

Fig. 30

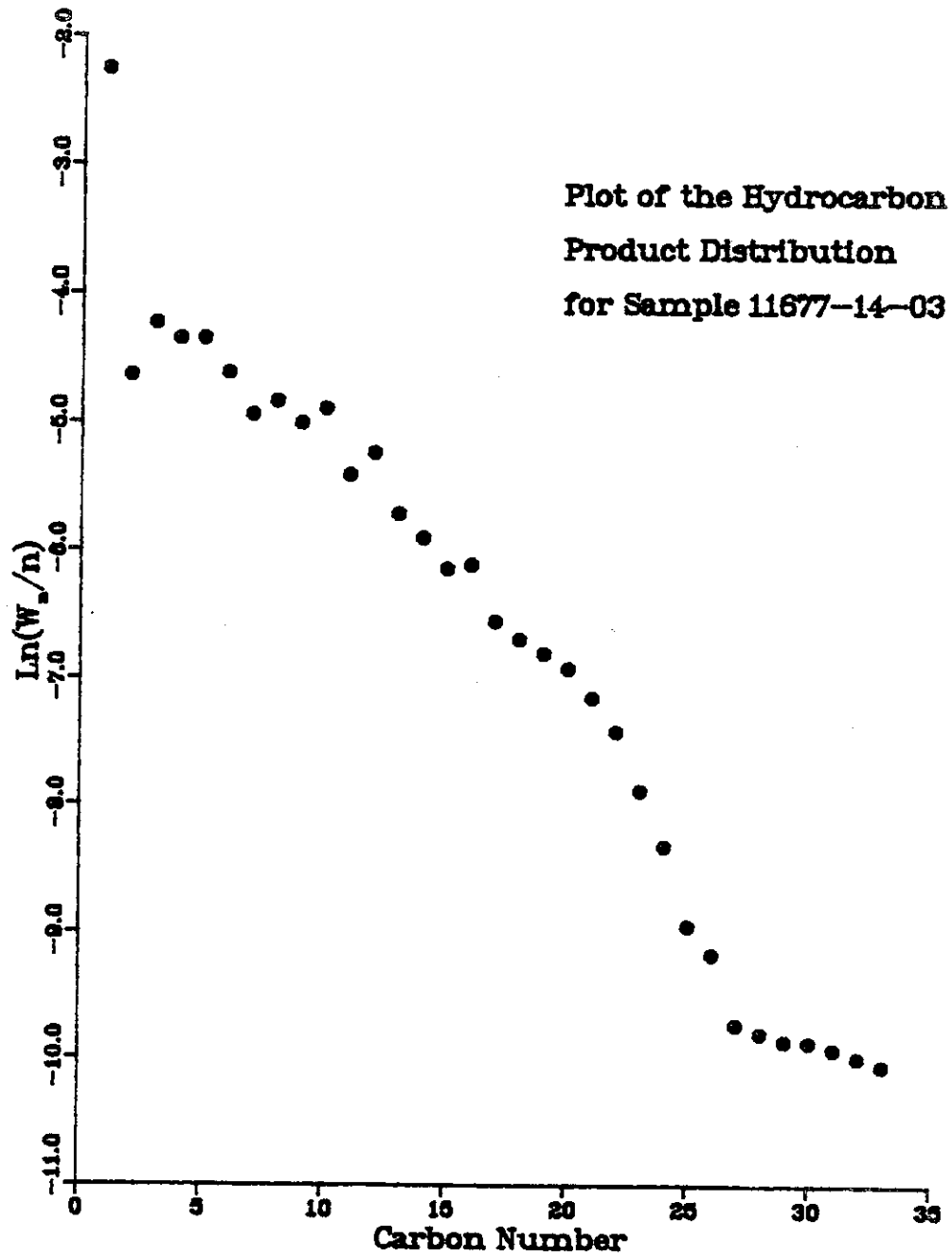


Fig. 31

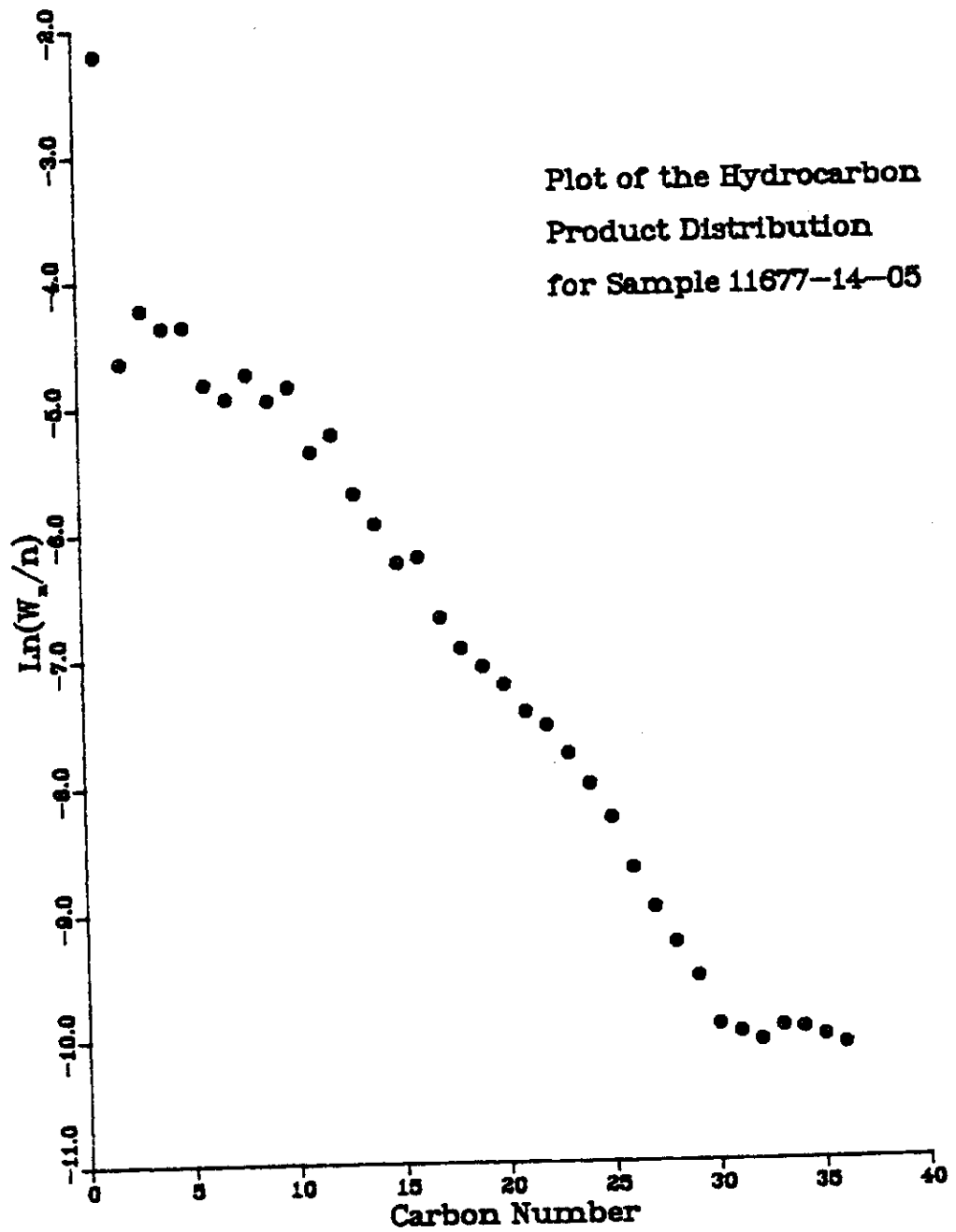


Fig. 32

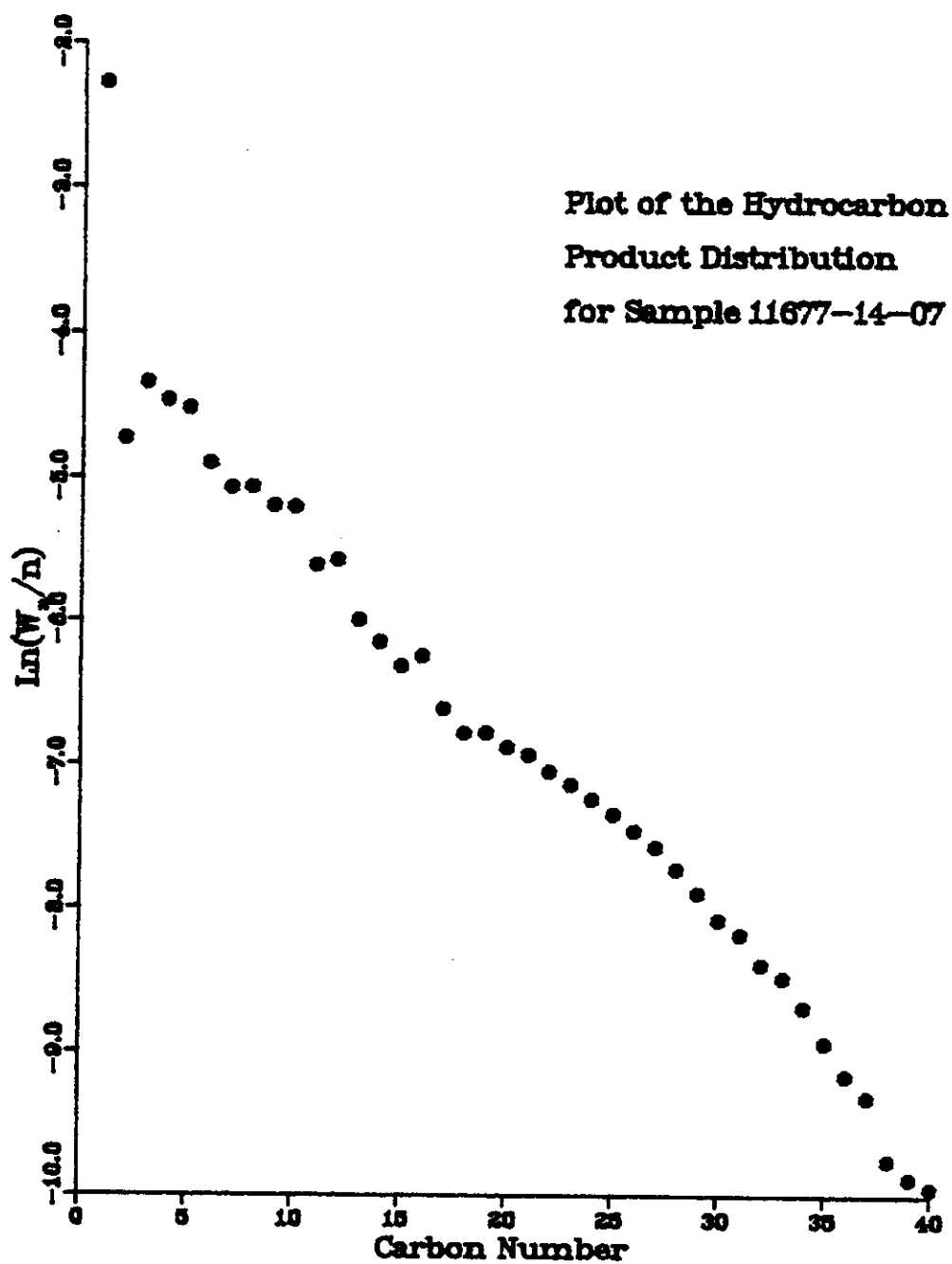


Table 5

RESULT OF SYNGAS OPERATION

RUN NO. 11677-14
 CATALYST HiCoThU103+U101 11684-70C 66 CC 26.5GM (31.8 AFTER RUN +5.3G)
 FEED H₂:CO:ARGON OF 50:50:0 @ 330 CC/MN OR 300 GHSV

RUN & SAMPLE NO.	11677-14-01	677-14-02	677-14-03	677-14-04	677-14-05
FEED H ₂ :CO:AR	50:50:0	50:50:0	50:50:0	50:50:0	50:50:0
HRS ON STREAM	19.5	26.0	44.0	49.5	68.5
PRESSURE, PSIG	299	303	299	304	303
TEMP. C	260	259	259	260	259
FEED CC/MIN	330	330	330	330	330
HOURS FEEDING	19.50	6.50	24.50	5.50	24.50
EFFLNT GAS LITER	122.45	45.70	179.75	42.50	194.60
GM AQUEOUS LAYER	51.39	17.09	64.41	14.00	62.37
GM OIL	19.28	6.92	26.08	5.53	24.62
MATERIAL BALANCE					
GM ATOM CARBON %	77.17	84.75	86.85	89.06	86.99
GM ATOM HYDROGEN %	81.45	86.43	88.97	89.16	91.52
GM ATOM OXYGEN %	88.64	93.70	94.49	96.01	94.50
RATIO CH _x /(H ₂ +CO ₂)	0.7232	0.7843	0.8136	0.8270	0.8108
