

Fig. 1 Details of Bertly Reactor

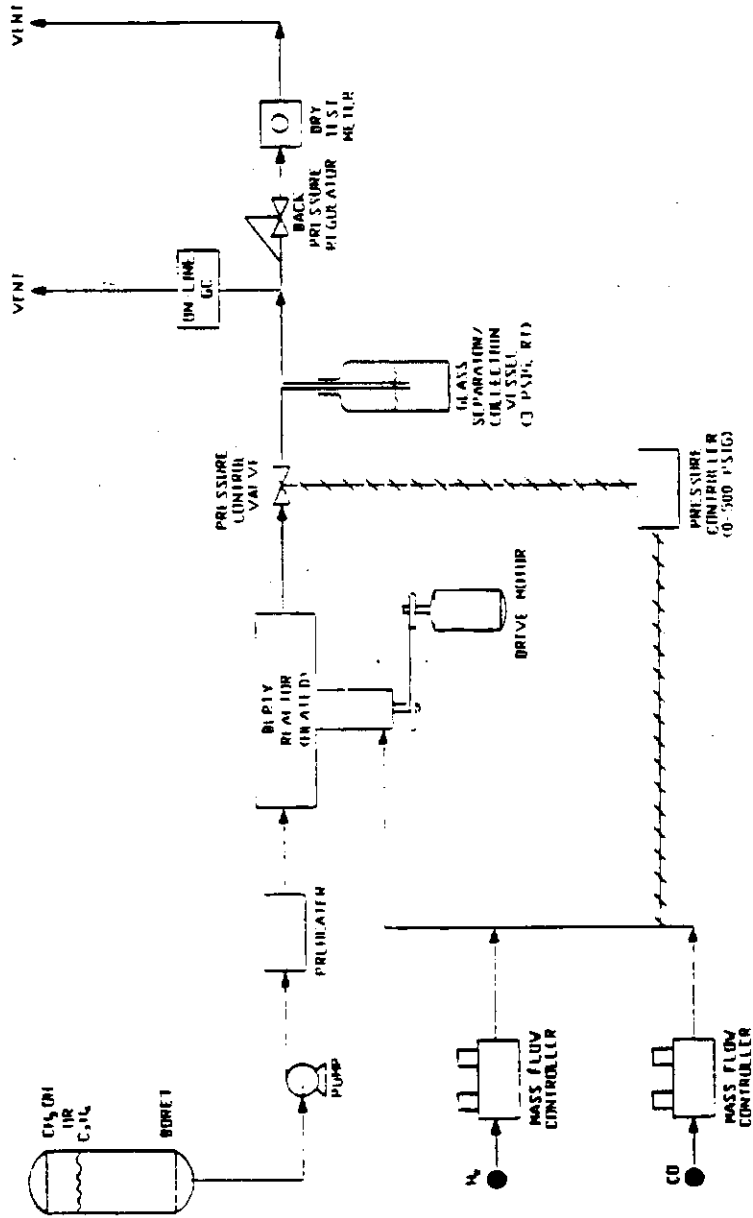


FIGURE 2
SCHEMATIC OF THE CATALYST TEST SYSTEM

FIG. 3
 STABILITY OF Y-82 WITH CH₃OH
 RUN 9710-7 300 PSIG, 400°C

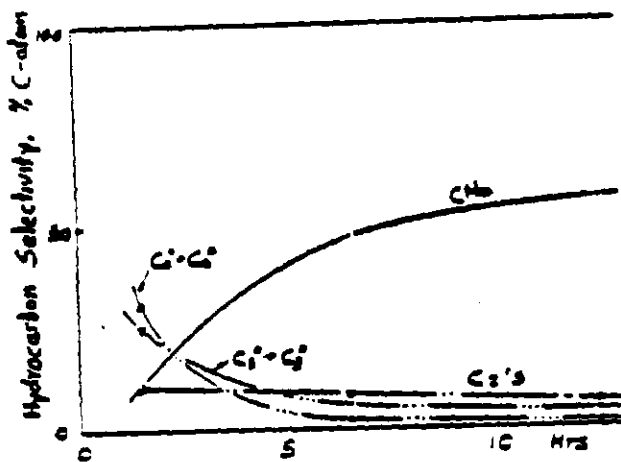
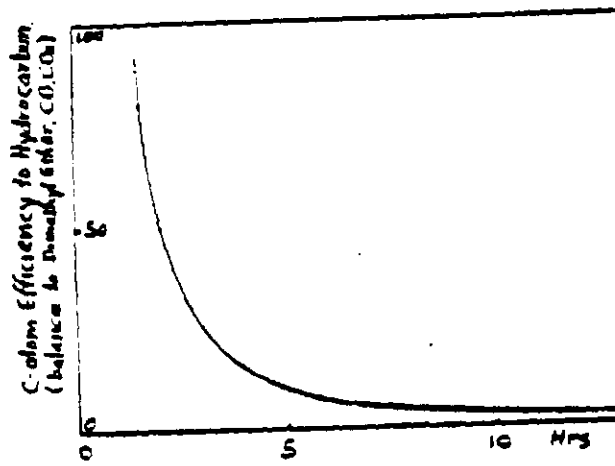
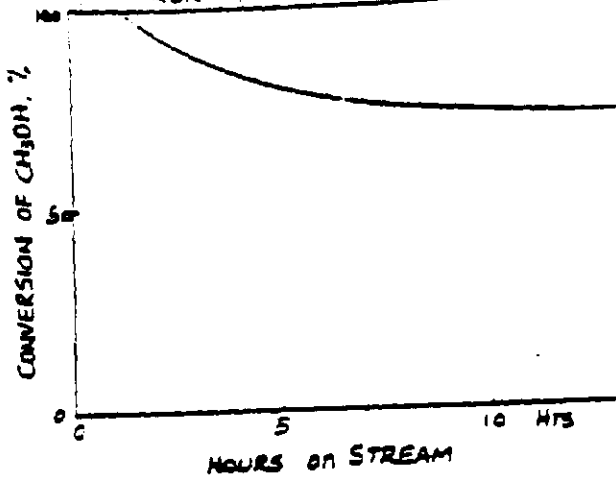


FIG. 4
 STABILITY OF UCC-101
 RUN 9710-8, 120 PSIG, 371-402°C

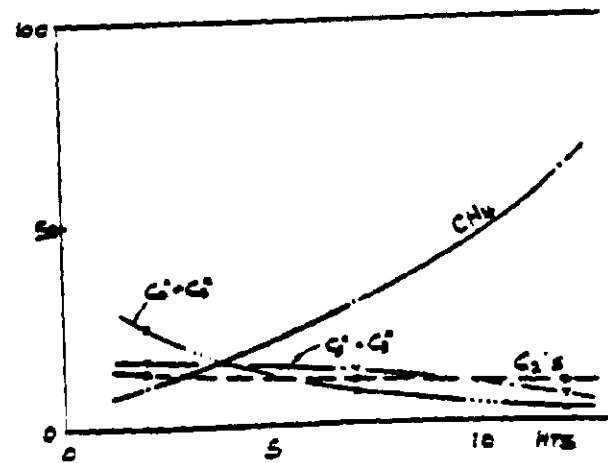
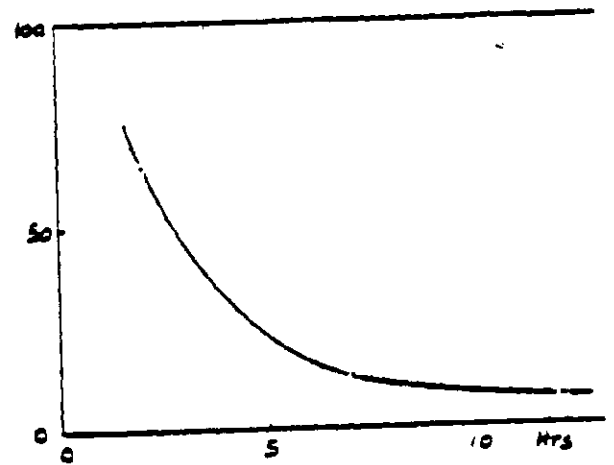
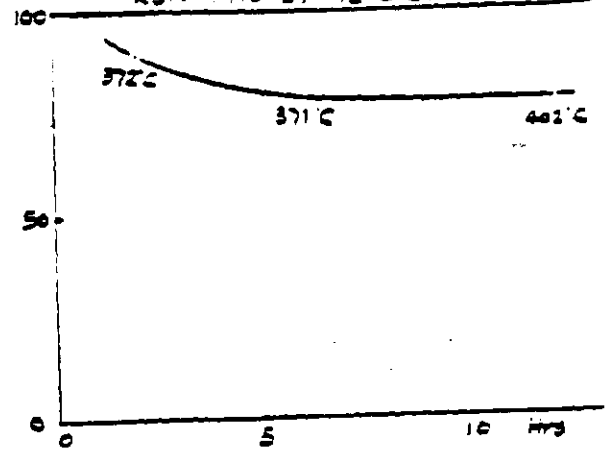
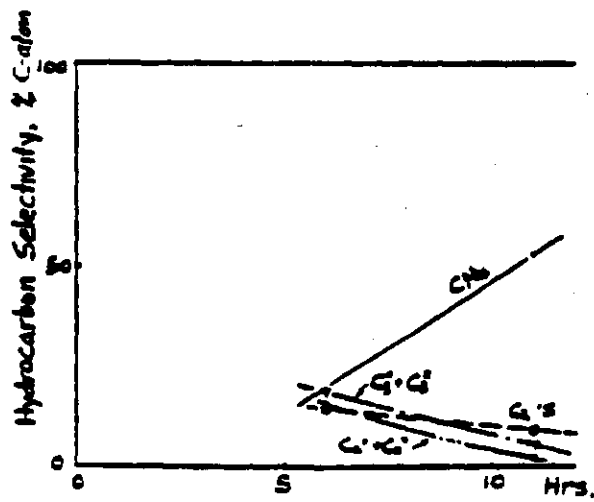
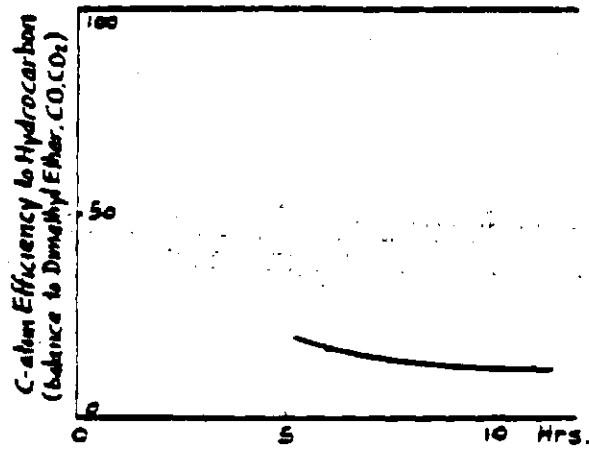
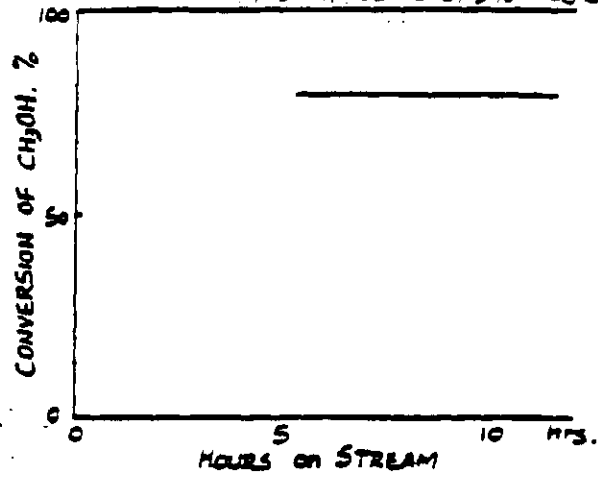
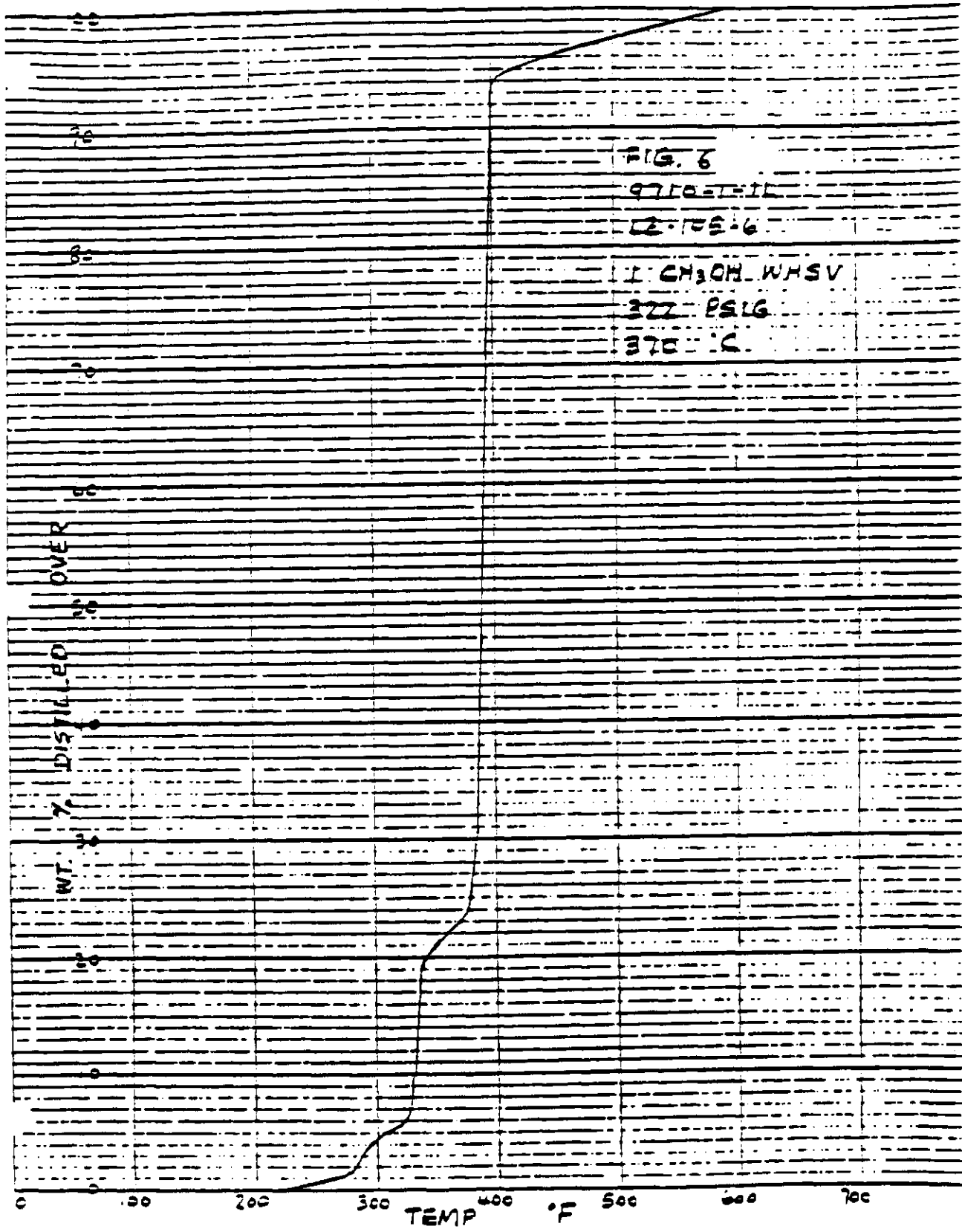
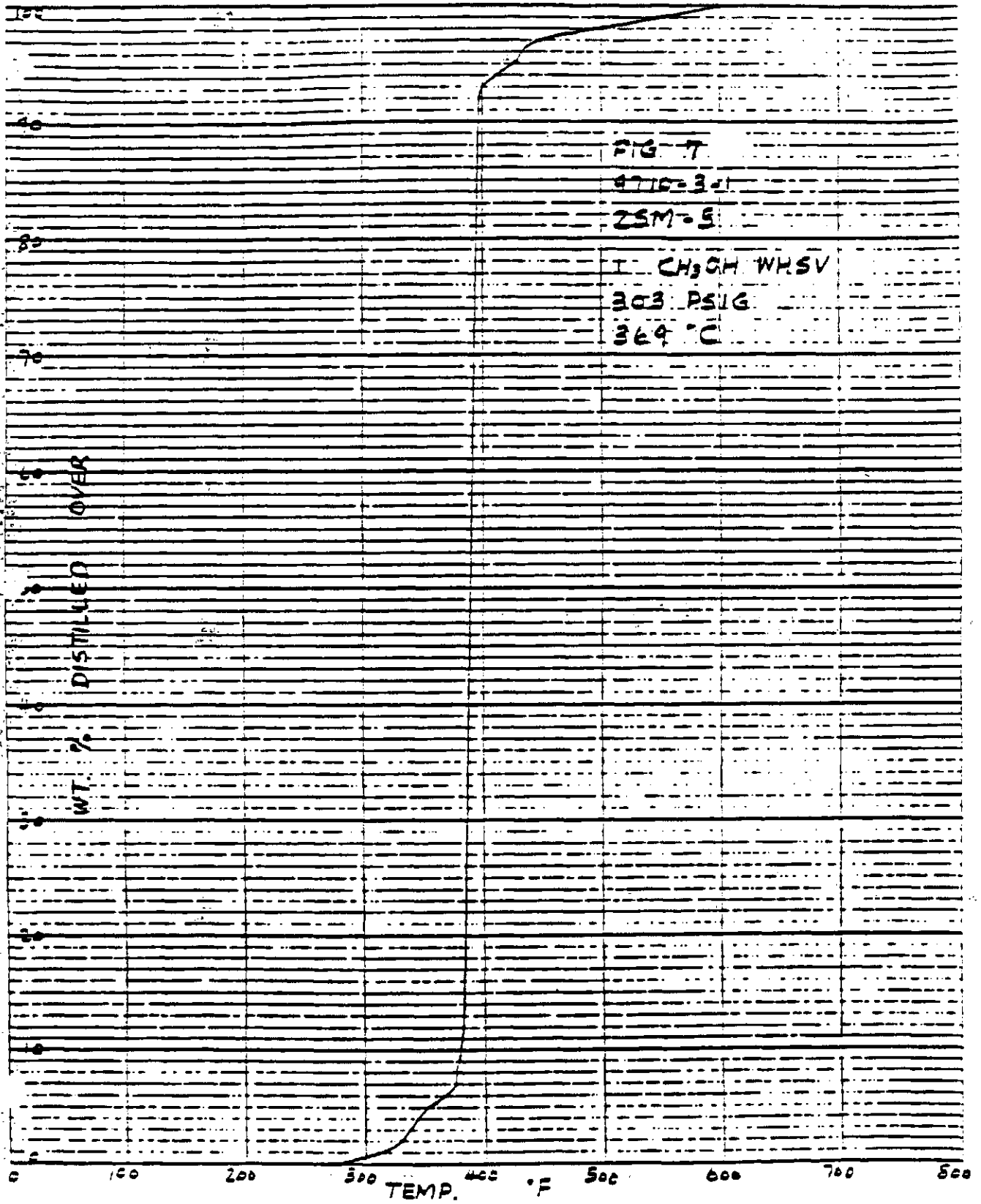
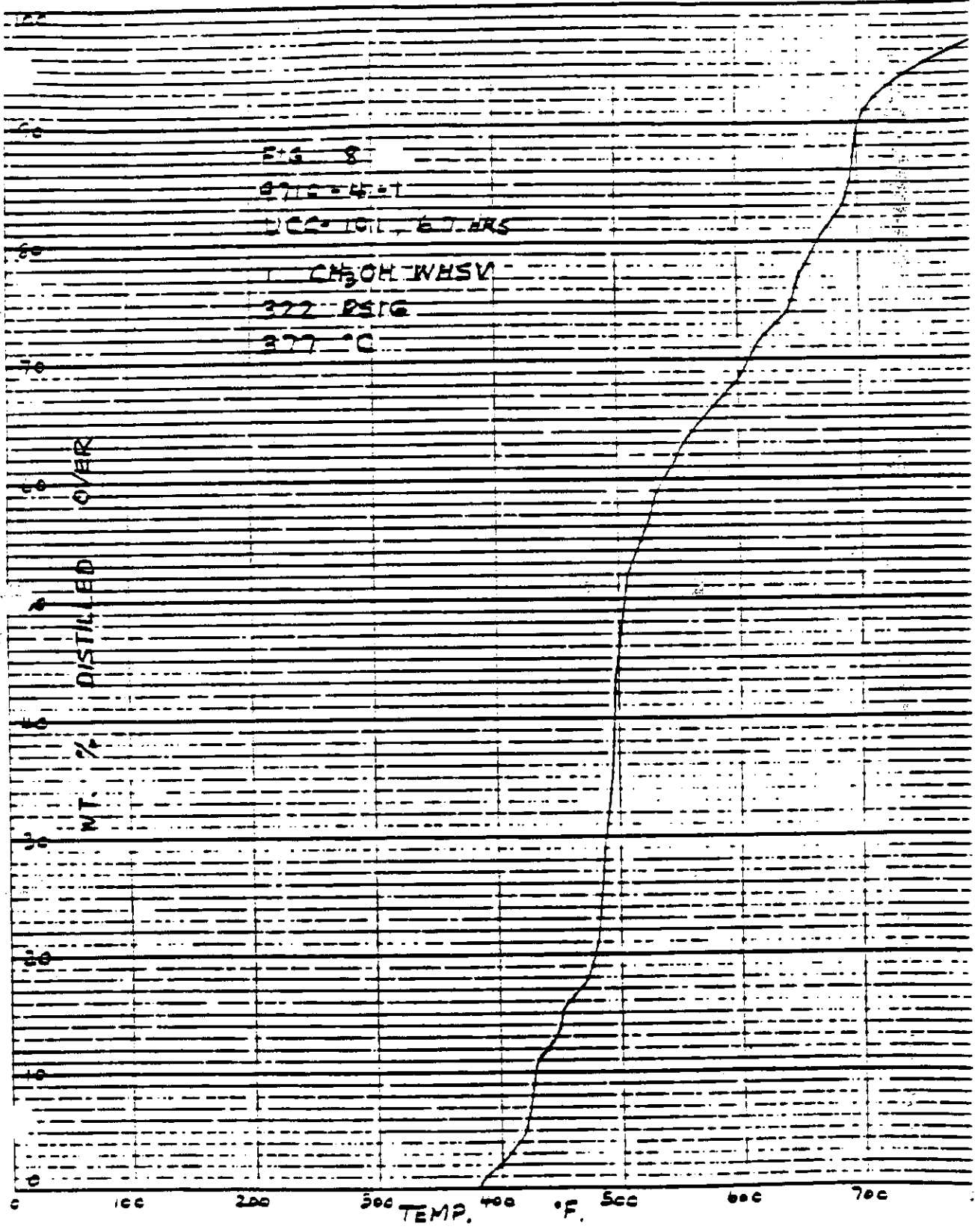


