



DETERMINATION OF OPERATING PARAMETERS AND DISSOLVED OXYGEN CONCENTRATION PROFILE IN A TAPERED, FLUIDIZED BED

MASSACHUSETTS INST. OF TECH., OAK RIDGE, TENN. SCHOOL OF CHEMICAL ENGINEERING PRACTICE

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SUBJECT: Determination of Operating Parameters and Dissolved Oxygen Concentration Profile in a Tapered, Fluidized Bed

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ABSTRACT

Minimum and maximum operating flowrates, axial holdups, pressure and dissolved oxygen profiles were determined for a 4.5 m, 273 liter, tapered, fluidized bed. Coal particles, totaling 36, 50, 75 and 110 kg were fluidized with nitrogen and water. The solid and gas holdup profile was most uniform at the minimum liquid flowrate and a liquid to gas flowrate ratio of ten. Oxygen mass transfer, calculated for the 36 kg loading only, was greatest at the minimum gas and liquid flowrate with an overall mass transfer coefficient of $\sim 0.08 \text{ min}^{-1}$.

* Rewritten by W.M. Ayers

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1.0 SUMMARY

Operating parameters for a 273 liter, tapered, fluidized bed were obtained for 36, 50, 75 and 110 kg of coal particles fluidized with nitrogen and water. The parameters included the minimum and maximum operating flowrates, the axial and radial pressure and dissolved oxygen profile and the axial volume fraction (holdup) profile. The minimum operating flowrate was independent of coal loading and the solics holdup decreased with height for the two phase, water-coal system. The solids holdup was constant with height in the three-phase system at liquidto-gas flowrate ratios of approximately ten. Oxygen profiles were obtained at four flowrates for the 36-kg loading. Oxygen transfer from the water to the nitrogen was greatest at the lowest gas and liquid flowrates (5.6 and 13.1 2/min) with an overall mass transfer coefficient of ~C.08 min⁻¹. No radial oxygen of pressure variation could be detected at any axial position.

It is recommended that oxygen mass transfer be investigated at liquid-to-gas flowrate ratios less than one for several bed loadings. Decreasing the inlet bubble size and installing redistributor plates within the column are also suggested to improve the mass transfer.

2. INTRODUCTION

2.1 Background

A tapered, cocurrent, fluidized bed has advantages as a reactor vessel for biological processes. Fluidization prevents biomass from plugging the reactor, and the increase in column cross sectional area with height reduces the superficial velocity and therefore instabilities such as slugging. In previous studies, degradation of phenol and nitrates and gas-liquid mass transfer as a function of flow rates, particle size and solids loading, have been studied with a bench-scale, tapered, fluidized bed (10, 11). A 273-liter, tapered, fluidized bed reactor has recently been installed at the Chemical Technology Division of ORNL to investigate scale-up parameters associated with these processes.

Basic operating parameters for this reactor such as the minimum fluidization velocity, maximum fluid flow rates, phase volume fractions (holdups) and oxygen and pressure profiles along the bed are needed as a function of solids loading and liquid-gas flow rates prior to starting up the reaction.

The minimum fluidization velocity is the lowest liquid and gas superficial velocity necessary to fluidize the bed, and the maximum operating velocity is defined as the superficial velocities that will place the height of the fluidized bed at the top of the column. This velocity is less than terminal velocity or maximum fluidization velocity. The phase holdups or volume fractions are the relative amounts of solid, liquid, and gas in a given volume. Since the microorganisms grow on the solid phase (coal particles), the solids holdup is a measure of the amount of biomass per unit volume. Thus, the phase holdups are needed to predict optimum coal loadings and nutrient and oxygen concentrations in the inlet streams. The pressure profile is used to calculate the bed height and the volume fraction profiles.

For this investigation, coal particles (~30 to +60 mesh) were fluidized with water and nitrogen. Since the inlet water was saturated with oxygen, the oxygen mass transfer was from the liquid to the gas. When operating as a bioreactor, the oxygen transfer will be from the gas to the liquid.

2.2 Objectives

The objectives were to determine: (1) maximum and minimum operating liquid and gas flow rates as a function of solids loading, (2) axial and radial pressure and oxygen concentration variation through the reactor, (3) axial phase holdup variation, and (4) estimation of the gas-liquid mass transfer coefficient.

2.3 Method of Attack

Minimum fluidization and maximum operating velocity for the twophase, solid-liquid system were determined at coal loadings of 36, 50, 75, and 110 kg. Gas and liquid flow rates that fluidized the bed near intermediate and minimum fluidization levels were also determined at these loadings. Oxygen concentrations and pressure were determined with probes inserted into nine ports along the column and samples of the bed composition were also obtained from these ports to determine the phase holdups. A summary of the operating conditions for each experiment is presented in Table 1 of Section 3.

3. APPARATUS AND PROCEDURE

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As shown in Fig. 1, the 4.57m (15-ft) column consists of two 1.22 m, tapered sections separated by a straight section of equal length. A top section and the adjacent 0.21 m³ (55-gal) drum allows entrained solids to



settle. A recycle pump supplements the inlet water flow rate. However, the fresh water feed was used for all flow rates below 76 liter/min. Nitrogen flow rates were measured with an orifice meter.

The axial pressure profile was measured by connecting nine manometers to the ports on the side of the column. A tube, with several small holes near its end, could be inserted into the column, through the ports to measure the radial variation in pressure. The oxygen concentration was obtained by attaching a 40 ml-chamber containing a YSI dissolved oxygen probe to this tube. Water flow through the sample chamber was controlled with a valve so that a constant residence time could be maintained for different positions along the column.

To determine the volume fractions, 500 ml of coal slurry was taken from each sample port. The sample was weighed then dried to determine the mass of the coal particles. The dried coal was then mixed with a known volume of water and the final volume of the coal-water slurry was measured to determine the density of the coal. With this density and the density of water, the solid and liquid volume fractions for the sample were calculated from the initial sample weight. It was necessary to determine the coal density at each port since there was an apparent stratification of coal particle size with bed height. It was assumed that the ratio of solid-to-liquid volume fraction in the bed was the same as that in the sample. The volume fractions in the bed were then determined with this ratio and the pressure drop across that section of the bed. Details of the calculation are presented in Appendix 9.1.