RATIO X IN CH _x	2.2904	2.2963	2.3091	2.3259	2.3237
USAGE H ₂ /CO PRDCT	2.0825	2.0243	2.0421	2.0109	2.0648
FEED H ₂ /CO FRM EFFLNT	1.0555	1.0198	1.0244	1.0012	1.0520
RESIDUAL H ₂ /CO RATIO	0.2335	0.2486	0.2663	0.2773	0.3491
RATIO CO ₂ /(H ₂ +CO ₂)	0.1045	0.1034	0.0913	0.0992	0.0874
K SHIFT IN EFFLNT	0.0273	0.0287	0.0268	0.0305	0.0334
SPECIFIC ACTIVITY SA	6.8252	6.1339	5.6323	4.7041	3.4957
CONVERSION					
ON CO %	44.45	43.43	42.69	41.76	40.97
ON H ₂ %	87.71	86.21	85.10	83.87	80.41
ON CO+H ₂ %	66.66	65.03	64.15	62.83	61.19
PRDCT SELECTIVITY, WT %					
CH ₄	9.56	9.77	10.44	11.36	11.07
C ₂ HC'S	1.67	1.82	1.95	2.00	1.93
C ₃ H ₈	2.07	2.18	2.30	2.46	2.53
C ₃ H ₆ =	2.14	2.01	2.06	2.10	1.87
C ₄ H ₁₀	1.67	1.71	1.79	1.96	1.97
C ₄ H ₈ =	3.41	3.44	3.36	3.58	3.14
C ₅ H ₁₂	1.98	2.11	2.11	2.28	2.32
C ₅ H ₁₀ =	4.38	4.25	4.35	4.41	4.10
C ₆ H ₁₄	2.27	2.21	2.33	2.36	2.51
C ₆ H ₁₂ = & CYCLO'S	3.76	3.66	3.59	3.73	2.34
C ₇ + IN GAS	10.54	10.76	11.05	11.93	12.79
LIQ HC'S	56.55	56.09	54.67	51.83	53.42
TOTAL	100.00	100.00	100.00	100.00	100.00