Fig. 5
 STABILITY OF UCC-101
 RUN 4710-9, 25 PSIG, 370-448°C









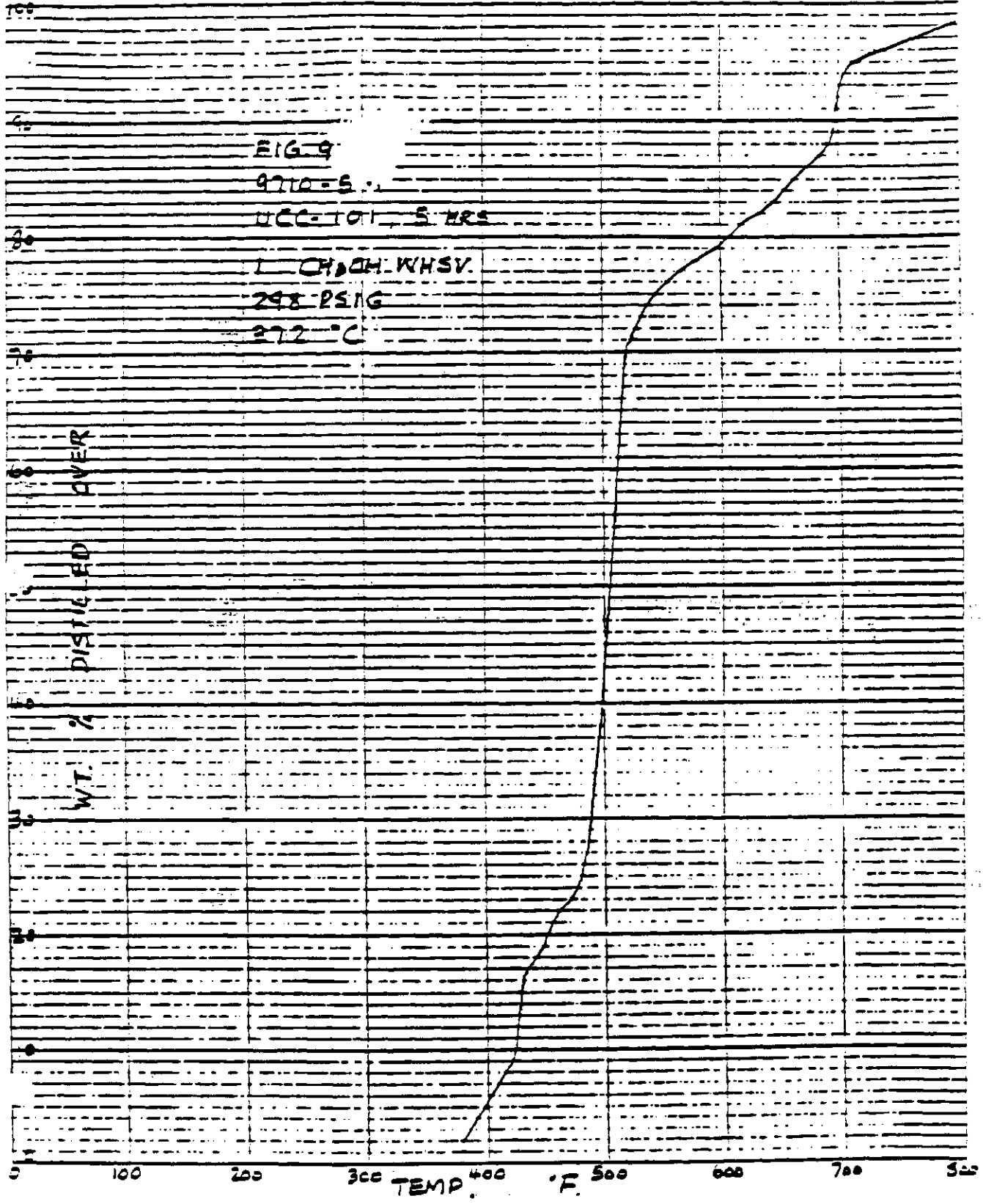
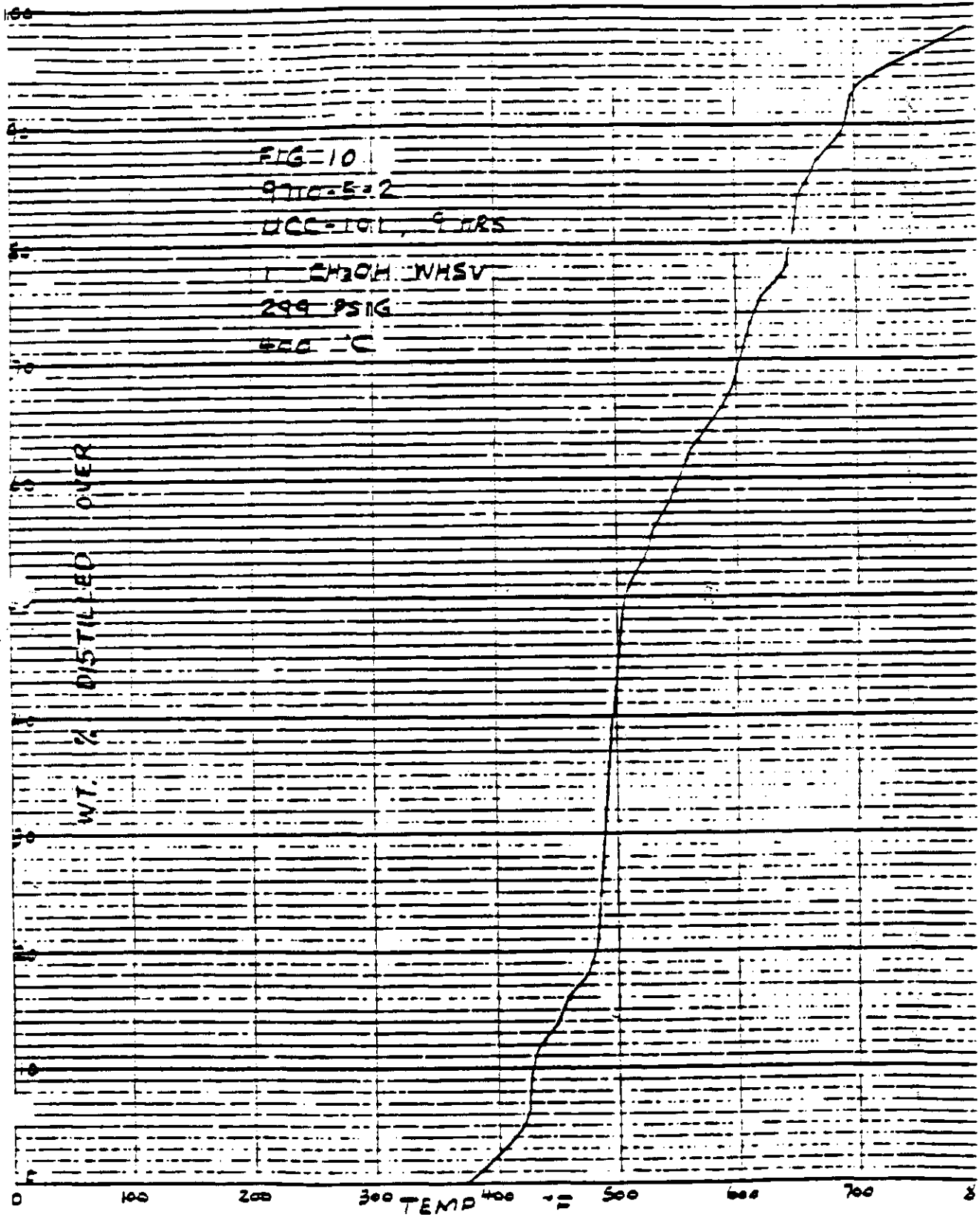
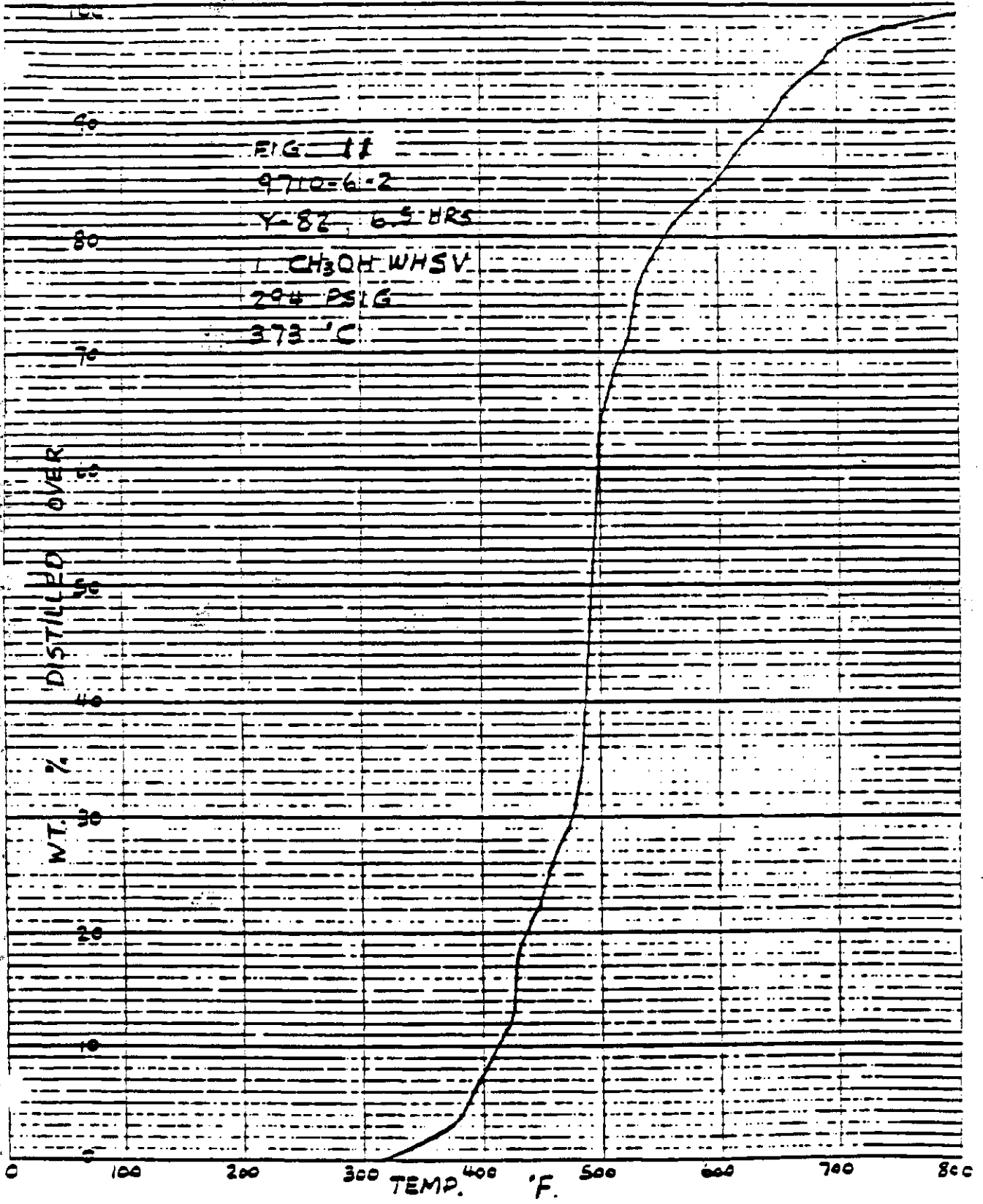
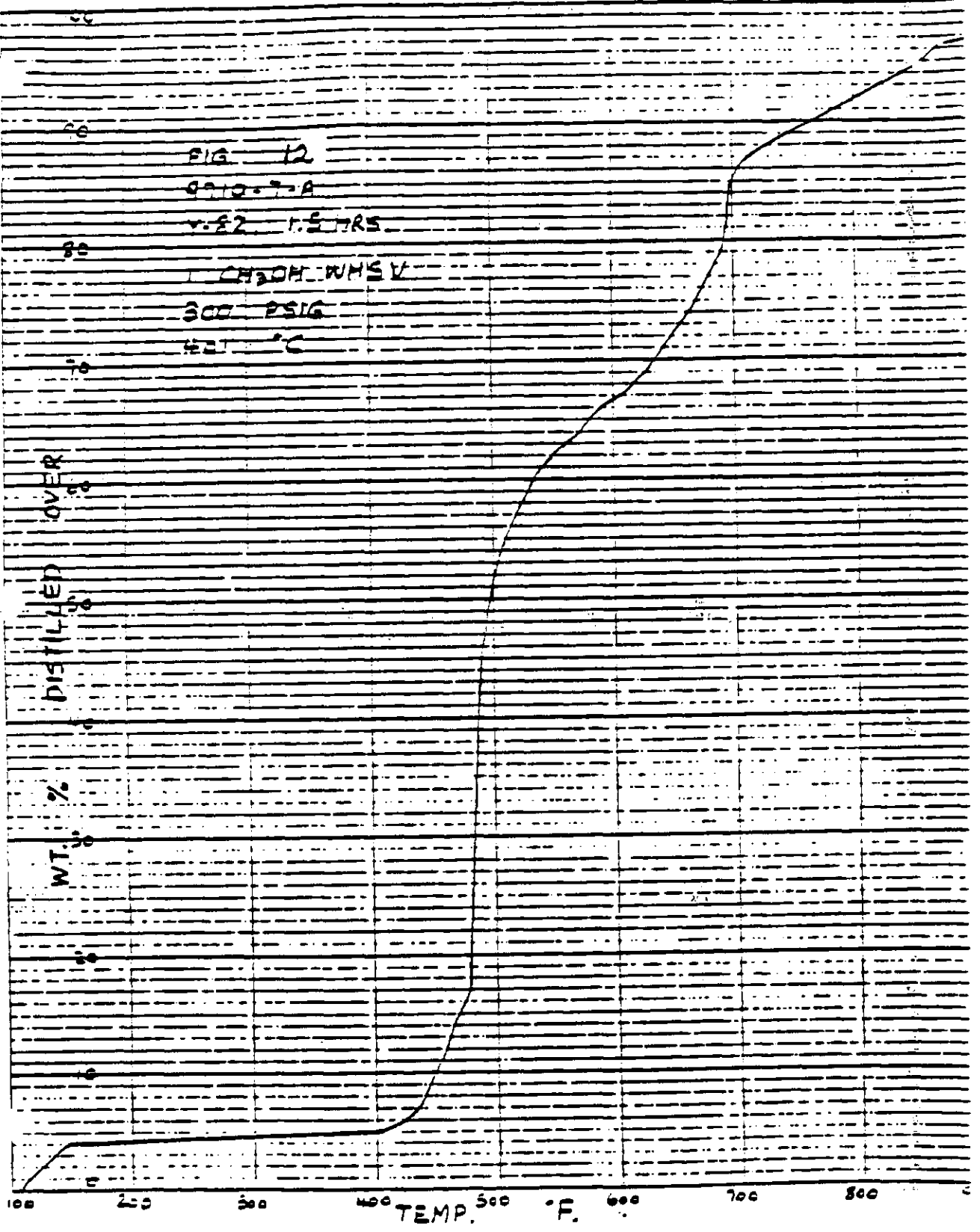
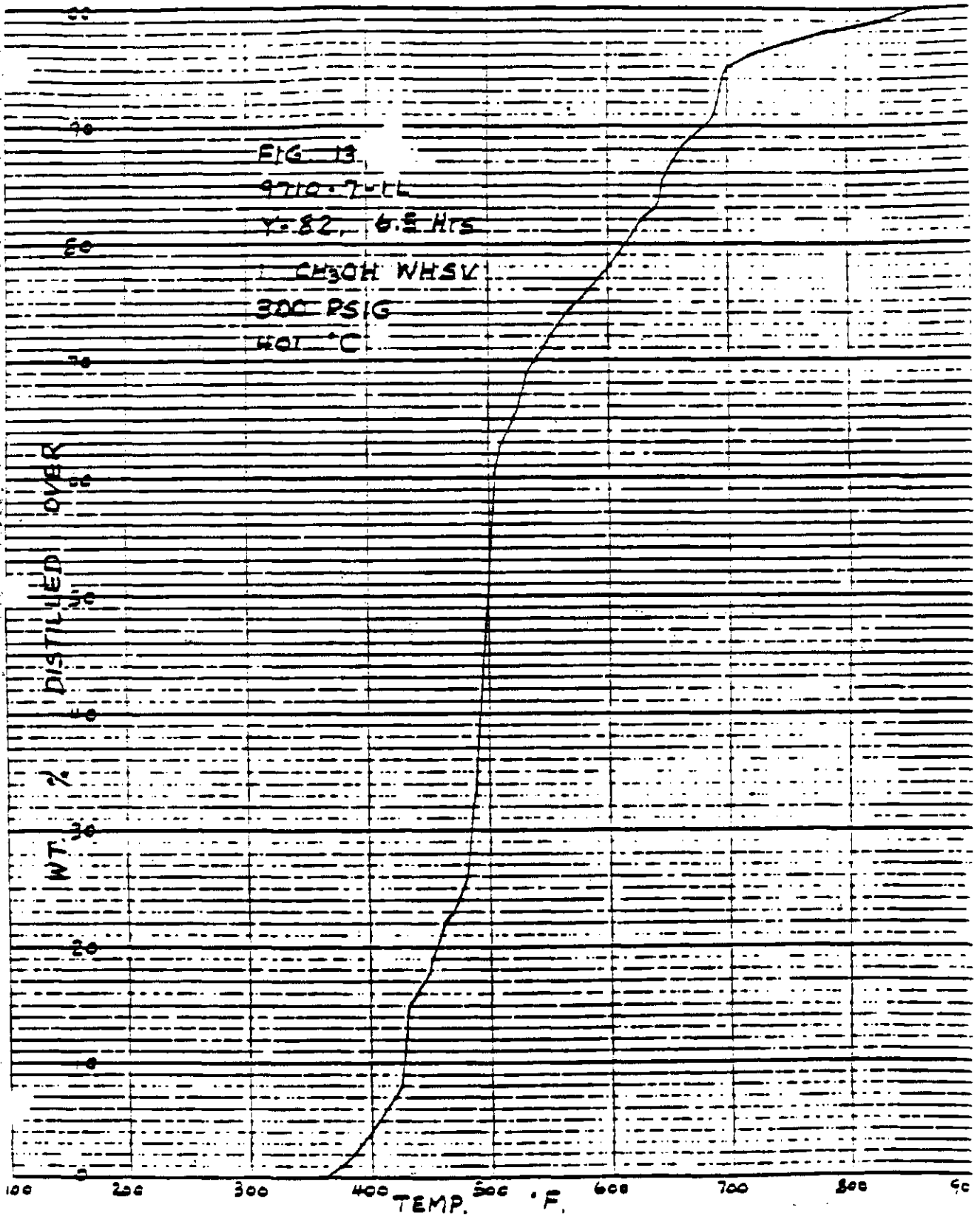


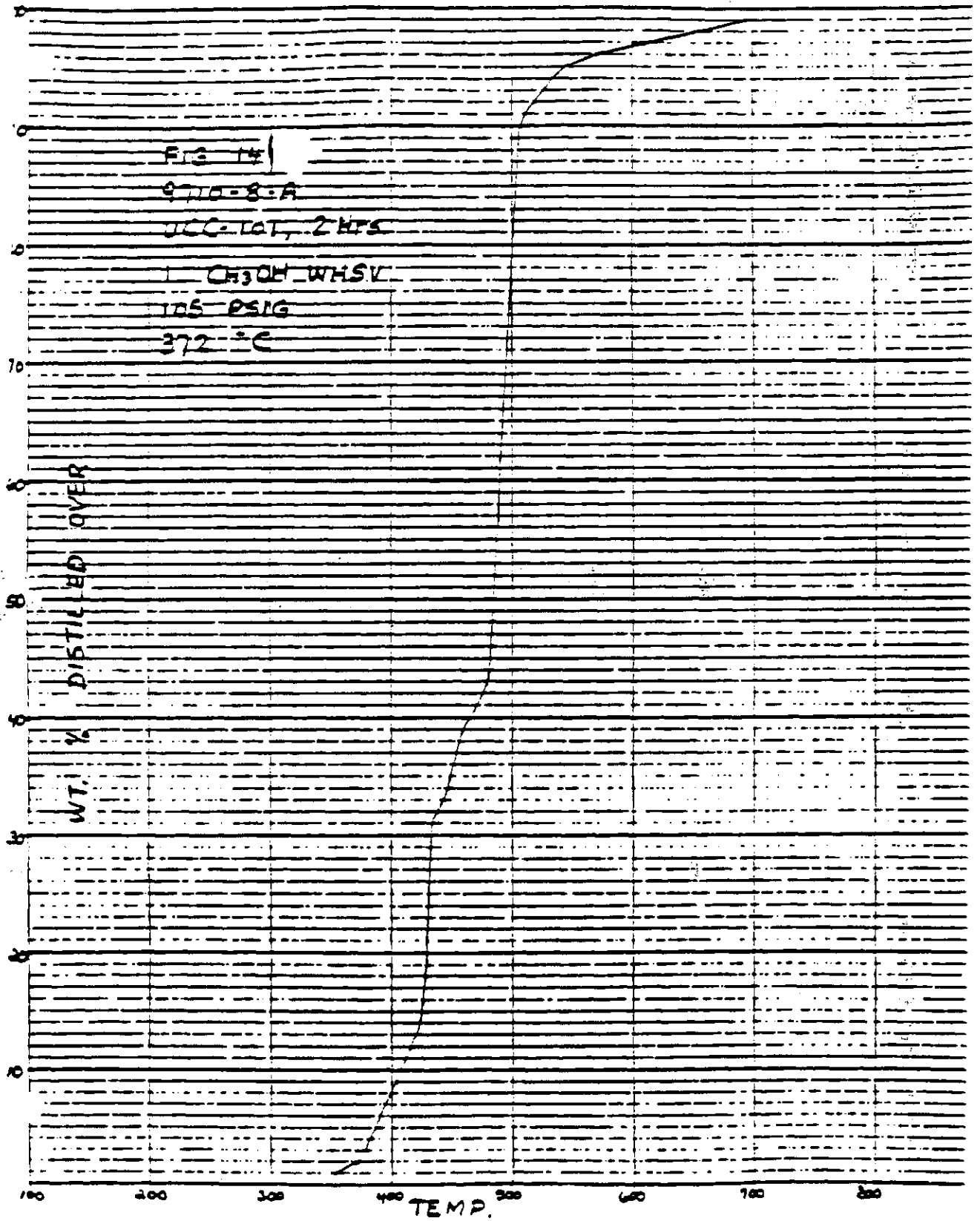
FIG 9
9710-5
UCC-101 5 HRS
1 CH₂OH.WHSV.
298 PSIG
272 °C

