	Kingsport LPMEOH [™] Demonstration Unit - Update of	3-6
	Operating Costs	
4.1-1	Test Conditions at LPMEOH [™] Demonstration Unit	4-3
4.6-1	LPMEOH [™] Demonstration Unit Availability and Causes of	4-10
	Unplanned Outages	
5.1-1	Kingsport LPMEOH [™] Operating Program Campaigns	5-1
5.2.1-1	Kingsport Catalyst Campaign 1 – Elemental Composition of	5-15
	Solids from Slurry Samples	
5.2.2-1	Elemental Analysis Results for Spent Activated Carbon from	5-17
	29C-40 Carbonyl Guard Bed (June 1999)	
5.2.2-2	Analysis of Fresh and Spent Manganese Dioxide from 29C-40	5-18
5 9 9 9	Carbonyl Guard Bed (11 August 1999)	5 10
5.2.2-3	Analysis of Fresh and Spent Activated Carbon from 29C-40	5-18
5 2 2 1	Carbonyl Guard Bed (11 August 1999)	c 00
5.2.3-1	Analysis of Fresh and Spent Cu/C Adsorbent from 29C-40	5-22
5 2 4 1	Carbonyl Guard Bed (20 February 2002)	5 22
5.2.4-1	Solida from Shurry Somplag	5-25
5 1 1	Solids from Slurry Samples	5 24
5.4-1 6 2 1	Spont Catalyst Slurry from L DMEOHTM Demonstration Unit	5-54
6.2-1	10C 20 Catalyst Guard Red Adapteents Used during	6.2
0.2-2	Operation of L DMEOHTM Demonstration Unit	0-3
623	20C 40 Carbonyl Guard Bed Adsorbents Used during	6.4
0.2-3	Operation of LPMFOH TM Demonstration Unit	0-+
6 2-4	Waste Oil from LPMEOH TM Demonstration Unit	6-5
6.2-5	Oil Collected in Distillation Equipment from LPMEOH TM	6-5
0.2 5	Process	0.5
7 2-1	Refined-Grade Methanol from LPMEOH TM Demonstration	7-3
, . _ 1	Unit - Ouality and Fitness for Use (1997)	10
7.3-1	Stabilized Methanol Composition - CO-rich Feed Gas	7-4
	$(H_2:CO = 0.7)$	
7.3.1-1	ARCADIS Geraghty & Miller FFV Fuel Economy Results	7-6
7.3.1-2	ARCADIS Geraghty & Miller FFV Emissions Results	7-6
7.3.1-3	West Virginia University - Tri-Boro Coach Emissions	7-8
	Results	
8.6-1	LPDME TM Goals and Performance Results (Laboratory and	8-9
	AFDU)	
9.2-1	Estimated Facility Investment Summary for Once-Through	9-4
	500 sT/D LPMEOH™ Plant	
9.3-1	Projected Operating and Maintenance Costs for Once-Through	9-5
	500 sT/D LPMEOH [™] Plant	
10.1.2-1	Market Forecast for Methanol	10-3
10.1.3-1	Baseload Coproduction, 500 sT/D, Fuel Grade Methanol	10-5

LIST OF TABLES (cont'd)

Title	Page
Summary of Streams to be Analyzed for Material Balance Calculations	A-2
Summary of Data for Typical Material and Energy Balance and Reactor Performance Assessment	C-2
Data Summary for Kingsport Catalyst Campaign 1	D-2
Data Summary for Kingsport Catalyst Campaign 2	D-5
Data Summary for Kingsport Catalyst Campaign 3	D-26
Data Summary for Kingsport Catalyst Campaign 4	D-30
	TitleSummary of Streams to be Analyzed for Material Balance CalculationsSummary of Data for Typical Material and Energy Balance and Reactor Performance AssessmentData Summary for Kingsport Catalyst Campaign 1 Data Summary for Kingsport Catalyst Campaign 2

LIST OF FIGURES

Figure	<u>Title</u>	Page
1.2.2-1	LPMEOH [™] Demonstration Project Organization	1-3
1.2.3-1	Integration of Existing Facilities with LPMEOH [™] Demonstration Unit	1-5
1.2.4-1	Photograph of Eastman Chemical Company's Chemicals-from-Coal Complex	1-6
1 2 5-1	LPMEOH TM Demonstration Project Schedule	1-9
1.5.2-1	Kingsport LPMEOH TM Demonstration Project Operating Phase Organization Chart	1-13
2 1-1	Gas-Phase Methanol Reactors	2-2
2.1-2	LPMEOH TM Reactor Schematic	2-2
2.4-1	LPMEOH TM Demonstration Unit Simplified Process Flow Diagram	2-5
4.1-1	LPMEOH [™] Demonstration Test Plan Schedule of Tests	4-4
5.1.1-1	Kingsport LPMEOH [™] Catalyst Life (Catalyst Campaign 1)	5-2
5.1.2-1	Kingsport LPMEOH [™] Catalyst Life (Catalyst Campaign 2: 1997-1998)	5-4
5.1.2-2	Kingsport LPMEOH [™] Catalyst Life (Catalyst Campaign 2: 1999)	5-6
5.1.2-3	Kingsport LPMEOH [™] Catalyst Life (Catalyst Campaign 2: 2000-2001)	5-7
5.1.3-1	Kingsport LPMEOH [™] Catalyst Life (Catalyst Campaign 3)	5-9
5.1.4-1	Kingsport LPMEOH [™] Catalyst Life (Catalyst Campaign 4)	5-10
5 2-1	Photograph of Alternative Fuels Field Test Unit (AFFTU)	5-14
5 2 2-1	Kingsport Catalyst Campaign 2 - Trace Contaminant	5-16
0.2.2 1	Concentrations on Slurry Samples	0 10
5.2.2-2	Effect of Arsine Concentration on Methanol Catalyst	5-17
	Deactivation Rate in Autoclave	
5.2.3-1	Kingsport Catalyst Campaign 3 - Trace Contaminant Concentrations on Slurry Samples	5-20
5.3.1-1	Sparger Resistance Coefficient – Sparger Design #1 (April - November 1997)	5-26
5.3.2-1	Sparger Resistance Coefficient – Sparger Design #2 (December 1997 - February 1999)	5-29
5.3.3-1	Sparger Resistance Coefficient – Sparger Design #3 (March 1999 - August 2001)	5-30
5.3.3-2	Sparger Resistance Coefficient – Sparger Design #3 (August 2001 - June 2002)	5-31
5.3.3-3	Sparger Resistance Coefficient – Sparger Design #3 (June 2002 - 31 December 2002)	5-32
5.5-1	Ramping Study at LPMEOH [™] Demonstration Unit (13 June 2000)	5-36
5.5-2	Ramping Study at LPMEOH [™] Demonstration Unit (13 March 2001)	5-36
5.6-1	Reduction Gas Úptake vs. Temperature – In-Situ Catalyst Activation (22-24 August 2001)	5-39

LIST OF FIGURES (cont'd)

<u>Figure</u>	<u>Title</u>	Page
5.6-2	Reduction Gas Uptake vs. Temperature – In-Situ Catalyst Activation (25-27 June 2002)	5-41
7.1-1	Once-through Methanol Coproduction with IGCC Electric Power	7-1
7.3.1-1	West Virginia University Transportable Heavy Duty Emissions Testing Laboratory	7-7
7.3.2-1	West Virginia University Stationary Gas Turbine Emissions Test Results – 1998 Testing	7-9
7.3.2-2	West Virginia University Stationary Gas Turbine Emissions Test Results – 2000 Testing	7-10
7.3.2-3	Aircraft Ground Support Equipment Average NO _x Emissions Reduction with WIF vs. JP-8	7-12
7.3.2-4	Low-NO _x Microturbine NO _x Emissions with Stabilized Methanol vs. Natural Gas	7-14
8.5-1	LaPorte Alternative Fuels Development Unit (AFDU)	8-5
8.5-2	1999 LPDME [™] Operating Run at LaPorte AFDU – Estimated Catalyst Activity	8-6
8.5-3	1999 LPDME [™] Operating Run at LaPorte AFDU – Production Results	8-7
8.6-1	LPDME TM Process Economics	8-10
9.5-1	Effect of Methanol Plant Size on Methanol Conversion Cost	9-7
9.5-2	Effect of Syngas Conversion on Methanol Conversion Cost	9-8
9.5-3	Effect of Inerts Concentration on Methanol Conversion Cost	9-9
10.1-1	LPMEOH [™] Process Flow Diagram – Design Options for IGCC Applications	10-2
A-1	Syngas Sampling Apparatus	A-8

LIST OF ABBREVIATIONS

AFB	-	Air Force Base
AFDU	-	Alternative Fuels Development Unit
AFFTU	-	Alternative Fuels Field Test Unit
AGSE	-	Aircraft Ground Support Equipment
Al/P	-	Aluminum to Phosphate
Air Products	-	Air Products and Chemicals, Inc.
APCI	-	Air Products and Chemicals, Inc.
AsH ₃	-	Arsine
CO	-	Carbon Monoxide
CO_2	-	Carbon Dioxide
COS	-	Carbonyl Sulfide
COTR	-	Contracting Officer's Technical Representative
Cu/C	-	Copper-impregnated Activated Carbon (following reduction of copper oxide)
CuO/C	-	Copper Oxide-impregnated Activated Carbon
DME	-	Dimethyl Ether
DOE	-	United States Department of Energy
DVT	-	Design Verification Testing
EA	-	Environmental Assessment
Eastman	-	Eastman Chemical Company
EIV	-	Environmental Information Volume
EMP	-	Environmental Monitoring Plan
EPA	-	United States Environmental Protection Agency
FFV	-	Flexible-Fuel Vehicle
FONSI	-	Finding of No Significant Impact
H_2	-	Hydrogen
H_2S	-	Hydrogen Sulfide
IGCC	-	Integrated Gasification Combined Cycle
ISO	-	International Organization for Standardization
LHV	-	Lower Heating Value
LPMEOH TM	-	Liquid Phase Methanol
LPDMEtm	-	Liquid Phase Dimethyl Ether
LPG	-	Liquefied Petroleum Gas
MeOH	-	Methanol
MTBE	-	Methyl Tertiary Butyl Ether
N_2	-	Nitrogen
NEPA	-	National Environmental Policy Act
NO _x	-	Nitrogen Oxides
NPDES	-	National Pollutant Discharge Elimination System
O_2	-	Oxygen
Partnership	-	Air Products Liquid Phase Conversion Company, L.P.
PFD	-	Process Flow Diagram(s)
PON	-	Program Opportunity Notice
Project	-	Production of Methanol/DME Using the LPMEOH [™] Process at an
5		Integrated Coal Gasification Facility
P&ID(s)	-	Piping and Instrumentation Diagram(s)
RCRA	-	Resource and Conservation Recovery Act
ROI	-	Return on Investment
SO_2	-	Sulfur Dioxide
Syngas	-	Synthesis Gas
TCLP	-	Toxicity Characteristic Leaching Procedure
TPD or sT/D	-	Short Ton(s) per Day (2,000 lbs/ton)
UHC	-	Unburned Hydrocarbons

LIST OF ABBREVIATIONS (cont'd)

S

LIST OF UNITS

Btu	-	British Thermal Units
Btu/hr	-	British Thermal Units per hour
Btu/lb	-	British Thermal Units per pound
Btu/SCF	-	British Thermal Units per standard cubic foot
°C	-	Degrees Centigrade
Ft	-	Foot
°F	-	Degrees Fahrenheit
g	-	Grams
Gal or gal	-	Gallons
gmol	-	Gram-mole
gpm	-	Gallons per minute
Нр	-	Horsepower
hr	-	Hour
KCAL	-	Kilocalories
KSCFH	-	Acronym for 1,000 standard cubic feet per hour; for this term, standard
		conditions are 14.7 psia and 60°F
kWh	-	Kilowatt-hour
1	-	Liter
lb	-	Pound
lb mole	-	Pound mole
lbs	-	Pounds
mg	-	Milligrams
mpg	-	Miles per gallon
MM or mm	-	Million
ppbv	-	Parts per billion (volume basis)
ppm	-	Parts per million
ppmw	-	Parts per million (weight basis)
psi	-	Pounds per square inch
psia	-	Pounds per square inch (Absolute)
psig	-	Pounds per square inch (Gauge)
SCF	-	Standard cubic feet (60°F, 14.7 psia)
SCFM	-	Standard cubic feet per minute
SCFH	-	Standard cubic feet per hour
SCFD	-	Standard cubic feet per day
Sl/hr-kg	-	Standard Liter(s) per Hour per Kilogram of Catalyst (oxide basis)
x'-y"	-	x feet - y inches

GLOSSARY OF TERMS

Availability	-	The percentage of time that the LPMEOH [™] demonstration unit was able to operate, with the exclusion of scheduled outages
Balanced Gas	-	A syngas with a composition of hydrogen (H ₂), carbon monoxide (CO), and carbon dioxide (CO ₂) in stoichiometric balance ($[H_2 - CO_2]/[CO + CO_2] = 2.05$) for the meduation of methanol
Carbon Monorid	a Car	A summer containing minority contain menorial (CO), also called CO Cos
Catalyst Activity		the rate at which the catalyst promotes the desired chemical reaction to proceed within the limitations of chemical equilibrium
Catalyst Age (η -	-eta) -	The ratio of the rate constant at any point in time to the rate constant for a freshly reduced catalyst (as determined in the laboratory autoclave)
Catalyst Concent	ration -	Synonym for Slurry Concentration
Catalyst Loading	-	Synonym for Slurry Concentration
CO Conversion	-	The percentage of CO consumed across the reactor
Crude-Grade M	ethanol	- Underflow from rectifier column (29C-20), defined as 80 wt% minimum purity;
		requires further distillation in existing Eastman equipment prior to use
DOE-NETL	-	The DOE's National Energy Technology Laboratory (Project Team)
DOE-HQ	-	The DOE's Headquarters - Coal Fuels and Industrial Systems (Project Team)
Fresh Feed	-	Sum of Balanced Gas, H ₂ Gas, and CO Gas
Gas Holdup	-	The percentage of reactor volume up to the Gassed Slurry Height which is gas
Gassed Slurry		
Height	-	Height of gassed slurry in the reactor
Hydrogen Gas	-	A syngas containing an excess of hydrogen (H_2) over the stoichiometric balance for
, C		the production of methanol; also called H_2 Gas
IGCC/OTM	-	An IGCC plant with a "Once-Thru Methanol" plant (the LPMEOH [™] Process) added-on
Inlet Superficial		
Velocity	-	The ratio of the actual cubic feet of gas at the reactor inlet (calculated at the reactor temperature and pressure) to the reactor cross-sectional area (excluding the area contribution by the internal heat evaluation) tuning units are feet are second.
V		by the internal near exchanger); typical units are feet per second
N M05	-	A full bland of 85 volume percent methodel and 15 volume percent unloaded geopline
Mathemal Dradua	-	The gram males of methanol produced per hour per bilegram establist (on an avide basis)
	uvity -	Melagular weight, nound nor nound mela
	-	Density, nounde non subjected
μ Drainat	-	Density, pounds per cubic 1001 Denduction of Mathemal/DME Using the LDMEOUTM Decoses at an Integrated Cool
Project	-	Gasification Facility
Raw Methanol	-	The sum of Refined-Grade Methanol and Crude-Grade Methanol; represents total methanol which is produced after stabilization
Reactor Feed	-	The sum of Fresh Feed and Recycle Gas
Reactor O-T-M		
Conversion	-	Percentage of energy (on a lower heating value basis) in the Reactor Feed converted to methanol (Once-Through-Methanol basis)
Reactor Volumet	ric	
Productivity	-	The quantity of Raw Methanol produced (tons per day) per cubic foot of reactor volume up to the Gassed Slurry Level
Recycle Gas	-	The portion of unreacted syngas effluent from the reactor "recycled" as a feed gas
Refined-Grade M	lethanol -	- Distilled methanol, defined as 99.8 wt% minimum purity; used directly in downstream Eastman processes
Slurry Concentra	tion -	Percentage of weight of slurry (solid plus liquid) which is catalyst (on an oxide basis)
Syngas Utilizatio	on -	Defined as the number of standard cubic feet of Balanced Gas plus CO Gas to the LPMEOH [™] demonstration unit required to produce one pound of Raw Methanol

GLOSSARY OF TERMS (cont'd)

Synthesis Gas	-	A gas containing primarily H_2 and CO, or mixtures of H_2 and CO; intended for "synthesis" in a reactor to form methanol and/or other hydrocarbons (synthesis gas may also contain CO_2 , water, and other gases)
Temperature		
Programming	-	The increase of reactor temperature as necessary to control the reactor purge flowrate and maintain reactor volumetric productivity
Tie-in(s) V	-	The interconnection(s) between the LPMEOH [™] demonstration unit and the Eastman Facility Volumetric flowrate, thousand standard cubic feet per hour

EXECUTIVE SUMMARY

This project, which was sponsored by the U.S. Department of Energy (DOE) under the Clean Coal Technology Program to demonstrate the production of methanol from coal-derived synthesis gas (syngas), has completed the 69-month operating phase of the program. The purpose of this Final Report for the "Commercial-Scale Demonstration of the Liquid Phase Methanol (LPMEOH[™]) Process" is to provide the public with details on the performance and economics of the technology. The LPMEOH[™] Demonstration Project was a \$213.7 million cooperative agreement between the DOE and Air Products Liquid Phase Conversion Company, L.P. (the Partnership). The DOE's cost share was \$92,708,370 with the remaining funds coming from the Partnership. The LPMEOH[™] demonstration unit is located at the Eastman Chemical Company (Eastman) chemicals-from-coal complex in Kingsport, Tennessee.

The technology was the product of a cooperative development effort by Air Products and Chemicals, Inc. (Air Products) and DOE in a program that started in 1981. Developed to enhance electric power generation using integrated gasification combined cycle (IGCC) technology, the LPMEOHTM Process is ideally suited for directly processing gases produced by modern coal gasifiers. Originally tested at the Alternative Fuels Development Unit (AFDU), a small, DOE-owned process development facility in LaPorte, Texas, the technology provides several improvements essential for the economic coproduction of methanol and electricity directly from gasified coal. This liquid phase process suspends fine catalyst particles in an inert liquid, forming a slurry. The slurry dissipates the heat of the chemical reaction away from the catalyst surface, protecting the catalyst, and allowing the methanol synthesis reaction to proceed at higher rates.

The LPMEOHTM Demonstration Project accomplished the objectives set out in the Cooperative Agreement with DOE for this Clean Coal Technology project. Methanol operation began on 02 April 1997, and startup was completed in two days. Nameplate production of 80,000 gallons per day (260 short tons per day [sT/D]) was achieved within four days, and production rates exceeding 115% of design were achieved within six days. Overall plant availability (defined as the percentage of time that the LPMEOHTM demonstration unit was able to operate, with the exclusion of scheduled outages) was 97.5%, and the longest operating period without interruption of any kind was 94 days. Over 103.9 million gallons of methanol was produced, including a one-month maximum of 2.5 million gallons. Eastman accepted all of the available methanol for use in the production of methyl acetate, and ultimately cellulose acetate and acetic acid.

The slurry bubble column reactor was successfully scaled from the 10 sT/D LaPorte AFDU to the 260 sT/D LPMEOHTM demonstration unit. The LPMEOHTM reactor did not exhibit any hydrodynamic instability, and the maximum catalyst loading of 151% of the design value of about 40,000 pounds (corresponding to a slurry concentration of greater than 49 weight % [wt%]) was achieved. The gas sparger at the inlet to the LPMEOHTM reactor was replaced with two modified units (using the same design principles as the original device), and stable performance was realized when a continuous flush of condensed process oil (in this case, by gravity) was utilized.

Catalyst performance (as determined by the rate of catalyst deactivation using an in-house kinetic model) improved during the demonstration test period. Initially, deactivation rates of greater than 1% per day were calculated at the design operating temperature of 250°C. As a basis of comparison, the calculated deactivation rate from the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89 was 0.4% per day. (This run was performed on carbon monoxide [CO]-rich syngas derived from natural gas.) The high deactivation rates were attributed to the presence of post-construction debris in the reactor system. During the second catalyst campaign that began in December of 1997, the rate of catalyst deactivation was calculated to be 0.6 to 0.7% per day at a reactor temperature of 235°C. The practice of catalyst withdrawal and addition (removing a portion of the spent catalyst from the LPMEOHTM reactor and adding fresh catalyst in 2,000 pound batches) to maintain the desired methanol production rate was successfully demonstrated.

In determining the reasons for the higher rate of catalyst deactivation, the presence of certain elements (in particular, arsenic and sulfur) was identified from samples of spent catalyst, spent adsorbent materials from catalyst guard beds, and syngas entering the LPMEOHTM demonstration unit. A commercially available copper oxide-impregnated activated carbon was identified as having capacity to remove arsenic (in the form of arsine) within an existing catalyst guard bed that was originally designed for removal of metal carbonyls at ambient temperature.

In parallel, laboratory testing was performed to develop a procedure to activate (or reduce) a full charge of methanol synthesis catalyst within the LPMEOH[™] reactor (so-called in-situ activation). The in-situ activation procedure prepared fresh methanol synthesis catalyst for production of methanol by reacting metal oxides (for example, copper oxide) with a reductant such as CO or hydrogen (H_2) to produce the base metal plus either carbon dioxide (CO_2) or water (H₂O). During the operating program, this feature was identified as having significant commercial interest, as capital cost savings of about 10% could be realized by elimination of the separate processing equipment for the catalyst withdrawal/addition procedure. Also, operating the LPMEOHTM reactor at low temperatures at the outset and slowly increasing temperature with time to maintain a constant production of methanol (called temperature programming) was believed to have the potential to extend the life of the methanol synthesis catalyst. Following the first in-situ activation of methanol synthesis catalyst in August of 2001 (during which initial performance of the catalyst was about 70% of the value expected from the laboratory testing), the concept of temperature programming was demonstrated (temperature was gradually raised from 216°C to 250°C in order to maintain the desired methanol production rate). Following changes to the storage conditions of the catalyst slurry prior to reduction, the in-situ activation procedure was successfully practiced in June of 2002, resulting in the catalyst meeting the initial performance targets.

During the final six months of operation under the demonstration test plan, the average rate of catalyst deactivation was calculated to be 0.17% per day, which was ½ of the design basis for the LPMEOHTM demonstration unit. The reactor was maintained at a constant low temperature of 215°C during this period, and the adsorbent used in the catalyst guard bed within the LPMEOHTM demonstration unit (copper oxide-impregnated activated carbon) was replaced with fresh material after 3 months of operation (including a novel thermal treatment that was applied after two months to extend the useful life of the adsorbent). Because the impact of catalyst cost

on the economics of methanol production is significant (reducing the rate of catalyst deactivation by up to ½ of the current basis can realize a 25 percent reduction in variable cost), this performance demonstrated the potential of the LPMEOHTM process to meet the targets for catalyst consumption utilizing coal-derived syngas.

Although the host site for the LPMEOHTM demonstration unit was a chemical production facility, testing was able to be performed to simulate the performance of the LPMEOHTM Process in an IGCC power plant. A total of 78 days of testing at a variety of H₂:CO ratios were performed to evaluate the effect of syngas compositions expected from coal-based IGCC units on the performance of the LPMEOHTM Process. All of the results met the expectations for methanol production rate based upon earlier studies in the laboratory. Catalyst performance for the LPMEOHTM reactor with lower H₂:CO ratios were not significantly impacted based on the calculated rate of catalyst deactivation. On/off and ramping studies were also completed, and the LPMEOHTM demonstration unit was capable of meeting the ramping requirements for an IGCC system by achieving the goal of a minimum 5% change in design flow per minute. Tests of the ability of the LPMEOHTM reactor to operate in a load-following environment were also successful.

Stabilized (degassed) methanol produced from CO-rich syngas in the LPMEOH[™] Process can have less than 1 wt% H₂O, and has the potential to be used directly in transportation and power generation applications. The cost of distillation from stabilized methanol to chemical-grade purity has been estimated to be \$0.02 to \$0.04 per gallon. Stabilized methanol from the LPMEOHTM Demonstration Project was made available to seven test locations to study the feasibility of its use these systems. Successful demonstration of the use of stabilized methanol in these applications can enhance the flexibility of and revenue from IGCC plants. In bus and flexible fuel vehicle (FFV) trials, stabilized methanol provided the same environmental benefits as chemical-grade methanol with no penalty on performance or fuel economy. Tests in a gas turbine and a diesel generator showed that levels of nitrogen oxides (NO_x) in the exhaust air were lowered when stabilized methanol or methanol emulsions are used instead of conventional oil fuels. As with chemical-grade methanol, lubrication additives will likely be required when stabilized methanol is fed to a gas turbine. Testing of stabilized methanol as the source of H₂ to a phosphoric acid fuel cell has shown that conventional steam reforming catalysts are not compatible with the trace mineral oil present in the stabilized methanol; when auto-thermal reforming or partial oxidation reforming are used, the stabilized methanol would possibly be acceptable as a fuel. Methanol from the LPMEOH[™] Process that is purified to chemical-grade specifications should be suitable for use in this application. Product-use testing proved that coalderived methanol, free of sulfur and other impurities, could be used one day as a replacement for petroleum in transportation, a peaking fuel in combustion turbines IGCC power plants, a source of H₂ for small fuel cells, or as a chemical feedstock.

As a secondary objective to the LPMEOHTM Demonstration Project, design verification testing was completed to evaluate capability of the slurry bubble column reactor to produce dimethyl ether (DME) as a coproduct with methanol using the Liquid Phase Dimethyl Ether (LPDMETM) Process. The LPDMETM Process uses a physical mixture of a commercial methanol catalyst and a commercial dehydration catalyst in a single slurry reactor. In addition to its current uses, DME has potential applications as a chemical building block. Also, since the physical properties of

DME are similar to those of liquefied petroleum gas (LPG), it has been speculated that DME could be used in large-scale power production, in home heating, in replacement of LPG for automobiles, and as a diesel fuel substitute or combustion supplement. Economic targets on the production cost of DME from the LPDMETM Process were established, and a 25-day design verification test was performed at the LaPorte AFDU in 1999. The rate of catalyst deactivation for both the methanol synthesis and dehydration catalysts was calculated to be 0.7% per day, which was a large improvement over the 4% per day rate of deactivation which was calculated from autoclave studies prior to the initial test of the LPDMETM Process at the LaPorte AFDU in 1991; these results were slightly higher than the results (0.5% per day) for the LPMEOHTM Process from the first 3 weeks of the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89. Follow-up economic analysis showed that the target DME cost of \$7 to \$8 per MMBtu (equivalent to costs for LPG in China) can likely be achieved at plant sizes approaching 1,100 sT/D. The program participants agreed that additional work on DME should be deferred so that all resources could be directed toward improving the catalyst performance for the LPMEOHTM Process during the remaining time within the operating program.

In evaluating the economics for the technology, parametric studies were also performed to determine the production cost of methanol from the LPMEOHTM Process in an IGCC power plant. The process can be designed to operate in a continuous, baseload manner, converting syngas from oversized gasifiers or from a spare gasifier. The process can also be designed to operate only during periods of off-peak electric power demand to consume a portion of the excess syngas and allow the electricity output from the combined-cycle power unit to be reduced. In this latter circumstance, the gasifiers continue to operate at full baseload capacity, so the IGCC facility's major capital asset is fully utilized. In either baseload or cycling operation, partial conversion of between 20% and 40% of the volume of H₂ and CO in the IGCC power plant's syngas is optimal on an economic basis, and conversion of up to 50% is feasible. Methanol conversion costs of about \$0.15 per gallon were determined for a coal-based IGCC plant using the once-through configuration; when the cost of syngas is added, the price of methanol is less than \$0.50 per gallon. As the economics for IGCC systems improve, the economics for coproduction of methanol will continue to look attractive in local markets when compared with methanol that is produced in off-shore locations from natural gas.

1.0 INTRODUCTION

1.1 PURPOSE OF THE PROJECT PERFORMANCE AND ECONOMICS REPORT

The purpose of the Project Performance and Economics Report for the "Commercial-Scale Demonstration of the Liquid Phase Methanol (LPMEOHTM) Process" project is to consolidate for public use all relevant nonproprietary information on the project, other than that already included in the Public Design Report (Final Report - Volume 1).

The scope of the report is limited to nonproprietary information. Therefore, although its content is insufficient to provide a complete tool for designing a LPMEOH[™] unit, it will serve as a reference for the design considerations involved in developing a commercial-scale facility.

1.2 OVERVIEW OF THE PROJECT

1.2.1 Background and History of the Project

The purpose of this project was to demonstrate the commercial viability of the LPMEOH[™] Process using coal-derived synthesis gas (syngas). This project was conducted pursuant to the U.S. Department of Energy (DOE) Clean Coal Technology Program.

The United States needs future sources of alternative liquid fuels. With domestic oil production declining and imports increasing, the potential of producing affordable liquid fuels from non-petroleum sources could one day prove both strategically and economically important. The LPMEOHTM Process offers an extremely attractive route to supplementing our liquid fuel supplies with methanol made from the abundant coal reserves of the United States.

Methanol also has a broad range of commercial applications. It can be substituted for or blended with gasoline to power vehicles. It is an excellent fuel for the rapid-start combustion turbines used by utilities to meet peak electricity demands. It contains no sulfur and has exceptionally low nitrogen oxide (NO_x) characteristics when burned. Methanol may be an economical source of hydrogen (H_2) for fuel cells. It can also be used as a chemical feedstock.

Air Products and Chemicals, Inc. (Air Products) and Eastman Chemical Company (Eastman) entered into a partnership known as Air Products Liquid Phase Conversion Company, L.P. (the Partnership). The Partnership and the DOE participated in the Clean Coal Technology Program demonstration of LPMEOH[™] technology. The Partnership designed, built, owned, and operated a nominal 80,000 gallons-per-day (260 short tons-per-day [sT/D]) LPMEOH[™] demonstration unit at Eastman's chemicals-from-coal complex in Kingsport, Tennessee. The program objectives were to demonstrate the scaleup and operability of the LPMEOH[™] Process under various coal-based syngas feed compositions and to gain operating experience for future syngas conversion projects.

The technology is the product of a cooperative development effort by Air Products and DOE in a program that started in 1981. Developed to enhance electric power generation using integrated gasification combined cycle (IGCC) technology, the LPMEOH[™] Process is ideally suited for

directly processing gases produced by modern day coal gasifiers. Originally tested at the Alternative Fuels Development Unit (AFDU), a small, DOE-owned experimental unit in LaPorte, Texas, the technology offers significant potential, compared to conventional gas-phase methanol production technologies, to reduce electric power generation costs at coal-based IGCC facilities with the coproduction of chemical feedstocks and alternative liquid fuels. The domestically developed LPMEOHTM technology can use domestic coal reserves to produce clean, storable, liquid fuels and chemical feedstocks. Eventual commercialization of the LPMEOHTM Process would provide chemical feedstock and electric power cost savings, lower sulfur dioxide (SO₂) and NO_x emissions, and could reduce the use of imported liquid fuels.

The LPMEOHTM technology could someday be used as an adjunct to an IGCC power plant—one of the cleanest and most efficient of the 21st century power generating options. When the IGCC power plant is not generating electricity at its full capacity, excess syngas can be diverted to make methanol in a fast ramping LPMEOHTM plant. The methanol could be stored on-site and used in peaking turbines or sold as a commercial fuel or a chemical feedstock. In this configuration, the cost of making methanol from coal is likely to be competitive with stand alone natural gas-to-methanol facilities. A discussion of the process options and economics is provided in Section 9 of this report.

The LPMEOH[™] demonstration unit was integrated with Eastman's chemicals-from-coal complex, accepting syngas and converting it to methanol, for use as a chemical feedstock within the Eastman facility. A portion of the methanol product was evaluated as a zero sulfur, low-NO_x combustion fuel for testing in stationary power applications and mobile transportation use (buses, flexible-fuel vehicles [FFV's], etc.).

These fuel-use tests provided a basis for the comparison of the product methanol with conventionally accepted fuels with regard to emission levels and economic viability. The program goal of demonstrating methanol as a fuel can lead to the potential for greater use of oxygenated fuels, which burn cleaner than conventional fuels, thereby reducing air emissions from mobile (e.g., buses and vanpools) and stationary (e.g., engines, turbines, and fuel cells) sources.

The DOE, under the Clean Coal Technology Program, provided cost-shared financial assistance for the design, construction, and operation of the commercial-scale LPMEOHTM demonstration unit by The Partnership. Air Products designed and constructed the LPMEOHTM demonstration unit and Eastman operated it. The demonstration unit is a nominal 260 ton-per-day (or 80,000 gallon per day) unit situated on a 0.6 acre plot within the existing Eastman chemicals-from-coal complex in Kingsport, Tennessee.

The Eastman coal gasification facility has operated commercially since 1983. Eastman currently both produces and purchases methanol for use at the site. At this site, it was possible to perform studies that demonstrate the unique load-following flexibility of the LPMEOH[™] demonstration unit for application to coal-based IGCC electric power generation facilities.

This project has also completed design verification testing (DVT), including laboratory- and pilot-scale research and market verification studies, to evaluate whether to include a

demonstration of the production of dimethyl ether (DME) as a mixed coproduct with methanol. DME has several commercial uses. In a storable blend with methanol, the mixture can be used as a peaking fuel in gasification-based electric power generating facilities, or as a diesel engine fuel. Blends of methanol and DME can be used as chemical feedstocks for synthesizing chemicals, including new oxygenated fuel additives.

1.2.2 Project Organization

As noted above, Air Products and Eastman formed the Partnership to execute the LPMEOHTM Demonstration Project. The Partnership subcontracted with Air Products to provide the overall management of the project, and to act as the primary interface with DOE. As subcontractor to the Partnership, Air Products provided the engineering design, procurement, construction, and commissioning of the LPMEOHTM demonstration unit, and provided the technical and engineering supervision needed to conduct the operational testing program required as part of the project. As subcontractor to Air Products, Eastman was responsible for operation of the LPMEOHTM demonstration unit, and for the interconnection and supply of syngas, utilities, product storage, and other needed services. Air Products also issued subcontracts for and managed the projects within the off-site testing of methanol from the LPMEOHTM Process. These roles and responsibilities are summarized in Figure 1.2.2-1.





1.2.3 Project Description

Air Products developed the two technologies that were demonstrated at various engineering scales during this demonstration project with the financial support of the DOE. The LPMEOHTM Process uses a slurry bubble column reactor to convert coal-derived syngas to methanol. Because of its superior heat management, the process can directly utilize the carbon monoxide (CO)-rich syngas characteristic of the gasification of coal, petroleum coke, residual oil, wastes, or other hydrocarbon feedstock. When added to a high-efficiency IGCC power plant, the LPMEOHTM Process converts a portion of the CO-rich syngas produced by the gasifier to methanol, and the unconverted gas is used to fuel the gas turbine combined-cycle power plant. The LPMEOHTM Process can also be applied to the conversion of syngas derived from natural gas. A block diagram showing the integration of the LPMEOHTM demonstration unit within the Eastman chemicals-from-coal complex is provided in Figure 1.2.3-1.

The Liquid Phase Dimethyl Ether (LPDMETM) Process uses a physical mixture of a commercial methanol catalyst and a commercial dehydration catalyst in a single slurry reactor to coproduce DME with methanol. This process provides high syngas conversion, efficient heat transfer, and directly converts a variety of syngas compositions.

1.2.4 Site

The demonstration unit, which occupies an area of 0.6 acre, is integrated into the existing 4,000acre Eastman chemicals-from-complex located in Kingsport, Tennessee. The Eastman complex employs approximately 8,600 people. In 1983, Eastman constructed a coal gasification facility utilizing Texaco technology. The syngas generated by this gasification facility is used to produce carbon monoxide and methanol. Both of these products are used to produce methyl acetate and ultimately cellulose acetate and acetic acid. The availability of this highly reliable coal gasification facility was the major factor in selecting this location for the LPMEOHTM demonstration unit. An aerial photograph of Eastman's chemicals-from-coal complex is provided in Figure 1.2.4-1.

Three different feed gas streams (H₂ Gas, CO Gas, and the primary syngas feed known as Balanced Gas) are available from existing operations to the LPMEOHTM demonstration unit, thus providing the range of coal-derived syngas ratios (H₂ to CO) needed to meet the technical objectives of the demonstration project.

Figure 1.2.3-1 Integration of Existing Facilities with LPMEOHTM Demonstration Unit



Note: Material Balance Points correspond to Stream Numbers on Table 2.5-1.

Figure 1.2.4-1 Photograph of Eastman Chemical Company's Chemicals-from-Coal Complex



1.2.5 Project Schedule

The project was divided into the following three phases:

Design Construction Operation

The design phase included all of the engineering needed to construct the demonstration unit. This activity started in October of 1993 and was completed in early 1996. The construction phase started in October of 1995 and included the fabrication of the equipment and the field construction tasks (foundations, steel erection, etc.). The demonstration unit was commissioned in March of 1997. The startup was extremely smooth, as demonstrated by the rapid progression from first introduction of syngas (02 April 1997) to stable operation at design conditions (06 April 1997) to a test at 115% of the design production rate of methanol (10 April 1997). The startup also proceeded without injury or environmental incidents.

Based upon the original timetable for the project, the four-year demonstration test period was scheduled to be completed on 31 March 2001, with project close-out to be concluded by 31 December 2001. However, operational issues were encountered with the gas sparger within the LPMEOHTM reactor and with the impact of trace contaminants present in the coal-derived syngas on the long-term performance and life of the methanol synthesis catalyst. Additional information on these items is provided in Section 3.1 of this report. The time required to study

and develop solutions to these technical challenges adversely impacted the ability to execute the Demonstration Test Plan according to the original schedule.

During the project execution, an update on the economics of the coproduction of methanol with electric power via the LPMEOHTM Process was performed. During these studies, modes of operation that had not been anticipated during the development of the original Demonstration Test Plan were identified. One example was the development of a procedure to activate the methanol synthesis catalyst within the LPMEOHTM reactor by using syngas which is diluted with nitrogen (N₂) (known as in-situ activation). If the separate processing equipment to activate catalyst on a batch-wise basis could be eliminated, a capital cost savings of about 10% would be realized. Also, operating the LPMEOHTM reactor at low temperatures at the outset and slowly increasing temperature with time to maintain a constant production of methanol can extend the life of the methanol synthesis catalyst.

Based upon these results, Air Products, on behalf of the Partnership, requested a 15-month nocost time extension on 25 February 2000. This extension was necessary to complete some of the key tests which were originally defined in the September 1996 Demonstration Test Plan, and to allow the opportunity to perform the new tests of significant commercial interest. The extension was approved by DOE on 24 April 2000 and accepted by Air Products on behalf of the Partnership on 08 May 2000. Air Products submitted the first draft for the update to the Demonstration Test Plan on 15 May 2000; the final version was approved by DOE on 15 February 2001.

Over the next year, significant progress was made toward achieving the objectives for operation of the LPMEOHTM demonstration unit during the 15-month period of extended operation. A new commercially available adsorbent material was charged into the catalyst guard bed within the LPMEOHTM demonstration unit, pretreated to remove oxygen (O₂), and brought on stream; initial results for the removal of arsenic from the Balanced Gas feed met performance expectations. Tests on operation in load-following and on-off modes, which are expected to occur in an IGCC complex where both methanol and electric power are produced, were successfully completed. Additional test campaigns on CO-rich syngas were performed. In August of 2001, the first test of the in-situ activation procedure was performed. Initial catalyst performance was less than the value expected from experiments in the laboratory.

Based upon this continued progress, Air Products, on behalf of the Partnership, requested an inscope technical redirection of funds on 01 February 2002. Pending results from a laboratory assessment, the Partnership was considering a recommendation to DOE that a second attempt to perform the in-situ catalyst activation procedure may be warranted. Based upon the redirection of the available project funds and the time necessary to complete the laboratory studies, implement changes, perform the in-situ activation, and assess the results, it was expected that demonstration operations could be supported through December of 2002. Project closeout activities would be completed by the project end date of 31 March 2003 (the project was extended to 30 June 2003 to allow for completion of all technical reports). DOE approved this in-scope technical redirection on 08 March 2002. Based upon this redirection, the list and schedule of tests within the Demonstration Test Plan was updated; these changes are discussed in Section 4.1. As detailed in Section 5.6.4, the performance following the second in-situ activation of catalyst matched the performance expected with properly activated catalyst from the laboratory autoclave.

The operating phase for the LPMEOHTM Demonstration Project was concluded on 31 December 2002. The final schedule, including activities associated with Design Verification Testing (DVT) of DME and the off-site fuel use testing program for stabilized methanol, is provided in Figure 1.2.5-1.

1.3 OBJECTIVES OF THE PROJECT

1.3.1 Primary Objective

The LPMEOHTM Process technology is expected to be commercialized as part of an IGCC electric power generation system. Therefore, the project incorporates the commercially important aspects of the operation of the LPMEOHTM Process which would enhance IGCC power generation. These include the following:

- The coproduction of electric power and of high value liquid transportation fuels and/or chemical feedstocks from coal. This coproduction requires that the partial conversion of syngas to storable liquid products be demonstrated.
- Using an energy load-following operating concept which allows conversion of off-peak energy, at attendant low value, into peak energy commanding a higher value. The load-following concept makes use of gasifier capacity that is under utilized during low-demand periods by using the LPMEOHTM Process to convert the excess syngas to a storable liquid fuel for use in electric power generation during the peak energy periods. This operating concept requires that on/off and electrical load following capabilities be demonstrated.

During operation, the instrumentation system allowed for the collection of engineering data, analysis, and reporting which was done by technical personnel. Typical reporting included onstream factors, material and energy balances, reactor and equipment performance, comparison with laboratory and AFDU results, conversion efficiencies, and catalyst activity. The resulting database will be used to quantitatively evaluate the LPMEOHTM Process technology compared to other commercially available methanol synthesis process technologies.

1.3.2 Secondary Objective

A secondary objective of the project was to demonstrate the production of DME as a mixed coproduct with methanol.

Subject to successful DVT, the Partnership proposed to enhance the project by including the demonstration of the slurry reactor's capability to produce DME as a mixed coproduct with methanol. DVT was required to address issues such as catalyst activity and stability and to provide data for engineering design and demonstration decision making.

Task Name	Duration in Months	Start Date	End Date	% Com	% Sched	19 93	19 94	19 95	19 96	19 97	19 98	19 99	20 00	20 01	20 02	20 03
PHASE 1: DESIGN	87.00 mon	10/1/1993	12/31/2000	<u>6</u>	90											
PROJECT DEFINITION (TASK 1)	16.78 mon	10/1/1993	9/30/1994	6	00											
CONTINUATION APPLICATION (B.P. #2)	0.41 mon	8/2/1994	8/10/1994	6	00		×									
PERMITTING (TASK 2)	47.31 mon	11/17/1993	9/10/1996	ē	100											
NEPA FONSI APPROVAL		6/30/1995	6/30/1995	ē	100			×								
DESIGN ENGINEERING (TASK 3)	38.62 mon	4/15/1994	8/1/1996	6	00											
VENDOR ENGINEERING	33.15 mon	8/10/1994	7/30/1996	6	100											
OFF-SITE TESTING (TASK 4)	82.17 mon	2/25/1994	12/31/2000	Ő	00											
UPDATED FUEL TEST PLAN APPROVAL		8/29/1997	8/29/1997	100	100					×						
DECISION TO CONTINUE DME TESTING		12/4/1996	12/4/1996	100	100				×							
PLANNING, ADMIN & DME DVT (TASK 5)	55.54 mon	10/1/1993	1/20/1997	10	100											
PHASE 2: CONSTRUCTION	65.44 mon	10/17/1994	12/31/2000	<u>10</u>	1 00											
PROCUREMENT (TASK 1)	30.02 mon	10/17/1994	7/30/1996	6	100											
CONSTRUCTION (TASK 2)	22.44 mon	10/2/1995	1/31/1997	ē	100											
TRAINING AND COMMISSIONING (TASK 3)	24.92 mon	9/5/1995	2/27/1997	100	100											
OFF-SITE TESTING (TASK 4)	41.00 mon	8/1/1997	12/31/2000	ē	100											
PLANNING AND ADMINISTRATION (TASK 5)	34.02 mon	6/1/1995	5/1/1998	00	100											
CONTINUATION APPLICATION (B.P. #3)	4.18 mon	5/31/1996	10/8/1996	ē	100											
PHASE 3: OPERATION	74.28 mon	1/20/1997	6/30/2003	1 00	00											
START-UP (TASK 1)	2.27 mon	1/23/1997	4/2/1997	100	100											
METHANOL OPERATION (TASK 2.1)	68.96 mon	4/2/1997	12/31/2002	100	100							-			-	
DISMANTLE PLANT (TASK 2.3)				0	0											
ON-SITE PRODUCT USE DEMO (TASK 3)	2.14 mon	11/3/1997	1/7/1998	100	100											
OFF-SITE PRODUCT USE DEMO (TASK 4)	37.00 mon	12/1/1997	12/31/2000	100	100							-				
DATA ANALYSIS/REPORTS (TASK 5)	71.33 mon	1/20/1997	6/30/2003	100	100										-	
PLANNING AND ADMINISTRATION (TASK 6)	74.28 mon	1/20/1997	6/30/2003	6	100					-	-	-	-	-		
PROVISIONAL DME IMPLEMENTATION	51.12 mon	4/1/1997	7/5/2001	<u>6</u>	1 00											
DME DVT (PDU TESTS) (TASK 3.6)	31.61 mon	4/1/1997	11/19/1999	100	100											
DECISION TO IMPLEMENT	22.47 mon	3/1/1998	1/14/2000	6	100											
DESIGN, MODIFY & OPERATE (TASK 3.2.2)				0	0											

1.4 SIGNIFICANCE OF THE PROJECT

This project demonstrated the commercial viability of Air Products' LPMEOHTM Process using coal-derived syngas, a mixture of H_2 and CO. The DOE's purpose for the demonstration project was to help fulfill the goals and objectives of the Clean Coal Technology Program by demonstrating the potential of a more efficient, liquid-phase reaction process as a preferred alternative to gas-phase reactions for methanol production.

The United States needs future sources of alternative liquid fuels. With domestic oil production declining and imports increasing, the potential of producing affordable liquid fuels from non-petroleum sources could one day prove both strategically and economically important. The LPMEOH[™] Process offers an extremely attractive route to supplementing liquid fuel supplies with methanol made from abundant coal reserves in the United States.

Methanol has a broad range of commercial applications. It can be substituted for or blended with gasoline to power vehicles. It is an excellent fuel for the rapid-start combustion turbines used by utilities to meet peak electricity demands. It contains no sulfur and has exceptionally low NO_x characteristics when burned. It is also used as a chemical feedstock.

The technology that has now completed the demonstration operations phase is ideally suited as an adjunct to a coal-based IGCC power plant - one of the cleanest and most efficient of the 21st century power generating options. When the power plant is not generating at its full capacity, excess syngas can be diverted to make methanol. The methanol could be stored on-site and used in peaking turbines or sold as a commercial fuel or chemical feedstock. In this configuration, the cost of making methanol from coal is likely to be competitive with stand-alone natural gas-tomethanol facilities.

The project objectives were to demonstrate the scaleup and operability of the LPMEOHTM Process under various coal-based feed gas compositions and to gain operating experience for future syngas conversion projects. DVT was also performed to evaluate the potential demonstration of the production of DME from syngas as a mixed coproduct with methanol.

The LPMEOHTM Process offers significant potential to economically produce methanol (using a technology developed in the United States over conventional technologies) and to reduce electric power generation costs with the production of alternative liquid fuels. The domestically developed LPMEOHTM technology can utilize the abundant coal reserves of the United States to produce clean, storable, liquid fuels and chemical feedstocks. Eventual commercialization of the LPMEOHTM Process in IGCC power plants would provide low priced chemical feedstocks and fuel leading to electric power generation cost savings, lower SO₂ and NO_x emissions, and the reduced use of imported liquid fuels.

1.5 DOE'S ROLE IN THE PROJECT

1.5.1 <u>Clean Coal Technology Program</u>

The DOE's Clean Coal Technology Program has been implemented through five competitive solicitations. Congress set the basic goals for the program and for each solicitation in the enabling legislation and accompanying report language. DOE subsequently translated the guidance into performance-oriented solicitations. For each solicitation, evaluation criteria were defined and weighted to reflect specific congressional guidance and the current program objectives. This process enabled industry to set the technical agenda by allowing companies to propose their own technologies as qualifying projects. This had the significant benefit of attracting higher levels of private-sector cost-sharing and increasing the likelihood of realizing commercialization objectives.

An important attribute to the solicitation approach used to implement the program was the use of multiple solicitations spread over a number of years. Allowing time between solicitations made it possible to adjust program objectives. At the end of each solicitation, Congress provided the flexibility as needed to effectively implement the program.

Each solicitation was issued as a Program Opportunity Notice (PON). Proposals for demonstration projects consistent with the objectives of each PON were submitted to DOE by a specific deadline. DOE evaluated the proposals and announced those projects selected for negotiation.

The objective of the Round III was to solicit cost-shared clean coal technology projects to demonstrate innovative, energy-efficient technologies capable of being commercialized in the 1990's. These technologies were to be capable of (1) achieving significant reductions in emissions of SO_2 and/or NO_x from existing facilities to minimize environmental impacts, such as transboundary and interstate pollution, and/or (2) providing for future energy needs in an environmentally acceptable manner. DOE received 48 proposals and selected 13 projects as best furthering the goals and objectives of the PON.

The LPMEOH[™] Demonstration Project was selected in December of 1989 under Round III of the Clean Coal Technology Program.

1.5.2 Management Plan

The DOE entered into a Cooperative Agreement with Air Products and Chemicals, Inc., which was later novated to the Partnership (between Air Products and Eastman) to conduct the project. The DOE monitored the project through the Contracting Officer and the Contracting Officer's Technical Representative (COTR). The Partnership managed the project through an Air Products Program Manager, who was assisted by a team of technical and managerial personnel from Air Products and from Eastman.

The organizational chart for the design and construction phases is included in the Public Design Report (Final Report - Volume 1). The organizational chart for the operating phase is shown in Figure 1.5.2-1.

1.5.2.1 DOE

The DOE was responsible for monitoring all aspects of the project and for granting or denying approvals required by the Cooperative Agreement. The DOE Contracting Officer was the authorized representative of the DOE for all matters related to the Cooperative Agreement.

The DOE Contracting Officer appointed a COTR, who was the authorized representative for all technical matters and had the authority to issue "Technical Advice." The COTR also approved those reports, plans, and technical information required under the Cooperative Agreement. The DOE COTR did not have the authority to issue any technical advice that assigned additional work outside the Statement of Work, increased or decreased the total estimated cost or time required for performance of the Cooperative Agreement, changed any of the terms, conditions, or specifications of the Agreement, or interfered with The Partnership's right to perform the terms and conditions of the Agreement. All Technical Advice was issued in writing by the DOE COTR.

1.5.2.2 The Partnership

The Air Products Program Manager coordinated the overall project and was responsible for all communication with the DOE and for interfacing with the DOE COTR.

The Air Products Program Manager was the primary focal point for this project having responsibility and authority for direction of the project subsequent to the signing of the Cooperative Agreement. The Air Products Program Manager was the principal representative between the Partnership and the DOE.

The Program Manager's responsibilities encompassed both technical and fiscal considerations, including the following:

- Overall technical coordination of the program
- Monitoring of program cost
- Monitoring of program planning
- Monitoring of program schedule
- Commitment of resources to optimize performance under the Cooperative Agreement
- Reporting requirements
- Final review of all contract deliverables

Throughout the course of this project, reports dealing with technical, cost, and environmental aspects of the project were prepared by the Partnership and provided to DOE.

Figure 1.5.2-1 Kingsport LPMEOH[™] Demonstration Project Operating Phase Organization Chart


2.0 TECHNOLOGY DESCRIPTION

2.1 DESCRIPTION OF THE DEMONSTRATED TECHNOLOGY

Methanol is a versatile commodity chemical with applications ranging from fuels to feedstocks for higher valued chemicals such as methyl methacrylate and methyl tertiary-butyl ether (MTBE, an octane booster). Many applications have evolved for methanol as a fuel including methanol as a gasoline extender, the methanol to gasoline process, methanol as a feedstock to fuel cell systems, and IGCC technology for the coproduction of methanol and electricity.

The LPMEOHTM Process represents a major departure from traditional gas-phase routes to methanol in the method of removing the heat of reaction. The reactions of H_2 and CO and carbon dioxide (CO₂) to form methanol are highly exothermic.

 $CO + 2H_2 \longrightarrow CH_3OH + 21.7 \text{ KCAL/gmol}$ $CO_2 + 3H_2 \longrightarrow CH_3OH + H_2O + 12.8 \text{ KCAL/gmol}$ TYPICAL REACTION CONDITIONS: 1,000 psig 440°-520°F

One of the most difficult design problems of the methanol synthesis process is removing the heat of reaction while maintaining precise temperature control to achieve optimum catalyst life and reaction rate. Catalyst life is seriously reduced by excessive temperatures.

Reactor schematics for conventional gas-phase reactors are shown in Figure 2.1-1. The catalyst is in the form of pellets. In these conventional gas-phase reactors, either cool unreacted gas is injected at stages in the catalyst bed, or internal cooling surfaces are used to provide temperature control. These schemes, however, were developed for diluted syngas which yields low conversion per pass. The LPMEOHTM Process (Figure 2.1-2) uses fine catalyst particles entrained in an inert hydrocarbon liquid, usually a mineral oil. The mineral oil acts as a temperature moderator and a heat removal medium, transferring the heat of reaction from the catalyst surface via the liquid slurry to boiling water in an internal tubular heat exchanger. As a result of this capability to remove heat and maintain a constant, highly uniform temperature through the entire length of the reactor, the slurry reactor can achieve much higher syngas conversion per pass than its gas-phase counterparts.

In addition, the LPMEOHTM Process is particularly well suited to coal-derived syngas which is rich in CO. These capabilities make the LPMEOHTM Process a potentially lower-cost conversion route to methanol, especially when methanol coproduction is added to a coal-based IGCC power plant. For a modest increase in complexity of an IGCC power plant, the methanol coproduction scheme produces a storable liquid fuel in parallel with electric power production, providing a significant turndown and peak-load capability for the IGCC power plant.

Figure 2.1-1 Gas-Phase Methanol Reactors



Information on the application of the LPMEOHTM Process to the IGCC power plant is provided in Section 9 of this report.

2.2 DESCRIPTION OF THE DEMONSTRATION FACILITY

The reactor used in the LPMEOH[™] Process is unlike conventional gas-phase reactors that use fixed beds of catalyst pellets and largely depend upon recycle diluent gas to both dilute the CO concentration and control the temperature rise caused by the heat of reaction. The LPMEOH[™] reactor is a slurry reactor with small, powder-size catalyst particles suspended in inert mineral oil. The syngas bubbles up through the slurry where the H₂ and CO dissolve in the oil and diffuse to the catalyst surface where the methanol reaction occurs. The product methanol diffuses out of the slurry and exits as a vapor with the unreacted syngas. The inert oil acts as a heat sink and permits isothermal operation. The net heat of reaction is removed via an internal heat exchanger which produces steam. Unlike gas-phase reactors that limit per-pass conversion of syngas to methanol to accommodate the reaction exotherm, the LPMEOH[™] reactor maintains isothermal operation. The methanol vapor leaves the reactor and is condensed to a liquid, sent to distillation columns for removal of higher alcohols, water (H₂O), and other impurities, and is then stored in lot tanks for sampling prior to being sent to Eastman's methanol storage. A portion of the unreacted syngas is sent back to the reactor with the recycle compressor, improving cycle efficiency. The methanol is used for downstream feedstocks and for off-site fuel-use testing.

Unlike gas-phase reactors, the LPMEOHTM reactor is tolerant to CO-rich gas. Shift and CO₂ removal are not required. Low H₂-to-CO ratios are acceptable as is any CO₂ content. Finally, in contrast to the gas-phase reactor in which the catalyst is sensitive to flow variations and changes from steady-state, the LPMEOHTM reactor is eminently suited for load-following and on/off operation in electrical generating applications.

The LPMEOH[™] demonstration unit is integrated with Eastman's coal gasification facility and inserted in parallel with an existing gas-phase methanol unit.

No new equipment items were added to the LPMEOHTM demonstration unit during the operating phase; however, a different design for the pump used to add fresh mineral oil to the process was installed to replace the pumps from the original design. Also, Air Products developed two alternative designs for the gas sparger within the LPMEOHTM reactor, the last of which has been in service since March of 1999. The adsorbent material in the catalyst guard bed within the LPMEOHTM demonstration unit was also changed to remove arsine in addition to metal carbonyls, and changes were made to the piping and instrumentation system around this catalyst guard bed to allow for the new adsorbent to be pretreated prior to use in the process. Details on the changes can be found in Section 3.0 of this report.

2.3 PROPRIETARY INFORMATION

As noted in Section 1.1, the scope of this report is limited to nonproprietary information. Therefore, although its content is insufficient to provide a complete tool for designing a LPMEOHTM unit, it will serve as a reference for the design considerations involved in developing a commercial-scale facility.

2.4 SIMPLIFIED PROCESS FLOW DIAGRAM

The LPMEOH[™] demonstration unit consists of five main process sections: Feed Gas Purification, Compression, Reactor Loop and Catalyst Reduction, Distillation, and Storage and Miscellaneous. The Feed Gas Purification area includes the feed gas supply tie-ins and a catalyst guard bed. The Compression area contains the recycle gas compressor and associated systems. The Reactor Loop and Catalyst Reduction area includes the reactor with its associated equipment and the slurry handling equipment and utility oil skid associated with catalyst preparation. The Distillation area includes two distillation columns and heat exchangers. The Storage and Miscellaneous area comprises oil storage and other process components. A detailed description of these sections is provided in Section 4.3 of the Public Design Report (Final Report - Volume 1). A block diagram for the LPMEOH[™] demonstration unit is provided in Figure 2.4-1. A glossary of syngas terminology is provided in the beginning of this document.

2.5 STREAM DATA

Table 2.5-1 shows the material balance for the points shown on Figures 1.2.3-1 and 2.4-1.

2.6 PROCESS AND INSTRUMENTATION DIAGRAMS

A set of Process and Instrumentation Diagrams (P&ID's) can be found in Appendix A of the Public Design Report (Final Report - Volume 1).

Figure 2.4-1 LPMEOHTM Demonstration Unit Simplified Process Flow Diagram



Note: Material Balance Points correspond to Stream Numbers on Table 2.5-1.

CASE:	L:\kingspr	t\ hmb\ 1feb95\ de	sign					
Stream No.		10	109	120	148	149	20	204
Press	PSI	875.0	785.0	735.0	722.0	722.0	765.0	85.0
Temp	С	37.8	169.5	121.1	40.5	40.5	37.8	40.1
Temp	F	100.0	337.1	249.9	105.0	105.0	100.0	104.2
Lbmol/hr								
H2		2.636	4402.484	2967.899	252.163	2712.126	77.996	3.610
CO		127.846	1818.957	1121.754	95.167	1023.566	11.594	3.021
N2		1.318	293.947	294.054	24.922	268.047	12.648	1.084
CO2		0.000	654.744	635.137	51.831	557.470	3.162	25.836
MEOH		0.000	42.496	/44.45/	3.952	42.504	0.000	698.001
DME		0.000	3.790	5.057	0.353	3.796	0.000	0.909
H2U		0.000	0.448	21.642	0.042	0.448	0.000	21.152
		0.000	0.016	0.447	0.002	0.016	0.000	0.429
C40H		0.000	0.001	0.005	0.000	0.001	0.000	0.003
IBOH		0.000	0.000	0.020	0.000	0.000	0.000	0.020
C50H		0.000	0.001	0.007	0.000	0.000	0.000	0.000
MFAC		0.000	0.065	0.242	0.006	0.065	0.000	0.010
MEFM		0.000	1.256	2.790	0.117	1.259	0.000	1.415
C1		0.000	12.228	13.449	1.137	12.234	0.000	0.077
C2		0.000	3.549	3.982	0.331	3.557	0.000	0.095
C3		0.000	1.007	1.171	0.094	1.010	0.000	0.067
OIL		0.000	0.000	0.032	0.000	0.000	0.000	0.032
Total molar flow		131.8	7235.0	5812.3	430.1	4626.1	105.4	756.1
Total mass flow	lb/hr	3,623	98,846	98,686	6,337	68,154	975	24,195
Enthalpy	MMBtu/hr	-6.083	-188.288	-221.879	-13.674	-147.071	-1.066	-79.011
mol%								
H2		2.000	60.850	51.062	58.627	58.627	74.000	0.477
CO		97.000	25.141	19.300	22.126	22.126	11.000	0.400
N2		1.000	4.063	5.059	5.794	5.794	12.000	0.143
CO2		0.000	9.050	10.927	12.051	12.051	3.000	3.417
DME		0.000	0.587	12.808	0.919	0.919	0.000	92.317
		0.000	0.052	0.067	0.062	0.062	0.000	0.120
		0.000	0.000	0.372	0.010	0.010	0.000	2.797
C3OH		0.000	0.000	0.000	0.000	0.000	0.000	0.007
C40H		0.000	0.000	0.001	0.000	0.000	0.000	0.000
IBOH		0.000	0.000	0.000	0.000	0.000	0.000	0.000
C50H		0.000	0.000	0.000	0.000	0.000	0.000	0.002
MEAC		0.000	0.001	0.004	0.001	0.001	0.000	0.023
MEFM		0.000	0.017	0.048	0.027	0.027	0.000	0.187
C1		0.000	0.169	0.231	0.264	0.264	0.000	0.010
C2		0.000	0.049	0.069	0.077	0.077	0.000	0.013
C3		0.000	0.014	0.020	0.022	0.022	0.000	0.009
OIL		0.000	0.000	0.001	0.000	0.000	0.000	0.004
Total		100	100	100	100	100	100	100
Vapor Fraction		1.000	1.000	1.000	1.000	1.000	1.000	0.025
Liquid Fraction		0.000	0.000	0.000	0.000	0.000	0.000	0.975
Mol. Wt. Mix		27.490	13.662	16.979	14.733	14.733	9.255	32.000
Mol. Wt. Vapor		27.490	13.662	16.979	14.733	14.733	9.255	32.286
Mol. Wt. Liquid	1.0.							31.992
Densty	LB/CUFT	4.092	1.206	1.621	1.701	1.701	1.114	13.386
vapor Density		4.092	1.206	1.621	1.701	1.701	1.114	0.461
uquia Density								48.981
1	1						1	1

Table 2.5-1Heat and Material Balance Summary

CASE:	L:\kingspr	t\ hmb\ 1feb95\ de	sign		
Stream No.		213	216	242	30
Press	PSI		40.0	80.0	790.0
Temp	С		40.7	100.8	37.8
Temp	F		105.3	213.4	100.0
Lbmol/hr					
H2		3.610	0.000	0.000	1608.013
со		3.021	0.000	0.000	656.961
N2		1.084	0.000	0.000	11.859
CO2		25.836	0.000	0.000	94.868
MEOH		3.601	520.901	173.499	0.000
DME		0.909	0.000	0.000	0.000
H2O		0.013	0.054	21.084	0.000
ETOH		0.000	0.043	0.386	0.000
СЗОН		0.000	0.000	0.063	0.000
C4OH		0.000	0.000	0.026	0.000
IBOH		0.000	0.000	0.085	0.000
C5OH		0.000	0.000	0.018	0.000
MEAC		0.170	0.002	0.000	0.000
MEFM		1.415	0.000	0.000	0.000
C1		0.077	0.000	0.000	0.000
C2		0.095	0.000	0.000	0.000
C3		0.067	0.000	0.000	0.000
OIL		0.000	0.000	0.032	0.000
Total molar flow		39.9	521.0	195.2	2371.7
Total mass flow	lb/hr	1,521	16,694	5,980	26,150
Enthalpy	MMBtu/hr	-5.147	-53.303	-19.938	-46.877
mol%					
H2		9.048	0.000	0.000	67.800
CO		7.572	0.000	0.000	27.700
N2		2.718	0.000	0.000	0.500
CO2		64.755	0.000	0.000	4.000
MEOH		9.026	99.981	88.885	0.000
DME		2.278	0.000	0.000	0.000
H2O		0.033	0.010	10.802	0.000
EIUH		0.001	0.008	0.198	0.000
C30H		0.000	0.000	0.033	0.000
		0.000	0.000	0.013	0.000
		0.000	0.000	0.044	0.000
MEAC		0.000	0.000	0.009	0.000
MEEM		3.546	0.000	0.000	0.000
		0 103	0.000	0.000	0.000
C2		0.133	0.000	0.000	0.000
02		0.207	0.000	0.000	0.000
0		0.000	0.000	0.000	0.000
Total		100	100	100	100
lotar		100	100	100	100
Vapor Fraction		1 000	0.000	0.000	1 000
Liquid Fraction		0.000	1.000	1.000	0.000
Mol. Wt. Mix		38.132	32.042	30.634	11.026
Mol. Wt. Vapor		-0.003			11.026
Mol. Wt. Liauid			32.042	30.634	
Density	LB/CUFT		48.306	45.200	1.377
Vapor Densitv	LB/CUFT				1.377
Liquid Density	LB/CUFT		48.306	45.200	
•					

Table 2.5-1Heat and Material Balance Summary (cont'd)

3.0 UPDATE OF THE PUBLIC DESIGN REPORT

3.1 DESIGN AND EQUIPMENT CHANGES

Over the first three years of operation, the LPMEOH[™] demonstration unit achieved an availability in excess of 97%. Engineering data was collected and analyzed so that the

engineering tools that are necessary for the design of larger-scale LPMEOH[™] plants could be improved. Much of the work associated with process optimization was completed, and ideas to reduce the number of rotating equipment items which simplify the process were successfully implemented.

Early performance of the proprietary gas sparger within the LPMEOH[™] reactor (7.5' internal diameter) met expectations for gas distribution and reactor hydrodynamics, but the



pressure drop across the device increased over time. Ultimately, the flow resistance across the sparger reached a value that severely limited the quantity of syngas that could be processed. After some proprietary studies conducted by Air Products, two updated spargers, designed by Air Products and fabricated by Eastman for the Partnership, were placed into service, the latest being in March of 1999. As detailed in Section 5.3, this device has shown excellent performance, with no increase in pressure drop over time.

Another technical challenge was the impact of trace contaminants present in the coal-derived syngas (which is the primary feedstock to the LPMEOHTM reactor) on the long-term performance and life of the methanol synthesis catalyst. These species, which are typically found in coal and are present in syngas streams in parts-per-billion levels, reduce the life of the catalyst, thus adversely affecting the economics of the LPMEOHTM Process. During the first year of operation, the rate of deactivation of the methanol synthesis catalyst was significantly higher that the target value which had been previously determined in a 4-month proof-of-concept test at the LaPorte AFDU. Significant progress was made in identifying both the species of catalyst guard bed vessels, can effectively and economically remove these contaminants. Additional information is provided in Section 5.2.

3.1.1 Condensed Oil (29G-01) and Oil Makeup (29G-03) Pumps

During plant commissioning activities, two sets of pumps required additional attention and impacted the mode of operation of the LPMEOHTM demonstration unit during startup.

The 29G-01 condensed oil pumps were designed to return condensed and entrained oil and catalyst from the 29C-05 secondary oil knock-out drum and 29C-06 cyclone to the 29C-01 LPMEOHTM reactor. A twin-screw design was selected due to the potential for high solids loadings at reactor temperatures and pressures. The 29G-03 oil make-up pumps were intended to provide seal flush to the condensed oil pumps, oil addition to the reactor loop, and high-pressure (1,100 psig) piping flush oil. The pumps also employed a twin-screw design with extremely tight clearances to generate high pressure differential, as the make-up oil was stored in the 29D-30 oil storage tank at low pressure (less than 10 psig). During initial commissioning activities in January of 1997, these pumps operated well during testing at the required discharge pressure of 700 to 900 psig. However, during subsequent testing, both oil make-up pumps exhibited difficulty in delivering oil at the required pressure. Notably, since the oil make-up pumps could not operate at full suction pressure.

During the plant design phase, in anticipation of operating problems with the condensed oil pumps, the elevation of the slurry collection equipment (secondary oil knock-out drum and cyclone) was raised high enough to permit any entrained or condensed material to gravity-drain back to the reactor. Because of the inability of the oil make-up pumps to provide seal flush for the condensed oil pumps, operation in the free-draining mode began at startup, and initial results were positive. Fresh make-up oil was added to the process by using the 29G-30 slurry transfer pump, which was designed to transfer catalyst slurry from the 29C-30 catalyst reduction vessel to the LPMEOHTM reactor. The slurry transfer pump packing also requires flush from the oil make-up pumps; however, it was determined that operation of the slurry transfer pump in clean-oil service without packing flush would not adversely affect the service life of the pump. Accordingly, make-up oil was batch-transferred from the oil storage tank to the catalyst reduction vessel, and then pumped to the reactor by the slurry transfer pump.

The gravity-drain line did exhibit intermittent plugging or vapor-locking during operation. Early in the operating campaign, blockages could be cleared by opening a utility line between the secondary oil knock-out drum and the catalyst reduction vessel and briefly blowing down to low pressure; piping connections to provide flush oil were rendered useless by the inoperable high-pressure oil make-up pumps. However, on 25 April 1997, a blockage occurred in the free-drain line that could not be removed by this method. Since the solids concentration of the condensed and entrained oil and catalyst slurry was relatively low, it was determined that the slurry transfer pump could pump this material without a packing flush on the pump. Condensed oil was batch-transferred from the secondary oil knock-out drum to the catalyst reduction vessel, and then pumped to the reactor. The frequency of the transfer to the catalyst reduction vessel was about once every three hours, and the catalyst reduction vessel was pumped to the reactor about once every 10 hours. The calculated accumulation rate of condensed and entrained slurry (1.5 to 2.0 gpm) matched the expected liquid traffic within the oil/catalyst collection equipment. This early

operating experience confirmed that the ability to flush piping systems in slurry service was an important operability requirement.

After an engineering review by Air Products and Eastman, a new 29G-03 oil make-up pump was specified with a simpler design than the original twin-screw configuration. Only one of these new pumps was purchased and installed to develop operational experience at minimum capital cost. The new pump, priced at \$7,900, was brought online in October of 1997 and met all operational requirements. The availability of this pump to flush out periodic blockages in the gravity-drain line established the viability of that concept to return condensed and entrained slurry to the LPMEOHTM reactor, and as a result, the condensed oil pumps were not put into service under operating conditions. The potential benefits are tremendous for future designs, including: significant capital cost savings from the elimination of two slurry pumps and their ancillaries (and perhaps the redundant oil make-up pump); increased operating flexibility; and lower maintenance costs by eliminating the seal system for the slurry pumps.

The long-term success of the gravity-draining system to return entrained slurry and condensed oil to the LPMEOHTM reactor eliminated the need for the condensed oil pumps and their continuous seal flush requirement. As a result, the single oil make-up pump has continued to operate in batch mode only. In this mode, the pump has added oil to the LPMEOHTM Process to provide flush oil to piping and instrumentation associated with the LPMEOHTM reactor and cyclone; the amount of flush oil can be optimized to match the required amount of make-up oil that is necessary to overcome the average oil loss rate with the methanol product (nominally 0.1 to 0.2 gpm). The new pump has operated well in over five years of operation in this batch mode.

3.1.2 Slurry Return Pump (29G-02)

During the initial startup of the LPMEOHTM demonstration unit, individual batches of methanol catalyst slurry were activated in the 29C-30 catalyst reduction vessel and then pressure-transferred into the 29D-02 slurry tank. After activation of the ninth and final batch, the entire charge was to be pumped from the slurry storage tank to the reactor using the 29G-02 slurry return pump. This pump included a packing design which required a continuous flush of clean oil from the oil storage tank using the 29G-34 oil feed pump. During the transfer operation, seal oil leaked through the packing at a significantly higher rate than expected. If this condition had continued throughout the transfer, the oil would have diluted the slurry considerably, perhaps to the point of overfilling the reactor vessel. The operations team decided to apply N₂ at 45-50 psig to the slurry tank and pressure-transfer the remaining slurry to the reactor. This procedure was accomplished successfully, and the technique was used for all remaining slurry transfer steps when the slurry tank was utilized. This pump will be eliminated in future designs, providing an opportunity for capital savings and improved operability.

3.1.3 Availability of Hydrogen Gas (H₂ Gas)

During the design phase for the LPMEOHTM demonstration unit, one of the feed syngas streams was the purge stream from the existing gas-phase methanol unit (called hydrogen gas or H₂ Gas). This stream contained an excess of H₂ over the stoichiometric balance for the production of methanol, and was expected to be used to adjust the H₂:CO ratio during several of the tests within the demonstration test plan. However, due to pressure control issues in the Eastman chemicals-from-coal complex, the operating pressure for H_2 Gas had to be reduced to less that the operating pressure of the LPMEOHTM demonstration unit. As a consequence, H_2 Gas was not available as a feed gas. The impact of this change from the plant design basis on the ability to meet the technical objectives for the LPMEOHTM Demonstration Project was minimal; details are provided in Section 4.1.

3.1.4 Gas Sparger

Because the performance of a slurry bubble column reactor depends heavily on the method of introduction of feed gas, the gas sparger design is a key feature of the overall LPMEOHTM reactor design. The gas sparger design for the LPMEOHTM demonstration unit was based on the device used during the successful 4-month proof-of-concept run at the LaPorte AFDU in 1988/89. In general, the process scaleup from the LaPorte AFDU to commercial scale was a significant objective for the LPMEOHTM Demonstration Project, and the gas sparger was one of the key components in this area.

Three different gas sparger designs were utilized during the operation of the LPMEOH[™] demonstration unit. The original gas sparger design was used from startup in April of 1997 until November of 1997. The second gas sparger design was a modification of the first sparger using similar design principles and was installed in December of 1997. This design was used until March of 1999. A modified sparger design, again using the same design principles as the original device, was installed in March of 1999 and was used for the remainder of the demonstration test program. Details on the performance of the three devices are provided in Section 5.3.

3.2 DEMONSTRATION PLANT EQUIPMENT COST UPDATE

The updated equipment cost summary for the LPMEOHTM demonstration unit is provided in Table 3.2-1. The only additional equipment that was purchased during the demonstration period was the replacement for the 29G-03 oil makeup pump (described in Section 3.1.1). It should be noted that, if the pumps and associated equipment described in Section 3.1 had not been installed, the equipment cost of the LPMEOHTM demonstration unit would have been reduced by almost 5% (\$210,000 out of \$4.4 million).

Table 3.2-1 Kingsport LPMEOH™ Demonstration Unit - Update of Equipment Cost Summary (Dollars in Thousands)

Area	Item Description	Total	Subtotal
А	Feed Gas Purification	\$77.0	\$77.0
В	Compression	\$776.8	
	Compression		\$699.6
	Exchangers		\$37.6
	Separators		\$39.6
С	Reactor Loop and Catalyst Reduction	\$2,202.1	
	Equipment	,	\$1,487.8
	Exchangers		\$332.3
	Pumps		\$382.0
D	Distillation	\$703.4	
	Equipment		\$309.5
	Exchangers		\$214.7
	Tanks		\$131.3
	Pumps and Misc.		\$47.9
E	Storage and Miscellaneous	\$648.6	\$648.6

Total \$4,407.9 \$4,407.9

3.3 <u>DEMONSTRATION PLANT OPERATING COST UPDATE</u>

Table 3.3-1 summarizes the updated operating costs for the LPMEOHTM demonstration unit. A new category of costs was added to cover the changeout and disposal of the adsorbent from the 29C-40 carbonyl guard bed (the costs and consumption utilized at the end of the demonstration period are used in the tabulation of costs). The costs were developed using the performance following the successful demonstration of the in-situ activation procedure and temperature programming (details are provided in Section 5.6) to extend the life of the methanol synthesis catalyst.

Table 3.3-1 Kingsport LPMEOH™ Demonstration Unit - Update of Operating Costs Base Year 2002

ANNUAL FIXED OPERATING COST

Number of Operators per Shift	1.5
Number of Shifts per Week	4.2
Operating and Technical Support Labor Rate, \$/hr	\$76.71

Total Annual Operating and Technical Support Labor Cost	\$671,971
Total Annual Maintenance Labor Cost	\$510,440
Total Annual Maintenance Material Cost	\$249,999
Total Annual Administration and Support Labor Cost	\$1,925,670
TOTAL ANNUAL FIXED O&M COST	\$3,358,081

VARIABLE OPERATING COST

Commodity	Unit	\$/Unit ⁽¹⁾	Quantity/hr	\$/hr
Compressed Air	1,000 SCF	0.22	18.00	3.95
Demineralized Water	1,000 Lb	0.66	2,500.00	1.65
Electricity	kWh	0.04	770.00	33.82
Filtered Water	1,000 Gal	0.49	0.30	0.15
High Pressure Steam	1,000 Lb	5.49	6.00	32.94
Low Pressure Steam	1,000 Lb	4.94	-1.00	-4.94
Medium Pressure Steam	1,000 Lb	4.39	12.50	54.90
Nitrogen	1,000 SCF	0.66	4.00	2.64
Synthesis Gas	1,000 SCF	4.48	850.00	3,805.62
Catalyst	1 Lb	10.29	5.21	53.58
Mineral Oil	1 Gal	3.03	7.18	21.78
Guard Bed Adsorbent (incl. disposal)	1 Lb	9.07	2.81	25.52
Waste Water	1 year	15,152.12		1.97
Slurry Reclaiming	1 Lb	0.38	7.81	2.95
Distillation of 29C-20 Underflow	1 Gal	0.05	781.25	42.89
Area Services (laboratory, general services)	1 month	86,985.60		135.92
TOTAL VARIABLE OPERATING COST				4,215.33
TOTAL ANNUAL OPERATING HOURS		7,680		

⁽¹⁾ - These unit costs reflect the costs incurred by the Air Products Liquid Phase Conversion Company, L.P. (the Partnership) and are typical of published utility costs (for example, Process Economics Program Report 136A, "Plant Utility Costs", published by SRI International, Menlo Park, CA), and have been indexed for inflation from 1997 to 2002 using data from the Producer Price Index – Chemicals and Allied Products.

4.0 DEMONSTRATION PROGRAM

4.1 TEST PLANS

A Demonstration Test Plan^(a) was prepared from the Technical Objectives set for the project. This original version was comprised of 37 specific tests. These are given in Table 3-2 of the Public Design Report (Final Report - Volume 1). The conditions imposed by each of these tests were incorporated in the process design of the LPMEOHTM demonstration unit.

Based upon the results from the initial three years of operation of the LPMEOHTM demonstration unit as well as the identification of new tests of significant commercial interest, an update to the Demonstration Test Plan was proposed by Air Products in 2000. The impact of catalyst poisons, both as post-construction debris and as present in the Balanced Gas feed, on the long-term performance of the catalyst altered the timeline for completing the objectives as stated in the Section 1.3.1 of this report. As a result, operating activities were focused upon studying the effects of catalyst concentration in the reactor (an important parameter in plant economics) on reactor hydrodynamics. As noted in Section 5.1, the weight of catalyst in the reactor reached a maximum of 151% of design during this step-wise approach to determining the operating limits for the slurry bubble column reactor.

Operation of the LPMEOH[™] demonstration unit was different relative to the original design concept, thus eliminating either the ability or the need to complete several of the tests which had been included in the original Demonstration Test Plan (refer to Table 3.2 of the Public Design Report [Final Report - Volume 1] for the relevant test number and description):

- In the original plant configuration, H₂ Gas was included as one of the feed syngas streams to the LPMEOHTM demonstration unit. However, as noted in Section 3.1, H₂ Gas was not available as a feed gas. This eliminated the ability to complete Test 12. Tests 13 and 14 also became redundant, as these conditions were covered in other tests.
- 2) Based upon the inspection of the cyclone separator and feed/product economizer at the vapor outlet of the LPMEOH[™] reactor, it was concluded that some form of separation device which limits the flow of entrained slurry from the reactor to the tubes of the heat exchanger will be required in future designs. This eliminated the need to perform the portion of Test 19 during which the cyclone would be bypassed.
- 3) A consequence of increasing the catalyst slurry concentration in the LPMEOH[™] reactor beyond the design value was that the extent of conversion of CO was increased. As the amount of CO is depleted in the reactor effluent and recycle gas streams, the H₂:CO ratio at the reactor feed was increased. Adequate data had already been acquired at elevated H₂:CO ratios such that a separate test for natural gas reformer-type syngas (Test 24) was no longer required.
- 4) Since the restart of the LPMEOH[™] demonstration unit on the second charge of fresh catalyst in December of 1997, the majority of operating time was spent at a reactor

temperature of 235°C. This allowed for additional study of catalyst performance as a function of operating temperature, which has importance for the LPMEOHTM Process in applications regarding coal-derived syngas (effect of operating temperature on the impact of catalyst poisons) and natural gas systems (extending catalyst life in the absence of poisons).