Table 5 (continued)

SUB-GROUPING					
C1 -C4	20.52	20.92	21.90	23.46	22.52
C5 -420 F	50.72	48.65	48.45	49.23	49.30
420-700 F	25.55	27.56	26.86	23.06	23.79
700-END PT	3.21	2.87	2.79	4.26	4.39
C5+-END PT	79.48	79.08	78.10	76.54	77.48
ISO/NORMAL MOLE RATIO					
C4	0.1119	0.0882	0.0694	0.0694	0.0622
C5	0.1666	0.1532	0.1204	0.1217	0.1114
C6	0.3604	0.2526	0.2427	0.2059	0.2087
C4=	0.0569	0.0573	0.0572	0.0601	0.0601
PARAFFIN/OLEFIN RATIO					
C3	0.9246	1.0350	1.0649	1.1178	1.2895
C4	0.4727	0.4785	0.5158	0.5271	0.6057
C5	0.4399	0.4828	0.4713	0.5025	0.5513
SCHULZ-FLORY DISTRBTN					
ALPHA (EXP(SLOPE))	0.8090		0.8115		0.8246
RATIO CH4/(1-A)**2	2.6217		2.9395		3.5975
LIQ HC COLLECTION					
PHYS. APPEARANCE	CLDY		CLR OIL		OIL/SLD
DENSITY	0.7562		0.7553		0.7621
N, REFRACTIVE INDEX	1.4276		1.4276		1.4276
SIMULT'D DISTILATN					
10 WT % @ DEG F	263		278		281
16	302		304		304
50	426		444		437
84	599		618		627
90	648		657		681
RANGE(16-84 %)	297		314		323
WT % @ 420 F	49.14	45.75	45.75	47.25	47.25
WT % @ 700 F	94.33	94.89	94.89	91.78	91.78

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Table 6

RESULT OF SYNGAS OPERATION

RUN NO. 11677-14
 CATALYST HiCoThU103+U101 11684-70C 66 CC 26.5GM (31.8 AFTER RUN +5.3G)
 FEED H₂:CO:ARGON OF 50:50:0 @ 330 CC/MN OR 300 GHSV

RUN & SAMPLE NO.	11677-14-06	677-14-07
	=====	=====
FEED H ₂ :CO:AR	50:50: 0	50:50: 0
HRS ON STREAM	73.0	91.5
PRESSURE,PSIG	305	301
TEMP. C	260	260
FEED CC/MIN	330	330
HOURS FEEDING	4.50	23.00
EFFLNT GAS LITER	38.15	202.70
GM AQUEOUS LAYER	10.70	54.73
GM OIL	6.75	34.48
MATERIAL BALANCE		
GM ATOM CARBON %	98.65	104.67
GM ATOM HYDROGEN %	101.82	101.181
GM ATOM OXYGEN %	95.05	99.83
RATIO CHX/(H ₂ O+CO ₂)	1.0951	1.1289
RATIO X IN CHX	2.2866	2.3089
USAGE H ₂ /CO PRODT	1.8025	1.8115
FEED H ₂ /CO FRM EFFLNT	1.0322	0.9666
RESIDUAL H ₂ /CO RATIO	0.3830	0.3078
RATIO CO ₂ /(H ₂ O+CO ₂)	0.0992	0.0918
K SHIFT IN EFFLNT	0.0422	0.0311
SPECIFIC ACTIVITY SA	3.3850	4.4495
CONVERSION		
ON CO %	45.73	43.81
ON H ₂ %	79.87	82.11
ON CO+H ₂ %	63.07	62.64
PRDT SELECTIVITY,WT %		
CH ₄	9.10	10.29
C ₂ HC'S	1.50	1.77
C ₃ H ₈	2.10	2.16
C ₃ H ₆ =	1.44	1.76
C ₄ H ₁₀	1.62	1.68
C ₄ H ₈ =	2.51	2.95
C ₅ H ₁₂	1.97	1.97
C ₅ H ₁₀ =	3.10	3.49
C ₆ H ₁₄	1.97	1.96
C ₆ H ₁₂ = & CYCLO'S	2.42	2.52
C ₇ + IN GAS	10.10	8.93
LIQ HC'S	62.17	60.52
TOTAL	100.00	100.00

Table 6 (continued)

SUB-GROUPING		
C1 -C4	18.27	20.60
C5 -420 F	39.62	38.41
420-700 F	25.51	24.83
700-END PT	16.60	16.16
C5+-END PT	81.73	79.40
ISO/NORMAL MOLE RATIO		
C4	0.0634	0.0579
C5	0.1097	0.1005
C6	0.1877	0.1685
C4=	0.0624	0.0613
PARAFFIN/OLEFIN RATIO		
C3	1.3889	1.1740
C4	0.6221	0.5505
C5	0.6175	0.5496
SCHULZ-FLORY DISTRBTN		
ALPHA (EXP(SLOPE))	0.8646	0.8748
RATIO CH4/(1-A)**2	4.9661	6.5581
LIQ HC COLLECTION		
PHYS. APPEARANCE		OIL/SLD
DENSITY		0.7649
N, REFRACTIVE INDEX		SOLIDIF'D
SIMULT'D DISTILATN		
10 WT % @ DEG F		300
16		337
50		540
84		776
90		828
RANGE(16-84 %)		439
WT % @ 420 F	32.27	32.27
WT % @ 700 F	73.30	73.30

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IV. Run 3 (11723-14) with Catalyst 3 (Co/Th/UCC-101+UCC-101)

This catalyst is similar to Catalyst 1, except that UCC-101 is used in place of the intimately mixed UCC-103, resulting in a formulation that has UCC-101 present as both intimately and physically mixed components. It is similar also to Catalyst 1 of the Third Annual Report, except with a larger proportion of UCC-101. The final cobalt content was approximately 8.3 percent.

Conversion, product selectivity, isomerization of the pentane, and percent olefins of the C₄'s are plotted against time on stream in Figs. 33-36. Simulated distillations of the C₅⁺ product are plotted in Figs. 37-42. Carbon number product distributions are plotted in Figs. 43-54. Chromatograms from simulated distillations are reproduced in Figs. 55-65. Detailed material balances appear in Tables 7-10.

Throughout the 260 hour run the conversion of syngas fell off slowly and steadily. The rate of deactivation, one percent every 36 hours, was two-thirds that of Catalyst 1 in this Report and almost exactly the same as that of Catalyst 1 in the Third Annual Report, but higher than that of the X₄-containing Catalyst 6 of the Third Annual Report. The usage ratio was high, more than 2:1, showing very little water gas shift activity. In this catalyst the cobalt was not used as effectively as in Catalyst 1; the level of cobalt loading was the same in both catalysts but the

specific activity of this catalyst was lower.