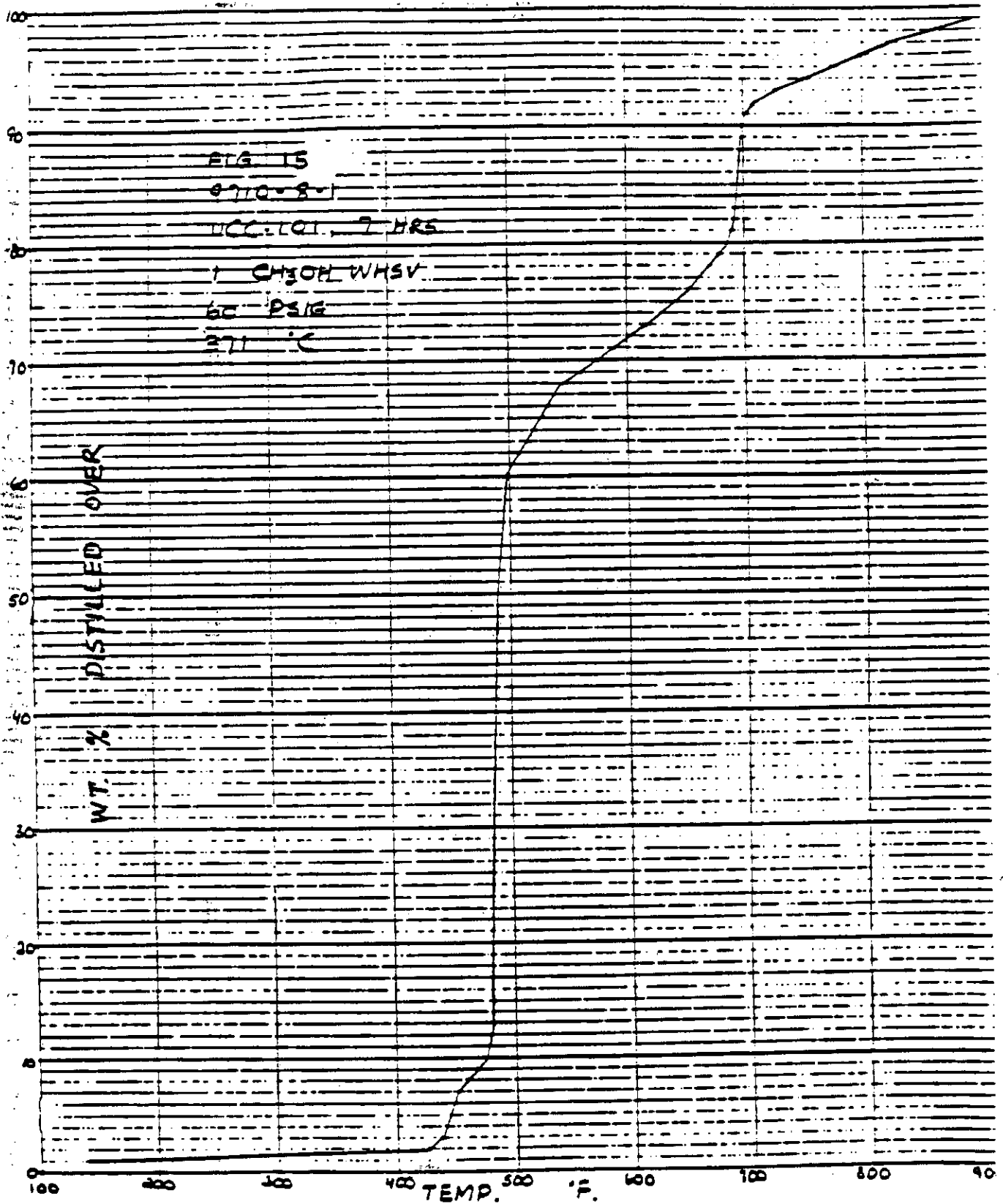


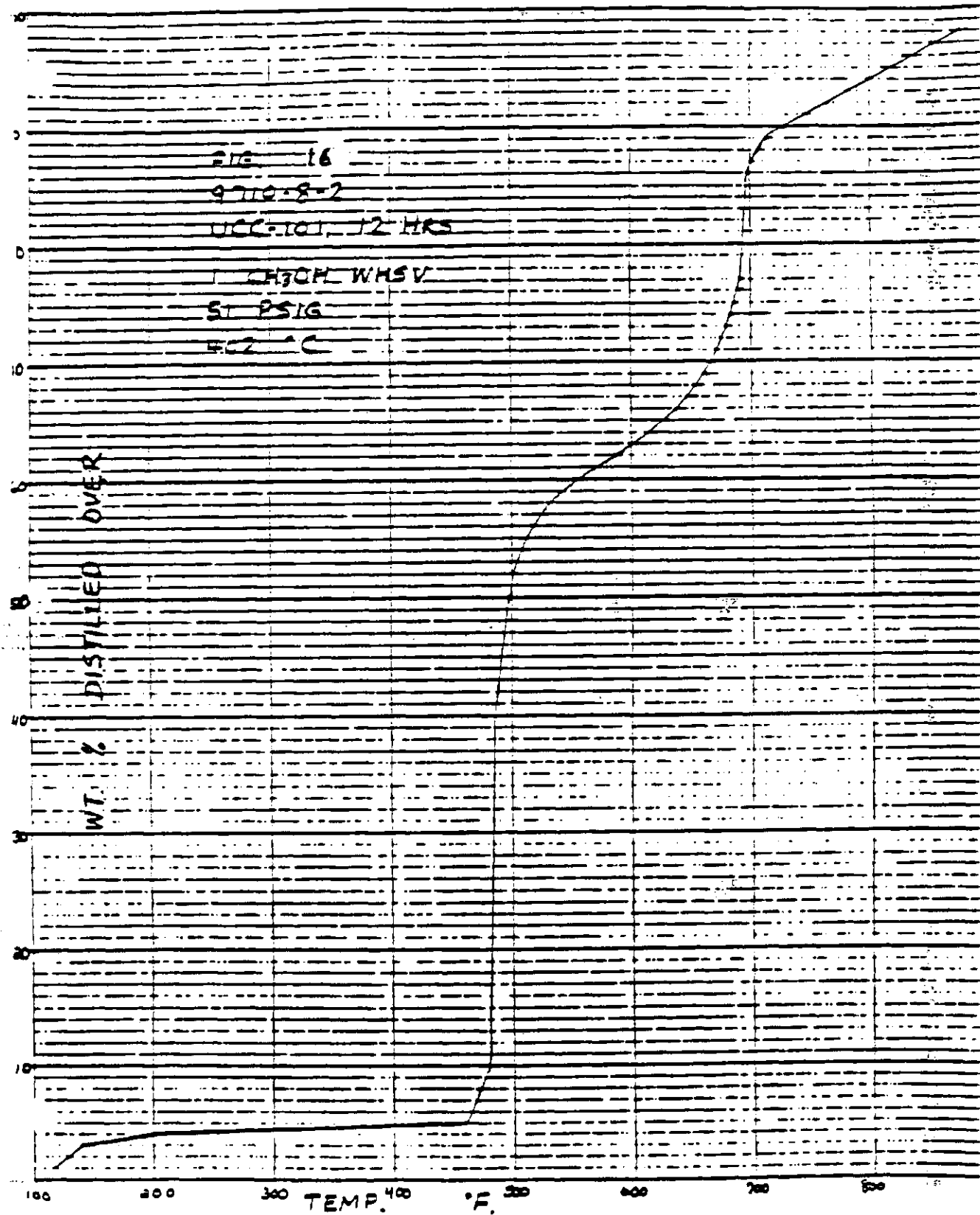






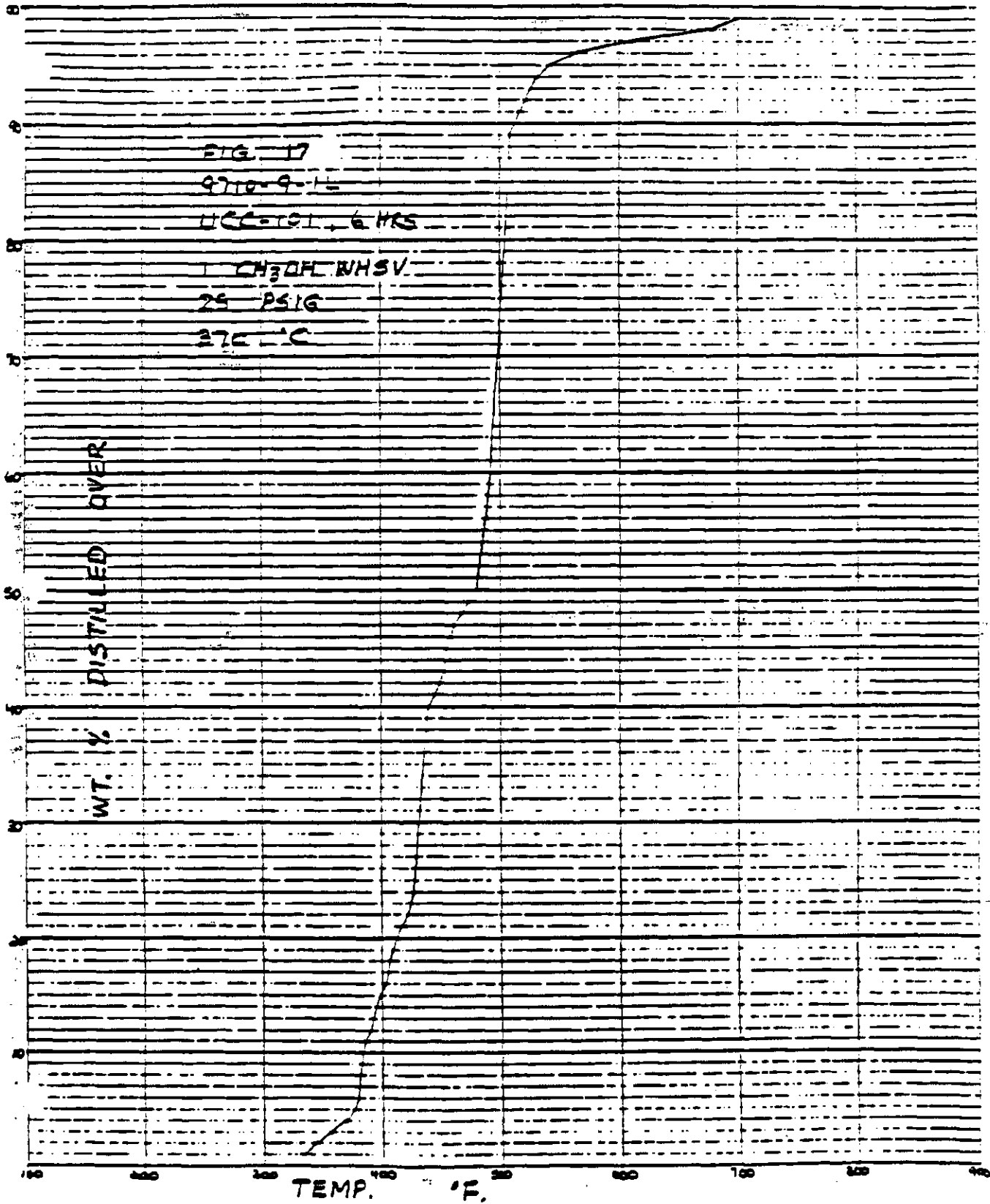


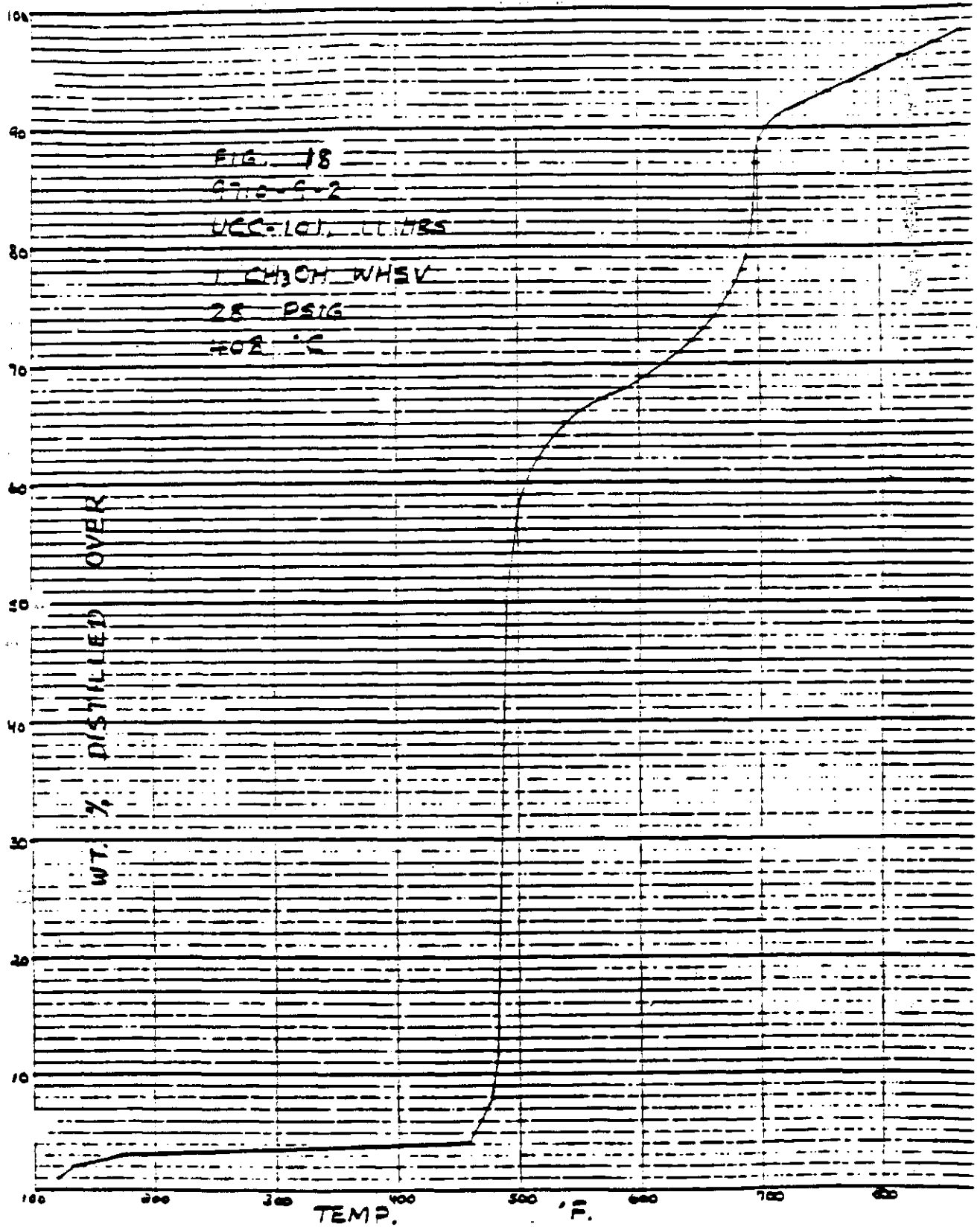


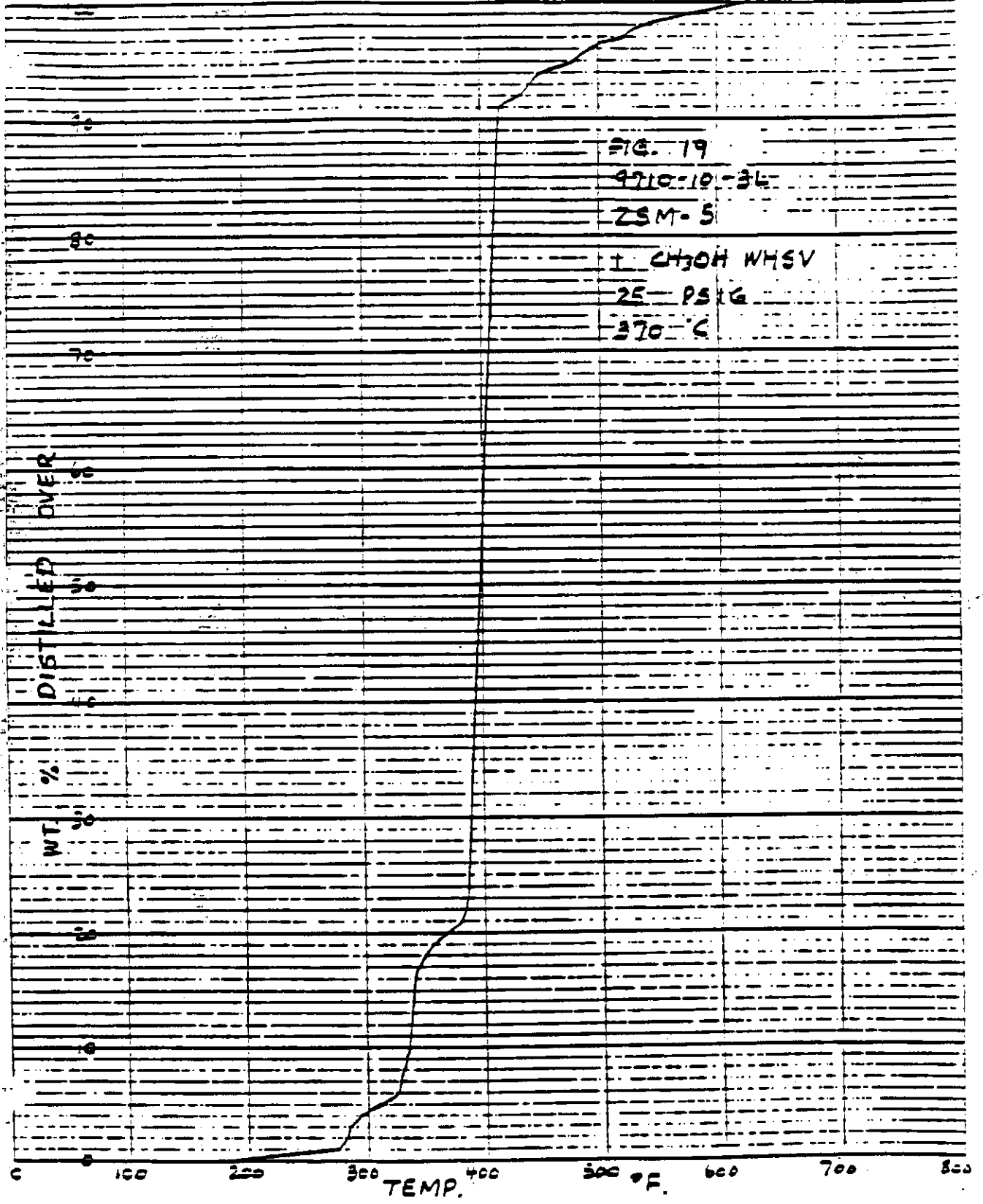


PIG 16
9710-8-2
UCC-101, 12 HRS

CH3OH WMSV
51 PSIG
402 °C







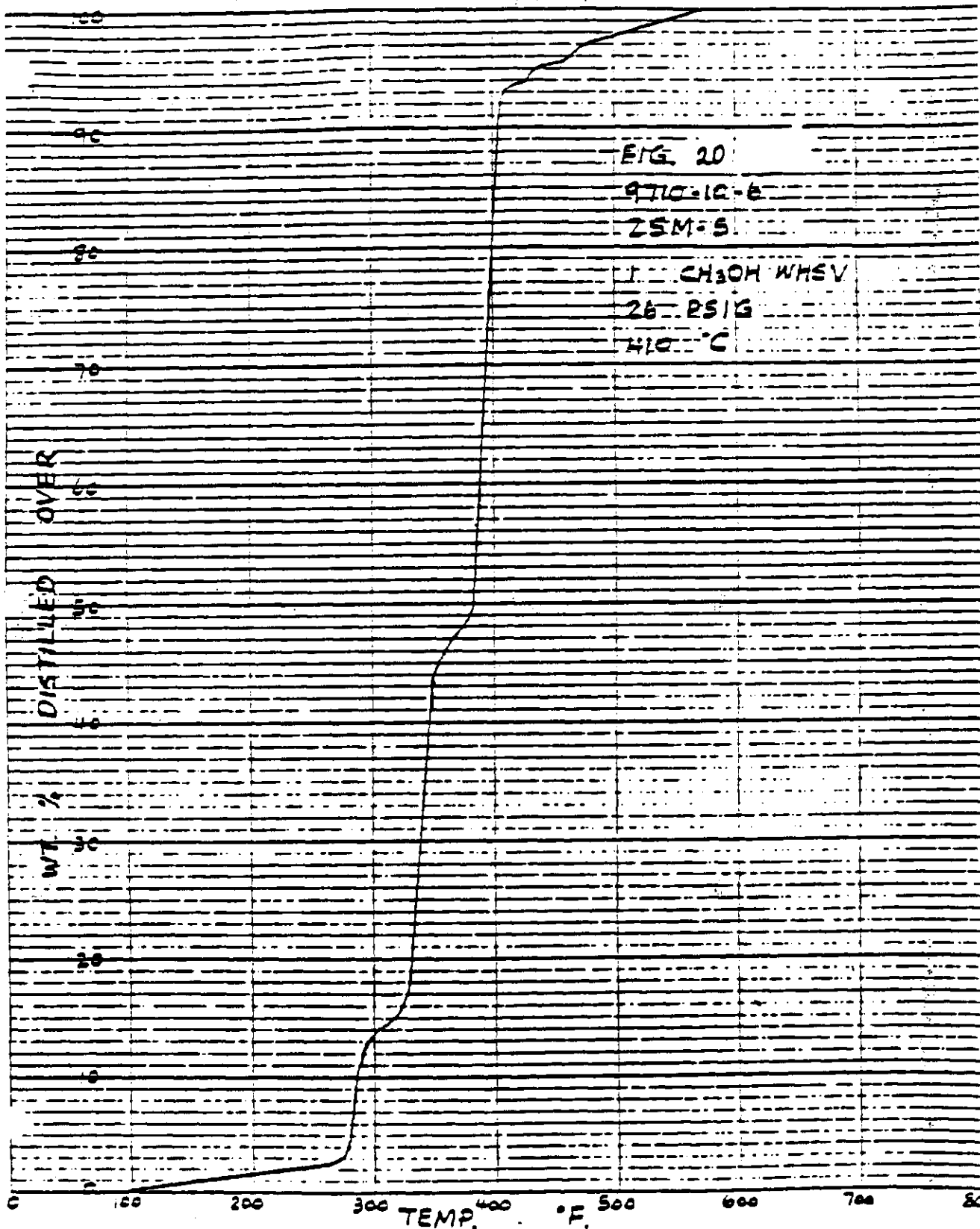


FIG. 20

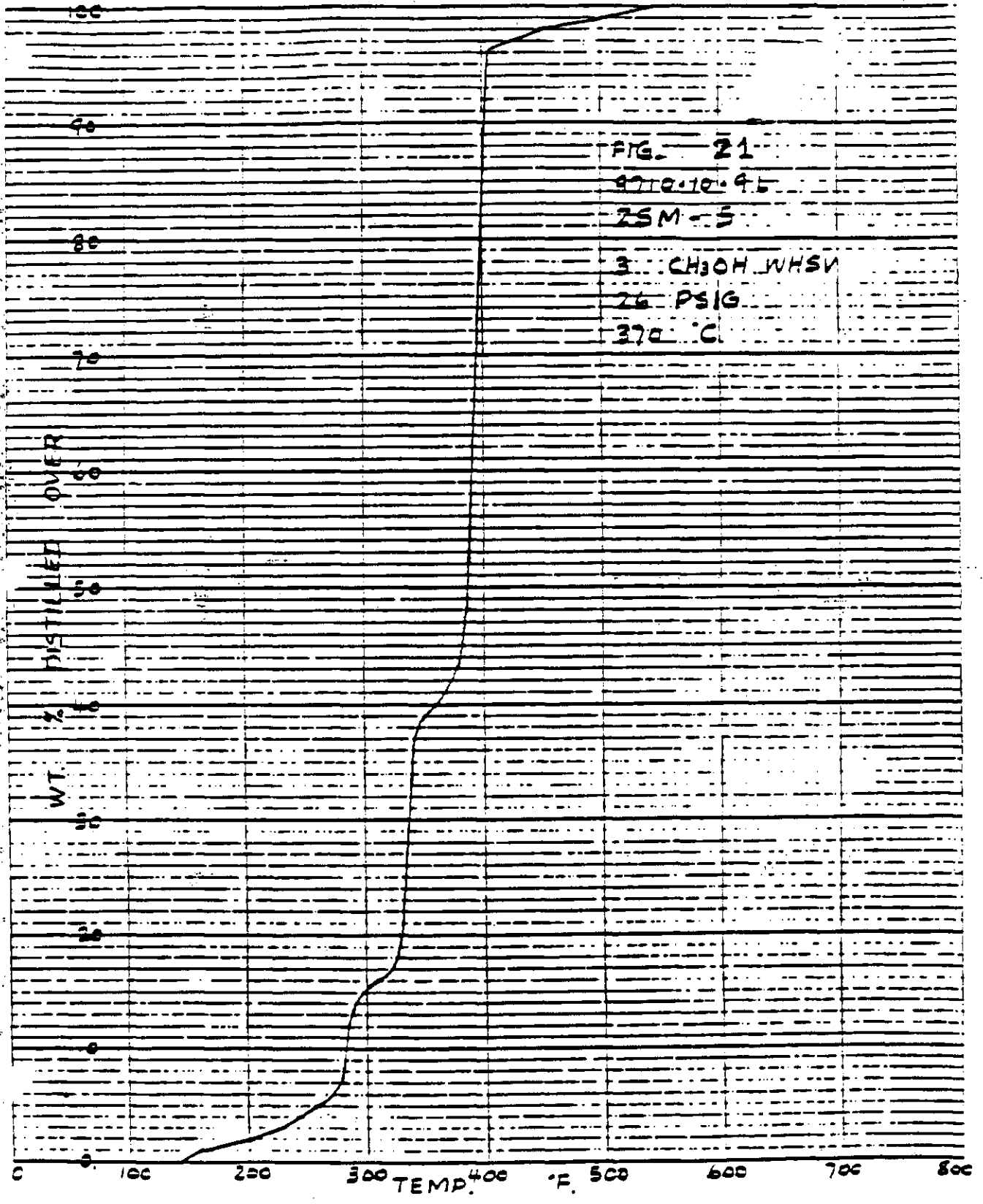
9710-10-6

25M-5

J. CH₃OH WHSV

26 PSIG

410 °C



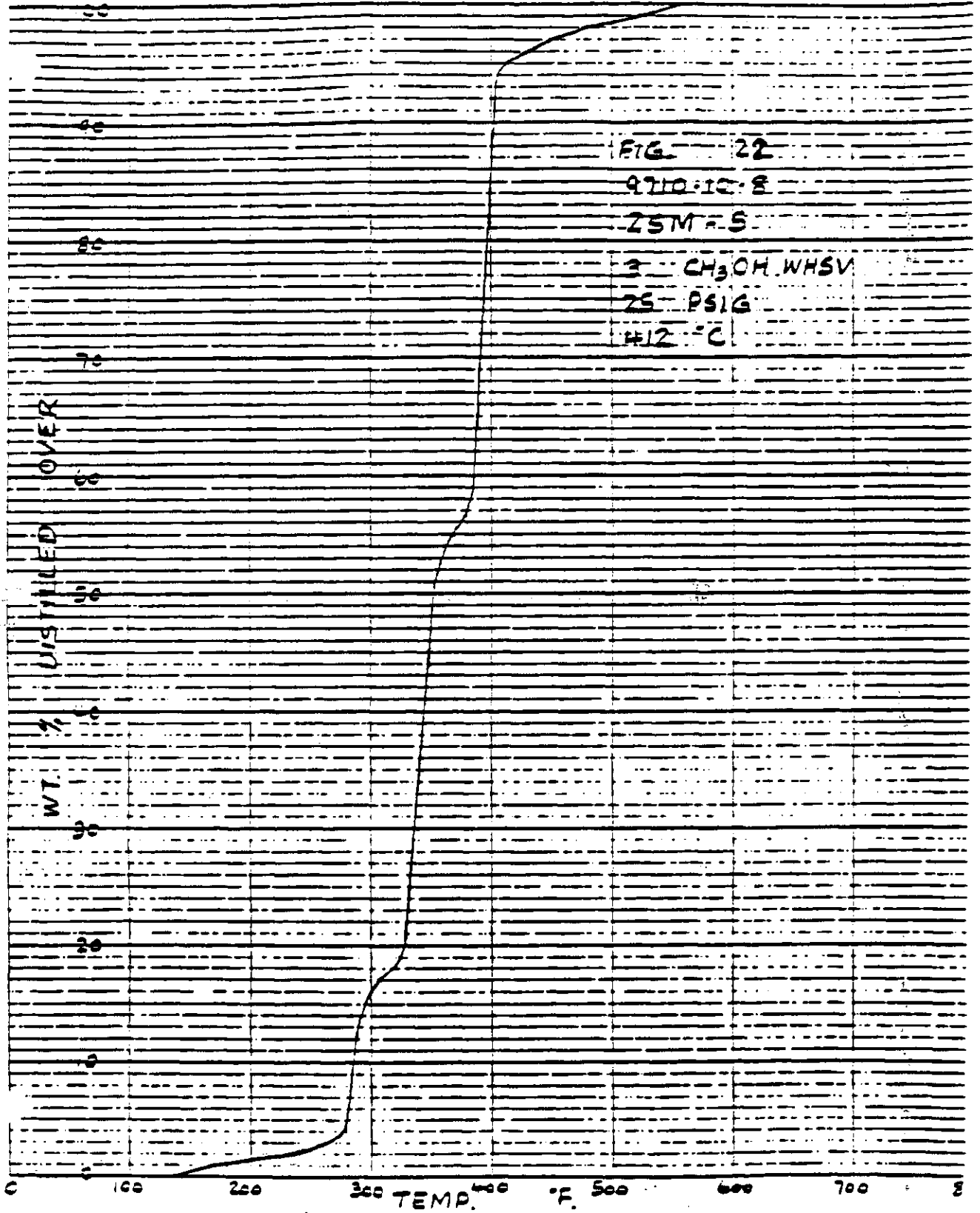
