

**APPENDIX D**

**DESCRIPTION OF CSTR REACTOR SYSTEM  
AND EXPERIMENTAL PROCEDURE**

## DESCRIPTION OF CSTR REACTOR SYSTEM AND EXPERIMENTAL PROCEDURE

### 1. Description of CSTR Reactor System

A schematic flow diagram of the CSTR reactor system that was operated at the University of Kentucky Center for Applied Energy Research located in Lexington, Kentucky is shown in Figure E-1.

The system consists of 3 mass flow meters, one each for H<sub>2</sub>, CO, and Argon. There are two CO purification traps for removing iron carbonyls from the CO feed stream. The reactor is a one liter stirred autoclave with two internal filters. The autoclave temperature is controlled by an external electrical heater with a programmable controller. The reactor pressure is controlled by a manually operated Tesome back-pressure regulator. The off gas and vapor products pass through a hot trap (200°C), a lower temperature hot trap (100°C), and a cold trap (0°C) that is cooled by an external refrigeration unit. The cold trap off gas is vented to a manifold that can direct this gas to the blowdown or to a gas measuring device connected to a gas chromatograph. The gas measuring device was a 500 ml volume bubble column. The gas volume was calculated from the measured rates corrected to standard temperature and pressure.

### 2. Reactor Charge Procedure

Approximately 16 gm of catalyst was hydrogen reduced in the fluidized column at EI's research lab in Pittsburgh, Pennsylvania. The catalyst was slurried in approximately 150 ml of synfluid and put in a plastic bottle protected under nitrogen. Four bottles of catalyst slurry were transported to Lexington, Kentucky by William Gall. Before charging the reactor, the bottles were shaken by hand to reslurry the catalyst. The autoclave vessel bottom was lowered and pivoted to one side. An argon purge line was placed in the autoclave and the vessel was purged for several minutes. A hand held purge line was placed into the top of the catalyst bottle and held there while the catalyst slurry was poured into the autoclave. The bottle was rinsed with about 100 ml synfluid under N<sub>2</sub> purge. The vessel bottom was pivoted back under the head and bolted shut. The reactor was charged and ready for startup.

### 3. Reactor Run Procedure

The reactor was brought up to the initial run conditions and held there for several hours. At the end of the initial startup conditions, an off gas rate was taken, an off gas G.C. analysis was done, and all the product collection vessels were drained. The first catalyst tested was run at six sets of reaction conditions, while the three additional catalysts were run at seven sets of reaction conditions.

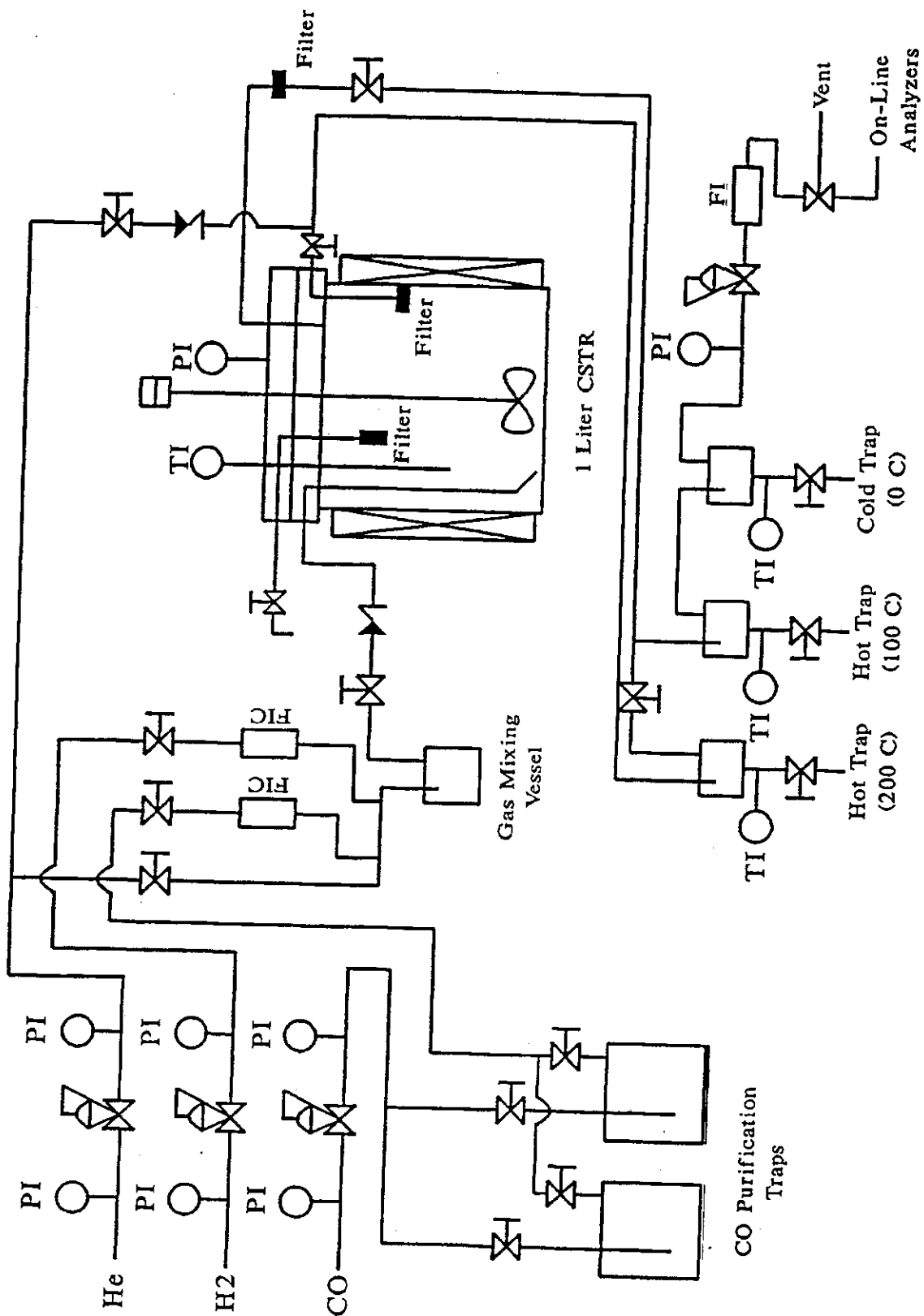
Liquid samples were removed only after the first, fourth, and last run periods. Each run lasted approximately 3 days. The following parameters were studied: reaction temperature, total gas feed rate and inert gas concentration. The reaction pressure was held at 450 psi and the H<sub>2</sub>/CO ratio at 2.0.