The selectivity, although not as good initially as that of Catalyst 1, deactivated more slowly; by about 150 hours on stream the two patterns were essentially the same. From an initial level of 13.1 percent, the production of methane increased at the rate of one percent every 72 hours to 16.7 percent at 260 hours on stream. This was accompanied by the customary decrease in production of C_5^+ but at a lower rate than usual, resulting in a declining output of C_2-C_4 over time. The yield of motor fuels, about 69 percent initially, fell off at the rate of one percent every 70 hours. Gasoline production, which fell at one percent every 39 hours, was partially offset by an increase in the yield of diesel fuel at one percent every 87 hours, resulting in an increasing ratio of diesel to gasoline with time on stream.

The products were considerably olefinic, with about 70 percent of the C_4 's in the form of butenes, and more highly isomerized than with most catalysts which contain UCC-103. Catalyst 1 of the Third Annual Report, which also contained UCC-101 in place of UCC-103, likewise yielded more highly isomerized pentanes than did the UCC-103 catalysts. Isomerization of the pentane is confirmed both in the liquid product and in the chromatograms from the simulated distillations; with time on stream, however, it decreased ultimately to its usual low level. Except for the excess of methane, plots of the carbon number product distributions follow the typical Schulz-Flory pattern.

While this is by no means the most stable catalyst yet test-

ed, its slow, steady deactivation rate is nevertheless remarkable for the low ratios of hydrogen to carbon monoxide in the syngas to which it was exposed. It was active initially in a syngas with H₂:CO ratio of only 0.28:1, and although the ratio increased steadily, by the end of the run it still did not exceed 0.4:1. These are very low-hydrogen environments for the maintenance of catalytic activity; usually in such an environment a Fischer-Tropsch catalyst will rapidly coke and deactivate.