Minimum fluidization velocities were obtained by monitoring the pressure drop through the bed as the gas and liquid flow was increased, until the pressure difference between the bottom and top pressure tap was constant. The maximum operating flow rates were determined by adjusting the liquid and gas flow rate until the bed could be seen in the top window. Liquid and gas flowrates and the fluidized bed height for all experiments are presented in Table 1.

At high gas velocities, entrainment of solids beomces a problem. A centrifugal pump was connected to the bottom of the collection drum to recycle solids back to the column.

4. RESULTS AND DISCUSSION OF EXPERIMENTAL CONDITIONS

4.1 Minimum and Maximum Operating Conditions

In Fig. 2 the maximum and minimum liquid superficial operating velocity (no gas flow) is plotted against coal loading. The velocity is calculated from the liquid flowrate and in the cross-sectional area at the bottom of the reactor. The minimum operating velocity is seen to be independent of

Solids Loading, M _S (Kg)	Flow Rates (1/min)		Bed Hé Static,Hbs	ight (cm) Fluidized,Hb	Degree of Fluidization Hb/Hbs	Axial Profile	Obtained
	Ļ	G				0 ₂	<mark>٤</mark> 5
36	13,6 52.3 60.7 88.4 13.1 48.3 62.1	0 0 0 5.6 5.2	104	140 270 330 360 141 217 280	1.3 2.6 3.2 3.5 1.3 2.1 2.7	X X Y Y Y	x × × × × ×
50	13.8 60.5 20.6 54.5	0 0 20.2 5.4	1 32	181 339 231 289	1.4 2.6 1.8 2.2	X X X	
75	13.7 28.7 42.2 13.4 12.3 31.9	0 0 1,4 3.2 3.2	182	213 291 335 230 234 319	1.2 1.6 1.8 1.3 1.3 1.3	X X X X X	//////////////////////////////////////
110	13,9 14,7 18,1 10,0 13,8 14,6	0 0 11.1 6.7 11.0	242	305 309 321 279 292 289	1.3 1.3 1.2 1.2 1.2	X X X X X X	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~

TABLE I: OPERATING CONDITIONS FOR EXPERIMENTS

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coal loading as is expected, since this velocity corresponds to the minimum fluidization velocity which is independent of the number of particles according to Wen and Yu (12). The maximum operating velocity, defined as the liquid flow rate at which the solid bed was at the top of the column, shows a decrease with increasing coal loading. The maximum operating velocity does not correspond to the terminal velocity, or the liquid velocity at which the particles leave the bed. The maximum operating velocity decreases with increasing coal loading since there are a larger number of particles in the bed as the solid loading increases.

The effect of gas flowrate on the minimum liquid fluidization velocity is presented in Fig. 3. Unfortunately, the influence of the gas flowrate at liquid flowrates much less than the minimum liquid fluidization rates (Fig. 2) was not determined. Thus the data in Fig. 3 only indicate the solids are chiefly fluidized by the liquid flowrate.

4.2 Pressure Profiles

The pressure drop across the bed increases until the fluid velocity is sufficient to suspend the particles. At this minimum fluidization point, the pressure drop across the bed is equal to the weight of the suspended particles. In Fig. 4 the pressure at each port minus its static bed pressure is plotted against column height at minimum fluidization for the four coal loadings. For each loading there is a distinct change in slope in the axial pressure profile. This change in slope corresponds to the solid bed height. The bed height at minimum fluidization increases with coal loading. This is expected since the static bed height also increases with coal loading.

At constant solids loading and no gas flow, increasing the liquid flow rate produces the pressure profile shown in Fig. 5. Once again, the change in slope of the profile corresponds to the height of the bed. Ideally, the pressure above the solid bed should be the same for all flow rates. However, limitations in the column draining capacity caused the water level above the bed to rise slightly with liquid flow rate.

The radial pressure variation at a constant liquid and gas flow rate and coal loading is shown in Fig. 6. The radial pressure readings are plotted against reduced radius (radial distance from center divided by the column radius at that height). Readings were taken with the probe at the bottom two ports (20 and 61 cm) and the port at the top of the middle section (223 cm). The vertical bars through the points represent the oscillation of the manometer reading during measurement. There was no measurable radial pressure variation at any of the probe positions.





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4.3 Volume Fraction Profiles

Volume fraction profiles were determined for 50, 75, and 110 kg loadings at each liquid and gas flow rate. Solid, liquid and gas volume fraction profile data are tabulated in Appendix 9.4. The variation of the solid volume fraction profile with loading and operating conditions is presented in Figs. 7 through 10. The solid fraction decreases with height at the minimum and maximum liquid operating velocities (no gas) as shown in Figs. 7 and 8, respectively. The volume fractions are, of course, smaller in Fig. 8 due to the larger bed volume. The introduction of gas into the bed made the distribution of solids more uniform at low gas flowrates (Fig. 9). At higher gas flowrates, the concentration of solids again decreased with height for 50 and 75 kg loadings but went through a maximum at approximately two-thirds the bed height for 110 kg (Fig. 10). The solids profile also went through a maximum at a lower liquid flowrate (10,0 %/min) and the same gas flow rate for the 110 kg loading (Appendix 9.4). The uniformity of solids distribution in Fig. 9 can probably be attributed to solids being drawn along in bubble wakes. Greatly increasing the liquid flowrate (e.g. 75 kg case, Fig. 10) at low gas rates led to a decrease in the solids profile again which might be due to bubble break-up as suggested by Michelsen and Ostergaard (8). Inability to set the gas at a desired level made it difficult to isolate the effect of this flowrate on the holdups. Repetition of these experiments with a more systematic variation of flowrates should be performed to develop a correlation.

4.4 Oxygen Concentration Profiles

Oxygen concentration profiles were only determined for the 36-kg loading. The axial concentration variation as a function of liquid and gas flowrate is presented in Fig. 11. The greatest transfer between the liquid and gas occurs at the lowest flowrates. The oxygen concentration in the exit gas at these flowrates is approximately three percent of the equilibrium value (Appendix 9.2). Thus, the transfer between phases could be further improved by decreasing the liquid flowrate, installing a sparger on the gas inlet line, and redistributors within the column.

Unfortunately, holdup data were only taken at one liquid and gas flowrate for this loading (48.3 and 5.6 ℓ/min). If the data in Fig. 9 are also representative of the 36 kg loading, the volume fraction profile should be more uniform at the lowest liquid and gas flowrates. This might also account for the greater mass transfer at these conditions.