4. Data Calculations

Mass balance data was obtained from the mass flow meter rates, off-gas rate, and off-gas G.C. analysis. These data were entered into a PC computer, and the gas conversions, reaction rates, and product selectivities were calculated. Some of the liquid products were analyzed by a liquid G.C. to determine the product distribution (alpha) value.

Figure E-1

# FTS CSTR Reactor System





University of Kentucky  
**CENTER FOR  
APPLIED ENERGY  
RESEARCH**

572 Iron Works Pike  
Lexington, KY 40511-8433  
FAX, 606-257-0302

August 24, 1995

William Gall  
Senior Project Engineer  
Energy International  
135 William Pitt Way  
Pittsburgh, PA 15238

Bill,

Here are the selectivity and rate data for the four runs we conducted. After looking at the data closely, it is clear that there is an analytical problem which makes calculating the rates and selectivities difficult. GC's taken at the conditions with high flow rates and high argon composition add up to greater than 100% with some as high as 120%. We do not encounter this problem with our runs which do not use argon and are conducted at much lower flow rates; in fact, the problem seems to disappear with the low flow rates used at the end of your runs. Possible causes for this problem that we can think of include:

1. Our standard gas used for calibrations contains only 2% argon whereas the argon composition of the exit stream is >60%.
2. Flow rates about 5X higher than we normally use might have entrained oil from the traps which would affect the GC detector.

We are confident that our mass flow controllers performed correctly because we calibrated them before and after each run. This makes it possible to calculate what the argon composition in the exit stream should be.

$$Ar_{exit} \text{ (mol\%)} = Ar_{in} / (\text{Total flow out})$$

To do the selectivity and rate calculations I assumed the remaining gases comprised  $100 - Ar_{exit}$  mol%. Each component was then corrected in proportion to the measured value from the GC. This does assume that each component was affected in the same ratio. If you have any questions please contact me at (606) 257-0324.