Fig. 33

RUN 11723-14

1:1 H₂:CO
300 PSIG
260°C

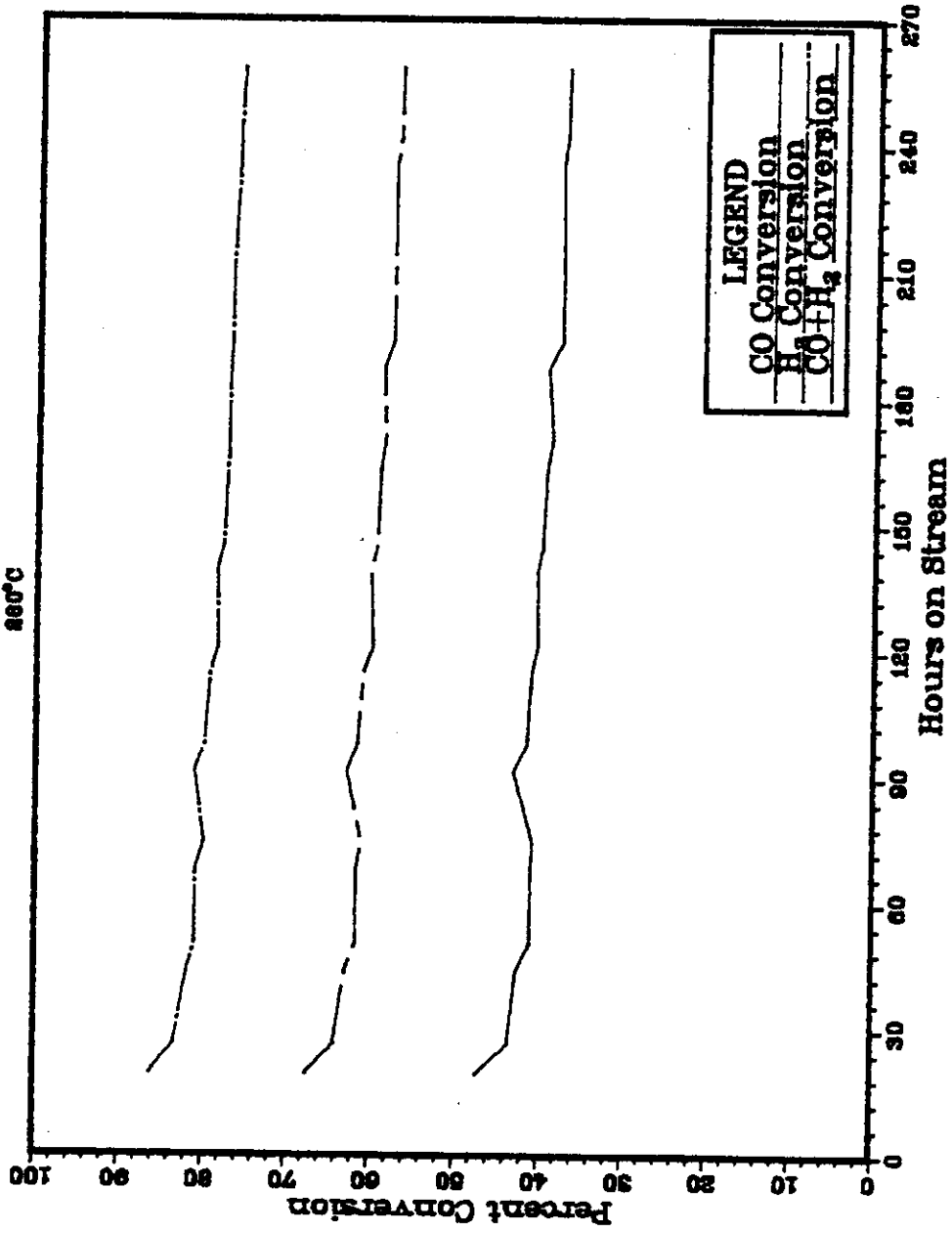


Fig. 34

RUN 11723-14

111 H₂CO
300 PSIG
260°C