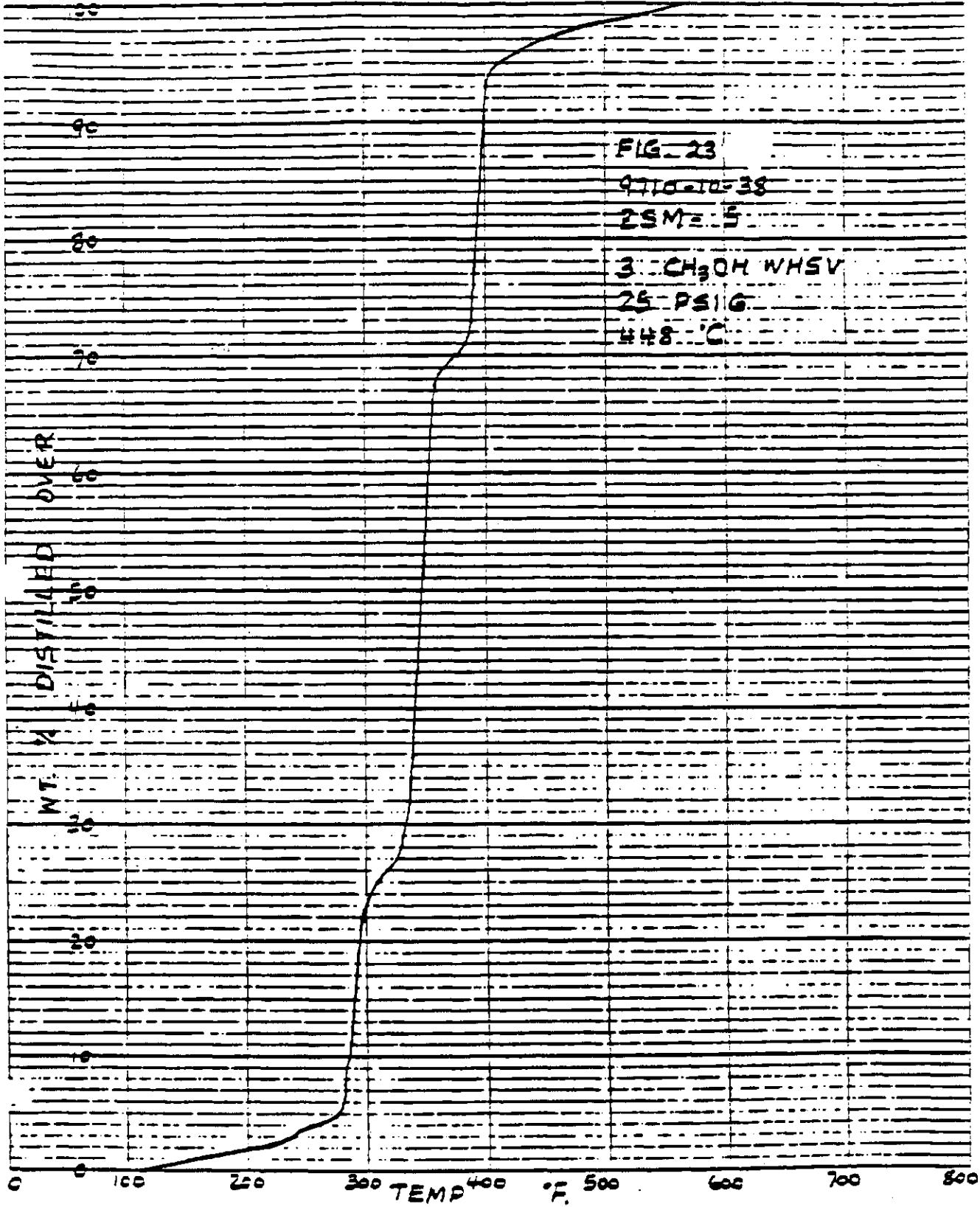
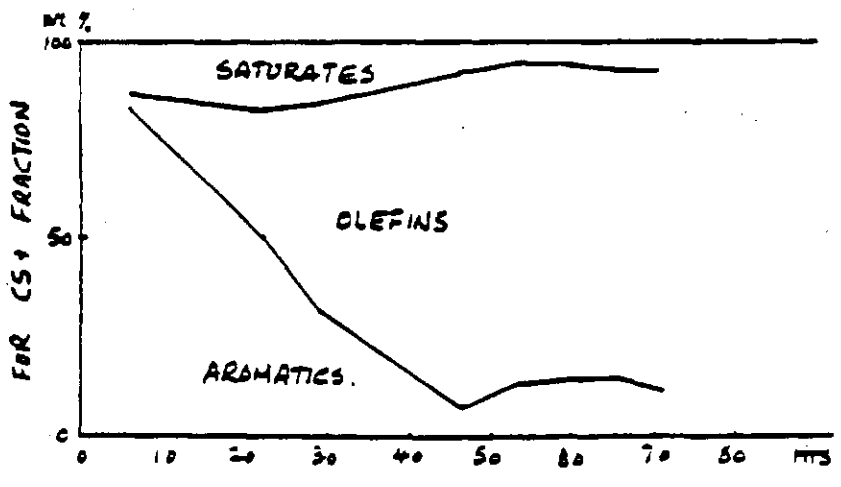
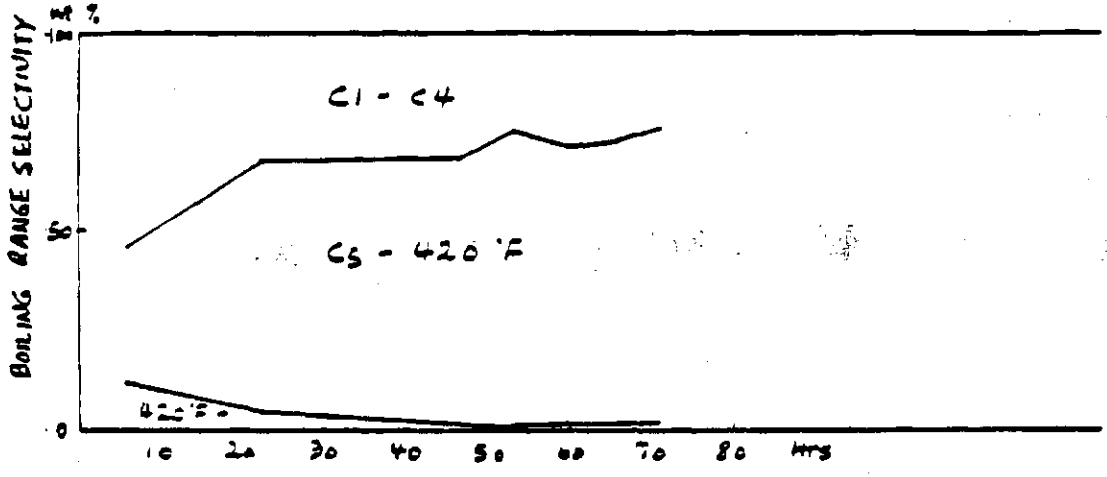
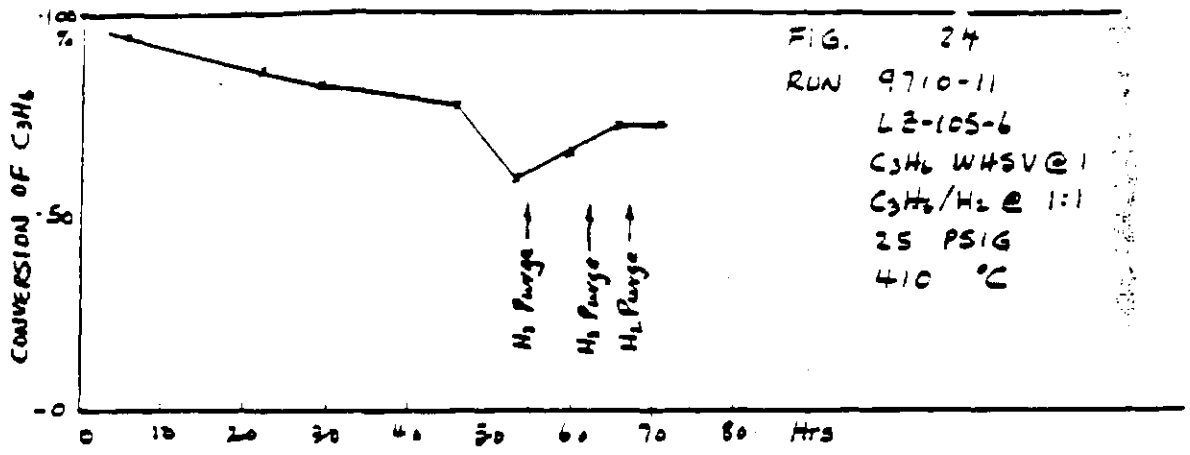
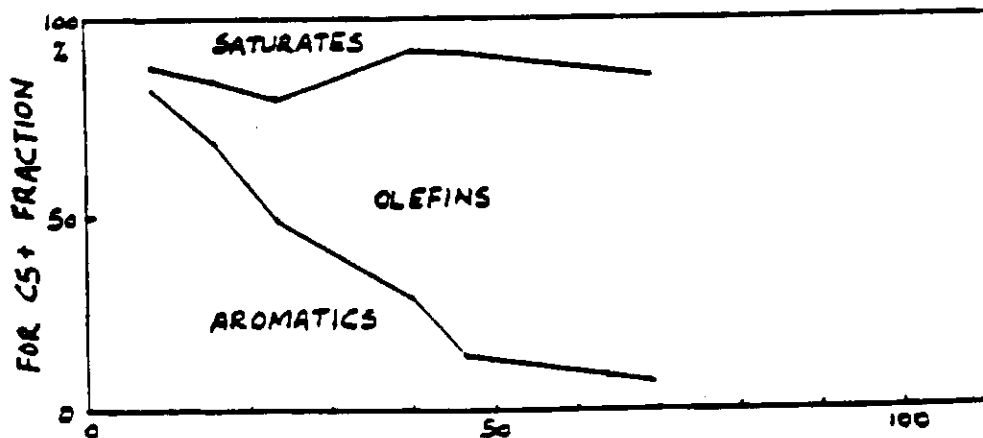
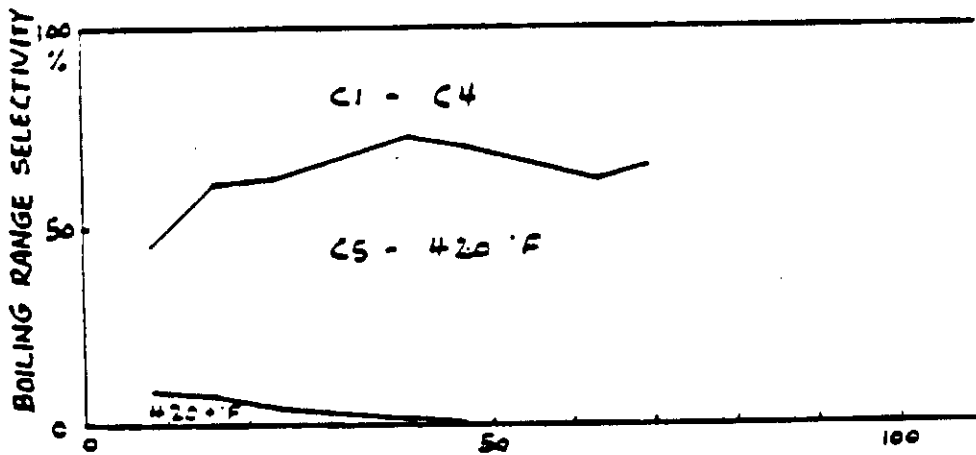
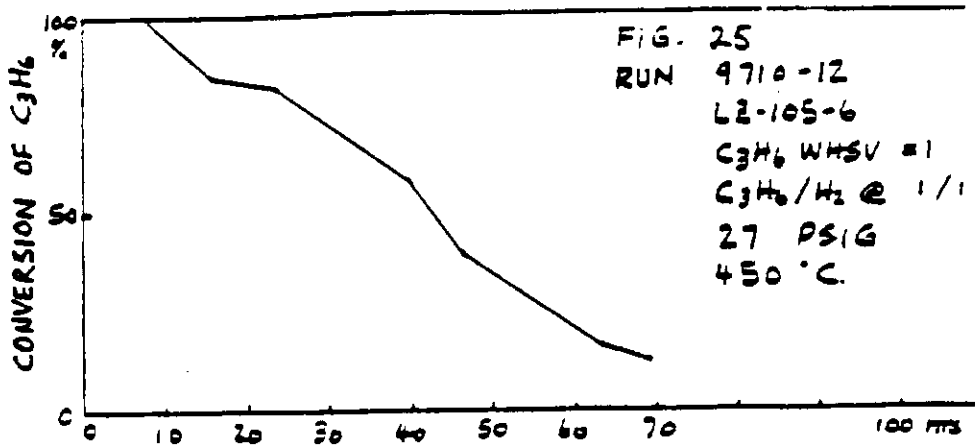


FIG. 22
9710-IC-8
ZSM-5
3 CH₃OH WHSV
25 PSIG
#12 °C







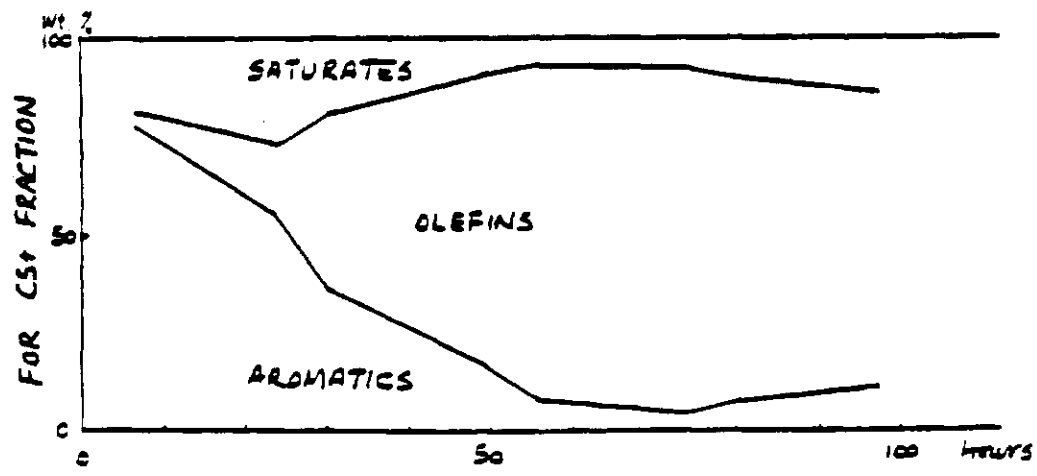
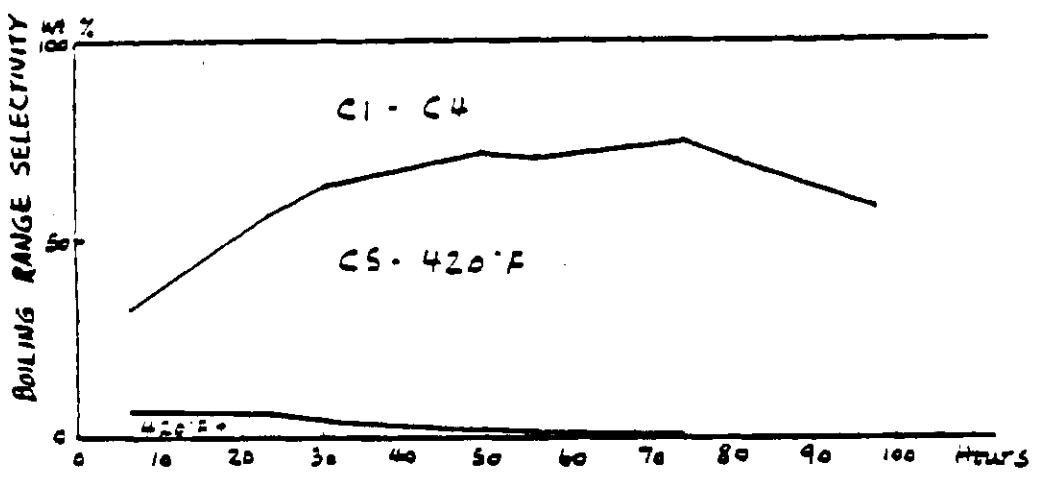
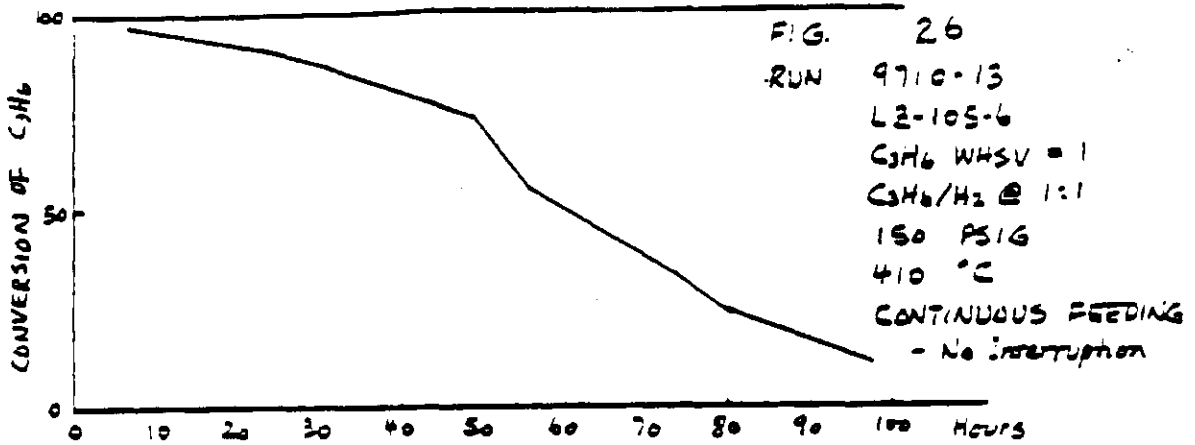