Sincerely,

Bob O'Brien

**UK**

Equal Opportunity/  
Affirmative Action University

One Liter CSTR Studies at the University of Kentucky CAER															
Run No. LGX-185 Catalyst No.- CAL-13 Charge- 16 g															
T (°C)	Flow Rates SLPH					CO	H <sub>2</sub> CO Usage	Reaction Rates (g/kg/hr)					H <sub>2</sub> O		
	P (psig)	Ar	H <sub>2</sub>	CO	CO Conv			H <sub>2</sub> Conv	CH <sub>4</sub>	C <sub>2</sub>	C <sub>2+</sub>	THC		CO <sub>2</sub>	
235	440		120	120	60	60.42	61.71	2.04	2831.40	73.36	8.94	1320.04	1402.83	48.92	1761.01
240	450		120	120	60	73.32	73.41	2.00	3436.13	80.06	10.58	1800.83	1691.28	82.51	2134.25
220	450		120	120	60	31.70	31.17	1.97	1485.55	38.56	3.64	698.89	741.09	8.96	946.12
240	450		120	120	60	68.83	68.30	2.04	3131.72	91.55	11.98	1438.05	1541.58	83.89	1945.53
240	450		60	60	30	90.12	85.38	1.89	2111.73	100.99	14.98	861.26	877.23	251.93	1151.94
240	450		0	60	30	99.00	87.28	1.78	2319.67	508.83	49.24	344.02	900.09	820.69	620.05
Temp	Flow Rates SLPH					CO <sub>2</sub>	Selectivity (mol%, based on C converted)					C <sub>2+</sub>			
	Press	Ar	H <sub>2</sub>	CO	CO Conv		CH <sub>4</sub>	C <sub>2</sub>	C <sub>2+</sub>	C <sub>3</sub>	C <sub>4</sub>		C <sub>5</sub>	C <sub>2-C<sub>6</sub></sub>	
235	440		120	120	60	1.10	4.52	0.59	1.84	2.27	2.31	7.01	94.89	88.47	
240	450		120	120	60	1.71	4.07	0.57	1.60	1.96	0.00	4.14	95.36	91.79	
220	450		120	120	60	0.38	4.53	0.48	1.74	2.28	0.00	4.48	95.01	90.98	
240	450		120	120	60	1.70	5.10	0.71	1.86	2.41	2.73	7.61	94.18	87.09	
240	450		60	60	30	7.59	8.36	1.32	2.77	2.97	3.15	10.22	90.33	81.43	
240	450		0	60	30	22.52	38.15	3.95	5.33	3.91	2.61	15.80	67.90	46.05	

One Liter CSTR Studies at the University of Kentucky CAER  
 Run No. LGX-186 Catalyst No.- Co.005 Charge- 17 g

T (°C)	Flow Rates SLPH			CO Conv	H <sub>2</sub> Conv	H <sub>2</sub> CO Usage	Reaction Rates (g/kg/hr)							
	P (psig)	Ar	H <sub>2</sub>				CO	CO	CH <sub>4</sub>	C <sub>2</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>	C <sub>6</sub>
235	440	120	120	60	62.98	64.35	2.04	2777.72	62.66	11.69	1311.33	1385.68	16.81	1772.77
240	450	120	120	60	61.08	62.83	2.06	2693.83	102.66	17.54	1214.01	1334.11	46.80	1694.28
220	460	120	120	60	32.65	33.92	2.08	1440.01	39.97	8.37	669.16	717.51	11.38	816.84
240	430	120	120	60	67.27	69.21	2.06	2867.04	78.83	14.48	1384.74	1478.05	24.49	1898.24
240	450	60	60	30	68.14	69.58	2.04	1502.68	114.57	23.58	577.27	715.41	116.43	871.16
240	450	30	60	30	78.26	78.80	2.01	1725.96	81.74	13.34	739.75	834.83	92.58	1034.28
240	450	0	60	30	69.72	70.30	2.02	1537.59	97.48	16.40	616.61	730.49	123.99	887.42
Temp	Flow Rates SLPH			CO <sub>2</sub>	CH <sub>4</sub>	Selectivity (mol%, based on C converted)								
	Press	Ar	H <sub>2</sub>			CO	C <sub>2</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>	C <sub>6</sub>	C <sub>7</sub>	C <sub>8</sub>	
235	440	120	120	60	0.39	3.94	0.78	1.32	1.60	1.95	5.66	85.28	90.41	
240	450	120	120	60	1.11	6.65	1.21	2.36	2.78	3.28	9.65	92.14	83.70	
220	460	120	120	60	0.50	4.85	1.08	1.92	2.32	2.57	7.89	94.07	87.26	
240	430	120	120	60	0.53	4.64	0.91	1.65	1.83	2.10	6.49	94.45	88.87	
240	450	60	60	30	4.93	13.31	2.82	4.87	6.36	5.66	18.81	83.77	67.88	
240	450	30	60	30	3.41	8.27	1.44	3.15	3.29	3.18	11.06	80.29	60.67	
240	450	0	60	30	5.13	11.07	1.99	5.16	4.90	3.75	15.78	86.94	73.14	

One Liter CSTR Studies at the University of Kentucky CAER  
 Run No. LGX-187 Catalyst No. - Co.041 Charge - 17 g