- 5) The amount of time allocated in the original Demonstration Test Plan to the study of catalyst performance above the original design temperature of 250°C was overestimated. Also, it was not logical to return to lower temperature in the reactor following operation at elevated temperature; this eliminated the need to perform Test 29.
- 6) During the operating program for the LPMEOH[™] demonstration unit, modes of operation which were not anticipated during the development of the original Demonstration Test Plan were identified. The ability to activate the methanol synthesis catalyst within the LPMEOH[™] reactor by using syngas which is diluted with N₂ (so-called in-situ activation) can result in capital cost savings of about 10% by elimination of the separate processing equipment which is used at the demonstration unit. Also, the life of the methanol synthesis catalyst can be extended by operating the LPMEOH[™] reactor at low temperatures at the outset and slowly increasing temperature with time to maintain a constant production of methanol (so-called temperature programming).

A tabulation of the conditions that were operated during the execution of the Demonstration Test Plan is provided in Table 4.1-1. This shows the test runs and the operating conditions. The time chart for execution of the Plan is provided in Figure 4.1-1. The changes to the Plan allowed for the satisfactory completion of the objectives from the Statement of Work while also providing the ability to perform new tests of significant commercial interest.

4.2 **OPERATING PROCEDURES**

Summaries of the main procedures associated with the startup, operation, and shutdown of the LPMEOHTM demonstration unit are provided. The P&ID's, located in Appendix A of the Public Design Report (Final Report - Volume 1), contain references to equipment and instrumentation systems.

Table 4.1-1 Test Conditions at LPMEOH[™] Demonstration Unit

	Typical Conditions									
Run #	Test Run Description	Temp (Deg C)	Wt% Catalyst	H2/CO Ratio at Inlet	Fresh Feed Rate (KSCFH)	MeOH Rate (TPD)	Test Run Objective			
Task 2.	I.1 - Process Shakedown and (Catalyst Aging	g:							
1	Initial Shakedown; and Design Production Tests	250	31	3.5	833	248	Initial facility start-up and shakedown			
2	Gassed Slurry Level	Data during t	his period co	vered within	other test runs		Maximize reactor volumetric productivity			
3	Reactor Feed: Texaco-Type Syngas	250	34	0.78	725	190	IGCC applications; Comparison with LaPorte AFDU and laboratory			
5	Check @ Test 1 Conditions	250	28	3.5	700	200	Catalyst life; Comparison with LaPorte AFDU and laboratory			
6	Catalyst Addition and Aging	235	36	4.1	750	225	Catalyst life			
7	Free-Drain Entrained/ Condensed Oil to Reactor	Data during t	his period co	vered within	other test runs		Enhance system flexibility and econonic benefit			
8	Operation @ Design Feed Gas Rates	235	43	4.0	850	256	Confirmation of operation at design condition			
9	Check for Limitation on Catalyst Slurry Concentration	235	49	3.7	750	207	Maximize reactor volumetric productivity			
10	Catalyst Addition to Reach Maximum Productivity	235	44	3.5	700	196	Maximize reactor volumetric productivity			
Task 2.	I.2 - Process Operational Test	Phase:								
11	Catalyst Withdrawal/ Addition Test	235	44	3.2	670	190	Catalyst life			
21	Reactor Feed: Destec-Type Syngas	235	45	1.0	700	200	IGCC applications			
2000-1	Catalyst Withdrawal/ Addition Tests	235	43	3.2	725	210	Catalyst life			
2000-2	Turndown and Ramping Tests	5% of design	syngas flow	per minute	ramp rate achie	ved.	IGCC applications			
2000-3	Load-Following and On/Off Tests	Feed gas off	for 10-14 ho	urs, plant ra	mped to re-star	t.	IGCC applications			
2000-4	Reactor Feed: H2/CO = 1.0	235	43	1.0	650	180	IGCC applications			
2000-5	Reactor Feed: H2/CO = 0.7	235	39	0.67	500	130	IGCC applications			
2000-6	Final Temperature Ramp of Methanol Synthesis Catalyst	235 -> 250	47	2.6	580	165	Catalyst life			
2000-7	Reactor Inspection	Data during t	his period co	vered within	other test runs		Physical inspection of key system components			
2000-8	In-situ Catalyst Activation 1, Temperature Programming	216 -> 250	40	3.0	500 -> 650	150 -> 200	Tests of recent commercial interest			
2000-8	In-situ Catalyst Activation 2, Temperature Programming	215	40	3.6	500 -> 650	170 -> 180	Tests of recent commercial interest			
2000-9	Heat/Mass Balance Check: Reactor Inlet H2/CO = 0.5	220	42	0.5	575	160	IGCC applications			

Figure 4.1-1 LPMEOHTM Demonstration Test Plan Schedule of Tests

Task Name	Duration in Weeks	19 97	19 98	19 99	20 00	20 01	20 02
TASK 2.1.1 - PROCESS SHAKEDOWN AND CATALYST AGING	110.57 wk						
TEST 1 - INITIAL SHAKEDOWIN AND DESIGN PRODUCTION TESTS	5.43 wk						
TEST 2- GASSED SLURRY LEVEL	28.29 wk						
TEST 3 - REACTOR FEED: TEXACO-TYPE SYNGAS	0.71 wk		×				
TEST 5 - CHECK @ TEST 1 CONDITIONS	3.43 wk	×					
TEST 6 - CATALYST ADDITION AND AGING	32.57 wk						
TEST 7 - FREE-DRAIN ENTRAINED/CONDENSED OIL TO REACTOR	53.00 wk						
TEST 8 - OPERATION @ DESIGN FEED GAS RATES	2.00 wk		X				
TEST 9 - CHECK FOR LIMITATION ON CATALYST SLURRY CONCENTRATION	20.29 wk						
TEST 10 - CATALYST AGING TO REACH MAXIMUM PRODUCTIVITY	12.71 wk						
TASK 2.1.2 - PROCESS OPERATIONAL TEST PHASE	189.29 wk						
TEST 11 - CATALYST ADDITION/WITHDRAW/AL TEST	93.43 wk						
TEST 21 - REACTOR FEED: DESTEC-TYPE SYNGAS	3.00 wk				×		
TEST 2000-1: CATALYST ADDITION/WITHDRAWAL TESTS	64.86 wk						
TEST 2000-2: TURNDOWN AND RAMPING TESTS	1.00 wk					×	
TEST 2000-3: LOAD FOLLOWING AND ON/OFF TESTS	1.00 wk					×	
TEST 2000-5: REACTOR FEED COMPOSITION TEST - H2:CO RATIO =0.7	4.00 wk					×	
TEST 2000-6: FINAL TEMPERATURE RAMP OF METHANOL SYNTHESIS CATALYST	24.71 wk						
TEST 2000-7: REACTOR INSPECTION	3.00 wk						
TEST 2000-8: IN-SITU CATALYST ACTIVATION, TEMPERATURE PROGRAMMING	62.71 wk						
TEST 2000-9: MATERIAL BALANCE TEST - H2:CO RATIO =0.5	As Available					x	

Final Report: Volume 2 (June 2003)

Kingsport LPMEOHTM CCT Project

4-4

Startup Procedure:

The following procedure describes the startup of the LPMEOHTM reactor system. It is assumed that the reactor system has been properly purged with N_2 through the syngas loop and 29C-01 LPMEOHTM reactor. If this procedure is being implemented for the first time, or if the reactor was open to atmosphere for a significant period of time during a shutdown, a hot function test with syngas and mineral oil (in the absence of catalyst) to at least 200°C should be performed followed prior to the startup to minimize the production of metal carbonyls during normal operation. If this is a re-start procedure, it will not be necessary to perform this part of the procedure. The distillation system should be started on total recycle (see below).

The 29K-01 recycle compressor is started on total recycle (complete internal recycle operation). N_2 is introduced into the bottom of the reactor. The reduced catalyst slurry is then pressure transferred from the 29D-02 slurry tank to the LPMEOHTM reactor. If an in-situ activation was used to reduce the catalyst, or if the reactor was shutdown for only a brief period, the catalyst slurry would already be in the reactor.

The flow from the syngas recycle compressor is then introduced to the reactor by closing the recycle flow control valve. The increased gas flow from the recycle compressor will assist in heat transfer in the reactor during the heat-up phase. Once positive gas flow through the reactor has been observed on the flowmeter at the inlet to the reactor, the N_2 flow introduced at the bottom of reactor can be stopped.

Boiler feedwater flow to the 29C-02 steam drum (including the reactor tubes) is introduced to heat the catalyst slurry at a maximum rate of 30°C/hour. Startup steam, at 600 psig, is injected into the heat exchanger risers, as needed, to maintain this heat-up rate until the reaction initiation temperature (a maximum of 220°C) is reached.

Fresh feed gas is slowly started to the unit. The main valves upstream and downstream of the 29C-40 carbonyl guard bed are closed initially. A small manual valve is opened to slowly pressurize the carbonyl guard bed to avoid fluidization and damage of the adsorbent. Once the carbonyl guard bed has reached the syngas header pressure (nominally 750 psig), the main inlet and outlet valves to the carbonyl guard bed can be opened. The LPMEOHTM reactor, the carbonyl guard bed, and the syngas loop are then pressurized to syngas header pressure. Methanol product flow can then be taken to the distillation system (see below).

Major LPMEOH[™] Reactor Control Loops:

The reactor pressure is controlled using the 29PC-150 control loop which achieves a desired reactor pressure set-point by controlling the rate of flow of unreacted syngas through the purge gas valve, 29FV-157A. LPMEOHTM reactor bulk temperature is tightly controlled to the desired setpoint by using the 29TC-109 temperature control loop to adjust the steam drum pressure using control valve 29PV-109. The 29E-02 syngas feed / product economizer effluent temperature is controlled using the 29TC-101 control loop which achieves the desired economizer effluent

temperature set-point by controlling the rate of reactor feed flow through the 29TV-101 syngas feed / product economizer bypass valve.

Distillation System Startup Procedure:

The 29C-12 stabilizer feed drum, the 29C-10 stabilizer column, and the 29C-20 rectifier column are first purged with N_2 . The N_2 pressure is maintained until the syngas is started to the unit. Cooling water flow is then introduced to the 29E-11 stabilizer condenser and the 29E-22 stabilizer subcooler, and the fans on the 29E-21 rectifier air cooler are started. The stabilizer feed drum, stabilizer column, and rectifier column are filled with methanol from the methanol lot tanks (29D-20 or 29D-21). The stabilizer and rectifier columns are then heated using steam in their respective reboilers (29E-10 and 29E-20).

A recycle flow loop from the stabilizer feed drum through to the stabilizer and rectifier columns is then established. The methanol product takeoff from the 29C-21 methanol rectifier reflux drum and the underflow from the rectifier column are returned to the stabilizer feed drum until syngas flow is started to reactor. The rectifier column is set on total reflux during this time. Once methanol is available from the process at the 29C-03 high-pressure methanol separator, the distillation system is taken off of total recycle operation and the crude-grade methanol is sent to distillation capacity within the Eastman chemicals-from-coal complex for further processing. Raw methanol is then introduced from the high-pressure methanol separator into the stabilizer feed drum.

Batch-wise Slurry Withdrawal and Catalyst Activation Procedure:

The 29C-30 catalyst reduction vessel is designed to replace a portion (in 2,000 pound increments) of spent methanol catalyst slurry in the reactor with an equal amount of freshly activated catalyst. Spent catalyst slurry is pressure-transferred from the reactor to the catalyst reduction vessel through the control valve (29PV-424) at the discharge of the 29G-30 slurry transfer pump. The slurry is then cooled to less than 45°C using the 29E-33 utility oil cooler, at which temperature it can be safely drained into drums.

Mineral oil is charged to the vessel from the 29D-30 fresh oil tank. This mineral oil is preheated and catalyst is added. The slurry is agitated using the 29Y-30 agitator, and a reducing gas (consisting of a dilute mixture of CO Gas in N₂) is introduced at the bottom of the catalyst reduction vessel. The agitator is stopped once gas flow has been established. The catalyst is then activated using a series of temperature steps which are programmed into the distributed control system logic. The heat of reduction is released to the slurry and is removed via an external heat exchanger jacket on the catalyst reduction vessel. Bulk temperature control is accomplished by varying the temperature of utility oil, which flows through the heat exchanger jacket and which is controlled at the outlet of the 29E-32 utility oil heater. The operating pressure for the catalyst activation procedure is controlled at the vapor outlet of the 29C-31 catalyst reduction vessel separator. Once the catalyst reduction procedure has been completed, the reduced catalyst is pumped to the reactor using the slurry transfer pump.

In-situ Activation Procedure:

In-situ activation is the reduction of an entire charge (nominally 40,000 to 50,000 pounds) of fresh catalyst within the reactor vessel. Following the removal and drumming of the spent catalyst from the reactor, batches of fresh catalyst and mineral oil are mixed in the catalyst reduction vessel and pressure transferred (without reducing or activating the batches) into the reactor. During the loading procedure, N₂ is introduced into the bottom of the reactor to assist in mixing and heat transfer. Once a full charge of fresh catalyst is added to the reactor, the recycle compressor is started to further mix the catalyst slurry. The catalyst slurry is then activated using a series of temperature program steps. A dilute mixture of Balanced Gas in N₂ is used as the reducing gas and enters the reactor at the bottom of the vessel. Once positive gas flow from the recycle compressor through the reactor has been observed on the flowmeter at the inlet to the reactor, the N₂ flow introduced at the bottom of reactor can be stopped. The heat of reaction is released to the slurry and is removed via the internal heat exchanger. Bulk temperature control is accomplished by initially manually varying the stream drum pressure. Once the steam drum pressure exceeds the steam header pressure sufficiently, the temperature ramp can be automatically controlled through the distributed control system. The operating pressure for the in-situ activation is controlled using the 29PC-150 control loop which achieves a desired reactor pressure set-point by controlling the rate of flow of unreacted syngas through the purge gas valve, 29FV-157A. Once the in-situ catalyst reduction procedure has been completed, the reactor is started up normally using the procedure described above.

Stand-by Conditions Procedure:

Following a shutdown of the LPMEOHTM demonstration unit, a standby condition is used to maintain conditions for subsequent re-start. The reactor temperature is held at 180°C during this stand-by condition, using 600 psig startup steam, as required. The recycle compressor is generally kept in operation during stand-by to assist in heat transfer and to allow for gas flow to continue through the reactor system. Generally, a reactor pressure of 95 psig is maintained with the recycle compressor in operation. N₂ makeup can be introduced to the system once the N₂ header pressure exceeds the compressor suction pressure of approximately 75 psig. The distillation system is placed in recycle operation. If the recycle compressor is not available, N₂ is introduced into the bottom of the reactor.

Shutdown Procedure:

To shutdown the LPMEOH[™] demonstration unit, the fresh feed gas valves controlling the flow of Balanced Gas and CO Gas are closed. If this is expected to be a short-term shutdown, the stand-by conditions are established (see above).

If this is a longer-term shutdown, for instance a maintenance inspection, the catalyst slurry in the LPMEOHTM reactor is pressure transferred to the slurry tank. The reactor can be depressurized very quickly to 95 psig. Once this pressure is reached, the recycle compressor can be put into internal recycle operation and subsequently shutdown, if required. N₂ can then be introduced into the bottom of the reactor to maintain mixing of the slurry while it is cooling. The slurry is

cooled in the reactor at a rate of 30-35°C per hour to a final temperature of 150°C. The slurry is then transferred either to the slurry tank for storage or to the catalyst activation vessel for further cooling using the utility oil cooler and subsequent draining into drums.

If the catalyst is to be reused after the shutdown, the catalyst will be pressure transferred from the slurry tank back to the reactor at the conclusion of the maintenance outage. The 29Y-02 agitator is used to maintain the slurry in suspension, and a steam jacket is used to minimize heat losses from the slurry tank. If the catalyst is to be replaced with freshly reduced catalyst, batches of spent catalyst are then pressure transferred with N_2 from the slurry tank to the catalyst reduction vessel for cooling to less than 45°C and drumming. Alternatively, these batches of spent catalyst can be directly transferred from the reactor to the catalyst reduction vessel for cooling and drumming, if desired.

4.2.1 Instrumentation and Data Acquisition

Within the LPMEOH[™] Process, measurement of the flow and pressure of the catalyst slurry or the three-phase reactor fluid was required in order to assess the performance of the technology at the commercial demonstration of 260 sT/D. Transmitters with diaphragm seal(s) and oil flush connections were used to provide accurate readings in this difficult environment. In addition, nuclear density gauges were used on the LPMEOH[™] reactor (to determine the expanded slurry height) and the 29C-30 catalyst preparation vessel (to measure the three-phase density and expanded slurry height). No other specialty instrumentation systems were required.

The LPMEOHTM demonstration unit utilized an existing computer network that controlled operating plants at the Eastman chemicals-from-coal complex. A distributed control system was used for automatic control, monitoring, and non-critical process and equipment shutdowns. A data acquisition was tied into the data highway for the control system so that real-time plant data could be accessed both at a office trailer (separate from the operating control room) at the LPMEOHTM demonstration unit as well as at Air Products' corporate headquarters in Allentown, PA. A dedicated programmable logic controller was used so that critical process shutdowns could be initiated independent of the control system hardware.

4.2.2 Test Methods

Appendix A, Table A-1 provides the streams, component lists and frequency of sampling for the routine analytical measurements. Appendix A also contains a description of the sampling and analytical methods used to quantify the concentration of arsenic (reported as arsine) and iron carbonyl in the fresh feed gas to the LPMEOHTM demonstration unit. The simplified process flow diagram, Figure 2.4-1, shows the measurement points.

4.3 ANALYSIS OF FEEDSTOCKS, PRODUCTS AND REAGENTS

In order to perform the routine material balance calculations, gas chromatographs were used to analyze the syngas feed streams, the streams entering and exiting the LPMEOHTM reactor, and the purge streams leaving both the main reactor loop and the distillation section. Analysis of the two methanol streams (Crude-Grade Methanol and Refined-Grade Methanol) was also

performed. Appendix B contains samples of material balance calculation sheets that include the typical analysis of these streams for two operating scenarios: Appendices B-1 and B-2 provides data when Balanced Gas is fed to the LPMEOHTM demonstration unit, and Appendices B-3 and B-4 provide data during operation when CO-rich syngas is fed to the LPMEOHTM reactor.

In addition, syngas sampling was performed to determine the levels of trace components within the various syngas streams that may be poisons to methanol synthesis catalysts. A summary of these results is provided in Section 5.2.

Throughout the operating program, samples were taken of catalyst slurry from the LPMEOH[™] reactor on a routine basis to analyze for the presence of known catalyst poisons and to track some of the key physical properties of the methanol synthesis catalyst. Section 5.2 contains the results of these analyses for the four different catalyst campaigns that were completed during the performance period.

4.4 DATA ANALYSIS METHODOLOGY

Appendix C, Table 1 contains the major field instrumentation signals, analytical measurements, and calculations which were used in determining the material and energy balance and LPMEOHTM reactor performance. The simplified process flow diagram, Figure 2.4-1, also shows the measurement points. Two sets of material balances were performed:

- Across the LPMEOH[™] reactor, using the Reactor Feed Flowrate (Stream 109) and Reactor Effluent Flowrate (Stream 120); and
- Around the LPMEOHTM demonstration unit, using the major plant feeds and product streams as noted in Appendix C, Table 1.

Part of the process of data reduction is the elimination of those operating periods when one or more key inputs to the material balance calculation is invalid (for example, a failure of a key process flowmeter or an upset of the analytical system). If the material balance cannot be closed due to such an instrument failure, a portion of a data period will be eliminated from the material balance calculation. This analysis will be performed at the spreadsheet level, so that the historian files will always contain the raw data from field instruments and analytical equipment.

A comprehensive set of the formulas used to calculate the parameters listed in Appendix C, Table 1 is provided in Appendix C-2. These calculations were based upon actual measurements from the LPMEOH[™] demonstration unit.

4.5 DATA SUMMARY

The operating program for the LPMEOH[™] demonstration unit can be broken into four operating campaigns based upon the use of a fresh charge (full or partial) of methanol synthesis catalyst in the LPMEOH[™] reactor. Kingsport Catalyst Campaign 1 was conducted between April of 1997 and November of 1997. This campaign was concluded on 03 November 1997 after 171 days of operation. Kingsport Catalyst Campaign 2 was conducted between December of 1997 and

August of 2001. The campaign was concluded on 06 August 2001 after 1,325 days of operation. Kingsport Catalyst Campaign 3 was conducted between August of 2001 and June of 2002. The campaign was concluded on 04 June 2002 after 284 days of operation. The fourth Kingsport Catalyst Campaign began on 28 June 2002 and included process and analytical data until the end of the operating program on 31 December 2002 (a total of 187 days of operation).

Appendix D, Tables D-1 through D-4 provide the results from the operation of the LPMEOH[™] demonstration unit during the 69-month operating phase. These data represent daily averages, typically from a 24-hour material balance period, and those days with less than 12 hours of stable operation are omitted. Since startup, about 103.9 million gallons of methanol was produced. Eastman accepted all of the available methanol for use in the production of methyl acetate, and ultimately cellulose acetate and acetic acid. The analysis of the key results is provided in Section 5.

4.6 **OPERABILITY AND RELIABILITY**

Table 4.6-1 contains the availability of the LPMEOH[™] demonstration unit during the 69-month demonstration test program (availability is defined as the percentage of time that the LPMEOH[™] demonstration unit was available to operate, with the exclusion of scheduled outages). The overall availability was 97.5%.

The causes of unplanned outages are also provided in Table 4.6-1. Startup was completed in two days, and nameplate production of 80,000 gallons per day (260 sT/D) was achieved within four days, and production rates exceeding 115% of design were achieved within six days.

Unexpected poisoning of the catalyst due to the presence of iron contamination in 1997 led to the

Operating Year ^a	1997	1998	1999	2000	2001	2002 ^b
Availability (%)	94.9	99.54	99.4	96.3	97.7	94.1
Total Hours of	325.6	45.1	52.8	311.4	166.3	198.9
Unplanned Outages						
Catalyst	149.3	0.0	0.0	0.0	38.7	0.0
Carbonyl Guard Bed	0.0	0.0	0.0	118.9	120.6	198.9
Electrical	20.7	24.6	0.0	42.3	0.0	0.0
Instrumentation	12.8	1.3	0.0	4.7	7.0	0.0
Mechanical	142.8	19.2	52.8	145.5	0.0	0.0

 Table 4.6-1

 LPMEOHTM Demonstration Unit Availability and Causes of Unplanned Outages

a. Operating year defined as April-March.

b. Through 31 December 2002.

only unexpected outage related to catalyst performance. Given the very high level of catalyst poisoning present, the team felt that the full replacement of the catalyst charge, and the outage that would result, was necessary to achieve overall program goals. During operation of the system, it was found that some of the oil return lines to the reactor were plugging and one heat exchanger developed a leak at a flange connection. Redesign and replacement were undertaken in 1997, creating a permanent solution.

In 1998, non-routine maintenance related to instrumentation tubing around the recycle compressor caused a portion of the reported unplanned outages. Also, electrical transients resulted in two additional short outages. The longest continuous operating run without interruption of any kind was 94 days (July-October of 1998); an earlier campaign of 65 days of continuous operation was completed in April of 1998.

In 1999, an unexpected leak developed in the boiler feed water piping system and two heat exchangers, causing the need to make repairs.

In 2000, a leak in the tubing associated with a flow transmitter in the boiler feed water system and an unrelated failure of a solenoid valve controlling boiler water feed to the steam drum resulted in downtime. An electrical outage outside of the LPMEOHTM demonstration unit caused the failure of rupture discs for the cooling water system.

As noted in Section 5.2, plant activities associated with the changeout of the adsorbent in the 29C-40 carbonyl guard bed began. The philosophy associated with the design of this equipment was that the adsorbent would not need to be changed during the demonstration test period; as a consequence, the ability to isolate the vessel from the remainder of the LPMEOHTM demonstration unit for maintenance was not included in the engineering design. In future designs, the required isolation valves would be installed, so that the outages that were taken in 2000 through 2002 would be eliminated.

In 2001, a controls signal error caused the unit to drop off line for a very short period of time. Also, operation could not be continued during a catalyst withdrawal/addition operation; the LPMEOHTM reactor was restarted following the addition of a batch (2,000 pounds) of fresh catalyst from the catalyst preparation vessel.

A monthly production record of 2,500,000 gallons of methanol was reached during October of 2002. For a 23-day period, production of nameplate capacity (80,000 gallons per day) or higher was achieved. The October monthly average production was 101% of nameplate capacity. Production rates as high as 117% of nameplate capacity were also achieved for shorter periods.

In addition to producing syngas for the LPMEOH[™] demonstration unit, the Eastman's coal gasification facility produces raw material for a gas-phase methanol production plant and a carbon monoxide facility. The fluctuating needs of the impurities removal facility causes some variance in the quantity of syngas produced. Given the ability of the LPMEOH[™] demonstration unit to tolerate changes in the composition and flow of syngas, this facility is used as the swing facility to manage changes in syngas production. Through all of this fluctuation, the

LPMEOHTM demonstration unit produced methanol to its full projected capacity based on the composition and flow of syngas available.

4.7 SECTION 4 BIBLIOGRAPHY

a. "Commercial Demonstration of the Liquid Phase Methanol (LPMEOH™) Process – Demonstration Test Plan", Prepared by Air Products Liquid Phase Conversion Co., L.P., DOE Cooperative Agreement No. DE-FC22-95PC93052, September 1996.

5.0 TECHNICAL PERFORMANCE

5.1 CATALYST PERFORMANCE

As indicated in Section 4.5, there were four catalyst campaigns which characterize the operating program of the LPMEOHTM demonstration unit. A comparison of the catalyst campaigns is given in Table 5.1-1. The first campaign involved batch-wise catalyst activation along with the addition of fresh slurry to maintain catalyst productivity. The second campaign involved batch-wise catalyst activation along with withdrawal of spent catalyst slurry and addition of fresh slurry to maintain catalyst productivity. The third and fourth campaigns used an in-situ activation procedure to prepare a full charge of methanol synthesis catalyst. Temperature programming was used in the third campaign to maintain productivity. During the fourth catalyst campaign, productivity was essentially maintained by virtue of the very low catalyst deactivation rates and, therefore, temperature programming was not used during the period of performance. Methanol synthesis catalyst (baseline and alternative) from two different suppliers was used during the operating program.

As shown in Figures 1.2.3-1 and 2.4-1, two catalyst guard beds were available to limit the levels of trace contaminants entering the LPMEOHTM demonstration unit. The long-term performance of the methanol synthesis catalyst was adversely impacted by the presence of species such as arsenic (detected in the Balanced Gas stream) and iron (apparently generated during certain operating periods within the LPMEOHTM demonstration unit). A review of the work on the adsorbents used during the period of performance of the LPMEOHTM Demonstration Project is provided in Section 5.2.

Kingsport				Catalyst	Starting Catalyst		Productivity	Operating
Campaign	Run Start	Run End	Days	Туре	Charge	Activation	Control	Temperature
#	Date	Date	Onstream	Used	(lbs)	Method	Method	(°C)
1	4/6/97	11/3/97	171	Baseline	20,300	Batch	Catalyst Addition	250 to 260
							Only	
2	12/20/97	8/6/01	1,325	Baseline + Alternative	19,500	Batch	Catalyst Withdrawal / Addition	235 ^a
3	8/24/01	6/4/02	284	Alternative	41,580	In-situ	Temperature Programming	216 to 242
4	6/28/02	Ongoing ^b	Ongoing ^b	Alternative	40,040	In-situ	-	215

 Table 5.1-1

 Kingsport LPMEOH[™] Operating Program Campaigns

a. Primarily 235°C with brief operation at other temperatures.

b. The LPMEOH[™] demonstration unit operating program ended 12/31/02 with 187 days onstream in Catalyst Campaign 4.

5.1.1 Catalyst Performance: Catalyst Campaign 1

An in-house kinetic model was used to assess the performance of the methanol synthesis catalyst and to eliminate the effects of varying feed composition or operating conditions. This model can be used to study changes in catalyst performance over stable periods of operation (typically two weeks or more); results over shorter durations must be considered in light of the influence of operating conditions (for example, variations in the bulk syngas composition). The "age" of the catalyst can be expressed in terms of a dimensionless variable eta (η), which is defined as the ratio of the rate constant at any time to the rate constant for freshly reduced catalyst (as determined in the laboratory autoclave). Figure 5.1.1-1 shows performance results from the LPMEOHTM reactor during Catalyst Campaign 1. Typical exponential decay appeared as a straight line on a semi-log-plot, as shown. The curve fit to data from a 4-month test at the LaPorte AFDU in 1988/89 (this run was performed on CO-rich syngas derived from natural gas at a reactor temperature of 250°C) and laboratory autoclave data from 1996 are included for reference. During this catalyst campaign, the LPMEOHTM reactor operated at 250°C, except for brief operation at 260°C at the very end of the campaign. Catalyst additions are noted on the figure.

A partial charge (20,300 pounds out of a design of 40,000 pounds) of fresh catalyst was activated in 9 batches within the 29C-30 catalyst preparation vessel and added to the LPMEOHTM reactor



Figure 5.1.1-1 Kingsport LPMEOHTM Catalyst Life (Catalyst Campaign 1)

in early April of 1997. The initial performance of the catalyst was equivalent to the results from the laboratory autoclave, indicating that the catalyst activation was successful. The highest methanol production rate over a 24-hour period occurred on 19 April 1997 (89,900 gallons per day, or 292.2 sT/D); for shorter balance periods (approximately 12 hours), methanol production rates of 92,900 to 94,500 gallons per day (302 to 307 sT/D) were measured.

Over the next several weeks, the catalyst life data indicated that an accelerated negative change in performance occurred. During the April/May of 1997 operating period, it was not immediately apparent whether this decline in η was a result of a decline in catalyst activity or hydrodynamic effects related to an increase in pressure drop across the gas sparger at the inlet to the LPMEOHTM reactor. However, upon restarting the LPMEOHTM demonstration unit following a planned outage in June of 1997, the value of η was determined to be unaffected by the performance of the sparger. Therefore, it was concluded that the catalyst activity was declining more rapidly than expected. The catalyst deactivation rate for the first 30 days of operation was 3.4% per day; the higher catalyst deactivation during this period was later linked to high concentrations of iron on the reactor catalyst samples. These higher iron concentrations were most likely due to construction debris (refer to Section 5.2 for additional information).

Following the initial operating period, the catalyst performance continued to decline more rapidly than expected; however, as noted on Figure 5.1.1-1, the rate of change was slower than during the April/May operation. During the remaining period of operation during Catalyst Campaign 1 from June through November of 1997, deactivation rates were higher than expected based on experience from the LaPorte AFDU (catalyst deactivation rate of 0.4% per day). In fact, the catalyst life was similar to the baseline deactivation results obtained during laboratory autoclave testing (catalyst deactivation rate of 1.0% per day). Eliminating the catalyst performance results for the first month due to catalyst hyperactivity and the presence of iron construction debris, the average deactivation rate for the entire campaign was 1.3% per day.

Fresh catalyst was activated and added to the LPMEOHTM reactor on six occasions during this first catalyst campaign to maintain catalyst productivity and increase catalyst loading. On 04 July 1997, the first catalyst batch of 2,300 pounds of fresh catalyst activated and added to the LPMEOHTM reactor. This same quantity of catalyst was activated and added to the reactor on 11 July, 23 July, 12 September, 17 September, and 03 October 1997.

Based upon the results of unit operation and catalyst performance, DOE accepted a recommendation by Air Products and Eastman in October of 1997 to drain the initial charge of catalyst from the reactor and replace the charge with fresh catalyst. Prior to this catalyst turnaround, the operating temperature of the LPMEOHTM reactor was raised from 250°C to 260°C for about one week of operation.

5.1.2 Catalyst Performance: Catalyst Campaign 2

Catalyst Campaign 2 was the longest of the operating program at 1,325 days. During most of this catalyst campaign, the operating temperature in the LPMEOHTM reactor was lowered to 235°C. By using the lowest possible temperature necessary to achieve the required methanol production rate, the life of the methanol synthesis catalyst was potentially extended. Catalyst life data during the first portion of Catalyst Campaign 2 (covering operations in 1997 and 1998) is shown in Figure 5.1.2-1. At the start of this campaign in December of 1997, only a partial charge (19,500 pounds) of fresh catalyst was activated in the catalyst preparation vessel and added to the reactor system; this limited the amount of material exposed to poisons at the outset. During this first portion of the campaign, fresh catalyst was activated and added to the LPMEOHTM reactor on a periodic basis (as noted on the figure) to achieve the desired reactor performance and to determine if hydrodynamic limitations could be reached at high slurry concentrations (the design value was 40 weight % [wt%]). The rate of catalyst deactivation for periods of stable operations (typically 2 weeks or longer) are indicated on the figure.

At a reactor temperature of 235°C, the decrease in η with time from the starting date of 20 December 1997 through Day 38 was calculated to be 1.0% per day, which represented a significant improvement over the 3.4% per day decline measured during the initial six weeks of operation of Catalyst Campaign 1. The deactivation rate also improved from the longer-term rate of 1.6% per day calculated throughout the summer and autumn of 1997.



Figure 5.1.2-1 Kingsport LPMEOH[™] Catalyst Life (Catalyst Campaign 2: 1997-1998)

Based on this improvement, DOE accepted a recommendation by Air Products and Eastman to further decrease the reactor temperature to 225° C. The initial operation at this temperature (from Day 42 through Day 60) showed a modest improvement in the deactivation rate to 0.7% per day. During two additional stable operating periods between Day 61 and Day 101, deactivation rates of 0.27% and 0.36% per day were calculated. These results compare favorably with the performance from the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89.

The reactor temperature was then lowered further to 220°C between Day 104 and Day 135. During a one-month test at a reactor temperature of 220°C and a Balanced Gas flowrate of 550-600 KSCFH, the rate of decline of η was 0.4% per day, which matched the performance at 225°C. Beginning on Day 139, the temperature of the LPMEOHTM reactor was increased to 235°C, which was the operating temperature after the December 1997 restart. After the catalyst addition on Day 182, the weight of catalyst in the LPMEOHTM reactor had reached 80% of the design value of 40,000 pounds.

The LPMEOHTM demonstration unit completed a 24-day stable period of operation on Day 206 at a reactor temperature of 235°C and a flowrate of Balanced Gas of approximately 700 KSCFH. Over this period, the rate of decline of η was 0.2% per day. This performance was superior to the original target from the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89 (0.4% per day).

On Day 244, a test at a reactor temperature of 250°C was initiated. The test at the higher reactor temperature was completed on Day 254 when accelerated changes in the calculated value for η were observed. From Day 261 to Day 267, three batches of fresh catalyst were activated and added to the LPMEOHTM reactor, which brought the catalyst loading to slightly above the design value of 40,000 pounds.

Six additional batches of fresh catalyst were activated and added to the LPMEOH[™] reactor between Day 285 and Day 376. At this point, the catalyst loading in the LPMEOH[™] reactor had been increased to about 140% of design. No hydrodynamic instability was detected as the catalyst slurry concentration in the reactor ranged from 46 to 48 wt%.

For the first year of operation under Catalyst Campaign 2, the average rate of catalyst deactivation was calculated to be 0.65% per day. This result was a significant improvement over the catalyst performance during Catalyst Campaign 1 (1.3% per day).

Catalyst life data during the second portion of Catalyst Campaign 2 (covering operations in 1999) is shown in Figure 5.1.2-2. The use of catalyst withdrawal and addition (as noted on the figure) was first practiced during this portion of the campaign in order to achieve the desired performance of the LPMEOHTM reactor.

Figure 5.1.2-2 Kingsport LPMEOH[™] Catalyst Life (Catalyst Campaign 2: 1999)



Between Day 377 and Day 466, the LPMEOHTM demonstration unit continued to perform well with high slurry concentrations in the LPMEOHTM reactor. Slurry concentrations greater than 48 wt% were achieved during this period with no apparent hydrodynamic instability. Three batches of fresh catalyst were activated and added during this time, bringing the catalyst loading to 151% of design, which was the maximum value achieved during the entire operating program.

Starting on Day 424, three separate transfers of catalyst slurry from the reactor were conducted in order to dilute the slurry concentration in anticipation of an upcoming outage in March of 1999.

These steps reduced the calculated slurry concentration in the reactor from in excess of 48 wt% to approximately 40 wt%. Reactor conditions were held at 235°C except for a brief production test, between Days 431 through Day 435, when reactor temperature was increased to 240°C. The

LPMEOHTM demonstration unit was shutdown between Day 437 through Day 449 in order to complete the planned outage. A second planned Eastman complex outage was taken between Day 483 and Day 494.

The average catalyst deactivation rate for calendar year 1999 was calculated to be 0.58% per day. This result was equivalent to the performance in 1998 and was slightly greater than the baseline

deactivation rate from the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89 (0.4% per day).

The catalyst life performance during the final portion of Catalyst Campaign 2 is provided in Figure 5.1.2-3. Catalyst withdrawals and additions are indicated as dotted vertical lines in the figure. A schedule for catalyst withdrawal and addition was developed that allowed for the desired methanol production rate to be maintained with minimum disruption to the operations staff. The overall schedule for catalyst withdrawals and additions was generally undertaken every two months, and the typical amount of catalyst that was exchanged was about 11,000 pounds.

A comparison of the performance of the methanol synthesis catalyst with different syngas compositions was able to be made. Between Day 825 and Day 844, during which a feed gas with a H₂/CO ratio of 1:1 was fed to the LPMEOHTM reactor, a catalyst deactivation rate of 1.24% per day was calculated. Shortly after that test, a catalyst deactivation rate of 0.89% per day was calculated for the period from Day 875 to Day 903, during which Balanced Gas alone was used as feed to the LPMEOHTM demonstration unit. An analysis of these results determined that the two data sets were statistically similar, given the scatter in the calculated values for the catalyst rate constant.





The catalyst performance was also used during this operating period to gauge the effectiveness of the 29C-40 carbonyl guard bed. As described in Section 5.2, the carbonyl guard bed was charged with manganese dioxide (targeting arsenic) and activated carbon (targeting iron and nickel) during this time. When the carbonyl guard bed was in service with aged adsorbents (approximately one year of service), a catalyst deactivation rate of 0.85% per day was calculated over a 12-day period (Days 917 through 928). When the carbonyl guard bed was bypassed over a 33-day period from Day 969 to Day 1002, a catalyst deactivation rate of 0.82% per day was calculated. These results were statistically the same and indicated that the aged adsorbents in the carbonyl guard bed was bypassed for the remainder of Catalyst Campaign 2. The results for catalyst deactivation rates shown on Figure 5.1.2-3 varied from 0.4% to 0.7% per day over this operating period, which could indicate that the levels of trace contaminants entering the LPMEOHTM demonstration unit were likely varying in concentration during this time.

The average rate of catalyst deactivation during the final portion of Catalyst Campaign 2 was calculated to be 0.64% per day. For the entire campaign, the catalyst deactivation rate was determined to be 0.63% per day, which was greater than the value calculated from the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89 (0.4% per day change in catalyst activity).

5.1.3 Catalyst Performance: Catalyst Campaign 3

Catalyst Campaign 3 involved a demonstration of in-situ activation procedure. A full charge (greater than 40,000 pounds) of fresh methanol synthesis catalyst was placed into the LPMEOH[™] reactor and activated at one time. This was in contrast to the batch-wise activation of smaller catalyst quantities (approximately 2,000 pounds each) used earlier in the demonstration.

As shown in Figure 5.1.3-1, the initial activity of the methanol synthesis catalyst for Catalyst Campaign 3 ($\eta \approx 0.7$) was less than expected when compared with data from the laboratory ($\eta = 1.0$). Information on the analysis of these initial performance results is provided in Section 5.6. Following the completion of the in-situ catalyst activation procedure, conditions were selected to deal with the hyperactivity that typically occurs when fresh methanol synthesis catalyst is initially exposed to syngas. During the first few weeks of operation, conditions were varying between a reactor pressure of 550-600 psig, reactor temperature of 212-220°C, and flowrate of Balanced

Gas of 500-525 KSCFH. Once the test had proceeded beyond the period of hyperactivity, operating conditions were set at 218°C reactor temperature, 685 psig reactor pressure, and 600-650 KSCFH of Balanced Gas.

Figure 5.1.3-1 Kingsport LPMEOHTM Catalyst Life (Catalyst Campaign 3)



As noted in Table 5.1-1, temperature programming (as opposed to catalyst withdrawal and addition) was used as a productivity control method during Catalyst Campaign 3. The reactor temperature was gradually increased as necessary to control reactor purge flowrate and maintain catalyst productivity as the operating campaign progressed. Temperature changes were managed such that sufficient performance data could be achieved at each temperature (minimum of two weeks of operation at each condition to determine catalyst life performance data). Over the entire period of Catalyst Campaign 3, the reactor temperature ranged from 216°C to 242°C.

The rates of catalyst deactivation that were computed during this period are indicated in Figure 5.1.3-1. Note that, for two performance periods during the first 100 days onstream, the calculated rate of catalyst deactivation was significantly lower than the average results during Catalyst Campaign 2 (0.63% per day), and was equivalent to or better than the results from the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89 (0.4% per day). Following the two periods of improved catalyst performance, the catalyst deactivation rates for Catalyst Campaign 3 approached historical values of 0.6 to 0.7% per day.

The first operating test using methanol synthesis catalyst activated in-situ (within the LPMEOHTM reactor as opposed to the 29C-30 catalyst preparation vessel) and temperature programming was completed on Day 213 when a catalyst withdrawal and addition campaign was undertaken to increase catalyst activity. A series of three withdrawals was conducted on Day 213 to Day 214. Four batches of fresh catalyst were activated and added to the reactor between

Day 215 and Day 220. At the conclusion of the catalyst withdrawal/addition catalyst campaign, the LPMEOH[™] demonstration unit was operated at a constant reactor temperature of 235°C and a reactor pressure of 700 psig.

5.1.4 Catalyst Performance: Catalyst Campaign 4

Based on the lower than expected activity achieved during the first in-situ activation, a decision was made to perform a second in-situ activation using a revised procedure. As detailed in Section 5.6.3, this revised procedure was developed to eliminate the step that resulted in the storage of the fresh slurry at elevated temperatures prior to activation.

The second in-situ activation was undertaken in late June of 2002. Performance results for the methanol synthesis catalyst following the second in-situ activation are presented in Figure 5.1.4-1. The initial catalyst activity was excellent, and matched the performance expected with properly activated catalyst from the laboratory autoclave. The initial operating conditions (reactor temperature of 214°C, reactor pressure of 450 psig, Balanced Gas flowrate of 500-600 KSCFH) were selected to deal with the hyperactivity that is typical of freshly activated methanol



Figure 5.1.4-1 Kingsport LPMEOH[™] Catalyst Life (Catalyst Campaign 4)

synthesis catalysts. Including the initial period of hyperactivity (approximately 3 weeks) that is typical of methanol synthesis catalysts, the catalyst deactivation rate over the first 52 days of operation was 0.52% per day; if the hyperactivity period is excluded, the rate of change of η over the first six months of this catalyst campaign was calculated to be in the range of 0.1 to 0.2% per day. These results were a significant improvement over the historical average of 0.6 to 0.7% per day and the results from the proof-of-concept run at the LaPorte AFDU in 1988/89 (0.4% per day), and may have been related to the performance of the adsorbent in the carbonyl guard bed, the removal of carbon steel components from the reactor internals (refer to Section 5.2), and the use of temperature programming to control the production rate across the LPMEOHTM reactor. Due to the low rate of catalyst deactivation, the reactor temperature was not increased during the 6-month operating period for this campaign (the operating program of the LPMEOHTM Demonstration Project ended on 31 December 2002). Reactor pressure was raised from 450 psig to 500 psig on 11 November 2002 to improve the conversion efficiency of the process.

The change in the rate of catalyst deactivation at Day 140 may have been caused by a breakthrough in arsenic (reported as arsine) through the adsorbent in the carbonyl guard bed. A thermal treatment of the adsorbent in the carbonyl guard bed (described in Section 5.2) may have resulted in the improvement in performance beginning on Day 160, and provides evidence of the impact of the presence of trace contaminants in coal-derived syngas on the life of methanol synthesis catalyst.

For the entire campaign (including the period of hyperactivity), the average rate of catalyst deactivation was calculated to be 0.17% per day, which was significantly lower than performance results from the laboratory, the LaPorte AFDU, or the earlier portions of the operating program for the LPMEOHTM Demonstration Project.

5.1.5 <u>Summary</u>

Catalyst performance was tracked throughout the 69-month operating program at the LPMEOHTM demonstration unit. The design for the project used the performance during the 4-month proof-of-concept run at the LaPorte AFDU (0.4% per day on CO-rich syngas derived from natural gas at a reactor temperature of 250°C) as the basis. One of the key objectives of the commercial demonstration was to determine the long-term performance of the methanol synthesis catalyst on coal-derived syngas, and to quantify the rate of catalyst deactivation for use in any future projects. An in-house kinetic model was used to calculate the ratio of the rate constant at any time to the rate constant for freshly reduced catalyst (as determined in the laboratory autoclave). The rate of change of this parameter (η) improved over the demonstration program. Following an accelerated rate of catalyst deactivation that was linked to the presence of post-construction debris in the system, the initial results at the design operating temperature of 250°C (averaging 1.3% per day) were significantly higher than the design basis.

The project team and DOE agreed that a second catalyst campaign be undertaken using a lower reactor temperature. During this 3+ year operating test, a long-term program of catalyst withdrawal and addition was demonstrated as an effective means to maintain catalyst
productivity. Most of the data during Catalyst Campaign 2 was taken at a reactor temperature of 235°C, and the rate of catalyst deactivation improved to 0.6% per day.

During the execution of the operating program for the LPMEOHTM Demonstration Project, modes of operation that were not anticipated during the development of the original Demonstration Test Plan were identified. The ability to activate the methanol synthesis catalyst within the LPMEOHTM reactor by using syngas that is diluted with N₂ (so-called in-situ activation) can result in lower capital cost. Also, the life of the methanol synthesis catalyst can be extended by operating the LPMEOH[™] reactor at low temperatures at the outset and slowly increasing temperature with time to maintain a constant production of methanol (so-called temperature programming). The third catalyst campaign used this in-situ activation procedure to activate a full charge (in excess of 40,000 pounds) of fresh catalyst slurry inside the reactor. The catalyst did not reach proper initial activity compared to the results from the laboratory autoclave. The selection of operating conditions not associated with the activation was identified as the cause of the lower initial catalyst activity (refer to Section 5.6). Temperature programming was first practiced during Catalyst Campaign 3 to minimize the operating temperature and increase as necessary to maintain catalyst productivity. Levels of iron, arsenic, and sulfur were increasing on the catalyst during this campaign. The average catalyst deactivation rate for Catalyst Campaign 3 was similar to the result for Catalyst Campaign 2 (0.6% per day).

Catalyst Campaign 4 also involved an in-situ activation with an improved procedure to avoid high temperature storage of fresh catalyst slurry prior to activation. The activation resulted in the proper initial activity of the methanol synthesis catalyst. The average rate of catalyst deactivation during the 6-month duration of this campaign was 0.17% per day, which was ½ of the design basis for the LPMEOHTM demonstration unit. The reactor was operated at a constant low temperature of 215°C during this campaign.

The improvement in the long-term performance of the methanol synthesis catalyst can likely be attributed to several factors that were investigated as part of the operating program for the LPMEOHTM Demonstration Project. The negative effects of certain trace contaminants in coalderived syngas may have had the most significant impact on catalyst life; Section 5.2 provides details on the results of analyses of catalyst samples from the LPMEOHTM reactor, laboratory work on trace contaminant removal, and the change of adsorbent materials within the existing catalyst guard bed system at the LPMEOHTM demonstration unit. The ability to maintain the desired rate of methanol production using the concept of temperature programming was likely another factor, and the development of the in-situ activation procedure allowed for the minimum operating temperature to be utilized; Section 5.6 provides additional details on the work in this area.

5.2 <u>IDENTIFICATION AND REMOVAL OF TRACE CONTAMINANTS IN COAL</u> <u>DERIVED SYNGAS</u>

One of the major technical risks associated with the design of the LPMEOH[™] demonstration unit was the magnitude of the impact of trace contaminants in coal-derived syngas on the longterm performance of the methanol synthesis catalyst. Catalyst consumption is a major component in the variable cost of operation of methanol production plants, and the results from the 4-month proof-of-concept test at the LaPorte AFDU were obtained using CO-rich syngas produced from natural gas. During the operation of the LPMEOH[™] demonstration unit, attempts were made to determine the impact of trace contaminants on the performance of the methanol synthesis catalyst. The presence of certain elements was identified from samples of spent catalyst, spent adsorbent materials from catalyst guard beds, and syngas entering the LPMEOH[™] demonstration unit. The effects of certain elements were identified in laboratory studies, and the adsorbents in the catalyst guard bed system were changed based upon these results.^(a)

A description of the Eastman upstream gas cleanup systems was provided in a topical report.^(b) These systems include a Rectisol gas clean-up plant located downstream of the Texaco coal gasification system and a catalyst guard bed (equipment number 10C-30), which was installed by Eastman upstream of the fixed-bed methanol plant. Figure 1.2.3-1 shows the location of this equipment in relation to the syngas generation system and to both the fixed-bed methanol plant and the LPMEOHTM demonstration unit. The Rectisol unit was primarily designed to remove sulfur compounds from the syngas stream. The Eastman catalyst guard bed was installed to remove trace contaminants (in particular, compounds of sulfur and arsenic) which adversely impact the long-term performance of methanol synthesis catalysts.

The 29C-40 carbonyl guard bed (as shown in Figure 2.4-1) is a fixed adsorbent guard bed that protects methanol synthesis catalyst in the LPMEOHTM reactor against possible upsets of contaminants from the upstream gas cleanup units. The unit is four feet in diameter with an active bed length of 12 feet. Normal process flow is downward; the design also included a thermal regeneration system to reclaim performance of the adsorbent following saturation with metal carbonyls.

During the design phase of the LPMEOH[™] Demonstration Project, the basis for the rate of catalyst deactivation (0.4% per day) was selected based upon the results from the 4-month test at the LaPorte AFDU. The experiences at the pilot unit had identified iron and nickel carbonyl as significant poisons to methanol synthesis catalyst. In addition, a field test was performed at the Kingsport chemicals-from-coal complex in 1996 using autoclaves housed in a transportable laboratory, the Alternative Fuels Field Development Unit (AFFTU); a photograph is provided in Figure 5.2.1-1. As documented in a topical report,^(c) the performance of the methanol synthesis catalyst in the autoclave was not impacted by the presence of trace quantities of iron, nickel, sulfur, arsenic (expected to be in the form of arsine [AsH₃]), and chloride that were detected on spent catalyst, even with no catalyst guard adsorbent materials in service.

Based upon these results, the basis for the design of the equipment to protect the methanol synthesis catalyst from trace contaminants in the coal-derived syngas was to utilize the existing

10C-30 catalyst guard bed within the Eastman chemicals-from-coal complex, which was charged with zinc oxide (targeting sulfur) and manganese dioxide on alumina (targeting arsenic), and to install the 29C-40 carbonyl guard bed utilizing activated carbon for removal of iron and nickel carbonyl. Both vessels were designed to operate at system pressure (nominally 750 psig) and ambient temperature.



Figure 5.2-1 Photograph of Alternative Fuels Field Test Unit (AFFTU)

5.2.1 Kingsport Catalyst Campaign 1

Although there were no spent catalyst withdrawals during the campaign, freshly reduced catalyst was added to the LPMEOH[™] reactor to maintain productivity starting in July of 1997. Analyses of early samples from Kingsport Catalyst Campaign 1 indicated a step-change increase in the concentration of iron on the catalyst surface during the initial six weeks, which could not be correlated to the presence of iron carbonyl in the feed gas streams. This finding was most likely related to the detection of post-construction debris within various parts of the facility. During this period, higher than expected levels of arsenic were also found on the catalyst samples. Table 5.2.1-1 lists elemental composition of freshly reduced and spent catalyst samples from Kingsport Catalyst Campaign 1. The presence of arsenic in sample K0597-2 indicated that the manganese dioxide in the Eastman catalyst guard bed was no longer effective in removing arsenic from the Balanced Gas feed. Given this result, Eastman replaced both adsorbents (zinc oxide and manganese dioxide) with fresh material in October of 1997.

5.2.2 Kingsport Catalyst Campaign 2

Fresh methanol catalyst was activated and transferred to the LPMEOHTM reactor in December of 1997 in preparation for this operating campaign. Given that the adsorbents in the Eastman catalyst guard bed had recently been replaced with fresh material, it was decided to replace the activated carbon in the carbonyl guard bed with fresh adsorbent. At the same time, the AFFTU was returned to the Eastman chemicals-from-coal complex to obtain real-time laboratory data in parallel with the unit operation.^(d) At the baseline operating conditions for laboratory testing (including a reactor temperature of 250°C), the rate of catalyst deactivation in the autoclave in the AFFTU was statistically similar to the results using contaminant-free syngas. During the parallel test for the LPMEOHTM reactor (at a reactor temperature of 235°C), the catalyst deactivation in the LPMEOHTM demonstration unit was about 1% per day (as noted in Section 5.1-2), which was still higher than the results from the 4-month proof-of-concept test at the LaPorte AFDU in 1988/89 (0.4% per day at a reactor temperature of 250°C). Analytical results

Table 5.2.1-1 Kingsport Catalyst Campaign 1 - Elemental Composition of Solids from Slurry Samples

		Concentration,							
		, I	parts per	million (we	eight basis	s) (ppmw)			
	Time Onstream,								
	days	As	S	Fe	Ni	Cl			
Several	0	<25	100	47	25	-			
K0597-2	31	446	190	281	61	< 200			
K0897-1	93	601	235	169	< 20	-			
K0997-1	110	779	575	261	37	-			
K1097-	136	711	330	189	28	-			
K9711-1b	173	699	340	194	37				

also detected the presence of arsenic in the Balanced Gas feed (> 31 parts per billion by volume [ppbv]), carbonyl sulfide (COS) (10-20 ppbv), and an average total sulfur concentration of about 37 ppbv.

During the test period, there were numerous withdrawals of aged catalyst slurry and additions of freshly activated catalyst to the reactor resulting in a mixture of aged and fresh catalyst as the run proceeded. This complicated the interpretation of analytical data. Catalyst slurry samples for analysis at Air Products were taken at least monthly and were routinely monitored for expected trace contaminants including arsenic, sulfur, chlorine, iron, and nickel. As shown in Figure 5.2.2-1, arsenic, sulfur, and iron were detected on the catalyst in significant concentrations and, of these, arsenic was by far the most abundant. Catalyst additions and withdrawals caused periodic decreases in arsenic concentration but these values subsequently increase with time on stream.

Given that arsenic was the largest species detected on the samples of spent catalyst, laboratory studies were performed to determine its impact on catalyst life. The results showed that arsenic (in the form of arsine) was a strong catalyst poison, and that its effects were not determined during earlier autoclave testing in the AFFTU due to the high background rate of catalyst deactivation that is inherent to the laboratory setting (around 0.05% per hour or 1.2% per day vs. 0.4% per day at the LaPorte AFDU) and the low concentration of arsenic in the Balanced Gas feed (measured at 39 ppbv in a sample taken upstream of the Eastman catalyst guard bed in June of 1999). Figure 5.2.2-2 shows the fit of autoclave data for the impact of arsine on the life of methanol synthesis catalyst.

Figure 5.2.2-1 Kingsport Catalyst Campaign 2 - Trace Contaminant Concentrations on Slurry Samples



Figure 5.2.2-2 Effect of Arsine Concentration on Methanol Catalyst Deactivation Rate in Autoclave



Given the increasing concentration of arsenic on the samples of spent catalyst, the adsorbents in both the Eastman catalyst guard bed and the carbonyl guard bed were changed. In June of 1999, the Eastman catalyst guard bed was charged with only manganese dioxide for arsine removal, and the carbonyl guard bed was split between manganese dioxide and activated carbon (as the carbon may have affinity for other species beyond iron and nickel carbonyl). The spent activated carbon was analyzed for presence of trace contaminants; the results are provided in Table 5.2.2-1. The fresh adsorbent contained substantial concentrations of sulfur, N₂, and iron and thus an accurate accounting of these elements in the spent samples was difficult. In particular, the fresh adsorbent sulfur concentration was too high to obtain meaningful sulfur loadings on the bed. The results indicated a fairly uniform arsenic concentration throughout the bed implying that activated carbon was largely ineffective for removal of arsenic-containing contaminants.

Table 5.2.2-1Elemental Analysis Results for Spent Activated Carbon from 29C-40 Carbonyl Guard Bed
(June 1999)

			Eler	nental Co	ncentrati	ions, ppm	w, vs. Fre	sh Adsor	bent ^a	
Sample No.	Description	As	S	Ν	Zn	Mn	Fe	Ni	Cr	Cu
AGB-57	fresh adsorbent	6.0	5,930	5,000	2.6	10.7	2,010	39.6	5.0	18.6
AGB-42	bed inlet	1,630	≤360	2,400	225	141	nd	nd	nd	nd
AGB-46	5' from bed inlet	1,570	≤330	-	213	135	nd	nd	nd	nd
AGB-50	12' from bed inlet	1,990	≤430	3,400	122	79	nd	nd	nd	nd

a. for spent adsorbent, elemental concentrations were corrected for concentrations on fresh adsorbent; nd = none detected. During the first two months of service, the pressure drop across the carbonyl guard bed began to increase gradually. By the end of July of 1999, the pressure drop was measured at over 20 psi; the expected pressure drop was less than 1 psi. During a vessel inspection on 12 August 1999, changes to the bottom support screen and its support were made. During this maintenance, samples of the two adsorbents were analyzed. The results for the manganese dioxide are provided in Table 5.2.2-2. Arsenic was found throughout the manganese dioxide, but sulfur above that on the fresh adsorbent was found only on the inlet sample. For the activated carbon portion of the bed, arsenic concentrations were determined to be greater than that of the fresh adsorbent throughout the bed with a gradient from inlet to outlet (Table 5.2.2-3). This implied that arsine had broken through the bed.

Based upon these results, a laboratory screening program was initiated to identify alternate adsorbent materials with affinity for arsine that could be utilized in the Eastman catalyst guard bed and/or the carbonyl guard bed within the LPMEOHTM demonstration unit. Of the adsorbents examined, an activated carbon impregnated with copper oxide (CuO/C) was viewed as the material most likely to remove AsH₃ and other contaminants from the Balanced Gas feed. It was reasoned that the CuO portion would remove AsH₃, COS, and hydrogen sulfide (H₂S) while the carbon portion would function as it did previously to remove contaminants such as iron and nickel carbonyl as well as trace organics. The carbonyl guard bed was determined to provide adequate contact time and mixing length of the CuO/C adsorbent.

Table 5.2.2-2Analysis of Fresh and Spent Manganese Dioxide from 29C-40 Carbonyl Guard Bed(11 August 1999)

		conc.	, wt%				Concen	tration, p	omw			
Sample	description	Al	Mn	As	As ^a	S	S^{a}	Mg	Zn	Mo	Ga	Ba
	fresh adsorbent	36.2	24.4	<46	6.3	150	210	180	nd	nd	nd	nd
AGB81199-1	Тор	34.2	22.2	940	200	280	410	270	90	40	nd	nd
AGB81199-3	2 ft down	34.6	21.7	800	130	80	260	<300	nd	nd	30	nd
AGB81199-6	5 ft down	34.4	21.9	580	66	90	250	240	40	nd	nd	120

a. Concentration determined by elemental analysis; nd = not detected.

Table 5.2.2-3Analysis of Fresh and Spent Activated Carbon from 29C-40 Carbonyl Guard Bed
(11 August 1999)

					Concen	tration,	ppmw			
Sample	Description	S	S^{a}	As	As ^a	Na	Mn	Mo	Ba	Zr
	fresh adsorbent	11,400	370 ^b	<16	2.1	400	12	nd	nd	nd
AGB 81199-9	8' down, top C	10,700	400^{b}	1,000	1,000	700	40	10	130	40
AGB 81199-11	10' down, middle C	9,800	330 ^b	240	260	800	30	10	130	30
AGB 81199-13	12' down, bottom C	8,300	320 ^b	320	410	300	50	10	80	nd

a. Concentration determined by elemental analysis.

b. Large errors because of high background S; nd = not detected.

Laboratory testing was performed to determine any safety implications of using this material at ambient temperature. Exposure of CuO/C adsorbent to Balanced Gas at 750 psig and 30°C resulted in temperature increases of 4.5° C or less, implying that no significant reduction (in this case, the reaction of the copper oxide with a reductant such as CO or H₂ to copper metal and either CO₂ or H₂O) to metallic copper occurred at near ambient temperature. No methanol or other organics were observed in the bed exit gases, implying that the adsorbent had no methanol synthesis catalytic activity. Following reduction to a copper-impregnated carbon (Cu/C) adsorbent using dilute H₂ in N₂, temperature increases upon syngas exposure were almost the same as for the unreduced material. Again, the bed outlet showed no sign of methanol or other organics meaning that the Cu/C material had negligible catalytic activity at near ambient temperatures. Additional testing at temperatures up to 250°C resulted in little to no formation of methanol or other organics.

The new adsorbent material was loaded into the carbonyl guard bed, and the vessel was operationally tested on 03 August 2000. Within 30 minutes of the introduction of Balanced Gas, temperatures in the carbonyl guard bed began to rise beyond the expected values based upon the results of the laboratory testing. The Eastman operations team moved quickly to depressurize the carbonyl guard bed by venting the syngas to the Eastman purge gas header. No environmental issues were experienced due to this incident. The carbonyl guard bed was then isolated from the feed syngas streams so that the LPMEOHTM demonstration unit could continue to operate.

The results of additional laboratory testing indicated that the probable cause of this temperature excursion was the reduction of copper oxide to copper metal. A procedure was then developed to reduce the copper oxide on the adsorbent to copper metal at 100°C using a dilute stream of Balanced Gas in N₂ over a 2-day period. The existing charge of adsorbent was subjected to the reduction procedure on 17-19 April 2001. Little uptake of H₂ or CO was measured, indicating that the adsorbent material had been essentially reduced during the initial operations in August of 2000. The carbonyl guard bed vessel was brought onstream on 26 April 2001. The carbonyl guard bed internal temperatures increased by about 30-40°C and then stabilized following the introduction of syngas. This was consistent with the temperature rise that would be expected due to the adsorption of CO on the guard bed material. However, sampling of the syngas entering and leaving the carbonyl guard bed indicated that breakthrough of arsine through the adsorbent had occurred within the first 6 days of operation versus an expected life of 1 to 2 months. Samples of spent material from the carbonyl guard bed indicated that the exposure to elevated temperatures during the upset in August of 2000 may have damaged the adsorbent.

Air Products and Eastman decided to replace the used adsorbent with fresh material. On 24 July 2001, the carbonyl guard bed was emptied and refilled with the same commercially available adsorbent. During this reduction procedure, a dilute stream of Balanced Gas in N₂ was preheated and used to reduce the copper oxide to copper metal in a temperature controlled manner. The temperature control during the procedure was excellent, and the uptake of CO and H₂ approached the theoretical value based upon the concentration of copper oxide on the adsorbent. The reduction was completed on 28 July 2001. After cooling with N₂, a pressure check to full supply pressure with Balanced Gas was conducted. The adsorbent temperatures increased by about 25°C and stabilized during this examination. The carbonyl guard bed was

bypassed after the pressure check, and was brought onstream at the start of the third catalyst campaign.

5.2.3 Kingsport Catalyst Campaign 3

Unlike previous campaigns, the third Kingsport campaign involved an in-situ activation of a full charge (greater than 40,000 pounds) of catalyst. Methanol synthesis catalyst in the oxide form and mineral oil were loaded into the LPMEOHTM reactor without prior activation. The entire contents of the reactor were then reduced, or activated, using dilute Balanced Gas in N₂. In previous campaigns at the LPMEOHTM demonstration unit, catalyst had been reduced with dilute CO in N₂ in smaller batches (approximately 2,000 pounds per batch) in the catalyst preparation system and transferred into the LPMEOHTM reactor. The initial performance of the catalyst was less than expected when compared with data from the laboratory. Additional information on this procedure is provided in Section 5.3.

Over the entire campaign, only two contaminants (iron and arsenic) were found on the catalyst. As shown in Figure 5.2.3-1, iron concentrations increased at a substantial rate while the increase in arsenic was much more modest. The final sample contained 534 ppmw iron. Arsenic concentrations reached a maximum of only 318 ppmw and 203 ppmw for the final sample analyzed (213 days). Sulfur was below its limit of detection (160 ppmw or lower) throughout the campaign. Although a direct comparison of data from Campaigns 2 and 3 is complicated by



Figure 5.2.3-1 Kingsport Catalyst Campaign 3 - Trace Contaminant Concentrations on Slurry Samples

differences in total syngas exposure, some conclusions can be drawn. The arsenic loading on the Campaign 2 catalyst contained 277 ppmw sulfur while no sulfur was detected on the Campaign 3 catalyst. However, iron on the Campaign 2 catalyst was only 89 ppmw.

The use of the Cu/C adsorbent in the carbonyl guard bed was instrumental in lowering the quantity of arsine entering the LPMEOHTM demonstration unit. In order to confirm the observations during the laboratory evaluations, the effectiveness of the adsorbent was followed by monitoring AsH₃ concentrations in the carbonyl guard bed outlet. Samples were collected by passing syngas through an activated carbon trap followed by subsequent analysis for arsenic. During the first 27 days of service, the outlet AsH₃ concentration was less than 4 ppbv while the inlet arsine concentration was 52 ppbv (6 days) and 63 ppbv (28 days). Thus, the bed was effective for AsH₃ removal. The outlet concentration was next determined on the 68th day and found to be 5.6 ppbv, indicating that some breakthrough had occurred. The outlet concentration appeared to increase with time on stream and reached 30 ppbv by Day 102, indicating substantial breakthrough.

During the laboratory assessment of the Cu/C adsorbent, it was determined that the bond between copper and arsine was irreversible, and it was theorized that the rate limiting step in the reaction of Cu/C with AsH₃ was the diffusion of bound arsenic away from the Cu surface and into the bulk. It was reasoned that a "purge" of 120°C N₂ would warm the bed sufficiently to increase the rate of surface arsenic diffusion into the bulk and thus provide fresh copper surface for AsH₃ adsorption. During Catalyst Campaign 3, a thermal treatment was conducted in order to extend the useful life of the Cu/C adsorbent. The results of syngas sampling at the outlet of the carbonyl guard bed showed that the AsH₃ outlet concentration was below its limit of detection following both treatments. Thus, this thermal treatment extended the useful life of the adsorbent. Arsine was detected in the outlet of the carbonyl guard bed after 42 days on stream, which indicated that the performance of the adsorbent was beginning to degrade. A second thermal treatment was conducted, and once again arsine was not detected in the outlet from the carbonyl guard bed.

Based upon the decrease in the effective onstream time of the current charge of adsorbent, it was decided to replace the adsorbent in the carbonyl guard bed with fresh material. The spent Cu/C adsorbent was removed from the carbonyl guard bed and sampled; results of the analysis are provided in Table 5.2.3-1. The arsenic concentration varied in a regular fashion down the bed with 9,260 ppmw and 2,280 ppmw at the inlet and outlet, respectively. Sulfur concentrations were corrected for that of the fresh adsorbent and, because of high background sulfur concentration, the level of sulfur could be calculated only for the inlet sample. Similarly, high background iron precluded quantification of iron adsorption. No other contaminants were found on the spent adsorbent samples.

The fresh charge of CuO/C adsorbent was reduced and brought onstream on 08 March 2002. Initial performance following the replacement of the adsorbent was acceptable, as the average concentration of arsine was reduced from an inlet value of approximately 55 ppbv to 3 ppbv. The performance of the adsorbent began to degrade by late April of 2002, as the concentration of

				Concer	ntration, p	pmw		
Sample no.	Description	As	S^{a}	Fe	Ni	Р	Cr	Mo
18141-88-1	inlet	9,260	5,900	≤1,400	≤27	≤150	<8	≤140
18141-88-3	3 rd from inlet	8,720	nd	≤1,400	≤27	≤150	<8	≤140
18141-88-6	6 th from inlet	5,100	nd	≤1,400	≤27	≤150	<8	≤140
18141-88-9	9 th from inlet	4,250	nd	≤1,400	≤27	≤150	<8	≤140
18141-88-12	12 th from inlet	3,240	nd	≤1,400	≤27	≤150	<8	≤140
18141-88-16	16 th from inlet	3,150	nd	≤1,400	≤27	≤150	<8	≤140
18141-88-19	outlet	2,280	nd	≤1,400	≤27	≤150	<8	≤140

Table 5.2.3-1 Analysis of Fresh and Spent Cu/C Adsorbent from 29C-40 Carbonyl Guard Bed (20 February 2002)

a. Concentration corrected for background on the fresh adsorbent; nd = none detected.

arsine at the outlet of the carbonyl guard bed had increased to 10 ppbv. This charge of adsorbent was used until the end of Catalyst Campaign 3 on 05 June 2002.

During this campaign, one result of note was the substantial increase in iron on the catalyst with time. The most likely and possibly the only source of iron contamination is iron carbonyl. The volatile carbonyl present in the syngas feed can enter the LPMEOHTM reactor if breakthrough of the adsorbent in the carbonyl guard bed occurred. A laboratory evaluation had earlier showed that the Cu/C adsorbent had a very high removal capacity for iron carbonyl. Furthermore, results from gas sampling performed on 23 April 2002 found < 8 ppbv iron carbonyl in the carbonyl guard bed inlet and outlet. Another possibility was that iron carbonyl was generated within the stainless steel metallurgy in the LPMEOHTM reactor or the associated recycle loop hardware. Based on laboratory experiments, this also seemed unlikely. Although the concentration of iron detected in slurry samples was likely sufficient to adversely impact the long-term catalyst activity during this campaign, its presence was not related to poor adsorbent performance in the carbonyl guard bed or to some problem with the in-situ activation procedure.

It was subsequently determined that the most likely source of the iron was the incorrect installation of carbon steel parts inside the reactor during a maintenance outage in April of 2001. The correct stainless steel parts were installed after the conclusion of Kingsport Catalyst Campaign 3 in June of 2002. A hot function test with a CO-rich syngas and mineral oil (in the absence of catalyst) showed low levels of iron carbonyl (approximately 30 ppbv) in the reactor feed gas stream after three days of testing.

5.2.4 Kingsport Catalyst Campaign 4

The in-situ activation procedure was successfully demonstrated during the final Catalyst Campaign at the LPMEOH[™] demonstration unit; details are provided in Section 5.3. The results of analysis of spent catalyst samples from this campaign are listed in Table 5.2.4-1. No significant quantities of any of the contaminants analyzed were found during the first three months of this catalyst campaign.

A fresh charge of CuO/C adsorbent was charged to the carbonyl guard bed, reduced, and brought onstream at the start of Catalyst Campaign 4. During the first week in service, gas sampling was performed to assess the initial performance of the adsorbent. The outlet concentration of arsine was determined to be less than the detection limit of the analysis (2 ppbv). These results provide evidence that the adsorbent in the carbonyl guard bed was performing well during the first week of operation.

Based upon results from Catalyst Campaign 3, an operating schedule for the adsorbent was developed. Fresh adsorbent can be used for about two months before breakthrough of arsine can be expected. A thermal treatment can then be performed on the adsorbent to provide increased capacity for arsine removal, and the material can be used for another month before the adsorbent needs to be replaced. Based upon this timetable, a thermal treatment was completed in late August of 2002. The spent adsorbent was removed after about 3 months of use, and fresh material was reduced and brought onstream in mid-October of 2002. (At the same time, Eastman replaced the manganese dioxide in the 10C-30 catalyst guard bed with fresh adsorbent.) The timing for the thermal treatment for this charge was accelerated by one week (to 06 December 2002) due to an observed increase in the rate of catalyst deactivation (as described in Section 5.1.4).

Table 5.2.4-1 Kingsport Catalyst Campaign 4 - Elemental Composition of Solids from Slurry Samples

	Concentration, ppmw						
	Time Onstream						
	days	As	S	Fe	Ni	Cl	
K0206-1	0	< 20	< 40	< 23	< 10	< 100	
K0109-1	12	< 2	< 19	28	< 7	na	
K0109-3	26	4.2	< 25	44	< 7	na	
K0110-1	40	5.4	< 39	40	< 6	na	
K0110-2	54	19	< 47	58	< 7	na	
120006 1		0	.1	1.1 0			

a. K0206-1 is catalyst in oxide form; others are solids from reactor slurry samples; na = not available.

5.2.5 Summary

Significant progress was made during the operating phase of the LPMEOHTM Demonstration Project regarding the quantification of the impact of trace contaminants in coal-derived syngas on the performance of methanol synthesis catalyst and the study and use of adsorbents to remove targeted catalyst poisons. The 29C-40 carbonyl guard bed was originally designed for ambient temperature operation and to target the removal of carbonyl contaminants that could be present in coal-derived syngas. With the exception of two instances of the accidental introduction of iron to the system in Catalyst Campaign 1 (construction debris) and Catalyst Campaign 3 (incorrectly installed carbon steel parts in the LPMEOHTM reactor), iron and nickel carbonyls did not adversely impact the performance of the methanol synthesis catalyst at the LPMEOHTM demonstration unit.

Arsenic, and to a lesser degree sulfur, were the main contaminants that were detected on the samples of catalyst slurry taken from the LPMEOHTM reactor. Two adsorbents to target arsenic and sulfur were evaluated at the LPMEOHTM demonstration unit during Catalyst Campaigns 2 through 4. During Catalyst Campaign 2, the carbonyl guard bed was charged with 50% by volume manganese dioxide and 50% activated carbon. The manganese dioxide adsorbent displayed affinity for arsenic removal but the adsorbent bed broke through (based on gas-phase arsine analysis) after approximately one month of service. During Catalyst Campaigns 3 and 4, a copper oxide-impregnated activated carbon adsorbent was reduced and placed into service. This adsorbent was effective in arsine removal as determined by gas-phase analysis and catalyst sample results. The Cu/C adsorbent had a service life of about two months at the LPMEOHTM demonstration unit. An innovative thermal treatment was employed to extend the service life of the adsorbent for two additional months.

The improved performance of the catalyst guard system may have had a direct impact on the results for catalyst life during Catalyst Campaign 4 (catalyst deactivation rate of 0.1 to 0.2% per day versus an average of 0.6 to 0.7% per day during the previous 3 years of operation). By controlling the exposure of methanol synthesis catalyst to poisons, the targeted economics for coproduction of methanol in an IGCC power plant (as detailed in Section 9) can be achieved.

5.3 GAS SPARGER PERFORMANCE

The sparger is a proprietary gas distribution device used to introduce the feed gas into the bottom of the LPMEOHTM reactor. The sparger was developed on a proprietary basis by Air Products and is intended to introduce the feed gas to the slurry bubble column reactor. It must also be able to withstand interruptions to gas flow without negative impact on operations. The principles for the design of the sparger at the LPMEOHTM demonstration unit were first utilized during the 4-month proof-of-concept test of the LPMEOHTM Process at the LaPorte AFDU, during which no sparger-related outages were encountered.

Pressure drop can be expressed by the following equation:

$$\Delta P = \frac{K * (V * MW)^2}{\rho}$$

where:

This equation shows that changes in gas composition or flowrate can influence pressure drop measurements dramatically. Tracking the resistance coefficient (K) over time can indicate any change in the gas flow path through the gas sparger. For a given gas volumetric flowrate and density, an increase in K (caused by a restriction in the flow path, for example), will cause an increase in pressure drop.

As noted in Section 3.1.3, three different gas sparger designs were utilized during the operation of the Kingsport LPMEOHTM demonstration unit. A review of the performance of the devices during the 69-month operating program is provided.

5.3.1 Initial Sparger Design (April – November 1997)

The LPMEOH[™] demonstration unit was first brought onstream on 02 April 1997. Initial results indicated that the scaled-up gas sparger did provide the necessary distribution of syngas at the bottom of the reactor so that actual hydrodynamic behavior matched design expectations. In addition, the heat transfer coefficient of the internal heat exchanger exceeded the design value, further indicating good mixing. However, the measured pressure drop across the gas sparger increased considerably throughout the first few weeks onstream. As shown in Figure 5.3.1-1, the value for the resistance coefficient also reflected this trend, indicating the possibility that an obstruction of the flow path of the reactor feed gas was occurring. During this same time, periods of operation with no vapor flow through the gas sparger appeared to have no impact the rate of change of pressure drop.

Figure 5.3.1-1 Sparger Resistance Coefficient - Sparger Design #1 (April - November 1997)



During a scheduled outage in May of 1997, an inspection of equipment associated with the reactor, particularly the gas sparger, was performed. About 800 pounds of residual catalyst was removed from the bottom head of the reactor during this exercise. A solid material (presumably methanol synthesis catalyst) appeared to block about 50% of the flow path through the sparger; a small amount of catalyst was found in the inlet piping to the sparger. There was no discernible pattern to the blockage by the catalyst, and no significant construction debris was found in the inlet piping or in the sparger. The sparger was removed from the reactor and cleaned. Structural modifications increased the maximum allowable pressure differential across the sparger, but no changes were made to its flow distribution characteristics. In addition to the inspection, a flush connection of fresh process oil from the 29D-30 oil storage tank via the 29G-30 slurry transfer pump to the gas inlet line to the reactor was added.

Upon restarting the LPMEOH[™] demonstration unit on 17 June 1997, the pressure drop across the gas sparger returned to its design value. However, the gas sparger pressure drop and resistance coefficient did continue to increase with time onstream, although not as rapidly as the April/May operation (see Figure 5.3.1-1). A flush by fresh oil through the new connection at the gas inlet line yielded a decline in pressure drop through the gas sparger. However, the effects were temporary, and the flow resistance continued to increase. Additional flushing with fresh oil was limited, because fresh oil can only be added to the process at an average of 0.1 to 0.2 gpm to match the rate of oil loss with the methanol product.

The resistance coefficient experienced a significant step-change increase during a week-long shutdown in early September of 1997. A practice of flushing the gas sparger with entrained slurry from the 29C-06 cyclone and 29C-05 secondary oil knock-out drum was then initiated; this was supplied at the average rate of liquid flow in the secondary oil knock-out drum and cyclone (1 to 2 gpm). This flushing procedure was a batch-wise pumping process using the 29C-30 catalyst reduction vessel and the slurry transfer pump. Flow resistance through the gas sparger was stabilized at a manageable level by flushing in this manner (refer to Figure 5.3.1-1). The rate of addition of the flush was 30 gpm, and it was performed 2 to 3 times per day at the average rate of liquid flow through the cyclone and secondary oil knock-out drum.

On 09 October 1997 (Day 114), a new procedure was tested to gravity-drain the condensed oil and entrained slurry continuously to the flush connection at the average rate of liquid flow in the reactor loop, thus eliminating the batch-transfer operating steps. An inventory of condensed oil was collected in the secondary oil knock-out drum, and the gravity-drain line to the flush connection was then placed in service. Level in the secondary oil knock-out drum dropped and the temperature at the bottom of the reactor fell slightly, confirming the flowing path of oil to the reactor despite the already high pressure drop through the gas sparger.

During subsequent operation, the gravity-drain line became obstructed about twice per day. This blockage, either resulting from accumulation of catalyst or from vapor-locking within the piping system, cleared easily with a brief flush of clean oil from the oil storage tank via the 29G-03 oil makeup pump. At this frequency of operation, the amount of fresh oil added to the process during line flushing was less than the average oil loss rate with the methanol product (0.1 to 0.2 gpm). During the gravity-drain operation, the flow resistance coefficient across the gas sparger showed some modest improvement, compared to the period in September when the condensed oil and entrained slurry were pumped batch-wise via the slurry transfer pump.

On 03 November 1997, the LPMEOH[™] demonstration unit was shut down in order to prepare a second batch of fresh catalyst slurry. After the system was drained of spent catalyst, the LPMEOH[™] reactor was inspected by video camera. Dried catalyst slurry was observed in the bottom head and around the gas sparger.

Preparations were then made to restart the LPMEOHTM reactor with a second partial charge of fresh methanol synthesis catalyst. Batches of fresh catalyst slurry were activated in the catalyst reduction vessel and stored in the slurry tank. On 26 November 1997, the first seven batches of freshly reduced catalyst slurry were pressure-transferred from the slurry storage tank to the LPMEOHTM reactor. N₂ flow was introduced through the gas sparger prior to beginning the transfer operation, and the recycle compressor was started under N₂ at the completion of the transfer. While this was occurring, the activation of the eighth batch of catalyst was proceeding in the catalyst reduction vessel. Heat-up of the reactor using 600 psig steam on the internal heat exchanger proceeded smoothly; however, the pressure drop at the bottom of the reactor increased slowly, from 1.4 psi to 2.1 psi over a four-hour period. When the gas sparger was flushed at the gas inlet piping connection using fresh oil from the oil storage tank, the pressure drop across the sparger rose to about 7 psi. Then, when the eighth batch of slurry was transferred to the reactor using the slurry transfer pump, the pressure drop rose to 14 psi. After this transfer, the recycle compressor began to approach surge, indicating significant blockage at the sparger. Attempts to

clean the flow path were unsuccessful, and the contents of the reactor were pressure-transferred back to the slurry tank.