As shown in Fig. 12, there was no significant radial concentration variation. Although it seems unlikely that such a variation would exist, the present method of drawing a sample from within the column through the detector cell might be too insensitive to detect it. Insertion of a probe into the column would provide a more accurate determination.

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There were significant problems in obtaining reproducible data with the oxygen concentration probe. The original Y.S.I. probe had an unsupported membrane that oscillated with the column pressure. This was replaced with an Instrumentation Laboratory probe that was accurate to approximately 0.5 ppm.

Although the oxygen concentration data are sparce, an estimate of an overall mass transfer coefficient was made assuming that each column section between sampling ports could be treated as a well mixed reactor. The values of KLa ranged from 0.02 to 0.14 min⁻¹ (Appendix 9.3). However, an order of magnitude estimate of 0.1 min⁻¹ is a more realistic volume when the uncertainty in the oxygen concentration measurements and assumption of the calculation are considered.

5. CONCLUSIONS

- 1. The minimum Operating flowrate is independent of loading and for this coal is approximately 14 1/min.
- 2. The maximum liquid flow rates are approximately 88, 60, 42, and 18 1/min at 36, 50, 75, and 110 kg coal, respectively.
- With no gas flow, the solids holdup decreases with height at both the minimum and maximum liquid flowrates.
- 4. The least variation of solid and gas holdup with height occurs at the minimum gas and liquid flowrate.
- 5. The maximum Oxygen mass transfer occurred at the lowest liquid and gas flowrate.
- 6. No radial variations in either pressure or oxygen concentration were detected.

6, RECOMMENDATIONS

- 1. Attempt to fluidize the bed at liquid-to-gas flowrate ratios (L/G) less than one.
- 2. Determine the oxygen concentration profile and mass transfer at low liquid and gas flowrates $(L/G \le 1)$ for several coal loadings.

- 3. Decrease the bubble size by installing redistributors and a sparger at the gas inlet.
- 4. Install a flowmeter on the gas inlet line to facilitate operating the column.
- 5. Install a cyclone separator to contain the solids if the column is to be operated at large flowrates.

7. ACKNOWLEDGMENTS

We wish to thank J. M. Begovich for his ideas, C. W. Hancher for his direction and laboratory expertise, and G. B. Dinsmore for his assistance in procurement of laboratory equipment.

8. LOCATION OF DATA

Original data are tabulated in notebook A-7555-G, p. 70-99, on file at the M.I.T. Practice School Office, Bldg. 3001, Oak Ridge National Laboratory.

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Therefore,

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$$K = \frac{\varepsilon_L}{\varepsilon_S} = \frac{(\rho_S - \rho_T)(\rho_S - \rho_L)}{(\rho_T - \rho_L)(\rho_T - \rho_L)} = \frac{\rho_S - \rho_T}{\rho_T - \rho_L}$$
(6)

Since ρ_L and ρ_T were known and since ρ_S was determined from the dried coal, K could be calculated. With K the volume fractions within the bed section can now be determined:

$$\varepsilon_{\rm G} + \varepsilon_{\rm L} + \varepsilon_{\rm S} = 1 \tag{7}$$

$$K = \frac{\varepsilon_L}{\varepsilon_S}$$
(8)

$$\Delta \rho = \left[\rho_{S} \varepsilon_{S} + \rho_{G} \varepsilon_{G} + \rho_{L} \varepsilon_{L} \right] g \Delta h \tag{9}$$

Therefore,

$$\varepsilon_{S} = \frac{\Delta \rho/g\Delta h - \rho_{G}}{K(\rho_{L} - \rho_{G}) + (\rho_{S} - \rho_{G})}$$
(10)

If $\rho_{G} < < \rho_{S}$ or ρ_{L} , Eq. (10) can be approximated with:

$$\varepsilon_{S} = \frac{\Delta \rho}{(K_{\rho_{L}} + \rho_{S})g\Delta h}$$
(11)

9.2 Approach of Gas and Liquid Oxygen Concentration. to Equilibrium

The oxygen mass transfer was greatest at 13.1 ℓ /min liquid and 5.6 ℓ /min nitrogen flowrate. The ratio of outlet liquid and gas concentration can be compared to the equilibrium value to determine if transfer between the phases was complete. From the data in Fig. 11, the liquid oxygen concentration varied from 9.3 to 6.4 ppm. At 20 °C the solubility ratio is:

$$H = 4.0 \times 10^4 = \frac{P_0^2}{X_0^2}$$
(12)

The outlet water concentration was 6.4 ppm or 2×10^{-4} mole/2. Thus Xo₂ was:

-

$$xo_{2} = \frac{2 \times 10^{-4}}{2 \times 10^{-4} + 55.5} = 3.6 \times 10^{-6}$$
(13)

and

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$$Po_2 = (4.0 \times 10^4)(3.6 \times 10^{-6}) = 1.4 \times 10^{-7} \text{ atm.}$$
 (14)

Now if the loss of oxygen from the water is assumed to be in the gas,

or

$$\frac{(2.9 \times 10^{-3} \text{gm } 0_2)(13.1 \& \text{H}_2 0)}{\text{min}} = 3.8 \times 10^{-2} \frac{\text{gm } 0_2}{\text{min}}$$
(15)

Assuming an ideal gas and I atmosphere total pressure,

$$3.8 \times 10^{-2} \frac{\text{gm 0}_2}{\text{min}} (\frac{1 \text{ mole}}{32 \text{ gm}}) (\frac{22.4 \text{ k}}{\text{mole}}) = 0.026 \text{ k/min}$$
(16)

Since the nitrogen flow rate was 5.6 *k*/min, the outlet oxygen mole fraction was,

$$Yo_2 = \frac{0.026}{0.026 + 5.6} = 4.62 \times 10^{-3}$$
 (17)

and

$$Po_2 = (4.62 \times 10^{-3})(1 \text{ atm}) = 4.62 \times 10^{-3} \text{ atm}$$
 (18)

Thus,

$$\frac{4.62 \times 10^{-3}}{1.44 \times 10^{-1}} = 3.2\% \text{ of the equilibrium value}$$
(19)

Application of mass transfer theory to the 15-ft tapered column must be prefaced with a consideration of which parameters vary through the column and how they vary. The tapering causes an increase in crosssectional area with height and a resulting decrease in superficial velocity. There is a significant pressure change with height and this pressure change affects the interfacial area per unit volume as well as the equilibrium oxygen concentrations in the liquid and gas.