T (°C)	Flow Rates SLPH			CO	H <sub>2</sub> CO			Reaction Rates (g/kg/hr)					H <sub>2</sub> O
	P (psig)	Ar	H <sub>2</sub>		CO Conv	H <sub>2</sub> Conv	Usege	CO	CH <sub>4</sub>	C <sub>2</sub>	C <sub>2</sub> +	THC	
235	450		120	60	71.14	72.44	2.04	3137.50	64.40	13.60	1487.67	1565.68	17.38
240	450		120	60	59.64	62.28	2.09	2630.36	128.68	25.03	1147.55	1301.28	50.16
220	450		120	60	20.69	22.86	2.21	912.36	54.76	12.83	387.40	454.98	6.01
240	440		120	60	58.00	60.08	2.07	2558.33	125.37	22.69	1120.45	1268.51	39.75
240	450		60	30	69.14	71.02	2.05	1524.76	141.12	25.56	553.48	720.15	136.26
240	450		30	30	33.35	37.18	2.23	735.43	182.60	37.48	96.65	316.72	161.80
240	450		0	30	23.09	27.41	2.37	509.25	88.62	12.36	148.18	249.16	18.42
Temp	Flow Rates SLPH			CO	Selectivity (mol%, based on C converted)			C <sub>2</sub> -C <sub>3</sub>					C <sub>3</sub> +
	Press	Ar	H <sub>2</sub>		CO <sub>2</sub>	CH <sub>4</sub>	C <sub>2</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>	C <sub>2</sub> -C <sub>3</sub>	C <sub>3</sub> +	
235	450		120	60	0.35	3.58	0.81	1.26	1.44	1.58	5.08	95.61	91.34
240	450		120	60	1.21	8.54	1.77	2.89	3.20	3.60	11.46	89.69	80.00
220	450		120	60	0.42	10.48	2.62	4.56	5.24	5.95	18.37	86.90	71.15
240	440		120	60	0.89	8.58	1.65	2.80	3.18	3.47	11.10	89.79	80.35
240	450		60	30	5.68	18.18	3.12	5.03	5.48	5.59	19.23	80.72	64.81
240	450		30	30	14.00	43.35	9.49	15.07	15.63	15.92	58.12	47.16	0.53
240	450		0	30	2.30	30.38	4.52	7.72	7.49	7.31	27.03	65.10	42.59



One Liter CSTR Studies at the University of Kentucky CAER														
Run No. LGX-188 Catalyst No. - Co.053 Charge- 17 g														
T (°C)	Flow Rates SLPH			CO Conv	H <sub>2</sub> Conv	H <sub>2</sub> CO Usage	Reaction Rates (g/kg/hr)						H <sub>2</sub> O	
	P (psig)	Ar	H <sub>2</sub>				CO	CH <sub>4</sub>	C <sub>2</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>		C <sub>6</sub>
235	450		120	64.19	66.31	2.07	2831.24	105.42	19.76	1276.11	1401.28	51.95	1778.42	
240	450		120	64.70	66.95	2.07	2853.65	161.15	21.17	1208.57	1390.89	119.77	1737.31	
220	450		120	25.55	27.47	2.15	1126.80	58.76	13.79	488.44	560.99	10.34	716.26	
240	450		120	69.61	71.19	2.05	3070.11	120.80	15.69	1380.07	1516.56	65.58	1920.89	
240	450		60	87.79	87.36	1.99	1935.92	82.03	13.00	838.77	933.80	111.95	1153.46	
240	450		30	87.62	86.14	1.97	1932.33	86.38	13.35	819.74	919.47	151.26	1118.97	
240	450		0	89.22	82.93	1.86	1987.51	135.39	17.98	724.10	877.49	338.27	988.50	
Temp	Flow Rates SLPH			Selectivity (mol%, based on C converted)										
	Press	Ar	H <sub>2</sub>	CO	CO <sub>2</sub>	CH <sub>4</sub>	C <sub>2</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>	C <sub>6</sub>	C <sub>7-C<sub>8</sub></sub>	C <sub>9</sub>	C <sub>10</sub>
235	450		120	120	1.17	6.50	1.30	2.25	2.69	3.00	9.24	92.20	84.28	
240	450		120	120	2.67	9.86	1.38	3.26	3.88	4.35	12.86	88.76	77.28	
220	450		120	120	0.58	9.10	2.28	3.79	4.77	5.85	16.69	88.62	74.21	
240	450		120	120	1.36	6.87	0.95	2.29	2.64	2.93	8.81	92.18	84.32	
240	450		60	60	3.68	7.40	1.25	2.24	2.49	2.66	8.54	91.35	84.08	
240	450		30	30	4.98	7.80	1.29	2.42	2.51	2.35	8.57	90.91	83.63	
240	450		0	60	10.94	12.01	1.70	3.31	2.85	1.91	9.77	86.28	78.21	