After completing the slurry transfer and securing the steam and syngas supplies, the manway on the bottom of the reactor was opened. The gas sparger was found to be plugged with fresh catalyst. In addition, a review of the data during the slurry transfer indicates that the N_2 flow to the reactor was quite low, and that there was evidence that the gas inlet piping to the reactor had filled with slurry during the transfer operation. Settling of the fresh catalyst appeared to have occurred in the reactor and gas inlet piping, which resulted in the plugging of the gas sparger.

5.3.2 Second Sparger Design (December 1997 – February 1999)

Following the attempted restart of the LPMEOHTM demonstration unit in November of 1997, a modified sparger design was developed by Air Products and fabricated and installed by Eastman. Part of the reasoning for draining the initial charge of catalyst slurry from the process was to demonstrate that the operation of the gas sparger could be maintained at the design flow resistance coefficient by gravity-draining the condensed oil and entrained slurry to the flush connection at the gas inlet piping to the LPMEOHTM reactor. Within 24 hours of restarting the LPMEOHTM demonstration unit, the gravity-draining flow path for this material was placed in service. As shown in Figure 5.3.2-1, the flow resistance through the gas sparger of the LPMEOHTM reactor could be stabilized by use of this continuous flush. Performance of the second sparger design showed significant improvement over the operating results from 1997; however, a slight increase in resistance coefficient over time was still apparent.

On 15 October 1998 (Day 299), maintenance work was performed on the oil flush system that cleans the taps used to measure the pressure drop across the gas sparger. Upon completion of this work, the measured pressure drop increased by 2.0 psi. An analysis of other pressure readings around the reactor was performed to determine the rate of increase of this pressure drop. It appeared that the increase in pressure drop across the sparger occurred around the time of the maintenance work. Another step-change increase in pressure drop occurred after restarting from a complex-wide power outage in early November of 1998.

As a result of the greater sparger pressure drop, condensed oil and entrained slurry could not be gravity-drained to the flush connection at the gas inlet line to the reactor. In order to overcome the pressure drop, the streams were batch-transferred to the catalyst reduction vessel and returned to the reactor via the slurry transfer pump. This procedure was last used with the initial sparger design in October of 1997. During periods when the catalyst reduction vessel was otherwise in use, the condensed oil and entrained slurry streams were free-drained directly to a connection on the reactor where the pressure drop was not limiting.

Figure 5.3.2-1 Sparger Resistance Coefficient - Sparger Design #2 (December 1997 - February 1999)



During January and February of 1999 (around Days 377-435), the sparger resistance coefficient across the LPMEOHTM reactor continued to increase beyond the step-change experienced in October of 1998 (refer to Figure 5.3.2-1). The associated higher pressure drop within the reactor loop limited the quantity of Balanced Gas that could be fed to the LPMEOHTM demonstration unit.

During a schedule outage in March of 1999, the second version of the gas sparger was removed from the LPMEOHTM reactor and inspected. It was estimated that the blockage in the sparger was consistent with the resistance coefficient that was calculated prior to shutdown.

5.3.3 Current Sparger Design (March 1999 – Present)

During the outage in March of 1999, a modified gas sparger was installed in the reactor; Air Products developed this design using the same principles as the previous two devices. During that outage, a gravity drain line piping modification was also made to separate the condensed oil from the oil knockout vessel from the entrained slurry from the cyclone. The condensed oil stream is relatively clean and was used to continuously flush the sparger; the slurry from the cyclone was sent directly into the reactor. The separation of these streams reduced plugging and improved the overall reliability of the sparger flush and the oil return. Upon restarting on 14 March 1999, the performance of the new sparger met or exceeded the design expectations for pressure drop and reactor operation, and stable operation was maintained for over 500 days (refer to Figure 5.3.3-1).

Figure 5.3.3-1 Sparger Resistance Coefficient - Sparger Design #3 (March 1999 - August 2001)



After the restart of the LPMEOHTM demonstration unit on 04 August 2000 (Day 958 of Catalyst Campaign 2), the resistance coefficient increased approximately 50% (from an average of about 4.3 to 6.6). Subsequent flushing of the sparger with clean oil resulted in a sizable reduction of the increased flow resistance (to an average of 5.3). During the operating period between March 2001 (Day 1193 of Catalyst Campaign 2), the resistance coefficient did not appear to be further increasing over time.

During this biennial outage, the sparger was removed from the LPMEOHTM reactor for inspection and cleaning. In addition, all differential pressure transmitters around the sparger and reactor were calibrated, and the pressure taps were cleaned. Following the restart of the unit (Day 1217 of Catalyst Campaign 2), the resistance coefficient was higher than the value that was calculated following the March 1999 outage. The most likely cause is that the recalibration of the differential pressure transmitters had experienced a shift in their baseline measurement. The sparger resistance continued to show no significant increase over time, which was consistent with the operating history with this device.

In August of 2001, Catalyst Campaign 3 was started, during which the catalyst inventory was removed and a new in-situ activation procedure was used to activate a full load of fresh catalyst inside the reactor vessel (refer to Section 5.1). Figure 5.3.3-2 shows the sparger performance for this operating period (August of 2001 to June of 2002). The sparger resistance showed no increase over time during this period.

Figure 5.3.3-2 Sparger Resistance Coefficient - Sparger Design #3 (August 2001 - June 2002)



During the preparations for Catalyst Campaign 4 in June of 2002, the sparger was inspected and cleaned. As shown in Figure 5.3.3-3, the performance of the sparger showed no significant increase in resistance between following the restart on 27 June 2002 through to the end of the operating program on 31 December 2002.

Figure 5.3.3-3 Sparger Resistance Coefficient - Sparger Design #3 (June 2002 - 31 December 2002)



5.3.4 Summary

The current design of the gas sparger for the LPMEOH[™] reactor has been demonstrated to meet the requirements for initial pressure drop and long-term performance. The device provides operating flexibility through all modes of operation. The use of condensed or entrained slurry, either by pump or by gravity draining, as a continuous or periodic flush through the sparger has contributed greatly to the reliability of the sparger operation. This flushing system and procedure is now a standard part of the design package for the LPMEOH[™] Process.

5.4 PROCESS VARIABLE TESTS: H₂:CO RATIO

Syngas ratios will vary with the type of gasifier and feed used in an IGCC power plant application. A key commercialization goal of the LPMEOHTM Demonstration Project was to operate the LPMEOHTM reactor utilizing a wide variety of syngas feed ratios. Three different feed gas streams were diverted from existing operations to the LPMEOHTM demonstration unit:

- H₂ Gas, containing an excess of H₂ over the stoichiometric balance for the production of methanol (as noted in Section 3.1, this stream was not used during the demonstration test program);
- 2) CO Gas, a syngas containing primarily CO; and
- 3) Balanced Gas, a syngas with a composition of H_2 , CO, and CO₂ in stoichiometric balance ($[H_2 CO_2]/[CO + CO_2] = 2.05$) for the production of methanol.

With the availability of these feed gas streams, a wide range of syngas feed ratios, including those from modern coal gasifiers, could be examined (subject to syngas availability). The H₂:CO ratio at the reactor inlet was set to the desired value by adjusting the total quantity of fresh feed syngas and the Recycle Gas.

As noted in Section 2.1, the LPMEOHTM Process is particularly well suited to coal-derived syngas which is rich in CO. Unlike gas-phase reactors, shift and CO₂ removal are not required to adjust the composition to a particular requirement. Low H₂-to-CO ratios are acceptable as is any CO₂ content. These capabilities make the LPMEOHTM Process a potentially lower-cost conversion route to methanol, especially when methanol coproduction is added to a coal-based IGCC power plant. As shown in Table 4.1-1, several test campaigns were undertaken to study the performance of the LPMEOHTM Process using CO-rich syngas. Three ranges of CO-rich feed gases were generally tested: a syngas with a H₂:CO ratio at the reactor inlet of approximately 0.5, a syngas with a H₂:CO ratio at the reactor inlet of about 0.7, and a syngas with a H₂:CO ratio at the reactor inlet of approximately 1.0. A table detailing all of the CO-rich testing performed during the demonstration test program at the LPMEOHTM demonstration unit is given in Table 5.4-1.

A total of 4 days of testing with a CO-rich syngas at a H_2 :CO ratio of about 0.5 (around 57 volume % [vol%] in the reactor feed gas stream) was conducted during the operating program. The results for methanol production matched the performance based upon studies in the laboratory autoclave. Due to the short duration of these tests, the rate of catalyst deactivation could not be calculated.

Date Started	Date Ended	Duration (days)	H2:CO Ratio @ Reactor Inlet	Reactor Temperature (°C)	Methanol Production (sT/D)	Catalyst Deactivation Rate (% per day)
5/7/97	5/8/97	2	0.43	250	70	na ^a
8/1/97	8/12/97	12	0.80	250	190	- 1.70
2/8/97	2/12/98	5	0.78	225	192	na
8/24/98	8/26/98	3	0.96	250	228	na
12/2/99	12/10/99	8	1.02	235	204	na
2/23/00	2/23/00	1	0.57	235	177	na
3/21/00	4/13/00	24	1.03	235	180	-1.24
5/9/00	5/10/00	1	~ 0.5	250	NA	na
11/15/00	11/18/00	4	1.08	235	220	na
5/9/01	5/13/01	5	0.65	235	117	na
10/29/01	10/29/01	1	0.64	218	156	na
12/12/01	12/17/01	6	0.68	224	142	na
12/24/01	12/29/01	6	0.76	226	143	na

 Table 5.4-1

 Kingsport LPMEOH[™] Process - H₂:CO Ratio Testing

a. na = operating campaigns were of insufficient length to calculate rate of catalyst deactivation.

A total of 35 days of testing with a CO-rich syngas at a H_2 :CO ratio of about 0.7 was conducted during the operating program. During a 12-day campaign that was undertaken during Catalyst Campaign 1 (early August of 1997), the rate of catalyst deactivation was calculated to be 1.7% per day. This result was statistically similar to the rates calculated during this campaign when Balanced Gas was used as the feed to the LPMEOHTM demonstration unit.

A total of 39 days of testing with a syngas at a H_2 :CO ratio of around 1.0 was conducted during the operating program. A 24-day test was performed from 21 March 2000 to 13 April 2000. The catalyst performance for this test period was calculated as a deactivation rate of 1.24% per day. This rate was also statistically similar to the rates calculated using Balanced Gas during this same period during Catalyst Campaign 2 (refer to Section 5.1.2).

In summary, a variety of H₂:CO ratios were tested in the Kingsport LPMEOHTM demonstration unit to evaluate the effect of syngas compositions on the performance of the LPMEOHTM Process. All of the results met the expectations for methanol production rate based upon earlier studies in the laboratory. Catalyst performance for the LPMEOHTM reactor with lower H₂:CO ratios were not significantly impacted based on the calculated rate of catalyst deactivation. These results provide additional confirmation (beyond the performance during the 4-month proof-of-concept test at the LaPorte AFDU in 1988/89 on CO-rich syngas) of the capability of the LPMEOHTM Process to process syngas directly from all modern syngas generation systems, including coal gasifiers.

5.5 RAMPING AND ON/OFF TESTING

The ability of the LPMEOHTM Process to start and stop rapidly is an important feature within an IGCC power plant application. In addition, the capability of the LPMEOHTM Process to perform in an electrical demand environment will be required for the technology to be successfully dispatched in these power generation applications. Although the host site for the LPMEOHTM demonstration unit was a chemical production facility, a test protocol was developed to evaluate the ability of the reactor to ramp up and shut down quickly, and cycle on and off. These ramping and on/off tests were conducted within the limitations of the rate of response of the existing Eastman complex, in particular the limits on rate-of-change on the parallel fixed-bed methanol plant and the impact of purge flow from the LPMEOHTM demonstration unit on the operation of the downstream boiler system.

Opportunities to test the ability of the LPMEOHTM process to stop quickly occurred throughout the operating program. Syngas could be stopped as quickly as the other systems within Eastman's chemicals-from-coal complex could handle the additional flow. It was also shown that the recycle compressor could continue to operate until the reactor pressure, which decreased either due to purge losses or continued consumption of H₂ and CO, reached the minimum flow based upon approach to surge (around 85 psig).

In designing the test for the restart of the LPMEOH[™] demonstration unit, a goal of 5 to 10% change in design flow per minute, which meets the requirements for an IGCC system, was established. The first test of the ability to ramp the LPMEOH[™] reactor from a standby condition to production rates was performed 13 June 2000. This test simulated the ramping that must be performed within the startup requirements of an IGCC facility. An average ramping rate of 3% of the design flow (990 KSCFH) per minute and a maximum ramp rate of 4% of the design flow per minute was achieved, which approached the minimum target rate of 5% per minute. Figure 5.5-1 shows the results of this initial ramping test.

Three additional ramping tests were performed between 04 and 13 March 2001. For the first two tests (on 04 and 06 March), the reactor standby temperature was 216°C and 212°C, respectively; from this starting point, a reasonable ramping rate could not be reached. During the third test on 13 March 2001, a standby reactor temperature of 227°C was used; for all of the tests performed in March of 2001, gas flow from the recycle compressor was maintained to the LPMEOHTM reactor. Balanced Gas was then introduced to the process as quickly as possible (within the limitations of other systems at the Eastman chemicals-from-coal complex). As shown in Figure 5.5-2, a maximum ramp rate of 5.7% full load per minute and an average ramp rate of 4.2% full load per minute were achieved during this testing. This ramp performance approached the minimum target ramp rate of 5% full load per minute. For each test, once Balanced Gas flowrate

Figure 5.5-1 Ramping Study at LPMEOH[™] Demonstration Unit (13 June 2000)



Figure 5.5-2 Ramping Study at LPMEOH[™] Demonstration Unit (13 March 2001)



was reestablished and reactor temperature and pressure were placed into control, the performance of the LPMEOH[™] demonstration unit was determined to have returned to the condition prior to the shutdown.

Two tests of the ability of the LPMEOH[™] reactor to operate in a load-following environment were performed during the week of 15 January 2001. For two consecutive days, the reactor was taken offline for 12 hours, and then Balanced Gas was introduced to the process as quickly as possible (within the limitations of other systems at the Eastman chemicals-from-coal complex). The attempts to perform these ramping tests were hampered by the standby conditions that were selected, including the lack of flow to the reactor (which lowered the reactor heat transfer coefficient) and the reactor temperature (180°C). Once Balanced Gas flowrate was reestablished and reactor temperature and pressure were placed into control, the performance of the LPMEOH[™] demonstration unit, including the gas sparger in the reactor, was determined to have returned to the condition prior to the shutdown.

During these tests, it was shown that the LPMEOHTM demonstration unit was capable of meeting the ramping requirements for an IGCC system by achieving the goal of a minimum 5% change in design flow per minute. Tests of the ability of the LPMEOHTM reactor to operate in a load-following environment were also successful; the performance of the LPMEOHTM demonstration unit, including the gas sparger in the reactor, was determined to have returned to the condition prior to the commencement of the on/off testing.

5.6 IN-SITU ACTIVATION AND TEMPERATURE PROGRAMMING

Depending upon the H₂:CO ratio and level of catalyst poisons in the fresh feed gas and sitespecific requirements, it may be economical to operate the LPMEOHTM Process in the same manner as conventional gas-phase reactor systems - that is, charge the maximum amount of unreduced catalyst slurry to the LPMEOHTM reactor, perform the catalyst reduction (reacting metal oxides [for example, copper oxide] with a reductant such as CO or H₂ to produce the base metal plus either CO₂ or H₂O) within the reactor vessel (so-called in-situ activation), and practice temperature programming (the increase of reactor temperature as necessary to control the reactor purge flowrate and maintain reactor volumetric productivity) to extend the catalyst life. In this scenario, eliminating the need for a separate catalyst activation vessel, slurry addition pump, and other subsystems can lower capital costs. There has been interest from potential customers for the LPMEOHTM Process to demonstrate in-situ catalyst activation at the LPMEOHTM demonstration unit; as a result, tests of both the activation procedure (a base case and an improved procedure) and the subsequent temperature programming were performed during the demonstration.

5.6.1 <u>Preparatory Engineering and Laboratory Work</u>

Preliminary process simulation work was performed in January of 2000 in order to determine the operating conditions that could be used (based upon site-specific limitations) during a potential test of in-situ catalyst activation within the LPMEOHTM reactor. For example, the amount of purge gas flow from the LPMEOHTM demonstration unit (Stream 148 of Figure 2.4-1) was limited based upon the capacity of the downstream boiler system; this established the maximum

quantity of gas (dilute Balanced Gas in N_2) that could be fed during the catalyst activation procedure. A test to determine the minimum operating pressure and corresponding flowrate of N_2 that could be generated by the recycle compressor was performed in March of 2000. This information was then used to modify the procedure and determine conditions for a confirmation test in the laboratory autoclave. Laboratory testing of the proposed procedure was then undertaken to verify the operating conditions.

In July of 2000, Air Products and Eastman initiated the engineering activities in support of the demonstration of the in-situ activation of methanol synthesis catalyst in the LPMEOHTM reactor. Piping and instrumentation changes were needed to allow for the introduction of the Balanced Gas and N₂ streams at the proper flowrates per the proposed procedure. Air Products released a process engineering specification for these changes in August of 2000. Eastman then developed a detailed design package that defined the necessary modifications to the piping and instrument systems. In order to minimize downtime, the field modifications were completed during a scheduled code biennial inspection of all pressure vessels within the LPMEOHTM demonstration unit in March of 2001. A detailed operating procedure was drafted by Air Products and reviewed by Eastman prior to the initiation of the first in-situ activation of methanol synthesis catalyst in August of 2001.

5.6.2 In-situ Activation Procedure and Results - August 2001

The LPMEOHTM demonstration unit was shutdown on 06 August 2001 to prepare for the first test of the in-situ catalyst activation procedure. The slurry contents of the reactor from Catalyst Campaign 2 were drained from the system. Beginning on 13 August 2001, batches of fresh catalyst were mixed with mineral oil in the catalyst reduction vessel, heated to 150°C, and transferred, without being reduced (or activated), to the slurry tank. A total of ten batches of fresh catalyst slurry, each containing about 4,000 pounds of fresh catalyst, were prepared in this manner. After the final transfer, the total catalyst inventory in the slurry tank was 41,580 pounds of fresh catalyst, which was approximately the design catalyst inventory for the LPMEOHTM reactor. The catalyst loading and transfer operation to the slurry tank was completed on 21 August 2001.

The common inlet and outlet line of the slurry tank (2" diameter) developed a plug during the transfer of the tenth batch of fresh catalyst slurry from the catalyst reduction vessel. After modifications to the piping system were made, Eastman operations personnel were able to clear this obstruction, and the contents of the slurry tank were then successfully transferred into the LPMEOHTM reactor. The transfer of fresh catalyst slurry to the reactor was completed 22 August 2001.

Immediately upon completion of the slurry transfer, the recycle compressor was started on N_2 to begin to cool the reactor contents to less than 90°C, which was the desired initial temperature for catalyst activation. After this step was completed, the in-situ catalyst activation procedure was started at 2300 hrs on 22 August 2001. Make-up N_2 flow was controlled at 52 KSCFH, and a small quantity of Balanced Gas was introduced into the discharge of the recycle compressor to ensure that sufficient reductant was present at all times in the reactor inlet. Temperature control was a manual ramping operation using the steam drum pressure as a control point. A reactor temperature ramp from 88°C to 215°C over a 27-hour period was used. The in-situ catalyst activation procedure ended at 0100 hrs on 24 August 2001.

The progress of the in-situ catalyst activation procedure was tracked by evaluating the uptake of reductant (H₂ and CO) by the catalyst. An uptake of 70% of the theoretical value was calculated based upon analytical data and measured flowrates. Figure 5.6-1 shows the uptake performance during the first in-situ activation as compared to the performance obtained in the laboratory autoclave during the testing in 2000. The LPMEOHTM demonstration unit was then restarted at 1200 hrs on 24 August 2001. As described in Section 5.1.3, the performance of the methanol synthesis catalyst was also lower than expected as calculated from an in-house kinetic model. Catalyst samples that were taken from the reactor following the in-situ catalyst activation did not provide conclusive results.

An assessment of the performance of the methanol synthesis catalyst following the completion of the in-situ activation procedure in August of 2001 was completed. Based on laboratory studies, it was found that storage of the fresh catalyst in the presence of mineral oil at elevated temperatures (the experiments were performed at approximately 200°C) prior to reduction was the cause of the lower than expected catalyst activity. During the initial demonstration of the in-situ activation procedure at the LPMEOHTM demonstration unit, the slurry was approached this temperature in the slurry tank, where batches of fresh catalyst slurry were stored during the catalyst loading procedure.



Figure 5.6-1 Reductant Gas Uptake vs. Temperature - In-Situ Catalyst Activation (22-24 August 2001)

5.6.3 <u>Temperature Programming (August 2001 – March 2002)</u>

Following the completion of the in-situ catalyst activation procedure, conditions were selected to deal with the hyperactivity that typically occurs when fresh methanol synthesis catalyst is initially exposed to syngas. During the first few weeks of operation, conditions were varying between a

reactor pressure of 550-600 psig, reactor temperature of 212-220°C, and flowrate of Balanced Gas of 500-525 KSCFH. Once the hyperactivity period had passed, operating conditions were set at 218°C reactor temperature, 685 psig reactor pressure, and 600-650 KSCFH of Balanced Gas.

As part of the operating protocol following the completion of the in-situ activation of methanol synthesis catalyst in the LPMEOHTM reactor, temperature programming continued during the operating period; this involved the increase of reactor temperature as necessary to control the reactor purge flowrate and maintain reactor productivity. Over the entire period spanning August of 2001 to March of 2002, the reactor temperature was increased from 218°C to the final temperature of 250°C (250°C was the design temperature for the LPMEOHTM demonstration unit). The reactor pressure was increased from 685 psig to 700 psig on 12 November 2001 and maintained at that level for the balance of the operating period. The flowrate of Balanced Gas was controlled at an average value of 630 KSCFH during this time.

The first operating test using methanol synthesis catalyst activated in-situ was completed on 25 March 2002 when a catalyst withdrawal and addition campaign was undertaken to increase catalyst activity.

5.6.4 Second In-situ Activation Procedure and Results (June 2002)

Based upon the results of the laboratory assessment following the August 2001 in-situ catalyst activation, operating steps were modified so that the fresh methanol catalyst would not be exposed to temperatures in excess of 100°C prior to the introduction of dilute syngas. A recommendation to DOE to perform a second test of the in-situ catalyst activation procedure was accepted.

Beginning on 21 June 2002, batches of fresh catalyst were mixed with mineral oil in the catalyst reduction vessel, heated to 100°C, and transferred, without being reduced or activated, directly to the reactor. Since the slurry tank was not used to collect the batches, the process was greatly simplified and shortened (from 8 days to 3 days) compared with the first in-situ activation procedure in August of 2001. A total of eleven batches of fresh catalyst slurry, each containing about 3,700 pounds of fresh catalyst, was prepared in this manner. After the final transfer, the total amount of fresh catalyst in the reactor was 40,040 pounds, which is approximately the design catalyst inventory for the LPMEOHTM demonstration unit. N₂ was introduced in the bottom of the reactor during the entire loading procedure, and the unreduced catalyst in the reactor was maintained at 100°C by adding steam as necessary to the steam drum and internal heat exchanger. The catalyst loading and transfer operation was completed on 24 June 2002.

The reactor was cooled overnight to the desired initial temperature for catalyst activation; the recycle compressor was started on N_2 to assist in cooling the reactor contents. By 0800 hours on 25 June 2002, the reactor inlet flow was set at approximately 300 KSCFH, and 52 KSCFH of make-up N_2 was introduced at the compressor suction. Reactor pressure was set at 85 psig. Cooling was completed after the reactor temperature reached 86°C.

The in-situ catalyst activation procedure was started at 1200 hrs on 25 June 2002. In addition to the 52 KSCFH of N_2 , a small quantity of Balanced Gas was introduced into the discharge of the recycle compressor to ensure that sufficient reductant was present at all times in the reactor inlet. A temperature ramp from 87°C to 215°C was conducted over a period of 51 hours, which was a slower ramp rate than was used during the first in-situ activation. The in-situ catalyst activation procedure ended at 0200 hrs on 27 June 2002.

The progress of the in-situ catalyst activation procedure was tracked by evaluating the uptake of reductant by the catalyst. A reductant gas uptake of 98% of the theoretical value was calculated based upon analytical data and measured flowrates, which was an indication that the methanol synthesis catalyst was properly activated during the in-situ catalyst activation procedure. Figure 5.6-2 shows the excellent uptake performance of the second in-situ activation as compared to the performance obtained in the laboratory autoclave.





The LPMEOHTM demonstration unit was then re-started at 1500 hrs on 27 June 2002. In addition, the initial value for η (as defined in Section 5.1) was calculated to be 1.23, which was representative of a properly activated catalyst. In addition, the startup conditions for this charge of methanol synthesis catalyst were less severe (pressure of 450 psig and 214°C reactor temperature) than those following the first in-situ catalyst activation (550-600 psig, reactor temperature of 212-220°C) while processing a greater quantity of Balanced Gas (603 KSCFH vs. 500-525 KSCFH). These results showed a clear improvement when compared with the performance of the methanol synthesis catalyst following the first in-situ activation procedure. As noted in Section 5.1.4, due to the low catalyst deactivation rate experienced following the remainder of the operating program of the LPMEOHTM Demonstration Project.

5.6.5 Summary

The in-situ activation procedure was successfully demonstrated at the LPMEOH[™] demonstration unit. Following the initial trial in August of 2001, revisions to the activation procedure, including a limitation on the storage temperature for the fresh catalyst slurry and a much slower temperature ramp rate during the activation procedure, were made. Temperature programming was found to be a very practical approach to maintaining reactor productivity during the

operating program following the first in-situ activation; although the reactor temperature did not have to be increased during Catalyst Campaign 4 (June-December of 2002), the selection of the minimum reactor temperature necessary to meet the desired methanol production rate may have been a factor in the low rate of catalyst deactivation that was calculated during this operating period (as described in Section 5.1.4).

5.7 SECTION 5 BIBLIOGRAPHY

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- c. "Design and Construction of the Alternative Fuels Field Test Unit and Liquid Phase Methanol Feedstock and Catalyst Life Testing at Eastman Chemical Company (Kingsport, TN)", Topical Report Prepared by Air Products and Chemicals, Inc., DOE Cooperative Agreement No. DE-FC22-95PC93052, March 1997.
- d. "Alternative Fuels Field Test Unit Support to Kingsport LPMEOH[™] Demonstration Unit Topical Report (December 1997 - January 1998)", Topical Report Prepared by Air Products Liquid Phase Conversion Co., L.P., DOE Cooperative Agreement No. DE-FC22-92PC90543, November 1998.

6.0 ENVIRONMENTAL PERFORMANCE

6.1 INTRODUCTION

As part of the requirements for DOE to fulfill its responsibility under the National Environmental Policy Act (NEPA) of 1969, the Partnership prepared a detailed Environmental Information Volume (EIV)^(a) describing the environmental aspects and projected impacts of the LPMEOHTM Demonstration Project. DOE then prepared an Environmental Assessment (EA)^(b) for the Project, and determined that an Environmental Impact Statement was not required. A Finding of No Significant Impact (FONSI) was issued on 30 June 1995.

The Partnership then prepared an Environmental Monitoring Plan (EMP)^(c) which described in detail the environmental monitoring activities to be performed during the operation of the LPMEOHTM demonstration



unit. In order to quantify the environmental performance of the LPMEOHTM demonstration unit, a comparison of the estimated quantities of emissions, effluents, and solids for disposal from the EA with the results from the LPMEOHTM demonstration unit is provided in this Section.

6.2 WASTE STREAMS AND THEIR DISPOSAL

6.2.1 Spent Catalyst Slurry

The LPMEOH[™] Process uses conventional low-temperature methanol synthesis catalyst (comprised primarily of oxides of copper and zinc) in powder form, slurried in a mineral oil that is inert at reaction conditions. Over time, these catalysts, whether used in conventional fixedbed reactor systems or the LPMEOH[™] Process, will show a decrease in catalytic activity. This aging phenomenon is caused by both the presence of the catalyst at the reaction conditions (most likely temperature) and by the chemical poisoning of the catalyst by trace contaminants within the system (for example: sulfur, iron, nickel, and arsenic are known catalyst poisons). In order to maintain the same level of production of methanol across the reactor, one of two methods can be used:

- 1) The process conditions can be altered to increase the rate of production of methanol (this is typically accomplished by slowly increasing the reactor temperature with time, and is the method that is used in the fixed-bed methanol processes)
- 2) A portion of the volume of catalyst slurry within the reactor can be removed and replaced with an equal volume of fresh catalyst slurry.

As noted in Section 5.1, the original design of the LPMEOHTM Process called for the use of catalyst withdrawal and addition to maintain the production rate of methanol. This method tends

to use larger quantities of methanol synthesis catalyst, as a portion of the catalyst that is removed from the LPMEOHTM reactor has greater activity than the average catalyst particle within the process.

However, the operating tests following the in-situ activation of methanol synthesis catalyst within the LPMEOHTM reactor have also demonstrated the ability to temperature program the reactor and maximize the on-stream time of the catalyst (refer to Section 5.6).

Table 6.2-1 provides the annual amount of spent catalyst slurry that was generated by the LPMEOHTM Process during the demonstration test period. The average annual generation rate is less than the projected rate as reported in the final EA for the LPMEOHTM Demonstration Project. During the period from April of 1997 through July of 2001, the catalyst withdrawal/addition method was used to maintain the rate of production of methanol across the LPMEOHTM reactor. In addition to this periodic transfer of a portion of the catalyst slurry from the process, the entire contents of the LPMEOHTM reactor were drained on two occasions (in November of 1997, following the initial accelerated rate of deactivation related to the presence of iron on the catalyst; and in July of 2001, in preparation for the first attempt to activate the methanol catalyst using the in-situ activation procedure). Between August of 2001 and March of 2002, temperature programming was used following the in-situ activation of the methanol synthesis catalyst within the LPMEOHTM reactor. A brief period of catalyst withdrawal and addition then followed until the contents of the reactor were drained in June of 2002 and the second attempt to demonstrate the in-situ activation procedure was successfully completed.

All of the spent catalyst slurry was shipped on a periodic basis from the LPMEOHTM demonstration unit to a company which reclaimed the metallic components (copper and zinc) from the catalyst.

Year of Operation	Spent Catalyst Slurry to Reclaimer, pounds	Comments
April 1997 – March 1998	91,100	Onstream 02 April 1997
April 1998 – March 1999	22,500	
April 1999 – March 2000	126,400	
April 2000 – March 2001	76,450	
April 2001 – March 2002	145,140	
April 2002 – December 2002	95,205	
Annual Average	92,800	Projected to March 2003
Expected Value	103,100 to 128,000	Refer to Final EA for Project ^(b)

Table 6.2-1Spent Catalyst Slurry from LPMEOH™ Demonstration Unit

6.2.2 10C-30 Catalyst Guard Bed

As shown in Figure 1.2.3-1, the 10C-30 catalyst guard bed is located within the Eastman chemicals-from-coal complex, upstream of both the LPMEOHTM demonstration unit and the existing gas-phase methanol reactor. Prior to the startup of the LPMEOHTM demonstration unit, this catalyst guard bed had been charged with both zinc oxide and manganese dioxide in order to remove trace quantities of sulfur and arsenic that are present in Balanced Gas. Table 6.2-2 shows the type and quantity of adsorbent materials that were used in this service during the demonstration period for the LPMEOHTM Process at Kingsport.

6.2.3 29C-40 Carbonyl Guard Bed

As shown in Figure 2.4-1, the 29C-40 carbonyl guard bed is located within the LPMEOHTM demonstration unit, and was designed to remove trace levels of metal carbonyls (iron and nickel) that could be present in either the Balanced Gas or CO Gas streams. Table 6.2-3 shows the type and quantity of adsorbent materials that were used in this service during the demonstration program for the LPMEOHTM Process at Kingsport. Details on the technical selection for the adsorbent materials used in the carbonyl guard bed are provided in Section 5.2. The expected quantity of adsorbent during the design phase of the project was a single charge of 10,000 pounds of activated carbon over a four-year period. Due to the adverse impact of trace contaminants (in particular arsenic) that are poisons to methanol synthesis catalysts, additional types and combinations of adsorbents were used in the carbonyl guard bed.

In Section 4.4.2 of the EMP, incineration was considered to be the means of disposal of the spent activated carbon. Given the change in the type of adsorbent and the presence of arsenic, Eastman performed the testing necessary to characterize the copper-impregnated activated carbon and determine the appropriate means of disposal. The Toxicity Characteristic Leaching Procedure

Table 6.2-210C-30 Catalyst Guard BedAdsorbents Used during Operation of LPMEOH™ Demonstration Unit

Adsorbent Material	Targeted Catalyst Poison	Time in Service
(1) Zinc Oxide (155 cubic feet)(2) Manganese Dioxide (75 cubic feet)	(1) Sulfur (2) Arsenic	1995 – September 1997
(1) Zinc Oxide (155 cubic feet)(2) Manganese Dioxide (75 cubic feet)	(1) Sulfur (2) Arsenic	October 1997 – June 1999
Manganese Dioxide (230 cubic feet)	Arsenic	June 1999 – October 2002
Manganese Dioxide (230 cubic feet)	Arsenic	October 2002 – 31 December 2002 (continuing)

Table 6.2-329C-40 Carbonyl Guard BedAdsorbents Used during Operation of LPMEOH[™] Demonstration Unit

Adsorbent Material	Targeted Catalyst Poison	Time in Service
Activated Carbon (6,300 pounds)	Iron, Nickel	April 1997 – November 1997
Activated Carbon (6,553 pounds)	Iron, Nickel	December 1997 – June 1999
(1) Manganese Dioxide (2,900 pounds)(2) Activated Carbon (2,200 pounds)	(1) Arsenic(2) Iron, Nickel	June 1999 – July 2000
Copper Oxide-impregnated Activated Carbon (4,892 pounds)	Arsenic, Sulfur, Iron, Nickel	August 2000 – July 2001
Copper Oxide-impregnated Activated Carbon (5,336 pounds)	Arsenic, Sulfur, Iron, Nickel	August 2001 – February 2002
Copper Oxide-impregnated Activated Carbon (5,073 pounds)	Arsenic, Sulfur, Iron, Nickel	March 2002 – June 2002
Copper Oxide-impregnated Activated Carbon (5,810 pounds)	Arsenic, Sulfur, Iron, Nickel	June 2002 – October 2002
Copper Oxide-impregnated Activated Carbon (5,225 pounds)	Arsenic, Sulfur, Iron, Nickel	October 2002 – 31 December 2002 (continuing)

(TCLP) was used to determine the metals and total concentration for organics; the results from the testing for various samples are provided in Environmental Monitoring Report Nos. 14, 19, and 22 (References 36, 39, and 44 in Section 12). Only one of the three samples exceeded the regulatory limit for arsenic. Even though the other material could have been incinerated, Eastman decided to classify the material as hazardous for arsenic (U.S. Environmental Protection Agency (EPA) Hazardous Waste Code D004) under the Resource and Conservation Recovery Act (RCRA) due to the uncertainty associated with the detection limit for arsenic. However, the incinerator did not accept material with this waste code. This led to the selection of a permitted hazardous waste landfill as the site to dispose of this spent adsorbent. Prior to landfilling, this facility stabilized the material to ensure that the leachate contained less than 5 ppmw arsenic.

6.2.4 Waste Oil

As part of the design of the LPMEOH[™] demonstration unit, an oil-water separator was included to collect any spill of the process mineral oil during operation or maintenance. During the demonstration program, oil was collected on a periodic basis and sent to incinerator equipment within the Eastman chemicals-from-coal complex for energy recovery. Table 6.2-4 shows the quantity of waste oil that was collected in the oil-water separator during the demonstration program for the LPMEOH[™] Process at Kingsport. During the design phase, the annual quantity of waste oil from the oil-water separator was expected to be 13,000 pounds; the average amount observed during the demonstration period should be used as the basis during the design of future plants.

Year of Operation	Waste Oil, pounds	Comments
April 1997 – March 1998	24,000	Onstream 02 April 1997
April 1998 – March 1999	24,000	
April 1999 – March 2000	42,000	
April 2000 – March 2001	6,000	
April 2001 – March 2002	0	
April 2002 – December 2002	48,000	
Annual Average	25,040	
Expected Value	13,000	Refer to Final EA for Project ^(b)

Table 6.2-4Waste Oil from LPMEOHTM Demonstration Unit

In addition, a one-time test was performed in June of 2002 to identify potential sources of iron contamination within the LPMEOH[™] Process. Mineral oil (in the absence of methanol synthesis catalyst) was heated in the presence of syngas, and sampling of the syngas stream was performed to determine the concentration of iron carbonyl. This quantity of oil (35,600 pounds) was also incinerated for energy recovery.

6.2.5 Recovered Distillation Liquids

As shown in Figure 2.4-1, a Crude-Grade Methanol stream containing about 25% of the methanol that was produced in the LPMEOHTM reactor was sent to existing distillation equipment within the Eastman chemicals-from-coal complex for further distillation. This stream contained a small quantity of the mineral oil that is used in the preparation of the methanol catalyst slurry. Table 6.2-5 shows the quantity of oil that was collected in the Eastman

Year of Operation	Oil Collected, pounds	Comments
April 1997 – March 1998	0	Onstream 02 April 1997
April 1998 – March 1999	178,269	
April 1999 – March 2000	34,100	
April 2000 – March 2001	116,533	
April 2001 – March 2002	221,855	
April 2002 – December 2002	110,557	
Annual Average	115,010	
Expected Value	324,000	Refer to Final EA for Project ^(b)

Table 6.2-5Oil Collected in Distillation Equipment from LPMEOH™ Process
distillation equipment during the demonstration period for the LPMEOHTM Process at Kingsport. The average annual generation rate is less than the projected rate as reported in the final EA for the LPMEOHTM Demonstration Project.

6.2.6 Waste Water

The chemistry for the production of methanol from syngas (refer to Section 2.1) also yields water as a byproduct. This water is separated from the Refined-Grade Methanol in the 29C-20 methanol rectifier column, and exits the LPMEOH[™] demonstration unit with the Crude-Grade Methanol stream. After further distillation in existing equipment within the Eastman chemicals-from-coal complex, the water from the LPMEOH[™] Process is combined with other streams and fed to the Eastman water treatment system. During the operation of the LPMEOH[™] demonstration unit, Eastman provided reports on the samples taken at the discharge point from the water treatment system to the Holston River. As noted in the Environmental Monitoring Reports (referenced in Section 12), there were no excursions for any of the parameters that were tested as required by the National Pollutant Discharge Elimination System (NPDES) permit for the Eastman chemicals-from-coal complex.

6.3 POTENTIAL ENVIRONMENTAL CONCERNS

Based upon the excellent environmental performance at the LPMEOHTM demonstration unit, the use of the technology in future applications should cause no significant impact on the local environment. Site-specific considerations should be made regarding the disposal options for the spent catalyst slurry (for example, the cost of and proximity to a catalyst reclaiming operation). In addition, the benefits of recovery and reuse of the oil streams from the oil-water separator and the distillation system (where used) should also be evaluated on a site-by-site basis.

6.4 SECTION 6 BIBLIOGRAPHY

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b. "Final Environmental Assessment for the Liquid Phase Methanol (LPMEOHTM) Project", DOE/EA-1029, June 1995.

c. "Commercial-Scale Demonstration of the Liquid-Phase Methanol (LPMEOH[™]) Process - Final Environmental Monitoring Plan", Prepared by Air Products Liquid Phase Conversion Co., L.P., DOE Cooperative Agreement No. DE-FC22-95PC93052, August 1996.

7.0 METHANOL PRODUCT-USE TESTING

7.1 INTRODUCTION

The LPMEOH[™] Process is a very effective technology for converting a portion of an IGCC electric power plant's coal-derived syngas to methanol, as depicted in Figure 7.1-1. The process has the flexibility to handle wide variations in syngas composition. It can be designed to operate in a continuous, baseload manner, converting syngas from oversized gasifiers or from a spare gasifier. Alternatively, the process can be designed to operate only during periods of off-peak electric power demand, consuming a portion of the excess syngas and reducing the electricity output from the combined-cycle power unit. In this scenario, the gasification unit continues to operate at full baseload capacity, so that the IGCC facility's major capital asset is always fully utilized.

In either baseload or cycling operation, partial conversion of between 20% and 33% of the IGCC plant's syngas is optimal, and conversion of up to 50% is feasible. The required degree of conversion of syngas, or the quantity of methanol relative to the power plant size, determines the design configuration for the LPMEOHTM unit. In its simplest configuration, syngas at maximum available pressure from the IGCC power plant's gasifier system passes once-through the LPMEOHTM unit and is partially converted to methanol without recycle, water-gas shift, or CO₂



Figure 7.1-1 Once-through Methanol Coproduction with IGCC Electric Power

removal. The unreacted gas is returned to the IGCC power plant's combustion turbines. If greater syngas conversion is required, different design options are available.

Methanol which is produced directly from CO-rich syngas has only 1 wt% H₂O in the condensed stream entering the first distillation column. Significant capital and operating costs (corresponding to a purification cost of \$0.02 to \$0.04 per gallon of methanol) can be avoided if stabilized methanol, which is produced via removal of dissolved gases such as CO₂ from the first column, can be directly used in some chemical and power systems. The Statement of Work from the Cooperative Agreement defines the scope of work for the studies associated with the testing of this as-produced methanol:

"The methanol product must be adequate for its intended use. Therefore, the economic upgrading of the methanol product will be demonstrated as part of Phase III, Task 2.1 (LPMEOHTM Process Demonstration Facility Operation – Methanol Operation), and the suitability of using the product will be demonstrated as part of Phase III, Tasks 3 (On-site Testing (Product Use Demonstration)) and 4 (Off-site Testing (Product Use Demonstration)). The product use demonstrations will include fuel use and chemical use product testing. The fuel use applications may include use in transit buses and car pools and for the generation of stationary power for public, industrial and electric utilities."

7.2 ON-SITE TESTING OF METHANOL

As noted in Section 1.2.4, the methanol that is produced at the LPMEOH[™] demonstration unit was used by Eastman in the production of methyl acetate and, ultimately cellulose acetate and acetic acid. Beginning with the startup of the LPMEOH[™] demonstration unit, Eastman monitored the quality of the Refined-Grade Methanol as determined by a set of in-house fitnessfor-use criteria. These criteria, as well as a summary of the results of the analysis of over 7.19 million gallons of Refined-Grade Methanol produced in 1997, are provided in Table 7.2-1. Occasionally, a lot tank sample fell outside an internal manufacturing limit. Through 31 December 1997, a total of 24 samples of refined methanol failed to meet one of the analytical tests. Refined-Grade Methanol is pumped from the methanol lot tanks within the battery limits of the LPMEOHTM demonstration unit to a primary methanol storage tank elsewhere within the Eastman complex. When results of a sample of Refined-Grade Methanol fall outside the fitnessfor-use criteria, Eastman evaluates the impact of this material on the contents of this primary storage tank. As long as the properties of the methanol in this tank stay within the methanol specification, the lot of Refined-Grade Methanol can be accepted. All 24 lots of Refined-Grade Methanol that have failed to meet all of the fitness-for-use criteria have been accepted in this manner. Over the entire 69-month operating program for the LPMEOH[™] demonstration unit, over 103.9 million gallons of methanol was produced, and Eastman accepted all of the available methanol for use in the production of methyl acetate, and ultimately cellulose acetate and acetic acid.

An opportunity to demonstrate the use of methanol directly from the LPMEOHTM Process in the production of methyl acetate occurred on 08 February 1998. During a planned five-day outage in Eastman's shift reactor, operations continued in the LPMEOHTM demonstration unit with a H₂:CO ratio in the reactor feed of 0.7, while Eastman's gas-phase methanol plant was forced to shut down. Furthermore, Eastman took the opportunity to shut down their methanol distillation

Table 7.2-1 Refined-Grade Methanol from LPMEOH[™] Demonstration Unit Quality and Fitness for Use (1997)

	1	2	3	4	5	6	7
Property	LIML	UIML	# Obs.	Avg.	Min.	Max.	# Out IMLs
Acetone, ppm		125	207	11.217	10	30	0
Appearance	pass	pass	207				3
Color by pcs		5	207	5	5	5	0
Decane, ppm		12	207	10.642	10	76	5
Ethanol, ppm		500	207	101.01	100	220	0
Water, % by Coulemetric			170	0.0094	0.0023	0.0571	
Water, % by Titration		0.03	96	0.0127	0.0023	0.0581	5
Acetic Acid, %		0.003	207	0.0004	0.0002	0.003	0
Methanol, %	99.9		207	99.985	99.92	100	0
Alkalinity, ppm		0.8	51	0.596	0.1	2.3	11
Specific Gravity @ 20/20	0.792	0.793	96	0.7928	0.792	0.793	0

Notes:

1) Terms:

Column 1 - Lower Internal Manufacturing Limit

Column 2 - Upper Internal Manufacturing Limit

Column 3 - Number of lot tank samples analyzed for this test

Column 4 - Average value of analytical test for all lot tank samples

Column 5 - Minimum value of analytical test for all lot tank samples

Column 6 - Maximum value of analytical test for all lot tank samples

Column 7 - Number of lot tank samples falling outside Internal Manufacturing Limit

 Analysis is performed on each lot with the exception of: Alkalinity - every 4th lot Specific Gravity - every other lot

3) The Coulemetric water test was replaced by a water titration in August of 1997.

equipment for maintenance, leaving no flow path for the Crude-Grade Methanol underflow from the second, rectifier column in the LPMEOHTM demonstration unit. However, because of the CO-rich feed, the rectifier column could be shut down for the production of stabilized methanol (98+ wt% methanol, <1 wt% H₂O). Approximately 12,000 gallons of stabilized methanol was loaded into trailers and shipped off-site for future product-use testing. During the remainder of this operating period, approximately 181,800 gallons (600 short tons) of stabilized methanol (99.3 wt% methanol, 0.3 wt% H₂O) was pumped directly from the LPMEOHTM demonstration unit to Eastman's methyl acetate plant, bypassing the normal route through the distillation equipment.

This methanol was considered acceptable by Eastman because of the low levels of all impurities. In particular, the H_2O concentration of the stabilized methanol was significantly lower than what is produced during operation on the normal Balanced Gas supply (typically 3-4 wt% H_2O). This is a result of operation on a CO-rich syngas with a relatively low CO_2 concentration. This test

was the first attempt to feed a stabilized methanol stream with low water content directly into one of Eastman's downstream chemical processes, providing another indication of the flexibility of the LPMEOHTM Process. Eastman's only alternative was to flare the syngas stream, which was undesirable both environmentally and economically.

7.3 OFF-SITE PRODUCT-USE TESTING OF METHANOL

The product-use test program attempted to develop broader market applications and commercial fuels comparisons by evaluating new utility dispersed electric power developments, and possibly new mobile transport engine developments. The objective of the product-use test program was to demonstrate commercial market applications for the stabilized methanol as a replacement fuel and as a fuel supplement. During the program definition phase, attempts were made to develop sites to test stabilized methanol in municipal, industrial, and utility applications and as fuel supplements for gasoline, diesel, and natural gas. These tests were envisioned based upon the U.S. energy market needs projected through the 2018 time period when the LPMEOHTM technology is expected to be commercialized.

Because the generation of electric power is not a feature of the demonstration project at Kingsport, the coproduction of methanol with electric power could not be demonstrated. However, the product-use test program was developed to provide data on potential applications of stabilized methanol in order to enhance the early commercial acceptance of central clean coal technology processing facilities, coproducing electricity and methanol to meet the needs of the local community. These applications required testing of the product to confirm its suitability.

A limited quantity of the stabilized methanol from the demonstration unit was made available for product-use tests. Product-use tests commenced during the first year of demonstration operations. As noted in Section 7.2, an initial inventory of approximately 12,000 gallons of stabilized methanol was produced at LPMEOHTM demonstration unit to supply the needs of the product-use test program. Analysis of the stabilized methanol from tests conducted using a CO-rich feed gas is shown in Table 7.3-1. Results from two Kingsport tests are quite comparable with results from a test campaign at the LaPorte AFDU. The difference in the concentrations of H_2O and higher alcohols of the two Kingsport samples is related to the age or activity of the catalyst and the amount of CO₂ in the reactor feed gas.

Table 7.3-1Stabilized Methanol Composition - CO-rich Feed Gas (H2:CO = 0.7)

	Kingsport Sample #1	Kingsport Sample #2	LaPorte AFDU Sample
Methanol	98.08	99.26	98.40
Ethanol	0.31	0.25	0.60
C3+	0.40	0.18	0.46
Water	1.22	0.31	0.54

The status of the seven projects which have been defined within the product-use test program is reviewed. Two categories of potential applications of stabilized methanol were under investigation, transportation systems and power generation systems.

7.3.1 <u>Transportation Systems</u>

A total of five vehicles have been tested on fuel blends made from stabilized methanol from the LPMEOHTM Demonstration Project. These tests, which have been performed at three different locations, were designed to determine if there are any differences in fuel economy, maintenance, or exhaust emissions when compared to the performance with fuels made with chemical-grade methanol.

7.3.1.1 Florida Institute of Technology Bus & Light Vehicles

Three of the vehicles are located at the Florida Institute of Technology, and all were tested on stabilized methanol from the LPMEOHTM Demonstration Project. A 1988 Chevrolet Corsica flexible-fuel vehicle (FFV) was driven a total of 4,455 miles on M-85 (85 vol% methanol/15 vol% gasoline) and M-100 (100 vol% methanol) fuel blends. Repairs were of a routine nature and independent of the type of methanol used or the fuel blend. A total of 32 refuelings were performed under the sponsorship of the product-use program, with an average fuel economy of 10.88 miles per gallon (mpg). When compared with the EPA rated gasoline mileage of 19.0 mpg, the methanol blends are less efficient on an energy density basis (methanol has a lower energy content per gallon than gasoline). Additional data on the base gasoline performance is necessary in order to complete this analysis.

The second vehicle is a 1993 Ford Taurus FFV which was acquired during the project. During the project, a total of 7,648 miles were accumulated on fuel blends with stabilized methanol content ranging from M-85 to M-15. Since the car was acquired in July of 1998, the car operated well; the fuel pump was replaced due to routine wear not related to the use of either stabilized methanol or chemical-grade methanol fuel blends. The fuel economy ranged from 14.68 mpg on M-85 to 19.81 mpg for M-15. This latter figure matched both the measured fuel economy with pure gasoline and the EPA rated gasoline mileage.

A bus which had been operated by the Jacksonville Transportation Authority was driven a total of 937 miles on M-100 during the project. Due to the costs associated with hiring a driver for this vehicle, no further tests beyond the initial shakedown of the bus were conducted.

7.3.1.2 ARCADIS Geraghty & Miller Flexible Fuel Vehicle

A second set of FFV trials was performed at ARCADIS Geraghty & Miller in a 1996 Ford Taurus FFV. During sponsorship by the product-use program, the FFV was driven a total of 3,206 miles on a M-85 fuel blended from chemical-grade methanol, and 3,479 miles were accumulated on M-85 produced from stabilized methanol. The fuel economy of about 16 mpg was the same for both fuel blends (Table 7.3.1-1).

Table 7.3.1-1ARCADIS Geraghty & Miller FFVFuel Economy Results

	Fuel Economy, mpg			
	Chemical -Grade M-85	M-85 w/ LPMEOH [™] Stabilized Methanol		
In-use	16.2	16.4		
Emission Test on Dynamometer	16.9	16.3		

Results of emissions testing on a dynamometer are presented in Table 7.3.1-2. The FFV exhibited higher emissions for total hydrocarbons, CO, CO₂, and methane for the stabilized methanol fuel blend; emissions of non-methane hydrocarbons and NO_x were higher for the M-85 fuel blended with chemical-grade methanol. For each of these parameters, emissions for both fuels were within the standards established by the State of California. Emissions of methanol and formaldehyde were greater for the stabilized methanol blend, and the readings for formaldehyde were above the California standard for both fuel blends (there is no standard for methanol emissions). It was concluded that further engine tuning would bring the formaldehyde emissions into compliance for both fuel blends.

Overall, the FFV operated by ARCADIS Geraghty & Miller exhibited similar performance characteristics for both fuel economy and emissions when the vehicle was operated on fuel blends from either stabilized methanol or chemical-grade methanol.

Table 7.3.1-2 ARCADIS Geraghty & Miller FFV Emissions Results

	Constituent Emissi	California	
	Chemical -Grade	M-85 w/ LPMEOH [™]	Emission
	M-85	Stabilized Methanol	Standard, grams/mile
Carbon Dioxide	334	346	No Standard
Carbon Monoxide	0.867	0.923	3.4
Nitrogen Oxides	0.138	0.121	0.4
Total Hydrocarbons	0.093	0.132	No Standard
Methane	<0.001	0.011	No Standard
Non-methane Hydrocarbons	0.120	0.118	0.25
Methanol	0.143	0.252	No Standard
Formaldehyde	0.0194	0.0337	0.015

7.3.1.3 West Virginia University Study of Stabilized Methanol in Transit Bus

West Virginia University (WVU) has developed a transportable laboratory facility which specializes in the measurement of emissions from heavy-duty vehicles. Figure 7.3.1-1 is a photograph of the laboratory, and one of the pieces of test equipment.

Under this effort, three buses operated on M-100 by Tri-Boro Coach in New York were tested. The buses were manufactured in 1993 and 1994 by Transit Motor Corp. The Detroit Diesel 6V92 methanol engines in the buses are rated at 253 horsepower. All buses were operated on the Central Bus District Cycle, which is intended to simulate the use of a transit bus in city service and is also used to rate the performance of new buses. One bus was also tested using a 5-mile cycle, which uses five different segments of acceleration, cruise, and deceleration.

The results from the average vehicle emissions for this product-use test are presented in Table 7.3.1-3. On average, emissions of hydrocarbons (primarily unburned methanol and aldehydes)

Figure 7.3.1-1 West Virginia University Transportable Heavy Duty Emissions Testing Laboratory



WVU mobile emissions laboratory



WVU mobile testing equipment

Table 7.3.1-3 West Virginia University - Tri-Boro Coach Emissions Results

Constituent Emissi	ion Rate, grams/mile	
Chemical -Grade M-85	M-85 w/ LPMEOH [™] Stabilized Methanol	New Diesel Engine w/ Diesel Fuel
2,773	2,811	2,611
10.41	9.60	4.9
5.60	5.11	30.1
5.39	6.88	0.13
0.21	0.24	0.24
	Constituent Emiss Chemical -Grade M-85 2,773 10.41 5.60 5.39 0.21	Constituent Emission Rate, grams/mile Chemical -Grade M-85 w/ LPMEOH TM M-85 Stabilized Methanol 2,773 2,811 10.41 9.60 5.60 5.11 5.39 6.88 0.21 0.24

and particulate matter were increased slightly when stabilized methanol is used as the fuel. Also shown for reference are recent test results as reported by the laboratory for a <u>new</u> diesel engine; methanol fuels offer substantial advantages in lower emissions of NO_x with minor changes in other components when compared with diesel fuel.

7.3.2 Power Generation Systems

As indicated in Figure 7.1-1, one of the primary applications for stabilized methanol derived from syngas streams rich in CO is in the generation of electric power. The more classical approach is to generate electricity in a methanol-fired gas turbine at the power plant during periods of peak power demand. In recent years, changes in the power generation and transmission industry and technological improvements in smaller turbine and fuel cell systems have spawned the concept of distributed power. A local power producing system will be able to provide the electricity for a small number of users (for example, the heating and air conditioning needs of an office complex or shopping center). Methanol, produced at a centrally located coal-fired IGCC plant, can be produced at prices which are competitive in these distributed power applications.

The product-use test program was designed to study the uses of stabilized methanol in a variety of power generation systems (turbines, fuel cells, generators).

7.3.2.1 West Virginia University Stationary Gas Turbine

This program was designed to compare the emissions and operational aspects of gas turbine operation on stabilized methanol compared with conventional liquid fuel. At WVU, a research aircraft was built using a GTC-85-72 gas turbine as the power source. This unit, which is rated at 235 horsepower in the form of compressor bleed air, was used in this study. Several of the turbine subsystems (fuel tank and pumps, speed and fuel flow controller, ignition system, bleed air load controller) had to be modified to accommodate the stabilized methanol fuel.

Testing has shown that the present configuration of the gas turbine does not allow operation on methanol at startup. Conventional fuel (in the form of Jet A) was used for initial turbine firing,

and a fuel emulsifier was added to assure a smooth transition from Jet A to stabilized methanol. At low power loads, flameout occurred during the transition to methanol; this appears to be the result of the higher heat of vaporization of methanol when compared with Jet A. Two glow plugs were added to overcome this problem. Initial testing was also performed without any additives in order to determine if the trace amounts of the inert oil from the LPMEOHTM Process were sufficient to provide the necessary lubrication for the fuel pump and controller. It was determined that the mineral oil levels in the stabilized methanol did not adequately lubricate these systems.

Testing was performed by the WVU Transportable Heavy Duty Emissions Testing Laboratory in order to compare the emissions generated from Jet A and stabilized methanol fuels. The results are provided in Figure 7.3.2-1, and demonstrate the capability of stabilized methanol to significantly reduce the levels of NO_x in gas turbine exhausts.

Following this work, a second set of performance tests and emissions measurements were performed with stabilized methanol and the selected lubricant to determine if any differences from the earlier testing could be determined. Figure 7.3.2-2 contains a representative set of results for a 2% mixture of the lubricant in methanol; emissions of NO_x were again substantially lower than for a kerosene fuel, and levels of CO, hydrocarbons, and particulates were also reduced.

The investigators recommend that additional work could be focused on optimizing the selection and concentration of the lubricant and solving the problems with operation at low power load.



Figure 7.3.2-1 West Virginia University Stationary Gas Turbine Emissions Test Results – 1998 Testing

Emissions during fuel type change over from Jet-A to



Figure 7.3.2-2 West Virginia University Stationary Gas Turbine Emissions Test Results – 2000 Testing

7.3.2.2 Aircraft Ground Support Equipment

In preparation for the possibility that combustion sources at U.S. Air Force bases may be required to reduce their emissions of NO_x , the Air Force had become interested in the evaluation of possible approaches to controlling NO_x emissions from aircraft ground support equipment (AGSE) diesel generators (part of a class of off-road power generation systems). These generators are very high NO_x emitters. It has been determined that the AGSE accounts for nearly 40 to 60 percent of a typical base's emissions of NO_x and the diesel generator accounts for 70 to 90 percent of the AGSE emissions. Replacement of these systems is expensive, and other alternative fuels (such as liquefied or compressed natural gas or liquefied petroleum gas) are not

feasible. One NO_x control approach applicable to these engines is the use of a diesel/water emulsion fuel in place of standard diesel fuel.

The use of diesel/water emulsions containing nominally 5-percent methanol have been shown to be effective in achieving nominally 40-percent NO_x reductions from some engines. However, the diesel/water emulsion experience base is quite limited, and the long-term effects of such emulsions on engine performance were unknown. Thus, there was a need to evaluate the effectiveness and the long-term performance of this NO_x control approach when applied to AGSE.

ARCADIS Geraghty & Miller developed a project to perform an extended evaluation of the use of a diesel/water/methanol emulsion, prepared using stabilized methanol from the LPMEOH[™]

Demonstration Project as the methanol component. Testing was planned on an Air Force flight line generator at Tyndall Air Force Base (AFB), Florida. Previous work supported by the Air Force Green AGSE Program developed an additive package that is effective in both stabilizing a diesel/water emulsion and preventing engine part corrosion. Two of these generators at Tyndall AFB were made available to this project. Of these, one was to be operated on the emulsion fuel during the evaluation, and the other was to be run on a military jet fuel (specification JP-8). JP-8 jet fuel is used in the mobility applications of the generators, and the Air Force has standardized the use of JP-8 in diesel engines to reduce the need to manage and maintain two fuel types and fueling systems. Engine performance and emissions testing were planned to be conducted before the start and after the end of this 6-month period to quantify emission reductions and performance impacts. Engine inspections were also planned, to address durability and corrosion issues.

The water-in-fuel (WIF) emulsion containing the stabilized methanol and additives was the emulsion fuel that was evaluated in these tests. The WIF emulsion was prepared to contain nominally 30 percent water, 5 percent methanol, and 1 percent additives, with the balance being JP-8.

When this project was originally planned, substantial support of the effort was offered by several Air Force and contractor organizations at Tyndall AFB. As the project proceeded, however, mission priorities of all the Air Force organizations supporting the evaluation changed to the point that further support of the project was not possible. As a result of these changes, support for the project could no longer be offered after May of 1999, and the project needed to be concluded. Up to that point, a series of initial performance and emissions tests had been completed.

In the initial performance and emissions tests that were completed, engine emissions of O_2 , CO, CO_2 , NO_x , and unburned hydrocarbons (UHC) were measured from engines fueled with both WIF and JP-8. Particulate emissions measurements were also performed using a dilution tunnel measurement technique. However, the 30-minute sampling time at each test load was not long enough to allow collecting measurable quantities of particulate on the sampling train filters.

Results of the tests were as follows:

- As shown in Figure 7.3.2-3, the use of the WIF emulsion reduced engine NO_x emissions by 21 to 57% over the engine load range, with the greater emission reductions achieved at lower engine loads. The International Organization for Standardization (ISO) cycle weighted average NO_x emissions were reduced 34% from 1,550 ppm at 15% O_2 with JP-8 fuel to 1,030 ppm at 15% O_2 with the emulsion.
- CO emissions from the engine did not vary significantly with load for the JP-8 fuel, and were nominally 500 ppm at 15% O₂. With the WIF emulsion, CO emissions were relatively constant at engine loads from full load to 50 percent load, but increased substantially at lower engine loads.



• Relative CO emissions for the two test fuels showed inconsistent behavior. For one series of tests, CO emissions with the WIF emulsion were lower than with JP-8 at engines loads of 50% or greater, but were higher at lower engine loads. ISO-cycle weighted average emissions for this test series were comparable for both fuels. For a second test series, CO emissions were substantially higher with the WIF emulsion than with JP-8 at all engine loads tested.

Stabilized methanol from the LPMEOHTM Demonstration Project was successfully used to produce the WIF emulsion which was used in this project. Although NO_x emissions reduction with the WIF were impressive, severe problems with cold starting and operation at low ambient temperatures were experienced. A few approaches to solve these problems were tried during this abbreviated project, but without success. Until the cold start problem is solved, WIF use in Air Force applications was not recommended.

7.3.2.3 Low-NO_X Stationary Microturbine Combustor

As originally proposed by ARCADIS Geraghty & Miller, the objective of the subject project was to demonstrate cost-effective volatile organic compound (VOC) destruction by a small gas turbine operating in a distributed power generation application fired with stabilized methanol from the LPMEOH[™] Demonstration Project. However, after an exhaustive search, no host site willing to participate in the project was identified.

At this point, a decision was made to shift the environmental focus of the project. California, as well as the EPA, regulates NO_x as an ozone precursor. As a consequence, California continues to pursue very aggressive NO_x control strategies to facilitate bringing California ozone nonattainment regions into attainment. Moreover, such strategies will become more commonplace in the Midwestern and Northeastern states in response to EPA's decision to implement a NO_x cap and trade program in both the Northeastern states as well as the Midwestern states that contribute to the ozone nonattainment status of regions of the Northeast via transported ozone.

Given these mandates, it was clear that any new distributed generation capacity will need to be low NO_x emitting units. In response to this need, Alzeta Corporation, with support from the California Energy Commission, the DOE's National Energy Technology Laboratory, and a number of gas turbine manufacturers, has been developing an advanced low NO_x surface stabilized combustor technology for stationary microturbines in distributed generation applications. The opportunity arose for the LPMEOHTM Demonstration Project to participate in this program and extend demonstration testing to stabilized methanol. Accordingly, a decision was made to redirect the project on VOC control to focus on completing a series of tests using stabilized methanol as a fuel for a low NO_x microturbine combustor targeted for use in a distributed generation application. In testing performed with natural gas fuel, it was possible to achieve combustor NO_x, CO, and UHC emissions approaching 2 ppm at 15 percent O₂. Parallel testing with stabilized methanol was performed in these tests to evaluate whether comparable performance could be achieved.

As shown in Figure 7.3.2-4, NO_x emissions with stabilized methanol from the LPMEOHTM Demonstration Project were comparable to and, for several conditions, lower than those with natural gas fuel. Results of the tests showed that combustor NO_x emissions could be held below 6 ppm at 15 percent O₂ over the range of combustor firing rates corresponding to turbine idle to full load. Emissions as low as 1 ppm at 15 percent O₂ were achieved at a number of test conditions, and were 3 ppm at 15 percent O₂ or lower for all but the highest load tested. The low NO_x emissions were achieved with CO emissions at 20 ppm (at 15 percent O₂) or lower. CO emissions when stabilized methanol was used as fuel were also comparable to those with natural gas fuel. In fact, CO emissions were 4 ppm at 15 percent O₂ or lower at all but low load (firing rate) and high load.

In summary, stabilized methanol from the LPMEOH[™] Process would seem to represent an acceptable liquid fuel for advanced low emission microturbines using the Alzeta combustor technology, offering emissions performance at or slightly better than the levels achieved with natural gas fuel.





7.3.2.4 University of Florida Fuel Cell

Over the past few years, progress has continued in the development of fuel cells, which can be used in either transportation or power applications.

While pure gaseous H_2 is the ideal fuel for fuel cell power systems, it is also relatively expensive, difficult to transport, and difficult to store for onboard transportation systems. Consequently, there has been, and is, considerable interest in utilizing liquid hydrocarbon fuels for transportation fuel cell applications. For a liquid fuel to be a feasible alternative to H_2 , it must be possible and practical to utilize a fuel processing system to transform the liquid fuel into a H_2 -rich gas. More specifically, the H_2 -rich gas must be compatible with long-term operation of the fuel cell system within which it would be utilized, with little or no degradation in performance. In earlier demonstrations with phosphoric acid fuel cells, chemical-grade methanol has been used successfully as the source of H_2 .

Clearly, the implication is that a suitable fuel-fuel processor combination must maintain the ability to provide almost complete conversion of the liquid fuel feedstock into an acceptable H_2 -rich gas over a long period of time and many cycles of operation. "Acceptable" H_2 -rich gas in this case means a maximum concentration of H_2 and virtually zero concentration of contaminants (primarily unreacted components from the liquid fuel) which would degrade the life and/or performance of the fuel cell stack. To meet this acceptability requirement, the fuel processor must also not undergo any significant decrease in its ability to convert the liquid fuel in a continuing and consistent manner. A prime mechanism for the fuel processor to experience a

significant decrease in conversion ability would be through a deactivation of the catalyst in a catalyst-driven processor such as the steam reformer.

Catalyst deactivation can occur for many reasons including excess temperature, rapid temperature transients, or the introduction of O_2 . However, with the introduction of a new liquid fuel such as stabilized methanol, it is obvious that deactivation can occur as a result of one or more contaminants (such as higher alcohols or the trace quantity of mineral oil from the LPMEOHTM Process) contained in the liquid fuel. Given the cost of distillation (\$0.02 to \$0.04 per gallon) of stabilized methanol to chemical-grade purity, a fuel cell test site of this potential source of H₂ was sought. The University of Florida had facilities and equipment in the Fuel Cell Research and Training Laboratory to perform this type of research, as well as two operational methanol-fueled fuel cell engines. As a result, an agreement was reached for the University of Florida to undertake the evaluation of the stabilized methanol as a suitable fuel for steamreformed methanol/phosphoric acid fuel cell systems.

Since it was obviously important not to damage the operational fuel cell engines, the procedure was to fabricate small steam reformers using catalyst pellets of the type (copper oxide-zinc oxide) typically used for methanol steam reformation. The apparatus was designed so as to allow determination of areas of deactivation if they occurred as a result of the stabilized methanol fuel. A parallel reformer operating simultaneously with chemical-grade methanol was used as a control.

Initial results showed very rapid degradation of the catalyst exposed to the stabilized methanol. Analysis of the results indicated that a likely cause of the deactivation was the presence of high boiling point components in the stabilized methanol feedstock. A potential scenario for this deactivation was a probable coating of the catalyst with trace quantities of the mineral oil which is used as the liquid medium in the LPMEOHTM reactor. Subsequent tests and modifications to the reformer unit were performed as well as the utilization of a higher temperature reformer catalyst to avoid rapid catalyst degradation. Higher temperatures did, as expected, result in less effect due to the mineral oil, but it also resulted in a more unfavorable balance among H_2 , methane, CO_2 , and CO in the reformate.

The results to date indicate that for the relatively low-temperature steam reforming of stabilized methanol, the trace mineral oil in the stabilized methanol would most likely have to be removed in order for it to be a suitable fuel. On the other hand, the limited higher temperature results indicate that possibly for auto-thermal reforming and probably for partial oxidation reforming, the stabilized methanol would be acceptable as a fuel. Methanol from the LPMEOHTM Process that is purified to chemical-grade specifications should be suitable for use in this application.

7.4 SUMMARY

Methanol from the LPMEOHTM demonstration unit was tested in a variety of chemical and fuel applications. A test methodology was developed by Eastman to satisfy the criteria for use of methanol in their chemicals-from-coal complex. Over the entire 69-month operating program for the LPMEOHTM demonstration unit, over 103.9 million gallons of methanol was produced,

and Eastman accepted all of the available methanol for use in the production of methyl acetate, and ultimately cellulose acetate and acetic acid.

Stabilized methanol from the project was also made available to seven test locations to study its feasibility as feedstock in transportation and power generation applications. Use in these applications is expected to enhance the flexibility of and revenue from IGCC electric power plants. In bus and FFV trials, stabilized methanol provided the same environmental benefits as chemical-grade methanol with no penalty on performance or fuel economy. Tests in a gas turbine and a diesel generator showed that levels of NO_x in the exhaust air can be lowered when stabilized methanol or methanol emulsions are used instead of conventional oil fuels. As with the chemical-grade methanol, lubrication additives will likely be required when stabilized methanol is fed to a gas turbine. Testing of stabilized methanol as the source of H₂ to a phosphoric acid fuel cell showed that conventional steam reforming catalysts are not compatible with the trace mineral oil present in the stabilized methanol; when auto-thermal reforming or partial oxidation reforming are used, the stabilized methanol would possibly be acceptable as a fuel. Methanol from the LPMEOHTM Process that is purified to chemical-grade specifications should be suitable for use in this application.

Product-use testing proved that coal-derived methanol, free of sulfur and other impurities, could be used one day as a replacement for petroleum in transportation, a peaking fuel in combustion turbines IGCC power plants, a H₂ source for small fuel cells, or as a chemical feedstock. Implementation of the LPMEOHTM technology and the use of stabilized methanol in transportation and power systems will add significant flexibility and dispatch benefits to IGCC electric power plants, which traditionally have been viewed as strictly a baseload power generation technology. Now, central clean coal technology processing plants, making coproducts of electricity and methanol, can meet the needs of local communities for dispersed power and transportation fuel. Methanol coproduction studies show that methanol can be produced at economically competitive levels from an abundant, non-inflationary local fuel source, such as coal. The coproduced methanol may be an economical H₂ source for small fuel cells, as a transportation fuel, and an environmentally advantaged fuel for dispersed electric power.

7.5 SECTION 7 BIBLIOGRAPHY

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8.0 LIQUID PHASE DIMETHYL ETHER (LPDME™) DESIGN VERIFICATION TESTING (DVT)

8.1 INTRODUCTION

The Statement of Work from the Cooperative Agreement defined a Secondary Objective for the LPMEOHTM Demonstration Project regarding the demonstration of Air Products' Liquid Phase Dimethyl Ether (LPDMETM) Process:

"Subject to Design Verification Testing (DVT), the Partnership proposes to enhance the Project by including the demonstration of the slurry reactor's capability to produce DME as a mixed co-product with methanol. The production of DME from synthesis gas is a natural extension of the LPMEOH[™] Process in that three reactions occur concurrently in a single liquid phase reactor, methanol synthesis, methanol dehydration and water-gas shift. This process enhancement can significantly improve the overall conversion of coal derived synthesis gas to a storable blend of methanol and DME...

"The enhanced (DME production) demonstration is complementary to ongoing studies being sponsored by DOE's Liquid Fuels Program... the enhancement will expand DOE's liquid fuels technology data base and will demonstrate the versatility of liquid phase synthesis.

"DVT is required to address issues such as catalyst activity and stability and to provide data for engineering design and demonstration decision making. The DVT plan will be coordinated with and utilize the resources of the DOE's Liquid Fuels Program as technology experts. The essential steps required for Project decision making regarding the methanol/DME co-product enhancement are:

- "DVT, including laboratory R&D to develop engineering data and to verify the market through engine tests and through market and economic study. (Phase I, Task 5).
- "DVT, including Process Development Unit Verification Testing utilizing the LaPorte AFDU for proof of concept tests. (Phase III, Task 6).
- "Final decision making to implement the full-scale methanol/DME co-product demonstration. (Phase III, Task 2.2).

"At the conclusion of each of the DVT steps, a joint Partnership/DOE decision will be made regarding continuation of methanol/DME demonstration."

The LPDME[™] Process uses a physical mixture of a commercial methanol catalyst and a commercial dehydration catalyst in a single slurry reactor to coproduce DME with methanol. This process provides high syngas conversion, efficient heat transfer, and directly converts a variety of feed gas compositions. There are three simultaneous reactions in this system, namely,

Methanol synthesis:	$CO + 2H_2 \Leftrightarrow CH_3OH$	(1)
Water gas shift:	$\rm CO + H_2O \Leftrightarrow \rm CO_2 + H_2$	(2)
Methanol dehydration:	$2CH_{3}OH \Leftrightarrow CH_{3}OCH_{3} + H_{2}O$	(3)

The methanol synthesis catalyst promotes Reactions 1 and 2, and the dehydration catalyst promotes Reaction 3. Both catalysts are micron-sized powders suspended in an inert hydrocarbon liquid, usually a mineral oil. The mineral oil acts as a temperature moderator and a

heat removal medium, transferring the heat of reaction from the catalyst surface via the liquid slurry to boiling water in an internal tubular heat exchanger. As a result of this capability to remove heat and maintain a constant, highly uniform temperature through the entire length of the reactor, the slurry reactor can manage the high syngas conversion per pass that can be achieved in the LPDMETM Process.

These capabilities make the LPDME[™] Process a potentially lower-cost conversion route to DME, especially when DME coproduction is added to a coal-based IGCC power plant. In this application, the overall reaction of syngas to DME is:

DME from CO-rich syngas: $3CO + 3H_2 \Leftrightarrow CH_3OCH_3 + CO_2$ (4)

This is the "once-through" CO-rich syngas concept for the LPDME[™] Process utilizing a single slurry reactor. Conversion per pass, with CO-rich syngas, can be higher than for the LPMEOH[™] Process. Methanol may also be produced, as a mixed coproduct with the DME, and can easily be separated and recovered. The separation of DME from CO₂ will be necessary for certain market applications.

8.2 DME MARKET STUDIES

A topical report^(a) provided information on the status of the current market for DME and an outlook on potential market developments through 2006. Over 50% of the demand for DME worldwide (currently between 100,000 and 150,000 metric tons per year) is for use as a projectile agent. DME is also being used as a methylating agent, and as a cleaning material for laboratory systems and some high precision, high value added applications, such as in electronics.

DME has potential applications as a chemical building block. In addition, since the physical properties of DME are similar to those of liquefied petroleum gas (LPG), it has been speculated that DME could be used in large scale power production, in home heating, in replacement of LPG for automobiles, and as a diesel fuel substitute or combustion supplement. In these applications, the specification for the purity of DME has been reported to be lower than the 99 wt% requirement for current uses, which may provide an opportunity for savings in capital and operating costs for product distillation. If these markets are developed, production capacity for DME will need to increase by an order of magnitude over current levels of about 143,000 metric tons per year. Projects to meet this future market are under development, including two large natural gas-based opportunities serving markets in India and Japan and a coal-based opportunity in China.

8.3 DVT ECOMONIC STUDIES

The economics studies, for once-through coproduction (with an IGCC power plant, for example) on syngas rich in carbon oxides, show that the LPDMETM Process will have an economic advantage greater than the LPMEOHTM Process. A once-through LPDMETM reactor is able to convert greater than 50% of such a syngas, whereas a once-through LPMEOHTM reactor can convert only about 30%. The economics, of course, depend upon the end-use (purity) of the

DME and upon the gasification plant's coproduct mix (amount of power, methanol, DME, etc.). The same liquid phase reactor design options to increase syngas conversion, such as feed gas compression and/or CO-rich gas recycle; are also applicable for the LPDMETM Process. So, the LPDMETM technology has the potential to improve on the \$0.05 to \$0.10 per gallon (methanol equivalent) advantage over the LPMEOHTM Process for the coproduction of DME to serve local markets.

As with the LPMEOHTM Process, gas-phase process technology must be considered as the economic competitor. The gas-phase DME process^(b) must run with H₂-rich syngas. In the IGCC coproduction flow sheet (shown in Figure 7.1-1), gas-phase technology is at an economic disadvantage, since separate shift and CO₂ removal are required. As is the case for methanol, inexpensive remote natural gas would therefore be the economic plant site choice for gas-phase technology. A comparison of coproduction of DME in an IGCC facility using the LPDMETM Process compared with DME imported from remote gas facilities, shows an advantage of 20-30% for locally produced DME relative to imported DME. The transportation cost to import DME is much higher than for methanol, and the LPDMETM coproduction advantage is even greater than that for the LPMEOHTM Process (vs. methanol import). Dehydration of imported methanol to make DME is not competitive. Therefore, for DME in local markets, coproduction using the LPDMETM Process should have a clear economic advantage over imported DME.

With H₂-rich syngas, the LPDMETM Process loses its (once-through, high conversion per pass) economic advantage. The overall reaction, with (> 2:1) H₂-rich syngas is:

DME from H₂-rich syngas: $2CO + 4H_2 \Leftrightarrow CH_3OCH_3 + H_2O$ (5)

Since water inhibits the methanol dehydration reaction, the slurry reactor must be staged, with water removal between stages. Staging could be by high ratio gas recycle, and/or with multiple reactors; but the once-through simplicity is lost. Therefore, it is unlikely that the LPDMETM Process would be developed for use in H₂-rich syngas applications.

A cost estimate of commercial-scale LPDME[™] plants has been performed. This work has helped quantify the targets for the laboratory research and development program (summarized below). From these studies, a commercially successful LPDME[™] system was defined for a Texaco-type syngas (35 vol% H₂, 51 vol% CO, 13 vol% CO₂) available at 500 psig. At a reactor operating pressure of 950 psig and a space velocity of 4,000 liters/hr-kg catalyst, the LPDMETM catalyst system must have a methanol equivalent productivity of 14 mol/kg catalyst-hr after 6 months of operation, producing at least 75% (by heating value) DME and 25% methanol.

8.4 DVT CATALYST STUDIES

A research and development program was initiated by Air Products in 1994, with DOE's sponsorship under the Liquid Fuels Program, to develop a catalyst system for the LPDMETM Process that met the economic targets. This work resulted in an invention of aluminum phosphate as the dehydration catalyst.^(c) The aluminum phosphate provided an active and stable catalyst system. The system met the program targets under standard test process conditions of an initial methanol equivalent productivity of 28 gmole per kg catalyst - hr, a CO₂-free, carbon

selectivity of 80% to DME, and stability of both catalysts equivalent to that of the methanol catalyst in the absence of the aluminum phosphate. Because the aluminum phosphate catalyst was not commercially available, a scaleup project was initiated with a commercial catalyst vendor to prepare a total of 800 pounds of aluminum phosphate to provide for two reactor charges and some additional material for testing.^(d) Eight 160 pound pilot batches of uncalcined catalyst were prepared. The reproducibility of the aluminum to phosphate (Al/P) ratio, an important physical property of the catalyst, from batch to batch was poor. Of these eight lots, two were qualified as vielding acceptable performance under the proposed LaPorte AFDU trial life test conditions. One of the batches was tested under conditions which allow direct comparison to the lab benchmark aluminum phosphate. This batch gave faster deactivation of the methanol catalyst than did the lab material, showing it to be of inferior quality. Two additional batches were prepared at half-dilution to alleviate any mixing problems which might have caused the poor reproducibility in the original eight batches. However, the Al/P ratio of both of these batches was high. One was tested, and shown to give unacceptable performance. All scaleup work on aluminum phosphate was suspended in November of 1998, as a commercially-available catalyst discussed below was identified which performed equivalently.

The deactivation rates of the methanol and the dehydration catalyst were correlated to reaction conditions. The correlations suggested a dependence of the deactivation rates on reaction conditions. The trends suggested by the correlations were confirmed by additional experiments. Better stability was obtained using the aluminum phosphate catalyst system by operating at favorable process conditions. The same improvement was also demonstrated with the gamma alumina catalyst system. However, there was a 10-20% trade-off in the productivity when shifting from the optimal conditions to the stable conditions for both the catalyst systems. As the gamma alumina was off-the-shelf commercial material, it was chosen for the LaPorte AFDU test. A methanol catalyst to gamma alumina ratio of 95:5 by weight was found to be optimum for catalyst stability.