FIGURE
CATALYST

27
UCC-101

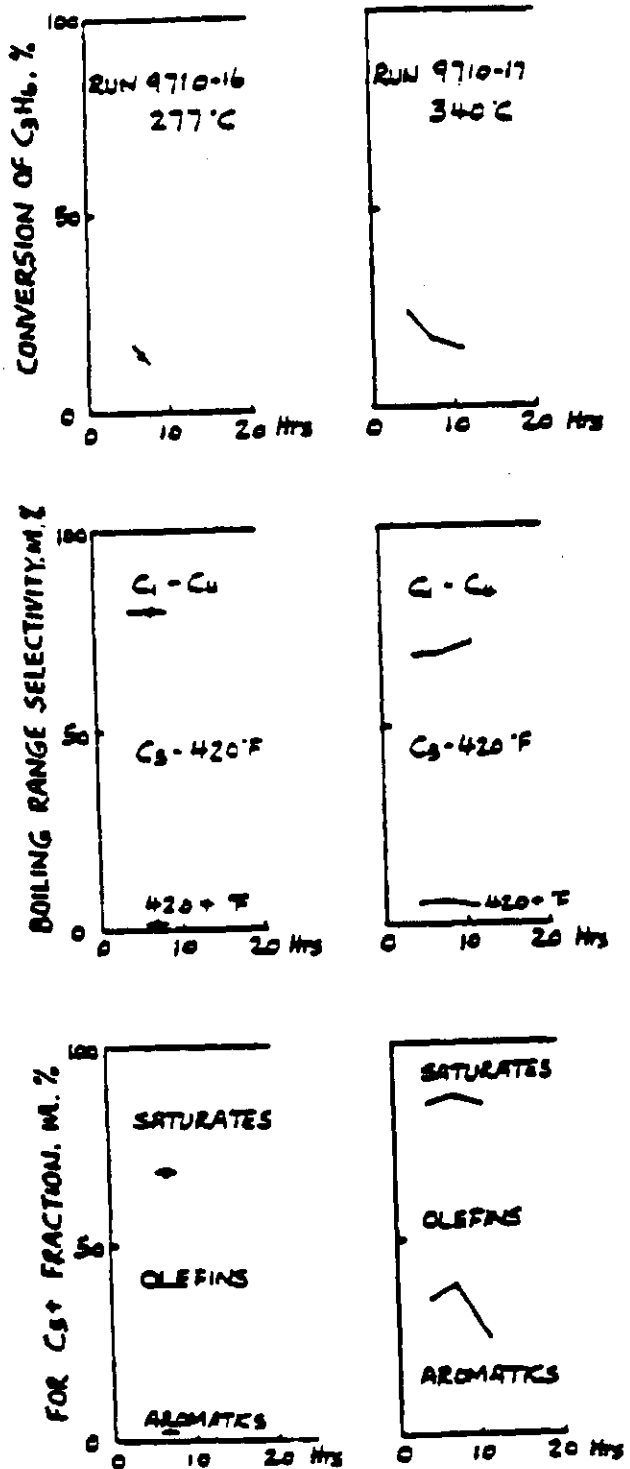


FIGURE 28
UCC-101

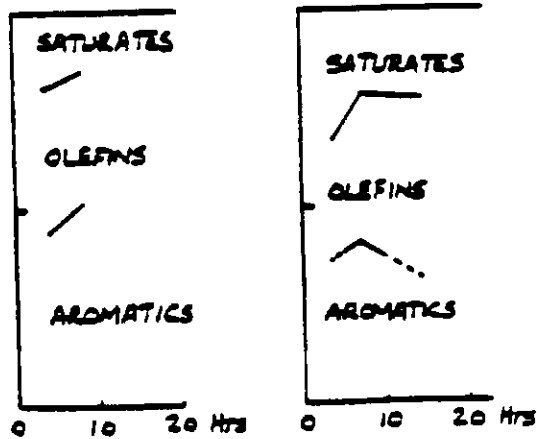
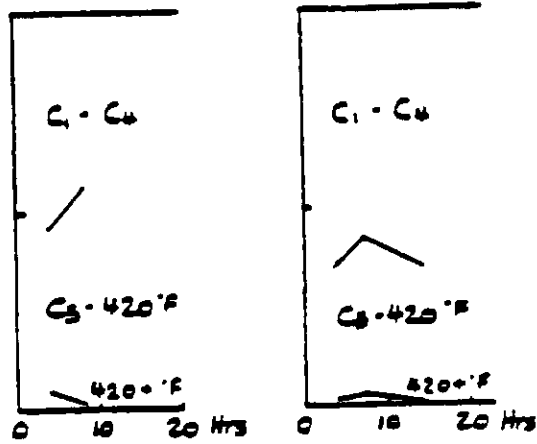
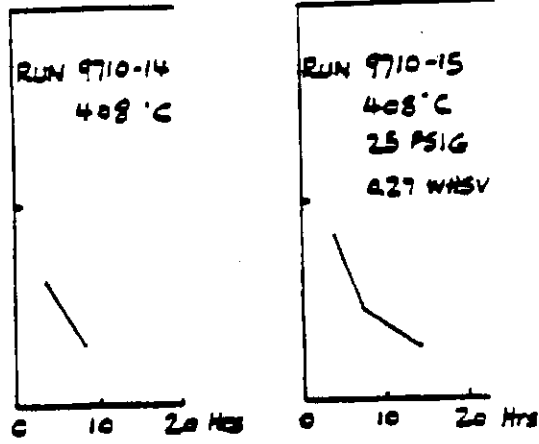


FIG. 29

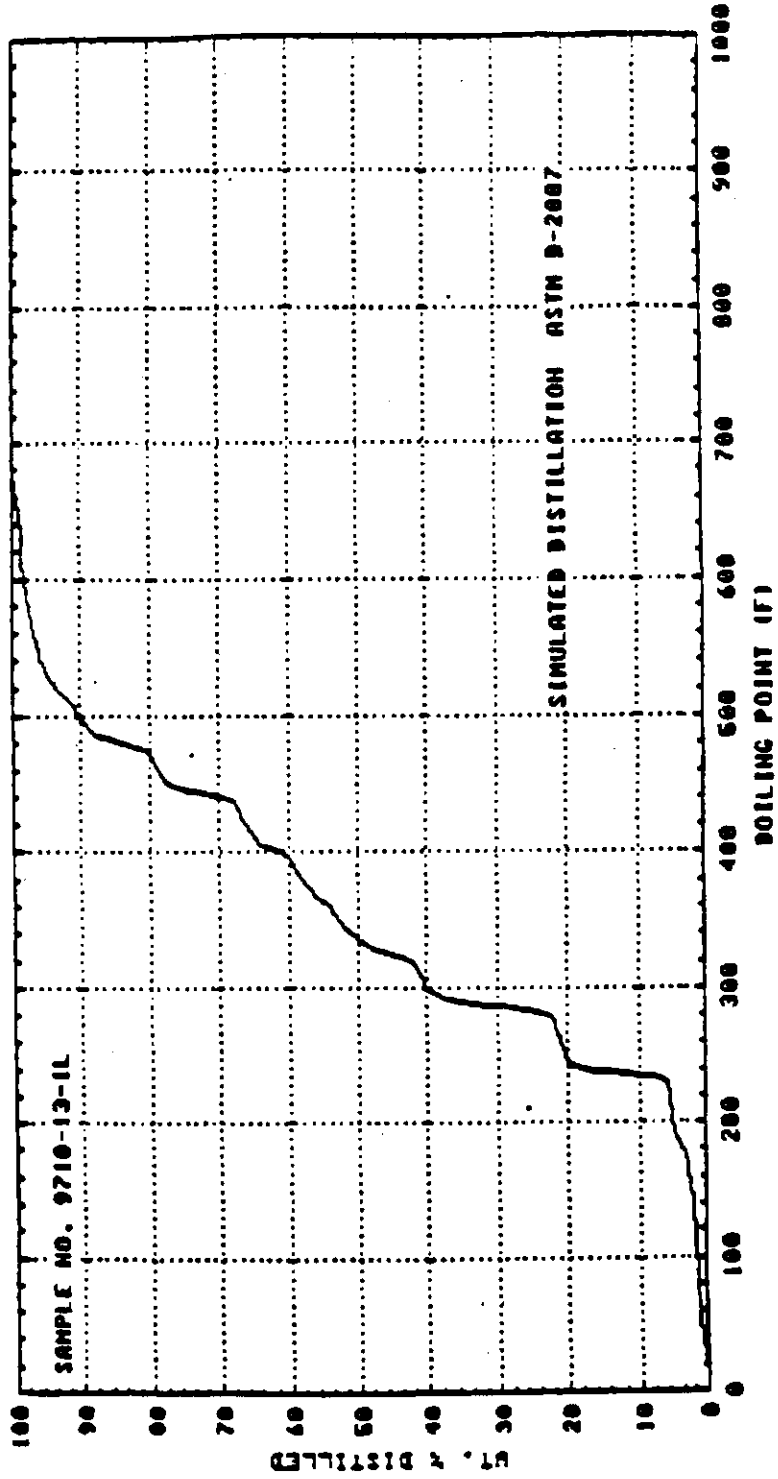


FIG. 30

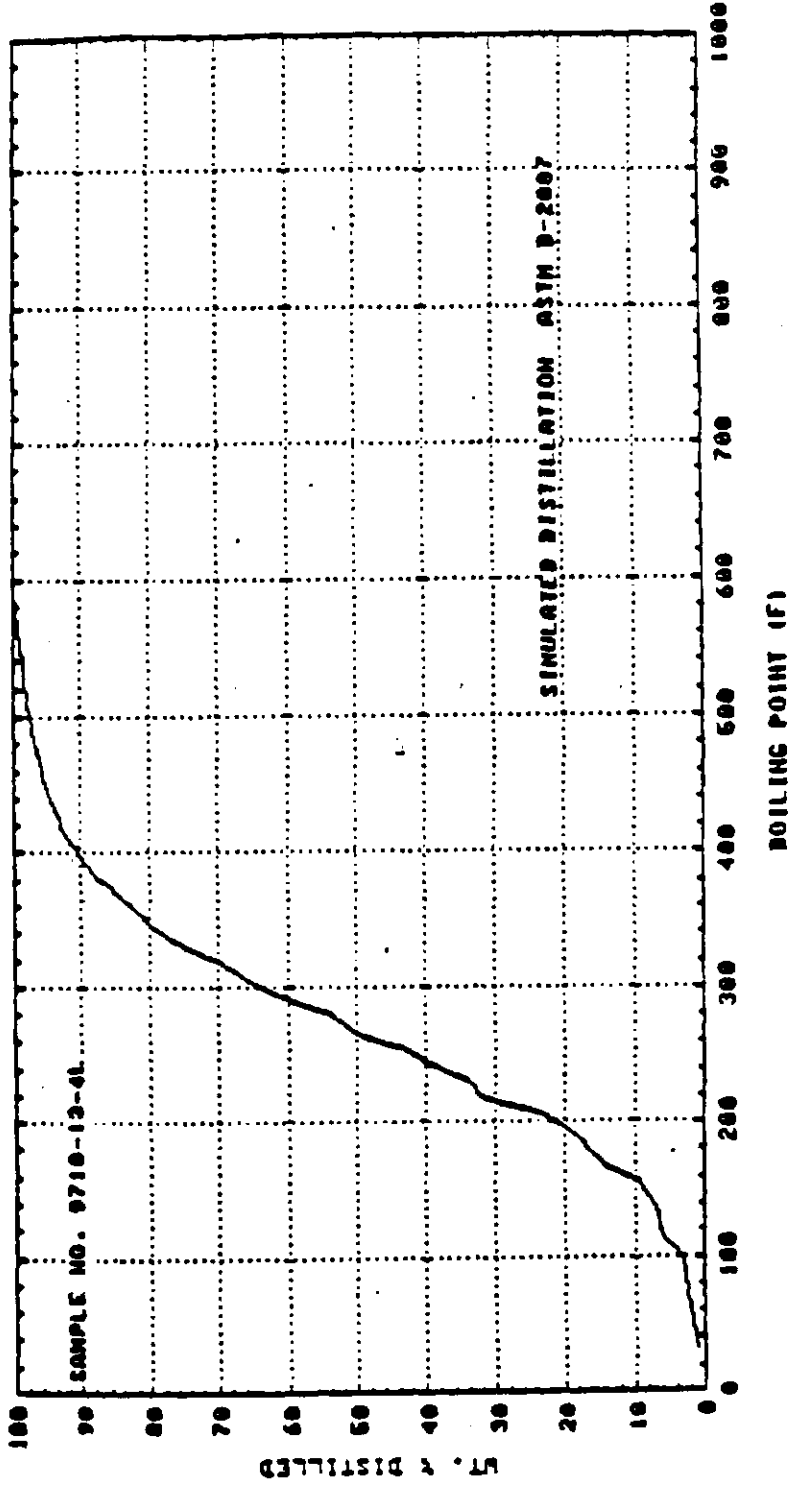


FIG. 31

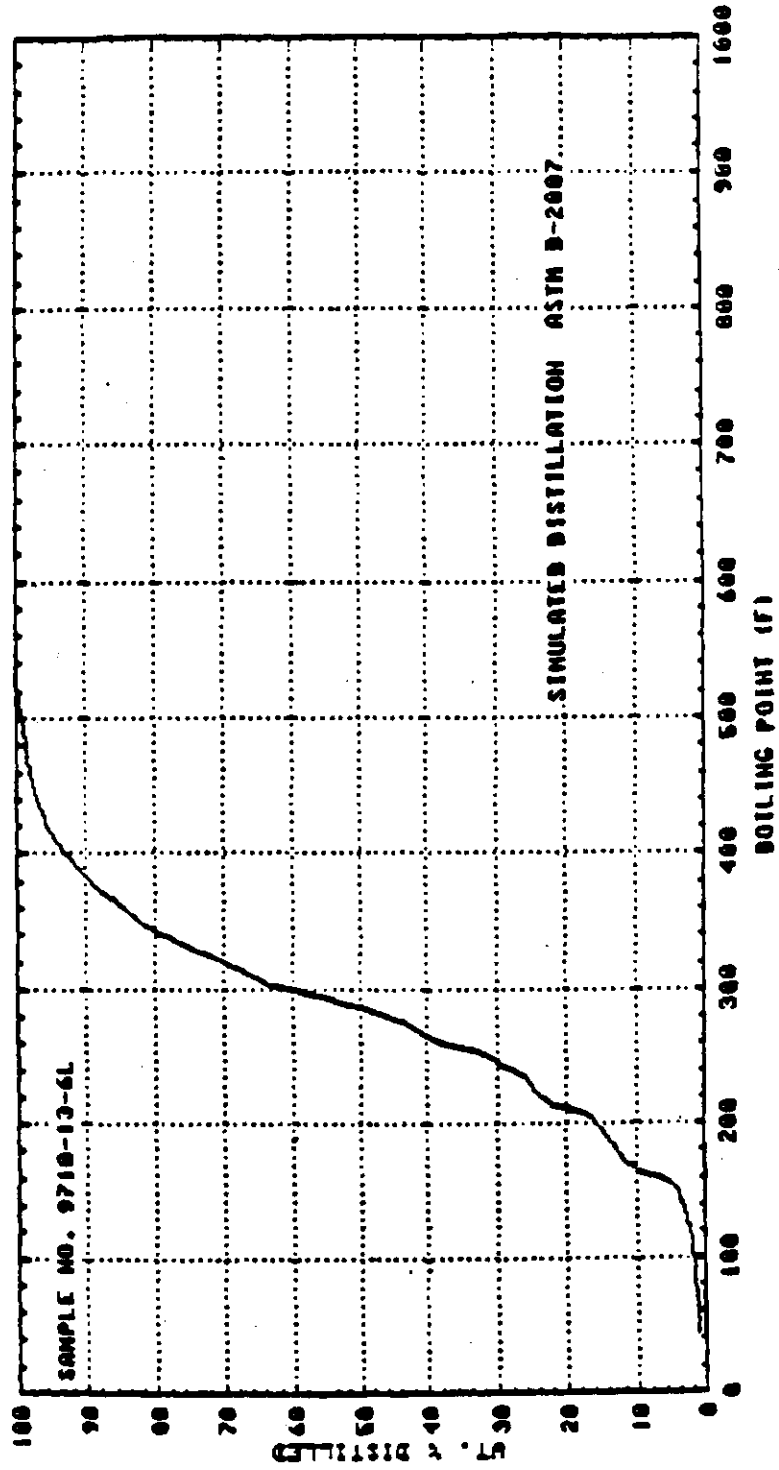


FIG. 32

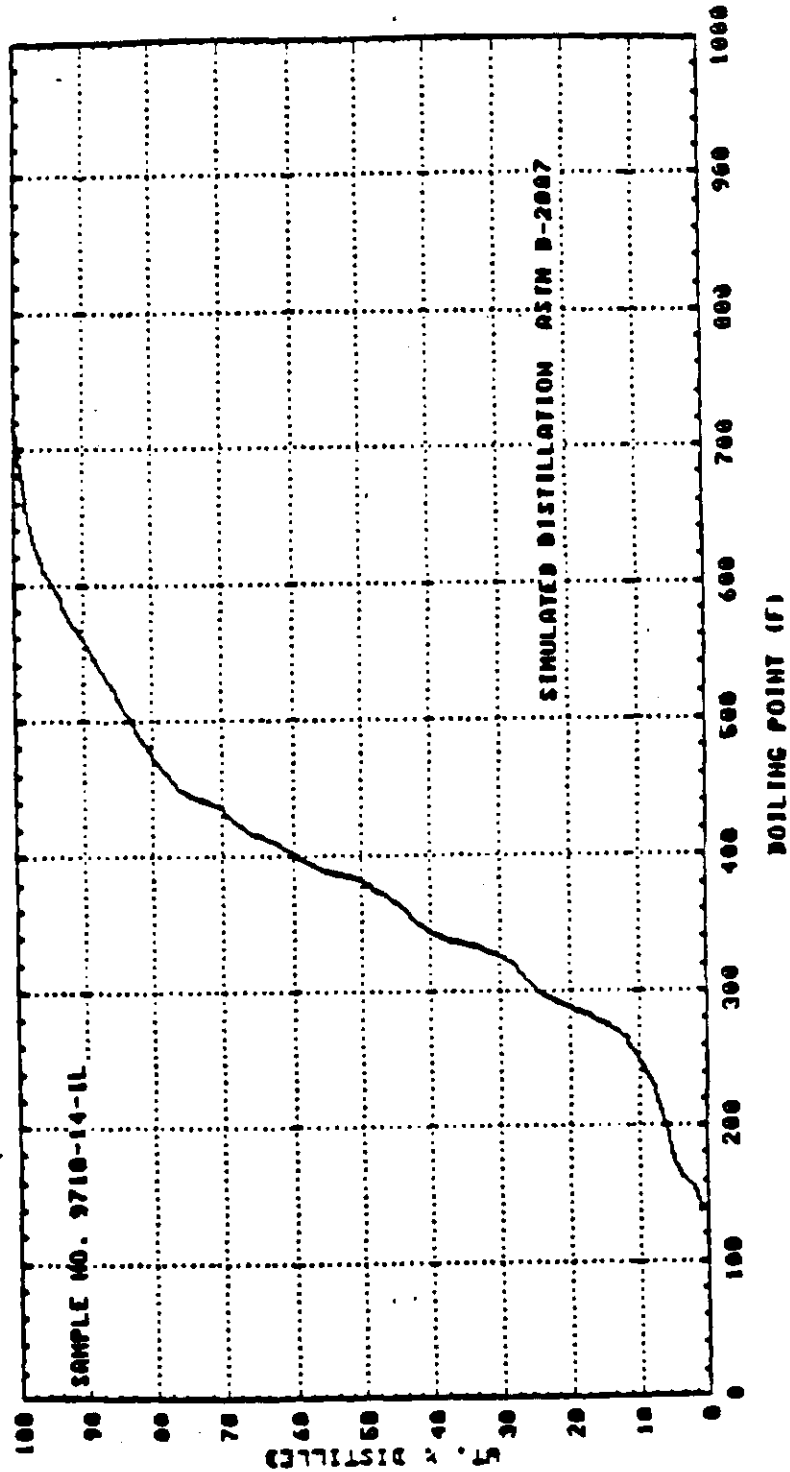
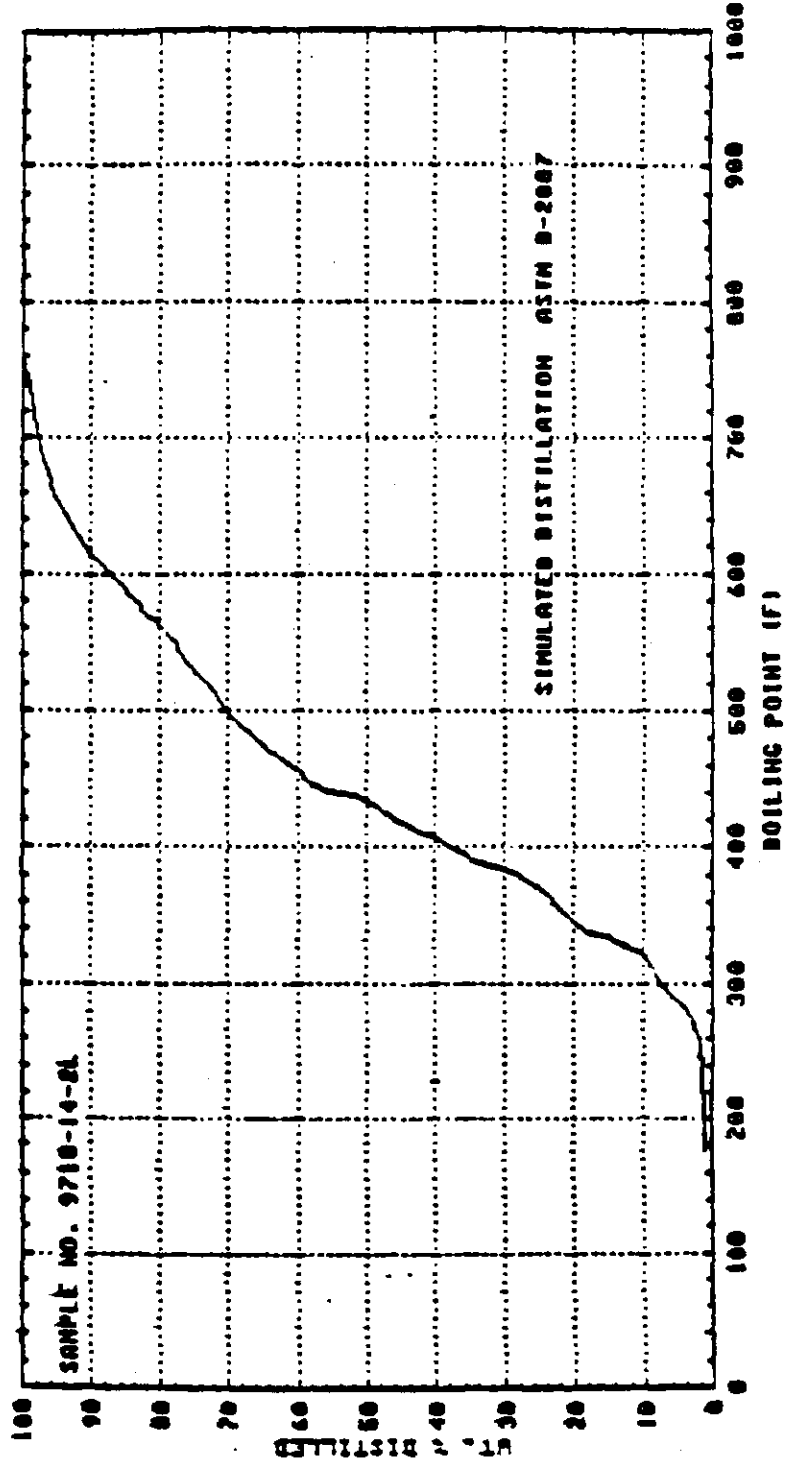