In the absence of holdup data or tracer tests to estimate dispersion within the column, an order of magnitude calculation for KLa, the overall mass transfer coefficient, can be estimated by equating the dissolved oxygen concentration change between two column positions to the rate of transfer to the gas phase. That is,

$$U_{L}A[C_{L_{1}} - C_{L_{2}}] = K_{L}a V_{S}[C_{L_{2}} - C_{L}^{*}]$$
 (20)

where C_L^{\top} would be the liquid oxygen concentration in equilibrium with the gas phase concentration and V_S is assumed equal to the reactor volume between the two measuring points. The assumption has also been made that the average liquid concentration is equal to the outlet concentration, i.e., a CSTR. The section inlet and outlet concentrations (C_{L_1} and C_{L_2}) are known but the section volume and C_L^{+} must be calculated. For a tapered section of column,

$$V_{\rm S} = \frac{\pi h}{3} (r_1^2 + r_1 r_2 + r_2^2)$$
(21)

The equilibrium concentration, C_L^* , can be estimated with Henry's Law.

$$x_{02}^{*} = \frac{C_{L}^{*}}{C_{L}^{*} + C_{H_{2}0}} = \frac{P_{02}}{H} = \frac{Y_{02}P_{T}}{H}$$
 (22)

$$c_{L}^{*} = \frac{c_{H_{2}0}}{(\frac{h}{Y_{02}P_{T}} - 1)}$$
(23)

Now, both Y_{02} and P_T vary with column position,

$$\underline{P}_{T} = P_{A} + P_{S} + P_{H}$$
(24)

assuming the water level was at 400 cm,

$$P_{T} = P_{A} + \rho g[400 - H + \Delta H]$$
(25)

where ΔH is the manometer reading and H is the height from the bottom of the column. The mole fraction in the gas, Y0₂, is calculated from the decrease in the liquid oxygen concentration and the nitrogen flow rate as shown in Appendix 9.2 with the exception that the gas molar volume is corrected for the pressure variation. Once the variables are determined as at each column position, K_La can be calculated from

$$K_{La} = \frac{U_{L}A(C_{L_{1}} - C_{L_{2}})}{V_{S}(C_{L_{2}} - C_{L^{*}})}$$

Data for one experiment and the calculated values of Kia are listed in Table 2.

H(cm)	P _T (atm)	V _G (1/mole)	C _L (ppm)	4C _L (ppm)	C _L *(ppm)	$(c_{L_1} - c_{L_2})/(c_{L_2} - c_{L^*})$	V _S (2)	K _L a(min ⁻¹)
20	1,42	17.2	9.3	0.7	0,05	8.15 X 10 ⁻²	15.6	6.84 X 10 ⁻²
61	1:37	17.8	8.6					
101	1.32	18.4	-					1 00 V 10 ⁻¹
155	1.26	19.3	8.2	1,4	0,11	2.08 X 10 ⁻¹	19.7	1,38 X 10 1
182	1.23	19,8	6,8	0,3	0.02	4.63 X 10^{-2}	29.9	2.03 X 10 2
223	1.19	20.5	6,5					

.

TABLE 2: DATA FOR CALCULATING KLa

M_S = 36 Kg L = 13.1 t/min

= 5.6 t/min G

29

9:4 Tabulation of Data and Operating Conditions

COAL LOADING = 35.0 KG Static Height = 184., CM VALUE OF ZERD MEANS NO MEASUREMENT L # 13.6 L/MIN ., G = 0.8 L/HIN BED HEIGHT = 148.7 CT CONDITION . HINIMUM FLUIDIZATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 20CH 61CH 141CH 155CH 182CH 223CH 264CH 384CH 345CH 46,6 21.8 28.6 21.8 21.8 22.1 22.8 21.5 8.8 DENSITY, G/CH3 22CM 61CH 171CH 155CH 182CH 223CH 264CH 384CH 345CH .227 .837 .848 .888 .888 .888 .888 .888 .888 HETGHT 22CM VOLUME FRACTIONS BETYEEN ES EL EG S.R CH ... 2 .998 .RAZ CITA S. N PPH OXYGEN HETGHT 20CH 61CH 1-1CH 155CH 182CH 223CH 264CH 384CH 345CH 3.7 8.9 3.3 8.3 8.8 8.8 8.8 0.0 8.0 L = 52.3 L/HIN G = 0.0 L/MIN BED HEIGHT = 270.3 C⁴ Condition = Internediate fluidization PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 2004 6104 1410H 1550H 1820H 2230H 2640H 3040H 3450H 49,2 44.4 38.8 33,3 31.2 28.9 26.7 26.7 8.8 DENSITY, G/CH3 26C1 61C4 191C4 155CM 182CH 223CH 264CH 304CH 345C4 .527 .888 .832 .888 .000 .800 .800 .800 .800 .800 HEIGHT VOLUME FRACTIONS JET JEEN **F.S** EL EG 2.0 CH .472 .998 .000 B.B AND PPH DXYGEN 200% 610H 1-104 155CH 182CH 223CH 264CH 384CH 345CH HEIGHT 6,7 2,7 .0 0.0 0.0 0.0 0.0 0.0 **0.0** L = 58.4 L/MIN G . B.B L/MIN BED HEIGHT - 363.0 CM CONDITION = MAXIMUN FLUIDIZATION PRESSURE ABGVE STATIC HEADS CM H20 HEIGHT 2204 5101 1+10M 1550H 1820H 2230H 2640M 3840H 3450H 54.2 51.4 48.6 45.3 43.7 41.6 39.8 37.4 35.2 DENSITY, G/CH3 HELGHT VOLUME FRACTIONS BET LEEN ES EL EG 2.2 CH .- 12 .998 .203 6.2 AND PPH OXYGEN HEIGHT 200H 610H 111CH 155CH 182CH 223CH 264CH 304CH 345CH 3.9 3.8 3.8 3.8 8.8 8.8 8.8 8.8 8.8 L # -62.1 L/MIN G # 8.2 L/HIN BED HEIGHT # 288.0 CH CONDITION . INTERMEDIATE FLUIDIZATION PRESSURE ABOVE STATIC HEADI CH H20 HEIGHT 20CH 61CH 141CH 195CH 182CH 223CM 264CH 384CH 345CH 52.3 48.6 44.7 39.7 37.7 36.1 (35.5 35.4 35.3 DENSITY, G/CH3