8.5 <u>1999 LPDME™ PROCESS DESIGN VERIFICATION TEST RUN</u>

A demonstration of the production of DME by the LPDMETM Process was successfully completed at the LaPorte AFDU in October-November of 1999. A photograph of the LaPorte AFDU is provided in Figure 8.5-1. The demonstration was conducted at a pilot scale of 10 sT/D to evaluate the commercial viability of the LPDMETM Process. Based upon the improvement in the life of the catalyst system at the laboratory scale, the participants in the LPMEOHTM Demonstration Project agreed that a trial at the LaPorte AFDU was the appropriate step to demonstrate these improvements. This run was co-funded by the LPMEOHTM Demonstration Project and the DOE's Liquid Fuels program. The objectives of the run were to demonstrate the operation of the LPDMETM Process with improved catalyst life at a 10 sT/D scale, using commercially produced catalysts, obtain information to correlate the scaleup of catalyst aging from autoclave to bubble column, conduct process variable testing, and perform experiments to better understand the reactor fluid dynamics. A topical report^(e) was written detailing the testing at the LaPorte AFDU.

Figure 8.5-1 LaPorte Alternative Fuels Development Unit (AFDU)



The methanol catalyst was successfully activated at the beginning of operations with dilute H_2 , resulting in an expected H_2 uptake. The reduction was conducted with 3 vol% H_2 in N_2 at 67 psig reactor pressure. The heat-up proceeded from 200 to 464°F, as planned. The reduction appeared normal, and a cumulative uptake very close to the theoretical maximum value of 2.68 SCF H_2 /lb oxide was obtained. The reduction was essentially complete at 390°F or 17 hours onstream. Nuclear Density Gauge measurements indicated an average gas holdup of 36.8 vol%, with a catalyst concentration of 40.1 wt% at 392°F during the reduction. The catalysts appeared to have good initial activity, with DME and methanol productivity slightly exceeding expectations. This confirmed that the catalyst activation was proper. The initial DME production rate was 5.1 sT/D compared to an expectation of 4.8 sT/D, while the methanol production rate was 3.6 sT/D vs 3.5 sT/D expected.

The LaPorte AFDU was operated for 25 days to compare catalyst aging in a pilot-scale slurry bubble column reactor with that in a laboratory autoclave. The proportion of two catalysts corresponded to a 95:5 methanol to dehydration catalyst ratio by weight. The methanol and dehydration catalyst activities, expressed as the ratio of the rate constant at any point in time to the rate constant for a freshly reduced catalyst (as determined in the laboratory autoclave), are plotted in Figure 8.5-2. These normalized rate constants were estimated based on a reaction model developed from laboratory data. After the expected initial aging, the catalysts appeared to be stabilizing, but there was significant scatter in the data. Gas chromatographic as well as sampling problems were discovered with methanol analysis that required use of liquid balance

Figure 8.5-2 1999 LPDME[™] Operating Run at LaPorte AFDU – Estimated Catalyst Activity



for calculations, contributing to the scatter. The problems were corrected in two days. The initial deactivation rate appeared high: 0.08% per hour (2% per day) for methanol catalyst with a

0.05% per hour standard error and 0.03% per hour (0.6% per day) for dehydration catalyst with a 0.009% per hour standard error. It was decided to extend the aging test to get a better estimate on catalyst deactivation rate, which was the main objective.

The LaPorte AFDU data appeared to follow the autoclave trends, with somewhat higher conversions than the autoclave throughout the test period. The methanol productivity level remained relatively constant, while the DME productivity showed a slight decline. The DME and methanol production rates through the run are shown in Figure 8.5-3. The DME production rate declined from 5.1 sT/D to 4.1 sT/D in 25 days onstream, while methanol production showed a scatter within the 3.1 to 3.8 sT/D range through the run. The scatter in data decreased significantly after the gas chromatograph and sampling problems were resolved (350 hours onstream). The performance also met the productivity and selectivity targets. The initial methanol equivalent productivity exceeded the target of 28 gmole/hr-kg catalyst, and the DME selectivity was at the target of 65% on a carbon basis. The reactor operated in a hydrodynamically stable manner, with uniform temperature profile and gas holdups. Differential pressure measurements indicated about 42 vol% gas holdup and 36 wt% catalyst concentration. Gas-, liquid-, and solid-phase mixing was studied using radioactive tracer injections. Washington University in St. Louis analyzed the tracer data as part of a DOE program on hydrodynamics of slurry bubble column reactors.

Figure 8.5-3 1999 LPDME[™] Operating Run at LaPorte AFDU – Production Results



The equipment for this test was operated for the first time following a 4-year hiatus; the operations were smooth. The initial startup was very quick, with the baseline condition reached in 12 hours after the introduction of syngas. A re-start after a syngas outage took only 4 hours. The speed of the startup and re-start demonstrates the ease and flexibility of the slurry technology in response to changes.

The major accomplishments of this run are summarized below:

- Commercial viability of the LPDME[™] Process was successfully evaluated on a 10 sT/D scale, using commercially produced catalysts. The LaPorte AFDU was operated for 25 days to compare catalyst aging in a pilot scale slurry bubble column with that in a laboratory autoclave. The catalyst life study was extended in favor of a planned process variable study to obtain additional data on catalyst aging. Hydrodynamic information was obtained at the baseline conditions by conducting a detailed survey of the reactor with radioactive tracer injections.
- The deactivation rate for both the catalysts was calculated to be 0.7% per day. This result was lower than the 1.2% per day which had been calculated for both the methanol synthesis and dehydration catalysts during experiments in the autoclave. The rate of deactivation was slightly higher than 0.5% per day rate which was achieved for the LPMEOHTM Process after 3 weeks of operation at the LaPorte AFDU in 1988/89. The methanol productivity remained relatively constant throughout the test period, while the DME productivity showed a slight decline. These trends were consistent with observations from the laboratory. The standard error for the methanol catalyst deactivation rate was high (0.25% per day) due to initial scatter in the data. The scatter decreased significantly after problems in the sampling and

8-7

analytical system were discovered and resolved at 350 hours on stream. The dehydration catalyst activity data have better statistics, with a standard error of 0.06% per day.

• The 1999 design verification test of the LPDMETM Process at the LaPorte AFDU represented a significant step forward in the development of the technology. The 0.7% per day rate of catalyst deactivation which was achieved during this campaign is a large improvement over the 4% per day rate of deactivation which was calculated from autoclave studies prior to the initial test of the LPDMETM Process at the LaPorte AFDU in 1991.

8.6 DME DVT CONCLUSIONS

During a Project Review Meeting on 12-13 January 2000, the preliminary results from the LPDMETM DVT were presented. A table which compared the performance of the LPDMETM catalyst system during the DVT with results from the recent work in the laboratory autoclave and the economic targets for the LPDME[™] Process was presented (Table 8.6-1). The results from the laboratory work had indicated that the targets for catalyst productivity and life can be met at lower selectivity to DME. The results from the DVT were applied to a cost estimate for a commercial-scale LPDME[™] plant (Figure 8.6-1). The case that was considered was the retrofit of an existing Texaco gasifier in China which is currently used in the production of ammonia. The results of this initial work were that, for this syngas (35 vol% H₂, 51 vol% CO, 13 vol% CO₂) available at 500 psig, the DME costs are competitive with LPG in China (\$7 to \$8 per MMBtu). The cost of syngas assumes constant utilization for each plant size, and economies of scale are indicated by changes in the conversion and separation costs. The target DME cost can likely be achieved at larger plant sizes by extrapolating the costs to the 1,100 sT/D production rates. The sensitivity of the cost of production of DME to the life of the LPDMETM catalyst system was shown for the results from LaPorte (0.7% per day deactivation), as well as the impact of changes to the value for catalyst life on the process economics.

As stated in the 30 July 1997 letter which transmitted Air Products' recommendation to proceed with LPDMETM DVT, the basis for commercialization of the technology must come from:

- 1) the results from testing of the LPDMETM catalyst system at the LaPorte AFDU; and
- 2) the reactor performance (methanol catalyst activity and life, hydrodynamics, and heat transfer) from the LPMEOH[™] demonstration unit.

After discussing the results from the LPDMETM DVT activities and the ongoing performance results from Kingsport, the meeting participants agreed that the available resources should be directed toward improving the catalyst performance for the LPMEOHTM Process during the remaining time within the operating program. The results of this economic analysis were forwarded to the DOE's Liquid Fuels Program in order to provide direction to the ongoing catalyst development efforts.

Table 8.6-1LPDMETM Goals and Performance Results (Laboratory and AFDU)

	Liquid Fuels Program Goals	Commercial Targets	Laboratory Results (July 1997)	Laboratory Results (June 1999)	AFDU Results (Oct./Nov. 1999)
Catalyst Productivity, mol/kg catalyst-hr (MeOH- equivalent)	> 28 (Initial Productivity)	> 14 (productivity for aged catalyst)	28 (Initial Productivity)	28 (Initial Productivity)	30.5 (Initial Productivity)
Catalyst Selectivity	DME Selectivity > 80% (% Carbon, CO ₂ -free)	DME = 75%, Methanol = 25% (heating value basis)	DME Selectivity = 79% (% Carbon, CO ₂ -free)	DME Selectivity = 65% (% Carbon, CO ₂ -free)	DME Selectivity = 66 (start) - 61 (end)% (% Carbon, CO ₂ -free)
Catalyst Life	 > 50% Initial Productivity after 1000 hours 	Target Productivity after 6 months of operation	57% of Initial Productivity after 1000 hours	61% of Initial Productivity after 1000 hours	Calculated Target Productivity after 3-1/3 months of operation

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

Figure 8.6-1 LPDME[™] Process Economics



TEXACO-TYPE SYNTHESIS GAS

8.7 SECTION 8 BIBLIOGRAPHY

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9.0 ECONOMICS

9.1 INTRODUCTION

A topical report on the economics of coproduction of methanol in an IGCC power plant was published as part of the LPMEOHTM Demonstration Project.^(a) This study was used as the basis for the capital and operating costs for the LPMEOHTM Process that are published within this report. The assumptions for key economic parameters are provided:

<u>Capital costs</u> for equipment, valves, and instrumentation are calculated using costs from the 260 sT/D LPMEOHTM demonstration unit as the basis (calendar year 1995). Different multipliers are used to scale the costs. For example, costs associated with methanol distillation to the desired grade of methanol might scale directly with methanol production rates, while some of the costs within the LPMEOHTM reactor system may depend upon the volumetric gas flowrate.

<u>Construction costs</u> assume the same schedule (15 months) and site preparation as the LPMEOHTM Demonstration Unit.

<u>Freight and Miscellaneous</u> refer to project execution costs, again based upon the costs from the LPMEOHTM Demonstration Project.

<u>Air Products Process Studies</u> refer to initial process evaluations which need to be performed in order to optimize the process cycle selected for a particular IGCC application. For example, optimization of feed and recycle compression and an evaluation of the need for water injection are performed during these studies.

<u>Air Products Technical Package</u> refers to the final process design package for a site-specific application. This includes all heat and material balances, detailed equipment specifications for the LPMEOHTM reactor, and process specifications for other equipment items.

<u>Project Engineering</u> refers to the overall management of the engineering, design, and construction work, based upon the costs from the LPMEOHTM Demonstration Project.

<u>Design Engineering</u> refers to the detailed design work (civil, mechanical, instrument and electrical), again based upon the costs from the LPMEOHTM Demonstration Project.

<u>Field Engineering</u> refers to all field engineering services, including construction management at the site. These costs are again based upon the costs from the LPMEOHTM Demonstration Project.

Travel and Living costs are included as a line item.

<u>Reserve</u> is provided as a line item, but is not included in these evaluations.

License Fee is provided as a line item, but is not included in these evaluations.

Once these items are computed, an allowance for <u>Owner's Cost</u> (25% of the equipment cost for the evaluated system) is provided. This is a nominal cost of project execution which includes costs for:

- a) Initial charge of catalyst
- b) Initial supply of chemicals and lubricants
- c) Any applicable taxes and insurance
- d) Cost of land
- e) Legal and other overhead costs

Costs associated with <u>Methanol Storage</u> and $\underline{CO_2 \text{ Removal}}$ (if required) are separated from other equipment items, since the costs are site-specific and are highlighted in this manner to provide a potential customer with the opportunity to optimize the required storage.

The sum of these items is the Capital Cost for the LPMEOHTM Facility. The costs are also tabulated by process area (excluding Owner's Cost, Methanol Storage Cost, and CO₂ Removal Costs) to provide information on the relative costs of the different process steps, including all valves, instruments, and associated construction costs.

A calculation of the <u>Plot Area Required</u>, based upon the layout from the LPMEOH[™] demonstration unit, is also given.

Operating costs are evaluated as annual costs (\$/Year) and as the cost per gallon of methanol. The annual cost is computed from the cost per unit time and the annual capacity factor. The final cost of methanol is computed as follows:

$$Gallon Methanol = Annual Costs ($/Year)365 * LF * 303 * X$$

where:

LF = Annual Capacity Factor X = Methanol Production Rate, short Tons per Day

The conversion factor 303 gallons per short ton of methanol is used in this equation.

The following is a summary of the components of the Operating Costs:

<u>Syngas</u> refers to the value (lower heating value [LHV] basis) of the syngas produced in the IGCC facility. When a specific application is defined, the cost (\$ per MMBtu) can be specified.

<u>Unreacted Gas</u> refers to the value (LHV basis) of the syngas returned from the LPMEOH[™] plant to the IGCC facility. When a specific application is defined, the cost (\$ per MMBtu) can be specified.

<u>Power</u> used by the LPMEOHTM Process is evaluated at \$0.04/kWh, which is assumed to be the electricity costs from the adjacent IGCC facility. Categories of power consumers (compressors, pumps, etc.) are tabulated separately.

<u>Steam</u> is a credit from the LPMEOHTM Plant; a typical value of \$4.00/1,000 pounds of steam production (including the cost for boiler feed water and the credit for condensate return) is used for the 200 psig steam produced by the LPMEOHTM Reactor.

<u>Cooling Water</u> is evaluated at a 20°F temperature rise across all cooling water heat exchangers. A cost of \$0.12/1,000 gallons of cooling water is used in this study.

Other <u>Miscellaneous Utilities</u> (such as instrument air and N_2) are calculated based upon the costs from the LPMEOHTM Demonstration Project.

<u>Catalyst, Chemicals and Lubes</u> refers to the estimated costs for the methanol synthesis catalyst, process mineral oil, and other lubricants. These values are based upon the costs from the LPMEOHTM Demonstration Project.

<u>Zinc Oxide</u> is used as the absorbent for the H_2S which is present with the fresh feed syngas or is produced by the <u>COS (carbonyl sulfide) Hydrolysis</u> catalyst within the LPMEOHTM Process. Each of these consumables is evaluated based upon present costs for each material.

<u>Operating Labor</u> is calculated from the original budgetary forecast for the LPMEOHTM Demonstration Project (which was met during the operating program).

<u>Maintenance</u> costs are evaluated based upon an annual budget of 2% of the total capital cost for the evaluated LPMEOHTM Facility.

<u>Property Taxes and Insurance</u> are estimated at an annual budget of 1.5% of the total capital cost for the evaluated LPMEOHTM Facility.

<u>Overhead</u> is computed at a nominal rate of 15% of the operating costs (not including the costs for syngas or the credit for the unreacted gas).

Recovery of <u>Capital Cost</u>, <u>Depreciation</u>, and <u>ROI</u> (Return on Investment) are computed based upon a 15 year depreciation and a ROI of about 14%, which results in a total charge of 20% of the investment per year.

9.2 ESTIMATED PROCESS CAPITAL COSTS

Table 9.2-1 provides a summary of the estimated capital costs for a 500 sT/D LPMEOHTM plant that coproduces a fuel-grade methanol (a minimum of 97 wt% methanol, a maximum of 1 wt% water, 1.5 wt% higher alcohols, and 0.5 wt% mineral oil from the LPMEOHTM Process) from CO-rich syngas (H₂:CO = 0.68) in a coal-based IGCC power plant. These costs were originally

Table 9.2-1Estimated Facility Investment Summary for Once-Through 500 sT/D LPMEOH™ Plant

INVESTMEN COMPRESS LPMEOH [™] E VALVES & IN CONSTRUC FREIGHT & I AIR PRODUC AIR PRODUC PROJECT EI DESIGN ENC FIELD ENGII TRAVEL & LI RESERVE LICENSE FE	T BREAKDOWN ION EQUIPMENT ISTRUMENTS TION MISCELLANEOUS CTS PROCESS STUDIES CTS TECHNICAL PACKA NGINEERING GINEERING VEERING	GE		<u>MM-\$</u> \$0.00 \$5.09 \$3.35 \$10.40 \$0.37 \$0.22 \$0.97 \$1.63 \$3.88 \$1.15 \$0.29 \$0.00 not incl
OWNER'S C METHANOL	OST STORAGE /AI	25% of Equipment 30 days	Sub-total Turnkey Plant 5.0 MM Gallons	\$27.35 \$1.27 \$2.46 \$0.00
			TOTAL CAPITAL	\$31.08
	LPMEOH TM = 0.11	Storage = 5.13	TOTAL	5.24
INVESTMEN	T BY AREA			
<u>AREA</u>	INVESTMENT		DESIGN BASIS	<u>MM-\$</u>
A B C D E F G	REACTOR LOOP & CAT FEED COMPRESSION RECYCLE COMPRESSI FRONT-END GAS CLEA COMMON EQUIPMENT SATURATOR DISTILLATION RESERVE AIR PRODUCTS PROCI AIR PRODUCTS TECHN LICENSE FEE	TALYST REDUCTION	500 sT/D 0 BHP 0 BHP 14552 LB-MOL/HR 500 sT/D 500 sT/D 0.0 % Sub-total Turnkey Plant	\$18.60 \$0.00 \$3.00 \$1.85 \$0.00 \$2.71 \$0.00 \$0.22 \$0.97 not incl \$27.35

Notes: 1) Syngas (H_2 :CO = 0.68) containing 5 ppmv COS is available at 1,000 psig.

2) Equipment from LPMEOH[™] demonstration unit, escalated to 2002\$, was used as basis.

published in the topical report on economics (referenced above). The capital cost of the LPMEOHTM demonstration unit, excluding the equipment items that were not used during the demonstration period (as described in Section 3.1), was used as a basis.

9.3 PROJECTED OPERATING AND MAINTANANCE COSTS

Table 9.3-1 provides a summary of the projected operating and maintenance costs for a 500 sT/D LPMEOH[™] plant in the IGCC plant location. The operating and maintenance costs for the distillation system can be separated from the other conversion costs to yield a methanol conversion cost (in this case, a methanol conversion cost of \$0.155 per gallon, including capital recovery, was calculated); this terminology will be used in developing the impact of changes in operating parameters on costs in Section 9.5.

Table 9.3-1 Projected Operating and Maintenance Costs for Once-Through 500 sT/D LPMEOH[™] Plant

MeOH Production Annual Load Factor Distillation to Fuel Grade Methanc Syngas H ₂ :CO = 0.68	500 90	sT/D %				
					M-\$/Yr	\$/Gal
Syngas (LHV)	1434	MMBTU/hr	\$0.00	/MMBTU	\$0	\$0.000
Unreacted Gas (LHV) Power	(995)	MMBTU/hr	\$0.00	/MMBTU	\$0	\$0.000
Feed Compressor	0	kW	\$0.04	/kWh	\$0	\$0.000
Recycle Compressor	0	kW	\$0.04	/kWh	\$0	\$0.000
Pumps, Heaters, etc.	382	kW	\$0.04	/kWh	\$121	\$0.002
Distillation	43	kW	\$0.04	/kWh	\$13	\$0.000
CO2 Removal	0	kW	\$0.04	/kWh	\$0	\$0.000
LP Steam, 100 psig	3515	lb/hr	\$3.00	/M-lb	\$83	\$0.002
MP Steam, 200 psig	(49300)	lb/hr	\$4.00	/M-lb	(\$1,555)	(\$0.031)
C Water, 20oF Delta T	652	gpm	\$0.12	/M-gal	\$37	\$0.001
Misc Utilities					\$240	\$0.005
Catalyst, Chemicals & Lubes					\$989	\$0.020
Sulfur Removal, Zinc Oxide	104	M-lb/yr	\$3.58	/lb	\$337	\$0.007
COS Hydrolysis Catalyst	6	M-lb/yr	\$3.63	/lb	\$18	\$0.000
Operating Labor					\$733	\$0.015
Maintenance	2%	of Investment	t/yr	_	\$622	\$0.012
				Sub-Total	\$1,638	\$0.033
Property Taxes, Insurance	1.5%	of Investment	t/yr		\$466	\$0.009
Overhead	15%	of Oper Costs	s (Less F	eed)	\$252	\$0.005
Capital Costs, Depr & ROI	20%	of Investment	/yr	_	\$6,212	\$0.125
		Methanol Co	nversion	Cost, Total	\$8,569	\$0.172

Notes: 1) Syngas (H_2 :CO = 0.68) containing 5 ppmv COS is available at 1,000 psig.

2) Costs from LPMEOH[™] demonstration unit, escalated to 2002\$, were used as basis.

9.4 SUMMARY OF PERFORMANCE AND ECONOMICS

The capital cost summary for a 500 sT/D LPMEOHTM plant (Table 9.2-1) was developed based upon the design of and operating results from the LPMEOHTM demonstration unit. The scaleup of the size of the LPMEOHTM reactor (7.5' internal diameter, 260 sT/D design, greater than 300 sT/D demonstrated performance) can be managed; a topical report was issued on the construction of the reactor at the LPMEOHTM demonstration unit and the lessons learned that can be applied to the fabrication of larger diameter vessels.^(b)

The operating cost summary (Table 9.3-1) was developed using the results from the 69-month operating program at the LPMEOHTM demonstration unit as the basis. Given the excellent onstream performance (availability of 97.5%), a highly reliable system can be designed for future plants. In particular, the elimination of the slurry return pumps within the LPMEOHTM Process (refer to Section 3.1-1) was a key factor in achieving high on-stream time and lower maintenance costs. The basis for catalyst consumption is a catalyst deactivation rate of 0.4% per day, which was readily achieved during Kingsport Catalyst Campaign 4 (refer to Section 5.1.4). All other parameters (such as steam production from the LPMEOHTM reactor and steam consumption for distillation) were demonstrated during the operating phase of the LPMEOHTM Demonstration Project.

9.5 EFFECT OF VARIABLES ON ECONOMICS

Cases were developed to illustrate the impact of key operating variables on the methanol conversion cost (as defined in Section 9.3). The space velocity in the LPMEOH[™] reactor was selected based upon an analysis contained in the topical report on the economics of coproduction of methanol with electric power. To illustrate the sensitivity, two syngas supply pressures (500 psig and 1,000 psig) were used.

9.5.1 Plant Size

Figure 9.5-1 shows the impact of methanol plant size on the methanol conversion cost for coalderived syngas available at 1,000 psig. The once-through process yields the lowest conversion cost; the case at 500 sT/D was used as the basis for the capital and operating costs that were provided in Sections 9.2 and 9.3.



Figure 9.5-1 Effect of Methanol Plant Size on Methanol Conversion Cost

9.5.2 Sulfur Content

In this analysis, the fresh syngas was assumed to contain 5 ppmv of COS. To remove this catalyst poison the COS is first hydrolyzed over a metal oxide catalyst and then the H_2S produced is removed with a zinc oxide bed. COS was selected to represent sulfur species in the syngas instead of H_2S because of the higher capital cost associated with the hydrolysis step (operating costs for absorption of H_2S and COS after hydrolysis are the same on a sulfur weight basis). Figure 9.5-2 shows the effect of an increase to 20 ppmv of COS in the syngas feed on the methanol conversion cost at 1,000 psig syngas supply pressure. There is an increase of \$0.05 to \$0.08/gal in the methanol conversion cost at lower syngas conversion, as a large amount of gas must be treated. As the amount of conversion is increased (independent of the methanol conversion technology), the effect on cost becomes more modest.



Figure 9.5-2 Effect of Syngas Conversion on Methanol Conversion Cost

9.5.3 Inert Content

In this analysis, the fresh syngas was assumed to contain 1 vol% inerts. Increased levels of inerts can result from such parameters as the O_2 purity from the air separation unit feeding the coal gasifier. The greater the syngas conversion, the greater becomes the cost penalty of inert build up in the synthesis loop. Figure 9.5-3 shows the effect of an increase to 10% inerts in the feed syngas on the methanol conversion cost at 500 psig syngas supply pressure. For the once-through case (Case 5-A) there is a \$0.044/gal increase in cost along with a small reduction in conversion. With a 1:1 recycle at 1,250 psig (Case 5-H), the cost increase becomes less, \$0.015/gal. For the gas-phase process, the effect is much greater. In the case shown the inerts in the synthesis loop were limited to 35% to limit the impact on conversion due to lower partial pressure of reactants; this increases the methanol conversion cost by \$0.027/gal and greatly reduced the syngas conversion to 77%.
Figure 9.5-3 Effect of Inerts Concentration on Methanol Conversion Cost



9.5.4 Capacity Factor

The depreciation and return on investment have a significant impact on the methanol conversion cost. These are magnified by the capacity factor, as this term is inversely related to the unit cost of methanol (\$ per gallon). All of the figures and tables presented to this point in this Section have used a capacity factor of 90%. An analysis was performed to determine the change in methanol conversion cost when the capacity factor is lowered to 70%. This is more representative of the coproduction of methanol with electric power in an IGCC load-following application. When syngas is available at 500 psig, the effect of lowering the capacity factor from 90% to 70% is to increase the methanol conversion cost by \$0.05 to \$0.06 per gallon of methanol. When the feed syngas is available at 1,000 psig, the methanol conversion cost increases by \$0.04 to \$0.05 per gallon of methanol when the capacity factor is lowered from 90% to 70%.

9.6 SECTION 9 BIBLIOGRAPHY

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10.0 COMMERCIALIZATION POTENTIAL AND PLANS

10.1 MARKET ANALYSIS

10.1.1 Applicability of the Technology

The LPMEOHTM Process is a very effective technology for converting a portion of the H₂ and CO in an IGCC electric power plant's coal-derived syngas to methanol. The process is very flexible in being able to process many variations in syngas composition. The LPMEOHTM Process can be used with an IGCC power plant to provide the once-through methanol production as depicted in Figure 7.1-1. The process can be designed to operate in a continuous, baseload manner, converting syngas from oversized gasifiers or from a spare gasifier. The process can also be designed to operate only during periods of off-peak electric power demand to consume a portion of the excess syngas and allow the electricity output from the combined-cycle power unit to be reduced. In this latter circumstance, the gasifiers continue to operate at full baseload capacity, so the IGCC facility's major capital asset is fully utilized. In either baseload or cycling operation, partial conversion of between 20% and 40% of the volume of H₂ and CO in the IGCC power plant's syngas is optimal on an economic basis, and conversion of up to 50% is feasible.

A simplified process flow diagram for the LPMEOHTM Process design options is shown in Figure 10.1-1. This shows several once-through LPMEOHTM Process design options, as described more fully in the Public Design Report (Final Report - Volume 1) and in a topical report on the economics of methanol coproduction as an add-on to an IGCC power plant.

In its simplest configuration, part or all of the CO-rich syngas at its maximum available pressure from the IGCC power plant (Stream 1) is passed once, without recycle, through the LPMEOHTM reactor. The unreacted syngas (Stream 3) is returned to the IGCC power plant's combustion turbines.

If greater amounts of syngas conversion are required, different design options are available:

- a) Use of Gas Recycle;
- b) Once-Through, with Feed Gas Compression;
- c) Once-Through, with Water Addition;
- d) Once-Through, with Water Addition and CO₂ Removal; and
- e) Staged Liquid Phase Reactors.

With any of these options, there is still no need for upstream stoichiometric adjustment of the feed gas by the water-gas shift reaction and CO_2 removal, so the simplicity of once-through CO-rich gas processing is retained.

The design configuration for the LPMEOHTM Process depends upon the degree of conversion of syngas (or the quantity of methanol produced relative to the power plant size).







The features of the LPMEOHTM Process that satisfy the requirements in the coproduction application can also be applied to other carbon-rich feedstocks such as petroleum coke and other environmentally disadvantaged fuels. The LPMEOHTM Process has also shown the potential to meet the economic requirements for large, natural gas-based methanol production plants.^(a)

10.1.2 Market Size and Barriers

Table 10.1.2-1 provides an assessment by the Methanol Institute of the forecast of the market for methanol through 2007.^(b) The decrease in demand for methanol to produce MTBE is expected to be offset by increases in other basic chemicals such as formaldehyde and methyl methacrylate. Beyond this analysis, the potential exists for future market growth for methanol as a source of H_2 for fuel cell systems.

As noted in Section 10.1.1, a key market for the LPMEOHTM Process is the coproduction of methanol with electric power in an IGCC power plant. Using the DOE's High Fossil Electricity Technology Case that relates to the output of DOE's Fossil Energy program, the potential market for IGCC power plants in the United States could exceed 60 GW by 2020.^(c) Assuming that coproduction could be a candidate for 10-20% of this total, the opportunity can exist for significant market participation for the LPMEOHTM Process.

The methanol marketplace is currently undergoing a period of transition due to the planned elimination of MTBE by several states in the United States as well as the development of large (greater that 5,000 metric tons per day), off-shore methanol plants utilizing low-cost natural gas as feedstock.

Future markets will depend on such factors as the rate of growth of the IGCC power generation

Table 10.1.2-1 Market Forecast for Methanol

(Thousands of Metric Tons; Source: Methanol Institute)

SUPPLY	2003	2004	2005	2006	2007
Capacity Available	35,680	38,617	40,692	39,944	40,910
Production	30,427	31,016	31,774	31,885	32,196
Excess Capacity	5,253	7,601	8,918	8,059	8,714
Operating Rate	85%	80%	78%	80%	79%
DEMAND					
Formaldehyde	11,000	11,390	11,708	12,029	12,386
МТВЕ	7,506	7,428	7,197	6,598	6,207
Acetic Acid	3,339	3,374	3,649	3,735	3,826
Dimethyl Terephthalate	590	590	603	618	632
Methyl Methacrylate	892	923	979	1,004	1,030
Gas/Fuels	859	867	896	946	988
Solvents	1,285	1,313	1,342	1,379	1,411
Others	4,956	5,131	5,401	5,577	5,716
TOTAL DEMAND	30,427	31,016	31,774	31,885	32,196

business, global methanol market prices, and government incentives aimed at encouraging IGCC and/or IGCC coproduction projects, consistent with DOE's Vision 21. Methanol coproduction with IGCC power generation, using the LPMEOHTM Process, will continue to benefit from the economies of scale provided by the gasification plant, and result in lower methanol costs than for a standalone methanol plant with the same methanol production capacity. Methanol transport economies, compared with imported methanol based on low cost stranded natural gas, can also be achieved by serving local inland markets. Outside the United States, most of the current interest in LPMEOHTM technology has come from China, because of China's abundance of coal, historic dependence on domestic chemical production, and the amenability of the LPMEOHTM Process to coal-based syngas. In the longer term, the development of clean coal projects in China utilizing IGCC technology would provide a platform for additional interest in the LPMEOH[™] Process. In the U.S., interest in LPMEOH[™] technology will also depend on the development and timing of the IGCC industry, which in turn will depend on a number of factors, notably, natural gas price and availability, further improvement in IGCC economics, new environmental regulations further affecting coal-based power generation, and government incentives.

10.1.3 Economic Comparison with Competing Technologies

As noted in Section 9, a topical report was written during the performance period that compared the cost of methanol as produced from the LPMEOHTM Process and from a conventional gasphase process as applied to a generic 500 sT/D methanol plant as part of an IGCC coproduction facility. The cost of methanol was calculated as the sum of three terms: the methanol conversion cost (which includes the fixed and operating costs for the methanol unit), the distillation cost, and the syngas cost from the IGCC facility. A proprietary cost estimation screening program was used to calculate the methanol conversion cost and the distillation cost from the LPMEOHTM Process and the gas-phase process for various syngas supply pressures and on-stream factors. The methanol conversion cost from the LPMEOHTM Process was \$0.02 to \$0.07 per gallon lower than from the gas-phase methanol process.

A major component of the methanol conversion cost in an IGCC complex is the cost to distill the as-produced methanol product in order to meet the final specification. It is typical for methanol to be stabilized (either by distillation or by deep flashing) to remove volatile components (such as CO₂) and permit shipment and transport in atmospheric vessels. Beyond stabilization, other distillation may be necessary so that the final methanol product meets the specification for the designated end-use. There are two grades of methanol product (Chemical-Grade AA and Fuel-Grade) that could be used in downstream chemical or power applications. These grades of methanol differ in the amounts of water and higher alcohols that are present in the final product. In particular, the Fuel-Grade product has a water specification of 1 wt%, while the Chemical-Grade AA methanol has a maximum water content of 0.1 wt%.

The LPMEOHTM Process, which can directly process coal-derived syngas which is rich in CO, produces a methanol product with nominally about 1 wt% water. Whereas, gas-phase methanol synthesis results in as-produced methanol stream with 2-20 wt% water, depending on the amount of CO₂ in the syngas which is converted to methanol and water. This resulted in lower purification cost for the LPMEOHTM Process for the Fuel-Grade product. By applying the same

cost estimation screening program, the distillation cost to produce Fuel-Grade methanol from the LPMEOHTM Process which directly utilizes CO-rich syngas was about \$0.02 per gallon less than from the gas-phase methanol process.

Sensitivity studies performed as part of this Report indicated that the magnitude of the advantage in the methanol conversion cost for the LPMEOHTM Process when compared with the conventional gas-phase process is increased when:

- a) the syngas is rich in CO,
- b) syngas is available at higher pressures,
- c) only modest syngas conversion to methanol is required,
- d) syngas is available with low H₂S and COS content,
- e) inerts in the syngas (such as N_2 in the O_2 from the air separation unit feeding the gasifier) are relatively high, and
- f) Fuel-Grade Methanol is required.

Table 10.1.3-1 summarizes several operating scenarios for a baseload 500 sT/D LPMEOHTM Facility, when compared with the more complex 500 sT/D gas-phase methanol facility. As in the prior discussion, the baseload coproduction assumes that the syngas is normally available 90% of the time and syngas is charged at \$4.50 per MMBtu. Cost advantages of \$0.04 to \$0.11 per gallon of methanol can be realized by utilizing the LPMEOHTM Process.

10.2 COMMERCIALIZATION PLAN

During the development phase for the LPMEOHTM technology, a team within Air Products was assembled to manage the various programs needed to achieve commercial readiness. With the conclusion of the LPMEOHTM Demonstration Project, these efforts have achieved this goal. Future commercial activities will be managed from the Worldwide Hydrocarbon Equipment business unit.

Any future offerings will be based on the results from both the 4-month proof-of-concept run at the LaPorte AFDU and the operation at the LPMEOH[™] demonstration unit. Future process flow

Table 10.1.3-1Baseload Coproduction, 500 sT/D, Fuel Grade MethanolCO-rich Syngas (H2:CO = 0.68) @ \$4.50 per MMBtu

Syngas Feed Pressure, psig	<u>500</u>	<u>500</u>	<u>1,000</u>	<u>1,000</u>
Flow Scheme	Once-Through	1:1 Recycle	Once-Through	1:1 Recycle
Syngas Conversion, %	34	46	34	46
Methanol Cost, \$/Gal	0.548	0.567	0.483	0.529
LPMEOH [™] Advantage	0.055	0.036	0.109	0.063
over gas-phase, \$/Gal				

diagrams will derive from the as-built flowsheet from the LPMEOHTM demonstration unit. Catalyst performance has improved since the completion of the second in-situ catalyst activation in June of 2002, and provides an indication of the potential for improved economics for the LPMEOHTM Process. Future project opportunities will likely be based on coal-derived syngas in either IGCC facilities or standalone coal-to-chemicals complexes, or syngas that is produced from environmentally disadvantaged feedstocks (such as petroleum coke). Facility availability at the LPMEOHTM demonstration unit has been excellent since initial startup, and consistent with the high demands that accompany these capital-intensive projects.

10.3 SECTION 10 BIBLIOGRAPHY

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- c. "Forecasting the Benefits of DOE Programs for Advanced Fossil-Fuel Electricity Generating Technologies: The EIA High Fossil Electricity Technology Case", U.S. DOE Office of Fossil Energy, National Energy Technology Laboratory, Office of Systems and Policy Support, October 2002.

11.0 CONCLUSIONS AND RECOMMENDATIONS

The LPMEOHTM Demonstration Project accomplished the objectives set out in the Cooperative Agreement with DOE for this Clean Coal Technology project. Methanol operation began on 02 April 1997, and startup was completed in two days. Nameplate production of 80,000 gallons per day (260 sT/D) was achieved within four days, and production rates exceeding 115% of design were achieved within six days. Overall availability of the LPMEOHTM demonstration unit was 97.5%, and the longest operating period without interruption of any kind was 94 days. Over 103.9 million gallons of methanol was produced, including a one-month maximum of 2.5 million gallons. Eastman accepted all of the available methanol for use in the production of methyl acetate, and ultimately cellulose acetate and acetic acid.

The slurry bubble column reactor was successfully scaled from the 10 sT/D LaPorte AFDU to the 260 sT/D LPMEOHTM demonstration unit. The LPMEOHTM reactor did not exhibit any hydrodynamic instability, and the maximum catalyst loading of 151% of the design value of about 40,000 pounds (corresponding to a slurry concentration of greater than 49 wt%) was achieved. The gas sparger at the inlet to the LPMEOHTM reactor was replaced with two modified units (using the same design principles as the original device), and stable performance was realized when a continuous flush of condensed process oil (in this case, by gravity) was utilized.

Catalyst performance (as determined by the rate of catalyst deactivation using an in-house kinetic model) improved during the demonstration test period. Initially high deactivation rates using Balanced Gas (greater than 1% per day vs. the design basis of 0.4% per day from a 4-month proof-of-concept run on CO-rich syngas derived from natural gas at the LaPorte AFDU [1988/89]) at the design operating temperature of 250°C were attributed to the presence of post-construction debris in the reactor system. During the second catalyst campaign that began in December of 1997, the rate of catalyst deactivation was calculated to be 0.6 to 0.7% per day at a reactor temperature of 235°C. The practice of catalyst withdrawal and addition (removing a portion of the spent catalyst from the LPMEOHTM reactor and adding fresh catalyst in 2,000 pound batches) to maintain the desired methanol production rate was successfully demonstrated.

In determining the reasons for the higher rate of catalyst deactivation, the presence of certain elements (in particular, arsenic and sulfur) was identified from samples of spent catalyst, spent adsorbent materials from catalyst guard beds, and syngas entering the LPMEOHTM demonstration unit. A commercially available copper oxide-impregnated activated carbon was identified as having capacity to remove arsenic (in the form of arsine) within the existing carbonyl guard bed that was originally designed for removal of metal carbonyls at ambient temperature.

In parallel, laboratory testing was performed to develop a procedure to activate (or reduce) a full charge of methanol synthesis catalyst within the LPMEOHTM reactor (so-called in-situ activation). During the operating program, this feature was identified as having significant commercial interest, as capital cost savings of about 10% could be realized by elimination of the separate processing equipment for the catalyst withdrawal/addition procedure. Also, operating the LPMEOHTM reactor at low temperatures at the outset and slowly increasing temperature with

time to maintain a constant production of methanol (called temperature programming) was believed to have the potential to extend the life of the methanol synthesis catalyst. Following the first in-situ activation of methanol synthesis catalyst in August of 2001 (during which initial performance of the catalyst was about 70% of the value expected from the laboratory testing), the concept of temperature programming was demonstrated (temperature was gradually raised from 216°C to 250°C in order to maintain the desired methanol production rate). Following changes to the storage conditions of the catalyst slurry prior to reduction, the in-situ activation procedure was successfully practiced in June of 2002, resulting in the catalyst meeting the initial performance targets.

During the final six months of operation under the demonstration test plan, the average rate of catalyst deactivation was calculated to be 0.17% per day, which was ½ of the design basis for the LPMEOHTM demonstration unit. The reactor was maintained at a constant low temperature of 215°C during this period, and the carbonyl guard bed adsorbent (copper oxide-impregnated activated carbon) was replaced with fresh material after 3 months of operation (including a novel thermal treatment that was applied after two months to extend the useful life of the adsorbent). Because the impact of catalyst cost on the economics of methanol production is significant (reducing the rate of catalyst deactivation by up to ½ of the current basis can realize a 25 percent reduction in variable cost), this performance demonstrated the potential of the LPMEOHTM process to meet the targets for catalyst consumption utilizing coal-derived syngas.

Although the host site for the LPMEOHTM demonstration unit was a chemical production facility, testing was able to be performed to simulate the performance of the LPMEOHTM Process in an IGCC power plant. A total of 78 days of testing at a variety of H₂:CO ratios were performed to evaluate the effect of syngas compositions expected from coal-based IGCC units on the performance of the LPMEOHTM Process. All of the results met the expectations for methanol production rate based upon earlier studies in the laboratory. Catalyst performance for the LPMEOHTM reactor with lower H₂:CO ratios were not significantly impacted based on the calculated rate of catalyst deactivation. On/off and ramping studies were also completed, and the LPMEOHTM demonstration unit was capable of meeting the ramping requirements for an IGCC system by achieving the goal of a minimum 5% change in design flow per minute. Tests of the ability of the LPMEOHTM reactor to operate in an electrical load-following environment were also successful.

An off-site, fuel-use test program was conducted to demonstrate the suitability of stabilized (degassed) methanol product as a transportation fuel and as a fuel for stationary applications for small modular electric power generators for distributed power. Stabilized (degassed) methanol produced from CO-rich syngas in the LPMEOHTM Process can have less than 1 wt% water, and has the potential to be used directly in these applications. Stabilized methanol from the LPMEOHTM Demonstration Project was made available to seven test locations. Successful demonstration of the use of stabilized methanol in these applications can enhance the flexibility of and revenue from IGCC plants. In bus and FFV trials, stabilized methanol provided the same environmental benefits as chemical-grade methanol with no penalty on performance or fuel economy. Tests in a gas turbine and a diesel generator showed that levels of NO_x in the exhaust air were lowered when stabilized methanol or methanol emulsions are used instead of conventional oil fuels. As with chemical-grade methanol, lubrication additives will likely be

required when stabilized methanol is fed to a gas turbine. Testing of stabilized methanol as the source of H_2 to a phosphoric acid fuel cell has shown that conventional steam reforming catalysts are not compatible with the trace mineral oil present in the stabilized methanol; when auto-thermal reforming or partial oxidation reforming are used, the stabilized methanol would possibly be acceptable as a fuel. Methanol from the LPMEOHTM Process that is purified to chemical-grade specifications should be suitable for use in this application.

As a secondary objective to the LPMEOH[™] Demonstration Project, design verification testing was completed to evaluate capability of the slurry bubble column reactor to produce DME as a mixed coproduct with methanol using the LPDMETM Process. In addition to its current uses, DME has potential applications as a chemical building block. Also, since the physical properties of DME are similar to those of LPG, it has been speculated that DME could be used in large scale power production, in home heating, in replacement of LPG for automobiles, and as a diesel fuel substitute or combustion supplement. Economic targets on the production cost of DME from the LPDME[™] Process were established, and a 25-day design verification test was performed at the LaPorte AFDU in 1999. The rate of catalyst deactivation for both the methanol synthesis and dehydration catalysts was calculated to be 0.7% per day, which was a large improvement over the 4% per day rate of deactivation which was calculated from autoclave studies prior to the initial test of the LPDME[™] Process at the LaPorte AFDU in 1991; these results were slightly higher than the results (0.5% per day) for the LPMEOH[™] Process from the first 3 weeks of the 4-month proof-of-concept run at the LaPorte AFDU in 1988/89. Follow-up economic analysis showed that the target DME cost of \$7 to \$8 per MMBtu (equivalent to costs for LPG in China) can likely be achieved at plant sizes approaching 1,100 sT/D. The program participants agreed that additional work on DME should be deferred so that all resources could be directed toward improving the catalyst performance for the LPMEOHTM Process during the remaining time within the operating program.

The commercial application of the LPMEOH[™] technology with the greatest long-term potential continues to be the coproduction of methanol within an IGCC power plant. The ability of the slurry reactor to ramp rapidly and to start/stop easily makes it amenable to peak shaving scenarios in an IGCC power plant. Of significant interest is that relatively small quantities (30,000 gallons per day) of methanol can be coproduced from coal at world scale (greater than 600,000 gallons per day) economics. Parametric studies were performed to determine the production cost of methanol from the LPMEOH[™] Process in an IGCC power plant. The process can be designed to operate in a continuous, baseload manner, converting syngas from oversized gasifiers or from a spare gasifier. The process can also be designed to operate only during periods of off-peak electric power demand to consume a portion of the excess syngas and allow the electricity output from the combined-cycle power unit to be reduced. In this latter circumstance, the gasifiers continue to operate at full baseload capacity, so the IGCC facility's major capital asset is fully utilized. In either baseload or cycling operation, partial conversion of between 20% and 40% of the volume of H₂ and CO in the IGCC power plant's syngas is optimal on an economic basis, and conversion of up to 50% is feasible. Methanol conversion costs of about \$0.15 per gallon were determined for a coal-based IGCC plant using the once-through configuration; when the cost of syngas is added, the price of methanol is less than \$0.50 per gallon. As the economics for IGCC systems improve, the economics for coproduction of

methanol will continue to look attractive in local markets when compared with methanol that is produced in off-shore locations from natural gas.

Further reductions to the cost of syngas from a modern coal gasification system will increase the market opportunities for coproduction of chemicals such as methanol from the LPMEOHTM Process. Developments of and advancements in the processes for the removal of trace contaminants in coal-derived syngas will extend catalyst life and lead to lower methanol conversion costs. Continued studies of the hydrodynamics of slurry bubble column reactors would facilitate the use of larger diameter reactors, which will improve the economies of scale for all potential feedstocks (coal, natural gas, environmentally disadvantaged fuels).

12.0 OTHER PROJECT REPORTS AND PUBLICATIONS

(since the Public Design Report [Final Report - Volume 1])

Note: Reports or publications on specific topics are included in individual chapters where appropriate.

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- 3. Technical Progress Report No. 32, April June 2002.
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- 9. Technical Progress Report No. 26, October December 2000.
- 10. Technical Progress Report No. 25, July September 2000.
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APPENDIX A

ANALYTICAL MEASUREMENTS FOR LPMEOHTM DEMONSTRATION UNIT

Stream No.	10	19
Description	CO Gas to LPMEOH [™]	Distillation Fuel Gas
	Demonstration Unit	
PFD Material Balance Point No.	10	
Gas Chromatograph Point No.	2	8
Reported in:		
EMRs (Compliance)		
EMRs (Supplemental)	X (Summary Data)	
Technical Progress Reports	Х	Х
Test Series Report (one time)		
Temperature	Х	Х
Pressure	Х	Х
Flow Rate	Х	Х
Composition		
LPMEOH [™] Gas Analysis (vol%)	Х	Х
Hydrogen		
Carbon Monoxide		
Nitrogen		
Carbon Dioxide		
Methanol		
Dimethyl Ether		
Methane		
Oxygen (Argon)		
Trace (by difference)		
Analysis Frequency	Continuous	Continuous
Sampling Technique	Piped to GC	Piped to GC
Analytical Method	GC	GC

Stream No.	20	30
Description	H_2 Gas to LPMEOH TM	Balanced Gas to LPMEOH TM
1	Demonstration Unit	Demonstration Unit
PFD Material Balance Point No.	20	30
Gas Chromatograph Point No.	3	1
Reported in:		
EMRs (Compliance)		
EMRs (Supplemental)	X (Summary Data)	X (Summary Data)
Technical Progress Reports	X	Х
Test Series Report (one time)		
Temperature	Х	Х
Pressure	Х	Х
Flow Rate	Х	Х
Composition		
LPMEOH [™] Gas Analysis (vol%)	Х	Х
Hydrogen		
Carbon Monoxide		
Nitrogen		
Carbon Dioxide		
Methanol		
Dimethyl Ether		
Methane		
Oxygen (Argon)		
Trace (by difference)		
Analysis Frequency	Continuous	Continuous
Sampling Technique	Piped to GC	Piped to GC
Analytical Method	GC	GC
Notes:		

a. Stream 20 was not available during operation of the LPMEOH[™] demonstration unit (refer to Section 3.1.3).

Stream No.	109	120
Description	Reactor Feed Gas	Reactor Section Effluent Gas
PFD Material Balance Point No.	109	120
Gas Chromatograph Point No.	5	6
Reported in:		
EMRs (Compliance)		
EMRs (Supplemental)	X (Summary Data)	
Technical Progress Reports	X	Х
Test Series Report (one time)		
Temperature	Х	Х
Pressure	Х	Х
Flow Rate	Х	Х
Composition		
LPMEOH [™] Gas Analysis (vol%)	Х	Х
Hydrogen		
Carbon Monoxide		
Nitrogen		
Carbon Dioxide		
Methanol		
Dimethyl Ether		
Methane		
Oxygen (Argon)		
"Other" (by difference)		
Analysis Frequency	Continuous	Continuous
Sampling Technique	Piped to GC	Piped to GC
Analytical Method	GC	GC

Stream No.	148	149
Description	Main (Plant) Purge	Recycle Gas
PFD Material Balance Point No.	148	149
Gas Chromatograph Point No.	7	7
Reported in:		
EMRs (Compliance)		
EMRs (Supplemental)	X (Summary Data)	
Technical Progress Reports	X	Х
Test Series Report (one time)		
Temperature	X	Х
Pressure	Х	Х
Flow Rate	Х	Х
Composition		
LPMEOH [™] Gas Analysis (vol%)	Х	Х
Hydrogen		
Carbon Monoxide		
Nitrogen		
Carbon Dioxide		
Methanol		
Dimethyl Ether		
Methane		
Oxygen (Argon)		
"Other" (by difference)		
Analysis Frequency	Continuous	Continuous
Sampling Technique	Piped to GC	Piped to GC
Analytical Method	GC	GC

Stream No. Description PFD Material Balance Point No.	204 Methanol to Distillation 204	216 Refined-Grade Methanol 216
Gas Chromatograph Point No.	N/A	N/A
Reported in: EMPs (Compliance)		
EMRs (Complemental)		X (Summary Data)
Technical Progress Reports	Х	X
Test Series Report (one time)		
Temperature	Х	Х
Pressure	Х	Х
Flow Rate	Х	Х
Composition		
Methanol Analysis (wt%)	Х	Х
Acetic Acid		
Water		
Decane		
Methanol		
Acetone		
Ethanol		
Analysis Frequency	Weekly	Daily
Sampling Technique	Grab	Grab
Analytical Method	GC/IR/TC	GC/IR/TC

Stream No.	242
Description	Crude-Grade Methanol to
PFD Material Balance Point No.	242
Gas Chromatograph Point No.	N/A
Reported in:	
EMRs (Compliance)	
EMRs (Supplemental)	X (Summary Data)
Technical Progress Reports	Х
Test Series Report (one time)	
Temperature	Х
Pressure	Х
Flow Rate	Х
Composition	
Methanol Analysis (wt%)	Х
Acetic Acid	
Water	
Decane	
Methanol	
Acetone	
Ethanol	
Analysis Frequency	Weekly
Sampling Technique	Grab
Analytical Method	GC/IR/TC

Test Methods for Sampling of Syngas

Test methods were developed to determine the average concentration of arsenic (reported as arsine), iron, and nickel in syngas streams; these techniques were typically applied to the fresh feed gas entering the LPMEOH[™] demonstration unit.

Gas-phase arsenic concentrations in a syngas feed were determined using a modified version of the National Institute for Occupational Safety and Health (NIOSH) Method 6001 for "Arsine" in which arsenic was trapped on a column of activated carbon. A diagram of the sampling set-up is given in Figure A.1. Sample gas was purged through a charcoal packed into a stainless steel tube for a predetermined length of time and flow. The sampling flow was set with the fine metering valve as illustrated. The charcoal was then washed with dilute nitric acid and arsenic was determined by Inductively Coupled Plasma Atomic Emission Spectroscopy (ICP-AES). The limit of detection was 1-5 ppbv arsenic (as arsine).

Gas-phase iron carbonyl concentrations in a syngas feed were determined using a modified NIOSH method 6007 for "Nickel Carbonyl," which is similar to the method for arsine previously described. Sample gas was purged through a packed bed of charcoal in a stainless steel tube for a predetermined length of time and flow. The charcoal was then washed with dilute nitric acid and the iron content was determined by ICP-AES. The limit of detection was relatively high due to the baseline presence of iron in charcoal, with a typical limit of 10 ppbv iron (as iron carbonyl).



Figure A-1 Syngas Sampling Apparatus

A gas scrubbing technique, using concentrated nitric acid as the scrubbing solution, was also utilized for determination of the concentration of iron carbonyl. Sample gas was purged through a set of bubbling impingers containing nitric acid scrubbing solutions for a predetermined length of time and flow. The nitric acid solution was then analyzed and the iron content was determined by ICP-AES. The limit of detection was generally in the range of 3-5 ppbv iron (as iron carbonyl).

APPENDIX B

SAMPLE MATERIAL BALANCES FROM LPMEOHTM DEMONSTRATION UNIT

RUN NO:	K5-7/2	IT	ILE: Recheck Test 1 Conditions	0		
Balance F	Period: Start Date End Date	02jul97 07: 03jul97 07:	00:00	Time From St	art of Run (days) Start End	86.7 87.7
Reaction	Conditions Temperature (°F) Pressure (psig) Space Velocity (sL/kg-hr) Vg (inlet)	481 707 8827 0.70		Slurry Data	Catalyst Weight (lb oxide) Slurry Concentration (wt %) Slurry Level (ft) Gas Holdup (vol %) Gassed Slurry Volume (ft3)	19,500 26.5 58.4 45.4 2459
Performa	nce Results Raw MeOH Production (ton/day) Raw MeOH Production (ton/day) Syngas Utilization (SCF/lb MeOH) Catalyst Life (eta) Catalyst Life (eta) CO Conversion to MeOH (%) CO Conversion to M2 (%)	195.5 195.6 0.54 0.54 30.7 32.5) (gas measurements) (liquid measurements)	Energy Balan	ce Steam Production (lb/hr) Steam Drum Pressure (psig) Steam Import/Export (lb/hr) Reactor O-T-M Conversion (% LHV) Wetted Tube Length (ft) Heat Transfer Area (ft2) Reactor Overall U (BTU/hr-ft2-F)	12027 373 1446 19.6 54.7 2120 166
	Syngas Conversion (% LHV) Syngas Usage (BTU/gallon MeOH Recycle Ratio MeOH Productivity (gmol/kg-hr) Rxr Volumetric Productivity (ton/dź Sparger "K"-value	77.4 () 69,80 3.22 3.22 25.36 ty-ft3) 0.080	8	Atom/Mass B	alance Closure (% of reactor inlet) C H O N Total Mass	99.14 98.92 100.75 99.65 100.60
Liquid Product Analysis (wt%)	7/2	REFINED 2 7/3 00 7:00	PRODUCT	7/2 19:00	CRUDE PRODUCT 7/3 7:00	
	Methanol 99.5 Ethanol 0.0 Water 0.0 Oil 0.0 Total 100.0	99.99 1 0.01 8 0.02 0 0.00 01 100.0	~ 5	85.62 0.25 14.02 0.15 100.04	86.58 0.25 13.08 0.15 100.06	

SAMPLE MATERIAL BALANCE – BALANCED GAS CASE (02 JULY 1997) **APPENDIX B-1**

Kingsport LPMEOHTM CCT Project

B-2

Final Report: Volume 2 (June 2003)

APPENDIX B-1 SAMPLE MATERIAL BALANCE – BALANCED GAS CASE (02 JULY 1997) (cont'd)

RUN NO: K5-7/2 TITLE: Recheck Test 1 Conditions

	FRESH FEED	CO MAKEUP	H2 MAKEUP	K-01 OUTLET	REACTOR FEED	C-05 OUTLET	MAIN PURGE	DISTILL. PURGE	CRUDE PRODUCT	REFINED PRODUCT
L	90	CO	10	1	rcc	010	100	7 4	000	70
L	60	70	5	~	t00	243	001	+	777	40
psig	754	864	486	742	733	669	686	ი	185	139
H2	67.14	1.93	77.66	69.27	68.72	63.64	69.27	8.00	0.00	0.00
co	29.89	97.05	9.12	18.27	19.85	15.82	18.27	7.00	0.00	0.00
N2	0.46	0.97	6.50	4.09	3.18	3.66	4.09	2.00	0.00	00.00
CH4	0.03	0.05	0.92	0.34	0.26	0.30	0.34	0.00	0.00	0.00
CO2	2.48	0.00	3.76	5.07	4.31	4.51	5.07	62.00	0.00	00.00
DME	0.00	0.00	0.01	0.00	00.0	0.01	0.00	4.00	0.00	0.00
MeOH	0.00	0.00	0.06	0.55	0.65	8.15	0.55	10.00	78.00	06.66
EtOH	0.00	0.00	00.0	00.0	00.0	0.00	0.00	0.00	0.16	0.01
H2O	0.00	0.00	0.00	00.0	00.0	0.46	0.00	0.00	21.83	0.09
others	0.00	0.00	1.97	2.40	3.03	3.44	2.40	7.00	0.01	00.00
TOTAL	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
lb/lbmol	10.953	27.501	7.770	10.849	10.897	12.096	10.849	39.217	29.048	32.028
MMBTU/hr	191.6	0.0	0.0	562.6	754.6	735.9	18.4	1.6	33.3	114.9
MMBTU/hr	-32.98	0.00	0.00	-104.51	-123.21	-149.39	-3.44	-2.98	-15.25	-40.68
SCFH	684,799	0	0	2,202,625	2,913,449	2,535,117	71,933	8,736	55,532	150,156
lbmol/hr	1,805.2	0.0	0.0	5,806.4	7,680.2	6,682.9	189.6	23.0	146.4	395.8
lb/hr	19,772	0	0	62,993	83,688	80,838	2,057	903	4,252	12,678

Final Report: Volume 2 (June 2003)

B-3

Kingsport LPMEOHTM CCT Project

RUN NO:	K6-09/13		TITLE	: Catalyst Addition and Agi	ing		
Balance F	Period: Start Date End Date		3sep98 07:00 4sep98 07:00	00.00	Time From St	art of Run (days) Start End	267.3 268.3
Reaction	Conditions Temperature (°F) Pressure (psig) Space Velocity (sL/kg-hr) Vg (inlet)		454 710 3811 0.64		Slurry Data	Catalyst Weight (Ib oxide) Slurry Concentration (wt %) Slurry Level (ft) Gas Holdup (vol %)	42,900 43.4 56.0 40.0
Performa	nce Results Raw MeOH Production (ton Raw MeOH Production (ton Syngas Utilization (SCF/lb n Catalyst Life (eta) CO Conversion (total) (%) CO Conversion to MeOH (%) CO Conversion to H2 (%)	day) (day) AeOH) 6)	262.6 256.4 39.8 39.8 0.54 48.0 50.7 -2.74	(gas measurements) (liquid measurements)	Energy Balan	ce Steam Production (Ib/hr) Steam Drum Pressure (psig) Steam Import/Export (Ib/hr) Reactor O-T-M Conversion (% LHV) Wetted Tube Length (ft) Heat Transfer Area (ft2) Reactor Overall U (BTU/hr-ft2-F)	21624 240 240 26.8 26.8 26.8 2026 199
	Syngas Conversion (% LHV Syngas Usage (BTU/gallon Recycle Ratio MeOH Productivity (gmol/k Rxr Volumetric Productivity Sparger "K"-value) MeOH) J-hr) (ton/day-ft3)	81.6 67,933 2.26 15.60 0.109 6.53		Atom/Mass B	alance Closure (% of reactor inlet) C O N Total Mass	100.60 100.17 100.23 101.50 100.53
Liquid Product Analysis (wt%)	Methanol Ethanol Water Oil Total	9/13 9/13 99.99 0.01 0.01 100.01	REFINED PR 9/14 7:00 99.99 0.01 0.01 100.00	ODUCT	9/13 19:00 91.10 0.09 8.80 0.06 100.05	CRUDE PRODUCT 9/14 7:00 91.29 0.09 8.53 0.06 99.97	

APPENDIX B-2 SAMPLE MATERIAL BALANCE – BALANCED GAS CASE (13 SEPTEMBER 1998)

Kingsport LPMEOHTM CCT Project

B-4

Final Report: Volume 2 (June 2003)

SAMPLE MATERIAL BALANCE – BALANCED GAS CASE (13 SEPTEMBER 1998) (cont'd) **APPENDIX B-2**

Catalyst Addition and Aging

TITLE

K6-09/13

RUN NO:

REFINED PRODUCT 167,205 440.8 14,122 32.039 0.00 0.00 0.00 0.00 0.00 0.00 99.98 100.00 128.1 -45.39 0.01 0.00 72 CRUDE PRODUCT 100,354 264.5 7,947 -27.16 100.00 30.041 0.00 0.00 0.00 0.00 0.00 85.49 0.06 14.45 65.8 0.01 220 185 DISTILL. PURGE 8.00 7.00 62.00 4.00 0.00 0.00 7.00 100.00 39.217 4,732 12.5 489 -1.61 0.9 76 PURGE 10.266 71,868 MAIN 189.5 1,945 72.47 13.38 4.71 0.83 7.18 0.00 0.00 0.00 0.00 0.00 100.00 -3.66 18.2 99 691 2,255,511 5,945.8 75,898 OUTLET -159.02 12.765 699.6 63.65 11.54 4.22 0.73 6.46 0.01 12.03 0.00 0.62 0.62 100.00 C-05 240 701 REACTOR 2,767,621 7,295.8 76,702 127.52 10.513 FEED 100.00 724.8 70.75 18.10 0.59 5.76 0.00 0.64 0.00 0.00 0.73 316 734 3.44 ,920,286 5,062.1 51,967 OUTLET 10.266 486.6 -91.37 100.00 72.47 13.38 K-01 4.71 0.83 7.18 0.00 0.60 0.00 0.00 0.82 269 743 MAKEUP 77.66 9.12 6.50 0.92 3.76 0.01 0.06 0.00 0.00 1.97 100.00 7.770 0.00 0.0 НZ 82 560 0 0.0 MAKEUP 27.501 0.0 0.00 100.00 1.93 97.05 0.05 0.00 0.00 0.00 0.00 0.97 <u>8</u> 80 866 0.00 FRESH 851,389 2,244.4 24,374 10.860 -41.05 FEED 67.63 29.23 100.00 237.6 0.05 2.70 0.00 0.00 0.00 0.40 0.00 101 754 MMBTU/hr **MMBTU/hr** SCFH Ibmol/hr b/lbmol TOTAL others MeOH EtOH DME H20 C02 lb/hr psig CH4 8 8 Ř Enthalpy (% lom) Mole Wt Comp LHV Flow ٩ ⊢

Final Report: Volume 2 (June 2003)

B-5

Kingsport LPMEOHTM CCT Project

RUN NO:	K3-8/06		TITLE	Texaco-type Syngas			
Balance P	eriod: Start Date End Date	06	aug97 07:00: aug97 07:00:	8 8	Time From St	art of Run (days) Start End	121.7 122.7
Reaction	Conditions Temperature (°F) Pressure (psig) Space Velocity (sL/kg-hr) Vg (inlet)		481 701 6507 0.70		Slurry Data	Catalyst Weight (Ib oxide) Slurry Concentration (wt %) Slurry Level (ft) Gas Holdup (vol %) Gassed Slurry Volume (ft3)	26,400 33.8 56.5 45.4 2377
Performa	nce Results Raw MeOH Production (ton/ds Raw MeOH Production (ton/ds Syngas Utilization (SCF/lb Mei Catalyst Life (eta) CO Conversion (total) (%) CO Conversion to MeOH (%) CO Conversion to H2 (%)	(A) (HC	193.0 193.0 45.5 0.56 1.2.8 12.8 -0.12	(gas measurements) (liquid measurements)	Energy Balan	ce Steam Production (lb/hr) Steam Drum Pressure (psig) Steam Import/Export (lb/hr) Reactor O-T-M Conversion (% LHV) Wetted Tube Length (ft) Heat Transfer Area (ft2) Reactor Overall U (BTU/hr-ft2-F)	13520 332 1341 18.5 52.7 2045 139
	Syngas Conversion (% LHV) Syngas Usage (BTU/gallon Mr Recycle Ratio MeOH Productivity (gmol/kg-hi Rxr Volumetric Productivity (to Sparger "K"-value	sOH)) n/day-ft3)	70.9 68,968 2.98 18.56 0.081 8.95		Atom/Mass B	alance Closure (% of reactor inlet) C H O N Total Mass	98.60 101.83 98.16 99.56 98.77
Liquid Product Analysis (wt%)		F 8/6 19:00	KEFINED PR (8/7 7:00	DDUCT	8/6 19:00	CRUDE PRODUCT 8/7 7:00	
	Methanol Ethanol Water Oil Total	99.98 0.01 0.02 0.00 100.01	99.99 0.01 0.00 0.00 100.01		95.11 1.32 2.42 1.15 100.00	95.61 0.97 2.56 0.65 99.79	

APPENDIX B-3 SAMPLE MATERIAL BALANCE – H₂:CO = 0.77 (06 AUGUST 1997)

Kingsport LPMEOHTM CCT Project

B-6

Final Report: Volume 2 (June 2003)

APPENDIX B-3 SAMPLE MATERIAL BALANCE – H_2 :CO = 0.77 (06 AUGUST 1997) (cont²d)

RUN NO: K3-8/06 TITLE: Texaco-type Syngas

		FRESH	S	H2	K-01	REACTOR	C-05	MAIN	DISTILL.	CRUDE	REFINED
		FEED	MAKEUP	MAKEUP	OUTLET	FEED	OUTLET	PURGE	PURGE	PRODUCT	PRODUCT
⊢	Ľ	06	77	74	122	336	243	89	66	217	76
д.	psig	755	861	455	753	738	688	669	6	185	140
Comp	H2	67.35	1.93	77.66	31.81	38.27	28.96	31.67	3.00	0.00	0.00
(mol %)	8	28.98	97.05	9.12	54.09	50.02	49.91	53.97	15.00	00.0	0.00
	N2	0.45	0.97	6.50	2.57	2.05	2.35	2.56	2.00	00.0	0.00
	CH4	0.06	0.05	0.92	0.45	0.36	0.41	0.45	0.00	00.0	0.00
	C02	3.16	00.00	3.76	10.27	8.36	9.49	10.25	59.00	00.0	0.00
	DME	0.00	00.00	0.01	0.01	0.01	0.01	0.01	3.00	00.0	0.00
	MeOH	0.00	00.0	0.06	0.51	09.0	8.00	0.82	11.00	94.71	99.97
	EtOH	0.00	00.00	00.0	00.0	00.0	0.02	00.0	0.00	0.79	0.01
	H20	0.00	00.00	00.00	00.0	00.0	0.10	00.00	0.00	4.40	0.02
	others	0.00	0.00	1.97	0.30	0.33	0.75	0.26	7.00	0.10	0.00
	TOTAL	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Mole Wt	lb/lbmol	11.004	27.501	7.770	21.360	19.391	22.197	21.405	39.897	31.853	32.038
LHV	MMBTU/hr	176.7	30.0	0.0	585.3	792.9	773.9	38.7	1.0	51.1	95.6
Enthalpy	MMBTU/hr	-31.93	-11.62	0.00	-250.85	-282.41	-309.74	-16.62	-1.90	-18.41	-33.86
Flow	SCFH	636,302	95,064	0	2,182,407	2,907,469	2,542,142	143,266	5,609	69,317	124,830
	lbmol/hr	1,677.4	250.6	0.0	5,753.1	7,664.4	6,701.4	377.7	14.8	182.7	329.1
	lb/hr	18,457	6,892	0	122,886	148,619	148,750	8,084	590	5,820	10,543

RUN NO:	K21-03/27	TITLE	: Reactor Feed: Destec Typ	e Syngas		
Balance P	eriod: Start Date End Date	27mar00 07:00 28mar00 07:00	00:	Time From St	art of Run (days) Start End	828.3 829.3
Reaction (Conditions			Slurry Data		
	Temperature (°F)	454			Catalyst Weight (Ib oxide)	44,687
	Pressure (psig)	200			Slurry Concentration (wt %)	42.2
	Space Velocity (sL/kg-hr)	3720			Slurry Level (ft)	49.0
	Vg (inlet)	0.66			Gas Holdup (vol %) Gassed Slurry Volume (ft3)	25.1 2057
Performan	ice Results					
	Raw MeOH Production (ton/day)	187.3	(gas measurements)	Energy Balan	ce	
	Raw MeOH Production (ton/day)	187.2	(liquid measurements)	;	Steam Production (lb/hr)	14609
	Syngas Utilization (SCF/lb MeOH)	42.5			Steam Drum Pressure (psig)	248
	Catalyst Life (eta)	0.370			Steam Import/Export (Ib/hr)	6697
					Reactor O-T-M Conversion (% LHV)	19.1
	CO Conversion (total) (%)	14.8			Wetted Tube Length (ft)	45.2
	CO Conversion to MeOH (%)	15.0			Heat Transfer Area (ft2)	1754
	CO Conversion to H2 (%)	-0.22			Reactor Overall U (BTU/hr-ft2-F)	158
	Syngas Conversion (% LHV)	75.6		Atom/Mass B	alance Closure (% of reactor inlet)	
	Syngas Usage (BTU/gallon MeOH)	68,524			, J	99.50
	Recycle Ratio	3.21			Т	100.39
	MeOH Productivity (gmol/kg-hr)	10.94			0	99.73
	Rxr Volumetric Productivity (ton/day-ft3)	0.091			Z	102.33
	Sparger "K"-value	4.09			Total Mass	100.83
Liquid Product	REFINED PRO	DUCT		CRUDE PROI	DUCT	
Analysis	3/27	3/28		3/27	3/28	
(wt%)	19:00	7:00		19:00	7:00	
	Methanol 99.99	<u> 66.66</u>		96.60	96.48	
	Ethanol 0.01	0.01		0.29	0.33	
	Water 0.01	0.01		3.00	3.07	
	Oil 0.00	0.00		0.06	0.06	
	Total 100.01	100.01	1	99.95	99.94	

APPENDIX B-4 SAMPLE MATERIAL BALANCE – H₂:CO = 0.96 (27 MARCH 2000)

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

B-8

SAMPLE MATERIAL BALANCE – H_2 :CO = 0.96 (27 MARCH 2000) (cont'd) **APPENDIX B-4**

RUN NO: K21-03/27 TITLE: Reactor Feed: Destec Type Syngas

		FRESH FEED	CO MAKEUP	H2 MAKEUP	K-01 OUTLET	REACTOR FEED	C-05 OUTLET	MAIN PURGE	DISTILL. PURGE	CRUDE PRODUCT	REFINED PRODUCT
⊢	ш	66	57	52	269	302	247	06	78	216	87
₽	psig	753	859	£	742	728	679	658	80	185	145
Comp	H2	65.86	1.93	77.66	35.54	42.04	33.21	35.54	8.00	00.0	0.00
(% Iom)	00	30.96	97.05	9.12	47.34	44.01	43.37	47.34	7.00	00.0	00.00
	N2	0.47	0.97	6.50	2.93	2.27	2.63	2.93	2.00	00.0	00.00
	CH4	0.16	0.05	0.92	0.91	0.71	0.82	0.91	00.0	00.0	00.00
	C02	2.55	00.0	3.76	9.98	7.96	9.10	9.98	62.00	00.0	00.00
	DME	0.00	00.0	0.01	0.02	0.01	0.02	0.02	4.00	0.00	0.00
	MeOH	0.00	00.0	0.06	0.24	0.19	7.85	0.24	10.00	94.50	99.98
	EtOH	0.00	00.0	00.0	0.00	00.0	0.01	00.0	00.0	0.21	0.01
	H2O	0.00	00.0	0.00	0.00	00.0	0.13	00.0	00.0	5.28	0.01
	others	0.00	00.0	1.97	3.04	2.80	2.86	3.04	7.00	0.01	00.00
	TOTAL	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Mole Wt	lb/lbmol	11.282	27.501	7.770	20.337	18.339	20.758	20.337	39.217	31.349	32.039
LHV	MMBTU/hr	174.9	12.7	0.0	551.7	742.7	723.9	25.7	1.0	50.5	91.2
Enthalpy	MMBTU/hr	-31.46	4.93	0.00	-223.49	-253.61	-277.87	-10.75	-1.79	-18.48	-32.23
Flow	SCFH	622,767	40,204	0	2,129,278	2,813,381	2,431,800	99,014	5,244	69,500	119,058
	lbmol/hr	1,641.7	106.0	0.0	5,613.0	7,416.4	6,410.5	261.0	13.8	183.2	313.9
	lb/hr	18.522	2.915	0	114.150	136.008	133.070	5.308	542	5.743	10.056

Final Report: Volume 2 (June 2003)

B-9

APPENDIX C

DETAILS ON DATA ANALYSIS METHODOLOGY

	Sgnal or Computed Value	Process Row 1 Diagram Sht.	Иа S Balance Point	Analytical Measurement	Comments
Material Balance:	CO Gas Flowrate	.	10	Yes - On-Line GC	Used in Overall Material and Energy Balance
	H2 Gas Flowrate	-	20	Yes - On-Line GC	Used in Overall Material and Energy Balance
	Balanced GasFlowrate	-	30	Yes - On-Line GC	Used in Overall Material and Energy Balance
	Reactor Feed Howrate	2	109	Yes - On-Line GC	Used in Material and Energy Balance Across Reactor
	Reactor Effluent Howrate	7	120	Yes - On-Line GC	Used in Material and Energy Balance Across Reactor
	Plant Purg e Howrate	ю	148	Yes - On-Line GC	Used in Overall Material and Energy Balance
	Recycle Gas Flowrate	-	149	Yes - On-Line GC	Used in Overall Material and Energy Balance
	Refined Grade Methanol Howrat	с	216	res-Batch Sample	Total Transfer from 29D-20 and 29D-21 Used in
					Overall Material and Energy Balance
	Crude Grade Methanol Howrate	ი	242	res-Batch Sample	Used in Overall Material and Energy Balance
	Distillation Fuel Gas Flowrate	0 0	èe Comment	Yes - On-Line GC	Sum of Vapor Hows from 29C-11, 29C-12, and 29C-21;
					Used in Overall Material and Energy Balance
	Total Methanol Production	с	204	No	Sum of Mass Balance Points 216 and 242
Energy Balance:					
5	Steam Production Howrate	ъ	409	No	Steam Produced at 29C-02 Steam Drum
	Steam Import/Export Howrate	ъ Ъ	430	No	As Required from Eastman Steam System
	Reactor O-T-M Conversion	n/a	n/a	No	-
Reactor Data:					
	Reactor Temperature	2	n/a	No	Specified for each Case
	Reactor Pressure	7	n/a	No	Specified for each Case
	GasHoldup	n/a	n/a	No	Calculated from Reactor Pressure-Drop Measurements
	Surry Concentration	n/a	n/a	No	Calculated from Catalyst Addition/Withdrawal Amounts
	Gassed Surry Level	2	n/a	No	From 29C-01 Nuclear Density Gauge
	Space Velocity	n/a	n/a	No	From Reactor Feed Flowrate and Surry Concentration
	Inlet Superficial Velocity	n/a	n/a	No	From Reactor Feed Flowrate
	Catalyst Age	n/a	n/a	No	Estimated Average Catalyst Age vs. Fresh Catalyst
	Reactor Volumetric Productivity	n/a	n/a	No	Tonsper Day Methanol per Cubic Foot Reactor Volume
	% CO Conversion	n/a	n/a	No	CO Consumption across 29C-01 Reactor
	Catalyst Methanol Productivity	n/a	n/a	No	Defined asgm-mole Methanol/hr-kg Catalyst Oxide
	Syng as Utilization	n/a	n/a	No	

Summary of Data for Typical Material and Energy Balance and Reactor Performance Assessment **Table C-1**

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

C-2

APPENDIX C-2

METHODS OF CALCULATION FOR KEY PROCESS PARAMETERS

Catalyst Age (η) = $k_0/k_0(t=0)$

- <u>where</u>: $k_0 =$ the pre-exponential kinetic rate constant for the methanol synthesis reaction at any time
 - $k_0(t=0) =$ the pre-exponential kinetic rate constant for the methanol synthesis reaction for fresh catalyst

The rate constants are determined using a proprietary model which computes the kinetic rate constant for the methanol synthesis reaction.

CO Conversion to Methanol [%] =

(Raw Methanol Flow + Methanol in Main Plant Purge + Methanol in Distillation Purge [all lbmol/hr]) Carbon Monoxide in Reactor Feed [lbmol/hr] * 0.01

<u>where</u>: Methanol in Main Plant Purge [lbmol/hr] = Main Plant Purge Flow [lbmol/hr] * Methanol Concentration in Main Plant Purge [mol%]

Methanol in Distillation Purge [lbmol/hr] = Distillation Purge Flow [lbmol/hr] * Methanol Concentration in Distillation Purge [mol%]

Carbon Monoxide in Reactor Feed [lbmol/hr] = *Reactor Feed Flow* [lbmol/hr] * *Carbon Monoxide Concentration in Reactor Feed* [mol%]

Gas Holdup (vol%) = (Liquid-Solid Density $[lb/ft^3]$ - 3-Phase Density $[lb/ft^3]$ * 100 (Liquid-Solid Density $[lb/ft^3]$ - Vapor Density $[lb/ft^3]$)

where: Vapor Density [lb/ft³] is calculated using the composition at the Reactor Effluent, and at the measured temperature and pressure of the Reactor

3-Phase Density $[lb/ft^3] = \frac{Pressure Drop [psi] * 144 [in^2/ft^2]}{Span between Taps of Pressure Drop Measurement [ft]}$

Liquid-Solid Phase Density [lb/ft³] is a function of the catalyst concentration [wt%], which is iterated until the calculated weight of catalyst matches the catalyst weight in the Reactor.

General Notes:

a. All terms in *Italics* refer to measured variables, either by field instruments or gas chromatographs.

APPENDIX C-2

METHODS OF CALCULATION FOR KEY PROCESS PARAMETERS (cont'd)

Gassed Slurry Volume $[ft^3]$ = Reactor Cylindrical Volume $[ft^3]$ + Reactor Head Volume $[ft^3]$

where: Reactor Cylindrical Volume $[ft^3] = A [ft^2] * Cylindrical Height [ft]$

A [ft²] = $\frac{\Pi}{4}$ * (Reactor ID [ft])² * (Fraction Open Area)

Reactor ID = Reactor Inside Diameter = 7.5 ft

Fraction Open Area = 0.9652 (3.48% of the cross-sectional area of the Reactor is occupied by the tubes of the internal heat exchanger)

Cylindrical Height [ft] = Gassed Slurry Height [ft] - $(\frac{\text{Reactor ID [ft]}}{4})$

Note: The term "(Reactor ID)/4" is the depth of the bottom head of the Reactor.

Reactor Head Volume [ft³] = $\frac{\Pi * (\text{Reactor ID [ft]})^3}{24} - 8 [ft^3]$

<u>Note</u>: 8 cubic feet is subtracted from the Reactor Head Volume to account for the displacement of the gas sparger and the header piping for the internal heat exchanger.

Inlet Superficial Velocity $[ft/s] = \frac{Reactor Feed Flow [lbmol/hr] * V [ft³/lbmol]}{3,600 [s/hr] * A [ft²]}$

where: $V [ft^3/lbmol] = 10.73 * (Reactor Temperature [°F] + 459.67)$ (Reactor Pressure [psig] + 14.1)

A [ft²] = $\frac{\Pi}{4}$ * (Reactor ID [ft])² * (Fraction Open Area)

Reactor ID = Reactor Inside Diameter = 7.5 ft

Fraction Open Area = 0.9652 (3.48% of the cross-sectional area of the Reactor is occupied by the tubes of the internal heat exchanger)

General Notes:

a. All terms in *Italics* refer to measured variables, either by field instruments or gas chromatographs.
APPENDIX C-2

METHODS OF CALCULATION FOR KEY PROCESS PARAMETERS (cont'd)

Methanol Productivity [gmol/kg-hr] = (Methanol in Reactor Effluent [lbmol/hr] - Methanol in Reactor Feed [lbmol/hr]) * 1,000 [g/kg] Catalyst Weight (lb oxide)

<u>where</u>: Methanol in Reactor Effluent [lbmol/hr] = Reactor Effluent Flow [lbmol/hr] * Methanol Concentration in Reactor Effluent Stream [mol%]

Methanol in Reactor Feed [lbmol/hr] = Reactor Feed Flow [lbmol/hr] * Methanol Concentration in Reactor Feed Stream [mol%]

Reactor O-T-M Conversion [%] = LHV of Raw Methanol * 100 LHV of Reactor Feed

<u>Note</u>: Lower heating values (LHV's) for Raw Methanol and Reactor Feed are calculated from compositions of each stream, in units of million Btu per hour.