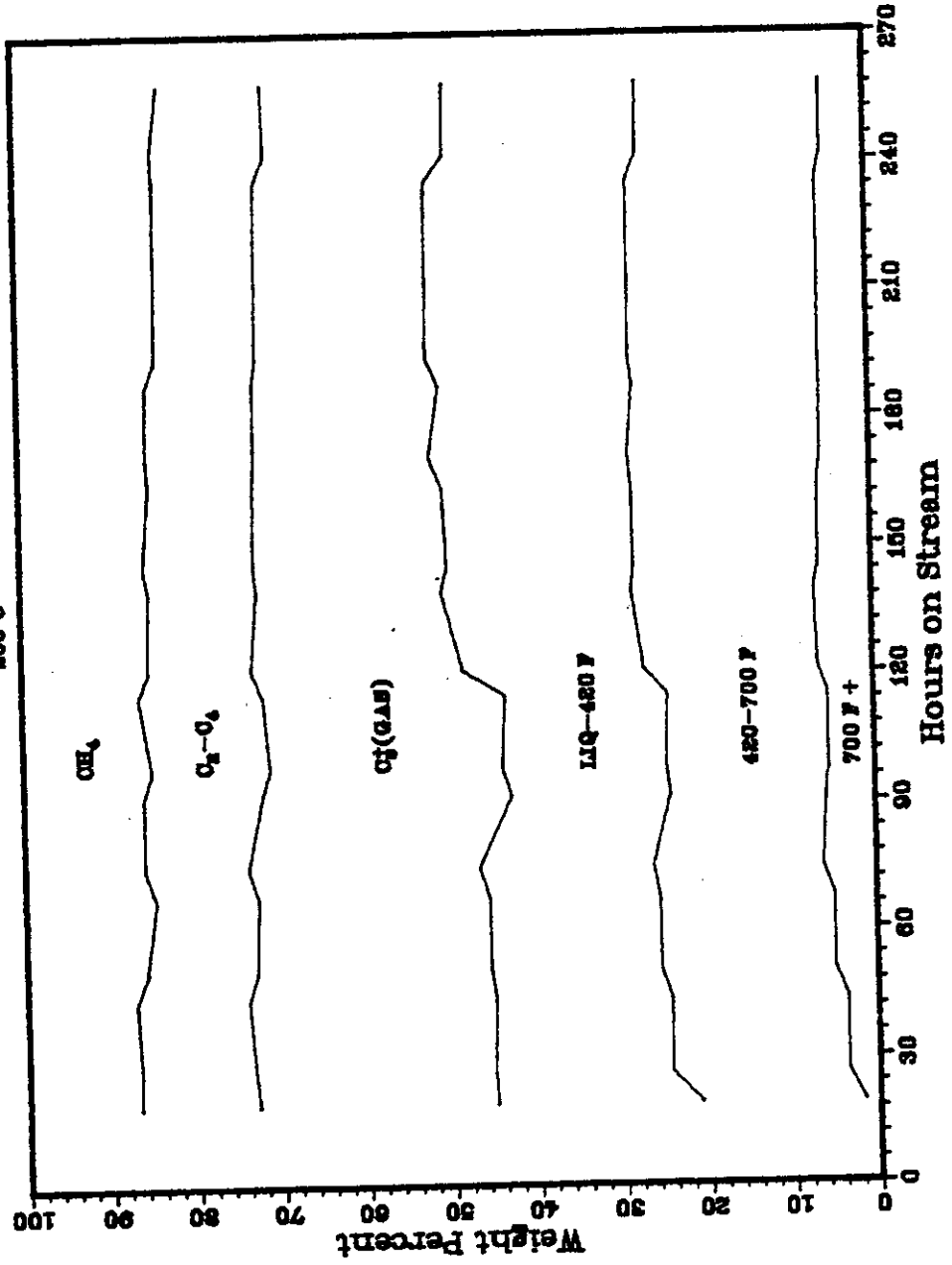


Fig. 35

RUN 11723-14

111 H₂O
300 PAIG
800°C

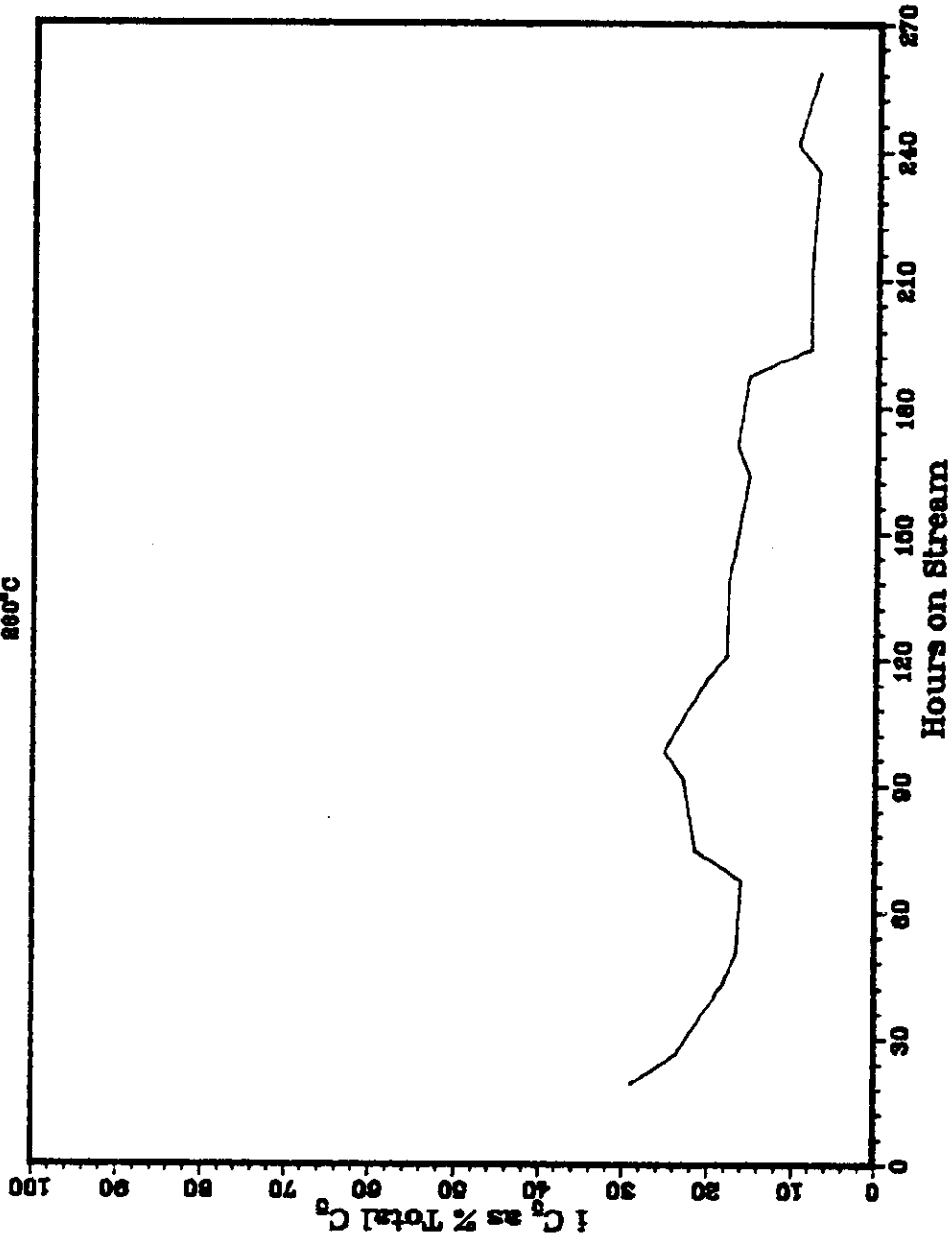


Fig. 36

RUN 11723-14

151 K₁CO
300 F₁10