28CH 61CH 171CH 155CH 182CH 223CH 264CH 384CH 345CH .888 .888 .888 .881 ,880 ,388 .888 .888 .994 HETGHT VOLUME FRACTIONS BETWEEN ES EL EG 3.8 CM S.S AND PPH OXÝGEN 9.3 9.2 0.6 8.9 0.8 8.2 8.8 8.8 8.8 HEIGHT L # 13.1 L/HIN G . 5.6 L/HIN BED HEIGHT . 101.8 CH CONDITION . HINIMUM FLUIDIEATION PRESSURE ABOVE STATIC HEAD, CH H20 28CH 61CH 181CH 155CH 182CH 223CH 264CH 334CH 345CH 51.3 42.5 33.6 22.3 21.4 21.5 21.7 21.7 81.3 HEIGHT DENSITY, G/CH3 23C" 61CM 141CH 155CH 182CT 223CH 264CH 384CN 345C4 .88C .683 .886 .849 .880 .383 .888 .898 .893 HETGHT VOLUME FRACTIONS BETYEEN ES EL EG 2.0 CH 9.0 AND PPH DXYGEN 23CH 61CH 141CH 155CH 182CH 223CH 254CH 384CH 345CH 9.3 8.6 18 8.2 6.8 6.5 6.4 6.4 8.8 HEIGHT = 48.3 L/HIN G . 5.6 L/HIN BED HEIGHT = 217,8 GH CONDITION = INTERMEDIATE FLUIDIZATION PRESSURE ABOVE STATIC HEAD, CH H20 2004 6104 14104 15504 18204 22308 26408 38408 34508 48,2 43,7 38,8 31,6 28,9 26,1 26,1 26,1 25,5 HEIGHT DENSITY, GZCH3 HETGHT ,888 1,279 ,888 ,888 ,883 VOLUME FRACTIONS
 BET HEEN
 ES
 EL
 EG

 28.0
 AND
 61.0
 CH
 .109
 .717
 .094

 41.0
 AND
 155.0
 CH
 .174
 .741
 .885

 195.0
 AND
 223.0
 CH
 .144
 .783
 .973
 PPH DXYGEN 2004 610H 1910H 1550H 1620H 2230H 2040K 3040H 3450H 8.6 8.4 3.2 8.2 9.8 7.5 2.8 7.5 8.8 NETGHT L = 60.7 L/MIN G & B.S L/MIN DED HEIGHT = 338 B C4 CONDITION = INTERMEDIATE FLUIDIZATION PRESSURE ABOVE STATIC MEAD, CH H20 HEIGHT 28CH 61CH 121CH 155CH 182CH 223CH 264CH 394CH 345CH 53.5 49.2 45.8 48.8 37.8 34.8 32.8 29.7 29.8 DENSITY, G/CH3 28CH 61CH 171CH 155CH 182CH 223CH 264CH 384CH 345CH 1.897 1.118 .888 1.871 1.883 1.878 1.869 .888 .889 HEIGHT VOLUME FRACTIONS BETHEEN ES ËL £G BETWEEN ES EL EG 28.8 AND 61.8 CH .184 .816 .808 61.8 AND 155.8 CH .154 .646 .838 158.8 AND 182.8 CH .137 .863 .888 168.8 AND 223.8 CH .144 .856 .888 283.8 AND 263.5 CH .131 .869 .888 PPH DXŸGEN HEIGHT 28CH 61CH 101CH 155CH 182CH 223CH 264CH 384CH 3450H 9.8 9.3 0.8 9.3 8.8 9.8 8.8 9.3 8.8

COAL LOADING B 58,8 KG STATIC HEIGHT # 132,0 CH VALUE OF ZERO MEANS NO MEASUREMENT L = 13,8 L/MIN B.B LIMIN 6 . BED HEIGHT # 161.8 CH CONDITION = HINIMUH FLUIDIEATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 20CH 61CH 141CH 155CH 182CH 223CH 264CH 384CH 345CH 53,2 44,3 35,4 25.0 21,9 22,1 22,3 21,8 21,8 DENSITY, G/CH3 23CH 61CH 141CH 155CH 182CH 223CH 264CH 384CH 345CH HEIGHT 1.222 1.212 1,216 1.138 1,080 ,200 , 699 ,966 ,968 VOLUME FRACTIONS BETHEEN ËS EL EG 28.0 AND 61.0 CH .387 .613 .000 61.0 AND 101.0 CH .302 .618 .000 SUI.0 AND 155.0 CH .301 .699 .000 PPH CXYGEN HEIGHT 2004 61CH 121CH 155CH 182CH 223CH 264CH 384CH 345CH 2.2 0.2 7,2 0,3 0,0 8.8 0,0.0.0 9.0 = 62.5 L/MIN 6 = 8.0 L/HIN BED HEIGHT = 339.8 CH Condition = Maximum FLUIDIEATION
 PRESSURE
 ABOVE
 STATIC
 HEAD,
 CH
 H20

 HEIGHT
 2001
 610H
 1210H
 1550H
 1820H
 2230H
 2640H
 3840H
 3430H

 56.1
 52.4
 46.5
 41.2
 39.1
 35.8
 32.9
 29.6
 27.3
 QENSITY, G/CH3 2004 6104 19108 15508 18208 22308 26408 38408 34508 1.116 1.128 1.138 1.674 1.875 1.878 1.879 1.858 .888 HEIGHT VOLUHE FRACTIONS BETHEEN ES EL EG 61.2 AND 61.8 CH .211 .789 61.2 AND 101.0 CH .223 .777 101.9 AND 155.8 CH .168 .832 155.2 AND 182.8 CH . 999 , 230 . 868 .000 155.2 AND 182.8 CH .133 .867 82.0 AND 223.8 CH .137 .863 .800 23.2 AND 263.5 CH .149 .860 .870 63.5 AND 384.8 CH .119 .881 .870 PPH OXÝGEN HEIGHT 28CH 61CH 121CH 155CH 182CH 223CH 264CH 384CH 345CH 8.6 3.8 1.0 9.9 0.0 0,0 8.6 1.6 8.8 = 54.5 L/HIN L = 54.5 L/MIN BED HEIGHT = 289.2 Ch CONDITION . INTERMEDIATE FLUIDIEATION PRESSURE ABOVE STATIC HEAD, CH H20 Height 28CH 61CH 141CH 155CH 182CH 223CH 264CH 384CH 345CH 52.3 49.2 43.6 38.3 35.7 32.2 29.2 27.4 26.7 DENSITY, G/CM3 20CH 61CH 131CH 155CH 182CH 223CH 264CH 384CH 345CH HETGHT 1.114 1.123 1.123 1.883 1,888 1,877 1.866 1.885 ,860 VOLUME FRACTIONS BETHEEN ES EL EG 28.0 AND 61.0 CH .198 .711 .899 61.0 AND 101.8 CH .198 .784 .897 101.8 AND 155.8 CH .163 .756 .881 155.0 AND 182.8 CH .136 .798 .866