Reactor Volumetric Productivity $[TPD/ft^3] = \frac{Raw Methanol Flow [TPD]}{Gassed Slurry Volume [ft^3]}$

 $Space Velocity [sL/kg-hr] = \frac{Reactor Feed Flow [lbmol/hr] * 10,175 [sL/lbmol @ 0°C]}{Catalyst Weight [lb oxide] * 0.454 [kg/lb]}$

Sparger Resistance Coefficient "K" =

<u>Sparger Pressure Drop [psi] * Reactor Feed Density [lb/ft³] * constant</u> (Reactor Feed Flow [KSCFH] * Reactor Feed Molecular Weight [lb/lbmol])²

Syngas Usage [Btu/gallon Methanol] =

Syngas LHV to Methanol [Btu/hr] * 24 [hr/day] * 6.642 [lb/gallon Methanol] Raw Methanol Flow [TPD] * 2,000 [lb/ton]

where: Syngas LHV to Methanol [Btu/hr] = the difference between the Lower Heating Value of the three feed gas streams (Balanced Gas, CO Gas, H₂ Gas) and the two purge gas streams (Main Plant Purge, Distillation Purge). Lower heating values are calculated from the compositions of each stream.

General Notes:

a. All terms in *Italics* refer to measured variables, either by field instruments or gas chromatographs.

APPENDIX C-2

METHODS OF CALCULATION FOR KEY PROCESS PARAMETERS (cont'd)

Syngas Utilization [SCF/lb Methanol] =

(Balanced Gas Flow [SCFH] + CO Gas Flow [SCFH]) * 24 [hr/day] Raw Methanol Flow [TPD] * 2,000 [lb/ton]

General Notes:

a. All terms in *Italics* refer to measured variables, either by field instruments or gas chromatographs.

APPENDIX D

DATA SUMMARY TABLES

	l Sparger Sparger	dP Resist. ("K") ("K")	6.23 10.87	8.69 12.72	10.50 14.17	11.05 14.88	14.17 18.78	17.61 20.28	19.17 21.35	20.47 21.69	21.59 19.36	21.64 20.51	21.59 23.04	24.25 23.16	23.48 24.80	25.38 26.81	26.18 27.29	24.49 27.11	26.13 27.67	26.26 27.32	25.91 29.72	26.05 29.24	26.24 29.58	26.26 28.88 76.76 78.83	14.81 33.22	2.99 3.37	3.05 3.52	3.45 3.71	4.00 3.90	3.96 4.11 4.51 4.38	5.04 4.93	5.33 5.31	5.64 5.38	7.34 5.74	8.12 6.67	9.33 7.65	9.29 7.90		9.47 8.15	9.47 8.15 10.66 9.21	9.47 8.15 10.66 9.21 12.10 10.78
;	U eactor Overal	l. Prod. (Btu PD/ft3) hr ft2 F	181 181	0.110 184	172 172	1112 170	123 168	171 011.0	0.119 173	128 153	133 158	117 172	0,110 175	0.126 186	0.126 164	0.130 147	134 154	/cl Ell.0	0.107 101	162 162	0.103 162	0.108 162	156 156	157 157	128 128	1079 161	0.077 175	171 080	0.086 170	0.082 168	154	160 160	0.088 163	0.083 161	0.085 168	170	168	071	101	0.080 166	.080 166 .085 168
	Catalyst Ro	OH Prod. Vo nol/hr-kg) (Tl	30.74 0	32.19 0	32.00 0	32.84 0	35.82 0	34.35 0	33.86 C	36.80 0	36.60 0	34.77 0	30.65	33.17 0	31.58 0	31.38 0	31.87 0	28.35 0	30.05 (29.74 (28.85	27.32 0	27.66 0	27.43 0	27.76 C	883	26.95 0	24.77 0	24.99 0	26.14 0	23.85 (27.35 (26.90 0	26.62 0	26.65 0	26.68 0	25.84 0	26.24 0	25.77 0	74.68 0	0014	25.39 0	25.39 C
	Kaw MeOH (Prod. Mc (TPD) (gr	242.3	253.8	252.4	258.9	281.9	270.5	267.0	288.8	292.2	275.5	261.1 242.7	261.0	248.2	247.8	252.2	224.4	238.0 230.8	228.0	215.4	219.5	218.1	218.9	69.0	205.3	188.3	189.4	198.9	181.5 208.8	205.4	201.5	203.5	206.5	199.6	202.3	198.4	193.2		195.9	195.9 193.9
	Syngas	Util. SCF/lb) (37.6	37.5	37.5	37.9	38.5	39.4	37.8	40.0	39.4	39.8	39.2	39.8	40.4	41.9	42.9	41.9	40.9 41 3	42.0	42.2	42.8	43.4	43.7	44.4	42.0	40.7	41.3	41.4	41.3 43.8	45.6	44.0	43.4	42.8	41.5	42.6	43.0	42.0		41.9	41.9 41.1
•	keactor O-T-M	Conv. (%) (33.1	31.7	30.5	31.2	33.6	30.0	29.1	30.7	29.2	26.4	24.6 24.6	26.2	26.2	26.0	26.9	24.4	25.6 25.1	24.5	23.7	23.9	23.4	23.4	151	22.5	21.4	20.2	20.8	19.5 21.5	21.2	21.0	20.7	19.5	19.3	19.5	19.1	19.0		19.6	19.6 19.4
	CO	Conv. (%)	54.8	52.5	50.5	51.7	55.8	49.8	48.2	51.1	48.3	47.1	6./4 49.0	44.6	44.6	41.6	39.7	42.7	42.1 43.0	41.1	42.8	40.5	38.9	37.5	700 8 6	39.3	42.6	39.2	35.5	33.4 33.6	33.2	33.7	33.0	29.7	31.0	30.0	29.9	31.5		32.5	32.5 32.8
•	Catalyst	Age (eta)	1.30	1.24	1.18	1.29	1.78	1.24	1.14	1.43	1.17	1.03	0.94	0.93	0.89	0.83	0.83	0.75	0.81	0.74	0.71	0.68	0.64	0.63	0.64	0.61	0.63	0.59	0.58	0.50	0.56	0.57	0.55	0.53	0.53	0.53	0.51	0.52		0.54	0.54 0.54
_	Catalyst	Inventory (Ib)	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20,300	20.300	19,500	19,500	19,500	19,500	19,500	19,500	19,500	19,500	19,500	19,500	19,500	19,500	19,500		19,500	19,500 19,500
)	Gassed	Slurry Hgt (ft)	54.9	54.9	56.1	55.1	54.5	54.3	53.4	53.8	52.4	55.8	52.5	49.5	47.0	45.5	45.0	47.5	53.0 52.0	50.5	50.0	48.5	49.0	49.5	43.0	61.4	57.8	56.0	55.0	52.7 51.6	50.9	49.4	54.8	59.0	56.0	53.5	52.5	59.4		58.4	58.4 54.0
	Gas	Holdup (vol%)	50.2	50.6	52.0	50.9	50.7	53.0	51.9	52.2	51.5	50.5	6.74 45.2	43.1	44.9	45.6	44.9	43.3	45.2 43.6	44.0	43.0	43.1	44.0	43.6 43.5	36.9	45.9	45.3	44.9	45.2	44.2	5.44	43.3	48.3	46.6	45.8	43.9	43.9	45.3		45.4	45.4 43.9
	Slurry	Conc. (wt% ox)	30.5	30.8	30.9	30.8	30.9	32.1	32.0	32.0	32.2	30.3	30.2 29.5	29.9	31.8	32.8	32.8	30.9	29.3	29.9	29.7	30.4	30.5	30.1	30.8	25.6	26.6	27.1	27.6	28.2 28.4	29.0	29.2	28.9	26.7	27.5	27.7	28.1	26.1		26.5	26.5 27.5
•	Space	Velocity (l/hr-kg)	6,203	6,783	7,014	7,019	7,094	7,629	7,762	7,970	8,562	8,771	8.376	8,419	8,019	7,980	7,842	7,817	166,1	7,945	7,728	7,767	7,831	7,856	3 813	8062	7905	8294	8465	8203 8456	8412	8338	8559	9252	9079	9042	9019	8944		8827	8827 8794
	Inlet Sup.	Velocity (ff/sec)	0.50	0.55	0.58	0.58	0.59	0.63	0.64	0.66	0.70	0.72	0.71	0.70	0.67	0.67	0.66	0.65	0.66	0.66	0.64	0.65	0.65	0.66	0.32	0.62	0.62	0.66	0.67	0.65	0.67	0.66	0.68	0.74	0.72	0.72	0.72	0.71		0.70	0.70 0.70
	Purge	Gas (KSCFH)	33.7	30.1	29.1	36.5	61.9	57.5	37.3	116.3	76.9	73.6	59.8 52.1	61.4	65.7	103.1	129.1	86.4	7.97	95.0	85.0	98.2	110.9	115.8	30.2	82.5	58.6	62.1	74.2	59.7	146.5	117.7	105.9	90.8	74.9	92.3	91.5	70.0		71.9	71.9 59.6
	Reactor	Feed (H2:CO)	NA	NA	NA	NA	NA	NA	NA	NA	3.47	3.78	4.32	3.52	3.52	3.27	2.95 2.55	3./U	3.58 3.58	3.49	3.82	3.55	3.48	3.29 3.37	0.43	3.69	4.34	4.15	3.55	3.55 3.19	3.19	3.26	3.28	3.12	3.33	3.13	3.22	3.54		3.46	3.46 3.48
	Recycle	Gas (KSCFH)	1,375	1,536	1,619	1,601	1,527	1,746	1,843	1,779	1,994	2,114	2,108 2.086	2,028	1,921	1,878	1,793	1,903	1,922	1,932	1,898	1,886	1,901	1,901	1.051	1,938	1,938	2,079	2,109	2,097	1,991	2,003	2,080	2,326	2,307	2,267	2,263	2,251		2,203	2,203 2,218
	Fresh	Feed (KSCFH)	758	792	789	818	904	887	841	964	958	913	968 793	865	835	864	902	783	794	798	757	782	789	798 277	255	718	638	651	687	625 762	781	739	737	736	169	719	117	676		685	685 664
		Pres. (psig)	719	716	705	705	705	704	705	703	60L	708	60/ 60/	702	700	694	069	696 200	00/	669	700	669	700	669	2002	724	711	707	707	707	708	707	707	706	707	706	706	707		707	707 707
		Temp (Deg C)	248	249	249	249	248	249	249	249	249	249	249 249	249	248	246	247	247	249	249	249	249	249	249	249	248	249	249	249	249	249	248	249	249	249	249	249	249		249	249 249
		Gas Type	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Ralanced	Balanced	Balanced	Balanced	Balanced	Balanced	CO-rich	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced		Balanced	Balanced Balanced
	Days	On Stream	9	4	∞	6	10	=	12	12	13	4	c 91	18	19	20	21	5 5	23	25	26	27	28	29	32	73	74	75	76	17 97	6/	80	81	82	83	84	85	86		87	87 88
			r-97	pr-97	pr-97	Apr-97	Apr-97	-Apr-97	-Apr-97	8-Apr-97	-Apr-97	0-Apr-97	22-Anr-97	24-Apr-97	25-Apr-97	26-Apr-97	27-Apr-97	28-Apr-97	29-Apr-97 30-Apr-97	1-May-97	2-May-97	3-May-97	4-May-97	5-May-97	8-Mav-97	18-Jun-97	19-Jun-97	20-Jun-97	21-Jun-97	22-Jun-97	24-Jun-97	25-Jun-97	26-Jun-97	27-Jun-97	28-Jun-97	29-Jun-97	30-Jun-97	1-Jul-97		2-Jul-97	2-Jul-97 3-Jul-97
		Date	12-Ap	13-A	14-A	15-/	16-	17	18	ĩ	51	00																													

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

Sparger Resist.	("K")	10.85	09.6	11.35	10.83	11.24	10.65	10.57	10.63	11.13	11.01	11.30	10.52	10.33	10.81	12.33	13.01	12.43	12.41	12.43	12.31	12.58	9.01	9.08	8.95	00.6	9.46	9.47	9.52	9.56	85.6	11 20	11.09	11.26	10.87	11.33	12.23	13.78	12.11	11.19	13.01	14.30	15.04
Sparger	(isd)	11.94	11.02	12.05	11.34	11.30	12.00	11.89	12.27	12.53	12.80	11.99	12.05	12.26	11.61	13.32	14.71	13.17	13.80	13.32	13.02	13.45	17.86	18.10	17.07	70.01	17.30	17.25	16.94	16.71	10.57	12.19	12.29	12.04	12.40	11.34	13.77	13.49	11.79	12.09	12.21	14.31	15.22
U Overall (Btu	hr ft2 F)	161	166	161	157	157	157	151	155	153	152	152	153	152	160	159	152	156	153	152	146	154	139	140	139	011	138	136	137	139	101	150	152	151	157	154	148	148	146	147	148	149	145
Reactor Vol. Prod.	(TPD/ft3)	0.099	0.089	0.082	0.093	0.093	0.101	0.104	0.100	0.101	0.101	0.102	0.104	0.103	0.093	0.092	0.088	0.100	0.104	0.103	0.100	0.102	0.077	0.079	0.081	0.084	0.078	0.078	0.081	0.083	0.001	160.0	0.089	0.087	0.087	0.089	0.083	0.084	0.086	0.089	0.083	0.085	0.086
Catalyst OH Prod.	nol/hr-kg)	24.25	21.28	21.09	21.54	22.17	23.42	23.24	23.14	23.07	22.71	21.92	22.25	22.50	22.15	21.67	21.30	21.42	21.64	21.62	20.58	21.71	18.11	18.25	18.56	C/.81	18.08	17.98	17.89	5671	19.46	18.78	18.03	17.56	17.89	17.20	17.36	16.59	16.36	16.65	16.03	16.13	16.61 14.88
Me Po	D) (gr	6.5	2.1	1.1	9.6	9.8	1.2	9.6	8.9	9.1	5.6	7.3	0.2	2.0	9.1	9.6	3.5	2.0	4.3	3.9	5.7	5.3	2.4	2.0	3.0	¢ ¢	2.0	7.7	6.0	5.6	0.0	0.0 2.5	8.0	1.6	6.3	8.3	3.1	3.8	1.3	5.0	7.7	7.1	2.1
gas Me il. Pr	(II)	.2 20	.2 18	.9 18	.4 20	.0 20	.4 22	.8 21	.8 21	.2 21	.5 21	.6 20	.8 21	.5 21	.8 20	.9 20	.6 20	.3 22	.6 22	.9 22	.6 21	.1 22	.0 19	.19	.5 19.	κ, τ 5		.0 18	.3 18	.0 18	<u>v</u>	6 19	.4 18	.8 18	.0 18	.8 17	.6 18	.1 17	.7 17	.7 1.7	.0 16	.9 16	- 17
or Syn	. (SCI	4	40	41	43	46	43	43	42	43	42	4	43	43	42	43	41	43	42	41	4 5	42	45	45	45	6 r	6 1 4	46	46	40	4	1 4	1 04	41	41	42	43	45	45	42	45	4 :	- 64 13
React O-T-P	(%)	20.5	19.0	18.5	21.0	21.0	21.8	21.5	21.4	21.4	21.1	20.5	20.5	20.6	1 20.8	20.5	20.4	22.3	22.4	22.7	22.1	22.8	18.1	18.2	18.5	0.01	18.4	18.6	18.3	18.2	1.61	18.9	18.5	17.9	18.2	17.7	17.8	17.2	16.8	17.0	16.8	16.7	1/1
st CO Conv	(%)	35.4	34.3	33.9	39.8	40.0	37.7	36.8	36.1	35.9	35.0	36.1	34.1	32.7	35.4	34.0	32.8	39.0	37.3	38.1	37.9	38.3	12.1	12.3	12.8		12.8	13.2	13.0	12.9	V. LC .	F.1 C	29.4	28.7	27.7	29.5	26.7	28.1	27.4	25.3	27.6	26.4	202
Cataly Age	eta)	0.53	0.50	0.47	0.53	0.52	0.54	0.53	0.52	0.51	0.50	0.49	0.48	0.48	0.48	0.46	0.46	0.51	0.51	0.52	0.50	0.52	0.56	0.57	0.56	4C.U	0.53	0.52	0.50	0.50	76.0	75.0	0.36	0.34	0.34	0.34	0.32	0.31	0.30	0.30	0.29	0.29	0.29
Catalyst Inventory	(Ib)	21,800	21,800	21,800	24,100	24,100	24,100	24,100	24,100	24,100	24,100	24,100	24,100	24,100	24,100	24,100	24,100	26,400	26,400	26,400	26,400	26,400	26,400	26,400	26,400	20,400	26,400	26,400	26,400	26,400	26,400	26 400	26,400	26,400	26,400	26,400	26,400	26,400	26,400	26,400	26,400	26,400	26,400
Gassed	Hgt (ft)	49.5	49.0	52.5	53.0	53.5	52.0	50.5	52.0	51.5	51.0	48.5	48.0	49.0	53.5	53.5	55.0	53.0	51.5	51.5	51.5	52.5	59.5	57.7	56.5	0.00	0.7 c	57.0	54.5	53.0	C.UC	0.02	50.5	49.5	51.0	47.5	52.5	49.0	47.5	47.0	48.0	47.0	47.5
Gas Holdun	(vol%)	41.9	42.0	42.3	42.4	41.8	41.4	41.3	42.1	41.6	41.6	40.3	40.2	40.9	42.1	40.6	42.5	40.8	39.8	39.6	39.4 20.5	5.65	47.5	46.2	45.4	4.04	45.4	45.9	45.9	42.4	4.66	40.4	41.1	40.3	43.2	41.7	45.8	41.2	40.1	40.7	39.0	39.3	39.3 40.3
Slurry Conc.	(wt% ox)	31.1	31.3	29.9	32.0	31.5	32.0	32.6	32.3	32.3	32.5	33.2	33.4	33.2	31.6	31.1	31.1	33.4	33.7	33.6	33.6	33.1	33.5	33.7	33.8	4.40	33.4	33.8	34.9	34.0	0.45	34.2	34.7	34.8	35.3	36.4	35.7	35.5	35.8	36.3	35.1	35.7	35.5
Space Velocity	(l/hr-kg)	7926	7788	7705	7013	7036	7177	7209	7228	7217	7231	7102	7213	7240	7067	7083	7058	6426	6460	6361	6334	6354	6600	6600	6507	6640	02/0 6383	6364	6320	6284	7000	14-21	6508	6454	6554	6383	6567	6368	6381	6469	6275	6291	6331 6403
In let Sup. Velocity	(ft/sec)	0.70	0.70	0.69	0.70	0.70	0.71	0.71	0.71	0.71	0.71	0.70	0.71	0.71	0.70	0.70	0.70	0.70	0.70	0.69	0.69	0.69	0.71	0.71	0.70	0.70	0.69	0.69	0.68	0.68	0.70	0.71	0.71	0.70	0.71	0.69	0.71	0.69	0.69	0.70	0.68	0.68	0.68
Purge Gas	(KSCFH)	103.0	49.7	60.0	100.1	145.4	116.5	116.0	101.0	108.2	96.7	125.2	115.9	108.8	103.4	111.9	74.5	106.6	98.9	92.3	78.2	99.5	138.9	137.3	143.3	144.2	145.9	151.5	154.3	155.6	6.67	1/1.4	65.3	82.3	77.2	101.1	107.3	125.8	129.7	99.2	124.1	121.1	186.5
teactor Feed	12:CO)	3.62	3.83	3.87	4.12	4.12	3.60	3.56	3.48	3.49	3.42	3.70	3.41	3.19	3.54	3.47	3.30	3.67	3.42	3.45	3.61	3.46	0.73	0.74	0.77	7970	0.78	0.82	0.81	0.80	5.13	321	3.26	3.28	3.08	3.48	3.10	3.43	3.43	2.94	3.47	3.21	2.99 3.10
tecycle F Gas	CSCFH) (1	2,161	2,277	2,235	2,118	2,063	2,125	2,139	2,170	2,154	2,189	2,124	2,173	2,187	2,137	2,160	2,197	2,070	2,090	2,062	2,082	2,048	2,233	2,238	2,182	2,148	2,129	2,118	2,101	2,088	202,2 202,2	CU2,2	2,278	2,244	2,296	2,224	2,283	2,205	2,204	2,287	2,176	2,187	2,115
resh I	SCFH) (F	761	610	632	748	805	800	801	781	788	763	770	767	768	746	755	706	801	796	781	748	06/	722	721	731	(†) 10.4	124 722	719	717	11/	740	10/	633	632	636	636	666	652	653	623	629	625	709
E L	psig) (K	706	695	694	695	669	701	702	703	704	703	702	702	702	702	702	702	702	700	700	700	/00	702	702	701	10/	203	702	702	702	00/	200	700	869	700	700	701	702	702	702	702	702	708
Temp	Deg C) (249	248	249	249	249	249	249	249	249	249	249	249	249	249	249	250	249	249	249	249	249	249	249	249	249	249 249	249	249	249	249	249 249	249	249	249	249	249	249	249	249	249	249	249 240
sä	Jype (1	anced	anced	lanced	lanced	lanced	lanced	lanced	lanced	anced	anced	lanced	anced	lanced	lanced	lanced	lanced	anced	anced	lanced	lanced	lanced	exaco	exaco	exaco	exaco	exaco	xaco	xaco	exaco	lanced	anced	anced	anced	anced	anced	anced	anced	anced	lanced	lanced	lanced	lanced
si e	am T	3 Bal	4 Bal	5 Bal	6 Bal	7 Bai	8 Bal	9 Bal	00 Bal	11 Bal)2 Bal	13 Bal	14 Bal)5 Bal)6 Bal	17 Bal)8 Bal	9 Bal	0 Bal	11 Bal	12 Bai	13 Bal	20 Te	21 Te	22 T.	23 - Tr	24 I. 57 T.		17 Te	28 Te	31 Bar	32 Ba.	14 Bal	15 Bal	36 Bal	37 Bal	38 Bal	39 Bal	40 Bal	41 Bal	t2 Bal	43 Bal	44 Bai
Da	s Stre	97 9.	97 9.	-97 9.	-9 <i>L</i> 6-	6 <i>L</i> 6-	-97 9.	-97 9.	.97 10	.07 10	.07 10	.07 10	.07 10	.97 10	.97 10	.07 10	.07 10	.97 10	11 26.	11 26-	-97 li	-97 1:	-97 12	-97 1	.1 70-	7.6- 	-10 10- 10- 11	-97 12	-97 12	-97 12	. 10 Lo	-07 I.	-97 13	-97 13	-97 15	-97 13	-97 13	1 13	t-97 14	-97 14	;-97 14.	<u></u> 3-97 1₄	2-97 1- 2-
	Date	s-Jul-	9-Jul-	10-Jul	11-Jul-	12-Jul	13-Jul-	14-Jul	15-Jul-	16-Jul-	-Jul-71	18-Jul-	-lul-91	20-Jul-	21-Jul-	22-Jul-	23-Jul-	24-Jul-	25-Jul-	26-Jul-	27-Jul	28-Jul	4-Aug	5-Aug	6-Aug	/-Aug	8-Aug 9-Aug	10-Aug	, 11-Aug	12-Aug	15-Aug	3uA-01 17-Aug	Juc-71	, 19-Aug	20-Aug	21-Aug	22-Aug	23-Aug	24-Aug	25-Aug	26-Aug	27-Aug	28-Au£
	Case	9	9	9	9	9	9	9	9	9	9	9	9	9	9	9	9	9	9	9	9 (9	ŝ	ŝ	m (n r	n (1	ŝ	б	n i	0 4	o vo	9	9	9	9	9	9	9	9	9	9	9 9

D-3

	Sparger Resist. ("K")	15.99	10.00 12.64	12.81 12.02	19.56	21.67	24.41 20.30	18.56	19.28	19.61	19.52	22.79 19.05	ce.61 17.72	18.54	18.41 18.21	17.11	17.88	22.81	17.47	16.14	15.19 14.35	14.31	14.25	14.66 14.85	15.03	14.44	14.36 13.68	13.66	14.49	14.09	13.85	14.74	14.70	15.03	14.42	14.93	14.70	14.99
	Sparger dP (psi)	16.66	10.65 13.24	13.01 12.77	14.82	16.90	20.28 15 76	16.11	16.80	18.24	16.17	19.31	15.76	16.91	17.50 17.24	15.23	15.08	17.92	16.12	14.49	13.97	13.01	13.28	12.83 14.08	14.45	13.34	12.54	12.79	12.73	12.88	12.98	12.12	12.77	13.15	12.31	13.12	12.16	12.27 12.59
=	Overall (Btu hr ft2 F)	146	149 147	142 143	166	151	147	155	148	151	148	143	140	153	154 155	143	146	141	140	142	135	140	143	142 154	144	147	146 146	149	149	152	150	145	150	159	157	156	152 152	155 159
	Reactor Vol. Prod. (TPD/ft3)	060.0	0.083 0.087	0.086 0.086	0.066	0.062	0.062	770.0	0.076	0.081	0.072	0.077	180.0	0.077	0.077 0.079	0.081	0.072	0.080	0.084	0.087	0.071	0.074	0.073	0.072	0.074	0.073	0.077	0.077	0.073	0.073	0.069	0.074	0.072	0.071	0.071	0.075	0.070	0.071 0.072
	Catalyst MeOH Prod. (gmol/hr-kg)	16.52	15.84 16.14	15.93 15.62	12.07	11.99	12.26	13.78	13.51	13.64	12.94	13.15	13 00	13.35	13.51 13.25	13.02	12.15	12.80	13.35	13.25	13.36 13.75	13.20	13.16	12.66 12.12	12.58	12.69	12.78 12.78	12.67	11.53	11.33	11.22	11.67	11.81	11.59	11.34	11.38	10.97	10.99 11.00
Daw	MeOH Prod. (TPD)	170.8	164.7 166.9	165.6 163.6	136.4	134.5	137.1	165.9	163.5	164.7	157.8	160.8	1.461	159.4	160.7 157.5	166.4	155.2	163.6	170.5	169.3	170.6	168.6	168.1	161.7 154.8	160.7	162.1	160.9 163.2	161.8	147.2	144.7	141.0 143.2	149.0	150.9	148.0	144.8	145.4	146.2	140.4 140.4
,	Syngas Util. (SCF/lb)	45.3	46.9 48.1	48.2 49.0	41.6	42.2	41.4	41.7	42.4	42.0	43.5	43.1	43.0 43.9	43.6	42.8 42.3	39.8	38.4	40.2 40.8	40.6	40.9	40.6 40.0	41.1	41.2	42.8 40.4	43.2	42.7	43.1 42.5	42.8	45.6	42.7	43.1 43.0	40.2	39.7	41.6	42.5	41.3	42.0 43.5	43.5 43.4
anotor	ceactor D-T-M Donv. (%) (17.0	15.8 16.1	16.0 15.7	15.7	15.7	16.0 17.6	18.0	17.7	17.7	17.5	7.71	17.2	17.2	17.3 17.0	17.6	17.2	18.2	18.2	18.0	18.3	18.0	18.0	17.5 16.7	17.1	17.3	17.2	17.3	15.9	15.8	6.61	16.7	16.8	16.5	16.2	16.1	15.6 15.6	15.7 15.5
•	CO CO (%)	24.6	23.7 23.4	23.3 22.4	28.3	27.1	25.2 35.6	33.3 33.3	31.6	29.4	30.3	29.5	28.1	28.0	27.5 26.8	34.1	35.1	34.4 31.5	29.9	30.0	30.2	29.7	28.9	29.3 27.2	26.7	27.6	28.1 27.1	26.6	26.5	25.4	24.9 24.8	29.6	28.7	28.0	28.0	26.9 28.0	27.2	27.1 26.4
	Catalyst Age (eta)	0.30	0.28 0.28	0.28 0.27	0.24	0.23	0.23	0.30	0.28	0.28	0.26	0.26	0.25	0.26	0.25 0.25	0.28	0.28	0.28	0.27	0.27	0.27	0.26	0.26	0.25 0.24	0.24	0.24	0.24	0.24	0.21	0.21	0.20	0.21	0.21	0.20	0.19	0.19	0.18	0.18 0.18
,	Catalyst nventory (lb)	26,400	26,400 26,400	26,400 26,400	28,700	28,700	28,700 31,000	31,000	31,000	31,000	31,000	31,000	31,000	31,000	31,000 31,000	33,300	33,300	33,300 33,300	33,300	33,300	33,300 22,300	33,300	33,300	33,300 33 300	33,300	33,300	33,300 33,300	33,300	33,300	33,300	33,300 33,300	33,300	33,300	33,300	33,300	33,300 33,300	33,300	33,300 33,300
-	Gassed Slurry l Hgt (ft)	45.5	47.5 46.0	46.0 45.5	49.0	51.5	52.5 50.5	51.0	51.0	48.5	52.0	49.5	46.5	49.5	50.0 47.5	49.0	51.5	48.5 49.0	48.5	46.5	57.5 57 5	54.0	54.5	53.5 51.0	52.0	52.5	50.5 50.5	50.0	48.0	47.5	49.5	48.0	50.0	49.5	48.5	46.5	49.v 47.5	47.0 46.5
D	Gas (Holdup I (vol%) I	39.3	38.5 38.9	46.4 42.8	38.6	39.0	39.5 38 5	38.6	40.5	39.2	38.3	41.1	36.6	37.5	36.9 40.8	40.7	39.5	38.0 38.7	38.0	37.1	40.9 40.0	39.8	40.5	39.4 42.0	39.6	39.4	39.0 38.9	39.3	38.6	39.8	38.9 39.4	37.2	38.3	42.4	39.1	38.4	36.8 36.8	38.4 38.2
	Slurry Conc. wt% ox)	36.5	35.1 35.1	39.3 37.9	36.4	35.4	35.1 37 5	37.3	38.1	38.8	36.7	39.1	38.8 38.8	37.6	37.2 40.1	41.0	39.2	40.1 40.0	40.2	40.9	37.0 37.0	38.1	38.2	38.2 40.5	39.0	38.7	39.5 39.5	39.9	40.7	41.5	41.6 40.2	40.3	39.7	41.7	40.8	41.6	40.4	41.4 41.5
	Ŭ														5	47	671	584 749	1766	1763	4753 1768	4743	4750	4698 4760	4785	4747	4717 1742	733	592	87	96 70	540	83	4598	4550	4568	534 534	512 566
•	Space /elocity l/hr-kg)	6311	6499 6418	6419 6516	5147	5093	5075	5047	5054	5115	4922	4971	5031	506	512 508	48	Ā	4 4	িব	Ā						-	4 4	4	4	4 ;	4 4	4	4				1 4	4 4
	et Sup. Space elocity Velocity it/sec) (1/hr-kg)	0.68 6311	0.70 6499 0.69 6418	0.69 6419 0.70 6516	0.63 5147	0.62 5093	0.62 5075 0.64 4872	0.67 5047	0.67 5054	0.68 5115	0.65 4922	0.66 4971	0.66 5031	0.66 506	0.66 512 0.66 508	0.69 48.	0.66 4	0.65 4	0.68 4	0.67 4	0.67	0.67	0.67	0.67	0.68	0.67	0.67	0.67 4	0.67 46	0.66 46	0.65 45	0.66 45	0.66 45	0.66	0.66	0.66	0.66 4	0.65 45
	arge Inlet Sup. Space Jas Velocity Velocity CFH) (ft/sec) (1/hr-kg)	27.5 0.68 6311	45.4 0.70 6499 54.3 0.69 6418	55.9 0.69 6419 73.7 0.70 6516	8.9 0.63 5147	3.3 0.62 5093	2.0 0.62 5075 7.1 0.64 4877	5.5 0.67 5047	9.6 0.67 5054	1.6 0.68 5115	6.9 0.65 4922	1.3 0.66 4971	9.0 0.0/0 20/0 7.0 0.66 5031	5.6 0.66 506	3.1 0.66 512 8.0 0.66 508	9.0 0.69 48	3.1 0.66 4	1.1 0.65 4 44 0.67 4	2.8 0.68 4	5.3 0.67 4	4.5 0.67	0.9 0.67	2.6 0.67	8.0 0.67 5.0 0.68	7.3 0.68	5.7 0.67	5.3 0.67 4 8.7 0.67 4	2.4 0.67 4	16.4 0.67 40	7.9 0.66 46	2.4 0.00 40 1 0 0.65 45	4.5 0.66 4	0.5 0.66 45	6.3 0.66	2.6 0.66	3.9 0.66	0.0 0.066 4 1.6 0.666 4	5.9 0.65 4: 6.9 0.66 4:
	tetor Purge Inlet Sup. Space eed Gas Velocity Velocity .CO) (KSCFH) (ft/sec) (I/hr-kg)	86 127.5 0.68 6311	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	92 165.9 0.69 6419 89 173.7 0.70 6516	94 48.9 0.63 5147	.68 53.3 0.62 5093	20 52.0 0.62 5075 47 571 0.64 4877	95 65.5 0.67 5047	81 79.6 0.67 5054	46 71.6 0.68 5115	71 86.9 0.65 4922	57 91.3 0.66 4971	45 89.6 0.07 5070 45 97.0 0.66 5031	33 95.6 0.66 506	22 83.1 0.66 512 .17 78.0 0.66 508	.12 49.0 0.69 48	37 23.1 0.66 4	01 51.1 0.65 4 50 64 0.67 4	34 62.8 0.68 4	42 65.3 0.67 4	37 64.5 0.67 4	38 71.2 0.67	26 72.6 0.67	47 88.0 0.67 . 31 55.0 0.68	14 87.3 0.68	23 95.7 0.67	36 95.3 0.67 4 13 88.7 0.67 4	08 92.4 0.67 4	43 116.4 0.67 40	28 77.9 0.66 46	25 82.4 0.00 40 14 81.0 0.65 45	68 54.5 0.66 4:	50 50.5 0.66 45	49 66.3 0.66	56 72.6 0.66	36 63.9 0.66	4 0.0 0.0 4 20 20 20 20 20 20 20 20 20 20 20 20 20	55 85.9 0.65 44 50 86.9 0.66 44
	yele Reactor Purge Inlet Sup. Space as Feed Gas Velocity Velocity THI) (H2:CO) (KSCFH) (#/sec) (//hr-kg)	84 2.86 127.5 0.68 6311	(64 3.02 145.4 0.70 6499 005 2.87 164.3 0.69 6418	11 2.92 165.9 0.69 6419 .46 2.89 173.7 0.70 6516	20 3.94 48.9 0.63 5147	02 3.68 53.3 0.62 5093	001 3.20 52.0 0.62 5075 05 442 571 0.64 4877	19 3.95 65.5 0.67 5047	58 3.81 79.6 0.67 5054	12 3.46 71.6 0.68 5115	112 3.71 86.9 0.65 4922	26 3.57 91.3 0.66 4971	152 5.43 89.6 0.07 2070 157 3.45 07.0 0.66 5031	81 3.33 95.6 0.66 506	18 3.22 83.1 0.66 512 15 3.17 78.0 0.66 508	61 4.12 49.0 0.69 48	25 4.37 23.1 0.66 4	117 4.01 51.1 0.65 4 08 3.50 644 0.67 4	98 3.34 62.8 0.68 4	93 3.42 65.3 0.67 4	92 3.37 64.5 0.67 .	82 3.38 71.2 0.67	89 3.26 72.6 0.67	67 3.47 88.0 0.67 45 3.31 55.0 0.68	07 3.14 87.3 0.68	884 3.23 95.7 0.67	2/2 3.36 95.3 0.67 2 181 3.13 88.7 0.67 2	71 3.08 92.4 0.67 4	70 3.43 116.4 0.67 44	15 3.28 77.9 0.66 46	04 3.25 82.4 0.06 46 662 3.14 81.0 0.65 45	43 3.68 54.5 0.66 4 ²	64 3.50 50.5 0.66 45	75 3.49 66.3 0.66	41 3.56 72.6 0.66	69 3.36 63.9 0.66	4 20.0 20.0 4.0 20.0 4.4 4.2 3.59 81.6 0.66 4.4	30 3.55 85.9 0.65 4! 159 3.50 86.9 0.66 4!
и 	h Recycle Reactor Purge Inlet Sup. Space d Gas Feed Gas Velocity Velocity FHJ (KSCFH) (H2-CO) (KSCFH) (ft/sec) (l/hr-kg)	1 2,184 2.86 127.5 0.68 6311	1 2,264 3.02 145.4 0.70 6499 8 2.205 2.87 164.3 0.69 6418	5 2,211 2.92 165.9 0.69 6419 0 2,246 2.89 173.7 0.70 6516	3 2,020 3.94 48.9 0.63 5147	3 2,002 3 .68 53.3 0 .62 5093	3 2,001 3.20 52.0 0.62 5075 5 2 075 4.42 571 0.64 4872	7 2,119 3.95 65.5 0.67 5047	3 2,058 3.81 79.6 0.67 5054	7 2,112 3.46 71.6 0.68 5115	2, 2,012 3.71 86.9 0.65 4922	7 2,026 3.57 91.3 0.66 4971	3 2,052 5.45 89.6 0.07 507 3 45 97.0 0.66 5031	2,081 3.33 95.6 0.66 506	3 2,118 3.22 83.1 0.66 512 5 2,115 3.17 78.0 0.66 508	2,161 4.12 49.0 0.69 48	7 2,125 4.37 23.1 0.66 4	3 2,017 4.01 51.1 0.65 4 2 2,008 3.50 2.4.4 0.67 4	7 2,098 3.34 62.8 0.68 4	8 2,093 3.42 65.3 0.67 4	7 2,092 3.37 64.5 0.67	3 2,082 3.38 71.2 0.67	3 2,089 3.26 72.6 0.67	7 2,067 3.47 88.0 0.67 2 2,145 3.31 55.0 0.68	8 2,107 3.14 87.3 0.68	2,084 3.23 95.7 0.67	6 2,075 3.36 95.3 0.67 2 7 2,081 3.13 88.7 0.67 2	3 2,071 3.08 92.4 0.67 4	0 2,070 3.43 116.4 0.67 44	5 2,115 3.28 77.9 0.66 46	5 2,104 5.25 82.4 0.66 46 5 2.062 3.14 81.0 0.65 45	2 ,043 3 .68 5 4.5 0 .66 4 ;	0 2,064 3.50 50.5 0.66 45	3 2,075 3.49 66.3 0.66	3.56 72.6 0.66	2,069 3.36 63.9 0.66	4 20.0 8.07 70.00 10.00 4 5 2.042 3.59 81.6 0.66 4	3 2,030 3.55 85.9 0.65 44 8 2,059 3.50 86.9 0.66 44
	Fresh Recycle Reactor Purge Inlet Sup. Space s. Feed Gas Feed Gas Velocity Velocity g) (KSCFH) (KSCFH) (H2.CO) (KSCFH) (Ih.sec) (Ih.h.kg.)	1 644 2,184 2.86 127.5 0.68 6311	5 644 2,264 3.02 145,4 0.70 6499 5 668 2.205 2.87 1643 0.69 6418	5 665 2,211 2.92 165.9 0.69 6419 6 669 2,246 2.89 173.7 0.70 6516	5 473 2,020 3.94 48.9 0.63 5147	5 473 2,002 3.68 53.3 0.62 5093	6 473 2,001 3.20 52.0 0.62 5075 5 536 2.075 442 571 0.64 487	5 577 2,119 3.95 65.5 0.67 5047	5 578 2,058 3.81 79.6 0.67 5054	5 577 2,112 3.46 71.6 0.68 5115	5 572 2,012 3.71 86.9 0.65 4922	5 577 2,026 3.57 91.3 0.66 4971	0 2/8 2,052 3.43 89.6 0.6/ 20/0 5 578 2.057 3.45 07.0 0.66 5031	0 579 2,081 3.33 95.6 0.66 506	0 573 2,118 3.22 83.1 0.66 512 1 555 2,115 3.17 78.0 0.66 508	6 552 2,161 4.12 49.0 0.69 48 [.]	5 497 2,125 4.37 23.1 0.66 4	5 548 2,017 4.01 51.1 0.65 4 5 578 2.08 3.50 5.4 0.67 4	5 577 2,098 3.34 62.8 0.68 4	5 578 2,093 3.42 65.3 0.67 4	5 577 2,092 3.37 64.5 0.67 4 A 577 2.008 2.35 770 0.68	5 578 2,082 3.38 71.2 0.67	4 578 2,089 3.26 72.6 0.67	4 577 2,067 3.47 88.0 0.67 . 2 522 2.145 3.31 55.0 0.68	4 578 2,107 3.14 87.3 0.68	5 577 2,084 3.23 95.7 0.67	5 5/8 2,0/5 3.36 95.3 0.6/ 2 4 577 2,081 3.13 88.7 0.67 2	3 578 2,071 3.08 92.4 0.67 4	5 560 2,070 3.43 116.4 0.67 44	5 515 2,115 3.28 77.9 0.66 46	5 513 2.062 3.14 8.10 0.65 46	5 499 2,043 3.68 54.5 0.66 4;	6 499 2,064 3.50 50.5 0.66 45	4 513 2,075 3.49 66.3 0.66	4 513 2,041 3.56 72.6 0.66	4 501 2,069 3.36 63.9 0.66	5 508 2,042 3.59 81.6 0.66 4	4 508 2,030 3.55 85.9 0.65 4! 3 508 2,059 3.50 86.9 0.66 4!
	Fresh Recycle Reactor Purge Inlet Sup. Space mp Pres. Feed Gas Feed Gas Velocity Velocity g.C. (psig) (KSCFH) (RSCFH) (H2-CO) (KSCFH) (It/sec) (Uhr-kg)	49 701 644 2,184 2.86 127,5 0.68 6311	4.9 705 644 2,264 3.02 145.4 0.70 6499 4.9 705 668 2,205 2.87 1643 0.69 6418	49 705 665 2,211 2,92 165.9 0.69 6419 49 706 669 2,246 2.89 173.7 0.70 6516	50 675 473 2,020 3.94 48.9 0.63 5147	51 675 473 2,002 3.68 53.3 0.62 5093	51 676 473 2,001 3.20 52.0 0.62 5075 51 675 536 2.075 4.42 571 0.64 4872	51 675 577 2,119 3.95 65.5 0.67 5047	51 675 578 2,058 3.81 79.6 0.67 5054	51 675 577 2,112 3.46 71.6 0.68 5115	51 675 572 2,012 3.71 86.9 0.65 4922	49 675 577 2,026 3.57 91.3 0.66 4971	49 675 578 2.057 3.45 87.0 0.66 5031 49 675 578 2.057 3.45 97.0 0.66 5031	49 680 579 2,081 3.33 95.6 0.66 506	49 690 573 2,118 3.22 83.1 0.66 512 49 681 555 2,115 3.17 78.0 0.66 508	49 676 552 2,161 4.12 49,0 0.69 48 [.]	49 675 497 2,125 4.37 23.1 0.66 4	49 675 548 2,017 4.01 51.1 0.65 4 49 675 578 2,008 3.59 24.4 0.67 4	49 675 577 2,098 3.34 62.8 0.68 4	49 675 578 2,093 3.42 65.3 0.67 4	50 675 577 2,092 3.37 64.5 0.67 5 50 674 677 2.008 3.35 64.5 0.67 5	50 675 578 2,082 3.38 71.2 0.67	49 674 578 2,089 3.26 72.6 0.67	49 674 577 2,067 3.47 88.0 0.67 49 672 522 2.145 3.31 55.0 0.68 .	50 674 578 2,107 3.14 87.3 0.68	50 675 577 2,084 3.23 95.7 0.67	49 6/5 5/8 2,0/5 3.36 95.3 0.6/ 2 49 674 577 2.081 3.13 88.7 0.67 2	49 673 578 2,071 3.08 <u>92.4</u> 0.67 4	50 675 560 2,070 3.43 116.4 0.67 44	50 675 515 2,115 3.28 77.9 0.66 46	49 6/5 508 2,104 5.25 82.4 0.06 40 50 675 513 2.062 3.14 81.0 0.65 45	59 675 499 2,043 3.68 54.5 0.66 4	59 676 499 2,064 3.50 50.5 0.66 45	58 674 513 2,075 3.49 66.3 0.66	59 674 513 2,041 3.56 72.6 0.66	59 674 501 2,069 3.36 63.9 0.66	59 675 508 2,042 3.59 81.6 0.66 4	59 674 508 2,030 3.55 85.9 0.65 44 559 673 508 2,059 3.50 86.9 0.66 44
	Fresh Recycle Reactor Purge Inlet Sup. Space is Temp Pres. Feed Gas Feed Gas Velocity Velocity pe (Deg C) (psig) (KSCFH) (KSCFH) (H2-CO) (KSCFH) (th/se) (I/hr-kg)	need 249 701 644 2,184 2.86 127.5 0.68 6311	nced 249 705 644 2,264 3.02 145,4 0.70 6499 nced 249 705 668 2.205 2.87 1643 0.69 6418	need 249 705 665 2,211 2.92 165.9 0.69 6419 Novel 249 706 669 2,246 2.89 173.7 0.70 6516	need 250 675 473 2,020 3.94 48.9 0.63 5147	need 251 675 473 2,002 3.68 53.3 0.62 5093	need 251 676 473 2,001 3.20 52.0 0.62 5075 251 675 536 2.075 4.42 571 0.64 4872	need 251 675 577 2,119 3.95 65,5 0.67 5047	need 251 675 578 2,058 3.81 79.6 0.67 5054	need 251 675 577 2,112 3.46 71.6 0.68 5115	nced 251 675 572 2,012 3.71 86.9 0.65 4922	need 249 675 577 $2,026$ 3.57 91.3 0.66 4971	nced 249 0/0 3/8 2,082 3.43 89.6 0.07 30/0 houd 249 675 578 2.057 3.45 07.0 0.66 5031	need 249 680 579 2,081 3.33 95,6 0.66 506	Inced 249 690 573 2,118 3.22 83.1 0.66 512 Vood 249 681 555 2,115 3.17 78,0 0.66 508	need 249 676 552 2,161 4.12 49,0 0.69 48	nced 249 675 497 2,125 4.37 23.1 0.66 4	need 249 675 548 2,017 4.01 51.1 0.65 4 249 675 578 2.098 3.50 2.4.4 0.67 4	need 249 675 577 2,098 3.34 62.8 0.68 4	need 249 675 578 2,093 3.42 65.3 0.67 4	need 250 675 577 2,092 3.37 64.5 0.67	$\frac{1000}{1000} = \frac{2.0}{250} = \frac{0.1}{675} = \frac{0.1}{578} = \frac{0.02}{2.082} = \frac{0.03}{3.38} = \frac{0.03}{71.2} = \frac{0.03}{0.67}$	need 249 674 578 2,089 3.26 72.6 0.67	need 249 674 577 2,067 3.47 88.0 0.67	need 250 674 578 2,107 3.14 87,3 0.68	nced 250 675 577 2,084 3.23 95.7 0.67	nced 249 6/5 5/8 2,0/5 3.36 95.3 0.67 2 nced 249 674 577 2,081 3.13 88.7 0.67 2	nced 249 673 578 2,071 3.08 92.4 0.67 4	need 250 675 560 2,070 3.43 116.4 0.67 44	nced 250 675 515 2,115 3.28 77.9 0.66 46	nced 249 6/5 508 2,104 5.25 82.4 0.66 46 hoved 250 675 513 2.062 3.14 8.10 0.65 45	need 259 675 499 2,043 3.68 54.5 0.66 44	nced 259 676 499 2,064 3.50 50.5 0.66 45	nced 258 674 513 2,075 3.49 66.3 0.66	nced 259 674 513 2,041 3.56 72.6 0.66	need 259 674 501 2,069 3.36 63.9 0.66	need 259 675 508 2,042 3.59 81.6 0.66 4	nced 259 674 508 2,030 3.55 85.9 0.65 4: reed 259 673 508 2,059 3.50 86.9 0.66 4:
	Fresh Resyle Reactor Purge Inlet Sup. Space 1 Gas Temp Pres. Feed Gas Feed Gas Velocity Velocity m Type (Deg.C) (psig) (KSCFH) (KSCFH) (H2:CO) (KSCFH) (th:sec) (thr.kg)	i Balanced 249 701 644 2,184 2.86 127,5 0.68 6311	8 Balanced 249 705 644 2,264 3.02 145,4 0.70 6499) Balanced 249 705 668 2,205 2.87 1643 0.69 6418	D Balanced 249 705 665 2,211 2.92 165.9 0.69 6419 Ralanced 249 706 669 2,246 2.89 173.7 0.70 6516	2147 230 675 473 2,020 3.94 48.9 0.63 5147	3 Balanced 251 675 473 2,002 3.68 53.3 0.62 5093	4 Balanced 251 676 473 2,001 3.20 52.0 0.62 5075 5 Delococod 251 675 536 2.05 4.42 571 0.64 4872	5 Balanced 251 675 577 2,119 3.95 65.5 0.67 5047	7 Balanced 251 675 578 2,058 3.81 79,6 0.67 5054	3 Balanced 251 675 577 2,112 3.46 71.6 0.68 5115	9 Balanced 251 675 572 2,012 3.71 86.9 0.65 4922) Balanced 249 675 577 2,026 3.57 91.3 0.66 4971	I Balanced 249 0/0 5/6 2,082 5.45 89.6 0.0/ 50/0 V Balanced 249 675 578 2.057 3.45 07.0 0.66 5031	1 Balanced 249 680 579 2,081 3.33 95.6 0.66 506	4 Balanced 249 690 573 2,118 3.22 83.1 0.66 512 : Relationed 249 681 555 2,115 3.17 78.0 0.66 508) Balanced 249 676 552 2,161 4.12 49,0 0.69 48	t Balanced 249 675 497 2,125 4.37 23.1 0.66 4	2 Balanced 249 675 548 2,017 4,01 51.1 0.65 4 5 Dalarced 240 675 578 2,098 3.50 24.4 0.67 4	t Balanced 249 675 577 2,098 3.34 62.8 0.68 4	5 Balanced 249 675 578 2,093 3.42 65.3 0.67 4	5 Balanced 250 675 577 2,092 3.37 64.5 0.67 c	Balanced 250 675 578 2,082 3.38 71.2 0.67	Datanced 249 674 578 2,089 3.26 72.6 0.67	D Balanced 249 674 577 2,067 3.47 88.0 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.67 0.68	i Balanced 250 674 578 2,107 3.14 87,3 0.68	3 Balanced 250 675 577 2,084 3.23 95.7 0.67	4 Balanced 249 6/5 5/8 2,0/5 3.36 95.3 0.6/ 2 5 Balanced 249 674 577 2,081 3.13 88.7 0.67 2	5 Balanced 249 673 578 2,071 3.08 92.4 0.67 4	3 Balanced 250 675 560 2,070 3.43 116.4 0.67 44	9 Balanced 250 675 515 2,115 3.28 77.9 0.66 46	D Balanced 249 0/5 508 2,104 5.25 82.4 0.06 44 r Balanced 250 675 513 2.062 3.14 81.0 0.65 45) Balanced 259 675 499 2,043 3.68 54,5 0.66 4	3 Balanced 259 676 499 2,064 3.50 50.5 0.66 45	t Balanced 258 674 513 2,075 3.49 66.3 0.66	5 Balanced 259 674 513 2,041 3.56 72.6 0.66	5 Balanced 259 674 501 2,069 3.36 63.9 0.66	(* 1) Datiatriced 259 675 508 2,042 3.59 81.6 0.66 4	9 Balanced 259 674 508 2,030 3.55 85.9 0.65 44 1 Balanced 259 673 508 2,059 3.50 86.9 0.66 44
	Days Fresh Recycle Reactor Purge Inlet Sup. Space On Gas Temp Pres. Feed Gas Feed Gas Velocity Velocity <td< th=""><th>.97 146 Balanced 249 701 644 2,184 2.86 127.5 0.68 6311</th><th>97 148 Balanced 249 705 644 2,264 3.02 145,4 0.70 6499 77 149 Balanced 249 705 668 2,205 2,87 1643 0,69 6418</th><th>97 150 Balanced 249 705 665 2.211 2.92 1659 0.69 6419 37 151 Ralanced 249 706 669 2.246 2.89 1737 0.70 6516</th><th>97 162 Balanced 250 675 473 2,020 3.94 48,9 0.63 5147</th><th>.97 163 Balanced 251 675 473 2,002 3.68 53.3 0.62 5093</th><th>-97 164 Balanced 251 676 473 2,001 3,20 52,0 0.62 5075 07 125 Delanced 251 675 536 7.075 4.47 571 0.64 4877</th><th>-71 105 Balanced 251 675 577 2,119 3.95 65,5 0.67 5047</th><th>97 167 Balanced 251 675 578 2,058 3.81 79.6 0.67 5054</th><th>.97 168 Balanced 251 675 577 2,112 3.46 71.6 0.68 5115</th><th>-97 169 Balanced 251 675 572 2,012 3.71 86,9 0.65 4922</th><th>-97 170 Balanced 249 675 577 2,026 3.57 91.3 0.66 4971</th><th>-9/ 1/1 Balanced 249 0/0 3/6 2,082 3.43 89.6 0.0/ 30/6 07 172 Balanced 249 675 578 2.057 3.45 07.0 0.66 5031</th><th>-97 173 Balanced 249 680 579 2,081 3.33 95.6 0.66 506</th><th>.97 174 Balanced 249 690 573 2,118 3.22 83.1 0.66 512 07 175 Ralanced 249 681 555 2,115 3.17 78.0 0.66 508</th><th>97 180 Balanced 249 676 552 2,161 4.12 49.0 0.69 48</th><th>97 181 Balanced 249 675 497 2,125 4.37 23.1 0.66 4</th><th>97 182 Balanced 249 675 548 2,017 4,01 51.1 0.65 4 27 102 Delensed 240 675 578 2,018 3.50 544 0.67 4</th><th>37 184 Balanced 249 675 577 2,098 3.34 62.8 0.68 4</th><th>97 185 Balanced 249 675 578 2,093 3.42 65.3 0.67 4</th><th>97 186 Balanced 250 675 577 2,092 3.37 64.5 0.67 6 07 107 D-1-1-12 250 674 677 2,098 3.25 670 0.68 7</th><th>97 188 Balanced 250 675 578 2,082 3.38 71.2 0.67</th><th>.97 189 Balanced 249 674 578 2,089 3.26 72.6 0.67</th><th>-97 190 Balanced 249 674 577 2,067 3.47 88.0 0.67 . 07 101 Balanced 249 672 572 2,145 3,31 55.0 0.68 .</th><th>97 192 Balanced 250 674 578 2,107 3.14 87.3 0.68</th><th>.97 193 Balanced 250 675 577 2,084 3.23 95.7 0.67</th><th>-97 194 Balanced 249 6/5 5/8 2/0/5 5.36 95.3 0.6/ / 97 195 Balanced 249 674 577 2.081 3.13 887 0.67 /</th><th>97 196 Balanced 249 673 578 2,071 3.08 92.4 0.67 4</th><th>.97 198 Balanced 250 675 560 2,070 3.43 116.4 0.67 4</th><th>.97 199 Balanced 250 675 515 2,115 3.28 77.9 0.66 46</th><th>-97 200 Balanced 249 6/5 508 2,104 3.25 82.4 0.56 44 97 201 Balanced 250 675 513 2.062 3.14 81.0 0.65 45</th><th>97 202 Balanced 259 675 499 2,043 3.68 54.5 0.66 4</th><th>97 203 Balanced 259 676 499 2,064 3.50 50.5 0.66 45</th><th>97 204 Balanced 258 674 513 2,075 3.49 66.3 0.66</th><th>.97 205 Balanced 259 674 513 2,041 3.56 72.6 0.66</th><th>-97 206 Balanced 259 674 501 2,069 3.36 63.9 0.66</th><th>-97 208 Balanced 259 675 508 2,042 3.59 81,6 0.66 4</th><th>97 209 Balanced 259 674 508 2,030 3.55 85,9 0.65 4; 97 210 Balanced 259 673 508 2,059 3.50 86,9 0.66 4;</th></td<>	.97 146 Balanced 249 701 644 2,184 2.86 127.5 0.68 6311	97 148 Balanced 249 705 644 2,264 3.02 145,4 0.70 6499 77 149 Balanced 249 705 668 2,205 2,87 1643 0,69 6418	97 150 Balanced 249 705 665 2.211 2.92 1659 0.69 6419 37 151 Ralanced 249 706 669 2.246 2.89 1737 0.70 6516	97 162 Balanced 250 675 473 2,020 3.94 48,9 0.63 5147	.97 163 Balanced 251 675 473 2,002 3.68 53.3 0.62 5093	-97 164 Balanced 251 676 473 2,001 3,20 52,0 0.62 5075 07 125 Delanced 251 675 536 7.075 4.47 571 0.64 4877	-71 105 Balanced 251 675 577 2,119 3.95 65,5 0.67 5047	97 167 Balanced 251 675 578 2,058 3.81 79.6 0.67 5054	.97 168 Balanced 251 675 577 2,112 3.46 71.6 0.68 5115	-97 169 Balanced 251 675 572 2,012 3.71 86,9 0.65 4922	-97 170 Balanced 249 675 577 2,026 3.57 91.3 0.66 4971	-9/ 1/1 Balanced 249 0/0 3/6 2,082 3.43 89.6 0.0/ 30/6 07 172 Balanced 249 675 578 2.057 3.45 07.0 0.66 5031	-97 173 Balanced 249 680 579 2,081 3.33 95.6 0.66 506	.97 174 Balanced 249 690 573 2,118 3.22 83.1 0.66 512 07 175 Ralanced 249 681 555 2,115 3.17 78.0 0.66 508	97 180 Balanced 249 676 552 2,161 4.12 49.0 0.69 48	97 181 Balanced 249 675 497 2,125 4.37 23.1 0.66 4	97 182 Balanced 249 675 548 2,017 4,01 51.1 0.65 4 27 102 Delensed 240 675 578 2,018 3.50 544 0.67 4	37 184 Balanced 249 675 577 2,098 3.34 62.8 0.68 4	97 185 Balanced 249 675 578 2,093 3.42 65.3 0.67 4	97 186 Balanced 250 675 577 2,092 3.37 64.5 0.67 6 07 107 D-1-1-12 250 674 677 2,098 3.25 670 0.68 7	97 188 Balanced 250 675 578 2,082 3.38 71.2 0.67	.97 189 Balanced 249 674 578 2,089 3.26 72.6 0.67	-97 190 Balanced 249 674 577 2,067 3.47 88.0 0.67 . 07 101 Balanced 249 672 572 2,145 3,31 55.0 0.68 .	97 192 Balanced 250 674 578 2,107 3.14 87.3 0.68	.97 193 Balanced 250 675 577 2,084 3.23 95.7 0.67	-97 194 Balanced 249 6/5 5/8 2/0/5 5.36 95.3 0.6/ / 97 195 Balanced 249 674 577 2.081 3.13 887 0.67 /	97 196 Balanced 249 673 578 2,071 3.08 92.4 0.67 4	.97 198 Balanced 250 675 560 2,070 3.43 116.4 0.67 4	.97 199 Balanced 250 675 515 2,115 3.28 77.9 0.66 46	-97 200 Balanced 249 6/5 508 2,104 3.25 82.4 0.56 44 97 201 Balanced 250 675 513 2.062 3.14 81.0 0.65 45	97 202 Balanced 259 675 499 2,043 3.68 54.5 0.66 4	97 203 Balanced 259 676 499 2,064 3.50 50.5 0.66 45	97 204 Balanced 258 674 513 2,075 3.49 66.3 0.66	.97 205 Balanced 259 674 513 2,041 3.56 72.6 0.66	-97 206 Balanced 259 674 501 2,069 3.36 63.9 0.66	-97 208 Balanced 259 675 508 2,042 3.59 81,6 0.66 4	97 209 Balanced 259 674 508 2,030 3.55 85,9 0.65 4; 97 210 Balanced 259 673 508 2,059 3.50 86,9 0.66 4;
	Days Fresh Recycle Ranctor Purge Inlet Sup. Space On Gas Temp Pres. Feed Gas Feed Gas Velocity Velocity <td< th=""><th>30-Aug-97 146 Balanced 249 701 644 2,184 2.86 127.5 0.68 6311</th><th>1-Sep-97 148 Balanced 249 705 644 2,264 3.02 145,4 0.70 6499 2-Sen-97 149 Balanced 249 705 668 2.205 2.87 1643 0.69 6418</th><th>3-Sep-97 150 Balanced 249 705 665 2,211 2,92 165,9 0,69 6419 4-Sen-97 151 Balanced 249 706 669 2,246 2,89 1737 0,70 6516</th><th>15-Sep-97 162 Balanced 250 675 473 2,020 3.94 48,9 0.63 5147</th><th>16-Sep-97 163 Balanced 251 675 473 2,002 3.68 53.3 0.62 5093</th><th>17-Sep-97 164 Balanced 251 676 473 2,001 3.20 52.0 0.62 5075 10 6-2007 145 Dalmanoi 251 675 536 7075 4.42 573 0.64 4872</th><th>19-5ep-97 105 Balanced 251 675 577 2,119 3.95 655 0.67 5047</th><th>20-Sep-97 167 Balanced 251 675 578 2,058 3.81 79,6 0.67 5054</th><th>21-Sep-97 168 Balanced 251 675 577 2,112 3.46 71.6 0.68 5115</th><th>22-Sep-97 169 Balanced 251 675 572 2,012 3.71 86,9 0.65 4922</th><th>23-Sep-97 170 Balanced 249 675 577 2,026 3.57 91.3 0.66 4971</th><th>24-Sep-9/ 1/1 Balanced 249 0/0 5/6 2/02 5/45 8/56 0/0 50/0 55.5em.07 179 Balanced 249 675 578 2.057 3.45 070 0.66 5031</th><th>26-56p-97 173 Balanced 249 680 579 2,081 3.33 95,6 0.66 506</th><th>27-Sep-97 174 Balanced 249 690 573 2,118 3.22 83.1 0.66 512 9x.Sen-07 175 Ralanced 249 681 555 2,115 3.17 78.0 0.66 508</th><th>3-Oct-97 180 Balanced 249 676 552 2,161 4.12 49,0 0.69 48</th><th>4-Oct-97 181 Balanced 249 675 497 2,125 4.37 23.1 0.66 4</th><th>5-Oct-97 182 Balanced 249 675 548 2,017 4,01 51.1 0.65 4 6 Oct 07 102 Deliverad 249 675 578 2,088 3.50 2.4.4 0.67 4</th><th>7-0ct-97 184 Balanced 249 675 577 2,098 3.34 62.8 0.68 4</th><th>8-Oct-97 185 Balanced 249 675 578 2,093 3.42 65.3 0.67 4</th><th>9-0ct-97 186 Balanced 250 675 577 2,092 3.37 64.5 0.67 c</th><th>10-001-97 160 Balanteed 250 675 578 2,082 3.38 71.2 0.67</th><th>12-Oct-97 189 Balanced 249 674 578 2,089 3.26 72.6 0.67</th><th>13-0ct-97 190 Balanced 249 674 577 2,067 3.47 88.0 0.67 . 14-0-4-07 101 Balanced 249 677 572 7.145 3.31 55.0 0.68 .</th><th>15-00t-97 192 Balanced 250 674 578 2,107 3.14 87,3 0.68</th><th>16-Oct-97 193 Balanced 250 675 577 2,084 3.23 95.7 0.67</th><th>17-0ct-97 194 Balanced 249 6/5 5/8 2/075 5.36 95.3 0.6/ / 18-0ct-97 195 Balanced 249 674 577 2.081 3.13 88.7 0.67 /</th><th>19-00t-97 196 Balanced 249 673 578 2,071 3.08 92.4 0.67 4</th><th>21-Oct-97 198 Balanced 250 675 560 2,070 3.43 116.4 0.67 44</th><th>22-Oct-97 199 Balanced 250 675 515 2,115 3.28 77.9 0.66 46</th><th>23-00f-97 200 Balanced 249 6/5 503 208 2,104 5.25 82.4 0.66 46 24.04-697 201 Balanced 250 675 513 2.062 3.14 81.0 0.65 45</th><th>25-00t-97 202 Balanced 259 675 499 2,043 3.68 54,5 0.66 4;</th><th>26-Oct-97 203 Balanced 259 676 499 2,064 3.50 50.5 0.66 45</th><th>27-Oct-97 204 Balanced 258 674 513 2,075 3.49 66.3 0.66</th><th>28-Oct-97 205 Balanced 259 674 513 2,041 3.56 72.6 0.66</th><th>29-Oct-97 206 Balanced 259 674 501 2,069 3.36 63.9 0.66</th><th>3U-U-U-97 201 Datanced 2-7 0/2 200 2/01 2/02 0/03 4 3.1-02-97 208 Balanced 259 675 508 2/042 3.59 81.6 0.66 4</th><th>1-Nov-97 209 Balanced 259 674 508 2,030 3.55 85,9 0.65 4; 2-Nov-97 210 Balanced 259 673 508 2,059 3.50 86,9 0.66 4;</th></td<>	30-Aug-97 146 Balanced 249 701 644 2,184 2.86 127.5 0.68 6311	1-Sep-97 148 Balanced 249 705 644 2,264 3.02 145,4 0.70 6499 2-Sen-97 149 Balanced 249 705 668 2.205 2.87 1643 0.69 6418	3-Sep-97 150 Balanced 249 705 665 2,211 2,92 165,9 0,69 6419 4-Sen-97 151 Balanced 249 706 669 2,246 2,89 1737 0,70 6516	15-Sep-97 162 Balanced 250 675 473 2,020 3.94 48,9 0.63 5147	16-Sep-97 163 Balanced 251 675 473 2,002 3.68 53.3 0.62 5093	17-Sep-97 164 Balanced 251 676 473 2,001 3.20 52.0 0.62 5075 10 6-2007 145 Dalmanoi 251 675 536 7075 4.42 573 0.64 4872	19-5ep-97 105 Balanced 251 675 577 2,119 3.95 655 0.67 5047	20-Sep-97 167 Balanced 251 675 578 2,058 3.81 79,6 0.67 5054	21-Sep-97 168 Balanced 251 675 577 2,112 3.46 71.6 0.68 5115	22-Sep-97 169 Balanced 251 675 572 2,012 3.71 86,9 0.65 4922	23-Sep-97 170 Balanced 249 675 577 2,026 3.57 91.3 0.66 4971	24-Sep-9/ 1/1 Balanced 249 0/0 5/6 2/02 5/45 8/56 0/0 50/0 55.5em.07 179 Balanced 249 675 578 2.057 3.45 070 0.66 5031	26-56p-97 173 Balanced 249 680 579 2,081 3.33 95,6 0.66 506	27-Sep-97 174 Balanced 249 690 573 2,118 3.22 83.1 0.66 512 9x.Sen-07 175 Ralanced 249 681 555 2,115 3.17 78.0 0.66 508	3-Oct-97 180 Balanced 249 676 552 2,161 4.12 49,0 0.69 48	4-Oct-97 181 Balanced 249 675 497 2,125 4.37 23.1 0.66 4	5-Oct-97 182 Balanced 249 675 548 2,017 4,01 51.1 0.65 4 6 Oct 07 102 Deliverad 249 675 578 2,088 3.50 2.4.4 0.67 4	7-0ct-97 184 Balanced 249 675 577 2,098 3.34 62.8 0.68 4	8-Oct-97 185 Balanced 249 675 578 2,093 3.42 65.3 0.67 4	9-0ct-97 186 Balanced 250 675 577 2,092 3.37 64.5 0.67 c	10-001-97 160 Balanteed 250 675 578 2,082 3.38 71.2 0.67	12-Oct-97 189 Balanced 249 674 578 2,089 3.26 72.6 0.67	13-0ct-97 190 Balanced 249 674 577 2,067 3.47 88.0 0.67 . 14-0-4-07 101 Balanced 249 677 572 7.145 3.31 55.0 0.68 .	15-00t-97 192 Balanced 250 674 578 2,107 3.14 87,3 0.68	16-Oct-97 193 Balanced 250 675 577 2,084 3.23 95.7 0.67	17-0ct-97 194 Balanced 249 6/5 5/8 2/075 5.36 95.3 0.6/ / 18-0ct-97 195 Balanced 249 674 577 2.081 3.13 88.7 0.67 /	19-00t-97 196 Balanced 249 673 578 2,071 3.08 92.4 0.67 4	21-Oct-97 198 Balanced 250 675 560 2,070 3.43 116.4 0.67 44	22-Oct-97 199 Balanced 250 675 515 2,115 3.28 77.9 0.66 46	23-00f-97 200 Balanced 249 6/5 503 208 2,104 5.25 82.4 0.66 46 24.04-697 201 Balanced 250 675 513 2.062 3.14 81.0 0.65 45	25-00t-97 202 Balanced 259 675 499 2,043 3.68 54,5 0.66 4;	26-Oct-97 203 Balanced 259 676 499 2,064 3.50 50.5 0.66 45	27-Oct-97 204 Balanced 258 674 513 2,075 3.49 66.3 0.66	28-Oct-97 205 Balanced 259 674 513 2,041 3.56 72.6 0.66	29-Oct-97 206 Balanced 259 674 501 2,069 3.36 63.9 0.66	3U-U-U-97 201 Datanced 2-7 0/2 200 2/01 2/02 0/03 4 3.1-02-97 208 Balanced 259 675 508 2/042 3.59 81.6 0.66 4	1-Nov-97 209 Balanced 259 674 508 2,030 3.55 85,9 0.65 4; 2-Nov-97 210 Balanced 259 673 508 2,059 3.50 86,9 0.66 4;

D-4

D-5

uilyst Reactor U All-kg) (TPD)fi3) hr.fh2 F) (ps) All-kg) (TPD)fi3) hr.fh2 F) (ps) 0.77 0.082 139 903 0.399 0.009 135 9454 0.25 0.094 136 887 0.88 0.107 127 1011 0.89 0.120 139 9464 0.84 0.107 127 1011 0.84 0.107 127 1011 0.84 0.106 148 5.53 0.111 147 5.63 5.11 0.87 0.106 148 5.53 0.112 0.106 148 5.53 0.113 0.106 148 5.53 0.112 146 5.53 5.11 0.113 0.106 148 5.53 0.111 147 5.64 5.12 0.122 0.103 146 5.53	20.12 0.112 156 555 19.88 0.100 138 5.27 19.32 0.104 143 5.04 18.81 0.106 143 4.96 18.81 0.106 144 4.96
Ialyst Reactor Ower H Prod. Vol. Prod. Ower Mr-kg) (TPD)f3) hrfl2 0.77 0.082 139 0.25 0.094 136 0.86 0.107 137 0.89 0.107 137 0.89 0.120 139 0.81 0.107 137 0.82 0.109 135 0.83 0.107 143 0.11 147 144 0.122 0.106 148 0.11 0.106 148 0.11 0.106 148 0.11 0.106 148 0.11 0.106 148 0.122 0.107 146 0.123 0.008 146 0.123 0.008 146 0.103 0.008 146 0.103 0.008 146 0.103 0.008 146 0.103 0.084	20.12 0.112 156 19.88 0.100 138 19.32 0.104 143 18.81 0.106 148
ialyst Reactor H.Prod. Vol. Prod. U.Mr-kg.) (TPD/ff) 0.77 0.082 0.39 0.093 0.25 0.093 0.88 0.107 0.80 0.1120 0.81 0.1120 0.82 0.101 0.84 0.102 0.84 0.102 0.84 0.102 0.84 0.103 0.84 0.103 0.84 0.103 0.84 0.103 0.87 0.103 0.84 0.003 0.84 0.003 0.93 0.033 0.93 0.033 0.93 0.093 0.93 0.093 0.93 0.093 0.93 0.093 0.94 0.093 0.95 0.093 0.93 0.093 0.94 0.093 0.95 0.093 0.94	20.12 0.112 19.88 0.100 19.32 0.104 18.81 0.106
Idialyst Idi	20.12 19.88 19.32 18.81
W W W W W W W W W W W W W W W W W W W	
Raw MedH Prod. (TPD) 1926.0 1926.0 197.6 197.6 197.6 211.9 211.9 211.9 212.3 219.9 221.3 2221.3 22222.3 2222.3 2222.3 2222.3 2222.3 2222.3 2222.3 2222.3 2	211.8 208.4 203.2 197.9
Syngas Utit. (SCF/Ib)	41.3 41.9 42.2 40.9
Reactor 6-1-M (%) (%) (%) (%) (%) (%) (%) (%) (%) (%)	20.1 19.9 19.6 19.3
CO Conv. (%) (%) (%) (%) (%) (%) (%) (%) (%) (%)	34.0 35.4 36.5 37.2
Catalyst Age (eta) Age (eta) Age (eta) (et	0.68 0.68 0.68 0.67
Catalyst Inventory (0b) 24,800 24,800 24,800 24,800 24,800 24,800 27,450 27,755 27,450 27,755 27,450 27,7555 27,7555 27,7555 27,75555 27,7555557575757575757575757575757575757	27,450 27,450 27,450 27,450
Gassed Gassed Flurty Hgt (f) Hgt (f) 49.0 49.0 49.0 49.0 50.0 59.0 59.0 59.0 59.0 59.0 59.0 5	45.0 49.5 46.5 44.5
Gas Collection (1997) Collecti	46.5 43.9 42.4 42.4
Slurry Cone. (wt%.ox) 35.1 35.1 37.0 37.6 37.6 37.6 37.6 37.6 37.6 37.6 37.6	40.3 36.7 38.1 38.8
Spance Velocity (I/Ihr-kg) 7,286 7,283 7,519 7,519 7,519 6,411 6,401 6,427 6,337 6,337 6,343 6,441 6,441 6,265 6,334 6,441 6,245 6,347 6,347 6,347 6,347 6,347 6,347 6,346 6,441 6,4	6,413 6,358 6,298 6,231
Inlet Sup. Velocity. (11/sec) 0.70 0.70 0.71 0.72 0.68 0.68 0.68 0.66 0.66 0.66 0.66 0.66	0.68 0.67 0.66 0.66
Purge Gas Gas 128:4 133.7 133.7 133.6 133.6 133.6 133.6 133.6 133.6 133.6 133.6 133.6 133.6 13.6 1	85.5 96.5 98.5 74.0
cantor 12:CO) 0 2:CO) 0 2:S85 0.81 0.81 0.74 0.81 4.45 4.45 4.45 4.45 4.45 4.45 4.45 4.4	3.40 3.63 3.88 4.02
ecycle F Gas Gas Cas Cas Cas Cas Cas Cas Cas Cas Cas C	2,247 2,219 2,223 2,235
resh F SCFH) (F SCFH) (F 7724 7724 7729 7729 7728 7728 7728 7728 7729 7729	729 715 574
Fress 1 psig) 710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7710 7700 7700 7700 7700 7700 7700	710 710 710
Temp (Dag C) 223 224 225 225 225 225 225 225 225 225 225	224 224 224 224
Type Cas Texuco Texuco Salanced Salance	3alanced 3alanced 3alanced 3alanced
Days Con On Stream 55 55 55 55 55 55 55 55 55 66 1 1 66 1 1 66 6 1 1 66 6 1 1 77 7 7 7	92 93 1 94 1 95 1
Date :: 1-Feb-98 5-Fe	Mar-98 Mar-98 +. Mar-98 Mar-98
Clase 0 2 2 5 5 5 4 3 2 2 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	6 23 6 24 6 25 7 25

D-6

		Sparger Dariet	("K")		0/.0	12.5	5.48	5.46	5.49	5.48	5.51	5.57	5.73 5.53	5.70	5.71	5.70	5.70	5.69	5.71	5.64	5.60	5.59	5.70	5.05 	5.67	50.C	5.98 6.01	5.60	5.62	5.64	0.00 5 40	5.29	5.15	5.17	5.18	5.20	5.07	5.08	5.09	5.12	5.24	5.14	5.21 5.16
		parger	u (psi)		4.4/	4.88	4.91	5.44	5.28	5.19	5.24	5.25	5.28 5.10	5.43	5.26	4.83	4.75	4.91	5.07	5.02	5.01	5.15	5.30	55.C	5.19	11.6	5.60 5.54	5.23	5.00	4.96	4.94	4.53	4.46	4.40	4.25	4.41	10.0	4 90	4.94	4.89	4.64	4.97	5.19 4.78
	n	verall S	ft2 F)		051	148	147	152	152	145	149	148	150	143	143	152	151	150	146	152	146	151	151	149	148	142	141	146	145	144	145	147	149	150	151	150	140	140	148	147	148	145	145 144
		tor Food	(ft3) hr		5	0.00	2 1	0	12	14	8	0	4 2		5	80	11	5	22	62	75	L1	5	5	8	6/	5 5	L1	6/	00	2 2	0 4	15	00	75	5	= 2	2 4	. 6	8	12	33	91
		Reac Vol D	(TPD)		0.16	JI 0	010	0.11	0.10	0.05	0.05	0.10	0.10	0.1.0	0.0	0.08	0.08	0.0	0.08	0.07	0.07	0.02	0.07	0.07	0.01	0.07	80.0 80.0	0.07	0.07	0.08	0.0	0.0	0.07	0.08	0.07	0.07	0.10	01.0	0.0	0.05	0.10	0.10	0.05
		Catalyst MoOtt Bead	(gmol/hr-kg)		18.35	19.07	18.96	19.20	16.90	16.07	15.79	15.63	15.31	15.09	15.31	15.40	15.28	15.28	15.14	15.36	15.24	15.24	15.25	15.17	15.06	14.98	14.84 14.76	14.94	14.76	14.68	14.72	14.00	13.74	13.70	13.73	13.89	21.01	19.40	19.09	19.05	18.83	18.75	19.00 19.07
(p,	Raw	MeOH Brod	(TPD)		6.291	0.061	199.4	202.1	194.6	185.0	181.8	179.9	176.2	173.6	176.4	1.77.1	176.0	176.0	174.4	176.9	175.5	175.5	175.7	174.5	173.0	172.5	170.9 169.9	172.1	170.0	168.9	169.4	159.9	158.2	157.7	158.0	159.9	C.617	224.1	220.0	219.4	216.8	215.9	218.8 219.6
(cont		Syngas	(SCF/lb)		41.9	41.4 71.5	014	40.5	42.2	41.6	41.6	42.0	41.2	40.6	39.7	39.2	39.7	39.7	39.6	39.4	39.5	39.5	39.2	39.0	40.0	40.6	41.0 41.2	40.7	41.2	41.4	41.3	41.0	41.6	42.2	42.1	41.7	40.2	C.95	39.7	40.4	40.5	39.6	40.1
gn 2 (Reactor	O-T-M	(%)		c.91	6.61 10.4	19.7	19.7	19.0	18.4	17.9	17.8	17.4	17.2	17.9	18.0	17.9	17.9	17.8	18.3	18.2	18.0	17.7	17.8	17.7	17.7	17.2 17.1	17.5	17.4	17.3	17.3	16.0	15.9	15.9	16.1	15.9	4.77	22.6	22.5	22.5	22.3	22.1	22.4 22.6
mpai		00	(%)		41.5 24 4	36.0	C.DC	33.9	32.8	31.9	30.2	29.5	30.0 20.8	28.6	32.1	35.4	35.2	33.5	31.8	33.1	32.7	31.7	31.4	31.3	31.3	30.8	28.7 28.5	29.4	30.3	30.0	29.7	31.1	30.8	30.5	32.2	31.8	4/.5	45.7	47.6	47.7	48.9	46.2	44.6 46.5
st Ca		Catalyst	(eta)	:	0.68	C0.0	0.68	0.66	0.61	0.60	0.60	0.59	0.59	0.58	0.61	0.61	0.60	0.59	0.58	0.60	0.59	0.59	0.59	0.59	0.57	0.56	0.55 0.55	0.55	0.55	0.55	0.55	0.54	0.54	0.54	0.55	0.55	C0.0	0.65 0.65	0.65	0.64	0.64	0.62	0.61
Cataly		Catalyst	(lb)		21,450	27.450	27,450	27,450	30,050	30,050	30,050	30,050	30,050 20.050	30.050	30,050	30,050	30,050	30,050	30,050	30,050	30,050	30,050	30,050	30,050	30,050	30,050	30,050 30,050	30,050	30,050	30,050	30,050	30.050	30,050	30,050	30,050	30,050	30,050	30,050	30.050	30,050	30,050	30,050	30,050 30,050
port		Gassed	Hgt (ft)		C.44	43.5	5.44	44.0	45.5	47.0	44.5	43.0	40.5	37.5	50.5	53.0	51.5	50.0	50.5	53.5	55.5	54.0	54.0	55.5	52.5	52.0	48.0 47.5	53.5	51.5	50.5	49.0 5 2 5	51.5	50.0	47.0	50.0	49.5	0.10 2.02	5.05	53.0	53.5	50.5	50.0	54.0 52.0
Kings		Gas	(vol%)		40.2	0.14 0.14	416	42.9	43.1	41.6	42.1	40.9	39.0 27.7	36.8	42.1	44.8	43.9	42.7	41.8	43.9	44.5	44.8	45.0	44.9	44.4	43.5	41.0 42.3	43.5	42.0	41.7	40.7	6.04 1.14	41.2	39.7	40.5	41.2	41./	42.0	42.4	42.1	40.2	40.2	41.4
for I		Slurry	(wt% ox)		5.1.5	2.40 2.95	38.4	39.2	40.7	39.2	40.8	41.2	41.9	43.0	37.6	37.5	37.9	38.1	37.4	36.9	36.3	37.1	37.2	36.5	37.6	37.5	38.4 39.2	36.7	37.1	37.4	37.7	36.7	37.4	38.4	37.1	37.7	5.15	30.9 37.6	36.8	36.5	37.1	37.4	35.9 36.5
Table		Space	l/hr-kg)		6,003	0,242	6 152	6,268	5,696	5,615	5,646	5,620	5,605 5,578	5.604	5,519	5,462	5,447	5,451	5,468	5,436	5,426	5,506	5,570	5,550	5,505	5,473	5,543 5,535	5,517	5,473	5,459	5,440 5 5 5 5	5.487	5,481	5,455	5,413	5,554	5,619	5,601	5.615	5,625	5,537	5,584	5,577 5, <u>516</u>
nary		t Sup.	/sec) (:	507	00.		.66	.66	.65	.65	.64	104	1 19	.63	.63	.63	.63	.63	.62	.62	.63	.64	.64	.63	.63	1.64	.63	.63	.63	0.63	102	.63	.63	.62	64	10.	.0/ 166		.67	.65	.66	1.66 1.65
Sumn		ge Inle v.	EH) (H				. 4	. 0	9	1	4	8	, e	5 6	5	0	0	0	3 (4 0	5 (-	5		4	0	0 0	3 (3	7 0	2 2	n -	7 0	3 (9 (~ ~	» (, 0	1 0	2 0	5	3 (
Data		or Pur	0) (KSC		.06	00. 84	62	71.	91.	82.	.67	85.	75.		50.	45.	50.	52.	49.	46.	46.	. 48.	46.	39.	50.	. 61.	70.	63.	67.	.69	70.	63.	62.	67.	67.	59.	-5.	- 4 9.	52	.09	67.	53.	54.
-2 -]		e React	I) (H2:O		10.4	70.0 70.0	3.93	3.46	3.51	3.51	3.35	3.33	3.47	3.30	3.61	4.03	4.05	3.80	3.61	3.68	3.65	3.57	3.58	5.5	3.57	3.52	3.35 3.34	3.38	3.55	3.53	3.47	3.98	3.96	3.93	4.15	4.16	4.40	4.20	4.48	4.55	4.73	4.40	4.13
ble D		Recycle	(KSCFI		2,111	277,2	211/2	2,247	2,234	2,240	2,267	2,261	2,269	2.264	2,227	2,195	2,185	2,192	2,208	2,189	2,183	2,227	2,258	2,267	2,229	2,198	2,240 2,229	2,230	2,201	2,193	2,182	2,259	2,249	2,233	2,206	2,289	2,145	2,1/6	2.139	2,139	2,096	2,138	2,107 2,087
Ta		Fresh	(KSCFH)	į	6/4	700	189	682	684	641	630	630	605	587	583	578	582	582	576	580	578	578	573	567	577	583	583 583	583	583	583	583	552	549	554	554	555	C6/	738	729	738	733	713	732
		Drac	(psig)		017	017	2002	709	710	710	710	709	709	710	709	710	710	711	710	710	710	709	710	710	710	710	710	710	710	209	700	407 709	710	710	709	710	60/	10/	710	710	710	710	710
		Tame	(Deg C)		224	+77 +77	F22	224	221	220	219	219	218	218	219	220	220	220	219	220	220	219	220	220	220	220	219 219	220	220	219	219	220	219	219	219	220	234	C62 734	235	235	234	234	235 235
		Coc	Type		Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced
		Days	Stream	:	8 8	66 U	101	102	103	104	105	106	107	001 106	110	Ш	112	113	114	115	116	117	118	611	120	121	122 123	124	125	126	127	131	132	133	134	135	951	140	142	143	144	145	146 147
			Date		28-Mar-98	20-Mar-08	31-Mar-98	1-Apr-98	2-Apr-98	3-Apr-98	4-Apr-98	5-Apr-98	6-Apr-98	8-Apr-98	9-Apr-98	10-Apr-98	11-Apr-98	12-Apr-98	13-Apr-98	14-Apr-98	15-Apr-98	16-Apr-98	17-Apr-98	18-Apr-98	19-Apr-98	20-Apr-98	21-Apr-98 22-Apr-98	23-Apr-98	24-Apr-98	25-Apr-98	26-Apr-98	29-Apr-96 30-Apr-98	1-May-98	2-May-98	3-May-98	4-May-98	8-May-98	9-May-98 10-Mav-98	11-Mav-98	12-May-98	13-May-98	14-May-98	15-May-98 16-May-98
			Case		0	0 V	o ve	9	9	9	9	9	9	0 0	9	9	9	9	9	9	9	9	9	9	9	9	99	9	9	9	9	0 0	9	9	9	9	0	0 V	9 9	9	9	9	6