FIG. 33



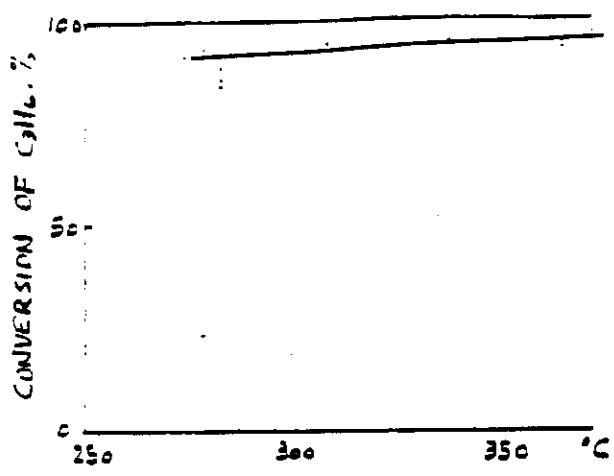
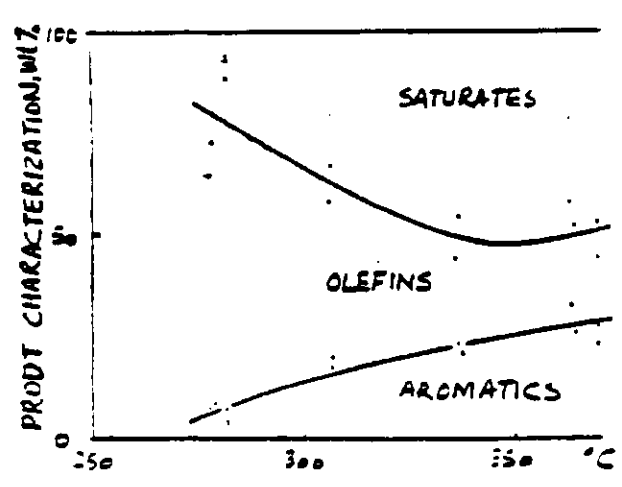
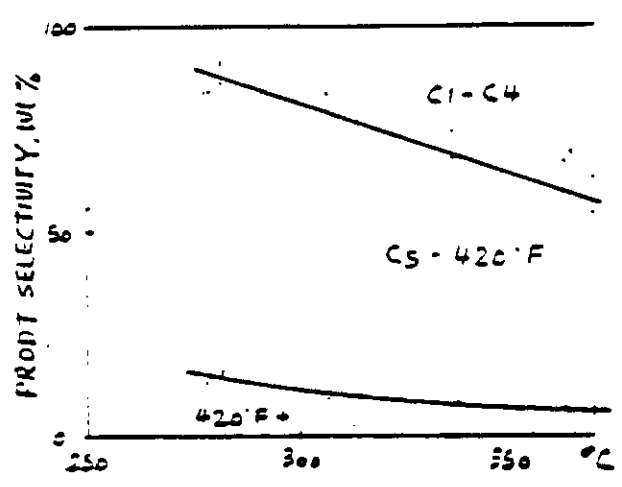


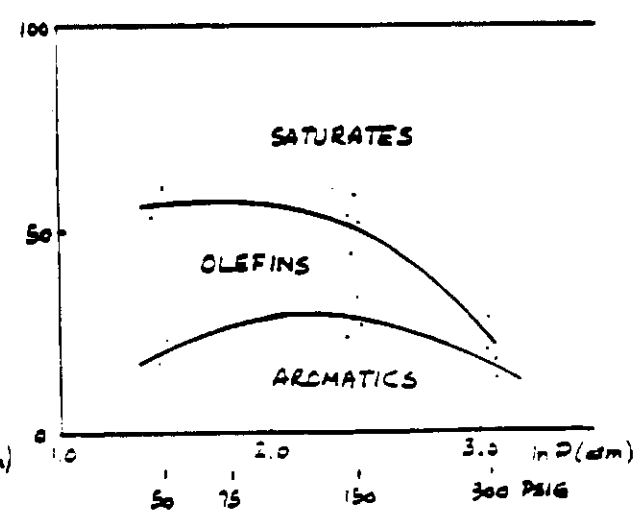
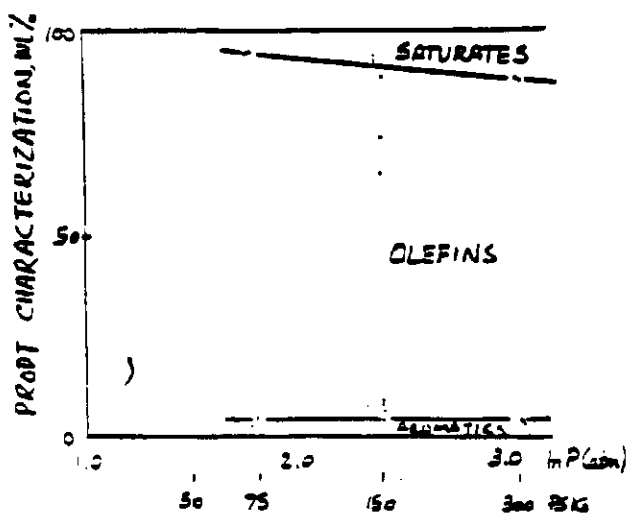
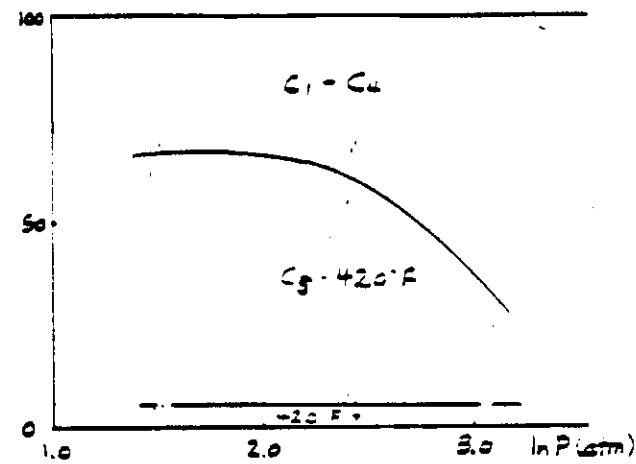
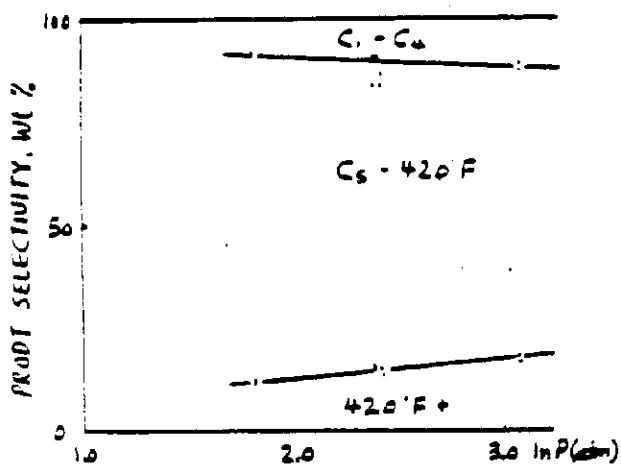
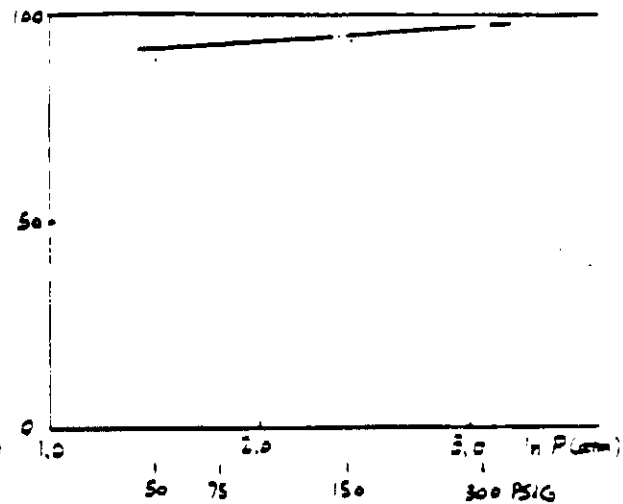
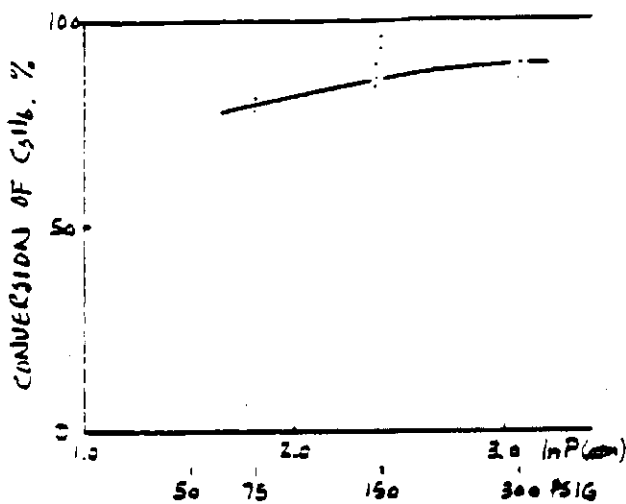
FIGURE 3+

EFFECTS OF TEMPERATURE
ON PROPYLENE OPERATION
WITH L2-105-6 AT 150 PSIG

RUN 9972-1, L2-105-6
H₂:C₃H₆ @ 1:1
C₃H₆ WHSV = 1.1
150 PSIG



6.35 EFFECT OF PRESSURE ON PROPYLENE OPERATION WITH LE-105-6
 AT 280 °C RUN 8972-1 AT 370 °C



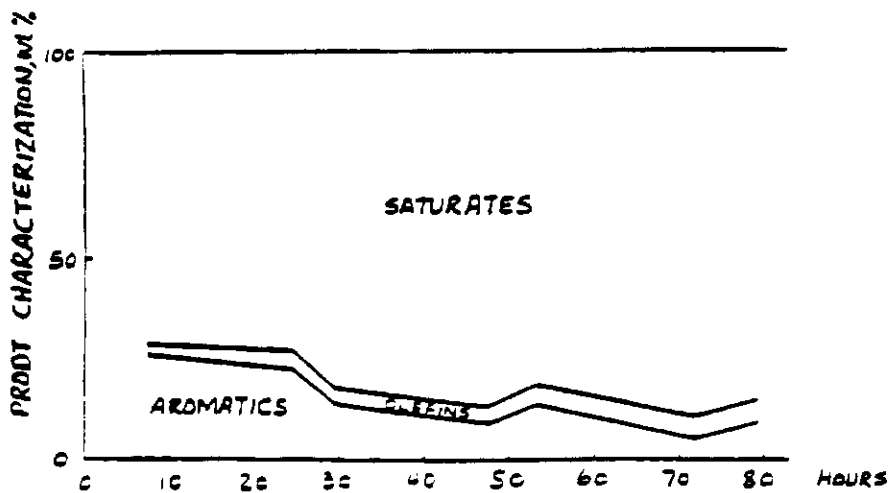
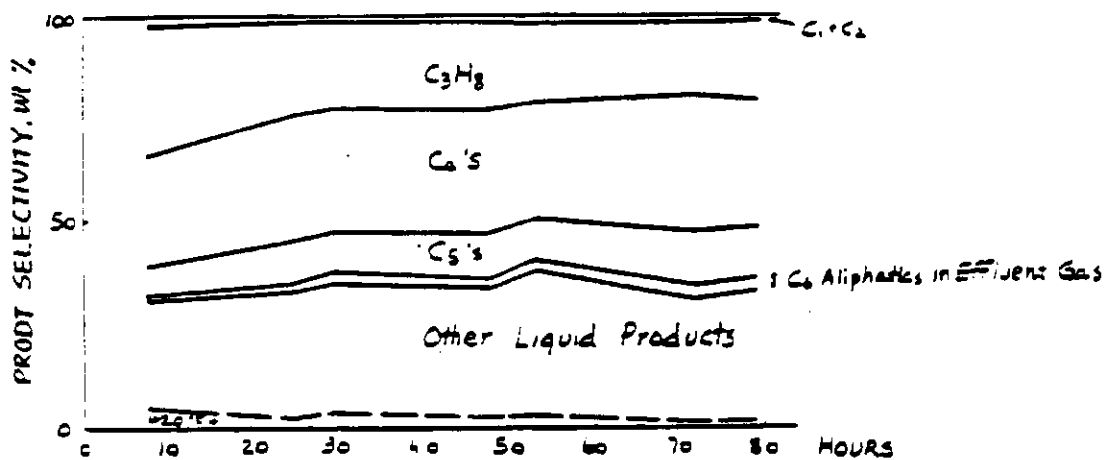
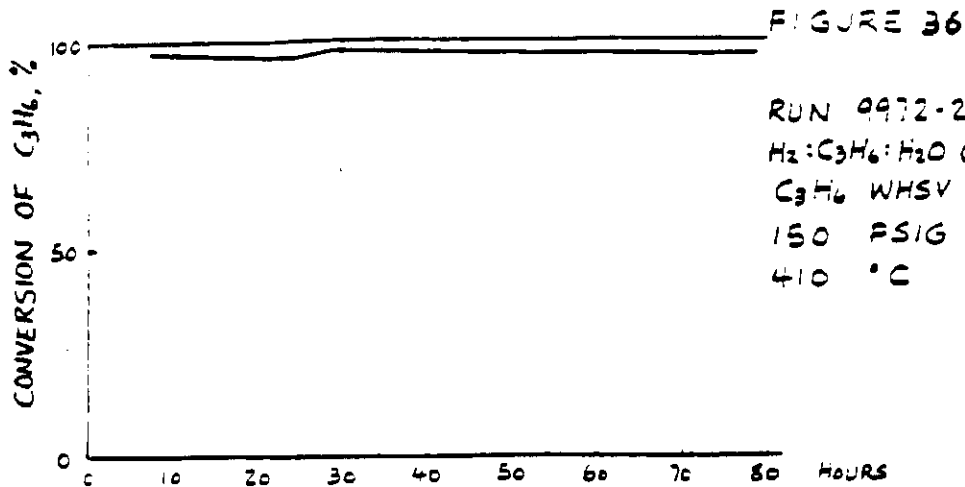


FIGURE 37

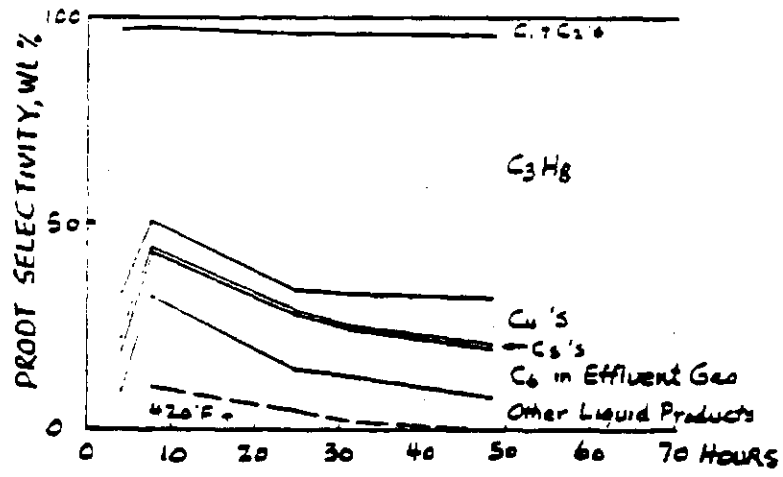
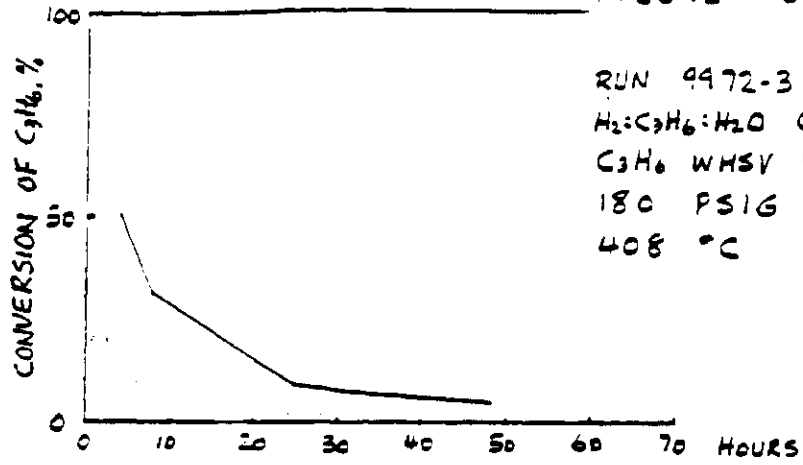


FIGURE 38

RUN 9972-4, UCC-101
 $H_2:C_3H_6:H_2O @ 1:1:3$
 C_3H_6 WHSV = 0.7
 150 PSIG
 280 & 340 °C

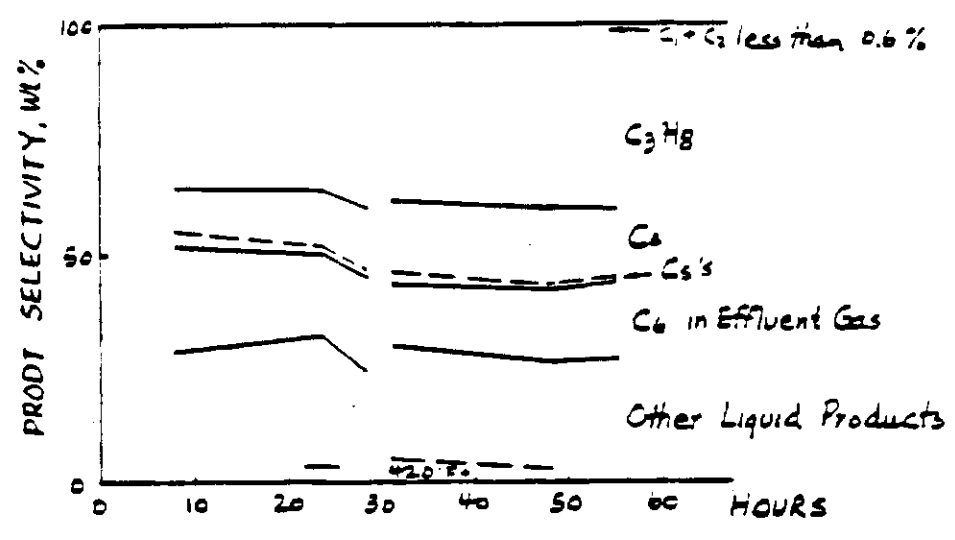
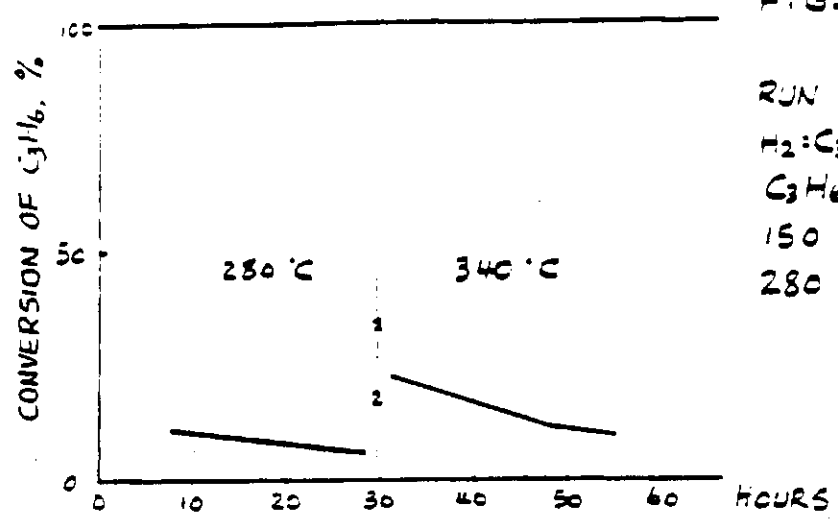