182.8 AND 223.8 CH ,131 ,854 ,865 123.8 AND 263,5 CH ,119 ,821 ,859 TPN OXYGEN HETCHT 28CH 41CH 141CH 155CH 182CH 223CH 264CH 384CH 343CH 7.8 2.8 8.8 ... 8,8 8,8 L = 28.6 L/MIN 8 = 28.2 L/MIN BED HEIGHT = 231.8 CM CONDITION . INTERMEDIATE FLUIDIBATION PRESSURE ABOVE STATIC MEAD, CH M20 HEIGHT 28CH 61CH 181CH 195CH 182CH 223CH 264CH 384CH 345CH 45,8 39,6 33,8 25,6 23,6 23,2 23,2 22,9 22,3 DENSITY, G/CH3 20CH 61CH 101CH 155CH 182CH 223CH 264CH 384CH 345CH NEIGHT 1.176 1.172 1.162 1.118 1.924 .788 .888 .888 .880 .800 VOLUME FRACTIONS BETHEEN ES EL EG 20.0 AND 61.0 CH .269 .596 .135 61.0 AND 101.0 CH .269 .612 .129 101.0 AND 155.0 CH .210 .687 .103 155.0 AND 182.0 CH .268 .899 .033 PPH DXÝGEN HETGHT 28CH 61CH 101CH 155CH 182CH 223CH 264CH 384CH 345CH 2.7 8.0 3.2 8.3 8.9 8.8 8.8 8.8 8.8 COAL LOADING # 75,8 KG STATIC HEIGHT & 182 ... CH VALUE OF ZERO MEANS NO MEASUREMENT L # 42.2 L/MIN G # B.C L/MIN BED HEIGHT # 335,8 C4 CONDITION = MAXIMUM FLUIDIZATION PRESSURE ABOVE STATIC HEAD, CM H20 HEIGHT 20CM 61CH 131CH 155CH 182CH 223CH 264CH 384CH 349CH 65.2 58.5 51.8 41.2 34.3 36.7 32.2 27.6 24.6 DENSITY, G/CH3 20CH 61CH 101CH 155CH 182CH 223CH 264CH 304CH 345CH 1.135 1.131 1.145 1.108 1.099 1.188 1.185 1.097 .088 HEI GHY VOLUME FRACTIONS BETHEEN ES EL EG 28.0 AND 61.8 CH .237 .763 .000 61.0 AND 101.8 CH .246 .754 .000 191.0 AND 155.0 CH .221 .779 .000 155.8 AND 182.0 CH .184 .816 .830 182.8 AND 223.8 CH ,178 ,822 ,808 223.8 AND 263.5 CH ,183 ,817 ,828 263.5 AND 364,8 CH ,188 ,028 ,888 PPH OXYGEN HEIGHT 20 20CH 61CH 1115H 155CH 182CH 223CH 264CH 384CH 345CH 8,4 8,8 3,8 2,8 2,8 8,8 3,8 8,8 8,8 = 28.7 L/MIN G # B.B L/MIN BED HEIGHT = 291.8 CM CONDITION & INTERMEDIATE FLUIDIBATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 28CH 61CH 101CH 155CH 182CH 223CH 264CH 384CH 348CH 65,8 62,6 59,5 48,5 36,9 31.6 26,7 23,5 23,4 DENSITY, G/CH3 28CH 61CH 191CH 155CH 182CH 223CH 26GCH 384CH 345CH 1.159 1.165 1.166 1.137 1.132 1.134 1.125 .888 .888 HEIGHT VOLUME FRACTIONS

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BETHEEN EL ES EG 28.8 AND 61.8 CH .289 .711 .808 61.8 AND 181.8 CH .296 .784 .836 181.8 AND 155.8 CH .268 .732 .886 155.0 AND 182.0 CH .242 .760 .000 142.0 AND 223.0 CH .239 .761 .900 223.0 AND 263.5 CH .233 .767 .200 PPH OXYGEN 20CH 61CH 1K1CH 155CH 182CH 223CH 264CH 304CH 349CH HEIGHT 2.0 4.8 1.0 3.6 0.3 0.9 0.0 8.8 8.8 L = 13.7 L/HIN G = B.E L/MIN BED HEIGHT # 213.0 CH CONDITION = MINIMUM FLUIDIZATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 20CH 61Ch 191CH 155CH 182CH 223CH 264CH 304CH 345CH 64,4 55,2 46,1 34,6 37,1 23,7 22,5 22,3 21,8 DENSITY, G/CI'3 200" 61CH 1 11CH 155CH 182CH 223CH 264CH 384CH 345CH HETGHT 1.222 1.227 1.222 1.185 1.156 1.122 .880 .883 .893 VOLUME FRACTIONS BETHEEN ES EG ΞL 28.8 AND 61.3 CH ,4 11 ,599 ,080 61.6 AND 181.8 CM .471 .599 .664 181.2 AND 155.2 CH .363 .640 .200 155.0 AND 162.0 CH .3 12 .698 .644 182.2 AND 223.2 CH .245 .755 .249 PPH CXYGEN HETGHT 20CM 61CH 1-1CH 155CH 182CH 223CH 264CH 384CH 345CH 0.7 6.7 ·• ø 3.0 2,0 8.0 8.8 2.0 0.8 L = 12.3 L/MIN 3.2 L/MIN 8 BED HEIGHT = 234.2 C" CONDITION # MINIMUM FLUIDIZATION PRESSURE ABOVE STATIC HEAU, CH H20 270" 61CH 1.1CH 155CH 182CH 223CH 264CH 384CH 345CH 66.4 57.2 48.8 37.2 31.8 23.7 22.4 22.2 21.7 HEIGHT DENSITY, G/CH3 2800 6104 1710M 15508 1820M 2230H 2640H 3840H 34504 1.223 1.226 1.214 1.284 1.288 1.182 .888 .888 .888 HETGHT VOLUME FRACTIONS BETWEEN ES EL EG 28.0 AND 61.0 CM .322 .520 .158 61.8 AND 181.9 CM .317 .529 .154 181.0 AND 155.0 CH .315 .529 .156 155.8 AND 182.8 CH .376 .543 .151 188.8 AND 223.8 CH .292 .565 .143 PPM CXYGEN HETCHT 22CM 61CH 191CH 155CH 182CH 223CH 264CH 304CH 345CH 2.2 9.0 1.3 9.9 3.9 8,8 6.0 8.0 9.9 L = 31.9 L/HIN 5 # 3.2 L/HIN BED HEIGHT = 319.9 CM CONDITION . INTERHEDIATE FLUIDIEATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 28CH 61CH 101CH 155CH 182CH 223CH 264CH 384CH 345CH 65,4 58,1 50,8 42,9 39,4 34,1 29,3 24,2 23,8 DENSITY, G/CH3 20CH 61CH 191CH 155CH 182CH 223CH 264CH 304CH 343CH 1.100 1.176 1.171 1.130 1.130 1.130 1.126 1.012 .080 HEIGHT VOLUME FRACTIONS