260°C

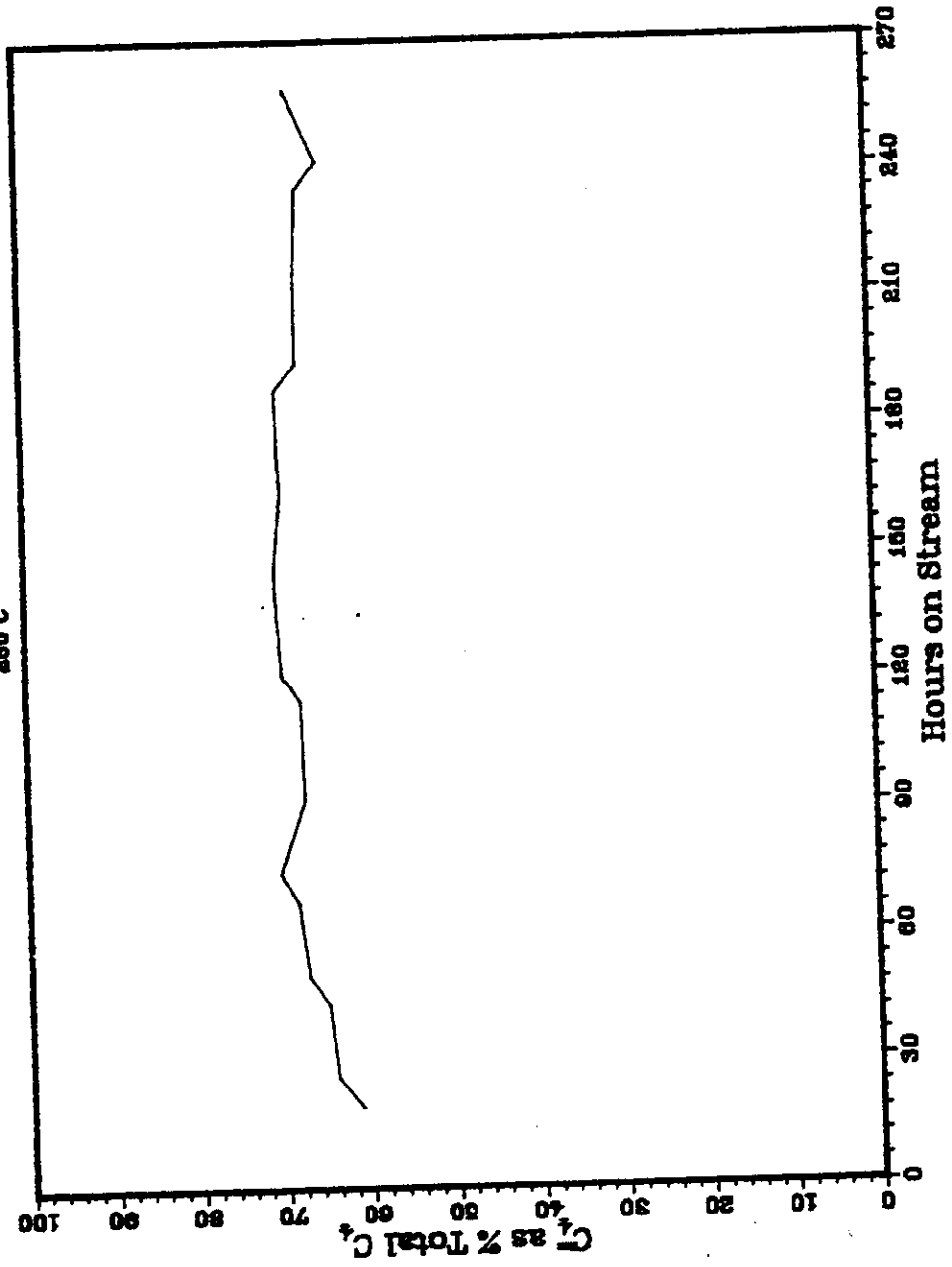


Fig. 37

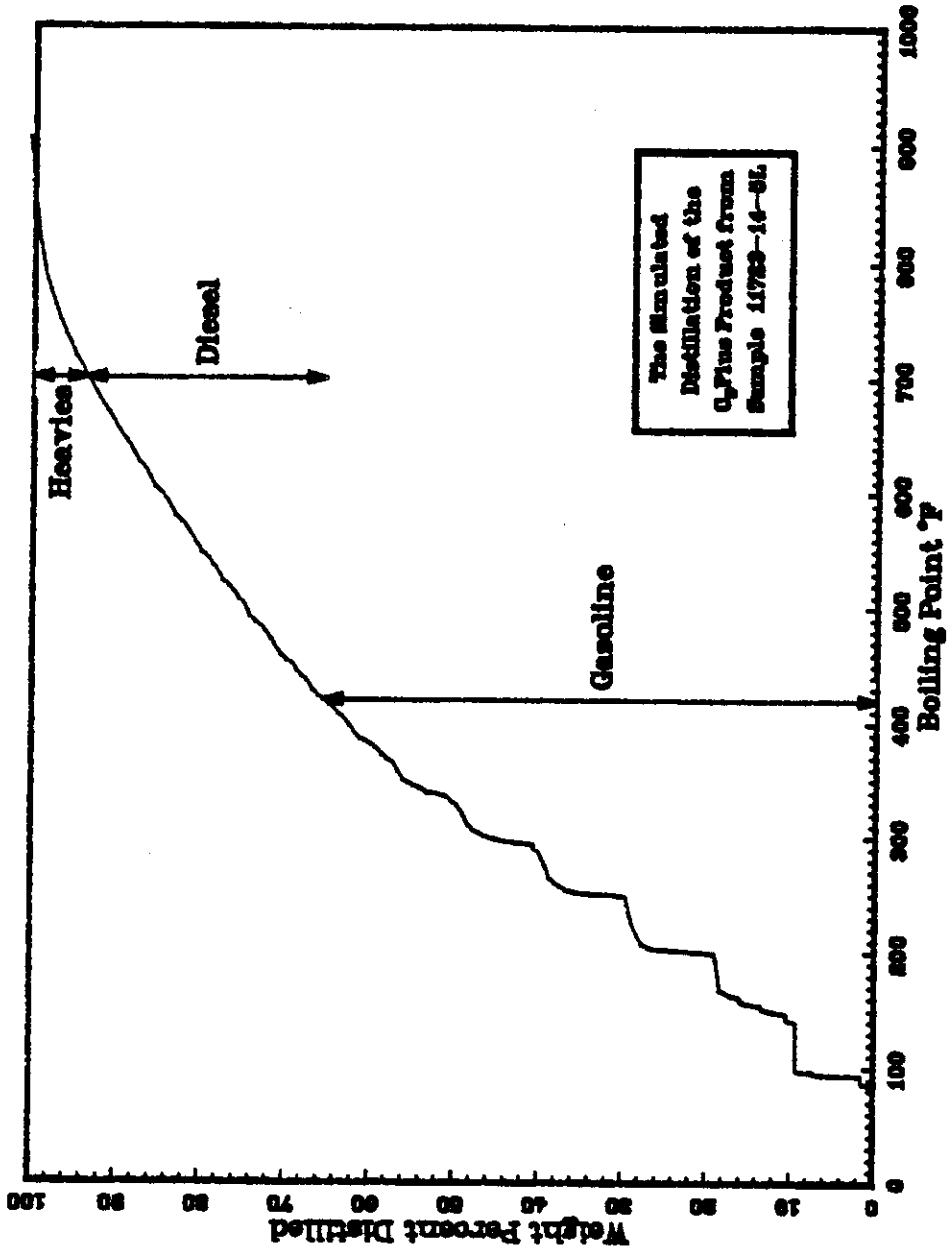


Fig. 38

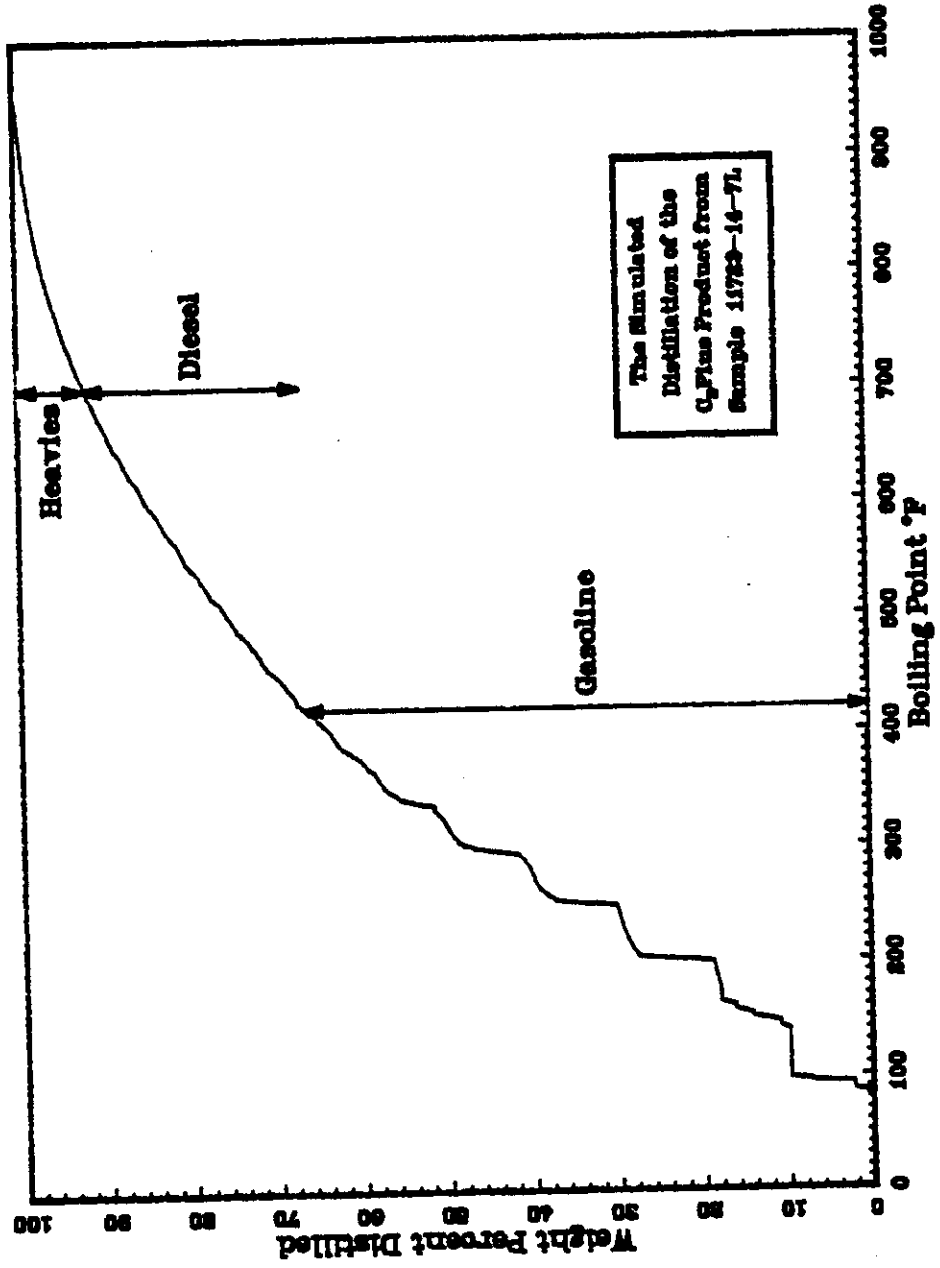


Fig. 39