FIGURE 39

RUN 9972-3, -4
UCC-101
 $H_2:C_3H_6:H_2O @ 1:1:2$
 $1:1:3$
150-180 PSIG
INITIAL CONVERSIONS

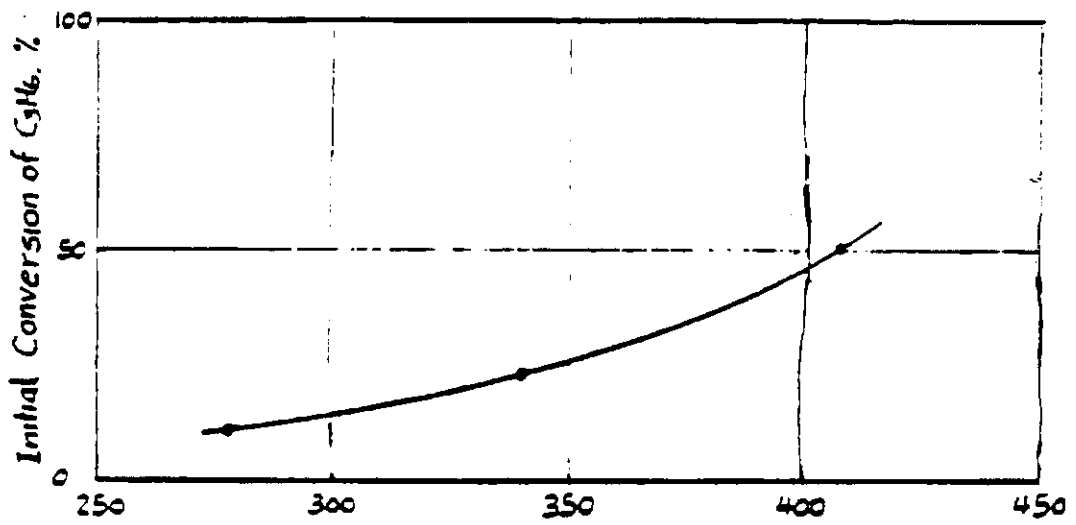


FIGURE 40

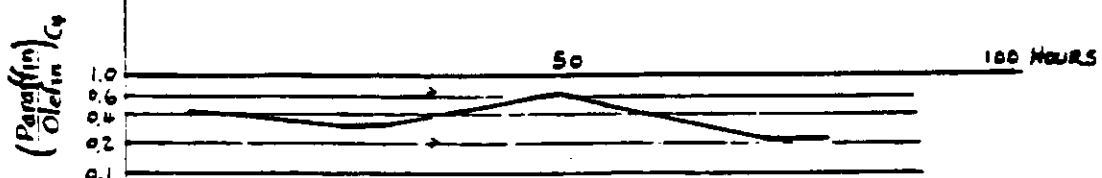
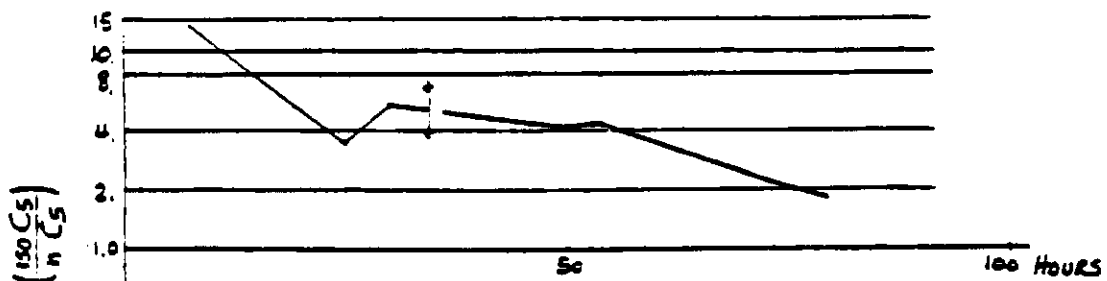
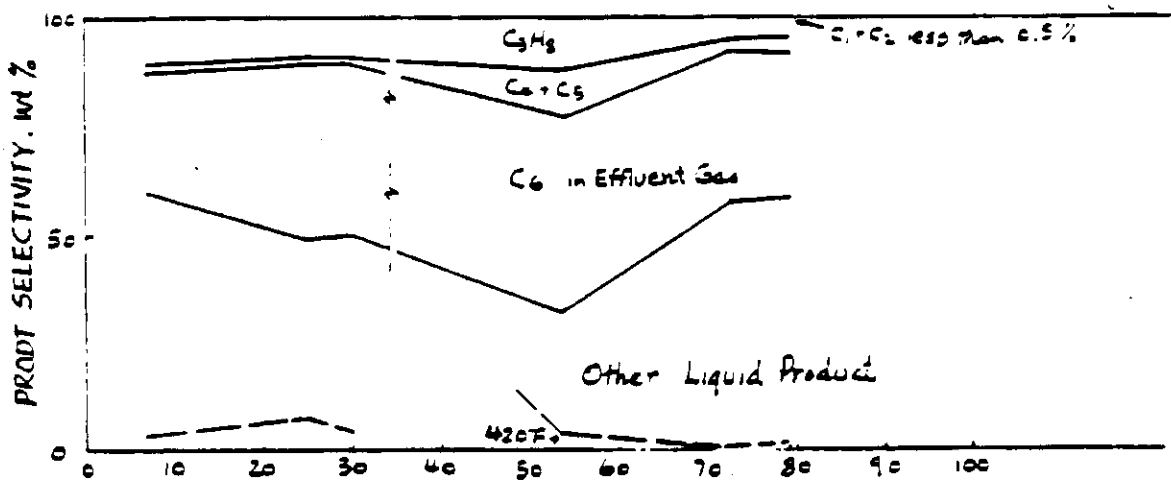
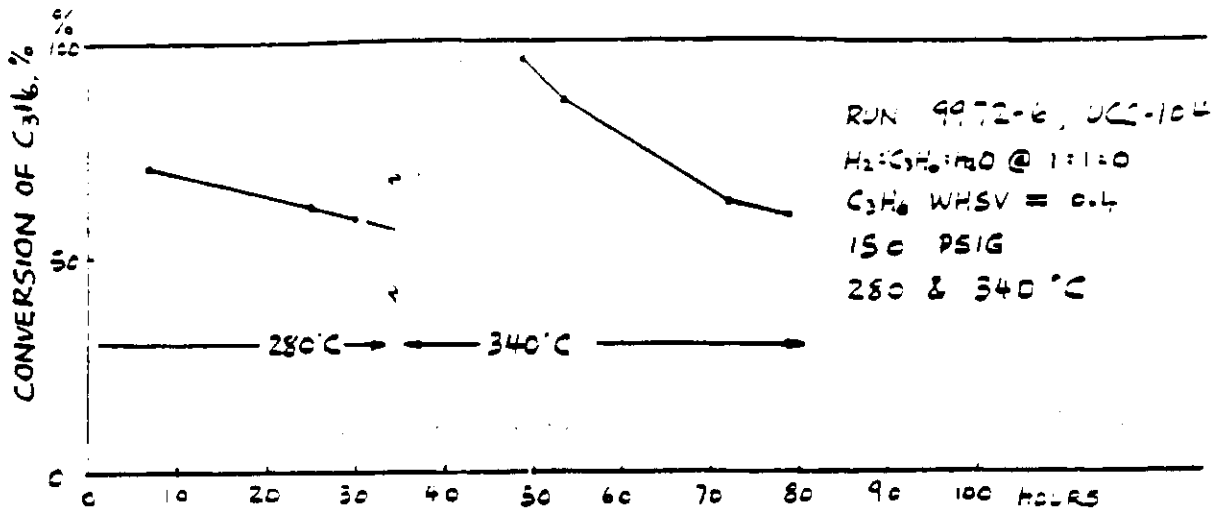


FIGURE 41

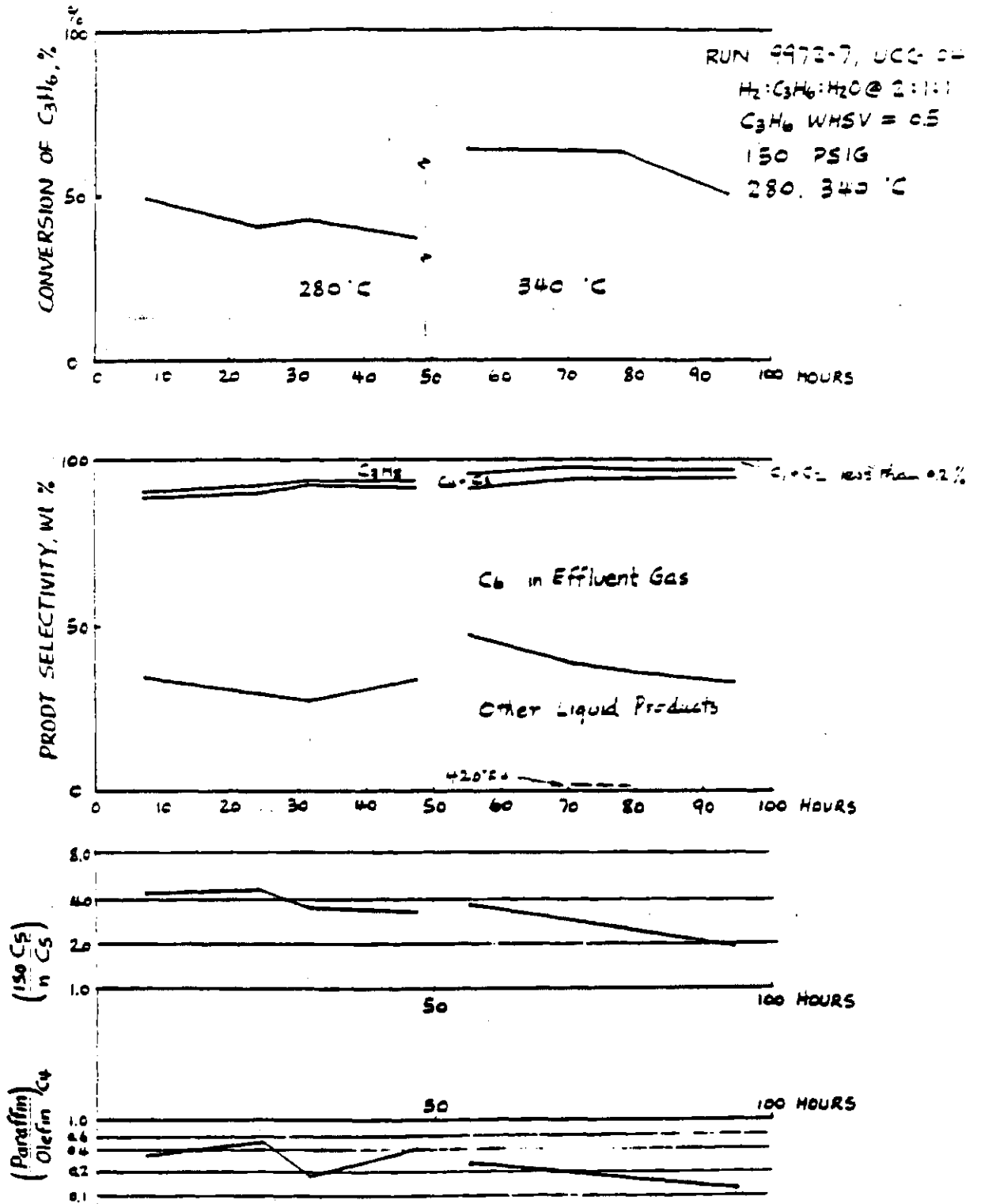
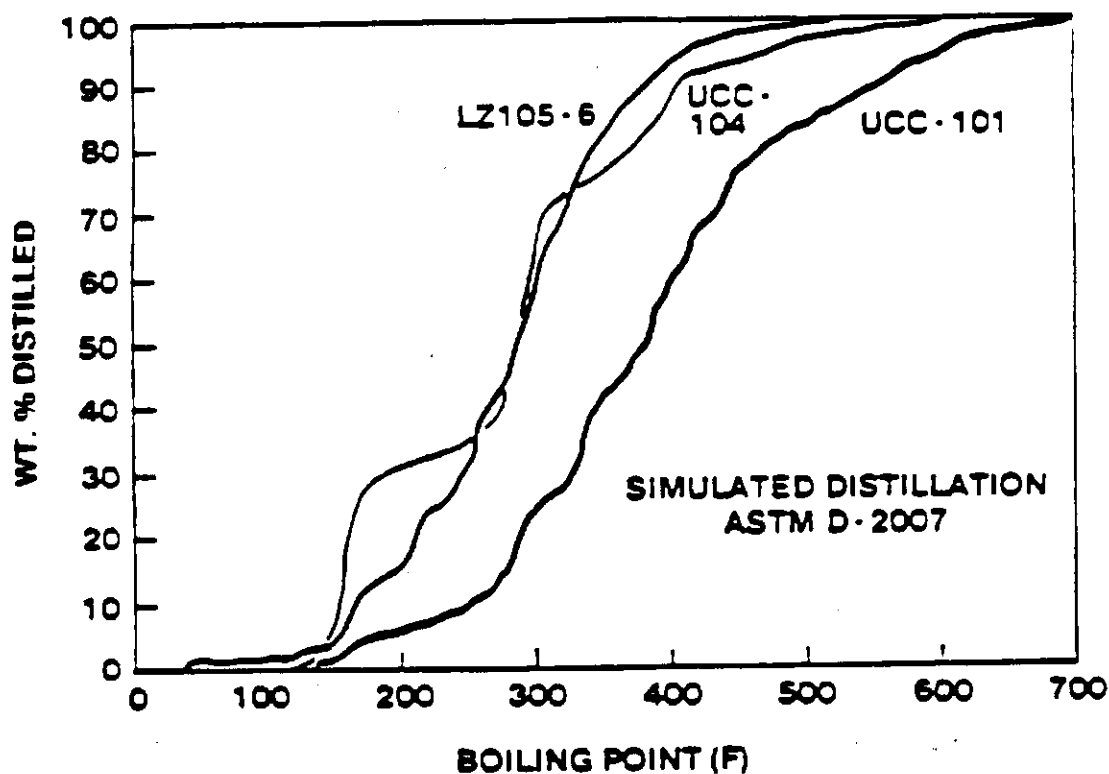


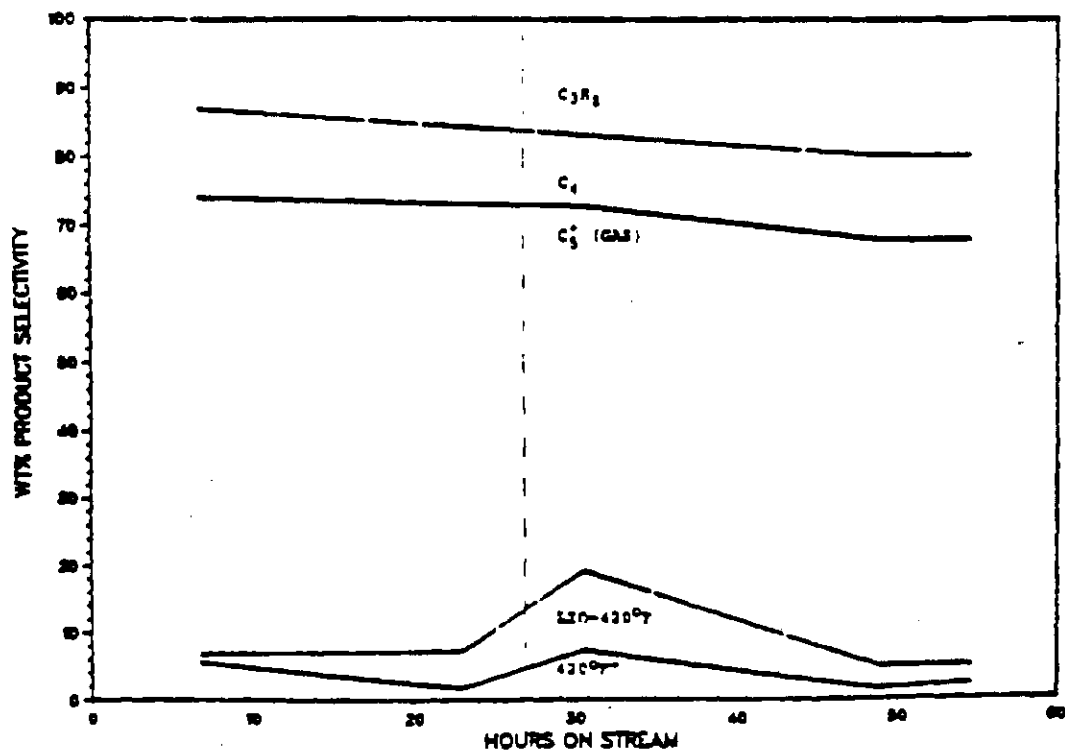
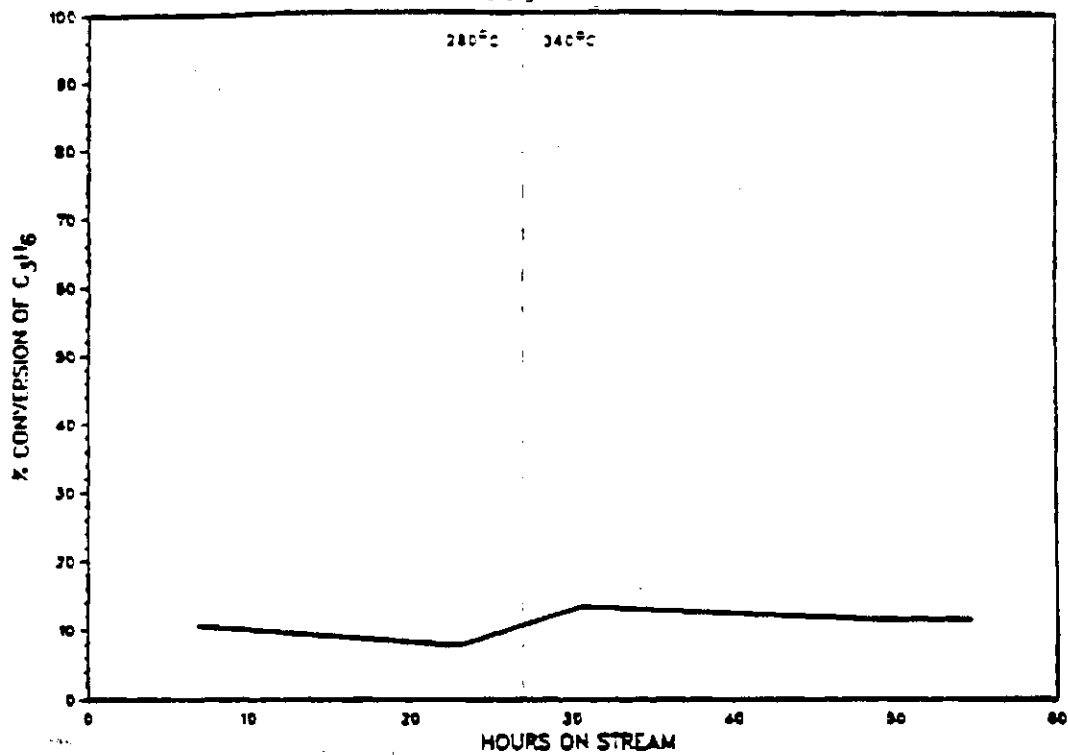
FIGURE 42

**BOILING POINT DISTRIBUTION OF
THE LIQUID PRODUCT OF PROPYLENE
OLIGOMERIZATION USING LZ105-6,
UCC 101, AND UCC 104 AS CATALYSTS.**



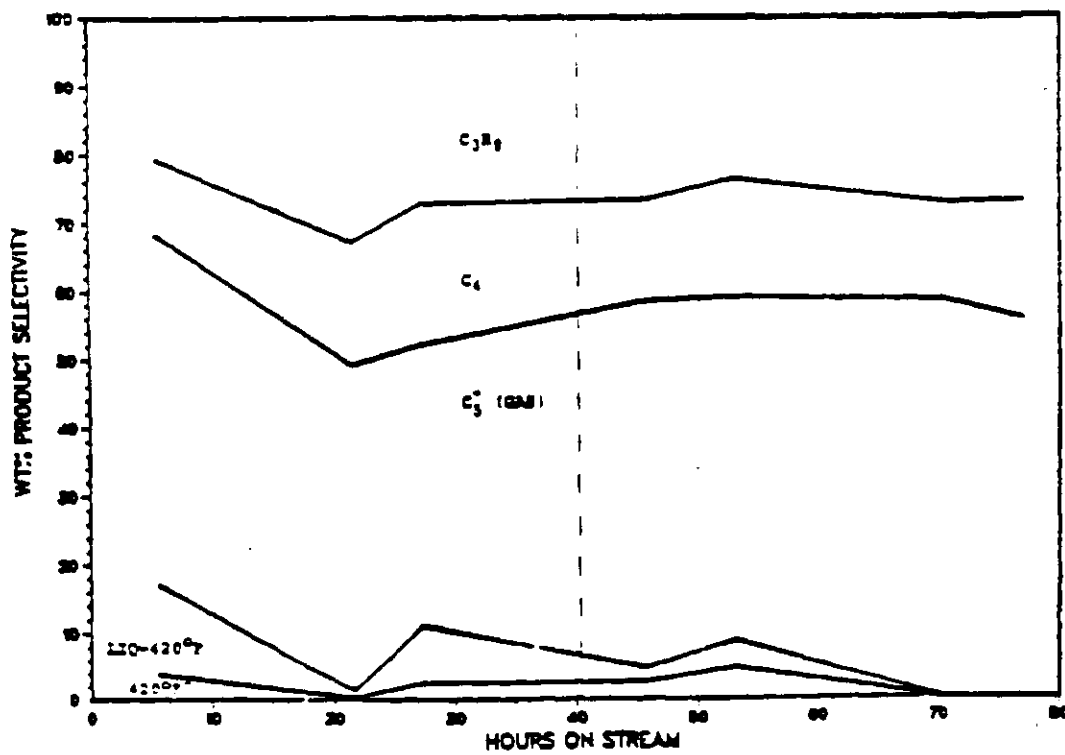
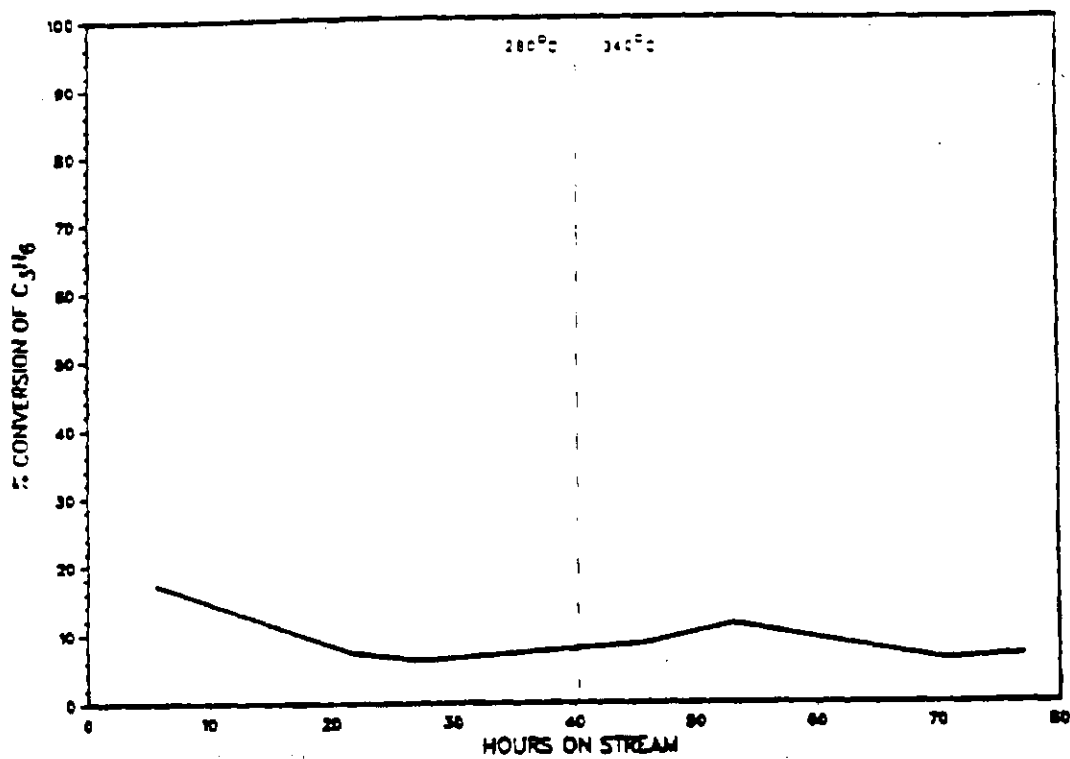
RUN NO. 9972-08