BETWEEN ES EL 20. P AND 61.8 CH .275 .589 61.8 AND 121.8 CH .269 .599 .136 ,132 191.8 ANS 155.8 CH ,233 .651 155.8 AND 182.8 CH .238 .688 .116 .134 182.6 AND 223.8 CH .236 .688 .184 223.6 AND 263.5 CM .2 15 .692 .183 263.5 AND 324.8 CH .: 39 .952 .889 PPH OXYGEN HEIGHT 28CH 61CH 1-1CH 155CH 182CH 223CH 264CH 384CH 343CH P.Ø 3,9 9,8 3.3 8.9 7.8 5.8 8.8 8.0 L = 13.4 L/HIN 6 = 1.4 L/HIN BED HEIGHT = 238.3 C1 CONDITION = HINIMUM FLUIDIZATION PRESSUPE ANCVE STATIC HEAD; CH H20 HEIGHT 20C° 61CH 1'1CH 155CH 182CH 223CH 264CH 304CH 345CH 65.1 56.3 46.8 35.9 30.4 22.6 20.3 21.7 21.5 DENSITY, G/CH3 2004 6104 1310M 1550M 1820H 2830M 2640M 3840M 3450M HETGHT 1,222 1,213 1,218 1,203 1,192 1,310 ,800 ,800 ,003 VOLUME FRACTIONS BETHEEN ES ٤L EG 28.0 AND 61.3 CH .326 .514 .160 61.2 AND 101.8 CH .324 .518 .158 181.0 AND 155.8 CH .316 .527 .157 155.8 AND 182.6 CH .378 .552 .148 188.8 AND 223.8 CH .= 34 .966 PPH OXÝGEN 28CH 61CH 141CH 155CH 182CH 223CH 264CH 384CH 345CH HEIGHT e.: e.g ',e e.e e.e e.e 8.0 8,8 8,8 COAL LOADING = 118,7 KG STATIC HEIGHT # 242,2 CH VALUE OF ZERO MEANS DO HEASUREMENT L # 18.1 L/MIN C . C.C.L/HIN BED HEIGHT = 321.0 C1 CONDITION & MAXIMUM FLUIDIBATION PRESSURE ABUVE STATIC HEAD, CH H20 Height 20CH 61CH 141CH 155CH 182CH 223CH 264CH 384CH 345CH 82.1 71.3 61.1 49.4 44.1 37.8 31.8 24.3 22.3 DENSITY, G/CH3 28CH 61CH 191CH 155CH 182CH 223CH 264CH 384CH 345CH HEIGHT 1.226 1.219 1.228 1.174 1.151 1.152 1.173 1.132 ,880 VOLUME FRACTIONS BETHEEN ES EL EG 20.0 AND 61.5 CM .347 .603 .908 41.8 AND 121.8 CM .399 .681 .808 151.3 AND 155.8 CM .348 .652 .808 155.0 AND 182.0 CH .286 .714 , 200 182.0 AND 223.0 CH ,271 ,729 ,800 223.0 AND 263.5 CH ,289 ,711 ,800 263.5 AND 304.8 CH ,267 ,733 ,800 PPH OXYGEN HEIGHT 28CH 61CH 141CH 155CH 182CH 223CH 264CH 384CH 345CH 0.0 0.0 ···· 0.0 0,0 8,8 6.8 9.9 8.9

EG

L = 13.9 L/MIN G = 0.0 L/HIN 8ED HEIGHT = 305.0 C1 CONDITION . MINIMUM FLUIDIZATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 22CH 61CH 1-1CH 155CH 182CH 223CH 264CH 384CH 345CH 82.1 71.1 67.7 48.3 42.6 35.1 28.1 22.2 21.7 DENSITY, G/Cº3 2004 61CH 171CH 155CH 182CH 223CH 284CH 384CH 345CH 1.264 1.232 1.213 1.179 1.144 1.146 1.145 .888 .888 HEIGHT MOLUNE FRACTIONS ES EL EG BETHEEN 20,0 AND 61.0 CH .441 .559 .700 61.0 AND 121.F CH .397 .673 .732 101.0 AND 155.0 CH .347 .653 .730 155.2 AND 182.9 CH ,285 ,715 .200 182.2 AND 223.9 CM .259 .741 .800 223.6 AND 263.5 CH .260 .740 .800 PPH DXYGEN 220" 6104 16108 15508 18208 22308 26408 38408 34508 5.7 9.0 7.0 9.8 6.8 9.8 8.8 8.8 8.8 HEIGHT L = 13.8 L/HIN G # 6.7 L/MIN BED WEIGHT # 292.7 C" CONDITION # "INIMUM FLUIDIZATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT 22CH 61CH 141CH 155CH 182CH 225CH 264CH 304CH 345CH 78.5 69.4 51.9 48.4 43.2 35.2 27.4 22.4 22.0 DENSITY, G/CH3 HEIGHT 28CH 61CH 191CH 155CH 182CH 283CH 264CH 384CH 345CH 1.213 1.239 1.210 1.210 1.201 1.205 1.203 1.008 ,900 VOLUME FRACTIONS BETHEEN ES. EL FG 28.2 AND 61.2 CM .321 .521 .158 61.0 AND 121.8 CH .316 .529 .156 191.3 AND 155.8 CH .317 .529 .154 155.0 ALD 162.0 CH .310 .535 .154 182.0 ALD 223.0 CH .377 .540 .152 223.0 AND 263.5 CH .378 .538 .153 263.5 ALD 324.0 CH .27 .970 .003 PPH SXYGEN 2004 6104 14104 15508 18208 22308 26408 38408 34508 2.2 8.8 7.8 8.8 8.8 8.9 8.8 8.8 8.8 8.8 HEIGHT L = 14.6 L/hIN # 11.0 L/HI Ĝ BED HEIGHT = 289.3 C% CONDITION = INTERNEULATE FLUIDIEATION PRESSURE ABOVE STATIC HEAD, CH H20 HEIGHT _ 20CH 61C1 121CH 155CH 182CH 223CH 264CH 384CH 345CH 75.7 67.6 63.6 49.4 33.9 35.7 28.2 22.7 21.9 DENSITY. G/C"3 230" 610H 1310H 1550H 1820H 2230H 2640H 3840H 3450H HEIGHT 1.214 1.135 1,206 1.208 1,202 1,197 1,196 1,037 1,806 VOLUME FRACTIONS BETHEEN ES EL EG 28.6 AND 61.6 CH ,224 .668 .100 61.0 AND 101.0 CH .222 .671 .107 101.0 AND 155.0 CH .222 .671 .107 135.0 AND 155.0 CH .313 .533 .154 155.0 AND 182.8 CH .321 .556 .122 137.0 AND 223.8 CH .290 .534 .170 223.8 AND 263.5 CH .299 .553 .149 263.5 AND 384.8 CH .176 .848 .846 384.8 AND 344,5 CM .018 .973 .808