D-7

	5	r >parger Resist.		01.c 91.5	5.16	5.23	5.10	5.08	5.08	5.12	5.13	5.10	5.25	5.24	5.19	5.63	5.17	5.18	5.22	5.23	5.25	5.24	5.25	5.28	5.32	5.31	5.47	5.37	5.44	5.36	5.32	5.45	5.42	5.39	5.43	5.40	5.43	5.46 5.48	7.48	4.90	4.87	. 100
	0	oparge dP (nei)	(and)	10.C	4.63	4.53	4.83	4.88	4.82	4.51	4.51	4.75	4.61	4.61	4.78	4.70	4.32	4.56	4.38	4.81	10.4	4.85	4.63	4.59	4.44	4.34	4.20	4.94	4.92	4.84	4.66	4.68	4.62	4.48	4.18	4.46	4.30	4.31	00T	4.00	4.12	441
	D	(Btu br ft2 E)		<u> </u>	141	144	144	144	148	44	145	145	<u> </u>	143	144	146	142	140	143	141	137	135	137	137	134	133	179	206	201	202	200	207	204	207	200	201	210	211	208	213	214	213
		Vol. Prod. 77PD/#3)		0.101	0 100	0.104	0.107	0.109	0.107	0.104	0.105	0.104	0 102	0.100	0.095	0.097	0.072	0.073	0.074	0.102	0.10	0.096	0.099	0.101	0.103	0.098	0.096	0.105	0.099	0.099	0.097	0.103	0.100	0.098	0.094	0.094	0.096	0.098	0.000	0.083	0.079	0.081
	1-1-1-0	Catalyst MeOH Prod.	(Su mond)	4C.81	19.06	18.91	19.09	19.06	18.73	18.98	18.52	18.70	18.04	18.32	18.44	17.51	13.18	13.24	13.01	18.01	86./1 17.20	17.48	17.44	17.04	16.95	16.84	16.77	82.CI 17 34	17.24	16.98	16.33	17.10	17.18	17.18	16.17	16.02	16.31	16.32	13.96	14.14	14.17	14.36
(p	Raw	Prod.		1315	219.5	217.8	219.9	219.6	215.8	218.6	213.4	215.4	207.8	210.9	212.4	201.4	151.7	152.5	149.7	207.2	200.7	219.0	218.5	213.6	212.3	210.7	210.1	0.173	216.0	212.8	204.7	214.2	215.2	215.2	202.6	200.7	204.4	204.5 199.4	F.6.1	176.9	177.3	179.8
cont'	5	byngas Util. (SCE/Ib)	(au tota)	40.6	30.0	40.3	39.6	39.4	39.2	40.1	40.8	41.0	41.1	41.8	41.4	43.6	39.3	39.1	39.1	42.4	47.0 0 04	40.2	40.2	40.5	41.8	42.2	41.3	40.2 40.2	40.6	39.9	39.8	40.9	40.8	40.8	40.1	40.4	39.5	39.7 40.1	1.07	42.8	42.9	42.3
gn 2 (Reactor	O-1-M Conv.		21.6	21.9	22.7	22.9	22.7	22.6	22.9	22.4	22.4 32.5	0.22	21.7	21.9	20.8	16.7	16.7	16.7	21.1	2.12	22.8 22.8	22.9	22.5	22.5	22.6	22.6	21.3	22.5	22.5	22.0	22.6	23.0	23.2	22.3	21.9	22.3	22.5 77 0	18.0	19.4	19.4	18.6
mpaig	ę	Conv.	6	41.7	6.24 64 3	45.6	42.9	42.8	43.3	45.2	45.1	42.5	41.9	41.0	40.1	44.3	40.4	38.5	39.6	38.4	2.65	6.64 6.45	45.6	45.7	45.6	46.4	46.9	40.8	40.9	41.2	42.2	42.3	42.8	43.5	45.8	43.5	45.3	45.6 43.6	0.0±	33.8	32.7	32.9
st Ca	1-1-2	Age (ata)	(112)	00.0	0.6.0	0.61	0.60	0.60	0.60	0.60	0.58	0.57	0.53	0.53	0.54	0.53	0.50	0.48	0.48	0.51	16.0	0.54	0.55	0.54	0.53	0.53	0.53	16.0	0.51	0.51	0.50	0.52	0.52	0.53	0.52	0.50	0.53	0.53	12:0	0.38	0.38	0.38
Cataly	11111	Catalyst Inventory (Ib)		30,050 020.05	30.050	30,050	30,050	30,050	30,050	30,050	30,050	30,050 30,050	30.050	30,050	30,050	30,050	30,050	30,050	30,050	30,050	0002.05	32,700 32,700	32,700	32,700	32,700	32,700	32,700	32,700 32,700	32,700	32,700	32,700 32,700	32,700	32,700	32,700	32,700	32,700	32,700	32,700 32,700	32,700	32,700 32,700	32,700	32,700
port		Gassed Slurry Hot (#)	(m) 4911	2.0C	50 0 52 0	50.0	49.0	48.0	48.0	50.0	48.5	49.5	0.0c	50.0	53.0	49.5	50.5	50.0	48.0	48.5	4/.0 52.0	54.0	52.5	50.5	49.0	51.0	52.0	0.1c	52.0	51.0	50.0	49.5	51.0	52.0	51.5	51.0	50.5	49.5 48 5	46.5	50.5	53.5	53.0
Kings	ļ	Holdup	(6/104)	40.4	40.4	39.5	39.8	39.9	37.9	39.7	39.4	39.9	40.4 30.0	40.3	41.2	42.7	41.1	39.0	38.6	40.2	C 285	4 7 1 2	39.4	40.7	38.6	38.0	33.9	28.6	29.7	29.4	28.9	30.4	31.3	32.4	33.9	34.4	34.0	34.1 34.5	55.7	36.8	38.3	38.4
e for]	10	Slurry Conc.	(vo s/au)	2.15	36.5	37.1	37.7	38.3	37.4	37.2	37.8	37.5	6.76 0.86	37.4	36.3	38.7	37.4	36.7	37.6	38.1	58.2 20.4	30.4 37.9	37.9	39.4	39.3	38.0	36.0	35.3	34.5	34.9	35.2	36.0	35.6	35.4	36.2	36.7	36.8	37.3 38.0	30.5	37.8	37.0	37.2
Table		Space Velocity (I/hr-ba)		5.07 5	5 456	5,412	5,469	5,498	5,442	5,388	5,412	5,460	5 461	5,447	5,444	5,420	5,304	5,359	5,229	5,482	5,470	5,013	4,953	4,942	4,882	4,826	4,780	4,/0/ 4 925	4,919	4,877	4,817	4845	4815	4772	4693	4762	4737	4705 4702	4738	4688	4700	4949
mary		/elocity (fi/sec)	(concent)	0.66	0.65	0.64	0.65	0.65	0.64	0.64	0.64	0.65	0.65 0.65	0.64	0.64	0.64	0.62	0.63	0.62	0.65	C0.U	0.65	0.64	0.64	0.63	0.62	0.62	0.63	0.63	0.63	0.62	0.62	0.62	0.61	0.60	0.61	0.61	0.61	10.61	0.60	0.60	0.64
Sum		urge n Gas '	(11100	80.4 71 5	63.6	68.8	57.3	56.1	49.3	66.4	75.2	79.9	6.16 8.07	89.1	80.5	112.2	27.6	26.2	26.6	96.2 100 5	<.001 2.03	59.8	62.4	64.8	84.0	87.2	77.8	65.0	65.4	55.8	48.5	70.9	71.7	74.2	53.7	55.1	49.9	54.0 52 5	5.00	90.4	89.1	88.7
Data	-	eed K		66.9 101	108	1.25	5.88	88.	6.94	1.13	1.28	5.97 00	1.05	5.94	3.76	1.58	5.24	1.94	80.0	5.74	001	1.05	1.17	1.29	1.30	1.36	1.43	5.61	99.66	69.9	1.90	.80	5.81	3.86	1.33	1.13	1.26	1.25	. 53	1.55	1.39	9.60
D-2 –	ŕ	cycle Ka Jas I CFH) (H		CII.	020	023	054	085	064	013	029	050	7 7/0	047	058	032 4	212	246	178	084	080	790	035	031	982	949	927	9/8	004	008	002	964	946	921	935 4	958	950 4	935 2	000	964	696	113
Fable	÷	ed Ke		2 2		2	26 2	21 2	15 2	30 2	25 22	36 2	0 0 0 0	34	33 2	32 2	7 2	97 2	37 2	33	2 2	1 C1	33 2	21 2	1 16	10	1	1 0	30	7 2	79 2	000	31 1	32 1	78 1	75 1	73 1	26	1 1		33	34
Γ,	đ	es. Fe es. (VSG	(9	2 F		10 7	10 72	10 72	10 70	10 7.	10 72	60 1		10 7	C 10	1 60	11	10 49	07 48	60 60	~ ~ ~	60	2 60	22 60	.2 60	⁷ - 7 ²	10	10	10 7	10 70	10	10	10 7.	10 7.	10 65	10 6	10 67	10 06 67	00 00	10	10 6	10
		emp P.	- -		7 CC2	234 7	234 7	235 7	234 7	235 7	234 7	234 7	/ CC7	234 7	234 7	234 7	230 7	230 7	230 7	235 7		235 7	235 7	235 7	235 7	234 7	234 7	7 052	234 7	235 7	235 7	7 1 234	234 7	234 7	234 7	235 7	234 7	234 7	234 7	234 7	234 7	234 /
		ias T		anced	inced 5	inced	inced 2	meed 2	inced 2	meed 1	meed	anced	nced 5	inced 2	meed 2	meed	meed 2	meed	nced	anced	anced .	inced 2	meed 2	meed	nced	nced	anced .	inced 5	meed	meed	inced	inced 5	inced	mced 2	meed	inced 2	nced	anced	anced	mced 2	meed	nced .
	!	85 E #		58 Bali o Dala	0 Bala	il Bala	2 Bala	3 Bala	4 Bala	55 Bala	56 Bala	57 Balí	00 Dali. 00 Bala	10 Bala	'l Bala	77 Bala	78 Bal£	79 Balź	30 Balí	31 Bali	52 Bali 2 Dalo	14 Bala	35 Bala	36 Balí:	37 Balź	38 Balź	91 Bali 2 5 - 5	42 Bali 13 Bala	14 Bala	15 Balá	96 Balí	 Bala Bala 	9 Bala	10 Balá	11 Balá	12 Bala)3 Balź)4 Balí	shalls Date	19 Date 20 Bala	21 Balź	22 Bala
	ć	10 <u>1</u>		-10 SV-/		-98 16	98 16	98 16	98 16	98 16	-98 It	-98 IC	96 IG	98 17	98 17	-98 17	-98 17	-98 17	-98 15	-98 15	-98 IS	-98 18	-98 15	-98 15	-98 15	-98 15	-98 -15	21 86- 98 19	98 15	98 15	98 IS	98 19	98 19	98 20	98 20	-98 2(-98 2(-98 2t	-98 21	-76 22	-98 22	-98 22
		Date		20 Mar	29-May	30-May	31-May	1-Jun-	2-Jun-	3-Jun-	4-Jun-	5-Jun-	-unr-0 -unl-2	-June -8	9-Jun	15-Jun-	16-Jun	17-Jun	18-Jun	19-Jun 20 Jun	unf-07	111-12 22-Jun-	23-Jun-	24-Jun	25-Jun	26-Jun	29-Jun	-102 1-101-0	2-Jul-	3-Jul-9	-Jul-	-mr-c	5-Iul-7	?-lul-?	9-Jul-9	10-Jul-	-11-Jul-	12-Jul-	13-Jul-	-Jul-28-Jul-	-Jul-	30-Jul-
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Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

Final Report: Volume 2 (June 2003)

D-10

		Days				Tab	le D-3 Recycle	2 – Da	ita Sur Purge	nmary Inlet Sup.	/ Tabl	e for]	Kings	port (Cataly: catalyst	st Cai Catalyst	npaig	gn 2 (c teactor s-T-M s	ont'd	aw foOH Cat	alyst	Reactor	U Dverall S ₁	S	parger
Case	Date	On Stream	Gas Type	Temp (Deg C)	Pres. (psig)	Feed (KSCFH)	Gas (KSCFH)	Feed (H2:CO)	Gas (KSCFH)	Velocity (ft/sec)	Velocity (l/hr-kg)	Conc. (wt% ox)	Holdup (vol%)	Slurry Hgt (ft)	Inventory (Ib)	Age (eta)	Conv. (%)	Jonv. (%) (S	CF/Ib) (1	rod. MeOl PD) (gmol	H Prod. Vhr-kg)	Vol. Prod. (TPD/ft3) h	(Btu r ff2 F)	, db (jst) F	tesist. "K")
10	13-Nov-98	328	Balanced	235	107	735	2,028	4.43	101.0	0.65	3443	44.6	35.2	55.0	47,400	0.32	38.5	21.7	41.9 2	10.4 11	.58	160.0	182 1	4.05	5.75
10	14-Nov-98	329	Balanced	235	695	761	2,134	1.88	91.4	0.68	3590	8. s	40.7	59.5 20.5	47,400	0.33	24.7	21.9	41.7 2	19.1 12	.06	0.087	162	2.13	16.68
9 9	15-Nov-98	330	Balanced	235	694	0//	2,075 1.090	1.88	100.1	0.68	3540 3302	45.3	40.3 20.0	58.0	47,400 47,400	0.33	25.4 26.7	22.4	41.5 2	22.5 12	2.25	0.091	167 2	1 00	12.79
2 2	17-Nov-98	166	Balanced Balanced	235	693	90/ 200/	2.031	1.91	110.1	0.06	3466	45.6	37.8	55.0	47,400 47,400	0.32	24.9	21.9	42.6 2	13.9 11		0.092	164 2	2.28	18.53
10	18-Nov-98	333	Balanced	235	692	730	1,967	1.91	98.3	0.65	3396	45.7	36.8	54.0	47,400	0.30	24.8	21.7	42.4 2	06.5 11	38	0.091	160 2	2.51	19.16
10	19-Nov-98	334	Balanced	235	695	693	2,111	1.94	52.6	0.66	3326	44.2	37.5	60.5	49,600	0.34	25.4	21.8	39.5 2	04.5 10).76	0.080	147 2	4.56	18.94
10	20-Nov-98	335	Balanced	235	695	692	2,066	1.89	77.3	0.67	3363	44.7	37.8	59.5	49,600	0.34	26.8	23.8	40.1 2	29.9 12	60.	0.092	163 2	5.56	17.21
10	21-Nov-98	336	Balanced	235	669	707	2,135	2.04	56.1	0.67	3370	46.3	35.2	54.0	49,600	0.32	26.8	22.5	39.3 2	16.1 11	.36	0.095	165 2	1.80	17.46
10	22-Nov-98	337	Balanced	235 235	708 705	745 753	2,060 2.055	2.15 2.30	66.5 87.1	0.65 0.66	3311 3327	45.3 45.3	36.9 34.0	57.5 55.0	49,600 49,600	0.32 0.31	28.8 29.0	23.4 22.5	40.2 2 41.2 2	22.2 11 19.4 11		0.092 0.095	166 2 164 1	0.05 9.76	17.25 17.79
2 2	25-Nov-98	340	Balanced	235	709	655	1.895	3.92	63.2	0.59	3035	45.7	33.1	53.5	49 600	0.29	40.5	21.3	41.0 1	91.8 10	.15	0.085	181	4.01	20.25
10	26-Nov-98	341	Balanced	235	709	728	1,956	3.31	95.2	0.62	3185	44.9	32.9	55.0	49,600	0.29	36.3	21.8	42.1 2	07.3 10	.95	0.090	167 1	5.52	19.10
10	27-Nov-98	342	Balanced	235	695	764	1,997	2.10	109.4	0.65	3268	45.9	34.3	54.0	49,600	0.30	26.7	22.1	42.5 2	15.6 11	.37	0.095	164 2	0.46	19.16
10	28-Nov-98	343	Balanced	235	695	753	1,990	2.07	111.7	0.65	3248	46.1	34.2	53.5	49,600	0.29	26.0	21.7	42.9 2	10.5 11	.12	0.094	160 2	0.64	19.46
10	29-Nov-98	344	Balanced	235	695	750	1,969	2.06	118.8	0.64	3223	46.1	33.6	53.0	49,600	0.28	25.5	21.4	43.6 2	06.6 10	.92	0.093	156 2	0.62	19.79
10	30-Nov-98	345	Balanced	230	696	520	2,117	2.08	27.1	0.61	3116	46.1	31.2	51.0	49,600	0.27	22.7	18.3	38.4 1	62.3 8.	.55	0.076	181 1	5.59	19.61
10	2-Dec-98	347	Balanced	230	695	501	2,027	3.73	51.0	0.58	2959	46.0	27.4	48.5	49,600	0.24	31.2	17.0	40.0 1	50.4 7.	.92	0.074	166 1	3.57	20.20
10	5-Dec-98	350	Balanced	231	700	515	1,850	4.75	28.0	0.55	2702	45.6	32.1	55.0	51,800	0.33	41.8	18.7	39.0 1	58.6 7.	98	0.069	161	0.96	20.63
10	6-Dec-98	351	Balanced	232	700	696 202	1,883	3.32	86.5	0.60	2921	45.6	32.1	55.0 55.0	51,800	0.29	36.4	21.7	42.0 I	98.7 10	0.10	0.086	161 1	5.47	20.91
10	7-Dec-98	352	Balanced	235	703	735	1,851	3.43	113.2	0.60	2932	46.2	33.9	55.5 52.0	51,800	0.28	37.6	22.1	43.4 2 c	02.9 10	0.26	0.087	151 1	5.45	21.14
2 2	8-Dec-98 9-Dec-98	252 254	Balanced Balanced	235	2017	741	1.935	3.27	112.6	0.62	3018	45.7	28.9	52.5	51 800	0.28	35.9	21.9	43.2 2	05.9 10	25.0 (42	0.093	176 1	5.96	+c.02
10	10-Dec-98	355	Balanced	234	710	712	1,920	3.40	102.6	0.61	2974	45.5	29.8	53.5	51,800	0.27	36.6	21.6	42.9 1	99.2 10	0.14	0.089	177 1	5.47	20.41
10	12-Dec-98	357	Balanced	235	710	750	1,893	3.69	88.2	0.61	2850	45.4	30.4	56.5	54,000	0.30	42.0	23.0	42.8 2	10.0 10	.37	0.088	171 1	5.69	21.11
10	13-Dec-98	358	Balanced	235	710	732	1,882	3.84	84.8	0.60	2826	45.1	30.3	57.0	54,000	0.29	42.7	22.9	42.2 2	08.2 10	0.15	0.087	171 1	5.17	21.26
10	14-Dec-98	359	Balanced	235	709	727	1,900	3.77	84.6 84.4	0.61	2843 2015	46.1 46.1	30.4	55.0 53.5	54,000	0.29	41.4 1 2	22.7	42.3 2	06.4 10 07.7 0).04 66	0.089	176 1	5.60	21.32 20.50
9	15-Dec-98	360	Balanced		110	000	1,027	10.0	t. to	0,.00	CT07	1.04	C 07		54,000	07.0	41.4	0.77	4 T T			0.000	1 101	t/ t	(C.U.
01 01	25-Dec-98 26-Dec-98	370 371	Balanced Balanced	235 235	705	753	1,995	3.58 3.81	89.3 95.6	0.64	2890 2844	47.2	29.9 28.9	53.5 54.0	56,200 56,200	0.29 0.29	39.8 41.6	22.3 22.4	42.4 2 42.3 2	15.0 10 13.5 10	0.17	0.096 0.094	180 1	5.38	20.28 19 63
10	27-Dec-98	372	Balanced	235	709	746	1,937	4.09	92.5	0.62	2817	45.7	29.2	57.2	56,200	0.29	43.8	22.4	42.3 2	11.5 9.	.93	0.088	175 1	5.38	19.25
10	28-Dec-98	373	Balanced	235	710	752	1,926	4.06	104.6	0.62	2802	45.3	25.7	55.5	56,200	0.28	43.2	22.4	42.9 2	10.2 9.	83	0.090	177 1	4.23	19.56
10	29-Dec-98	374	Balanced	235	709	753	1,901	4.06	113.9	0.62	2791	45.8	26.5	55.0	56,200	0.28	42.7	22.1	43.7 2	06.8 9.	.68	0.089	179 1	4.08	19.52
10	31-Dec-98	376	Balanced	235	710	160	1,940	4.73	108.3	0.63	2828	45.1	29.7	59.0	56,200	0.28	41.6	22.4	43.1 2	11.8 9.	.93	0.085	179 1	4.67	19.45
9 9	1-Jan-99 2-Jan-99	378	Balanced Balanced	235	710	0C/ 266	1,904	3.48	102.3	0.62	2819 2819	46.1	31.2	58.0 58.0	56,200 56,200	0.28	39.1 39.1	22.6 22.6	42.7 2	15.2 10 15.2 10	.10	0.088	179 1	5.50	0C.VI
9	3- Ian-00	370	Balanced	235	710	778	1 915	3.52	107.5	0.63	2835	47.6	318	555	56,200	0.28	39.5	22.8	12.8	18.4 10	1.23	0.094	184	5 16	77 61
2 2	4-lan-99	380	Balanced	235	710	0/1	1.952	3.47	107.5	0.64	2867	48.6	34.3	55.5	56.200	0.28	38.8	22.3	42.9 2	15.7 10	24	0.092	180	5.53	19.68
10	5-Jan-99	381	Balanced	235	710	735	1,909	3.97	132.1	0.61	2767	49.0	35.4	55.5	56,200	0.25	39.4	21.0	44.7 1	97.5 9.	.18	0.085	165 1	4.01	20.78
10	6-Jan-99	382	Balanced	235	710	773	1,848	3.88	137.5	0.61	2745	47.5	31.6	55.5	56,200	0.27	41.1	22.2	44.7 2	07.4 9.	-67	0.089	186 1	3.97	20.97
10	7-Jan-99	383	Balanced	235	710	752	1,949	3.43	116.1	0.63	2837	47.2	30.9	55.5	56,200	0.25	36.6	21.6	43.6 2	07.1 9.	.66	0.089	172 1	6.52	21.02
10	8-Jan-99	384	Balanced	235	710	749	1,949	3.38	124.2	0.62	2822	45.9	29.4	57.0	56,200	0.25	35.9	21.3	44.1 2	03.7 9.	.51	0.085	167 1	6.30	20.67
10	9-Jan-99	385	Balanced	235	710	741	1,972	3.47	123.9	0.63	2851	45.6	27.3	56.0	56,200	0.24	35.5	20.9	44.0 2	02.0 9.	.40	0.086	170 1	6.06	20.79
10	10-Jan-99	386	Balanced	235	710	741	1,983	3.32	131.6	0.63	2865	46.7	30.8	56.5	56,200	0.24	33.9	20.7	44.4 2	00.3 9.	.31	0.084	165 1	6.61	20.71
10	11-Jan-99	387	Balanced	235	209	663	1,970	3.88	98.1	0.61	2766	47.0	27.1	53.0	56,200	0.24	37.3	19.8	43.0 1	84.8 8.	.71	0.083	190	5.50	21.00

D-11

	rrger Sparger	IP Resist. ssi) ("K")	.95 21.21	21.86	1.27 21.19	.58 20.37	.22 20.49	.74 20.60	1.92 20.57	.27 22.25	1.21 24.50	18 25.25	.84 23.37	93 24.54 36 33.60	20.22 02.	21.89	21.86	112 21.72	1.87 21.70	1.19 21.28	.06 21.74	1.99 23.65	1.81 23.46	.43 23.16	:43 23.26	.76 23.36	1.49 23.40 1.23 24.21	17.12 07.1	.20 21.26	17 4.23	15 4.29	20 4.33	0/ 4.29 56 4.19	96 4.14	33 4.07	39 4.15	34 4.12	39 4.02	56 4.13	26 4.14	.09 4.15		.04 4.18
:	verall Spi	(Btu (J)	175 14	169 15	168 1/2	182 13	164 15	161 12	166 12	171 12	171 12	171 12	160 13	9 971 180	1/16 1/16	164 14	172 1/2	172 15	180 1/2	164 15	173 15	173 12	172 1/2	165 15	179 15	174 15	175 1/ 168 1/	1 177 15	168 16	190 3	193 3	193 3	176 3	180 3	175 4	187 4	171 4	182 4	184 4	178 4	181 4	160	100
	Reactor C	ol. Prod. (PD/ft3) hi	0.081	0.076	0.073	0.073	0.073	0.073	0.075	0.060	0.066	0.066	0.066	0.067	0.060	0.073	0.071	0.074	0.073	0.072	0.074	0.070	0.073	0.072	0.076	0.077	0.073	0.064	0.073	0.042	0.044	0.044	0.053	0.056	0.061	0.062	0.062	0.058	0.057	0.059	0.060		0.00/
	ıtalyst	DH Prod. V bl/hr-kg) (8.37	8.23	8.10	8.53	8.56	8.50	8.49	7.02	7.00	7.02	5.87	5.58	5.68	7.24	5.80	5.67	6.69	5.85	5.98	5.67	5.90	10.7	7.43	7.32	7.56	5.83	8.10	60.9	5.17	5.01	5.96	7.05	7.30	7.37	7.19	7.32	7.19	5.92	5.78		18./
	son c	od. Me(PD) (gm	8.3	5.8	9.4	4.8	6.6	8.8	8.5	3.2	2.7	3.4	6.6	- 5	1.0	84	8.1	4.8	5.4	9.2	2.3	4.9	0.3	3.8	3.7	13	6.9 0.6	2.6	7.4	6.5	8.1	5.0	20 20	1.0	5.9	17.3	3.7	6.3	13.6	8.1	5.3		8.1.8
`	gas Me	il. Pr //b) (T	5 17	.8	9 17	.3 18	.7 18	.6 18	.2 18	.7 16	.7 16	9 16	.7 15	21 51 21	5 F	191	2 15	2 15	.1 15	.3 15	9 16	9 15	.5 16	.6 15	.3 15	4	-7 14 9 14	0	61 51 51 15	.3 11	.5 11	LI	0.0	1	.3 14	.3 14	.3 14	9 14	.7 14	.5 13	.4 13		.6 16
,	M Syn	. Uti	42	43.	41.	43.	43.	43.	44.	39.	39.	38	38	- 40		40	4	41	41.	42	42	41.	43.	45.	46.	46	45.45	52	2 4	50.	49	. 51	c 77	45.	51.	51.	51.	50	51.	54.	55.		45.
)	O-T-I	nv. Conv () (%)	6 19.2	6 19.1	5 20.0	7 20.5	7 20.5	3 20.5	6 20.5	0 19.3	7 20.2	8 21.1	4 18.6	20.5	2.02 5	4 191	1 17.7	4 17.4	8 17.1	9 16.8	5 17.0	4 17.0	9 18.0	3 17.2	.1 17.0	7 16.6	6 16.3 3 15.8	9 153	8 17.2	3 12.6	6 12.9	2 12.4	- 12-4 5 13-6	4 13.5	2 13.5	5 13.7	0 13.3	5 13.5	9 13.1	5 12.9	9 12.8		4 15.5
•	lyst C(ge Cor a) (%	22 35.	22 35.	23 41.	24 43.	3 37.	38. 38.	23 37.	22 42.	22 43.	21 38.	21 36	22 249	1 77 1 77	35 04	9 36	8 34.	9 33.	8 30.	8 31.	8 33.	8 33.	8 31.	9 30	9 28	9 30. 8 30.	12 0	8 28	4 19.	4 19.	19	- 4C - C	7 22	7 20.	7 20.	6 20.	6 19.	6 18.	5 19.	5 19.		9 27.
,	st Cata	ry Ag (et	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.7	0.7	0.2	7.0	0.0	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	10	0.1	0.1	0.1	0.1		0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1		0.1
	Catalys	Invento (Ib)	56,200	56,200	58,400	58,400	58,400	58,400	58,400	60,600	60,600	60,600	60,600	60,600	000'000 60 600	60.600	60.600	60,600	60,600	60,600	60,600	60,600	60,600	57,303	53,951	53,951	50,676	50.676	50,676	49,988	49,988	49,988	49,700 57 188	52,188	52,188	52,188	52,188	52,188	52,188	52,188	52,188		54,388
l	Gassed	Slurry Hgt (ft)	52.5	55.0	58.5	61.5	61.5	61.5	60.0	64.5	58.5	59.0	58.0	54.5	52.5	545	53.0	50.0	50.5	52.5	52.0	52.5	52.0	51.0	48.5	47.0	48.0 47.5	40.0	51.5	65.5	63.0	62.5	c 10 5 <i>C</i> 9	59.5	56.5	56.5	55.5	60.09	60.0	56.0	53.5		C.7.C
	Gas	Holdup (vol%)	24.9	28.2	28.4	29.3	28.5	29.2	26.7	24.7	23.7	25.0	25.8	21.7	C C C C	272	22.6	21.4	22.5	24.4	24.6	24.2	22.2	25.7	24.7	25.2	25.8 25.5	255	26.4	31.1	30.5	31.4	30.62	30.1	30.5	30.4	31.9	31.8	31.9	30.2	27.5		28.7
	Slurry	Conc. (wt% ox)	46.5	46.4	45.8	44.8	44.5	44.8	44.5	42.9	45.1	45.4	46.1	46.4	17.1	47.3	47.4	48.6	48.7	48.4	48.7	48.3	47.8	48.1	47.5	48.5	46.6 46.7	45.8	6.6 4	39.8	40.6	41.1	42.0	43.1	44.6	44.6	45.6	43.5	43.5	44.7	44.9		44.6
	Space	Velocity (l/hr-kg)	2743	2722	2605	2604	2659	2636	2635	2397	2275	2118	2356	2013	2430	7447	2456	2455	2488	2566	2560	2451	2415	2579	2744	2773	2904 2870	2775	3008	3019	2985	3008	3163	3296	3390	3381	3387	3397	3421	3348	3323		3187
	Inlet Sup.	Velocity (ft/sec)	0.61	0.60	0.60	0.60	0.61	0.61	0.61	0.57	0.54	0.51	0.56	0.48	0.50	0.58	0.59	0.59	0.59	0.61	0.61	0.58	0.58	0.58	0.58	0.59	0.58 0.58	0.55	0.61	0.59	0.59	0.59	0.65	0.68	0.70	0.69	0.70	0.70	0.40	0.69	0.68	0 00	0.68
	Purge	Gas SCFH)	91.8	103.2	71.7	78.6	100.4	103.5	113.5	30.5	33.6	33.0	24.5	34.8	C. C.2	515	52.7	51.0	51.0	64.0	71.2	58.1	80.5	103.2	113.6	109.8	105.6 112.6	156.3	100.0	123.0	122.9	131.3	84.9	93.5	173.7	175.2	172.7	170.6	174.9	199.0	205.3	0.00	7.01
	cactor	Feed 2:CO) (K	3.81	3.83	4.42	4.43	3.77	3.86	3.77	4.64	4.60	3.78	4.02	1.82	1 30	27.5	4.20	4.02	4.05	3.75	3.76	4.01	3.87	3.72	3.58	3.50	3.88 4 01	1.43	4.45 3.35	3.02	2.99	3.04	20.0 2 63	3.36	2.92	2.94	2.94	2.78	2.76	2.96	3.10		5.60
	cycle R	Gas SCFH) (H	982	994	.950	. 940	948	935	923	. 631	802	648	926	542	010	040	985	994	014	062	039	976	904	926	939	956	942	783	010	066	029	061	160	401	370	370	379	385	393	331	310		775
	sh Re	d (KS FH) (KS	2	2	6 1	9 1	1 1.	5 1	4 1	9 1	8	9	5		+ -			1	3	8	1 2	1	1	5 1	4	1	7 6 1		3 2	8 2	7 2	2 2 2 7	0 - 1 C	0	3	0 2	4	1 2	8	8	5 2	·	0
	Fre	s. Fe g) (KSC	0 63	0 64	0 62	0 63	69 6	0 68	9 69	1 53	0 53	0 52	9 51	- 0	1 02	0 29	0 54	0 53	1 53	0 55	9 58	0 54	0 58	0 58	0 59	9 58	9 55 9 54	0 57	0 28 85	0 48	0 48	0 49	- 1 - 49 - 53	0 53	0 62	0 63	0 61	0 62	0 61	1 62	1 62		10
		mp Pre g C) (psi	35 71	35 71	35 71	35 71	35 70	35 71	35 70	35 71	35 71	35 71	35 70	55 71	11 CC 21	35 71	35 71	35 71	35 71	35 71	35 70	35 71	35 71	35 71	35 71	35 70	37 70 37 70	11 12	1/ 01 01	35 71	35 71	35 71	11 23	35 71	35 71	35 71	35 71	35 71	35 71	35 71	35 71	10	1/ 00
		e (De	ced 2	ced 23	ced 2:	ced 2:	ced 2:	ced 23	ced 2:	ced 2:	ced 23	ced 23	ced 2.	ced		eed 2	ced 2	ced 23	ced 2:	ced 2:	ced 25	ced 2	ced 2:	ced 2:	ced 2	ced 23	ced 20	red 20	ced	ced 20	ced 2:	ced .	red 2.2	ced 2	ced 2	ced 2	ced 23	ced 2	ced 2:	ced 2:	ced 2	, c	ced pao
		n Typ	Balanc	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Balanc	Balanc	Balanc	Balane	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Balan	Baland	Balan	Balan	Balan	Balan P. 1	Baland	Balanc	Balanc	Balan	Balan	Balan	Balan	Balan	Balan	-	Balanc
	Days	On Strear	388	, 389	391	392	394	395	396	405	406	408	410	412	214	014 114	417	418	419	, 420	, 421	, 422	423	424) 425) 426	427	428	, 450 , 435	9 451	9 452	9 453	454 6	457	9 458	9 459	9 460	9 461	9 462	9 463	9 464		, 405
		Date	12-Jan-99	13-Jan-99	15-Jan-99	16-Jan-99	18-Jan-99	19-Jan-99	20-Jan-99	29-Jan-99	30-Jan-99	1-Feb-99	3-Feb-99	5-Feb-99	0-FCD-99	0-Eah 00	10-Feh-99	11-Feb-99	12-Feb-99	13-Feb-99	14-Feb-99	15-Feb-99	16-Feb-99	17-Feb-99	18-Feb-99	19-Feb-99	20-Feb-95	21-Feb-95 22 Eab 00	23-Feb-95 28-Feb-99	16-Mar-95	17-Mar-99	18-Mar-99	71-Mar-95	22-Mar-95	23-Mar-95	24-Mar-95	25-Mar-95	26-Mar-95	27-Mar-95	28-Mar-95	29-Mar-99	00 11 00	50-Mar-95
		Case	10	10	10	10	10	10	10	10	10	10	10	0 9	0 9	2 2	01	10	10	10	10	10	10	11	Ξ	Ξ	= =	:	= =	10	10	9 9	2 2	10	10	10	10	10	10	10	10	01	2

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

						Tal	ble D-	- 2 – D ^a	ita Su	mmar	y Tab]	le for	King	sport	Cataly	st Ca	mpai	gn 2 (cont'	(p					
		Dave				Fresh	Recorde	Reactor	Puroe	In let Sun	Shace	Slurry	Gas	Gased	Catalvet	Catalvet	00	Reactor O-T-M	Svnøgs	Raw MeOH	Catalvet	Reactor	U Overall	Snaroer	Snaroer
ase)	Date	On	Gas	Temp (Deg ()	Pres.	Feed	Gas	Feed (H2-CO)	Gas	Velocity (#/sec)	Velocity (I/hr-ka)	Conc.	Holdup (%)ov)	Slurry Hot (ft)	Inventory (Ib)	Age (eta)	Conv.	Conv.	Util. (SCF/h)	Prod.	MeOH Prod.	Vol. Prod. (TPD/#3)	(Btu hr ft? F)	dP (isu)	Resist.
2000	202	IIII	~d(; *	(2 927)	(and)	(111001)	(11001)		(magain)	(and an)	(Qu mm)	(vo a/am)	(6/104)	(m) 1911	(m)	(mo)	677	(0/)	(01100)	(211)	(9 m m m m m m m m m m m m m m m m m m m	(2004 11)	((md)	(*)
10	2-Apr-99	468	Balanced	235	710	650	2,261	3.57	137.8	0.67	3147	45.3	26.1	54.0	54,388	0.19	27.5	15.8	47.9	162.9	7.84	0.072	175	3.97	4.19
9	3-Apr-99	469	Balanced	235	710	909	2,236	4.38	84.5	0.66	2953	0.44	27.3	61.0	56,588	0.21	34.7	16.9	43.3	167.8	7.74	0.065	161	3.67	4.22
0 9	4-Apr-99	470	Balanced	235	210	669 553	2,201	4.24	1.801	0.66	2101	44.1	26.6	5.9.0 5.2 5	88C,0C	0.21	33.2	17.6	44.6 14.5	1.751	8.11	0.070	101	3.73	4.25
0 0	6-Apr-00	4/1	Balanced	235	710	651	2.218	4.09	103.9	0.00	2985	C.C.4 8.44	26.5	5.6.5	56.588	0.21	33.6	17.3	0 1 4	174.9	8.06	0.074	181	3.75	4.20
10	7-Apr-99	473	Balanced	235	710	654	2,225	4.07	109.3	0.67	2990	45.6	29.2	58.0	56,588	0.21	33.3	17.2	45.0	174.2	8.04	0.071	177	3.93	4.38
10	8-Apr-99	474	Balanced	235	710	652	2,188	4.07	120.8	0.66	2940	45.5	29.0	58.0	56,588	0.20	33.0	17.1	46.2	169.6	7.82	0.069	171	3.88	4.41
10	9-Apr-99	475	Balanced	235	710	645	2,149	3.93	122.2	0.65	2902	45.9	30.1	58.0	56,588	0.20	32.6	17.4	45.6	169.9	7.84	0.070	181	3.90	4.44
10	10-Apr-99	476	Balanced	235	710	641	2,181	3.98	117.8	0.65	2914	47.0	29.8	55.5	56,588	0.20	32.4	17.0	45.7	168.5	7.82	0.072	176	3.75	4.41
10	11-Apr-99	477	Balanced	235	607	653	2,199	3.72	104.0	0.66	2954	47.5	31.0	55.5	56,588	0.21	32.1	17.8	43.9	178.6	8.24	0.077	185	3.92	4.37
10	12-Apr-99	478	Balanced	235	607	651	2,249	3.80	116.0	0.67	3001	47.8	31.9	55.5	56,588	0.20	30.7	16.8	45.5	171.7	7.92	0.074	179	3.91	4.38
10	13-Apr-99	479	Balanced	234	709	651	2,263	3.48	108.6	0.67	3025	47.6	30.7	55.0	56,588	0.20	29.1	17.0	44.7	174.9	8.06	0.076	184	4.24	4.43
10	14-Apr-99	480	Balanced	235	710	535	2,304	3.89	45.5	0.66	2954	47.1	27.4	53.5	56,588	0.19	30.3	16.0	40.4	158.8	7.32	0.071	161	4.13	4.51
10	15-Apr-99	481	Balanced	235	711	535	2,303	3.80	44.2	0.65	2943	46.7	27.7	54.5	56,588	0.19	28.4	16.2	39.9	160.8	7.41	0.070	191	4.21	4.53
10	16-Apr-99	482	Balanced	235	710	510	2,313	3.90	48.4	0.65	2942	47.1	28.2	54.0	56,588	0.18	28.2	15.0	41.0	149.4	6.88	0.066	178	3.83	4.48
10	3-May-99	499	Balanced	235	111	554	2,234	3.39	69.7	0.65	2898	45.7	29.4	58.0	56,588	0.17	26.7	15.7	42.7	155.8	7.19	0.064	184	4.38	4.83
10	4-May-99	500	Balanced	235	711	601	2,156	3.21	106.7	0.64	2874	46.2	30.9	58.0	56,588	0.18	26.2	16.2	45.2	159.5	7.36	0.065	188	4.43	4.89
10	5-May-99	501	Balanced	235	710	209	2,122	3.34	110.1	0.63	2844	45.1	27.9	58.0	56,588	0.18	27.2	16.4	45.3	160.6	7.41	0.066	191	4.16	4.90
01	6-May-99	502	Dallalloou	C C 7	017	010	000.4	07.0	1.102	0.00	1400	40.0	1.62	0.40	20,200	61.0	7.07	0.01	/.++	6.001	00.1	0.072	107	4.45	67.4
01 01	8-May-99	504	Balanced	252	710	180	2,219	18.7	1.88	C0.U	2934	46.5	5.62	C.0C	56,588 56,500	0.18	74.7	8.CI	43.7	0.161	7.45	0.068	C 101	4.75	4.05 62 4
0 0	9-May-99	505	Balanced Balanced	C62 735	710	/8C	2,198	3.13	8/.6 103 1	0.64 0.64	29162	40.7	28.5	0.00 53.5	56,588 56,588	0.18	24.7	15.8	45.4 44.8	158.9	7 3.4	0/0/0	197	62.4 6.2.4	4 56
01	11-May-90	205	Balanced	235	210	965	2,121	60 E	9 66	0.63	2846	46.2	LLC	5.55	56,588	0.18	25.6	16.2	0 8 4	159.5	1 37 F	0.068	193	4.06	4 59
10	12-Mav-99	508	Balanced	235	709	592	2,143	3.01	99.2	0.64	2858	46.6	27.4	54.5	56,588	0.18	24.9	16.1	44.8	158.9	7.33	0.069	193	4.07	4.52
10	13-May-99	509	Balanced	235	709	581	2,177	2.92	99.2	0.64	2873	46.6	27.4	54.5	56,588	0.17	23.8	15.8	44.2	157.6	7.27	0.069	187	4.07	4.52
10	14-May-99	510	Balanced	235	710	585	2,161	2.82	104.7	0.64	2891	47.4	27.6	53.0	56,588	0.17	23.3	15.7	44.5	157.7	7.32	0.071	193	4.15	4.47
10	15-May-99	511	Balanced	235	607	588	2,179	2.73	107.1	0.64	2889	47.5	27.7	53.0	56,588	0.171	22.5	15.7	45.0	157.0	7.25	0.071	184	4.21	4.46
10	16-May-99	512	Balanced	235	40 <i>L</i>	588	2,177	2.74	111.0	0.65	2890	47.8	27.2	52.0	56,588	0.169	22.4	15.6	45.2	155.9	4.20	0.071	183	4.20	4.46
Ξ	18-May-99	514	Balanced	235	209	546	2,191	2.69	107.0	0.64	3048	47.9	26.3	48.5	53,494	0.170	21.1	14.9	44.5	149.3	7.30	0.073	188	4.16	4.44
= :	19-May-99	515	Balanced	235	710	566	2,193	3.72	63.8	0.64	2907	46.9	27.4	53.0 23 î	55,694	0.206	30.7	16.9	41.1	165.5	7.75	0.074	186	3.75	4.57
= =	21-May-99	110	Balanced	235	210	554	2728	5.0.5 71.5	4.10 8.89	0.04	2690	47.2	112	52 O	55 694	0.12.0	6.00 8.75	16.6	40.3	164.8	00''	0.075	186	5.72 4.15	4 37
=	23-May-99	519	Balanced	234	710	558	2,207	3.10	48.6	0.64	2921	47.5	28.0	52.0	55,694	0.198	26.9	16.7	40.8	164.1	7.69	0.075	190	4.20	4.38
Ξ	24-May-99	520	Balanced	235	709	560	2,208	3.06	52.0	0.64	2927	46.8	28.8	54.0	55,694	0.197	26.9	16.8	40.5	165.8	7.77	0.073	194	4.26	4.44
Ξ	25-May-99	521	Balanced	235	710	564	2,182	3.24	63.5	0.63	2896	46.8	28.2	53.5	55,694	0.193	27.4	16.6	41.8	162.3	7.60	0.072	190	4.01	4.45
Ξ	26-May-99	522	Balanced	235	710	566	2,187	3.21	68.3	0.64	2909	47.3	28.0	52.5	55,694	0.192	26.8	16.4	42.1	161.3	7.56	0.073	188	3.98	4.45
Ξ	27-May-99	523	Balanced	235	710	569	2,195	3.24	51.0	0.64	2907	46.5	29.3	55.0	55,694	0.202	28.0	16.9	40.9	166.9	7.82	0.072	161	3.96	4.44
Ξ	28-May-99	524	Balanced	235	710	561	2,204	3.19	50.3	0.64	2922	46.3	28.1	54.5	55,694	0.198	27.3	16.7	40.7	165.4	7.75	0.072	190	3.97	4.40
Ξ	29-May-99	525	Balanced	235	710	565	2,155	3.32	55.5	0.63	2878	46.2	27.3	54.0	55,694	0.203	28.3	16.8	41.2	164.5	7.73	0.072	192	3.78	4.43
Ξ	30-May-99	526	Balanced	234	710	566	2,178	3.25	55.9	0.63	2883	46.5	27.3	53.5	55,694	0.205	27.3	16.7	41.3	164.4	7.72	0.073	195	3.84	4.43
Ξ	31-May-99	527	Balanced	234	710	562	2,172	3.11	54.4	0.63	2888	46.9	27.1	52.5	55,694	0.202	26.9	16.7	41.2	163.9	7.68	0.074	198	3.95	4.40
= :	1-Jun-99	528	Balanced	234	710	566	2,142	3.41	68.0	0.63	2854	46.9	26.4	52.0	55,694	0.202	28.4	16.6	42.1	161.2	7.55	0.074	193	3.65	4.41
= :	2-Jun-99	529	Balanced	235	710	562	2,129	3.43	72.4	0.62	2839	47.0	27.3	52.5	55,694	0.195	28.0	16.3	42.8	157.5	7.38	0.071	181	3.65	4.45
= =	3-Jun-99	530	Balanced	235	710	565	2,141	3.26	65.9	0.63	2858	46.9	27.6	53.0	55,694	0.196	27.5	16.6	42.1	160.9	7.54	0.072	187	3.79	4.41
=	4-Jun-99	531	Balanced	235	710	267	2,155	3.12	67.8	0.63	2877	46.8	26.1	51.0	55,694	0.194	26.4	16.5	42.3	161.0	7.55	0.075	161	3.90	4.41

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		Davs				Fresh	Recycle	Reactor	Purge	Inlet Sun	Shace	Shurry	Gas) passed	atalvst	atalvst	CO Ke	actor T-M Sv	Ra Neas Me	w OH Ca	talvst Re	actor	U verall Sns	rger Sha	røer
Case	Date	On Stream	Gas Type	Temp (Deg C)	Pres. (psig)	Feed (KSCFH)	Gas (KSCFH)	Feed (H2:CO)	Gas (KSCFH)	Velocity (ft/sec)	Velocity (l/hr-kg)	Conc. (wt% ox)	Holdup (vol%)	Slurry In let (ft)	(lb)	Age (eta)	Conv. C	%) (SC	Tel III Tel III Tel IIII	od. MeOl PD) (gmol	H Prod. Vol. (/hr-kg) (TP)	D/ft3) hr	Btu ff2 F) (r	P Re	C) ist
)								Ì) `)				
= :	5-Jun-99	532	Balanced	235	710	565	2,162	2.93	65.2	0.63	2885	47.5	26.6	51.0	55,694	0.190	25.2	6.4	2.1 16	1.1	.55 0.	075	186 4	02	5
= =	6-Jun-99 7 Jun 00	533	Balanced Balanced	235	11/	0.005	2,147 2,146	3.03 2.03	6.79 60.8	0.63	28/0	47.8 8.78	2.02	50.5	55.694 55.694	0.187	1 0.02	6.3 4 6.4 4	CI 1.2 21 1.2	1.6	49 0.	c/0 270	197 3.	95 4. 00 4	42
= =	90-mil-/	400 535	Balanced	235	710	200	2,154 2,154	2.89	0.20 69.4	0.63	2868	48.2	0.02		55.694	0.185	1 0 7 0 7	- T O T	CI 1.2		47 0.	074		06	2 %
Ξ	96-nul-9	536	Balanced	235	710	562	2,141	2.75	67.0	0.63	2865	46.2	25.2	52.5	55,694	0.185	24.2	6.4 4	2.1 16	0.1 7	.51 0.	073	196 4	17 4.	38
Ξ	10-Jun-99	537	Balanced	235	710	550	2,151	2.70	6.99	0.63	2863	46.4	26.5	53.0	55,694	0.180	23.2	6.0 4	2.4 15	5.6 7	.30 0.	070	191 4.	22 4.	37
11	11-Jun-99	538	Balanced	235	705	559	2,112	2.91	9.77	0.62	2823	47.2	27.1	52.0	55,694	0.181	24.5	6.1 4	3.5 15	4.2 7	.23 0.	071	186 3.	94 4.	38
Ξ	12-Jun-99	539	Balanced	235	704	554	2,093	3.51	86.8	0.62	2799	47.0	26.7	52.0	55,694	0.181	25.3	5.8 4	4.1 15	0.7 7	.07 0.	690	181 3.	72 4.	43
11	13-Jun-99	540	Balanced	235	706	559	2,140	2.75	82.0	0.63	2847	47.2	27.1	52.0 5	55,694	0.178	23.2	5.9 4	3.6 15	4.1 7	.23 0.	071	189 4.	07 4.	39
Ξ	14-Jun-99	541	Balanced	235	705	554	2,109	3.01	87.5	0.62	2816	47.3	27.3	52.0	55,694	0.177	24.3	5.7 4	4.2 15	0.5 7	.07 0.	690	189 3.	77 4.	6
Ξ	15-Jun-99	542	Balanced	235	705	507	2,130	3.21	56.6	0.62	2789	47.3	26.8	51.5	55,694	0.177	25.4 1	5.2 4	1.7 14	5.9 6	.84 0.	067	196 3.	71 4.	46
Ξ	16-Jun-99	543	Balanced	235	705	497	2,160	3.11	68.0	0.62	2953	48.0	27.8	48.5	53,074	0.169	23.0 1	4.4	3.2 13	8.1 6	.80 0.	068	192 3.	72 4.	4
Ξ	17-Jun-99	544	Balanced	235	705	478	2,172	3.28	69.7	0.62	2943	48.3	25.3	46.5	53,074	0.162	22.5	3.6 4	4.1 13	0.1 6	.40 0.	067	177 3.	58 4.	4
= :	19-Jun-99	546	Balanced	235	700	499 	2,109	4.17	32.1	0.60	2715	45.7	27.2	54.5	55,274	0.210	32.2	6.0	1.0	5.8	.89 0.	064	190 3.	16	: 23
= :	20-Jun-99	547	Balanced	235	00/	197	2,109	4.13	31.4	0.60	2/16	45.7	212	0.40 0.21	55,274	612.0 701.0	52.5	6.3 1.1 1	0.2 14	- r - r	-00 10 ⁻	005 070	200	16	5 ;
= =	23-Jun-99	550	Balanced	662	169	60C	160,2	5.17	0.0/	0.65	8282	C.04	512	0.50	4/7,00	0.197	1.12		2.0 IO	1.0	.0 65.	7/0	CSI 01	94 27	1
= :	24-Jun-99	195	Balanced	407	C60	coc	2,128	3.11	04.0 C 6	0.04	1/97	4/.0	C17	0.70	4/7,00	cu2.0	0.12	1.01		0.1	.0 .0.	0/4	104	- + ·	÷ ;
= =	25-Jun-99 26-Jun-99	552 553	Balanced Balanced	235 234	693 692	577 575	2,061 2,045	3.07 3.11	73.3 76.5	0.63	2816 2788	46.9 46.6	27.0 26.7	52.5	55,274 55,274	0.193 0.198	26.9 I	7.1 4 4	3.0 16 2.7 16	1.0 7 1.6 7	.61 .64 0.0	074 073	191 3. 205 3.	90 86 4.	49 45
Ξ	28-Jun-99	555	Balanced	235	705	569	2,081	3.47	81.5	0.62	2817	47.7	27.0	50.5	55,274	0.196	28.2	6.4 4	3.6 15	6.7 7	.41 0.	074	196 3.	45 4.	47
11	29-Jun-99	556	Balanced	235	702	547	2,108	3.01	62.0	0.62	2835	47.2	27.6	52.0	55,274	0.185	25.5 1	6.3 4	2.0 15	6.2 7	.38 0.	072	189 3.	92 4.	51
Ξ	30-Jun-99	557	Balanced	235	969	540	2,074	3.05	66.5	0.62	2797	46.8	25.8	51.5	55,274	0.182	25.8 1	6.4 4	2.0 15	4.4 7	.30 0.	071	194 3.	84 4.	50
Ξ	1-Jul-99	558	Balanced	234	694	536	2,059	3.25	74.2	0.61	2763	46.9	24.6	50.5	55,274	0.182	26.6 1	6.1 4	2.8 15	0.3 7	.10 0.	071	202 3.	64 4.	50
11	2-Jul-99	559	Balanced	234	692	550	2,050	3.12	86.9	0.62	2771	47.0	27.1	52.0	55,274	0.182	25.5	6.0 4	4.0 15	0.1 7	.10 0.	690	198 3.	68 4.	51
11	4-Jul-99	561	Balanced	235	692	497	2,070	3.21	57.7	0.61	2729	46.6	24.7	51.0 5	55,274	0.176	25.5 1	5.5 4	1.4 14	4.2 6	.81 0.	067	193 3.	56 4.	55
Ξ	5-Jul-99	562	Balanced	234	069	527	2,064	2.90	72.3	0.62	2754	47.0	25.8	51.0	55,274	0.178	23.9	5.8 4	2.7 14	8.1 7	.01 0.	690	194 3.	77 4.	54
Ξ	8-Jul-99	565	Balanced	235	689	536	2,036	3.37	93.3	0.61	2744	46.1	26.8	53.5	55,274	0.180	26.1 1	5.4 4	4.3 14	5.1 6	.87 0.	065	190 3.	44	90
Π	96-lul-9	566	Balanced	235	689	536	2,037	3.10	95.5	0.62	2747	46.7	26.3	52.0	55,274	0.177	24.2	5.2 4	5.1 14	2.7 6	.76 0.	065	189 3.	56 4.	56
Ξ	10-Jul-99	567	Balanced	235	690	543	2,066	3.01	9.66	0.62	2782	47.0	27.2	52.0	55,274	0.180	23.4	5.1 4	5.5 14	3.2 6	.77 0.	066	192 3.	61 4.	49
= :	11-Jul-99	568	Balanced	234	169	528	2,068	2.99	92.0 85.2	0.62	2771	47.4	26.0	50.5	55,274 47 228	0.172	23.2	5.1 4	4.4	2.7 6	.75 0.	067	198 3.	60	۹ ک
= =	72-101-61	0/5	Balanced	234	690	452	2,123	2.63	2.00 90.9	0.02	3223	48.2	32.4	46.0	47.338	0.155	17.9	2.7	5.4 II	9.5 6	.0. 0.	062	182 3. 9		21 2
11	15-Jul-99	572	Balanced	235	069	457	2,125	2.75	99.1	0.62	3237	47.6	30.9	46.0	47,338	0.167	18.3	2.5 4	6.8 11	9.5 6	.65 0.	062	181 3.	76 4.	52
Ξ	16-Jul-99	573	Balanced	235	690	505	2,027	3.67	43.6	0.61	3020	46.1	29.1	49.5	49,538	0.216	30.0	6.4 4	0.7 14	9.0 7	.86 0.	072	192 3.	43 4.	58
Ξ	17-Jul-99	574	Balanced	234	169	495	2,042	3.73	46.3	0.60	3020	46.2	28.9	49.0	49,538	0.208	29.5	6.0 4	0.8 14	5.4 7	.66 0.	071	189 3.	39 4.	58
Ξ	18-Jul-99	575	Balanced	234	690	500	2,020	3.93	53.6	0.60	2988	46.2	29.2	49.5	49,538	0.210	30.7	6.1 4	1.3 14	5.2 7	.65 0.	070	193 3.	21 4.	61
Π	19-Jul-99	576	Balanced	235	690	493	1,981	3.83	53.6	0.59	2935	46.3	29.5	49.5 4	49,538	0.204	30.8 1	6.4 4	1.0 14	4.2 7	.60 00.	690	189 3.	25 4.	60
Ξ	20-Jul-99	577	Balanced	235	690	492	2,057	3.59	47.0	0.61	3041	46.7	29.1	48.5	49,538	0.202	28.3	5.8 4	0.8 14	4.7 7	.63 0.	071	189 3.	49 4.	57
Ξ	21-Jul-99	578	Balanced	235	690	493	2,085	3.46	47.6	0.61	3064	46.7	29.3	48.5	49,538	0.199	27.3	5.7 4	0.9 14	4.7 7	.62 0.	071	192 3.	61 4.	56
Ξ	22-Jul-99	579	Balanced	234	690	488	2,070	3.34	47.5	0.61	3050	47.1	28.7	47.5	49,538	0.195	26.2	5.5 4	1.3 14	1.8 7	.48 0.	071	190 3.	66 4.	55
Ξ	23-Jul-99	580	Balanced	235	689	495	2,026	3.30	61.7	0.60	3002	46.2	28.6	49.0	49,538	0.196	25.6]	5.4 4	2.3 14	0.3 7	.40 0.	068	190 3.	4	57
Ξ	24-Jul-99	581	Balanced	234	690	505	2,054	3.26	70.7	0.61	3056	46.9	29.1	48.0	49,538	0.194	25.0 1	5.3 4	2.7 14	1.8 7	.47 0.	020	195 3.	51 4.	57
= =	25-Jul-99	582	Balanced	234 234	069	500	2,072 2,000	3.13	64.7 50.2	0.61	3059 2041	46.3 16.1	28.8 20.7	49.0	49,538 40,538	0.192	24.4	5.2	2.4 14 1.4 13	1.5 7.7	.46 0.	069	194 3.	59 4.	53
= =	26-Jul-99	583	Balanceu	+07	040	4/0	2,090	3.14 2.20	7700	10.0	5041	40.1	7.67	- 0.44	49,000 1000	0.190	24.1	4.7	CT 0'T	- <u>-</u>	-74 -72	000		+ +	5
Ξ	27-Jul-99	584	Balanced	234	689	518	2,069	3.28	115.1	0.62	3468	46.9	27.8	42.0	43,980	0.211	22.3	3.6 4	8.3 12	8.7 /	.0	073	198 3.	42 4.	54

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

									•			0	_	•		0 ²²	actor	Ra	*			D		
Days					Fresh	Recycle	Reactor	Purge	Inlet Sup.	Space	Slurry	Gas (Gassed (Catalyst C	Catalyst	0 CO	T-M Syı	ngas Me	OH Cata	alyst F	Reactor O	werall Spa	rger Spa	arger
On C (0 L	jas ype	Temp (Deg C)	Pres. (psig)	Feed (KSCFH)	Gas (KSCFH)	Feed (H2:CO) (Gas (KSCFH)	Velocity (ft/sec)	Velocity (l/hr-kg) (Conc. wt% ox)	Holdup (vol%) F	Slurry Ir Hgt (ft)	rventory (lb)	Age (eta)	Conv. Co	onv. U %) (SC	Jtil. Pro JF/lb) (TI	d. MeOH D) (gmol/	H Prod. V(/hr-kg) (T	ol. Prod. ((PD/ft3) hr	(Btu d : ft2 F) (p	P Re	sist. K")
648 Bi	ä	alanced	235	710	803	2,003	2.81	105.1	0.65	3565	42.9	31.8	54.5	46,488	0.346	34.2 2	3.3 4	1.7 23	0.8 12.	96	0.101	186 4.	52 4	36
649 Bi	ñ	lanced	235	710	795	2,016	2.89	113.2	0.65	3558	43.3	31.5	53.5	46,488	0.334	34.0 2	2.8 4.	2.3 22	5.8 12.	.68	0.100	185 4.	35 4	35
650 Ba	ĝ	lanced	235	710	171	2,031	2.90	103.1	0.65	3559	43.3	30.9	53.0	46,488	0.326	33.4 2	2.2 4	2.0 22	0.5 12.	.38	0.099	184 4.	39 4	39
651 Bi 657 Bi	ää	alanced	234 234	710 710	767 763	2,019 2,025	3.06 2.97	103.0 104.0	0.64 0.64	3515 3522	43.2 43.5	30.8 30.2	53.0	46,488 46,488	0.328 0.325	34.7 2 33.5 2	2.2	2.1 21	8.9 12. 7.3 12.	29	0.098	187 4. 186 4.	14 18 4 4	.40 40
655 B	m	alanced	235	710	669	2,065	3.21	72.6	0.64	3495	42.7	30.6	54.0	46,488	0.313	34.4 2	1.2 4	0.5 20	5.9 11.	.61	0.091	184 3.	98 4	34
656 F	щ	salanced	235	710	745	2,039	2.92	98.0	0.65	3528	42.6	30.4	54.0	46,488	0.309	32.3 2	1.4 4.	2.1 21	2.4 11.	.93	0.094	180 4.	20 4	37
657 E	щ	3alan ced	235	711	765	1,991	3.05	122.6	0.64	3484	43.4	33.5	54.5	46,488	0.311	33.5 2	1.5 4.	3.4 21	1.5 11.	88.	0.092	180 4.	01 4	.42
658 E	ща	alanced	235	711	0//	1,996	3.00	125.1	0.64	3488	43.3	30.1	53.5	46,488	0.314	33.3 2	1.7 4.	3.3 21	3.3 11.	98	0.095	185 4.	05 4	.40
659		Balanced	+62	017	00/	2,021	3.08	C 10.7	0.63	0700	C.C+ C.AA	20.0	5.05	40,400	60C.0	2 1.10	4 00	0.4 10	1.7 10	-74 76	060.0	180 4.	4 4	27
002 663		Balanced	235	711	699	2,092	3.10	80.6	0.64	3480	43.9	30.5	51.5	46,488	0.279	31.3 1	9.9	1.6 19	3.2 10.	80	0.089	184 4	04	38
665	_	Balanced	235	710	663	2,049	3.18	85.9	0.63	3423	43.3	29.5	52.0	46,488	0.276	31.7 1	9.7 4.	2.1 18	9.1 10.	.62	0.087	180 3.	82 4	34
666		3alan ced	235	710	654	2,034	3.04	79.3	0.62	3407	43.6	29.1	51.0	46,488	0.273	30.7 1	9.8 4	1.4 18	9.5 10.	.64	0.088	185 3.	84 4	.45
667		Balanced	234	711	675	2,037	3.36	110.0	0.63	3417	44.1	29.5	50.5	46,488	0.275	32.3 1	9.4 4.	3.4 18	5.5 10.	.47	0.088	189 3.	59 4	34
668		Balanced	234	710	665	2,088	2.88	95.4	0.64	3486	45.0	30.6	49.5	46,488	0.264	28.5 1	9.1 4	2.6 18	7.5 10.	.53	060.0	183 4.	02 4	33
699		Balanced	234	710	682	2,063	3.27	122.8	0.63	3442	44.9	29.7	49.0	46,488	0.264	30.9 1	9.0 4	4.4 18	4.4 10.	.35	060.0	189 3.	64 4	36
670		Balanced	234	710	682	2,066	3.15	122.8	0.63	3461	45.4	29.4	48.0	46,488	0.261	30.0 1	9.0 4	4.3 18	4.8 10.	.37	0.092	195 3.	76 4	33
671		Balanced	234	710	674	2,050	3.01	120.3	0.63	3439	45.5	29.7	48.0	46,488	0.259	29.1 1	9.0 4	4.3 18	2.6 10.	.26	0.091	193 3.	87 4	28
672		Balanced	234	710	675	2,078	2.91	125.6	0.64	3478	46.2	30.0	47.0	46,488	0.251	27.7 1	8.6 4	4.8 18	0.9 10.	.16	0.092	188 3.	95 4	29
673 1	_	3alan ced	234	710	683	2,093	2.89	139.0	0.64	3507	46.5	29.3	46.0	46,488	0.245	27.1 1	8.2	5.4 18	0.4 10.	.13	0.094	191 3.	92 4	29
674		Balanced	235	710	684	2,059	2.88	139.8	0.64	3494	45.9	30.2	47.5	46,488	0.241	26.8 1	8.1 4.	5.9 17	8.7 10.	-04	060.0	185 3.	98 4	37
675		Balanced	235	710	659	2,066 2,042	3.06	129.9	0.63	3438	45.9	28.5	46.5	46,488 46 488	0.237	27.8 1	8.0 9.6	5.4 17	4.1 9.5	78	0.089	187 3.	78 4	8, 8
0/0		Balanced	235	117	648	2,042	50 C	7.001 116.8	<i>co.</i> 0	3387	43.1	30.4	63.0	46.488	0.738	1 0.02	0./ 8 8	4.6 17	15 27	80 08	0.078	187 3	00 90	45
6%0 680		Balanced	235	711	660	2,019	2.76	123.3	0.62	3387	43.4	30.0	52.0	46,488	0.243	26.2	8.1	5.0 17		68	0.081	184 3.	81 4	45
681		Balanced	235	710	567	2,042	2.97	61.7	0.61	3312	43.8	28.8	50.5	46,488	0.253	26.9 1	7.7 4	0.7 16	7.1 9.3	38	0.079	182 3.	51 4	48
682		Balanced	235	711	566	2,136	2.98	49.2	0.63	3419	45.3	30.8	49.0	46,488	0.248	27.2	8.0 3	9.8 17	.6 0.0	56	0.083	199 3.	79 4	29
683		Balanced	234	710	537	2,175	3.02	38.1	0.63	3470	43.9	31.8	52.5	46,488	0.233	26.7 1	7.4 31	9.2 16	4.8 9.2	24	0.075	207 3.	99 4	28
686		Balanced	235	710	624	2,027	3.11	112.1	0.61	3353	42.9	30.0	53.0	46,488	0.244	27.0 1	7.3 4.	5.1 16	5.3 9.2	35	0.075	187 3.	43 4	.45
687		Balanced	235	710	616	2,060	2.92	101.1	0.62	3386	43.2	30.1	52.5	46,488	0.246	25.9 1	7.4 4.	3.8 16	8.7 9.4	48	0.076	185 3.	61 4	4
689		Balanced	234	711	529	2,171	2.57	112.6	0.63	4407	42.0	31.2	43.5	36,124	0.240	18.9 1	3.9 4	6.4 13	5.8 9.8	- 68	0.075	195 3.	87 4	41
069		Balanced	234	711	529	2,144	2.52	114.9	0.62	4375	42.4	30.4	42.5	36,124	0.236	18.5 1	3.8 4	7.1 13	4.7 9.7	75	0.076	192 3.	89 4	41
169		Balanced	234	712	557	2,140	2.51	128.3	0.62	4411	43.6	29.4	40.0	36,124	0.247	19.0	4.1	7.9 13	9.6 10.	.10	0.083	215 3.	91 4	38
692		Balanced	235	712	594	2,140	2.98	73.6	0.63	4215	41.9	32.2	47.0	38,324	0.290	26.7 1	7.6 4	1.8 17	2.0	F. 1	0.087	185 3.	89 4 ·	39
693		Balanced	255	717	966	2,130	2.90	8.0/	0.63	4200	41.9	32.8	6./4	38,324	0.288	20.2	7.6 4.	2.0 17	.11 61	17.	0.086	186 3.	94 4	40
694		Balanced	235	711	672	2,022	3.74	65.6	0.62	3890	40.8	32.9	52.5	40,524	0.373	0.4	1.4	0.5 19	5.7 12.	80	0.091	187 3.	50 4	33
695		Balanced	235	711	667 201	2,091	3.37	68.2	0.64	3992	41.7	38.9	55.5	40,524	0.340	34.1 2	0.6	0.4 19	7.9 12.	.73	0.085	163 3.	80	33
697		Balanced	C67	cu/	/84	2,101	1.44	08.8	0./0	C114	47.4	20.2	C.4C	42,124	0.421	7 7.17	H C.2		5.9 14.	17.	0.102	1// 0.	4	54 1
700		Balanced	235	710	807	2,050	3.31	77.6	0.66	3727	41.0	32.8	57.5	44,924	0.413	40.6 2	4.7 4	0.1 24	1.7 14.	.03	0.100	172 4.	45 4	38
701		Balanced	235	710	808	1,994	3.61	78.2	0.65	3660	40.1	27.1	55.0	44,924	0.417	43.4 2	4.8	0.1 24	1.8 14.	-04	0.105	177 3.	83 4	26
704		Balanced	225	700	577	2,157	2.86	24.7	0.61	3488	40.8	30.4	55.5	44,924	0.408	29.9 2	ic 0.0	8.2 18	1.1 10.	.50	0.078	165 4.	05 4	.45
710		Balanced	235	710	756	2,104 2,002	3.40 3.77	86.1 79.0	0.65	3688	43.7 43.5	30.2	50.0	44,924	0.362	38.5 2	3.0 4	0.8 22	0.5 12.	16.	0.106	164 4.	20 4	35 26
111/		Balanced	C52	710	cc/.	2,092	5.27	78.9	0.66	5/39	43.5	31.7	51.5	44,924	0.346	36.2 2	2.5	27 I.I.		18.	0.102	165 4.	53	36
712		Destec	235	695	722	2,157	1.06	96.9	0.69	3816	45.5	33.6	45.5	44,924	0.368	15.9 1	9.8 4.	2.6 20	3.4 11.	- 81	0.099	158 7.	28	38

						Tal	ble D-	2 – D£	ita Su	mmar	y Tabl	e for	Kings	port	Cataly	st Cai	npaig	;n 2 (c	ont'd						
		Days				Fresh	Recycle	Reactor	Purge	Inlet Sup.	Space	Slurry	Gas	Gassed	Catalyst	Catalyst	0 F	keactor)-T-M S	yngas M	Raw 1eOH	Catalyst	Reactor	U Overall	Sparger	Sparger
Case	Date	On Stream	Gas Type	Temp (Deg C)	Pres. (psig)	Feed (KSCFH)	Gas (KSCFH)	Feed (H2:CO)	Gas (KSCFH)	Velocity (ff/sec)	Velocity (l/hr-kg)	Conc. (wt% ox)	Holdup (vol%)	Slurry Het (ft)	Inventory (lb)	Age (eta)	Conv. ((%)	Conv. (S)	Util. F CF/lb) (1	Prod. M (PD) (g	feOH Prod. mol/hr-kg)	Vol. Prod. (TPD/ft3)	(Btu hr ft2 F)	db (psi)	Resist.
5				1.0							(a)					0100) t			į į	
17 6	3-Dec-99 4 Dec 00	714	Destor	CC7	569	771	2,140 2,115	1 06	08.0	0.09	2006	44.0	0.00 0.15	0.0c	44,924	275.0	16.5	20 3 C.61	42.9 2 11.8 2	1.202	6/ 11 00 CI	0.004	661 166	07.7	60.4 4 35
21	5-Dec-99	715	Destec	235	695	715	2,121	1.00	97.0	0.68	3770	44.0	33.0	51.5	44,924	0.381	15.8	20.0	41.9 2	04.7	11.90	0.095	160	7.18	4.33
21	8-Dec-99	718	Destec	235	695	730	2,112	1.03	120.6	0.69	3813	44.7	32.9	50.0	44,924	0.370	15.5	19.5	43.1 2	03.2	11.81	0.097	184	7.05	4.29
21	9-Dec-99	719	Destec	234	969	703	2,127	1.02	107.5	0.68	3797	45.2	32.0	48.5	44,924	0.346	14.9	18.7	43.3 1	94.9	11.34	0.096	148	7.06	4.29
21	10-Dec-99	720	Destec	234	697	703	2,127	1.00	112.5	0.68	3766	46.0	32.6	47.5	44,924	0.353	15.0	18.9	43.2 1	195.7	11.37	0.098	151	6.92	4.32
= :	16-Dec-99	726	Balanced	235	710	662	1,925	4.44	107.7	0.59	3323	40.5	25.7	53.0	44,924	0.325	41.3	20.4	43.4 1	83.2	10.63	0.082	170	2.80	4.62
= =	17-Dec-99 18-Dec-99	727 728	Balanced Balanced	235 235	710	659 665	1,942	4.08 3.98	76.3 77.6	0.60	3390	37.5 38.1	24.3 23.3	58.5	44,924 44,924	0.329 0.329	40.3 40.1	21.1	41.1 1	92.2 93.0	11.16	0.083	176	3.18 3.24	4.54 4.48
: =	19-Dec-99	729	Balanced	235	710	658	1,936	3.86	75.3	0.60	3389	40.0	28.9	56.5	44,924	0.321	38.9	21.1	41.4 1	50.7	11.08	0.080	170	3.32	4.47
Ξ	20-Dec-99	730	Balanced	235	710	731	1,953	3.24	94.6	0.62	3511	39.9	25.3	54.0	44,924	0.326	35.4	22.0	42.3 2	207.3	12.04	0.091	177	3.75	4.41
Ξ	22-Dec-99	732	Balanced	235	710	869	2,060	3.02	81.2	0.64	3622	41.6	32.1	55.5	44,924	0.297	31.6	20.9	41.5 2	1.20	11.72	0.087	172	4.11	4.47
Ξ	23-Dec-99	733	Balanced	235	710	724	1,992	3.24	104.9	0.63	3566	41.7	30.4	54.0	44,924	0.303	33.8	21.2 .	42.8 2	202.8	11.77	0.089	177	3.79	4.46
н	24-Dec-99	734	Balanced	235	710	721	1,996	3.18	102.8	0.63	3572	42.3	31.4	53.5	44,924	0.300	33.1	21.1	42.8 2	202.2	11.73	0.090	177	3.80	4.46
Ξ	25-Dec-99	735	Balanced	235	710	727	2,005	3.01	103.4	0.64	3604	43.0	33.2	53.5	44,924	0.297	31.8	21.1	42.5 2	205.2	11.90	0.091	177	3.92	4.47
Ξ	26-Dec-99	736	Balanced	235	710	738	1,946	3.31	135.4	0.62	3508	43.0	30.7	51.5	44,924	0.300	33.2	20.6	44.8 1	196.7	11.42	0.091	173	3.47	4.49
=	27-Dec-99	737	Balanced	235	710	719	1,981	3.29	117.0	0.63	3538	42.5	30.0	52.0	44,924	0.296	33.5	20.8	43.6 1	0.861	11.49	0.091	177	3.65	4.4
= :	28-Dec-99	738	Balanced	235	710	742	1,964	3.05	125.6	0.63	3561	42.0	35.5	57.5	44,924	0.297	32.3	21.3	43.8	203.2	11.79	0.084	168	3.87	4.42
=	29-Dec-99	739	Balanced	C62	01/	C60	/ 10'7	5.10	90.0	0.65	6065	42.1	5.0.5 2.0.5	0.25	44,924	0.288	32.1	20.0	42.0	0.06	11.37	060.0	180	3.88	4.41
= =	31-Dec-99 1-Ian-00	747	Balanced Balanced	235	710	67/ 177	979 1 1 979	2.91 2.91	114.1	0.63	1000	41.5	305	54.5	44,924 44 924	0.287	30.8	20.8	43.0 I 43.7 I	- 766 J - 766 J	11 50	160.0	163	4.04 3.95	4.02 4 43
=	2-Jan-00	743	Balanced	235	710	717	1,952	3.05	124.8	0.62	3501	41.4	27.6	52.5	44,924	0.286	31.5	20.5	14.2	94.5	11.31	0.088	165	3.71	4.39
Ξ	3-Jan-00	744	Balanced	235	710	692	1,946	2.93	98.6	0.61	3462	41.4	26.8	52.0	44,924	0.292	31.2	20.8	42.8 1	93.8	11.26	0.089	163	3.85	4.42
Ξ	4-Jan-00	745	Balanced	235	710	694	1,991	2.82	97.6	0.62	3532	42.6	26.5	49.5	44,924	0.286	29.8	20.6	42.6 1	195.4	11.35	0.094	179	4.29	4.75
Ξ	5-Jan-00	746	Balanced	235	710	694	2,009	3.06	113.2	0.63	3567	43.0	26.9	49.0	44,924	0.273	29.9	19.6	43.8 1	1.061	11.03	0.092	172	3.57	4.43
= =	6-Jan-00	747	Balanced	235	710	200	1,986	3.25	123.8	0.62	3504	42.5	29.8 32.6	52.0	44,924	0.279	31.6 20.0	19.8	44.4	189.4	10.99	0.087	169	3.49	4.50 4.48
= =	/-Jan-00 8-Ian-00	749	Balanced	235	710	160	2.014	2.82	109.1	0.63	3549	42.9	35.5	55.5	44,924	0.271	28.6	19.9	43.8 1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-	0.061	11.09	0.082	153	3.87	4.41
Ξ	9-Jan-00	750	Balanced	235	710	695	2,005	2.65	107.7	0.63	3555	43.3	35.7	55.0	44,924	0.271	27.4	20.0	43.6 1:	91.4	11.10	0.083	152	4.03	4.37
Ξ	10-Jan-00	751	Balanced	235	710	697	1,996	2.65	112.1	0.63	3538	42.9	32.2	53.0	44,924	0.276	27.9	20.1	43.5 1	92.2	11.16	0.086	157	4.04	4.38
Ξ	11-Jan-00	752	Balanced	235	710	701	2,027	2.79	118.9	0.64	3592	42.9	32.1	53.0	44,924	0.273	28.2	. 19.7	43.9 1	9716	11.13	0.086	154	4.04	4.41
= =	12-Jan-00	753	Balanced	235	710	701	1,996 2.015	2.71	122.2	0.63	3535 3576	42.5	33.8	55.0 40.5	44,924 44.024	0.271	27.8 36.0	19.7	44.3 14.0	6.681	11.03	0.082	149	3.92	4.40
: =	14-Ian-00	755	Balanced	235	710	697	2,054	2.68	120.8	0.64	3593	43.2	26.6	48.5	44,924	0.263	26.6	19.2	14.4	.88.3	10.93	0.093	165	4.01	4.48
=	20-Jan-00	761	Balanced	235	710	563	2,134	2.42	113.5	0.63	4721	35.9	21.6	45.5	33,687	0.264	19.8	15.2 .	45.8 1.	47.4	11.41	0.077	164	3.96	4.31
Ξ	21-Jan-00	762	Balanced	235	710	555	2,117	2.43	115.5	0.62	4680	35.7	21.2	45.5	33,687	0.257	19.4	14.9	46.4 1	143.5	11.11	0.075	144	3.88	4.32
Ξ	22-Jan-00	763	Balanced	234	710	548	2,089	3.13	62.6	0.61	4328	37.3	25.4	48.0	35,887	0.297	27.7	17.5	40.8 1	161.2	11.71	0.080	135	3.59	4.19
Ξ	24-Jan-00	765	Balanced	234	710	646	2,004	3.95	76.8	0.61	4057	39.0	22.7	46.0	38,087	0.374	38.2	20.4	41.3 1	87.7	12.85	0.097	162	3.13	4.27
Ξ	3-Feb-00	775	Balanced	235	710	863	1,885	4.21	82.1	0.63	3795	38.7	32.8	56.5	42,487	0.491	44.8	27.4	40.1 2	258.1	15.83	0.109	157	4.02	4.23
= :	4-Feb-00	776	Balanced	235	710	859	1,960	3.29	81.6	0.65	3905	41.3	31.8	53.0	42,487	0.479	43.5	26.5	40.1 2	256.8	15.75	0.115	167	4.21	4.21
= :	5-Feb-00	LLL	Balanced	235	017	854	1,921	3.27	85.8	0.04	3842	41.0	31.0	53.0	42,487	0.454	42.4	26.1	40.8	5.1.3	15.41	0.113	157	4.27	3.97
= =	6-Feb-00 7 Eab-00	8/1	Balanceu Balancad	C62 734	110	558 558	1,940	3.51 3.16	87.3	0.04	3830	40.8	50.4 7.7.0	0.55	42,481	0.470 0.470	42.1	25.6	40.4 10.2 10.2	245.5	15.68	0.110	147	5.95 4.03	4.24
: =	/-reb-00 8-Feh-00	780	Balanced	234	710	824	1.998	3.27	88.5	0.65	3895	40.8	27.6	51.0	42.487	0.427	40.1	24.8	10.9 2.	41.6	14.82	0.113	150	4.03	427
Ξ	9-Feb-00	781	Balanced	234	710	840	1,917	3.29	99.1	0.64	3802	42.7	26.0	46.5	42,487	0.442	41.9	25.7	41.2 2	44.4	14.99	0.126	169	3.92	4.25