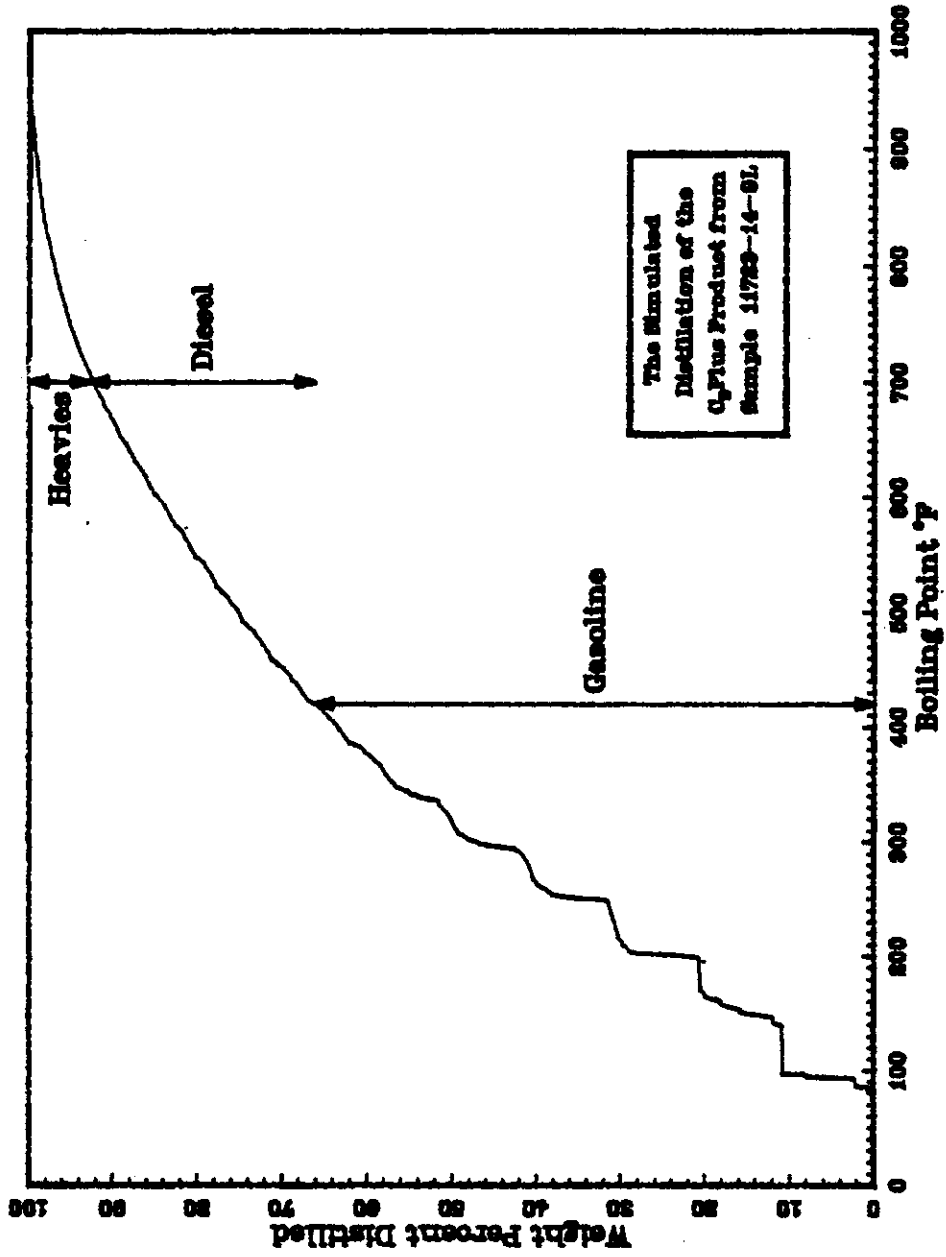


Fig. 40

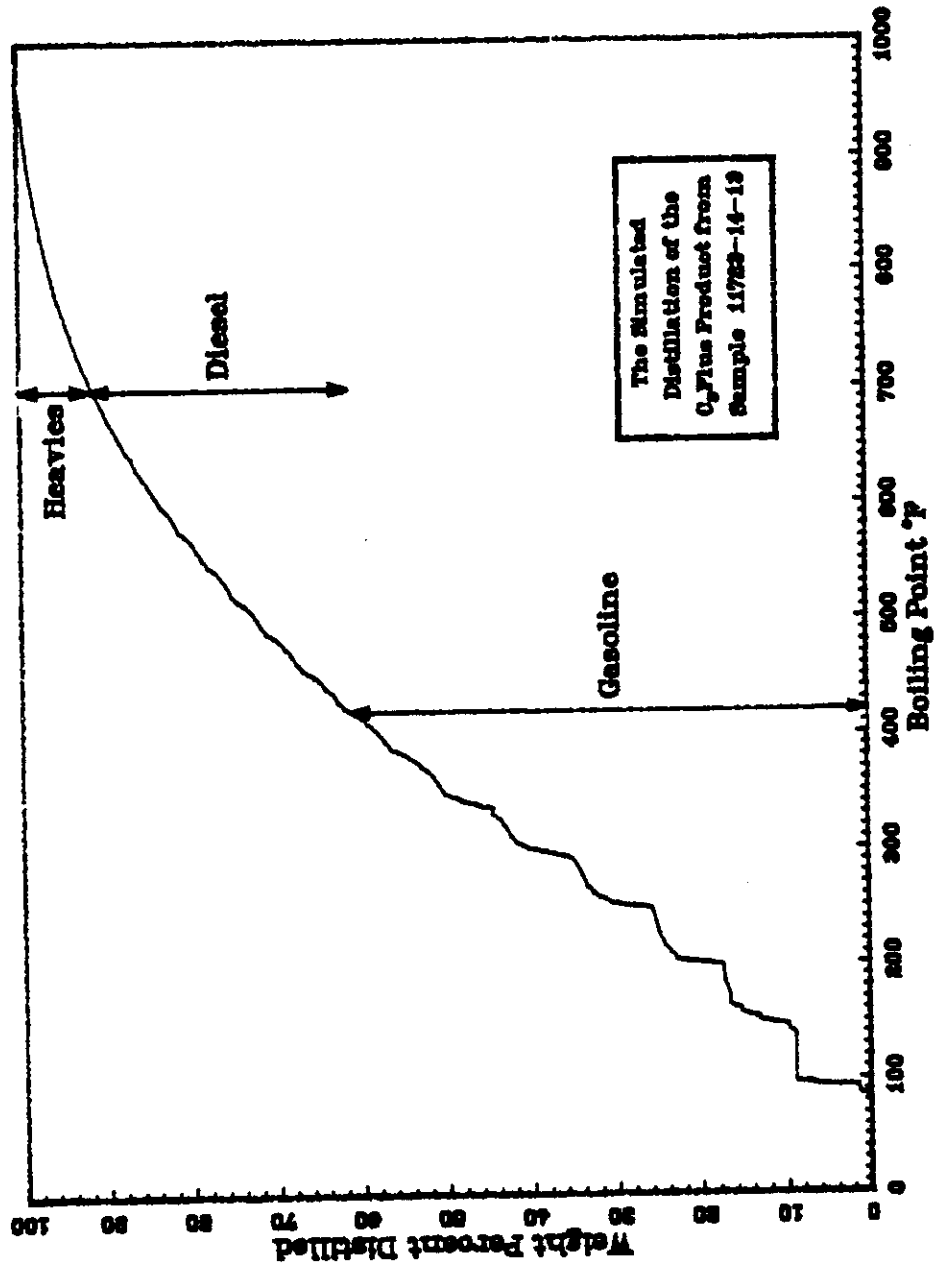


Fig. 41

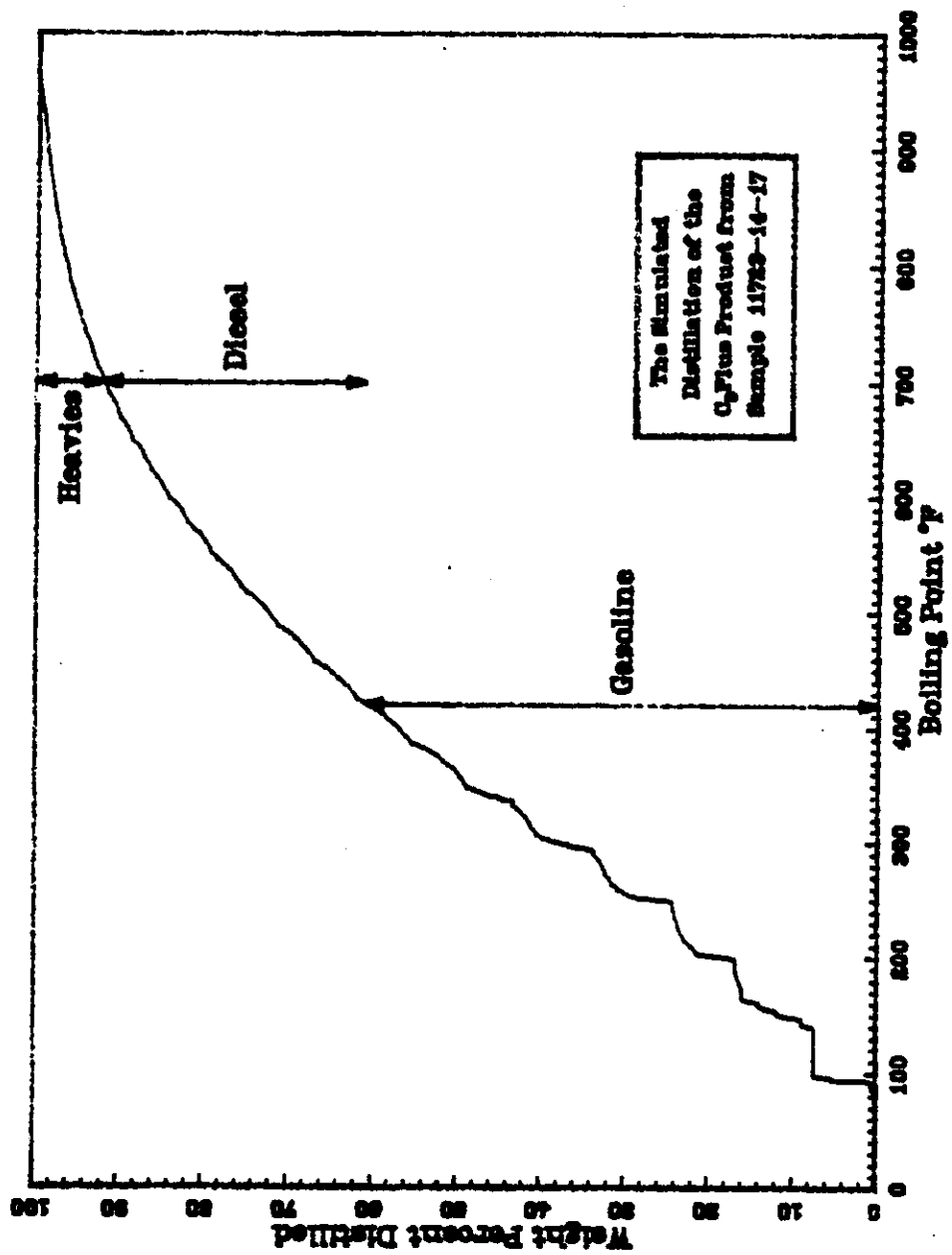


Fig. 42