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PPH OXÝGEN 28Ch 61Ch 141Ch 155Ch 182Ch 223Ch 264Ch 384Ch 345Ch HEIGHT 8,6 B.C 7,8 9,8 9,8 3,8 8,5 8,8 8,8 L = 18.8 L/HIN 6 # 11.1 L/HIN G B 11.1 LFT.11 DED HEIGHT = 279.0 CH CONDITION = MINIMUM FLUIDIEATION PRESSURE ABOVE STATIC HEAD, CH H2O 28CH 61CH 121CH 155CH 182CH 223CH 264CH 384CH 345CH 78.2 68.3 59.8 47.6 41.8 32.9 24.8 22.2 21.5 HEIGHT DENSITY, G/CH3 HEIGHT 28CH 61CH 131CH 155CH 182CH 223CH 264CH 304CH 343CH 1.219 1.217 1.226 1.222 1.218 1.210 1.185 1.811 .808 VOLUME FRACTIONS BETHEEN F.S EL EG 28.3 AND 61.8 CH .327 .514 .159 61.2 AND 181.9 CH .331 .586 .164 181.8 AND 155.8 CH .334 .581 .164 155.0 AND 182.5 CH .329 .568 .163 182.8 AND 223.8 CH .325 .514 .168 223.8 AND 263.5 CH .303 .548 .149 263.5 AND 384.8 CH .837 .949 .814 PPH DXVGEN HETGHT 200H 610H 1910H 1550H 1820H 2230H 2640H 3840H 3450H 8,8 8,9 7,8 8,8 8,8 8.8 8.8 8.8 8.8 L = 14.7 L/HIN G = 0.2 L/HIN BED HEIGHT = 389.0 C1 CONDITION W INTERNEDIATE FLUIDIZATION PRESSURE ABOVE STATIC MEAD, CH H20 HEIGHT 2004 610H 1710H 1550H 1820H 2230H 2640H 3840H 3450H 81.5 78.2 59.3 46.8 41.9 34.8 28.2 22.1 21.8 DENSITY, G/C+3 HEIGHT 20CH 61CH 141CH 159CH 182CH 223CH 264CH 304CH 349CH 1.243 1.222 1.175 1.148 1.149 1.158 1.162 ,860 .009 VOLUME FRACTIONS BETHEEN ES EL EG BETALEN 20.0 AND 61.0 CH 414 586 600 61.0 AND 181.0 CH 349 651 800 191.0 AND 155.0 CH 284 714 800 155.0 AND 182.0 CM 265 735 800 182.0 AND 223.0 CH 267 733 800 223.0 AND 263.5 CH 278 722 800 PPH DXYGEN HEIGHT _ 28CH 61CH 191CH 155CH 182CH 223CH 264CH 384CH 345CH 8.0 8.8 9.8 9.8 9.8 8.8 0.0 0.0 8.9

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9.5 Nomenclature

A	column cross sectional area, cm ²
с _{н2} 0	liquid phase water concentration, ppm
с _L	liquid phase oxygen concentration, ppm
C _{L1}	C _L entering column section, ppm
CL2	C _L exiting column section, ppm
c _L *	C_L in equilibrium with gas phase oxygen concentration, ppm
g	gravitational acceleration, 980 cm/sec ²
∆h	length between two column ports, cm
H	distance from bottom of column, cm
ΔH	manometer reading, cm H ₂ O
нь	height of fluidized bed, cm
Hbs	height of static coal bed, cm
К	liquid to solid holdup ratio
К _L а	overall mass transfer coefficient, min ⁻¹
MS	mass of coal in bed, kg
M _T	total mass in sample, gm
PA	atmospheric pressure, atm
P _H	pressure during operation at height H up column, cm H_2O , atm
P _S	pressure of static bed/water at height H, cm H_2O , atm
PT	total pressure, atm
P _T	average column section pressure, atm
ΔP	pressure difference, cm H ₂ O
P ₀₂	partial pressure oxygen in gas, atm
r ₁ ,r ₂	column radius at inlet and outlet to section, cm
υ _L	liquid superficial velocity, cm/sec
-	

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Υ _G	gas molar volume, 1/mole
₽G	average molar volume for column section, 2/mole
×02	mole fraction in liquid
У ₀₂	mole fraction in gas
۷ _S	volume of column section, cm ³
^ε L, ^ε L'	liquid volume fractions in column and port sample
ε ε S, S'	same for solid volume fraction
ρ _G	gas density, gm/cm ³
۴L	liquid density, gm/cm ³
٩S	solids density, gm/cm ³
- Т ⁰	total density of port sample, gm/cm ³
θ	column angle from vertical, degree

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