D-17

		Sparger Resist.	("K")	4.23	4.23	4.22	4.23	4.29	4.29	4.31	4.30	4.28 4.42	4.19	4.17	4.21	4.19	4.22	4.17	4.13	4.19	4.22	4.23	4.21	4.25	4.20	4 33	4.22	4.22	4.24	4.23	4.16	4.14	4.21	4.19 4.13	4.12	4.09	4.14	3.96	4.01	4.25 4.33	4.25	4.36	4.18
		Sparger dP	(psi)	4.12	4.09	4.12	4.22	4.19	3.81	3.91	4.15	3.95 4.30	8.28	3.42	4.27	4.17	4.15	4.22	4.31	4.11	4.19	4.20	4.11	4.12	4.41 4.47	4.33	4.16	4.27	4.24	431	3.97	4.22	4.14	c0./	6.98	6.84	6.83	6.59	6.52	6.89 7.00	7.61	7.62	7.18
	n	Overall (Btu	hr ft2 F)	152	140	139	132	147	159	159	168	166 174	164	160	152	151	153	155	151	156	157	153	153	155	159	160	155	157	149	148	152	148	14/	167	150	158	149	150	143	111	130	122	117
	1	Reactor Vol. Prod.	(TPD/ft3)	0.125	0.122	0.123	0.119	0.128	0.117	0.120	0.126	0.123 0.130	0.074	0.090	0.115	0.118	0.117	0.119	0.119	0.117	0.117	0.116	0.119	0.117	0.120	0.111	0.108	0.112	0.109	0.106	0.108	0.104	0.103	0.094 0.085	0.084	0.091	0.092	0.091	0.089	0.090 0.081	0.091	0.097	0.082
		italyst DH Prod.	ol/hr-kg)	5.25	5.19	4.93	5.14	5.06	5.23	5.12	5.45	5.29 5.44	0.28	0.45	3.92	3.97	3.89	4.19	4.13	4.13	4.18	4.16	4.09	3.76	3.86 3.67	3.31	3.41	3.50	3.24	2.96	2.59	3.02	2.00	0.92	0.89	0.94	0.93	0.91	0.88	0.82 0.70	0.64	0.71	0.50
	8	d. MeC C	D) (gmo	5 1	1 60	1	3	-	L L	-	.8	63	9	1	1 1	3	6.	6	9		6	. 1	7	2	 n -		1 61	4.	4		5	5		0.8	3	1	.0	L L	2		.8	4	1.8
ıt'd)	Rav	IS MeC	b) (TP	261	260	256	259	258	261	259	264	262 264	176	179	238	239	237	242	241	242	242	242	241	235	237	228	229	231	227	222	215	222	210	186	186	187	187	186	186	185 183	181	183	180
(co1	1	Synga Util.	(SCF/I	40.6	41.4	41.9	41.9	41.6	41.3	42.0	40.8	41.8	42.4	38.1	43.4	43.8	43.0	42.5	42.2	42.3	42.2	42.2	42.5	42.1	42.3	42.6	42.7	42.7	43.2	42.3	41.5	43.5	C.44	41.3	41.9	42.5	42.8	42.2	46.2	43.8 43.6	44.8	43.0	42.4
uign (Reactor	O-T-M Conv.	(%)	27.5	27.3	26.7	27.1	26.8	27.9	27.5	27.6	27.9 24.9	17.9	20.1	24.6	24.2	24.3	24.8	24.5	25.1	25.1	24.9	24.4	24.5	24.5 24.3	24.1	23.6	23.6	23.3	22.6	22.4	23.0	1.77	1.9.1	19.0	19.1	19.0	19.5	16.2	19.3 18.4	17.3	17.7	17.7
ampa		CO Conv.	(%)	43.8	42.8	41.7	41.0	41.2	47.5	46.1	43.7	45.7 43.4	11.4	46.3	39.3	38.5	38.8	38.2	36.8	38.7	38.7	38.0	38.0	38.8	36.8 36.8	37.5	36.6	35.2	34.9 36.5	33.7	35.8	34.8	0.65	1.61	14.9	15.0	15.0	16.0	16.2	15.4 15.8	14.6	15.2	14.9
vst C:		Catalyst Age	(eta)	0.475	0.468	0.449	0.460	0.453	0.482	0.472	0.468	0.475 0.467	0.500	0.402	0.400	0.392	0.393	0.404	0.396	0.404	0.404	0.401	0.398	0.398	0.396	0.376	0.364	0.364	0.357	0.346	0.346	0.353	0.344	0.374	0.375	0.370	0.365	0.373	0.364	0.371 0.346	0.327	0.325	0.327
Cataly		Catalyst Inventory	(ql)	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687 44.687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687 44 687	44.687	44,687	44,687	44,687	44.687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687 44,687	44,687	44,687	44,687
port		Gassed Slurry	Hgt (ft)	50.0	51.0	49.5	52.0	48.0	53.0	51.5	50.0	51.0 48.5	56.5	47.5	49.5	48.5	48.5	48.5	48.5	49.5	49.5	50.0	48.5	48.0	47.0 49.0	49.0	50.5	49.0	49.5 18 5	50.0	47.5	51.0	0.06	48.0 52.5	52.5	49.0	48.5	49.0	50.0	49.0 54.0	47.5	45.0	52.5
Kings		Gas Holdup	(vol%)	25.2	30.5	23.4	32.3	32.9	29.3	26.0	23.9	24.4 24.9	35.6	22.7	26.1	26.1	26.0	25.9	26.6	26.4	26.2	27.3	26.1	25.0	22.6	24.1	27.7	25.1	26.1 73.8	24.9	23.3	27.2	27.0	27.8	27.8	25.1	25.2	27.9	28.1	25.1 31.8	22.4	31.0	32.2
for F	;	Slurry Conc.	wt% ox)	41.7	43.1	41.4	43.2	45.6	41.7	41.2	41.3	40.9 42.4	42.4	42.2	42.3	42.8	42.8	42.6	43.0	42.4	42.3	42.4	42.8	42.7	42.5 42.0	42.6	42.4	42.2	42.2	41.6	42.4	41.9	42.4	41.0 41.4	41.4	42.2	42.5	43.2	42.7	42.2 42.1	42.1	46.6	43.0
Table		Space Velocity	l/hr-kg) (3620	3627	3653	3645	3646	3563	3573	3630	3570 3649	3760	3423	3689	3726	3689	3688	3724	3644	3661	3656	3692	3615	3663 3645	3609	3695	3701	3674 3675	3720	3635	3663	3622	3719 3719	3713	3720	3713	3712	3697	3639 3740	3940	3934	3867
mary		let Sup. elocity	fi/sec) (0.64	0.64	0.64	0.64	0.64	0.63	0.63	0.64	0.63 0.64	0.66	0.61	0.65	0.65	0.65	0.65	0.65	0.64	0.64	0.64	0.65	0.64	0.64	0.63	0.65	0.65	0.65	0.65	0.64	0.64	0.64	0.66 0.66	0.66	0.66	0.66	0.66	0.66	0.65 0.67	0.70	0.70	0.69
Sumi		urge In Gas V	SCFH) (94.7	13.0	16.0	23.1	19.5	11.6	8.60	01.5	22.7 12.2	85.1	6.7	29.5	35.6	33.8	25.5	18.6	21.4	20.7	18.1	26.2	14.3	12.2	18.2	24.0	24.2	30.4	0.60	31.0	33.6	47.0	8.0	89.3	0.6	98.1	94.7	04.9	05.3 05.0	19.5	98.1	5.4
Data		actor F eed	2:CO) (K	4	10 1	1 11.	.96 1	.02	.48 1	.39 1	.12 1	1 29	157	1.74	17 1	.15 1	.16 1	00	1 06.	03	02	.96	.05	60. 5	101	05	.08	.92 1	1 10	16	.16	1 1	.03	66 66	.94	96	.97	.05	80.	96. 11.	.08	.15	Π.
D-2 -		ycle Re as F	TH) (H	70 3	57 3	84 3	64 2	37 3	19 3	41 3	57 3	90 62 9	37 0	41 4	49 3	56 3	42 3	44 E0	74 2	10	8:	191	55	82 :	64 6 64 6 7 7 7	32 3	95 3	81 2	66 2 2 2	26 2	97 3	74	F 70	21 0	30 0	30 0	21 0	32 1	13	69 91 10	46 1	72 1	55 1
able	1	G	H) (KS	1,8	1,8	1,8	1,8	1,8	1,8	1,8	1,8	3, 3, 1	2,2	2,0	1,9	1,5	1,9	1,9	1,5	5 <u>,</u>	- -	<u> </u>	<u> </u>	<u>.</u>	2, 1	1.0	1,9	1,9	1,5	2.0	1,0	2,1,5	1,5	- - - - - - - - - - - - - - - - - - -	2,1	2,1	2,1	2,1	2,1	0, U	2,3	5,5	2,2
Ē		Fresh Feed	(KSCF)	884	868	894	905	896	006	908	006	916 908.7	624	570	854	860	852	861	850	854	854	852	855	825	836	808	818	824	818	783	746	808	804	65U 662	651	663	667	656	657	675 626	638	657	640
		Pres.	(psig)	710	710	710	710	710	710	710	710	710	710	700	710	710	710	710	710	710	710	710	712	710	710	710	710	710	710	710	710	710	/10	700/	700	700	700	700	700	700 695	200	700	700
		Temp	(Deg C	234	233	233	233	233	235	235	235	235 235	235	234	234	234	234	234	234	234	234	234	234	234	234	234	234	234	234	234	234	234	234	235 235	235	234	234	234	234	234 233	234	234	234
		Gas	Type	Balanced	Balanced Balanced	Shell	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	= =	1:1	1:1	1:1	1:1	E	ΞΞ	1:1	Ξ	E							
	1	Days On	Stream	782	783	784	785	786	787	788	789	701	795	797	799	800	801	802	803	804	805	806	807	808	810	817	813	814	815	818	819	820	821	825 826	827	828	829	830	831	832 833	834	836	837
			Date	10-Feb-00	11-Feb-00	12-Feb-00	13-Feb-00	14-Feb-00	15-Feb-00	16-Feb-00	17-Feb-00	18-Feb-00	23-Feb-00	25-Feb-00	27-Feb-00	28-Feb-00	29-Feb-00	1-Mar-00	2-Mar-00	3-Mar-00	4-Mar-00	5-Mar-00	6-Mar-00	8-Mar-00	9-Mar-00	11-Mar-00	12-Mar-00	13-Mar-00	14-Mar-00	17-Mar-00	18-Mar-00	19-Mar-00	20-Mar-00	24-Mar-00 25-Mar-00	26-Mar-00	27-Mar-00	28-Mar-00	29-Mar-00	30-Mar-00	31-Mar-00 1-Apr-00	2-Apr-00	4-Apr-00	5-Apr-00
			Case	Ξ	Ξ	Ξ	Ξ	Ξ	Ξ	Ξ	Ξ	= =	25	Ξ	Π	Ξ	Ξ	Ξ	Ξ	= :	= :	= :	= :	= :	= =	: =	Ξ	Ξ	= =	=	Ξ	= :	=	212	21	21	21	21	21	21	21	21	21

D-18

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

	Sparger	Resist. ("K")	4.29	4.28	4.49	4.17	4.09	4.10	4.12	4.13	4.37 4.38	4.56	4.32	4.40	4.56	4.40	10.4	4.40	4.36	4.47	4.52	4.00	3.97	3.96	3.95	4.06	4.00	3.96	3.97	4.22	4.15	4.11	4.11	4.06	4.29	4.20	4.26	4.34	4.33	4.27	4.33 4.26	4.68 4.56
	Sparger	dP (psi)	7.24	7.38	7.90	7.08	6.90	6.80	6.98	4.19	420 429	4.53	4.18	4.28	3.28	197	10.4	4.03	4.74	4.45	3.97	4.33	4.43	4.31	4.49	4.68	4.70	4.16	4.08	4.3	4.19	3.96	3.95	4.07	4.28	4.33	4.17	4.29	4.71	4.39	4.63 4.54	4.86 4.73
	U Overall	(Btu hr ff2 F)	113	126	135	120	131	106	66	130	122	124	113	139	139	132	101	135	135	129	149	141	142	130	119	131	128	132	061	151	126	131	131	123	119	122	125	138	146	127	138	120 122
	Reactor	'ol. Prod. TPD/ft3)	0.079	0.082	0.088	0.081	0.081	0.083	0.080	0.092	0.085 0.089	0.083	0.074	0.076	0.077	0.0/8	0.000	0.103	0.110	0.110	0.112	0.115	0.121	0.125	0.120	0.121	0.121	0.099	0.104	0.115	0.101	0.107	0.107	0.103	0.107	0.109	0.110	0.111	0.119	0.112	0.119 0.118	0.107 0.108
	lyst	Prod. V hr-kg) (26	14	22	10	13	10	90	52	50 37	02	80	54	04	51 26	07	36	35	63	27	62	67	05	08	97	08	53	646	92	76	70	63	70	71	68	30	55	79	59	60 60	48 52
	I Cata	MeOH (gmol/	10.	10.	5.6	10.	10.	10.	10.	10.	10.	10	3.6	10.		2	1 2	1 1 1	14.	13.	13.	13.	13.	14.	14.	13.	4		12.	12.	12.	12.	12.	12.	12.	12.	12.	12.	12.	12.	12.	12.
it'd)	Raw s MeOF	Prod. (TPD)	174.9	173.2	163.8	172.7	173.2	172.7	172.0	180.1	179.6 177.4	1715	169.0	137.9	154.0	0.171	1.101	209.0	224.3	224.6	218.7	236.0	236.8	243.1	243.6	241.8	243.7	6.991 1.031	223.9	223.8	221.4	220.0	218.6	219.8	220.1	219.6	212.9	217.2	221.7	218.0	220.0 218.1	216.1 216.7
con	Syngas	Util. (SCF/lb	46.8	48.0	43.5	43.2	42.8	44.5	45.0	43.2	42.9 44.3	44.5	44.8	47.9	42.8	41.1	1.01	41.7	41.1	41.4	40.3	39.9	40.2	43.4	43.0	42.4	42.4	38.3	42.6	41.0	41.5	41.7	42.0	41.5	41.7	41.7	42.2	41.2	40.1	41.8	41.4 42.1	41.3 41.7
uign 2	Reactor O-T-M	Conv. (%)	16.5	16.3	16.6	16.7	17.2	16.9	17.0	17.8	17.5	16.9	16.9	13.6	15.8	16.4	10.0	20.9	21.9	21.9	22.4	23.9	23.5	23.6	23.6	23.8	23.9	21.2	22.7	21.7	21.4	21.4	21.4	21.4	21.2	21.1	20.7	21.2	21.2	21.0	21.3 21.2	20.9 20.9
amps	6	Conv. (%)	12.5	12.1	13.9	14.1	14.3	14.5	13.7	26.0	26.5 25.5	24.9	25.8	19.1	32.7	27.2	0.02	37.0	34.6	37.5	42.6	40.5	38.8	37.5	26.2	37.1	36.0	40.0	40.8	37.3	36.7	37.1	36.9	35.7	34.3	33.5	33.7	34.5	32.4	32.4	31.4 30.9	32.0 32.4
yst C	Catalyst	Age (eta)	0.312	0.312	0.305	0.306	0.327	0.308	0.316	0.271	0.266 0.258	0 244	0.234	0.247	0.319	0.201	100.0	0.380	0.395	0.384	0.411	0.415	0.405	0.397	0.397	0.399	0.397	0.307	0.387	0.360	0.349	0.347	0.344	0.346	0.342	0.337	0.326	0.333	0.336	0.329	0.332 0.329	0.320 0.320
Catal	Catalyst	Inventory (lb)	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687	44,687 44.687	44 687	44,687	34,207	36,407	36,407 36,407	104,00	40.807	40,807	43,007	43,007	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207	45,207 45,207	45,207 45,207
sport	Gassed	Slurry Hgt (ft)	52.5	50.0	44.5	50.5	51.0	49.5	51.0	46.5	50.5 47.5	49.0	54.0	43.5	47.5	6.14 2.65	0.0 1	48.5 48.5	48.5	48.5	46.5	49.0	46.5	46.5	48.5	47.5	48.0	48.0	51.5	46.5	52.0	49.0	48.5	51.0	49.0	48.0	46.0	46.5	44.5	46.5	44.0 0.44	48.0 48.0
Kings	Gas	Holdup (vol%)	31.5	27.3	26.6	31.8	33.0	32.6	34.7	22.9	29.6 24.4	28.8	33.9	23.7	30.4	33.2	0.76	20.2 20.2	21.0	20.8	18.2	26.7	29.6	28.8	20.4	18.3	27.2	27.5	25.2	25.9	28.8	22.4	19.9	23.4	18.6	18.8	23.2	25.2	26.4	24.6	25.6 24.0	26.0 25.8
for]	Slurry	Conc. (wt% ox)	42.7	42.5	45.3	43.9	44.1	44.7	44.7	42.8	43.0 42.8	43.6	43.0	38.0	38.2 :: -	40.7 40.8	40.9	38.5	38.8	40.1	40.3	43.1	45.5	45.5	41.2	41.1	43.8	43.8	41.2	44.2	42.3	41.6	41.0	40.9	40.3	41.0	43.5	44.0	45.6	43.7	45.6 45.0	43.4 43.3
Table	Space	Velocity (l/hr-kg)	3849	3843	3842	3885	3802	3841	3813	3739	3697 3723	3750	3717	4851	4418	0174	4465	4126	4245	4013	3830	3780	3833	3844	3864	3846	3850	3701	3706	3792	3792	3752	3739	3758	3770	3807	3747	3752	3842	3800	3775 3763	3760 3783
mary	tlet Sup.	/elocity (ft/sec)	0.69	0.68	0.68	0.69	0.68	0.68	0.68	0.66	0.66 0.66	0.67	0.66	0.66	0.64	0.66	0.00	0.67	0.68	0.68	0.65	0.67	0.68	0.68	0.69	0.69	0.69	0.66	0.66	0.67	0.67	0.67	0.66	0.67	0.67	0.68	0.67	0.67	0.68	0.68	0.67 0.67	0.67
Sum	urge Ir	Gas V SCFH)	47.2	60.4	95.0	97.2	97.0	04.9	12.8	00.5	96.8 09.8	08.7	10.5	24.5	74.3	0.4c	07.4 70.5	c. <i>e</i> /	85.7	88.9	63.5	0.69	70.0	38.7	24.7	09.2	11.3	27.7	02.9	82.1	87.6	91.6	92.2	91.0	93.5	93.6	96.9	81.1	65.2	94.2	87.0 93.0	79.3 88.1
- Data	actor I	feed 2:CO) (K	16.0	.86	60.1	1.12	90.1	1.14	1.02	2.77	2.87	98 (3.00	2.74	1.48	5.50 FF (11-2	5.04 3.65	3.12	3.50	3.96	3.43	3.31	3.16	3.01	3.07	2.93	3.88		3.41	3.41	3.47	3.46	3.29	3.17	3.11	3.21	3.21	2.93	3.02	2.79	2.92 3.00
D-2 –	ycle Rc	as I CFH) (H	229 (225 (275	288	229	245	230	173	19 18	16	96	266 2	236 2	587 587	767	243 162	183	182	072	130	147	94	082	160	680	195	2 280	[4]	159	134	960	112	116	[4]	120	122	180	[42	126	137 137
able	h Rec	d G FH) (KS	2,5	2,2	3 2,5	2,2	9 2,5) 2,	5,2,	3, 2,	6 6 6 6	2	6	2,2) 2,	5 C	4 c	4 6	°,	5 2,	3 2,0	4 2,	3 2,	2,1	3 2,	4 2,	0, 2	сі с к	5 F	6	5	5 2,	5 2,	2,	4 2,	4	3	5	1, 2,	,2,	6 6	6 6 6
	Free	s. Fee ig) (KSC	0 68	0 69	0 59	0 62	0 61	0 64	0 64	0 64	0 0 65 65	0 63	0 63	0 55	0 55	0 0		02 02	0 76	0 77	0 73	6 78	6 79	88	8 87.	85	98 98	0 63	62 0	0 76	0 76	0 76	0 76	0 76	0 76	0 76	0 74	0 74	0 74	0 75	0 75 0 76	0 74 0 75
		emp Pro eg C) (ps	34 7(34 7(34 70	33 7(34 7(33 7(34 7(33 7(34 7(34 7(34 7(35 7(34 7(34 7(34 /(34 70	10 10	34 7(34 71	34 71	34 71	34 7(34 7(33 7(33 7(34 7(34 7(11 11 11 11 11 11 11 11 11 11 11 11 11	34 71	34 71	34 71	34 71	34 71	34 71	33 71	34 71	34 71	34 71	34 71	34 71	34 71 34 71	35 71 35 71
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	8	Ga Tyj	1:	ï	÷	Ξ		11		Balar	Balar Balar	Balar	Balar	Balar	Balar	Balar Balar	Data	Balar Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar	Balar Balar	Balar Balar
	Day	On Strea:	9 838	0 839	0 840	0 841)0 842	00 843	00 844	00 845	00 846 vo 847	10 853	0 854	00 856	00 858	00 855	00 901 	00 861 00 862	0 863	0 864	0 865	00 875	D0 876	00 878	90 879	90 880	00 881	00 882 vo 882	00 88, 00 889	00 890	168 00	00 892	D0 893	0 894	0 895	0 896	0 897	968 0	0 895	0 901	0 902 0 903	706 00 008
		Date	6-Apr-0	7-Apr-0	8-Apr-0	9-Apr-0	10-Apr-(11-Apr-(12-Apr-(13-Apr-(14-Apr-(21-Anr-(22-Apr-(24-Apr-(26-Apr-(27-Apr-1	-1dV-97	29-Apr-/ 30-Anr-(1-May-0	2-May-6	3-May-C	13-May-(14-May-(16-May-t	17-May-	18-May₄	19-May-	20-May-	-27-Mav-(28-Mav-(29-May-(30-May-(31-May-t	1-Jun-0	2-Jun-0	3-Jun-0	4-Jun-0	5-Jun-0	6-Jun-0	8-Jun-0	9-Jun-0 10-Jun-0	14-Jun-(15-Jun-(
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		st por	Ċ	E	ļ	LICSI	Verycle	Incatciol	r uige	Welceiter	apace .	o mic	spo	- nasseu	atatyst ,	-atalyst		fe INI-1-	ingds init		OUT De - 4	Neactor	Overall	ur	Della
Case	Date	Stream	Type	(Deg C)	rres. (psig)	(KSCFH)	uas (KSCFH)	reeu (H2:CO)	(KSCFH)	velocity (ft/sec)	velocity (l/hr-kg)	conc. (wt% ox)	Holdup (vol%) H	Sturry Ir Hgt (ft)	(lb)	Age (eta)	сопу. (%)	onv. (S(%) (S(TH (I) (T	od. Me PD) (gn	out 1700. nol/hr-kg)	VOI. PTOU. (TPD/ft3)	(Bu hr ff2 F)	ar (psi)	resist. ("K")
=	16-1un-00	606	Balanced	236	710	755	2.148	3.04	86.8	0.68	3808	43.8	27.1	48.0	45.207	0.321	32.8	5 0.9	1.6 21	8.0	12.59	0.108	129	4.53	4.32
11	17-Jun-00	910	Balanced	236	710	758	2,156	3.02	87.6	0.68	3815	43.4	27.0	48.5	45,207	0.320	32.6	9.09	1.7 21	8.1	12.60	0.107	128	4.71	4.46
Π	18-Jun-00	116	Balanced	236	710	758	2,163	3.02	90.3	0.68	3828	43.4	25.2	47.5	45,207	0.319	32.4	20.8 4	1.7 21	8.0	12.59	0.109	132	4.2	3.94
Π	19-Jun-00	912	Balanced	236	710	680	2,192	3.25	56.1	0.67	3754	44.3	23.7	45.0	45,207	0.309	33.2	20.0 3	9.9 20	14.3	11.79	0.108	140	4.4	4.35
= :	20-Jun-00	913	Balanced	236	710	672	2,193	3.16	56.1	0.67	3742	43.6	25.8	47.5	45,207	0.301	32.2	9.7 4	0.1 20	01.0	11.60	0.101	128	4.42	4.32
=	21-Jun-00	914	Balanced	235	710	676	2,195	3.08	60.2	0.67	3768	42.3	22.7	48.0	45,207	0.300	31.3	9.6 4	0.2 20	01.8	11.65	0.100	129	4.37	4.21
=	24-Jun-00	917	Balanced	234	710	682	2,136	3.35	71.5	0.65	3661	43.8	28.0	48.5	45,207	0.313	34.0	20.1	1.3 15	8.1	11.45	0.097	134	4.14	4.19
= :	25-Jun-00	918	Balanced	234	710	681	2,141	3.27	71.9	0.65	3670	42.9	23.9	47.5	45,207	0.310	33.2	20.1	1.3 15	08.0 	4. : :	0:099	136	4.32	4.29
= :	26-Jun-00	919	Balanced	234	017	180	2,141	3.27	74.4	0.02	306/	45.5	25.4	6.74	45,207	0.309	33.2	20.0 1 0		1.4	11.41	660.0 200.0	138	4.25	4.22
= =	27-Jun-00	920	Balanced	234	710	668	2,137	3.30 2.78	74.2	C0.U	3654 3661	45.4 A5.5	26.9	C.84	45,207	0.308	33.6	20.1 4 0.6	21 21	1.4	11.00	/60.0	241	4.11	4.15 1 11
= =	20 Inn 00	126	Balanced	234 234	017	600 691	2,130 2,136	82.C	01 1	co.u	1005	6.64 44.3	255	46.0	45,207	267.0 797 0	C.2C	9.0	21 1.1 10 10 10	1.2	11 29	001.0	142	4.42	4.44
= =	30-Jun-00	923	Balanced	234	710	683	2,150	3.17	89.2	0.66	3696	42.3	23.6	48.5	45,207	0.291	31.2	9.4	2.4 19	3.1	11.16	0.095	133	4.26	4.21
П	1-Jul-00	924	Balanced	234	710	684	2,158	3.11	90.5	0.66	3699	42.3	23.7	48.5	45,207	0.290	30.8	9.4 4	2.6 19	12.5	11.13	0.095	132	4.36	4.22
Ξ	2-Jul-00	925	Balanced	234	710	685	2,135	3.10	88.3	0.65	3680	42.5	24.2	48.5	45,207	0.291	30.7	9.5 4	2.6 15	12.7	11.14	0.095	130	4.25	4.19
Ξ	3-Jul-00	926	Balanced	234	710	687	2,127	3.06	94.5	0.65	3656	42.6	25.3	49.0	45,207	0.289	30.5	9.5 4	3.1 19	1.3	11.06	0.093	119	4.14	4.07
Ξ	4-Jul-00	927	Balanced	234	710	663	2,124	3.05	76.5	0.65	3637	42.9	23.9	47.5	45,207	0.290	30.7	9.5 4	1.7 19	90.8	11.02	0.096	142	4.35	4.29
П	5-Jul-00	928	Balanced	234	710	667	2,116	3.31	84.8	0.65	3641	40.8	14.0	45.5	45,207	0.285	31.7	8.9 4	2.6 18	87.8	10.85	0.098	148	4.3	4.51
Ξ	8-Jul-00	931	Balanced	234	710	670	2,174	3.01	82.8	0.66	3731	44.4	26.4	46.5	45,207	0.279	29.1	8.6 4	2.0 19	1.4	11.06	0.098	140	4.52	4.55
11	9-Jul-00	932	Balanced	234	710	683	2,156	3.08	89.5	0.66	3691	42.7	26.3	49.5	45,207	0.284	30.1	8.9 4	2.6 19	12.3	11.12	0.093	133	4.15	4.25
Ξ	10-Jul-00	933	Balanced	234	710	677	2,143	3.04	89.5	0.66	3695	42.1	28.9	50.0	45,207	0.286	30.0	9.0 4	12.1 19	12.9	11.15	0.092	134	4.25	4.25
Ξ	11-Jul-00	934	Balanced	234	710	682	2,174	2.82	89.3	0.66	3725	40.5	18.3	48.5	45,207	0.279	28.0	8.9 4	2.6 19	02.0	11.09	0.094	133	4.82	4.64
Ξ	12-Jul-00	935	Balanced	234	710	675	2,169	2.85	88.1	0.66	3706	42.2	25.7	50.0	45,207	0.277	28.1	8.8	2.6 19	00.4	11.00	0.091	130	4.18	4.08
Ξ	13-Jul-00	936	Balanced	234	710	668	2,164	2.88	84.8	0.66	3712	43.7	25.2	47.0	45,207	0.274	28.1	8.6 4	2.5 18	8.9	10.92	0.096	141	4.55	4.41
Ξ	14-Jul-00	937	Balanced	234	710	679	2,057	3.24	6.69	0.64	3581	41.4	16.6	46.0	45,207	0.279	31.3	9.1 4	3.4 18	7.7	10.84	0.097	165	4.35	4.88
Ξ	16-Jul-00	939	Balanced	234	710	681	2,158	2.81	90.3	0.66	3716	42.8	25.9	49.0	45,207	0.276	27.6	8.7 4	2.9 15	0.5	11.01	0.093	135	4.36	4.24
Ξ	17-Jul-00	940	Balanced	234	710	699	2,150	2.94	88.0	0.66	3686	42.9	25.5	48.5	45,207	0.273	28.4	8.5 4	2.6 18	83	10.89	0.093	136	3.8	3.88
Ξ	18-Jul-00	941	Balanced	233	710	643	2,214	2.69	105.0	0.66	3984	41.2	21.5	46.0	42,278	0.265	24.1	6.9 4	4.2 17	14.6	10.80	0.091	133	4.55	4.49
Ξ	19-Jul-00	942	Balanced	233	710	614	2,220	2.82	95.3	0.65	4237	43.7	39.7	50.5	39,349	0.276	24.3	6.4 4	4.0 16	57.5	11.13	0.079	114	4.4	4.44
Ξ	20-Jul-00	943	Balanced	234	710	655	2,215	3.07	63.6	0.67	4094	42.1	29.4	48.5	41,549	0.316	29.8	8.8	0.8 15	02.6	12.10	0.095	123	4.41	4.27
= :	22-Jul-00	945	Balanced	235	710	676	2,169	3.53	40.7	0.66	3848	42.3	26.9	49.0	43,749	0.361	37.3	20.6 20.6	9.3 20 20)6.5 X	12.41	0.100	150	4.4	435
= =	00-INC-67	0476	Dalalicou	+67	017	0/0	1/1/7	04.0	0.71	10.0	0/00	t (25.1		01-10	0.000	0.00			0.0	20.21	101.0	12 5		0 0 0
= =	25-141-00	947	Balanced	+C7	017	713	2,172 2,165	40.0 80.6		0.67	3897	43.2	27.8	48.0	43 749	240.0	34.4	0.6	17 DI	7.0	12.55	0 104	130	3.98	3.87
Ξ	26-Iul-00	070	Balanced	234	710	787	2 122	3 34	57.2	0.68	3766	43.0	22.0	47.0	45 949	0 392	39.0	3.0	9.7 23	8.7.8	13 50	0 121	146	4 8	451
Ξ	27-Jul-00	950	Balanced	234	705	808	2,091	3.24	66.3	0.68	3737	43.8	25.3	47.5	45,949	0.392	38.4	3.3	0.3 24	10.7	13.67	0.121	144	4.72	4.53
Ξ	28-Jul-00	951	Balanced	234	705	812	2,096	3.16	70.6	0.68	3741	44.3	30.9	50.5	45,949	0.394	37.9	23.5 4	0.3 24	11.6	13.73	0.114	134	4.53	4.28
Ξ	29-Jul-00	952	Balanced	234	705	816	2,098	3.14	79.5	0.68	3742	44.3	27.5	48.0	45,949	0.392	37.5	23.3 4	0.7 24	10.5	13.67	0.119	137	4.46	4.18
Ξ	10-Aug-00	964	Balanced	234	706	622	2,147	4.69	48.3	0.64	3539	42.9	24.2	48.5	45,949	0.322	42.5	9.1 4	0.3 18	\$5.3	10.53	0.083	121	5.88	6.94
Ξ	11-Aug-00	965	Balanced	234	709	618	2,177	4.63	46.3	0.65	3570	43.4	26.5	49.0	45,949	0.317	41.6	8.8	0.4 18	3.5	10.51	0.082	119	5.87	6.86
Ξ	12-Aug-00	996	Balanced	234	701	641	2,147	4.32	47.1	0.65	3569	44.6	31.1	50.0	45,949	0.332	40.5	9.5 4	0.5 19	0.1	10.80	0.084	115	5.9	6.61
Ξ	13-Aug-00	796	Balanced	235	709	668	2,177	4.34	54.2	0.66	3649	44.5	28.7	48.5	45,949	0.331	41.0	9.8 4	0.7 15	1.7	11.19	0.087	123	6.05	6.66
Π	14-Aug-00	968	Balanced	235	705	708	2,229	3.62	52.9	0.69	3767	44.3	28.1	48.5	45,949	0.342	37.0	20.7 4	0.2 21	1.0	11.98	0.090	123	6.78	6.46
Ξ	15-Aug-00	696	Balanced	235	705	750	2,171	3.41	67.4	0.69	3772	44.6	27.5	47.5	45,949	0.354	36.5	21.5 4	0.6 22	21.6	12.58	0.096	121	6.78	6.43
Ξ	16-Aug-00	026	Balanced	234	705	760	2,176	3.34	74.0	0.68	3758	44.8	27.2	47.0	45,949	0.365	36.9	2.1 4	0.4 22	25.5	12.81	0.098	125	6.9	6.38

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

						Tal	ble D-	2 – D:	ata Su	mmar	y Tab	le for	King	sport	Cataly	st Ca	mpai	gn 2 (cont'	(p					
		Č				Econde	Domila	Decetor	Duese	In lot Cun	Canada	Chunnel	000	Comod	Cottaline	Cotoline	S	Reactor O.T. M	Cumano	Raw	Cotoline	Doctor	U	Canadae	Caseson
		On	Gas	Temp	Pres.	Feed	Gas	Feed	Furge Gas	In let Sup. Velocity	Space Velocity	Sturry Cone.	Holdup	Slurry	Latalyst Inventory	Age	Conv.	U-1-M Conv.	syngas Util.	Prod.	Catalyst MeOH Prod.	Vol. Prod.	Overall (Btu	sparger dP	sparger Resist.
Case	Date	Stream	Type	(Deg C)	(psig)	(KSCFH)	(KSCFH)	(H2:CO)	(KSCFH)	(ft/sec)	(l/hr-kg)	(wt% ox)	(%lov)	Hgt (ft)	(lb)	(eta)	(%)	(%)	(SCF/lb)	(TPD)	(gmol/hr-kg)	(TPD/ft3)	hr ft2 F)	(isd)	("K")
Ξ	17-Aug-00	179	Balanced	234	705	760	2,138	3.49	78.9	0.68	3730	44.7	26.8	46.5	45,949	0.364	37.9	22.0	40.9	223.0	12.67	0.098	124	6.64	6.35
Ξ	18-Aug-00	972	Balanced	234	705	735	2,137	3.50	78.4	0.67	3667	44.9	26.5	46.5	45,949	0.348	37.2	21.5	41.1	214.4	12.19	0.094	121	6.56	6.47
Ξ	20-Aug-00	974	Balanced	234	705	743	2,194	3.14	64.5	0.69	3767	44.4	29.1	49.0	45,949	0.354	34.7	21.6	40.0	223.0	12.66	0.098	117	6.98	6.36
Ξ	21-Aug-00	975	Balanced	234	705	739	2,190	3.29	76.9	0.68	3757	44.2	31.5	51.0	45,949	0.342	34.8	21.0	41.1	216.1	12.27	0.095	110	6.85	6.38
Ξ	22-Aug-00	976	Balanced	234	705	735	2,168	3.33	79.6	0.68	3736	44.6	33.2	51.5	45,949	0.340	35.0	20.9	41.3	213.8	12.19	0.094	109	6.73	6.39
= :	23-Aug-00	776	Balanced	234	708	727	2,172	3.09	76.7	0.68	3734	45.7	29.7	47.0	45,949	0.321	32.6	20.6	41.0	212.9	12.10	0.094	107	6.77	6.46
= =	24-Aug-00	978	Balanced	234	210	6/0	2,168	3.68	1.44.1	0.66	3635	9.44 2.05	25.8	46.0	45,949	0.332	37.4	20.4	39.4	203.8	11.57	160.0	121	6.28	6.52
= =	25-Aug-00 26-Aug-00	979	Balanced	234 234	710	0C0 655	2,188	3.55	40.0 46.1	0.06	3638 3638	42.5	23.8 23.8	49.0 48.0	45,949 45.949	0.315	35.3	19.8 19.8	39.6 39.8	197.3	11.28	0.098	1126	6.24 6.32	0.48 6.42
Ξ	27-Aug-00	981	Balanced	234	710	655	2,168	3.45	49.0	0.66	3637	43.8	29.7	50.5	45,949	0.302	33.7	19.7	40.0	196.6	10.93	0.087	105	6.41	6.40
Ξ	28-Aug-00	982	Balanced	234	707	692	2,166	3.12	63.4	0.66	3661	44.0	34.9	54.0	45,949	0.314	32.6	20.3	40.5	205.0	11.65	0.090	107	6.72	6.52
11	29-Aug-00	983	Balanced	234	706	732	2,131	3.10	82.8	0.66	3659	44.0	30.2	50.5	45,949	0.321	32.9	20.6	41.7	210.5	11.97	0.099	114	6.55	6.54
Ξ	30-Aug-00	984	Balanced	234	706	738	2,116	3.16	84.0	0.66	3648	42.7	26.0	50.0	45,949	0.326	33.8	21.0	41.6	212.9	12.10	0.101	123	6.46	6.51
Ξ	31-Aug-00	985	Balanced	235	706	730	2,111	3.20	77.6	0.66	3622	42.0	23.2	49.5	45,949	0.326	34.5	21.2	41.2	212.5	12.08	0.102	128	6.39	6.49
Ξ	1-Sep-00	986	Balanced	235	706	726	2,123	3.19	76.5	0.66	3634	42.1	22.7	49.0	45,949	0.326	34.3	21.1	41.0	212.4	12.07	0.094	118	6.45	6.45
Ξ	2-Sep-00	786	Balanced	235	706	725	2,123	3.09	75.7	0.66	3647	42.4	23.6	49.0	45,949	0.327	33.7	21.1	41.1	211.6	12.13	0.092	116	6.64	6.39
Ξ	3-Sep-00	988	Balanced	234	706	726	2,138	3.13	85.2	0.66	3654	42.9	25.7	49.5	45,949	0.316	33.1	20.8	41.8	208.4	11.85	0.100	124	6.58	6.37
Ξ	4-Sep-00	686	Balanced	234	706	726	2,127	3.17	96.6	0.66	3651	43.5	28.1	50.0	45,949	0.308	32.7	20.4	42.5	204.7	11.64	0.098	118	6.49	6.43
Ξ	5-Sep-00	066	Balanced	234	706	735	2,147	3.12	104.6	0.67	3694	43.7	29.5	50.5	45,949	0.303	31.8	20.1	43.1	204.4	11.63	0.096	114	6.58	6.41
Ξ	6-Sep-00	166	Balanced	235	709	705	2,169	3.27	77.0	0.67	3708	41.1	19.6	49.0	45,949	0.304	33.1	20.1	41.4	204.2	11.60	0.099	137	7.58	6.45
Ξ	7-Sep-00	992	Balanced	235	707	705	2,156	3.17	81.0	0.67	3696	41.4	25.9	52.5	45,949	0.304	32.6	20.2	41.6	203.1	11.55	0.092	128	6.68	6.43
Ξ	8-Sep-00	993	Balanced	235	707	697	2,156	3.15	79.4	0.67	3670	41.4	25.9	52.5	45,949	0.298	31.9	19.9	41.7	200.6	11.41	0.091	127	6.68	6.45
Ξ	9-Sep-00	994	Balanced	235	707	969	2,140	3.13	84.5	0.66	3642	42.1	24.1	50.0	45,949	0.292	31.5	19.8	42.2	198.0	11.19	0.094	125	6.63	6.42
Ξ	10-Sep-00	995	Balanced	235	707	692	2,151	3.11	88.1	0.66	3637	42.5	24.7	49.5	45,949	0.290	31.2	19.6	42.4	195.6	11.14	0.094	123	6.64	6.44
=	11-Sep-00	966	Balanced	235	707	688	2,156	3.06	89.7	0.66	3646	42.8	25.5	49.5	45,949	0.287	30.5	19.4	42.6	194.0	11.04	0.093	121	6.67	6.36
= :	12-Sep-00	266	Balanced	234	708	684	2,157	3.01	91.7	0.66	3641	43.4	27.9	50.0	45,949	0.283	29.8	19.2	42.8	191.7	10.91	0.091	116	6.72	637
= :	13-Sep-00	866	Balanced	234	807	(89	2,149	2.98	7.19	0.66	3640	43.4	27.8	50.0	45,949	0.284	29.8	19.3	42.8	192.7	10.97	0.092	811	6.71	6.33
= =	15 Sep-00	0001	Balanced	462 234	710	660 666	2,140 2 198	20.5 3.08	0.00	0.00	2605	4.64 2.4.5	321	6.UC	45,040	0.274	4.62 4.00	18.7	42.6	5.061	10.67	0.088	c11 113	20.0	10.0
=	16-Sen-00	1001	Balanced	234	710	661	2,229	3.08	91.7	0.67	3697	44.8	33.0	51.0	45,949	0.266	28.7	18.3	42.9	184.8	10.51	0.086	Ξ	6.61	6.35
Ξ	17-Sep-00	1002	Balanced	235	710	675	2,180	3.02	85.7	0.66	3669	42.4	25.8	50.5	45,949	0.271	29.2	18.9	42.6	190.1	10.81	0.090	124	6.60	6.39
= =	19-Sep-00	1004	Balanced	234 734	710	599 570	2,240 2.186	2.87	9.111 9.001	0.66 0.64	4614 4504	39.3 30.4	24.3 76.3	44.0 45.0	36,163 36.163	0.273	23.3 73.7	15.7	45.6 45.3	157.7	11.41	0.086	142	6.54 6.34	6.45 6.30
	00-de-07	C001	Balanced	235	017	6.44	2. 140	69 5	X6.4	0.64	4748	6.65	015	0.05	101 11	0 317	0.55	18.4	42.8	977	2	580.0	49	5 44	10.0
=	00-cpc-22	1012	Balanced	235	206	817	2.126	3.89	62.0	0.69	3859	42.2	32.8	55.0	44.963	0.458	45.7	24.3	40.4	242.8	14.10	0.105	174	5.89	5.34
Ξ	28-Sep-00	1013	Balanced	235	706	851	2,095	3.52	65.7	0.69	3866	41.4	30.1	54.5	44,963	0.458	43.6	24.9	40.0	255.5	14.84	0.112	160	5.81	5.34
F	30-Sep-00	1015	Balanced	230	206	9/9	2,122	4.57	33.1	0.64	3636	41.8	26.8	51.0	44,963	0.451	47.8	21.8	39.0	208.2	12.30	160.0	167	4.86	5.64
Ξ	2-Oct-00	1017	Balanced	235	206	811	2,422	3.51	46.8	0.68	3838	40.2	27.4	55.0	44,963	0.465	43.7	25.0	38.9	250.4	14.53	0.108	148	5.87	5.27
Ξ	3-Oct-00	1018	Balanced	235	706	832	2,091	3.39	53.1	0.68	3845	42.2	27.5	51.0	44,963	0.464	42.9	25.1	39.2	254.6	14.77	0.119	155	5.90	5.30
Ξ	4-Oct-00	1019	Balanced	235	706	856	2,088	3.48	59.7	0.69	3849	42.1	27.9	51.5	44,963	0.476	44.4	25.6	39.5	260.1	15.09	0.120	158	5.76	5.27
Ξ	5-Oct-00	1020	Balanced	235	705	867	2,027	3.65	67.5	0.67	3780	39.4	29.0	54.0	44,963	0.491	46.9	26.2	39.7	262.5	15.23	0.116	156	5.52	5.40
Ξ	6-Oct-00	1021	Balanced	235	705	116	2,071	3.49	67.8	0.70	3715	45.4	33.9	52.0	47,163	0.492	46.3	26.8	39.6	275.7	15.25	0.126	167	5.86	5.29
= :	7-Oct-00	1022	Balanced	235	705	934	2,068	3.60	73.9	0.70	3726	44.7	31.0	51.0	47,163	0.501	47.7	27.0	39.9	281.3	15.55	0.131	172	5.63	5.30
= =	8-Oct-00 9-Oct-00	1023 1024	Balanced Balanced	235 235	705	925 900	2,057 2,057	3.55 3.73	80.2 79.0	0.69 0.68	3710 3680	43.3 43.9	32.0 33.5	54.5 54.5	47,163 47,163	0.485 0.468	46.6 47.1	26.6 26.0	40.2 40.1	275.8 269.0	15.27 14.88	0.120 0.117	144 137	5.63 5.37	5.30 5.42

D-21

						Tab	ole D-	2 – Da	ita Su	mmary	/ Tabl	e for l	Kings	port (Cataly	st Cai	npaig	gn 2 (c	ont'd						
		Davs				Fresh	Recvcle	Reactor	Purge	Inlet Sup.	Space	Slurry	Gas	Gassed	Catalvst	Catalvst	8	ceactor D-T-M S	F Muneral M	taw leOH (Catalvst	Reactor	U Overall	Sparger	Sparger
(ő	Gas	Temp	Pres.	Feed	Gas	Feed	Gas	Velocity	Velocity	Conc.	Holdup	Slurry	Inventory	Age	Conv.	Conv.	190	rod. Me	OH Prod.	Vol. Prod.	(Btu	ਿਚ	Resist.
Case	Date	Stream	Type	(Deg C)	(big)	(KSCFH)	(KSCFH)	(H2:CO)	(KSCFH)	(ft/sec)	(l/hr-kg)	(wt% ox)	(vol%)	Hgt (ft)	(ql)	(eta)	(%)	(%) (S)	CF/lb) (1	(PD) (gn	nol/hr-kg)	(TPD/ft3)	hr ft2 F)	(isd)	("K")
Ξ	10-Oct-00	1025	Balanced	235	709	916	2,038	3.63	67.5	0.69	3682	44.5	33.8	53.5	47,163	0.494	47.8	26.8	39.6 2	77.6	15.36	0.123	151	5.42	5.35
Ξ	11-Oct-00	1026	Balanced	235	707	116	2,053	3.46	64.9	0.69	3709	44.4	27.9	49.5	47,163	0.487	45.9	26.7	39.5 2	76.9	15.32	0.133	157	5.63	5.26
11	12-Oct-00	1027	Balanced	235	709	928	2,074	3.41	66.3	0.70	3742	44.8	31.8	51.5	47,163	0.491	45.6	26.8	39.6 2	81.1	15.55	0.130	155	5.85	5.41
11	13-Oct-00	1028	Balanced	235	707	918	2,085	3.31	68.1	0.70	3764	44.3	33.3	53.5	47,163	0.480	44.1	26.5	39.4 2	79.4	15.46	0.124	140	5.87	5.29
Ξ	14-Oct-00	1029	Balanced	234	706	922	2,083	3.30	73.8	0.70	3745	44.3	33.2	53.5	47,163	0.474	43.9	26.6	39.6 2	79.2	15.44	0.124	136	6.05	5.51
11	15-Oct-00	1030	Balanced	234	708	920	2,070	3.43	81.3	0.69	3729	44.1	27.5	50.0	47,163	0.476	45.1	26.6	39.9 2	76.5	15.31	0.132	145	5.83	5.40
11	23-Oct-00	1038	Balanced	234	706	865	2,103	3.27	76.8	0.69	3717	44.1	30.1	51.5	47,163	0.430	41.0	24.9	40.1 2	59.0	14.33	0.120	124	5.69	5.14
= =	25-Oct-00	1040	Balanced	238 734	710	993 888	1,906 2.008	3.35	140.8 70.3	0.68	3622 3742	45.2 44.4	36.9 26.5	55.0 48.5	47,163 47 163	0.445	45.2 41.0	27.0 25.4	43.1 2	76.6	15.34 14.68	0.120	117	5.3 5.07	5.19
=	26-Oct-00 30-Oct-00	1041	Balanced	234	710	792	2.248	3.31	54.9	0.71	3824	45.5	34.4	52.0	47.163	0.393	38.2	23.0	39.3 2	41.9	13.37	0.1110	123	6.18	5.18
Ξ	3-Nov-00	1049	Balanced	234	710	786	2,178	3.57	59.0	0.69	3703	42.1	24.4	51.5	47,163	0.385	40.4	23.1	39.9 2	36.3	13.08	0.109	127	5.67	5.32
11	4-Nov-00	1050	Balanced	234	710	788	2,189	3.49	59.8	0.69	3705	44.6	30.6	51.0	47,163	0.395	40.5	23.5	39.5 2	39.3	13.25	0.112	131	5.84	5.32
11	5-Nov-00	1051	Balanced	234	710	162	2,191	3.45	63.1	0.69	3712	45.6	33.3	51.0	47,163	0.393	39.9	23.4	39.7 2	38.9	13.22	0.112	127	5.81	5.26
11	7-Nov-00	1053	Balanced	234	710	786	2,146	3.56	60.1	0.68	3658	44.1	27.9	50.0	47,163	0.400	41.6	23.7	39.4 2	39.4	13.25	0.114	139	5.68	5.34
11	8-Nov-00	1054	Balanced	234	705	775	2,095	3.41	55.9	0.67	3605	44.2	28.8	50.5	47,163	0.394	40.3	23.7	39.5 2	35.2	13.02	0.111	134	5.80	5.41
11	9-Nov-00	1055	Balanced	234	705	782	2,118	3.61	65.0	0.68	3616	44.9	28.6	49.0	47,163	0.391	41.6	23.5	39.8 2	35.5	13.03	0.115	136	5.61	5.48
Ξ	10-Nov-00	1056	Balanced	236	705	800	2,162	3.23	62.3	0.70	3726	45.6	33.2	51.0	47,163	0.381	38.2	23.4	39.7 2	41.6	13.37	0.113	126	5.98	5.26
Ξ	11-Nov-00	1057	Balanced	234	705	792	2,148	3.21	61.8	0.69	3708	45.0	30.3	50.0	47,163	0.385	37.7	23.1	39.9 2	38.4	13.20	0.114	130	5.94	5.26
= :	12-Nov-00	1058	Balanced	234	705	792	2,143	3.11	59.0	0.69	3698	44.6	28.5	49.5 51.0	47,163	0.387	37.1	23.3	39.6 2	40.0	13.28	0.116	129	6.02 5.02	5.31
= =	13-Nov-00	6601	Balanced	234	207	800	20172	3.49	0.00 78.0	0.69	3683	40.7	1.05	0.15	47,163	0 376	2.05 30.3	23.0	40.7 2	36.2	13.06	0110	11/	553	5.44
17	17-Nov-00	1063	Balanced	235	704	126	2,350	1.08	62.4	0.74	3958	46.3	35.7	6.16 C.16	47,163	0.403	16.8	20.3	39.6 2	202	12.19	0.102	122	9.94	4.91
	00-A0NI-/ T	C001	Ralanced	AVC.	12.1	0101	CUD	5.6 V	L CL	0.66	0952	1.51	0.04	2.07	291 11	042.0	101	1.51	6 D L V	5 5 5	1114	2.61.0	2.7.1	9L.V	10.5
= =	21-Nov-00 25-Nov-00	1067	Balanced	234 234	705	650	2,149	4.12	36.3	0.65	3471	44.5	29.9 31.0	51.5	47,103 47,163	0.345	41.3	21.0	38.8 2	600	14.14	0.093	134	4./0	5.39
11	26-Nov-00	1072	Balanced	235	705	650	2,169	3.88	31.6	0.65	3499	45.1	30.5	50.0	47,163	0.347	40.0	21.3	38.3 2	03.4	11.24	0.097	138	5.09	5.33
11	27-Nov-00	1073	Balanced	234	701	693	2,155	3.33	39.4	0.68	3599	43.3	25.7	50.0	47,163	0.340	35.9	21.5	38.9 2	13.9	11.83	0.102	129	5.45	5.21
Ξ	28-Nov-00	1074	Balanced	234	701	704	2,141	3.52	47.8	0.67	3553	43.6	28.6	51.5	47,163	0.347	37.9	21.8	39.2 2	15.6	11.92	0.100	125	5.17	5.28
11	30-Nov-00	1076	Balanced	236	402	669	2,114	3.22	47.6	0.66	3556	41.4	21.6	51.0	47,163	0.346	41.1	21.7	38.9 2	15.7	11.93	0.101	123	5.07	5.39
Π	1-Dec-00	1077	Balanced	236	710	869	2,182	3.68	37.7	0.67	3618	44.2	28.7	50.5	47,163	0.346	39.4	21.5	38.6 2	17.1	12.01	0.102	122	5.27	5.22
= :	2-Dec-00	1078	Balanced	236	710	200	2,194	3.41	34.2	0.68	3642	44.3	28.3	50.0	47,163	0.348	37.6	21.7	38.3	19.3	12.12	0.104	124	5.6	5.19
: =	4-Dec-00	1080	Balanced	236	710	107	2.217	3.31	444	0.09	3671	64	29.9	50.0	47.163	0.332	35.9	21.4	39.1 2	181	12.07	0.104	118	5.63	5.19
Ξ	5-Dec-00	1081	Balanced	234	710	686	2,231	3.28	43.8	0.68	3666	43.9	28.9	51.0	47,163	0.325	34.5	20.8	39.0 2	11.2	11.68	0.099	119	5.56	5.14
11	6-Dec-00	1082	Balanced	235	710	701	2,190	3.51	49.3	0.67	3629	43.1	30.1	53.5	47,163	0.333	37.0	21.2	39.3 2	14.0	11.84	0.095	122	5.3	5.17
Π	7-Dec-00	1083	Balanced	235	710	714	2,121	3.59	51.0	0.66	3561	42.4	26.7	52.5	47,163	0.342	38.6	21.7	39.9 2	15.0	11.90	0.097	131	5.18	5.24
11	8-Dec-00	1084	Balanced	235	710	701	2,200	3.54	49.7	0.67	3631	42.8	27.3	52.0	47,163	0.328	36.8	21.1	39.3 2	13.8	11.82	0.098	125	5.27	5.28
Ξ	9-Dec-00	1085	Balanced	234	710	101	2,139	3.43	49.3	0.66	3576	43.1	28.7	52.5	47,163	0.337	37.0	21.6	39.2 2	14.3	11.86	0.097	126	5.3	5.20
Ξ	10-Dec-00	1086	Balanced	234	710	700	2,191	3.36	48.3	0.67	3616	43.1	27.3	51.5	47,163	0.326	35.6	21.0	39.6 2	12.4	11.75	0.098	123	5.36	5.18
Ξ	11-Dec-00	1087	Balanced	235	710	741	2,108	3.37	60.7	0.66	3559	43.9	31.5	53.0	47,163	0.351	37.9	22.4	39.8 2	23.3	12.36	0.100	138	5.29	5.26
Ξ	12-Dec-00	1088	Balanced	235	710	758	2,167	3.25	62.4	0.68	3657	44.3	33.7	54.0	47,163	0.342	36.2	22.1	40.0 2	27.5	12.58	0.100	133	5.47	5.32
Ξ	13-Dec-00	1089	Balanced	235	710	750	2,147	3.10	62.4	0.67	3623	44.8	31.2	51.0	47,163	0.340	35.1	22.1	40.1 2	24.6	12.43	0.105	137	5.49	5.17
11	14-Dec-00	1090	Balanced	235	710	749	2,921	3.01	62.3	0.68	3659	45.5	31.7	50.0	47,163	0.332	33.6	21.7	40.4 2	22.7	12.32	0.106	135	5.66	5.22
= :	15-Dec-00	1001	Balanced	235	710	735	2,195	3.06	60.6	0.68	3663	45.5	31.6	50.0	47,163	0.325	33.5	21.4	40.0	20.5	12.20	0.105	131	5.59	5.24
= :	18-Dec-00	1094	Balanced	234	710	743	2,148	3.20	82.7	0.67	3620	44.7	31.5	51.5	47,163	0.316	33.8	21.1	41.3	15.8	11.94	0.100	115	5.12	5.17
Ξ	19-Dec-00	1095	Balanceu	234	710	745	2,138	3.04	79.0	0.67	3620	43.8	31.2	53.0	47,163	0.319	33.0	21.3	41.0 2	17.4	12.03	0.098	116	5.42	5.25

	Sparger	Resist. ("K")	5.38	5.32	5.28	5.35	5.31	5.28	5.29 5.31	5.25	5.16	5.12	5.80	5.54	5.41	5.33	5.40	5.36	5.65	5.23	5.25	5.25	5.21	5.75	14:0	72.5 25.3	5.16	5.23	5.38	5.33	5.42	5.25 5.18	5.28	5.19	5.31	5.50	5.27	5.47	5.42	5.40	5.43	87°C	5.60 5.60
	Sparger	db (isd)	5.04	5.1	5.44	5.19	5.33	5.42	5.23 5.41	541	5.57	5.37	4.55	4.79	4.87	4.99	5.97	5.62	6.14	5.38	5.51	5.53	5.32	5.45	56.C	00°.0	5.52	5.8	5.95	5.7	5.94	5.57 5.59	5.48	5.5	5.48	5.92	5.88	5.68	5.55	5.48	5.49	1.0	6.02
	U Overall	(Btu hr ft2 F)	129	134	135	139	135	141	128	118	124	129	154	142	151	129	136	131	128	135	137	137	131	130	101	121	121	124	129	126	131	130 128	112	126	128	107	130	128	133	132	134	151	122
	Reactor	Vol. Prod. (TPD/ft3)	0.100	0.102	0.096	0.099	0.099	0.100	0.097 0.092	0 097	0.094	0.095	0.088	0.085	0.088	0.110	0.106	0.102	0.105	0.109	0.111	0.119	0.119	0.120	611.0	0.126	0.123	0.116	0.119	0.118	0.121	0.121 0.120	0.109	0.124	0.110	0.083	0.095	0.097	0.102	0.099	0.101	0.005	0.097
	Catalvst	MeOH Prod. (gmol/hr-kg)	12.26	12.62	11.82	11.66	11.54	11.69	11.42 10.79	11.24	11.08	11.11	10.99	10.93	11.33	14.20	14.50	13.67	13.99	14.03	13.90	14.50	14.27	14.20	14.15	14.30	14.33	14.31	14.36	14.03	13.63	13.48 13.09	11.62	12.32	12.73	10.29	11.49	11.34	11.46	11.42	11.27	10.11	11.09
(p	Raw MeOH	Prod. (TPD)	221.8	228.3	213.7	210.8	208.7	210.5	206.5 195.1	203.1	200.3	200.8	161.6	160.6	166.5	244.8	262.3	247.2	253.1	253.7	251.5	262.2	257.4	256.5	4.05C	0.4C2	259.1	258.8	259.7	253.7	246.5	243.8 236.6	210.2	222.8	230.2	185.8	207.7	204.9	207.1	206.4	203.7	0.661	200.5
cont'e	Syngas	Util. (SCF/lb)	40.4	40.6	41.1	41.0	41.4	41.1	41.5 41.8	42.7	44.1	43.4	43.7	44.1	43.7	41.5	39.3	39.4	40.0	39.7	38.6	39.7	40.5	41.4	C.14	40.8	41.3	41.1	41.2	41.9	40.5	40.9 41.9	42.0	42.0	41.3	42.7	42.3	43.4	43.4	43.9	44.2	44.4 15.0	42.9
;n 2 (teactor D-T-M	Conv. (%)	21.5	22.0	20.9	20.8	20.3	20.6	20.3 19.0	20.2	19.2	19.1	16.5	16.1	16.6	23.8	25.5	24.8	24.6	24.7	24.5	25.7	25.2	25.0	24.8	24.9 30 1	24.8	25.4	25.6	24.8	24.7	24.1 24.1	22.6	21.8	23.3	19.3	21.3	20.5	20.6	20.7	20.5 17 8	17.8	19.8
npaig	202	Conv. (%)	36.3	37.0	31.5	32.8	31.4	30.7	31.1 28.6	30.5	27.4	28.3	28.3	26.2	26.0	42.4	42.3	42.7	40.0	41.4	40.9	41.7	42.0	43.0	1.04	41.0	38.7	38.8	41.2	38.4	37.4	38.0 37.4	32.3	33.6	37.7	32.0	34.2	32.9	33.0	33.8	33.5 21 E	31.5 316	29.3 29.3
st Cal	Catalvst	Age (eta)	0.330	0.335	0.299	0.291	0.294	0.298	0.290 0.265	0.281	0.266	0.267	0.272	0.262	0.270	0.401	0.433	0.412	0.403	0.404	0.405	0.426	0.421	0.415	CU4.U	0 384	0.383	0.397	0.405	0.391	0.386	0.382 0.372	0.304	0.329	0.356	0.271	0.311	0.292	0.294	0.296	0.292	0.284	0.277
Catalys	Catalvst	Inventory (Ib)	47,163	47,163	47,163	47,163	47,163	47,163	47,163 47,163	47,163	47,163	47,163	38,360	38,360	38,360	44,960	47,160	47,160	47,160	47,160	47,160	47,160	47,160	47,160	47,160	47,160 47,160	47,160	47,160	47,160	47,160	47,160	47,160 47.160	47,160	47,160	47,160	47,160	47,160	47,160	47,160	47,160	47,160	47,160 47 160	47,160
port (Gassed	Slurry Hgt (ft)	53.0	53.5	53.0	50.5	50.0	50.0	50.5 50.5	50.0	50.5	50.5	44.0	45.0	45.0	53.0	59.0	57.5	57.5	55.5	54.0	52.5	51.5	51.0	0.1C	0.05	50.0	53.0	52.0	51.0	48.5	48.0 47.0	46.0	43.0	50.0	53.0	52.0	50.5	48.5	49.5	48.0	48.U 18.5	49.0
Kings	Gas	Holdup (vol%)	32.2	33.4	34.2	35.5	29.6	33.5	30.2 23.5	24.1	31.6	31.2	29.4	28.6	30.1	21.6	23.4	30.3	30.1	27.6	26.7	25.1	30.7	23.3	16.4	10./ 29.1	30.6	24.5	27.5	26.3	28.8	25.2 29.4	22.6	18.0	22.7	23.4	24.2	25.6	21.7	18.4	26.0 17.8	1/.8	14.0
for F	Slurry	Conc. wt% ox)	43.8	44.4	44.9	46.8	44.7	46.2	44.7 42.3	42.7	45.2	45.1	42.6	41.8	42.3	39.2	38.2	41.3	41.2	41.2	41.6	41.8	44.3	42.0	40.5	40.1	45.1	41.4	42.9	43.0	45.2	44.2 46.3	44.4	44.7	42.3	41.0	41.8	43.0	42.8	41.1	44.5	41.8	40.1
Table	Space	/elocity l/hr-kg)	3633	3668	3613	3584	3623	3607	3592 3624	3582	3682	3703	4196	4309	4326	3834	3721	3632	3714	3684	3693	3666	3634	3625	1705	3707	3719	3652	3641	3642	3597	3558 3555	3633	3641	3559	3533	3578	3586	3584	3573	3561	35/0	3589 3589
nary '	t Sup.	locity /	.67	.68	.67	.67	.67	.67	.67 167	99	.68	.68	.63	.65	.65	.68	69'	.67	69	.68	.69	.68	.67	.67	/0/	60. 09	69	.68	.68	.68	.67	.66 .66	/.91	.67	.66	.66	.66	.67	.67	.66	.66	66	.00 167
Sumn	ge	IS Ve	5 0	.4	0	.6 0	.5 (.6 (9.0	2 0		0.0	3 (7 0	9 9	0.0 0	.1 0	3	2	6	0	5	8	0.0	<i>.</i>	ن م م	99	.6 (8.	0.1 0	7 0	5 C 5 C	2 (0	.8 (.7 C	0 0	.8	4 0	0.0	5	2 2 2 2	. o
Data	or Pur	d (KSC	4 71	4 77	3 75	4 73	5 83	8 76	2 80 5 29	103	11	107	4 92	2 90	5 96	2 100	3 61	1 57	60	0 65	5 49	5 67	4 89	100	70 IO	2 X	102	2 90	8 94	8 100	8 72	888	3 78	5 89	5 74	5 73	3 86	4 101	2 99	2 115	101	21 I I I I I I I I I I I I I I I I I I I	
-7-]	le Reac	Fee H) (H2:C	3.4	1 3.4	3.3	3.1	3.0	3.2	0.0 0.0 0.0	2.9	5 2.7	2.9	3.5	3.3	3.1	3.7	3.3	3.5	3.2	3.4	3.3	7 3.2	3.3	3.5		7.0	3.1	3.0	2.9	3.0	7 2.9	3.0	7.1	3.0	3.2	3.3	3.2	3.2	3.2	1 3.3	5. 	1.0	3.2
ıble I	Recve	Gas (KSCF	2,113	2,162	2,146	2,143	2,168	2,158	2,156 2,218	2.112	2,196	2,232	2,141	2,226	2,218	2,088	2,12(2,113	2,129	2,11(2,13(2,067	2,048	2,02	2,015	2,085	2,067	2,083	2,052	2,061	2,077	2,069	2,22	2,183	2,118	2,22(2,186	2,182	2,182	2,162	2,152	2,156	2,186
Ţ	Fresh	Feed (KSCFH	747	772	731	720	720	721	714 679	723	735	726	589	590	909	847	859	811	833	838	808	867	869	884	\$84 000	688 883	68	886	168	886	823	832 827	736	781	792	662	733	742	749	755	736	151	716 71
		Pres. (psig)	710	710	710	710	710	710	710	710	715	715	715	715	714	710	710	710	710	710	710	710	710	710	01/	710	710	710	710	710	710	710	710	710	710	607	710	710	710	710	710	710	708
		Temp (Deg C	235	235	235	236	234	234	234 235	235	235	235	234	234	234	235	235	235	235	235	235	235	234	234	462	236	236	235	234	234	234	234 234	234	233	234	235	235	235	235	235	235	234	235
		Gas Type	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced
	Days	On Stream	1097	1098	1118	1119	1120	1121	1122	1126	1129	1131	1133	1134	1135	1140	1146	1147	1148	1149	1150	1152	1153	1154	5611	1150	1158	1159	1160	1161	1166	1167	1172	1173	1174	1179	1181	1183	1184	1185	1186	1187	1189
		Date	21-Dec-00	22-Dec-00	11-Jan-01	12-Jan-01	13-Jan-01	14-Jan-01	15-Jan-01	10-1an_01	22-Jan-01	24-Jan-01	26-Jan-01	27-Jan-01	28-Jan-01	2-Feb-01	8-Feb-01	9-Feb-01	10-Feb-01	11-Feb-01	12-Feb-01	14-Feb-01	15-Feb-01	16-Feb-01	17-Feb-01	10 Ech 01	20-Feb-01	21-Feb-01	22-Feb-01	23-Feb-01	28-Feb-01	1-Mar-01	6-Mar-01	7-Mar-01	8-Mar-01	13-Mar-01	15-Mar-01	17-Mar-01	18-Mar-01	19-Mar-01	20-Mar-01	21-Mar-01	22-Mar-01 23-Mar-01
		Case	=	Ξ	Ξ	Ξ	Ξ	Ξ	= =	=	: =	Ξ	Ξ	Ξ	Ξ	=	Ξ	Ξ	Ξ	Ξ	Ξ	Ξ	Ξ	= :	= =	= =	: =	Ξ	Ξ	Ξ	Ξ	= =	=	Ξ	=	F	=	Ξ	Ξ	=	= =	= =	: =

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

Sparger Resist. ("K")	5.46	5.79	5.33	7.96	7.43	9.55	6.78	5.89	6.34	0.95 7.48	7.37	6.74	7.26 7.00	6.98	7.04	6.30	6.67	6.44	6.21	6.59	6.08 6.43	6.53	6.84	6.82	6.74	0.92 6.67	6.74	7.12	6.96	6.64	7.13	4.7 103	6.90	6.49	6.66 7.07	10.1	0.04 6.46	7.13	7.30
Sparger dP (psi)	6.02	6.43	5.81 6.05	8.20	8.14	9.3	6.05	5.62	6.18	7 T	6.19	7.02	6.77 6.56	6.43	5.98	5.84	6.6	5.99	6.01	6.67	6.11	6.54	6.8	6.7	6.7	6.68	6.8	7.17	6.82	6.49	6.99 7.01	6.94	6.54	6.27	6.32	c/.0	1.1	7.0	7.86
U Overall (Btu hr ft2 F)	132	124	130 124	108	102	105	106	107	126	105	118	66	103 99	104	104	105	109	103	26	103	110	119	121	117	118	c11 221	152	118	185	195	160 202	159	185	187	190	102	161	147	141
Reactor Vol. Prod. (TPD/ft3)	0.102	0.101	0.104	0.075	0.085	0.076	0.075	0.075	0.083	0.000	0.086	0.082	0.082 0.08	0.072	0.072	0.074	0.076	0.074	0.072	0.076	0.080	0.08	0.08	0.08	0.08	0.073	0.073	0.072	0.071	0.070	0.068	0.0.68	0.068	0.068	0.069	0.000	0.096	0.075	0.085
Catalyst OH Prod. nol/hr-kg)	11.65	11.48	11.55 11.17	9.03	10.04	8.31	8.68	8.82	9.07	9.5 10.35	10.33	9.79	9.55 9.42	8.8	9.16	9.16	9.4	9.16	9.16	9.32	9.52 0.10	9.33	9.36	9.35	9.36	9.16 9.28	9.19	9.06	8.91	8.80	8.27	8.23	8.23	8.17	8.16 7.01	16.1	10.06	7.70	8.99
taw leOH rod. M (g) (g)	10.5	07.5	08.8 02.0	61.7	1.67	48.5	54.8	38.6	42.6	0.10	79.9	70.8	66.5 64.2	53.5	59.7	59.8	64.1	59.9	59.8	62.7	66.2 60.4	62.9	63.5	63.2	63.4	6.19 6.19	60.5	58.1	55.5	53.5	44.4 16.1	43.6	43.6	42.4	42.3	0.00	90.8 83 0	41.0	64.8
F mgas M Jtil. P CF/lb) (1	12.4 2	12.7 2	12.8 2	1 200	1.5 1	44 1	16.7 1	1 16.4	1 18.1	1 7.01	1 1.7 1	12.7	H.1 1 H.9 1	13.4 1	13.8 1	13.3 1	12.7 1	13.6 1	13.5 1	12.8	1 12.7	1 1	1.8 1	1.1 1	1.4 1.	42 1	1.8 1	12.8 1	12.4 1	12.3	13.6 I		1.5 1	1.6 1	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 1.64	1.4	1 1.0	1.3 1
actor Sy T-M Sy nv. U %) (SC	2 6.0	0.3 2	0.5 2	5.2	7.8	5.3	5.4 2	4	4.4	0.0	9.2	7 L.T	7.5 4	5.8 2	5.7 4	5.5 4	5.9 4	5.5	5.3	5.5	5.9 2.2	8.9	5.9 2	5.7 4	8.6	4. 0 8. 0 7	5.5 4	5.4 2	5.0 2	5.2	2.1		4.2	4.2	4.6	0.4	4 - c	5.3	7.2 4
CO CO Co CO CO (%)	30.7 21	29.9 20	30.4 21 28.9 11	24.5	24.9 I	22.0 1:	23.4 10	18.1	18.4	1 7.12	91.4	25.0 1	26.5 1'	24.0 1:	27.1 10	25.6 10	24.8 10	25.0 10	23.8 1	23.2	23.8	23.6 10	23.7 10	23.6 10	23.5 10	24.0	23.4 10	22.6 10	24.0 14	24.2	20.9	20.7	21.6 1-	20.6 1-	22.5 1.		0.82	26.3 1:	24.9 1
atalyst Age C (eta)	0.297	0.290	0.293	0.249	0.245	0.198	0.212	0.222	0.222	0.266	0.276	0.242	0.237	0.219	0.237	0.233	0.236	0.233	0.227	0.227	0.233	0.237	0.241	0.242	0.238	0.234	0.228	0.223	0.222	0.218	0.206	0.199	0.202	0.196	0.193	0.104	0.240	0.217	0.172
talyst C entory (lb)	7,160	7,160	7,160 7 160	5.716	5,716	5,716	5,716	1,216	1,216	2,800 5,620	5,620	5,620	5,620 5,620	5.620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	5,620	070	7 820	7,820	7,820
urry Inv t (ft)	9.0 4'	9.0 4	8.0 4	1.5	0.5 4	6.5 4	9.5 4	44 44	4 4	4 4 4 4	50	9.5 4:	8.5 9.0 4.4	0.5 4:	3.0 4:	1.5 4:	1.5 4:	1.5 4:	3.0 4.	1.0	9.5 5.0 4.4	8.5	8.5 4:	8.5 4:	6.5 4:	7.5 4. 4. 4.	7.0 4:	6.0 4:	5.5 4:	6.0 4	3.0	4 4 4 4 4 4	4.0 4.	3.5 4:	3.5	4 o t	1.0 5.5 4 4 4	5.0 4 4	6.0 4
Gas Ga oldup SI ol%) Hg	24.9 4	23.3 4	24.3 4 18.9 4	10 T T T T T T T T T T T T T T T T T T T	27.4 5	21.1 4	28.4 4	. 14.4	, <u> </u>	, , , ,	52	25.7 4	25.1 4 24.7 4	32.4 5	22.2 5	9.6 5	23.3 5	22.5 5	24.8 5	22.7 5	18.7 4	23.5 4	21 4	25 4	25.6 4	29.5 4 4 4	26.4 4	26.8 4	21.9 4	20.2 4	22.4 4	25.5	20.3 4	18.6 4	18.2 4	4 C I 2	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	25.4 4	27.6 4
slurry Conc. H 1% ox) (1	43.5	43.0	42.8	43.4	43.4	43.4	44.3	39.5	43.1	c.c4 1 C4	41.2	42.7	43.0	44.6	39.8	39.7	40.9	40.6	40.6	40.9	40.4 43.4	42.5	41.7	43	44.4	44./	44.3	45.1	43.6	42.8	45.3	45.5	4	44	43.7		42.4	46.4	47.0
pace 5 elocity (nr-kg) (w	585	648	630 657	100	584	420	302	859	847	08U 474	372	500	446 467	462	411	468	506	478	512	551	519	477	469	480	477	484 458	485	461	489	593	433 530	05C	589	543	477 404	494	317	219	329
Sup. S Sup. S scity V(67 3	5 89	67 3		99	63 2	61	62	8 8	7 6	19	63 3	2 23	62 3	61	62	63 3	63	63	64	63 63 64 64	63	62 3	63 3	62	65 63 65	63 3	62 3	63	65	5 5		65 3	64 3	69 69	00	2 29	3 6	65 3
ge Inlet s Vel FH) (ft/	2 0.	6 0.	6 0. 1 0.	2 0	.0 0.	7 0.	.8 0.	.8	0 0	. r	.6 0.	7 0.	8 4 0.0	6 0.	4 0.	1 0.	9 0.	6 0.	4	0.	0 0	5 0	8	3 0.	1 0.	6 0 0 0	1 0.	0	0.	0.	0 0	0	7 0.	2	0.0			0 0	3 0.
or Pur _f 1 Ga 0) (KSC	.79	.16	-06	40.	65.	74.	115	138	13	46.	105	.89	87.	72.	83.	76.	.89	78.	78.	72.	73.	62	.09	64.	. 67.	/1.	62.	.69	62.	99	73.	72	80.	80.	55.	.60	00 80	27.	47.
le React Fecc H) (H2:C	2.86	2.88	2.90	2.92	2.88	2.73	2.7(2.36	2.3(7C.7	3.2	2.6(2.93	2.9(3.19	2.98	2.78	2.91	2.76	2.63	2.61	2.58	2.58	2.81	2.58	2.61	2.61	2.52	2.8(3.05	2.55	2.54	2.92	2.71	2.85	20.7 20.7	58.7 CF C	2.62	2.67
Recyc Gas () (KSCF	2,171	2,212	2,201	2.342	2,258	2,224	2,063	2,192	2,178	161,2	2,010	2,127	2,097	2.170	2,121	2,154	2,161	2,163	2,210	2,221	2,175	2,160	2,152	2,163	2,150	2,168 2,146	2,171	2,155	2,145	2,100	2,181	2.181	2,118	2,154	2,189	201(2	2,002 2010	2,196	2,179
Fresh Feed (KSCFF	744	739	746	543	621	544	603	571	572	613	671	608	612 614	556	583	577	584	581	579	580	592 568	573	570	573	578	5/5	560	564	549	542	524	524	533	529	510	C64	800	471	566
Pres. ((psig)	708	710	710	210	014	710	710	710	710	710	710	710	710 710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	710	017	710	710	710
Temp (Deg C	235	234	234 734	230	235	235	234	233	234	234 234	235	235	235 235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	235	237	162	240	235	250
Gas Type	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Dataticed	Balanced	Balanced	Balanced							
Days On Stream	1190	1611	1192	1776	1231	1258	1261	1263	1264	1265	1272	1273	1274	1277	1278	1279	1280	1281	1282	1283	1284	1288	1289	1290	1291	1292	1294	1296	1297	1298	1299	1301	1302	1303	1304	1305	1307	1311	1316
Date	24-Mar-01	25-Mar-01	26-Mar-01	2/-Mar-01	4-May-01	31-May-01	3-Jun-01	5-Jun-01	6-Jun-01	10-uut-/ 8-lun-01	14-Jun-01	15-Jun-01	16-Jun-01 17-Jun-01	19-Jun-01	20-Jun-01	21-Jun-01	22-Jun-01	23-Jun-01	24-Jun-01	25-Jun-01	26-Jun-01 28-Jun-01	30-Jun-01	1-Jul-01	2-Jul-01	3-Jul-01	4-Jul-01 5-Jul-01	6-Jul-01	8-Jul-01	9-Jul-01	10-Jul-01	10-Iuf-11	13-Jul-01	14-Jul-01	15-Jul-01	16-Jul-01	10-Inf-/1	19-Jul-01 22-Jul-01	23-Jul-01	28-Jul-01
Case	Ξ	F	= =	=	F	=	н	Ξ	= =	= =	: =	Π	= =	=	Н	П	н	Ξ	= :	=	= =	: =	Ξ	Π	= :	= =	Π	Ξ	Ξ	Ξ	= =	: =	Ξ	Ξ	= =	= =	= =	= =	=

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

		Sparger	("K")	7.41	6.88	6.98	6.81	8.72	6.92	6.28
		Sparger	ar (psi)	8.36	7.85	69.9	7.56	8.47	197	6.94
	n	Overall	(Bu hr ft2 F)	148	154	160	155	156	157	152
		Reactor	VOI. 1700. (TPD/ft3)	0.085	0.087	0.089	0.089	0.087	0.087	0.087
		Catalyst	(gmol/hr-kg)	9.14	9.51	9.46	9.21	8.72	8.6	8.45
(p	Raw	MeOH	(TPD)	167.4	173.9	171.2	168.4	159.5	157.3	154.5
(cont'		Syngas	UIII. (SCF/lb)	41.4	42.7	42.4	42.8	42.1	42.4	42.3
ign 2 (Reactor	M-T-O	CONV. (%)	17.2	17.9	17.4	17.2	16.6	16.4	16.1
ımpai		00	CONV. (%)	24.9	24.7	24.2	23.7	23.2	23.0	22.5
yst C ²		Catalyst	Age (eta)	0.174	0.174	0.172	0.166	0.158	0.156	0.154
Catal		Catalyst	Inventory (1b)	47,820	47,820	47,820	47,820	47,820	47,820	47,820
sport		Gassed	Sturry Hgt (ft)	47.0	47.5	46.0	45.0	44.0	43.0	42.5
King		Gas	(vol%)	33.5	30.4	26.6	23.3	22.7	24.4	18.8
le for		Slurry	conc. (wt% ox)	48.7	47.2	46.6	46	46.4	47.7	46.6
/ Tabl		Space	velocity (l/hr-kg)	3393	3349	3356	3347	3322	3311	3299
nmary		Inlet Sup.	velocity (ft/sec)	0.66	0.65	0.65	0.65	0.64	0.64	0.64
ita Sur		Purge	uas (KSCFH)	48.3	70.6	71.3	75.4	62.8	61.7	59.2
2 – Da		Reactor	(H2:CO)	2.68	2.47	3.27	2.48	2.52	2.53	2.50
le D-2		Recycle	(KSCFH)	2,221	2,144	2,176	2,156	2,170	2,169	2,177
Tab		Fresh	(KSCFH)	577	619	909	600	560	556	544
		ć	psig)	710	710	710	710	710	710	710
		F	(Deg C)	250	251	250	250	250	250	250
		Ċ	Type	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced
		Days	Stream	1317	1319	1320	1321	1322	1323	1324
			Date	29-Jul-01	31-Jul-01	1-Aug-01	2-Aug-01	3-Aug-01	4-Aug-01	5-Aug-01
			Case	11	11	11	11	11	11	11

	arger	csist. 'K")	80	02	3 4	.13	.34	.36	.45	.48	41	37	38	.47	.51	.46	41	.43	.53	.50	29	.23	.35	37	89. 5	68. 5	99	n ≘	46	.86	02	.73	- 26	5 6	37	24	69	41	.55	.93	23	20	5 5	40
	ger Sp	(, K	~			5	6	3 (8	6	8	6	÷	0	4 9	4	9	-	7	~	5	3	8	2	е.		× •	+ 0		5	9	9	ין ש ו ו			5	4	5	0	5	4	_	0,	+ -
	l Spar	dr (psi	53	909	5.7	4.9	6.9	6.7	6.5	6.4	63	6.4	6.4	6.4	6.4	6.4	6.4	6.4	6.4	6.4	6.3	6.3	14.5	73	6.4	9.9 1	77	60	6.8	6.1	6.3	7.0	6.7	70	6.4	6.6	5.8	5.4	5.6	5.7	6.8	6.8	17	6.0 7.0
D	Overal	(Btu hr ft2 F	105	100	107	107	II	104	66	100	105	102	108	109	108	102	107	108	106	109	104	102	128	Ξ	114	113	114	701	106	108	106	105	101	108	108	110	118	130	119	117	121	115	11	621 112
	Reactor	Vol. Prod. (TPD/ft3)	0.086	0.00	0.094	0.091	0.092	0.092	0.094	0.095	0.101	0.093	0.092	0.096	0.098	0.096	0.098	0.096	0.097	0.093	0.092	0.097	0.080	0.087	0.093	0.090	160.0	0.000	0.086	0.084	0.087	0.087	0.085	0.084	0.084	0.084	0.082	0.085	0.078	0.079	0.085	0.084	0.083	0.081 c80.0
	Catalyst	MeOH Prod. (gmol/hr-kg)	10.17	11 87	12.08	11.47	12.38	12.40	12.35	12.23	12.27	12.26	11.85	12.43	12.39	12.28	12.54	12.43	12.18	12.07	11.89	11.89	9.76	12.31	12.19	12.12	06.11	11 43	11.48	11.42	11.28	11.25	11.38	11 71	11.74	11.72	11.49	11.63	11.67	11.55	16.11	11.85	11.11 11.11	11.05
Raw	MeOH	Prod. (TPD)	161.8	1 80.0	192.4	182.6	197.2	197.5	196.8	194.8	195.4	195.3	188.7	198.0	197.3	195.6	199.7	198.0	193.9	192.2	189.2	189.7	155.5	196.4	194.4	193.3	190.8	182.2	183.0	182.1	179.8	179.5	181.5	10/.0	187.3	186.9	183.3	185.5	186.1	184.2	190.0	188.9	187.7	181.3 181.3
	Syngas	Util. SCF/lb)	43.0	c 17	41.3	41.7	40.0	39.9	41.2	40.6	40.0	40.5	39.8	41.1	41.2	41.6	41.1	41.4	41.4	41.0	40.9	40.7	44.5	40.2	40.2	40.3	40.9	41.0	41.9	42.0	42.5	42.5	42.3	41.4	41.5	41.3	42.2	42.4	42.3	42.7	41.6	41.6	41.7	42.3 42.7
eactor	M-T-0	ONV. (%) (9.91	20.6	20.6	19.5	20.2	20.3	20.0	19.9	20.1	20.4	20.0	20.4	20.1	19.7	20.2	20.2	19.8	20.1	20.0	18.9	14.1	20.2	20.0	19.8	19.0	18.7	18.5	18.6	18.2	17.6	17.7	16.9	18.8	19.0	19.6	19.6	19.8	19.2	19.3	19.1	18.9	19.0 18.0
R	8,8	Conv. (%)	32.5	25.4	36.2	40.7	31.7	32.5	31.7	32.7	34.3	32.8	32.8	32.4	31.2	30.0	31.1	31.0	30.8	31.8	31.4	28.3	9.3	30.4	29.9	29.6 27.7	1.12	26.0	28.2	29.2	27.7	27.1	26.2	0.26	31.1	29.8	33.3	36.5	36.8	36.6	31.3	28.3	27.8	28.1 26.0
	Catalyst	Age (eta - corr)	1159	1 0.05	1.090	2.090	3.090	4.090	5.090	6.090	7.090	8.090	9.090	12.090	13.090	14.090	15.090	16.090	18.090	19.090	20.090	25.090	26.090	29.090	30.090	31.090	32.090 22.090	34.090	35.090	36.090	37.090	38.090	39.090	41.000	42.090	43.090	44.090	45.090	46.090	47.090	48.090	49.090	060.05	51.090 52.090
	atalyst	Age (eta) (1 537	527	0.530	.543	0.521	0.518).510	0.507	0.509	0.512	0.516).513	0.508	.497	0.506	0.505	.498	0.503	.489	0.475	.542	0.515	0.505	0.498	1.4/8	0.456	0.463).460	.449).436).437	0 440	.444	0.440).444	0.472	0.476	0.473).452	0.439	1.435).451).409
	talyst C	entory [b)	580 (200	,580	,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580 (,580	084	, 000 580	,580	,580 (,580 (,580 (,580	0.000,000,000,000,000,000,000,000,000,0	,580	,580 (,580 (,580 (,580 (,580 (,580 (,580 (084	,580
	ssed Ca	irry Inv (ft) (6 41	11		3.0 41	.0 41	.0 41	0.0 41	0.0 41	5.0 41	0.0 41	0.0 41	0.0 41	3.0 41	8.5 41	3.5 41	0.0 41	7.5 41	0.0 41	0.0 41	5.5 41	5.5 41	H.0 41	0.0	1.0	0.0	20 41 41 41	.5 41	.5 41	.5 41	0.0 41	.0 0	14 0.0	5.0 41 41	5.0 41	5.0 41	2.0 41	5.5 41	5.5 41	5.0 41	5.5 41 	14	1.0 4.1 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0
	as Ga	dup Sh 1%) Hgr	8		9 60 4 44	.8 48	.1 5	.8 5:	.7 5(.0	.7 40	.9 5(.6 49	.5	8.	.6 45	.1 48	.0	.0	4.	9 49	.1 40	.5 4(.1 52	. Si		7. 4 N 1	ن 17 14	- 1 - 2(.5 5.	94 60	.6 49	2 1 2 2	0 c 9 y	9 9 9	.8	2 5	.0 52	.4 50	.1 51	.7 5:	4 5	0.0	0. F.
	5 : 5	c. Hol ox) (vo	7 20	52	33	1 26	3 26	2 22	3 21	5 21	5 21	1 20	2 19	1 22	1 23	4 22	5 23	3 23	4 21	8 21	3 19	7 20	5 25	5 25	1 20	0 2 2	2 2	17	54	2 20	3 20	8 24	24	C7 C7 L	. 8	9 24	7 27	1 24	5 25	2 26	9 24	5 24	6 6 7 8	25 4
	Slur	/ Con (wt%	40	12	43.0	41.	39.	38.	38.	38.	40.	38.	38.	39.	40.	39.4	39.	39.	39.	38.	38.	39.	41.:	37.	38.	38.0	40.	.00	38.	37.	38.	39.	38.	.00	37.	37.	38.	38.	36.3	37.	37.	37.0	5/.	38.
	Space	Velocity (1/hr-kg	3320	2804	3841	3756	4053	4037	4021	3997	4005	3947	3930	3951	3990	4010	3989	3960	3955	3920	3890	4025	4220	4026	4016	4030	4067	4055	4036	4024	4016	4108	4151	101	4053	4043	3813	3849	3830	3880	4009	4039	4049	4000 4081
	Inlet Sup.	Velocity (ft/sec)	0.65	0.64	0.64	0.62	0.66	0.66	0.66	0.66	0.66	0.65	0.64	0.65	0.65	0.66	0.65	0.65	0.65	0.64	0.64	0.66	0.69	0.66	0.66	0.66	/ 970	0.06	0.66	0.66	0.66	0.66	0.67	C0.U	0.65	0.65	0.61	0.62	0.62	0.63	0.65	0.65	0.65	ca.u 990
	Purge	(KSCFH)	71.8	540	58.5	60.0	38.4	39.9	52.3	50.0	38.9	46.5	37.2	57.6	62.2	67.2	59.5	64.8	65.7	55.9	52.9	61.2	99.3	50.2	48.8	50.6	61.0	6.00	75.9	74.1	78.1	76.3	76.4	0.00	6.69	659	73.8	81.3	82.3	83.4	72.6	72.3	13.2	دد/ 83.1
	Reactor	reed (H2:CO)	3 33	2.5.2	3.64	4.55	3.09	3.20	3.15	3.24	3.39	3.12	3.30	3.09	3.09	3.00	3.02	3.01	3.06	3.13	3.10	2.92	0.64	2.90	2.88	2.90	2.81	0 TG	3.02	3.14	3.00	3.06	2.91	3 57	3.35	3.13	3.49	3.94	3.92	4.06	3.27	2.88	2.86	2.79
	Recycle	Gas (KSCFH)	1.798	080.6	2,100	2,096	2,244	2,217	2,213	2,201	2,215	2,159	2,191	2,154	2,177	2,198	2,166	2,159	2,173	2,140	2,129	2,234	2,461	2,240	2,229	2,244	2,260	2 2 8 I	2,266	2,255	2,251	2,317	2,350	202,2 7 7A6	2,286	2,287	2,108	2,122	2,102	2,141	2,219	2,253	2,259	2,308 2,284
	Fresh	reed (KSCFH)	580	640	663	634	658	657	675	658	651	660	627	677	677	677	684	684	699	657	645	643	577	658	652	649	000	040	640	637	637	636	639	040	648	644	645	655	656	655	658	655	652	645 645
	4	Pres. (psig)	573	666	675	685	685	685	685	685	685	683	685	685	685	685	685	685	685	685	685	685	684	685	685	685	680 282	685	685	685	685	969	700	00/	700	700	700	969	695	700	700	700	00/	700
	ŧ	I emp (Deg C)	216	916	218	217	218	219	218	218	220	219	218	218	218	218	218	218	218	218	218	218	218	218	218	218	218	218	218	218	218	218	218	077	220	220	220	220	220	220	220	220	220	220 220
	C	Type	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	H2:CO = 0.6	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced
	Days	On Stream	PC	5 7	32	39	40	41	45	46	47	48	49	52	53	54	55	56	58	59	60	65	66	69	70	12	74	C 7	07 12	78	79	80	8	84	298 98	87	90	16	92	93	94	95	96	79 99
		Date	7_Sen_01	1 6 01	5-Sep-01	2-Oct-01	3-Oct-01	4-Oct-01	8-Oct-01	9-Oct-01	0-Oct-01	1-Oct-01	2-Oct-01	5-Oct-01	6-Oct-01	7-Oct-01	8-Oct-01	9-Oct-01	11-Oct-01	2-Oct-01	3-Oct-01	'8-Oct-01	9-Oct-01	l-Nov-01	2-Nov-01	3-Nov-01	5-Nov-01	10-001-/	10-2021-0	0-Nov-01	1-Nov-01	2-Nov-01	3-Nov-01	10-vov-0	10-vov-01	9-Nov-01	2-Nov-01	3-Nov-01	4-Nov-01	5-Nov-01	[0-vov-9]	7-Nov-01	8-Nov-01	:9-Nov-01 1-Dec-01
		Case	2000-8	c 3 000c	2000-8 2	2000-8	2000-8	2000-8	2000-8	2000-8	2000-8 1	2000-8 1	2000-8 1	2000-8 1	2000-8 1	2000-8 1	2000-8 1	2000-8 1	2000-8 2	2000-8 2	2000-8 2	2000-8 2	2000-9 2	2000-8	2000-8	2000-8	2000-8	2000-8	2000-8 5	2000-8 1	2000-8 1	2000-8 1	2000-8 1	1 0-0002	2000-8 1	2000-8 1	2000-8 2	2000-8 2	2000-8 2	2000-8 2	2000-8 2	2000-8 2	2000-8 2	2000-8
						•															1												1											

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

Case	Date	Days On Stream	Gas Type	Temp (Deg C)	Pres. (psig)	Fresh Feed (KSCFH)	Recycle Gas (KSCFH)	Reactor Feed (H2:CO)	Purge Gas (KSCFH)	Inlet Sup. Velocity (ft/sec)	Space Velocity (l/hr-kg)	Slurry Conc. (wt% ox)	Gas Holdup (vol%) 1	Gassed C Slurry In Hgt (ft)	Catalyst C ventory (Ib)	Catalyst Age (eta)	Catalyst Age	CO CO (%) CO CO CO CO	cactor S-T-M S onv. (S) (S)	jyngas M Util. P. CF/Ib) (T	Raw feOH (rod. Me .PD) (gn	Catalyst eOH Prod. mol/hr-kg)	Reactor Vol. Prod. (TPD/ft3)	U Overall (Btu hr ft2 F)	Sparger dP (psi)	Sparger Resist. ("K")
a 0000	, A				001	101	3700	20 0	000		F.204	¢ 0¢	2.20	0.53	002 11	0110	50 000	376	1 01		0.00	ę	000 0	-	101	
2000-8	2-Dec-01 3-Dec-01		Balanced	223	200/	645	2.274	3.05	90.0 68.8	0.66	4084	39.2 39.2	26.1	51.5	11.580	0.402	54.090	28.8	10.1 18.7	41.4	87.1	11.42	0.086	100	7.12	6.65
2000-8	4-Dec-01	102	Balanced	224	700	654	2,296	2.89	71.0	0.67	4113	38.5	23.2	51.0 4	41,580	0.397	55.090	27.8	18.7	41.5 18	89.2	11.87	0.088	103	7.44	6.67
2000-8	5-Dec-01	103	Balanced	224	700	651	2,277	2.90	69.1	0.66	4083	38.5	23.5	51.0 4	41,580	0.397	56.090	28.2	18.9	41.4 10	88.7	11.84	0.088	110	7.05	6.36
2000-8	6-Dec-01	104	Balanced	224	700	650	2,258	2.83	73.5	0.66	4050	37.8	21.3	51.0 4	41,580	0.395	57.090	27.8	18.9	41.8 1.	86.7	11.71	0.087	108	7.06	6.27
2000-8	7-Dec-01	105	Balanced	224	700	646	2,250	2.81	74.6	0.66	4038	38.9	23.0	50.0 4	41,580	0.391	58.090	27.5	18.8	41.9 1.	84.8	11.59	0.088	110	7.19	6.36
2000-8	8-Dec-01	106	Balanced	224	700	651	2,240	2.83	71.6	0.65	4028	39.4	25.9	51.0 4	41,580	0.399	59.090	28.2	19.3	41.6 1.	87.6	11.77	0.088	118	7.25	6.39
2000-8	9-Dec-01	107	Balanced	224	700	647	2,285	2.79	71.6	0.66	4082	38.9	26.0	52.0 4	41,580	0.388	60.090	27.1	18.8	41.7 1.	86.2	11.68	0.085	114	7.24	6.41
2000-8	10-Dec-01	108	Balanced	224	700	651	2,273	2.75	1.77.	0.66	4080	40.6	30.7	52.0 4	41,580	0.386	61.090	26.8	18.7	42.0 1.	85.9	11.66	0.085	Ξ	7.19	6.34
2000-8	11-Dec-01	109	Balanced	224	700	654	2,272	2.68	79.3	0.66	4084	39.1	25.7	51.5 4	41,580	0.381	62.090	26.1	18.5	42.3 1	85.4	11.63	0.086	111	7.19	6.32
2000-5	12-Dec-01	110	H2:C0 = 0.7	224	700	643	2,413	1.06	107.4	0.69	4220	38.0	29.2	54.0	41,580	0.407	63.090	13.3	16.3	44.9 1	71.6	10.78	0.076	83	10.51	5.93
2000-5	14-Dec-01	112	H2:CO = 0.7	224	069	513	2,449	0.58	91.1	0.67	4084	37.6	27.7	56.0 4	41,580	0.435	65.090	8.7	13.7	44.2 1.	39.4	8.76	0.059	142	12.14	6.01
2000-5	15-Dec-01	113	H2:CO = 0.7	224	069	506	2,456	0.54	0.66	0.67	4072	38.9	25.2	51.5 4	41,580	0.408	66.090	7.9	12.7	45.3 1.	34.0	8.42	0.062	150	11.91	6.05
2000-5 2000-5	16-Dec-01	114	H2:CO = 0.7 H2:CO = 0.7	224	069	492	2,402 2,515	0.50	98.5 108.8	0.66	3999 4218	39.1 47 7	29.5 30.8	51.5 4	41,580	0.412 0.354	67.090 68.090	7.5 8.5	12.3	45.8 L: 47.1 L:	35.6	8.09 8.51	0.060	144	11.43 11.70	5.90 6.09
6-0007	1/-Dec-01	ci i			001	100	0.10.4	11.0	100.0	0.0	D171	1.47	0.00	0.01	000/11	10000	00000	0.0	1.20		0.00	10.00	0.001		2 10	0.0
2000-8	21-Dec-01 22-Dec-01	119	Balanced	226	00/ 200/	200 255	2,298	3.57	40.0 51.7	0.65	410/	38.5	19.8	49.0 4	41,580	0.323	60.090 57.090	28.3	16.4	41.2 16	07.0 61.7	10.20	0.079	116	5.40	7.12 5.92
2000-5	24-Dec-01	122	H2:CO = 0.7	226	700	512	2,519	0.65	106.8	0.68	4183	42.0	28.8	48.0 4	11,580	0.341	63.090	8.2	11.6	47.2 1:	30.2	8.17	0.065	106	13.03	5.96
2000-5	27-Dec-01	125	H2:C0 = 0.7	225	700	572	2,469	0.73	125.9	0.68	4195	43.2	26.0	44.0 4	41,580	0.400	64.090	8.9	12.9	47.5 14	44.3	9.05	0.078	164	11.23	6.11
2000-5	28-Dec-01	126	H2:CO = 0.7	225	700	542	2,455	0.70	116.3	0.68	4164	43.1	25.5	44.0 4	41,580	0.363	65.090	8.6	12.6	46.5 1:	39.9	8.78	0.076	139	10.87	5.89
2000-5	29-Dec-01	127	H2:CO = 0.7	226	700	584	2,423	0.95	100.5	0.68	4147	39.0	27.5	53.0 4	41,580	0.354	66.090	11.3	14.7	44.3 1.	58.1	16.6	0.071	106	11.05	6.55
2000-8	1-Jan-02	130	Balanced	226	002	629	2,233	2.72	87.4	0.65	4006	38.6	26.5	53.0 4	41,580	0.332	060.69	24.7	17.4	43.0 1	75.8	11.02	0.079	127	6.88	6.83
2000-8	2-Jan-02	131	Balanced	226	700	600	2,280	2.67	61.8	0.66	4039	38.8	25.6	52.0 4	41,580	0.327	70.090	24.3	17.2	41.3 1.	74.5	10.94	0.080	127	7.02	6.60
2000-8	3-Jan-02	132	Balanced	226	700	605	2,276	2.68	63.4	0.66	4023	39.4	27.2	52.0 4	41,580	0.332	71.090	24.5	17.4	41.2 1	76.1	11.04	0.081	134	7.00	6.75
2000-8	4-Jan-02	133	Balanced	226	700	575	2,309	2.65	61.6	0.66	4033	39.2	26.7	52.0	41,580	0.306	72.090	22.8	16.3	41.8 1.	65.1	10.35	0.076	120	16.9	6.56
2000-8	5-Jan-02	134	Balanced	226	700	550	2,334	2.65	58.8	0.65	4011	37.4	24.2	54.0 4	41,580	0.291	73.090	21.9	15.6	41.8 1.	57.7	9.89	0.069	106	6.61	6.32
2000-8	6-Jan-02	135	Balanced	226	002	550	2,290	2.79	62.8	0.65	3982	37.7	25.0	54.0 4	41,580	0.290	75.000	22.3	15.4	42.4 I. 42.0 1.	55.6	9.76	0.069	107	6.55	6.66 6 03
2000-8	20-118L-/ 8-19n-02	137	Balanced	226	2007	555	0:339	2.56	639	0.65	4020	38.5	26.1	54.0	11 580	0.290	76.090	213	15.6	42.0 1	58.7	50.6	0.070	110	6.84	6.51
2000-8	9-Jan-02	138	Balanced	226	700	550	2,317	2.56	68.5	0.65	4011	38.8	25.6	52.0 4	41,580	0.283	060.77	20.8	15.2	42.6 1:	55.1	9.73	0.071	108	7.30	6.93
2000-8	10-Jan-02	139	Balanced	226	700	545	2,315	2.54	71.0	0.65	4003	37.3	21.8	52.5 4	41,580	0.280	78.090	20.7	15.2	42.4 1:	54.2	9.67	0.070	102	6.31	5.99
2000-8	12-Jan-02	141	Balanced	226	700	547	2,311	2.52	74.1	0.65	3985	36.6	23.0	55.0 4.	41,580	0.278	80.090	20.4	15.1	43.0 1:	52.9	9.59	0.066	100	6.84	6.60
2000-8	13-Jan-02	142	Balanced	226	700	550	2,290	2.56	75.2	0.65	3975	38.8	25.7	52.0 4	41,580	0.280	81.090	20.7	15.2	43.0 1.	53.5	9.63	0.070	108	66.9	6.83
2000-8	14-Jan-02	143	Balanced	226	700	550	2,283	2.68	73.7	0.65	3957	38.9	25.9	52.0 4	41,580	0.282	82.090	21.6	15.3	42.8 1.	54.0	9.66	0.071	114	6.37	6.41
2000-8	16-Jan-02	145	Balanced	228	700	550	2,273	2.80	61.2	0.65	3948	37.4	25.5	55.0 4	41,580	0.278	83.090	23.0	15.8	42.0 1.	57.3	9.86	0.072	115	6.51	6.58
2000-8	17-Jan-02	146	Balanced	228	700	555	2,263	2.81	61.1	0.64	3932	37.5	24.4	54.0 4	41,580	0.282	84.090	23.5	16.1	41.6 1,	60.0	10.03	0.070	114	6.14	6.23
2000-8	19-Jan-02	148	Balanced	228	700	575	2,287	2.70	71.4	0.65	3967	38.7	25.9	52.5 4	41,580	0.280	85.090	22.7	16.2	42.9 1,	60.8	10.08	0.073	118	6.75	6.65
2000-8	24-Jan-02	153	Balanced	230	700	555	2,205	3.24	69.69	0.63	3849	38.6	30.0	56.0 4	41,580	0.267	90.090	25.9	16.1	42.8 1.	55.7	9.76	0.066	115	6.32	7.21
2000-8	25-Jan-02	154	Balanced	230	700	565	2,227	3.10	73.6	0.64	3893	37.7	26.9	55.5	41,580	0.270	91.090	25.2	16.2	42.8 1.	58.5	9.94	0.068	115	6.38	6.99
2000-8	26-Jan-02	155	Balanced	230	700	558	2,240	3.07	72.2	0.64	3910	37.6	25.2	54.5	41,580	0.267	90.090	24.4	15.8	42.9 1.	55.9	9.78	0.068	III	6.32	6.63
2000-8	27-Jan-02	156	Balanced	230	700	559	2,261	2.89	70.1	0.65	3929	36.4	20.8	54.0 4	41,580	0.267	93.090	23.6	16.0	42.4 1.	58.2	9.92	0.070	113	5.95	6.17
2000-8	28-Jan-02	157	Balanced	230	700	548	2,287	2.66	72.5	0.65	3963	36.9	21.6	53.5 4	41,580	0.254	94.090	21.5	15.3	42.9 1.	53.3	9.62	0.068	105	6.52	6.36
2000-8	29-Jan-02	158	Balanced	230	700	543	2,292	2.59	71.9	0.65	3961	38.0	23.7	52.5	41,580	0.251	95.090	20.9	15.2	42.9 1.	51.9	9.53	0.069	105	6.84	6.58
2000-8	30-Jan-02	159	Balanced	231	700	550	2,288	2.61	77.4	0.65	3961	37.6	22.4	52.5	41,580	0.249	96.090	20.9	15.1	43.6 1.	51.4	9.50	0.069	102	6.86	6.64
2000-8	31-Jan-02	160	Balanced	231	700	550	2,254	2.92	85.0	0.65	3922	37.6	20.9	51.5 4	41,580	0.249	97.090	21.7	15.0	44.2 1-	49.3	9.37	0.069	76	6.46	6.64

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

Sparger Resist. ("K")	7.24 6.67 6.66	6.47	5.93	6.09	6.UI 6.10	5.94	6.14	6.05 6.00	7.34	7.15	7.27	7.17	7.39	8.78	6.76	6.46	6.59	6.78	8 30 8 30	7.08	6.68	6.49	7.03	6.09 7.26	7.01	6.91	6.56	7.49	8.79	6.48	6.24	0.00	0.08	6.53	6.79	6.51	6.58 6.58
Sparger dP (psi)	7.11 6.07 6.06	5.56	10.89	11.03	27.01 10.75	10.64	11.09	10.70 10.59	6.67	6.67	6.82	6.65	7.29	8.82	6.76	6.53	6.70	6.78	15.0	7.00	7.07	6.80	7.64	0.08 7.47	7.21	5.68	5.95	7.65	8.74	6.42	6.33	0./9	77.0	7.20	7.74	7.44	7.47 7.47
U Overall (Btu hr ft2 F)	108 1107	108	86	97	00 92	97	98	96 100	112	107	112	109	117	118	110	Ξž	56	103	111	112	117	112	611	/ 11	115	108	109	118	III	107	112		671	115	116	112	106 103
Reactor Vol. Prod. (TPD/ft3)	0.068 0.065 0.067	0.073	0.060	0.057	0.059	0.060	0.057	0.061	0.071	0.067	0.069	0.072	0.065	0.064	0.067	0.068	0.060	0.062	0.063	0.064	0.067	0.066	0.064	0.067	0.065	0.067	0.069	0.070	0.068	0.067	0.068	0.06/	0.000	0.087	0.092	0.095	0.093 0.092
Catalyst eOH Prod. mol/hr-kg)	9.43 9.41 9.42	9.82	8.17	8.10	7.80	8.09	8.11	8.05 7.86	9.34	9.29	9.21	9.26	8.77	8.76	9.11	9.01	8.88	8.80	8.65	8.61	8.95	8.71	8.24	8.15	8.56	8.42	8.31	8.35	8.04	7.92	7.98	1.09	10.9/	11.97	12.45	12.19	12.10 11.88
Raw AeOH Prod. M TPD) (g	150.3 150.0	56.6	130.2	129.1	0.621	128.8	129.2	128.2 125.2	149.0	148.0	146.7	147.6	39.7	139.7	145.3	143.6	141.5	140.2	0.761	137.2	42.6	138.8	131.2	129.7	136.2	134.0	132.2	132.9	128.1	126.2	127.2	4.77	02.5	6.061	195.4	191.2	189.7 186.3
Syngas N Util. 1 SCF/lb) (43.7 44.0 45.0	42.0	45.7	46.5	46.6	46.2	45.9	46.5 46.5	43.8	44.3	45.2	44.7	46.9	46.7	45.2	45.6	46.2	8.6	47.3	47.9	46.4	46.4	50.1	50.5	49.0	50.1	50.0	49.9	47.1	48.1	47.8	49.0 1	45.4	40.9	40.5	41.3	41.5
Leactor D-T-M Conv. (1 (%) (1)	15.0 15.1	16.0	12.1	11.7	11.6	12.1	12.3	12.2 11.8	15.8	15.9	15.5	15.5	14.7	14.5	15.0	14.9	14.8	14.8	14.6	14.2	15.3	14.9	8.51	13.3	14.2	15.5	14.8	14.0	13.4	13.1	13.3	13.0	C:/1	18.9	19.4	19.4	19.2 18.9
R CO CO ((%)	21.2 22.7	26.1 26.1	8.0	7.8	7.8	8.3	8.3	8.4 8.1	24.0	23.6	22.5	23.0	21.2	20.5	21.8	21.4	21.2	21.5	20.4	19.7	21.4	20.6	20.3	19.0	19.9	21.7	20.6	19.5	19.2	18.5	18.4	18.0	5.22	24.0	26.6	27.7	27.4 26.9
Catalyst Age (eta - corr)	060.86 090.090	102.090	108.090	108.090	060.001	112.090	113.090	114.090 115.090	060.96	060.76	98.090	060.66	102.090	103.090	104.090	105.090	106.090	060.701	109 090	110.090	112.090	107.090	108.090	110.090	111.090	112.090	113.090	114.090	115.090	116.090	117.090	118.090	060.011	112.090	113.090	114.090	115.090 116.090
atalyst Age (eta)	0.249 0.257 0.258	0.248	0.263	0.287	0.278	0.270	0.266	0.258 0.253	0.229	0.226	0.221	0.206	0.181	0.177	0.166	0.162	0.157	0.157	0.158	0.154	0.152	0.147	0.138	0.137	0.144	0.145	0.142	0.139	0.130	0.126	0.128	0.120	0.205	0.295	0.308	0.312	0.308 0.301
talyst C entory (Ib)	,580 ,580	, <u>-</u> 80 (,580 (,580	.580	,580	,580	,580	,580 (,580	,580	,580 (,580 0	,580	,580	,580	,580	, 580 082, 085	. 00C.	,580	,580	,580	,580	580	,580	,580	,580	,580 (,580	,580	,580	080,	1 104	904 1	,904	,904	,904 ,904
ssed Ca urry Inv t (ft)	2.5 41 5.0 41 2.5 41	1.0 - 4	2.0 41	4.0 7 41	14 c.u 14 0.1	1.0 41	4.0 41	0.0 2.0 41	0.0 41	2.5 41	1.0 41	9.0 41	1.5 41	2.0	2.0 41	0.0	5.5 41	4.0 4.1 4.1 4.1 4.1 4.1 4.1 4.1 4.1 4.1 4.1	(+ 0.0 0 0 0	1.0 41	1.0 41	1.0 41	8.5 7 4]	6.5 41 6.5 41	0.0 41	8.0 41	5.0 41	5.0 41	5.0 41	5.0 41	4.5 41			2.0	0.5 40	8.0 40	8.5 8.5 40 40
Jas Ga oldup SI o1%) Hg	8.6 3.8 4.6 5.5 5.5 5.5 5.5 5.5 5.5 5.5 5.5 5.5 5	1.3 5	7.5 5	6.5 2.9 2.9	0.0 4.6 5 5	1.3 5	6.5 5.	2.6 5 4.0 5	2.5 5	7.4 5	7.8 5	3.9 4	7.5 5	4.6 5.	6.0 5	2.3 5	3.8	2.4	0.0	5.0 5	2.8 5	1.2 5	8.4.	5.3 4	4.0 5	7.8 4	8.7 4	9.4 4	8.8	7.7 4	9.2		9.0.9 0 1 2 1 2 1 2	6.6 5	5.9 5	4.6 4	0.9 4 0.9
urry 6 onc. Ho % ox) (v	9.7 6.9 7 8 7 8 7 8	<u>8.1</u> 2	9.7 2	8.4	1.6 1.6 1.6	8.1 2	8.3 2	9.0 8.5 2	9.0 2	6.2 1	7.0 1	0.0 2	0.0 2	8.8	9.4 2	9.2	0.6	7.3	- 60	9.5 2	8.9 2	8.9 2	0.9	2.2 2	9.8	8.9 1	0.3 1	4.5 2	8.3 3	0.5 1	0.2	1.4	1.6	7.2 1	9.4 2	0.3 2	8.9 2 2
ace Sl ocity C. -kg) (wt	48 19 16 2	81 - 3	43 3	77 3 56	54 34 3	37 3	18 3	25 3 13 3	55 3	33 3	68 3	97 4	11	50 3	80 3	, 3 , 60	14 -	01 6	0 T	55 3	04 3	70 3	18 1	49 63 5 4	34 3	26 3	58 4	88 4	16 4	69 4	14	ין ז 1 ק	ין 19	70 3	21 3	98 4	98 98 3
up. Sp ity Velo c) (I/hr	33 39	+ + 38	8 40	8 9 9 9	7 40	7 40	7 40	7 40	2 37	2 37	3 37	37	1 38	38	38	, % , %	88	85 5		38.	38	5 37	88 8	0 X0	38	34	1 35	5 37	38	38	8 1	5 1 2		39 4	7 41	7 40	7 40 40
In let S Veloc (ft/se	0.65 0.65	0.62	39.0	0.68	20:0 0:67	0.67	0.67	0.67 0.67	0.62	0.62	0.63	0.63	0.62	0.65	0.66	0.66	0.65	0.02	-0.0 0.64	0.65	0.65	0.65	0.66	0.0	0.66	0.59	0.61	0.65	0.65	0.66	0.65			0.06	0.67	0.67	0.63
Purge Gas (KSCFH	76.5 82.7 86.0	64.7	95.7	101.7	101.8	100.5	99.4	101.0 99.8	81.2	88.9	96.4	87.4	114.3	115.8	6.06	97.8	104.0	110.3	113.3	117.1	103.1	103.7	133.2	140.2	128.0	139.5	138.6	136.7	103.4	108.0	106.6	120.0	0.4%	57.7	60.8	61.7	64.3 63.2
Reactor Feed (H2:CO)	2.71 2.98 2.04	3.32	0.54	0.66	0.69	0.65	0.70	0.73 0.71	2.98	2.88	2.79	2.91	2.78	2.72	2.83	2.76	2.76	2.80	2 66	2.64	2.67	2.61	2.85	2.73	2.67	2.65	2.61	2.62	2.75	2.72	2.87	707	25.2	2.32	2.58	2.72	2.72 2.72
Recycle Gas (KSCFH)	2,282 2,263 7.746	2,171	2,393	2,425	2,406	2,393	2,377	2,384 2,371	2,160	2,139	2,163	2,202	2,201	2,219	2,226	2,221	2,218	2,200	2,105 2,198	2,223	2,198	2,177	2,260	2,250	2,208	1,916	2,012	2,179	2,251	2,556	2,246	4777	275,2	2,192	2,239	2,244	2,242 2,252
Fresh Feed (KSCFH)	547 550 575	548	495	501	494	496	495	497 485	543	547	553	550	546	544	548	545	545	547	245 543	547	551	537	547	546 546	556	559	551	553	503	505	507	900 1900	680 650	920 651	659	657	655 643
Pres. (psig)	700 700	700	693	695 605	c60	700	700	700 700	700	700	700	700	700	700	200	700	200	00/	00/	700	700	700	00/	00/	700	700	700	700	700	700	700	00/	00/	700	700	700	700
Temp (Deg C)	230 230	235	235	235	235	235	236	235 235	235	235	235	238	240	240	246	246	247	247	245	245	250	250	250	250	250	250	250	250	251	251	251	007	252	235	235	235	235 235
Gas Type	Balanced Balanced Balanced	Balanced	12:CO = 0.7	12:CO = 0.7	12:CO = 0.7 12:CO = 0.7	12:CO = 0.7	12:CO = 0.7	D:CO = 0.7 D:CO = 0.7	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced	Balanced Balanced							
Days On Stream	161 162	165	168 F	171 H	173 F	175 F	176 F	177 F 178 F	179	180	181	183	190	191	192	193	194	195	197	198	200	201	202	203	205	206	207	208	209	210	211	212	219	221 221	222	223	224 225
Date S	Feb-02 Feb-02	Feb-02	Feb-02	-Feb-02	-Feb-02 Feb-02	-Feb-02	-Feb-02	-Feb-02 .Feh-02	-Feb-02	-Feb-02	-Feb-02	-Feb-02	Mar-02	Mar-02	Mar-02	Mar-02	Mar-02	Mar-02	Mar-02	Mar-02	Mar-02	-Mar-02	-Mar-02	-Mar-02 Mar-02	Mar-02	Mar-02	-Mar-02	-Mar-02	-Mar-02	-Mar-02	-Mar-02	-Mar-02	-Mar-02	Apr-02 Apr-02	Apr-02	Apr-02	Apr-02 Apr-02
Case	2000-8 1- 2000-8 2- 2000-8 2-	2000-8 5-	2000-5 8-	2000-5 11	2000-5 13.	2000-5 15	2000-5 16	2000-5 17 2000-5 18.	2000-8 19	2000-8 20	2000-8 21	2000-8 23	2000-8 2-	2000-8 3-	2000-8 4-	2000-8 5- 2000-8 5-	2000-8 6-	2000-8 7-	2000-8 0.0	2000-8 10-	2000-8 12-	2000-8 13	2000-8 14	2000-8 15-	2000-8 17.	2000-8 18.	2000-8 19.	2000-8 20	2000-8 21.	2000-8 22	2000-8 23	2000-8 24	2000 0 21	2000-8 2-	2000-8 3-	2000-8 4-	2000-8 5- 2000-8 6-
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Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

		Sparger	Resist. ("K")	6.48	6.43	6.61	6.66	6.93	6.93	6.50	6.51	6.52	6.61	6.44	6.45	6.44	6.40	6.37	6.51	6.44	6.39	6.44	6.40	6.33	6.39	6.37	6.54	6.46	6.46	6.49	6.48	6.60	6.48	6.86	6.82	6.86	6.64	6.83	6.97	6.95
		Sparger	(isd)	7.28	7.17	7.44	7.45	7.68	17.7	7.36	7.39	7.53	7.55	7.12	6.83	6.85	6.77	6.86	7.04	6.75	6.57	6.46	6.68	6.76	6.76	6.58	6.73	6.70	6.57	6.59	6.67	69.9	6.57	7.03	7.12	7.04	6.90	7.00	7.02	7.12
	n	Overall	(Btu hr ft2 F)	108	104	111	107	104	105	103	115	114	108	105	107	109	106	108	106	106	106	110	109	103	112	107	105	76	101	101	110	101	100	81	127	102	111	66	100	88
		Reactor	/ol. Prod. TPD/ft3)	0.095	0.095	0.095	0.093	0.093	0.095	0.095	0.103	0.100	0.094	0.093	0.089	0.087	0.088	0.091	0.090	0.090	0.009	0.086	0.086	0.084	0.088	0.086	0.087	0.085	0.087	0.083	0.088	0.080	0.082	0.072	0.091	0.080	0.083	0.082	0.083	0.080
		lyst	Prod. V hr-kg) (02	88	98	85	58	54	43	54	79	84	4	4	76	63	59	52	37	17	=	4	5	6	1	5	=		3	F	F	=	6	45	2	5	9	=	=
		I Cata	MeOH (gmol/l	12.0	Ш	П.	TI .	. II		П.	Ξ	Ξ.	10.1	10.4	10.4	10.	10.0	10.3	10.	10.	10.	9.6	9.6	9.6	9.6	9.6	9.7	9.4	9.5	9.7	9.7	6.6	9.6	8.1	10.4	9.6	9.5	9.3	9.5	9.0
	Raw	MeOF	Prod.	188.4	186.2	187.8	185.7	181.4	180.8	179.2	180.9	184.7	169.9	163.6	163.6	168.7	166.7	165.9	164.8	162.5	159.5	153.7	154.2	154.1	154.9	154.6	152.9	147.5	148.9	152.5	152.2	155.5	153.8	128.2	163.8	150.7	149.5	146.5	148.9	141.0
		Syngas	Util. (SCF/lb)	41.7	42.1	41.8	42.3	43.2	43.1	43.6	43.7	42.6	41.6	42.2	42.1	41.1	41.6	41.7	43.5	42.7	42.8	43.4	42.4	42.9	42.6	42.8	43.7	43.7	44.5	42.3	42.6	41.9	42.3	46.5	41.7	42.7	43.4	45.4	43.5	43.4
,	Reactor	M-T-O	Conv. (%)	19.1	18.9	19.3	19.0	18.5	18.6	18.4	18.7	19.1	17.8	17.2	17.4	18.0	17.7	17.6	20.1	16.9	16.8	16.5	16.4	16.2	16.3	16.3	15.6	14.8	14.9	14.6	15.7	15.8	16.0	13.0	17.2	16.6	15.7	15.2	15.3	14.6
)		8	Conv. (%)	27.2	26.8	27.4	27.0	26.4	26.3	25.3	25.7	26.0	24.7	24.2	26.2	27.1	26.4	25.6	24.5	24.5	24.8	24.0	23.2	22.2	22.5	22.4	21.7	20.3	20.6	22.1	22.2	22.6	21.8	17.5	24.5	22.6	22.0	21.3	21.5	20.6
(Catalyst	Age (eta - corr)	117.090	118.090	119.090	120.090	121.090	122.090	123.090	124.090	125.090	126.090	127.090	128.090	129.090	130.090	131.090	132.090	133.090	134.090	135.090	136.090	137.090	138.090	139.090	140.090	141.090	142.090	143.090	144.090	145.090	146.090	143.090	144.090	145.090	146.090	147.090	148.090	149.090
,		Catalyst	Age (eta)	0.306	0.300	0.305	0.299	0.291	0.291	0.289	0.293	0.303	0.279	0.265	0.268	0.279	0.272	0.270	0.263	0.258	0.255	0.243	0.241	0.239	0.241	0.242	0.235	0.224	0.228	0.235	0.235	0.237	0.233	0.188	0.229	0.205	0.205	0.198	0.200	0.177
		atalyst	ventory (lb)	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904	0,904
(assed C	lurry In gt (ff)	17.5 4	16.5 4	17.0 4	17.5 4	16.5 4	15.5 4	15.0 4	12.0 4	14.0 4	13.0 4	12.0 4	14.0 4	16.5 4	15.0 4	13.5 4	13.5 4	13.0 4	13.0 4	12.5 4	13.0 4	14.0 4	12.0 4	13.0 4	12.0 4	11.5 4	11.0 4	14.0 4	11.5 4	16.5 4	15.0 4	12.5 4	13.0 4	15.0 4	13.0 4	12.5 4	13.0 4	12.0 4
)		as G	dup [%] H	2	3	4	8.	7 67	5	7 0.	.2	7 63	8.	9.9	8.9	7 0.0	3 4	4.	7 E	.3	L.7 L	.6	11	.5 4	.3	.5	8.	7 6.	.5 4	.6	7 0.3	4.	7 0.9	.3 2	8.3	.4	8.	.5	11 2	.5
		rry G	nc. Hol 6 ox) (vo	.1 23	.7 20	.5 26	17 24	.6 22	.9 22	.5 17	.7 21	.7 18	.0 17	.2 16	10	.6 20	3 16	.6 17	.0 20	.5 16	.0 14	.3 14	.4 19	15 15	.1 16	.0 14	.6 17	.7 13	.0 10	.4 14	.0 18	.1 21	16	.1 20	.7 22	.7 20	.7 15	91 6.	.8 20	.3 19
		ce Slu	city Co kg) (wt%	5 40	7 35	7 41	1 40	9 40	2 40	4 35	2 42	а 40	2 41	5 41	3 40	4 35	5 35	3 40	2 42	4 40	5 40	8 40	5 41	5 35	0 41	2 40	2 41	6 40	0 40	2 35	3 42	2 40	2 35	4 42	7 42	1 40	5 41	1 41	2 41	8 42
,		p. Spa	y Veloc (I/hr-	407	404	404	405	402	401	401	400	400	397	395	391	391	391	392	398	397	393	385	389	393	391	388	398	402	401	399	395	401	403	396	391	392	389	391	393	392
		Inlet Su	Velocit (ft/sec	0.67	0.66	0.66	0.66	0.66	0.66	0.66	0.65	0.65	0.65	0.65	0.64	0.64	0.64	0.64	0.64	0.65	0.64	0.63	0.64	0.64	0.64	0.63	0.65	0.66	0.65	0.65	0.65	0.66	0.66	0.65	0.65	0.65	0.64	0.64	0.65	0.65
		Purge	Gas (KSCFH)	69.69	76.8	74.6	77.2	88.7	89.1	99.3	100.4	87.6	66.3	68.9	63.4	61.2	62.1	65.0	69.2	73.0	69.5	78.6	65.0	69.8	71.9	74.4	74.7	77.0	87.6	62.3	65.5	59.0	63.9	96.6	60.8	64.7	72.6	83.0	74.6	66.6
		Reactor	Feed (H2:CO)	2.71	2.67	2.69	2.71	2.70	2.66	2.57	2.57	2.52	2.57	2.65	2.89	2.88	2.85	2.76	2.72	2.79	2.46	2.80	2.69	2.59	2.59	2.58	2.64	2.60	2.62	2.72	2.68	2.75	2.71	2.53	2.78	2.74	2.64	2.65	2.67	2.68
		Recycle	Gas (KSCFH)	2,234	2,220	2,216	2,233	2,220	2,209	2,213	2,195	2,205	2,245	2,238	2,214	2,213	2,213	2,210	2,252	2,250	2,237	2,186	2,218	2,249	2,232	2,214	2,277	2,300	2,292	2,283	2,258	2,286	2,296	2,291	2,214	2,239	2,225	2,232	2,233	2,255
		Fresh	Feed (KSCFH)	654	654	654	655	653	650	651	659	655	589	576	574	578	577	577	581	578	569	556	545	550	550	552	556	538	553	537	541	543	543	497	569	536	541	555	540	510
			Pres. (psig)	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700	700
			Temp (Deg C)	235	235	235	235	235	235	234	235	235	234	234	235	235	235	235	235	234	234	234	235	235	235	234	234	234	234	235	235	235	235	235	240	240	240	240	240	242
			Gas Type	Balanced	3alan ced	3alan ced	3alan ced	3alanced	3alanced	3alanced	Balanced	Balanced	3alanced	Balanced	Balanced	Balanced	Balanced	Balanced	3alanced	3alanced	3alanced	Balanced	3alanced	3alanced	3alan ced	3alanced	3alanced	3alanced	3alanced	Balanced	Balanced	Balanced	Balanced	3alan ced	3alanced	Balanced	Balanced	3alanced	3alanced	3alan ced
		Days	On Stream	226	227	228	229	230	231	232	233	234	235	236	237	238	239	240	241	242	243	257	258	259	260	261	262	263	264	265	266	267	268	277	278	279	280	281	282	283
			Date	Anr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	Apr-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	May-02	Jun-02	Jun-02	Jun-02
			ase	00-8 7-7	00-8 8-7	00-8 9-7	00-8 10-	-11 8-00	00-8 12-	00-8 13-	00-8 14-	00-8 15-	00-8 16-	00-8 17-	00-8 18-	00-8 19-	00-8 20-	00-8 21-	00-8 22-	00-8 23-	00-8 24-	V-8 8-00	00-8 9-N	00-8 10-	-11 8-00	00-8 12-	00-8 13-	00-8 14-	00-8 15-	00-8 16-	-21 8-00	00-8 18-0	.00-8 19-	00-8 28-	00-8 29-	00-8 30-	00-8 31-	.00-8 1	00-8 2	00-8 3
			0	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20	20

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

		_	1																																										1	
	Canada	Resist	("K")	8.75	8.56	8.49	8.32	8.46	8.41	8.37	8.51	8.43	8.41	8.39	8.40	8.24	8.19	8.15	8.13	8.13	8.27	8.25	8.30	8.23	8.22	8.25	8.32	8.33	8.34	8.33 07 9	8.31	8.33	8.35	8.32	8.34	8.27	8.25	8.41	8.25	8.59	9.14	9.28	9.17	9.05 8.80	7.30	7.75
	Change	dP	(psi)	5.78	5.83	5.76	5.93	5.78	5.77	5.72	5.81	5.81	5.92	6.02	6.08	6.28	6.41	6.52	6.60	6.58	6.26	6.52	6.36	6.65	6.48	6.53	6.26	6.24	6.28	6.34 6.42	6.36	6.32	6.25	6.29	6.24	6.43	6.55	6.06	6.56	5.35	4.42	4.27	4.46	4.58 4.90	5.04	4.44
	U I	(Bhi	hr ft2F)	132	126	127	126	125	119	113	119	113	118	117	117	115	115	114	115	117	125	124	118	122	130	112	123	124	125	124	121	119	120	121	117	124	122	121	122	124	123	117	119	123 123	811	119
	an of our	d Prod	PD/ft3)	0.086	0.083	0.084	0.080	0.082	0.083	0.081	0.082	0.081	0.083	0.081	0.082	0.082	0.082	0.080	0.082	0.079	0.081	0.077	0.078	0.080	0.079	0.077	0.076	0.077	0.076	0.070 0.070	0.081	0.081	0.080	0.078	0.077	0.078	0.076	0.076	0.077	0.077	0.076	0.073	0.073	0.076 0.076	0.084	0.082
		- No	kg) (T																																											
	Cotolise	MeOH Pr	(gmol/hr-	11.60	11.68	11.66	11.00	10.87	10.89	10.80	10.83	10.91	10.86	10.49	10.48	10.45	10.52	10.52	10.62	10.56	10.51	10.58	10.49	10.53	10.53	10.54	10.50	10.49	10.46	10.01	10.45	10.39	10.38	10.39	10.34	10.36	10.35	10.45	10.52	10.48	9.94	9.81	9.92	10.04	11.00	10.79
	Raw	Prod	(TPD)	177.9	179.0	178.8	168.7	166.7	167.0	165.6	166.2	167.3	166.5	161.0	160.8	160.3	161.3	161.6	163.2	162.2	161.4	162.6	161.1	161.7	161.7	161.9	161.3	161.1	160.6	161.5	160.5	159.6	159.4	159.5	158.8	159.0	158.9	160.4	161.6	160.8	152.4	150.3	152.2	154.0 155.7	168.8	165.3
	Cranado	Util	(SCF/lb)	40.4	40.6	40.6	39.7	39.6	39.7	39.9	39.7	39.4	39.6	39.0	39.2	39.1	39.1	38.8	38.4	38.6	38.9	38.6	38.9	38.8	38.8	38.7	38.9	38.9	38.9	38.9 38.0	38.9	39.1	39.0	39.1	39.3	39.1	39.1	38.7	38.5	39.4	41.5	42.5	41.6	41.3 40.8	40.5	42.3
)	Reactor O T M	Conv.	(%)	28.5	28.4	28.2	27.1	26.8	26.7	26.7	26.8	26.8	26.5	26.0	25.9	25.9	25.6	25.9	26.0	26.1	25.8	26.1	25.8	25.9	26.2	26.2	26.0	25.9	25.9	6.07	25.6	25.4	25.5	25.6	21.2	25.6	25.6	25.8	25.8	26.3	25.7	25.6	25.7	25.8 25.8	26.4	25.9
•	ę	Conv	(%)	52.4	51.1	51.1	50.3	50.5	49.8	50.2	50.9	50.0	48.0	47.8	47.0	45.5	44.5	44.2	42.6	43.4	44.8	44.1	44.0	41.9	44.1	43.9	45.1	45.1	8.4	43.6	41.4	40.9	41.7	42.0	42.2	41.4	40.5	43.5	39.8	48.8	54.9	56.0	54.9	54.2 51.5	44.5	49.2
,	-	Catalyst Age	(eta - corr)	1.159	1.095	1.090	2.090	3.090	4.090	5.090	6.090	7.090	8.090	060.6	10.090	11.090	12.090	13.090	14.090	15.090	16.090	17.090	18.090	19.090	20.090	21.090	22.090	23.090	24.090 25.000	060.62	27.090	28.090	29.090	31.090	32.090	33.090	34.090	35.090	36.090	37.090	38.090	39.090	40.090	41.090 42.090	43.090	44.090
	ato lunt	Age	(eta)	1.231	1.226	1.220	1.181	1.141	1.107	1.093	1.115	1.102	1.074	1.067	1.043	1.023	1.015	1.021	1.020	1.045	1.057	1.073	1.039	1.027	1.067	1.020	1.063	1.055	1.044	1.027	0.996	0.980	0.988	6660	0.984	0.986	0.976	1.011	1.002	1.037	1.018	1.015	1.014	1.019	666.0	0.992
	ter les	entory	(lb)),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040	040	040	0,040	0,040),040),040),040),040),040),040),040),040),040),040),040),040),040),040),040
)	č,	ascu Ce	(ii)	5.6	1.5 4	.5 4	0.0	8.5 44	8.0 4	9.0	8.5 4	9.0 44	8.0 4	7.5 4	5.5	5.5	7.0 4	8.0 44	7.5 41	9.0 44	7.5 44	0.0	9.5 44	8.0 44	9.0 44	0.0 44	.5 4	9.0).5 4 4	6 4 4 4	15 4	7.0 4	7.5 4	4 0.6	9.0	8.5 4	5. 4	.5 4	0.0 44	9.5 44	7.5 4	9.0	8.5 44	8.5 9.0 4 4	8.0 44	8.0 44
	Ű	dun Sh	1%) Hg	94	1.0 5.	.6 5(2 20	5 4	1 48	44 74	+.0 45	.7 49	.0	9	4	4	4	9 48	1.5 47	8.	4	.1 5(94	.1 46	2 4	.0 5(.0 5(.4 5(. 5 . 5	0 - 4 4	F 4	.9 4	8.	.8	2	1.0 45	.6		50 50	.7 49	4	.5 49	1.7 48	4 vi 4 4	.5 48	.4
	i	Hol H	ox) (vo	0 25	3 24	7 23	8 23	5 23	7 22	4 23	9 24	2 22	5 22	7 21	4	7 23	1 22	8 22	6 24	6 23	1 23	8 23	7 21	8 23	0 25	5 25	5 23	9 23	24	7 73 73	8 6 8 6	0 21	7 21	2 22	7 21	9 24	2 23	3 22	7 22	9 22	1 20	8	8 20	0 21 7 21	3 21	3 21
•	G	v Suu	g) (wt%	39.	37.	37.	37.	38.	38.	38.	38.	38.	38.	38	39.	39.	39.	38.	39.	38.	39.	37.	37.	38.	39.	38.	37.	37.	36 39	38. 38. 38.		39.	38.	38.	37.	38.	38.	37.	37.	37.	38.	36.	37.	38. 37.	38.	38.
	Canon	. Veloci	(l/hr-k	2723	2744	2750	2764	2747	2755	2739	2735	2751	2765	2763	2766	2782	2828	2822	2831	2814	2789	2806	2791	2805	2793	2793	2769	2771	2773	78/7	2785	2774	2767	2772	2766	2771	2780	2742	2788	2664	2518	2484	2520	2546 2595	2766	2695
	Inlat Cure	Velocity	(ft/sec)	0.64	0.65	0.65	0.65	0.65	0.65	0.65	0.65	0.65	0.66	0.65	0.66	0.66	0.67	0.67	0.67	0.67	0.66	0.66	0.66	0.66	0.66	0.67	0.65	0.66	0.66	00.00	0.66	0.66	0.65	0.66	0.65	0.65	0.66	0.65	0.66	0.63	0.59	0.59	0.60	0.60 0.61	0.65	0.64
	Diverse	Gas	(KSCFH)	48.9	48.6	49.2	32.6	34.0	35.3	34.4	34.2	32.7	34.2	27.8	28.4	26.6	26.8	25.1	23.0	22.1	25.4	23.5	25.4	24.9	23.9	23.6	24.8	24.8	24.4	26.0	26.6	27.9	26.9	26.5	29.6	29.2	28.5	27.8	25.3	32.4	59.3	69.3	58.2	55.6 49.3	47.0	72.0
	Decoders	Feed	(H2:CO)	3.81	3.69	3.71	3.82	3.91	3.86	3.90	3.93	3.85	3.70	5.01	3.69	3.53	3.48	3.40	3.22	3.27	3.46	3.34	3.39	3.15	3.33	3.31	3.46	3.48	3.46	3.34 3.10	3.16	3.14	3.21	3.23	3.27	3.15	3.07	3.34	2.94	5.37	4.62	4.77	4.62	4.52 4.23	3.41	3.99
	Docelo	Gas	(KSCFH)	1,288	1,305	1,308	1,366	1,353	1,353	1,353	1,337	1,354	1,368	1,389	1,391	1,402	1,434	1,431	1,434	1,417	1,406	1,418	1,407	1,420	1,404	1,410	1,397	1,396	1,395	1 410	1,408	1,407	1,401	1,401	1,401	1,408	1,410	1,384	1,412	1,323	1,224	1,194	1,215	1,235 1,271	1,339	1,279
	Dench	Feed	(SCFH)	599	606	605	558	550	553	550	550	550	550	523	525	522	525	523	522	522	523	523	523	522	522	523	523	523	520	520	520	520	518	520	520	518	518	518	518	528	527	532	528	530 529	569	583
		Pres	(psig) (1	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	750	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450	450 450	450	450
		Temn	Deg C)	214	214	214	214	214	215	214	215	215	214	214	214	215	215	215	215	215	215	215	215	215	215	215	215	215	215	515	215	214	215	215	215	215	215	215	215	215	214	214	215	215 215	215	215
		Gas	Type (lanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	ılanced	ılanced	alanced	alanced	alanced	alanced	alanced	alanced	ılanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	alanced	lanced	ılanced	ılanced	lanced	lanced	alanced	alanced	alanced
		s no	ream	1 B	2 Bi	3 Bi	4 Bi	5 B	6 Bi	7 B	8 Bi	9 Bi	10 Bi	11 Bi	12 Bi	13 Bi	14 Bâ	15 Ba	16 Ba	17 Ba	18 Ba	19 Ba	20 Bi	21 Bi	22 Bt	23 Bt	24 Bi	25 Bi	26 26	27 B	2 6 2 8	30 Bi	31 Bi	33 Bt	34 Bi	35 Bi	36 Bi	37 Bi	38 Bi	43 Bt	47 Bå	48 Bi	49 Ba	50 B B B	62 Bi	71 Bt
	L	-	te St	n-02	n-02	n-02	1-02	1-02	1-02	1-02	1-02	1-02	1-02	1-02	1-02	ıl-02	ıl-02	il-02	-1-02	1-02	-1-02	-1-02	-1-02	-1-02	-1-02	-1-02	1-02	ıl-02	1-02 . 22	1.02	70-11	1-02	1-02	-1-02	d-02	g-02	g-02	g-02	g-02	g-02	ıg-02	ıg-02	ıg-02	ıg-02 г02	ıg-02	2 -02
			Dai	3 28-Jui	8 29-Jui	8 30-Jui	8 1-Jul	8 2-Jul	8 3-Jul	8 4-Jul	8 5-Jul	8 6-Jul	8 7-Jul	8 8-Jul	8 9-Jul	8 10-Ju	8 11-Ju	8 12-Ju.	8 13-Ju.	8 14-Ju.	8 15-Ju.	8 16-Ju.	8 17-Ju	8 18-Ju	8 19-Ju	8 20-Ju.	8 21-Ju.	8 22-Ju	8 23-Ju	8 24-Ju 25 Iu	10-02 8	8 27-Jui	8 28-Ju.	8 30-Ju	8 31-Ju.	8 I-Au	8 2-Auչ	8 3-Auչ	8 4-Aսչ	ցսA-9 8	8 13-Au	8 14-Au	8 15-Au	8 16-Au 3 17-Au	8 28-Au	8 6-Sep
			Case	2000-{	2000-8	2000-	2000-1	2000-1	2000-1	2000-1	2000-1	2000-	2000-1	2000-	2000-	2000-	2000-1	2000-1	2000-	2000-1	2000-	2000-1	2000-	2000-	2000-	2000-	2000-1	2000-1	2000-	-0002	2000-	2000-{	2000-	2000-	2000-1	2000-1	2000-1	2000-	2000-1	2000-1	2000-1	2000-	2000-	2000-1	2000-8	2000-

Kingsport LPMEOHTM CCT Project

Final Report: Volume 2 (June 2003)

	Reactor Raw U Cataluer Construer CD O.T.M Summer MoOH Cataluer Basertor Duorell Sno	Age Age Conv. Conv. Util. Prod. McOHProd. Vol. Prod. (Btu d	(eta) (eta - corr) (%) (%) (SCF/lb) (TPD) (gmol/hr-kg) (TPD)/ft3) hr ft2 F) (pr	0.976 $45,090$ 46.2 25.9 42.0 166.4 10.86 0.083 119 4.0	0.960 46.090 47.2 25.6 42.4 163.7 10.69 0.083 119 4.5	1.005 47.090 46.8 26.3 41.7 169.1 11.04 0.084 123 4.7	1.011 48.090 46.7 26.6 41.7 169.2 11.04 0.085 125 4.	0.992 49.090 50.1 26.1 42.8 164.1 10.72 0.083 121 4.	1.027 50.090 50.2 26.8 41.9 169.9 11.08 0.086 125 4.2	1.044 52.090 52.8 26.8 42.6 167.3 10.92 0.081 123 4.0	1.031 53.090 50.9 26.9 42.1 169.0 11.03 0.081 123 4.2	1.036 54.090 51.8 26.8 42.4 167.8 10.95 0.082 123 4.	1.020 55.090 50.2 26.8 42.4 169.4 11.05 0.082 121 4.2	1.014 56.090 44.7 27.0 40.9 173.5 11.32 0.084 125 4.6	1.012 57.090 44.3 27.1 44.3 173.1 11.30 0.084 124 4.5 1.020 2000 240 200 105 105 1000 100 200		1025 59,090 49.7 26.8 41.4 171.1 11.15 0.085 127 4.2	1.020 59.090 50.1 26.8 41.8 169.7 11.07 0.083 125 4.2	0.976 59.090 44.5 26.3 41.3 170.9 11.14 0.085 125 4.5	0.956 59.090 42.7 26.4 42.4 170.3 11.11 0.084 121 4.9	0.965 59.090 43.0 26.5 42.4 170.3 11.11 0.085 124 4.5	1.031 59.090 48.9 27.4 45.7 177.8 11.63 0.085 124 4.2	0.996 59.090 42.3 27.0 41.9 176.4 11.51 0.082 123 5.	0.989 59.090 39.8 27.4 42.4 181.9 11.88 0.086 121 5.2 0.000 0.000 0.00 0.27 0.00 0.000 0.000 0.000 0.00	0.985 99.090 40.8 2/4 42.9 180.5 11.76 0.088 126 53. 0.986 59.090 42.3 27.4 43.0 180.1 11.76 0.088 125 4.	0.937 59.090 42.2 25.4 40.7 163.5 10.66 0.081 121 5.0	0.945 59.090 41.9 25.5 40.4 164.8 10.74 0.080 120 3. 0.958 59.090 476 26.4 40.2 165.7 10.79 0.081 123 5.	0.949 59.090 42.6 26.3 40.4 164.7 10.73 0.082 123 5.	0.955 59.090 41.7 25.6 40.4 165.1 10.75 0.084 127 5.	0.957 59.090 41.5 25.6 40.0 166.4 10.84 0.079 124 6.	0.936 59.090 43.4 25.4 40.9 164.2 10.70 0.079 123 5. 	NG CZI 1800 1C.01 219 21.5 121 10.00 10.02 10.00 10.01 10.00 10.00 10.00 10.00 10.00 10.00 10.00 10.00 10.00 10	0.970 59,090 44.2 26.3 41.5 166.8 10.87 0.083 128 5.2	0.929 59.090 41.2 25.6 40.6 166.5 10.85 0.083 126 6.	0.907 59.090 40.7 25.2 40.7 165.4 10.77 0.083 123 5.1	0.932 59.090 41.3 25.5 41.1 166.7 10.86 0.083 128 5.4	0949 59.090 42.3 25.8 41.1 166.7 10.87 0.080 127 5.0 0027 50.000 42.3 25.8 41.1 166.7 10.87 0.080 127 5.0	2, 20,000 18,01 6,001 1,15 0,07 7,57 0,000 6,000 1,000	0.942 59.090 45.3 24.8 39.1 174.8 11.38 0.085 132 6.5 0.012 59.000 43.2 24.6 302 174.7 11.37 0.085 130 5.5	0.902 59.090 43.8 24.3 39.4 172.9 11.26 0.085 129 6		0.929 59.090 44.3 24.8 38.9 174.6 11.37 0.085 133 5.8
D	Gae Gaesad Catalvet	Holdup Slurry Inventory	(vol%) Hgt (ff) (lb)	21.5 48.0 40.040	21.3 47.0 40,040	21.9 48.0 40,040	22.3 47.5 40,040	21.5 47.0 40,040	21.7 47.0 40,040	22.3 49.5 40,040	23.0 49.5 40,040	22.4 49.0 40,040	22.4 49.0 40,040	23.8 49.0 40,040	23.4 49.0 40,040	21.7 49.0 40.040	22.9 48.0 40.040	22.8 48.5 40,040	24.7 48.0 40,040	24.3 48.5 40,040	24.4 47.5 40,040	23.1 50.0 40,040	25.2 51.0 40,040	23.9 50.5 40,040	24.8 49.0 40,040 24.0 49.0 40,040	23.1 48.0 40,040	29.4 49.0 40,040 26.4 49.0 40.040	26.3 48.0 40,040	27.7 47.0 40,040	30.8 50.0 40,040	26.3 49.5 40,040	2/.5 40,040 C.14 40,040	30.2 48.0 40,040	30.8 48.0 40,040	26.9 47.5 40,040	30.0 48.0 40,040	21.1 49.5 40,040	30.1 49.0 40,040	30.6 49.0 40,040 30.5 40.0 40.040	29.6 48.5 40.040	000 000	25.5 49.0 40,040
3	Inlet Sun Snace Slurry	Velocity Velocity Conc.	(ft/sec) (1/hr-kg) (wt% ox)	0.65 2732 38.3	0.64 2702 38.8	0.65 2736 38.4	0.64 2726 38.8	0.62 2638 38.8	0.63 2666 38.9	0.62 2610 37.8	0.62 2643 38.0	0.62 2419 38.1	0.63 2656 38.1	0.65 2749 38.5	0.65 2738 38.4	6.15 0057 10.0 0.67 7000	0.64 2695 38.8	0.63 2664 38.5	0.66 2788 39.4	0.65 2759 38.9	0.65 2737 39.5	0.64 2692 37.8	0.66 2782 38.0	0.66 2808 37.8	0.65 2741 38.6 0.65 2741 38.6	0.65 2741 38.8	0.65 2753 40.5 0.65 7752 39.4	0.65 2747 39.9	0.65 2756 40.9	0.66 2775 40.4	0.65 2738 39.1	C.04 60/2 40.0 C.14 2775 A.0.0	0.64 2715 41.3	0.66 2783 41.5	0.66 2782 39.4	0.65 2771 41.2	0.65 2750 37.4	0.64 2/1/ 40./	0.65 3068 40.9	0.66 3076 40.8		0.65 3060 39.0
	h Revuela Reactor Durga	1 Gas Feed Gas	TH) (KSCFH) (H2:CO) (KSCFH)	1.306 3.68 66.6	1,287 3.84 72.2	1,309 3.66 68.4	1,279 3.61 67.0	1,223 4.05 80.6	1,256 3.95 73.2	1,189 4.19 77.0	1,223 3.99 71.2	1,206 4.26 76.7	1,232 3.92 71.9	1,295 3.33 58.3	1,298 3.28 57.4	1 208 4 25 84 4	1.264 3.90 67.1	1,243 3.94 70.2	1,334 3.47 58.5	1,312 3.24 73.9	1,303 3.24 76.8	1,190 3.70 129.3	1,322 3.09 69.2	. 1,331 2.78 82.4	1,248 2.89 88.8 1,248 3.04 93.0	1,343 3.28 49.8	1,544 5.2.3 48.1 1344 3.28 463	1,342 3.29 47.9	1,349 3.18 45.4	1,352 3.15 43.0	1,333 3.42 50.7	1304 350 605	1,308 3.38 60.2	1,371 3.19 53.6	1,360 3.20 59.0	1,342 3.18 55.1	1,336 3.23 54.0	1,318 3.2/ 3/.8	1 1,554 3.64 29.3 1 560 3.47 29.0	1.553 3.58 30.9	1 547 2 52 20.0	2015 CC.C 04C,1
	Tras	Gas Temp Pres. Feet	Type (Deg C) (psig) (KSC)	Balanced 215 450 583	Balanced 215 450 578	Balanced 215 450 588	Balanced 215 450 588	Balanced 214 450 586	Balanced 214 450 594	Balanced 215 450 593	Balanced 215 450 593	Balanced 215 450 593	Balanced 215 450 598	Balanced 215 450 591	Balanced 215 450 591	Balanced 215 450 591 Balanced 215 450 591	Balanced 215 450 591	Balanced 215 450 591	Balanced 215 450 588	Balanced 215 450 602	Balanced 215 450 602	Balanced 215 450 678	Balanced 215 450 616	Balanced 216 450 645	Balanced 215 450 04: Balanced 215 450 646	Balanced 215 450 555	Balanced 215 450 555 Balanced 215 450 555	Balanced 215 450 555	Balanced 215 450 556	Balanced 215 450 555	Balanced 215 450 560	Balanced 215 450 55: Balanced 215 450 576	Balanced 215 450 577	Balanced 215 450 563	Balanced 215 450 573	Balanced 215 450 570	Balanced 215 450 570	Balanced 215 450 565	Balanced 215 500 57(Balanced 215 500 570	Balanced 215 500 568	275 COD 216 Provided 600	Lalancea 212 Dalancea
	Dave	On	Case Date Stream	2000-8 7-Sen-02 72	2000-8 8-Sep-02 73	2000-8 9-Sep-02 74	2000-8 10-Sep-02 75	2000-8 11-Sep-02 76	2000-8 12-Sep-02 77	2000-8 14-Sep-02 79	2000-8 15-Sep-02 80	2000-8 16-Sep-02 81	2000-8 17-Sep-02 82	2000-8 18-Sep-02 83	2000-8 19-Sep-02 84	2000-8 21-Sep-UZ 80 2000-8 22 San 02 87	2000-8 23-Sen-02 88	2000-8 24-Sep-02 89	2000-8 25-Sep-02 90	2000-8 26-Sep-02 91	2000-8 27-Sep-02 92	2000-8 29-Sep-02 94	2000-8 1-Oct-02 96	2000-8 3-Oct-02 98	2000-8 4-Oct-02 99 2000-8 5-Oct-02 100	2000-8 25-Oct-02 120	2000-8 26-Oct-02 121 2000-8 37 Oct 02 133	2000-8 28-Oct-02 123	2000-8 29-Oct-02 124	2000-8 30-Oct-02 125	2000-8 31-Oct-02 126	2000-8 1-Nov-02 127 2000-8 4 Ni 62 120	2000-8 5-Nov-02 131	2000-8 6-Nov-02 132	2000-8 7-Nov-02 133	2000-8 8-Nov-02 134	2000-8 9-Nov-02 135	2000-8 10-Nov-02 136	2000-8 12-Nov-02 138 2000-8 12 Nov-02 120	2000-8 14-N0V-02 140	1000-2010-2010-2010-2010-2010-2010-2010	141 70-A0NI-CT 0-A0A7

Final Report: Volume 2 (June 2003)

D-31

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																		-	Reactor	-	Raw			n		
		Days	Gas	Temn	Pres	Fresh Feed	Recycle Gas	Reactor Feed	Purge Gas	Inlet Sup. Velocity	Space	Slurry Cone	Gas Holdun	Gassed Shurry 1	Catalyst 'nventorv	Catalyst A <i>oe</i>	Catalyst	CO	O-T-M	Syngas A	MeOH Prod Mi	Catalyst eOH Prod	Reactor Vol Prod	Overall	Sparger dP	Sparger Resist
Case	Date	Stream	Type	(Deg C)	(psig)	(KSCFH)	(KSCFH)	(H2:CO)	(KSCFH)	(ft/sec)	(l/hr-kg)	(wt% ox)	(vol%)	Hgt (ft)	(lb)	(eta)	Age (eta - corr)	(%)	(%) (1	SCF/lb) ((gi (gi	mol/hr-kg)	(TPD/ft3)	hr ft2F)	u (psi)	("K")
2000-8	19-Nov-02	145	Balanced	215	500	570	1,555	3.59	29.4	0.66	3071	38.8	26.3	50.0	40,040	0.900	59.090	43.9	24.6	39.3	174.2	11.34	0.083	130	6.08	7.99
2000-8	20-Nov-02	146	Balanced	215	500	570	1,535	3.67	31.0	0.65	3055	38.4	24.9	50.0	40,040	0.919	59.090	45.2	24.9	39.3	174.1	11.33	0.083	131	6.00	7.90
2000-8	21-Nov-02	147	Balanced	215	500	560	1,556	3.48	32.6	0.66	3076	40.1	28.4	49.0	40,040	0.897	59.090	43.0	24.7	39.1	171.9	11.19	0.084	132	7.03	8.72
2000-8	22-Nov-02	148	Balanced	215	500	585	1,537	3.59	35.3	0.65	3064	38.0	22.1	49.0	40,040	0.939	59.090	45.2	25.6	39.4	178.3	11.61	0.087	136	6.55	8.56
2000-8	24-Nov-02	150	Balanced	215	500	575	1,566	3.45	29.3	0.66	3098	40.3	28.3	48.5	40,040	0.904	59.090	43.0	24.8	38.9	177.5	11.56	0.087	134	6.97	8.96
2000-8	25-Nov-02	151	Balanced	215	500	550	1,529	3.89	32.5	0.64	3012	40.0	28.1	49.0	40,040	0.836	59.090	44.8	23.7	39.6	166.7	10.86	0.081	118	5.66	8.31
2000-8	26-Nov-02	152	Balanced	215	500	560	1,536	3.95	33.9	0.65	3027	38.4	23.3	49.0	40,040	0.852	59.090	45.5	23.3	39.6	169.6	11.04	0.082	121	5.94	8.84
2000-8	27-Nov-02	153	Balanced	215	500	550	1,582	3.57	30.6	0.66	3088	38.6	24.1	49.0	40,040	0.813	59.090	41.3	23.4	39.4	167.6	10.91	0.081	117	6.21	8.32
2000-8	28-Nov-02	154	Balanced	215	500	550	1,571	3.58	32.3	0.66	3077	40.5	29.5	49.0	40,040	0.809	59.090	41.3	23.4	39.4	167.4	10.90	0.081	116	6.32	8.59
2000-8	29-Nov-02	155	Balanced	215	500	535	1,581	3.26	32.0	0.66	3078	39.3	24.6	48.0	40,040	0.781	59.090	38.0	22.9	39.2	163.9	10.67	0.081	115	5.58	7.13
2000-8	30-Nov-02	156	Balanced	215	500	548	1,541	3.73	38.4	0.64	3020	42.4	30.1	46.0	40,040	0.794	59.090	42.3	23.2	39.9	164.5	10.71	0.085	120	6.30	9.21
2000-8	1-Dec-02	157	Balanced	215	500	538	1,594	3.35	32.7	0.66	3108	38.9	24.1	48.5	40,040	0.778	59.090	38.4	24.1	39.1	164.9	10.73	0.081	115	6.35	8.16
2000-8	2-Dec-02	158	Balanced	215	500	550	1,605	3.16	32.9	0.66	3108	38.9	24.1	48.5	40,040	0.791	59.090	37.5	23.3	39.3	167.8	10.93	0.082	116	5.82	7.30
2000-8	3-Dec-02	159	Balanced	215	500	548	1,584	3.16	30.6	0.66	3093	38.7	24.4	49.0	40,040	0.785	59.090	37.4	24.4	39.5	166.3	10.83	0.081	114	6.48	8.23
2000-8	4-Dec-02	160	Balanced	215	500	548	1,584	3.16	30.6	0.66	3106	38.3	23.2	49.0	40,040	0.799	59.090	37.9	23.3	39.0	168.6	10.97	0.082	118	6.31	7.96
2000-8	9-Dec-02	165	Balanced	215	500	535	1,596	3.32	32.8	0.66	3090	40.4	27.1	47.5	40,040	0.753	59.090	37.6	22.7	39.2	163.6	10.65	0.082	111	5.74	7.64
2000-8	10-Dec-02	166	Balanced	215	500	550	1,590	3.39	35.7	0.66	3092	37.3	20.0	49.0	40,040	0.782	59.090	39.0	23.1	39.4	167.7	10.92	0.082	112	5.08	6.84
2000-8	11-Dec-02	167	Balanced	215	500	548	1,584	3.15	29.6	0.66	3104	37.7	21.9	49.5	40,040	0.789	59.090	37.5	23.4	39.1	168.0	10.94	0.081	119	6.59	8.31
2000-8	13-Dec-02	169	Balanced	215	500	550	1,534	3.14	31.6	0.66	3079	37.9	20.3	48.0	40,040	0.807	59.090	38.1	23.7	38.9	169.6	11.04	0.084	123	6.36	8.05
2000-8	14-Dec-02	170	Balanced	215	500	548	1,578	3.28	31.7	0.66	3089	39.3	24.5	48.0	40,040	0.783	59.090	38.3	23.2	39.1	168.2	10.95	0.084	118	6.17	8.22
2000-8	15-Dec-02	171	Balanced	215	500	558	1,559	3.48	32.7	0.65	3059	38.4	23.3	49.0	40,040	0.814	59.090	41.1	23.8	39.3	170.4	11.09	0.083	121	5.80	8.07
2000-8	16-Dec-02	172	Balanced	215	500	560	1,544	3.66	36.2	0.65	3033	36.6	19.4	50.0	40,040	0.827	59.090	42.9	23.9	39.3	171.1	11.14	0.082	118	5.34	7.82
2000-8	23-Dec-02	179	Balanced	215	500	550	1,604	3.25	27.0	0.67	3122	37.6	24.8	51.5	40,040	0.808	59.090	38.4	23.3	39.2	168.3	10.96	0.078	115	6.43	8.17
2000-8	25-Dec-02	181	Balanced	215	500	551	1,569	3.57	35.3	0.65	3067	39.0	24.5	48.5	40,040	0.782	59.090	40.5	23.1	40.0	165.2	10.75	0.081	113	5.83	8.13
2000-8	26-Dec-02	182	Balanced	215	500	558	1,534	3.89	41.7	0.65	3045	37.7	19.5	48.0	40,040	0.797	59.090	43.3	23.3	40.3	166.1	10.81	0.082	113	5.47	8.23
2000-8	29-Dec-02	185	Balanced	215	500	545	1,526	4.02	36.7	0.65	3027	37.3	25.3	52.5	40,040	0.813	59.090	44.8	23.4	39.7	164.8	10.73	0.075	114	7.07	10.53
2000-8	30-Dec-02	186	Balanced	215	500	550	1,537	3.82	37.3	0.65	3025	37.2	24.2	52.0	40,040	0.813	59.090	43.6	23.7	39.6	167.0	10.87	0.076	115	5.81	8.52

D-32