Fig. 43



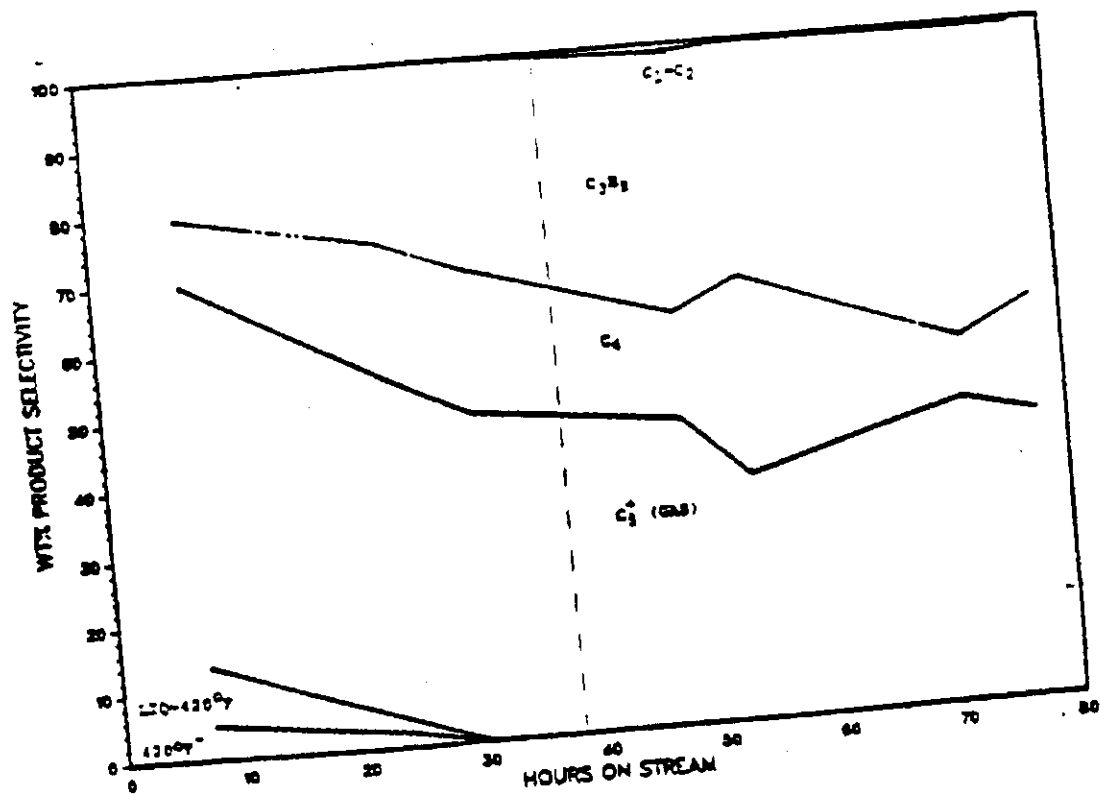
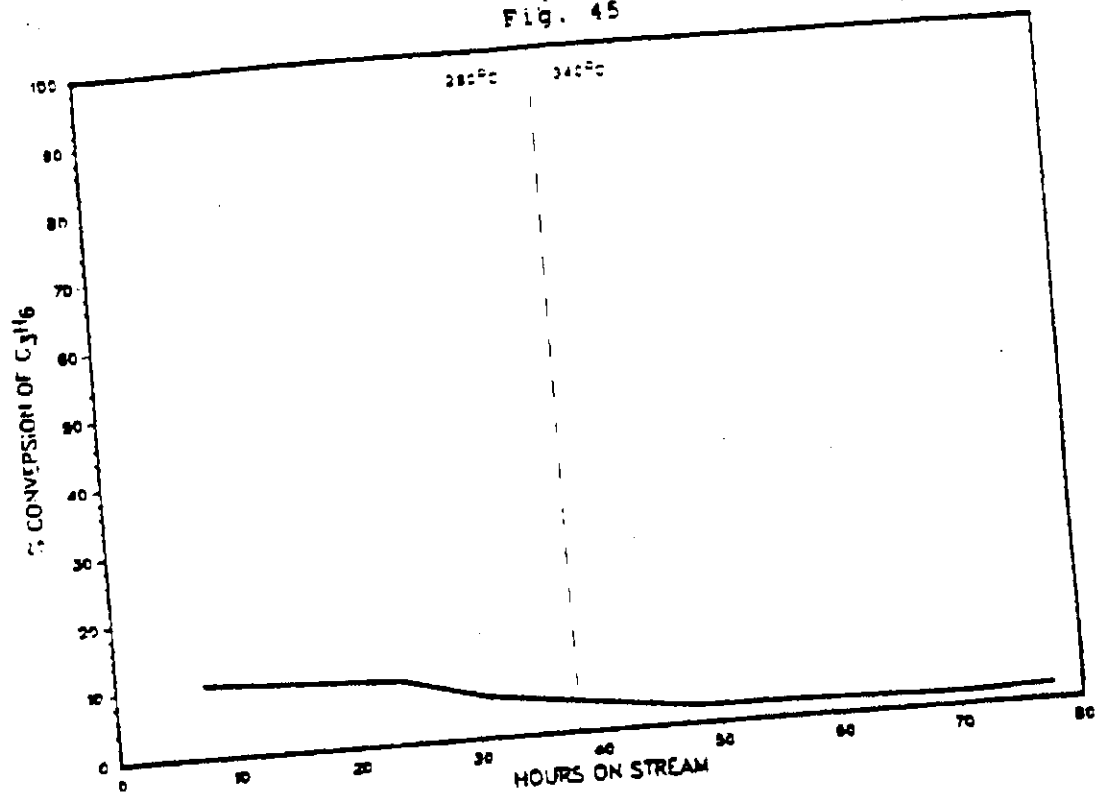
RUN NO. 9972-09

Fig. 44



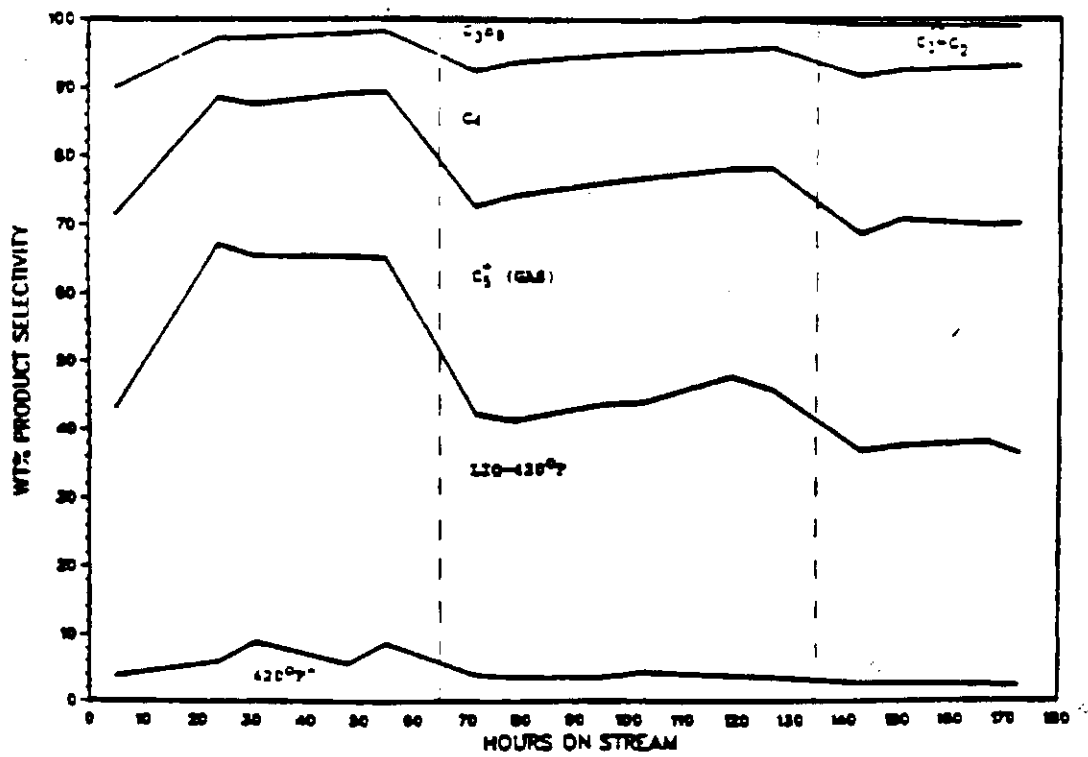
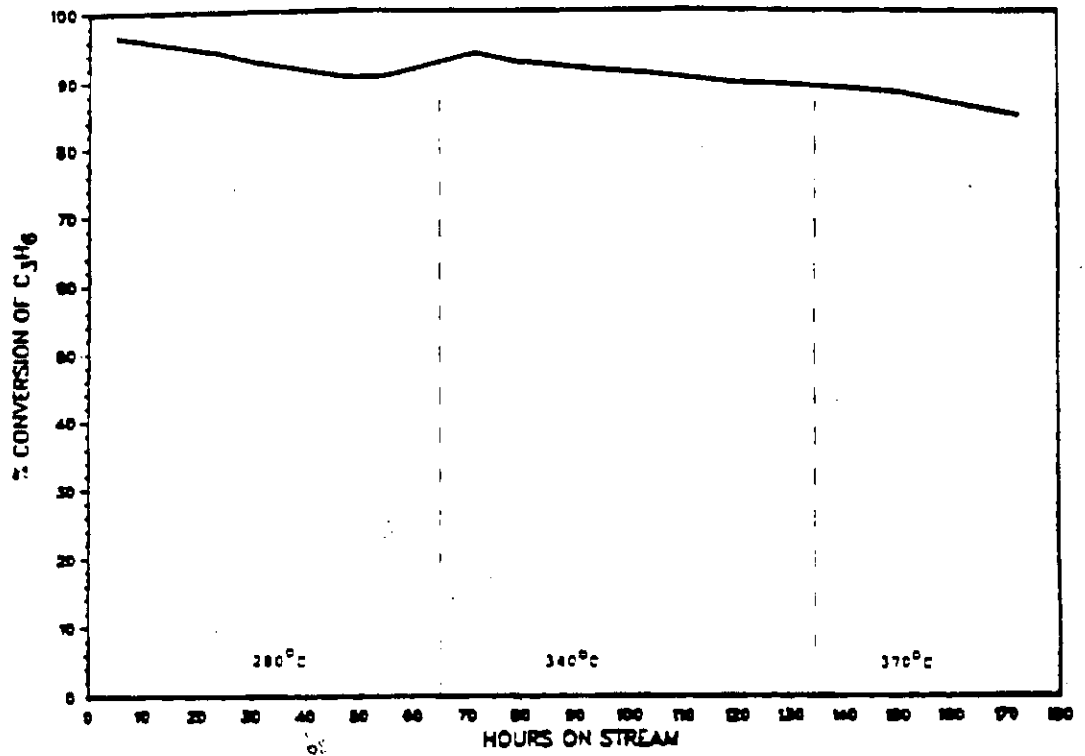
RUN NO. 9972-10

Fig. 45

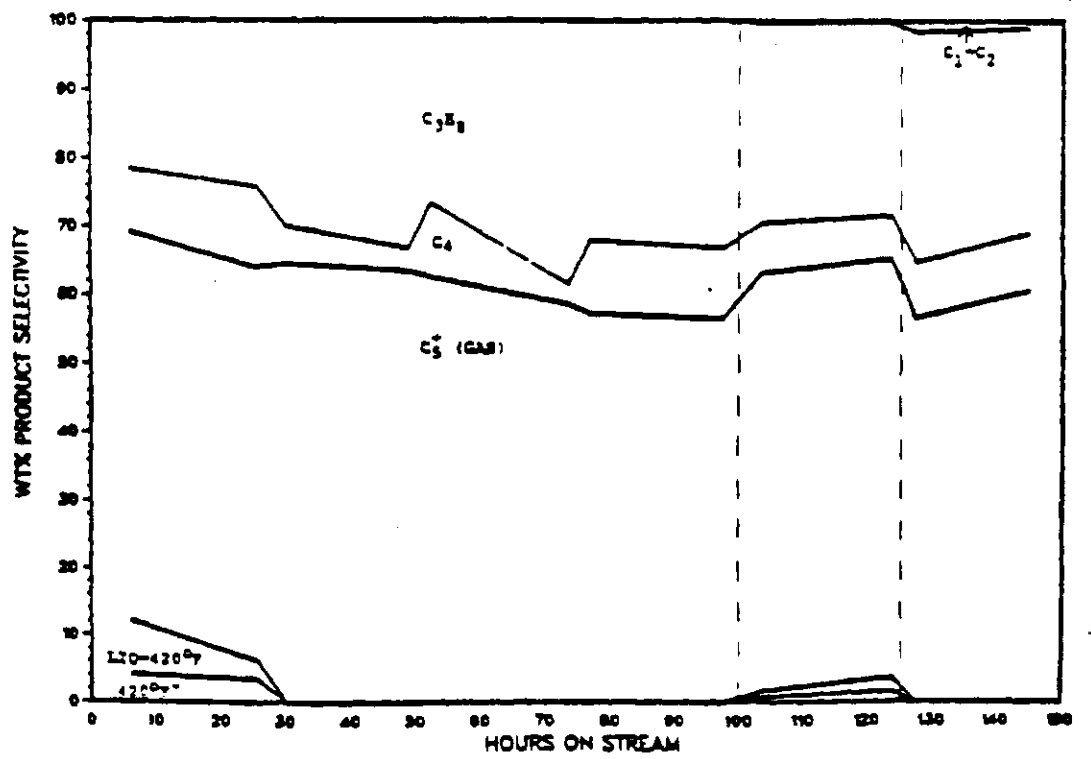
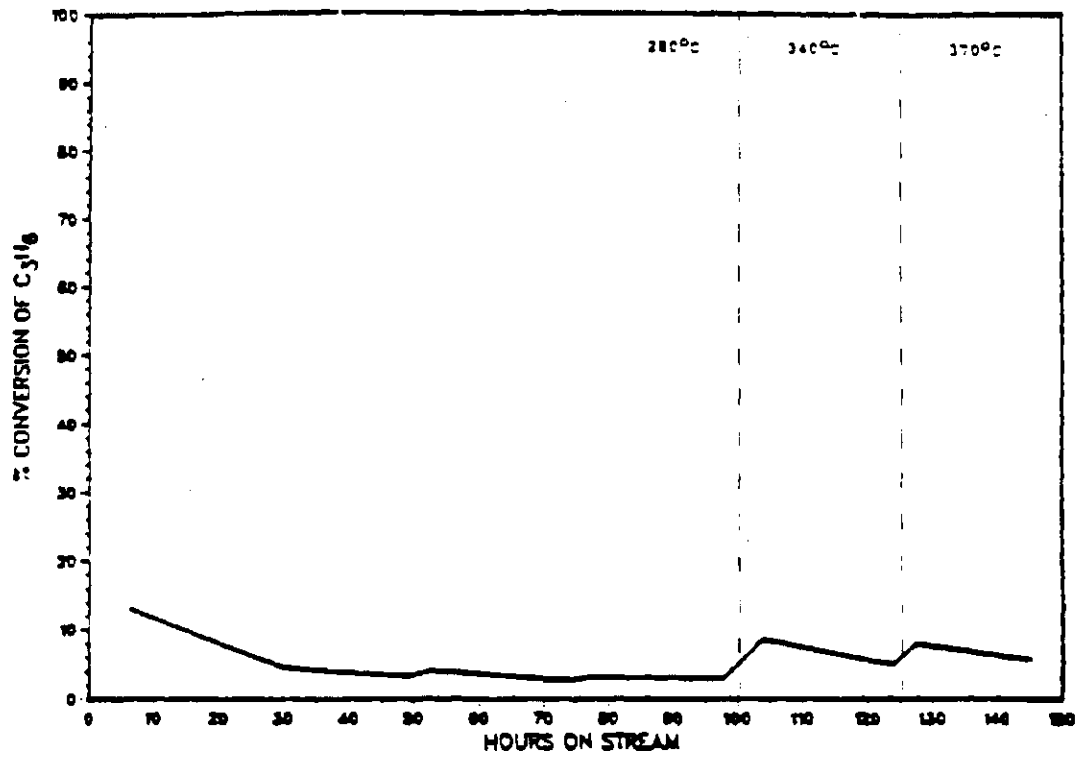


RUN NO. 9972-13

Fig. 46

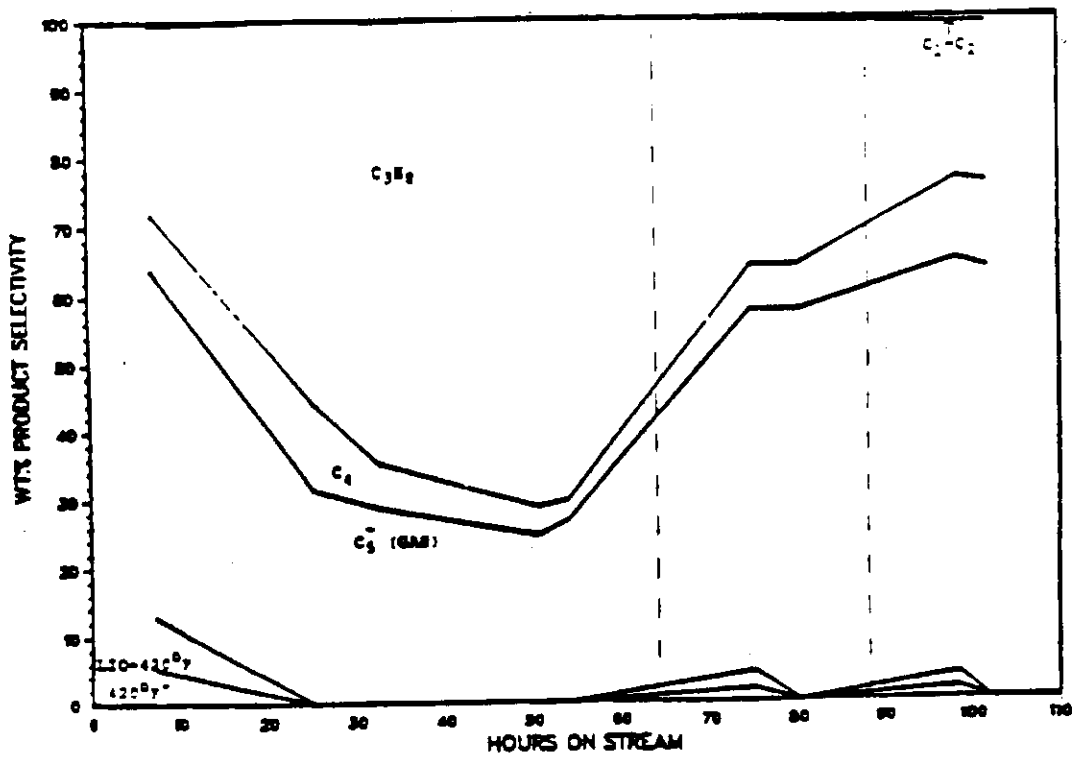
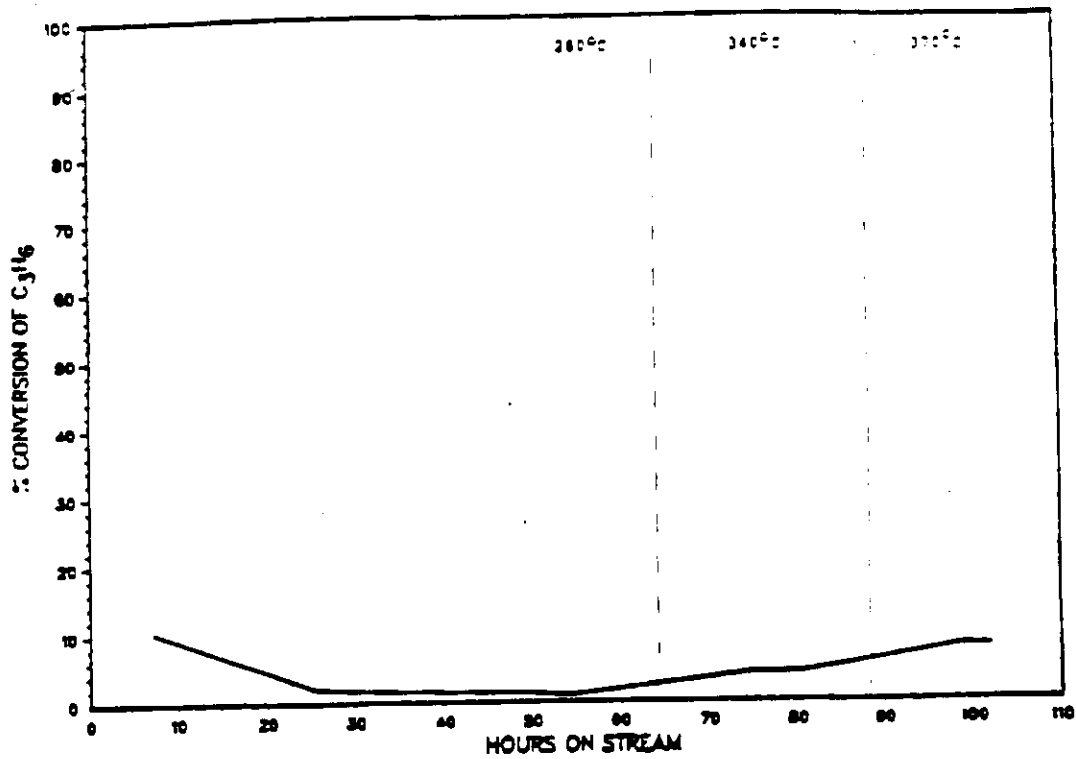


RUN NO. 9972-14
Fig. 47



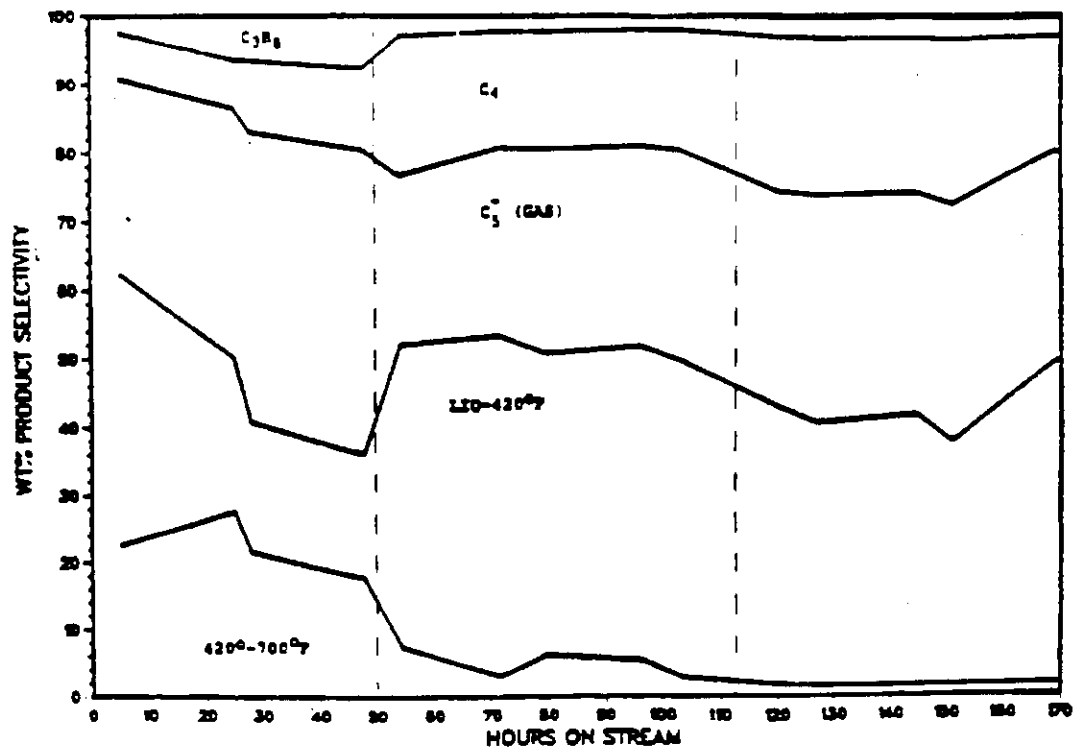
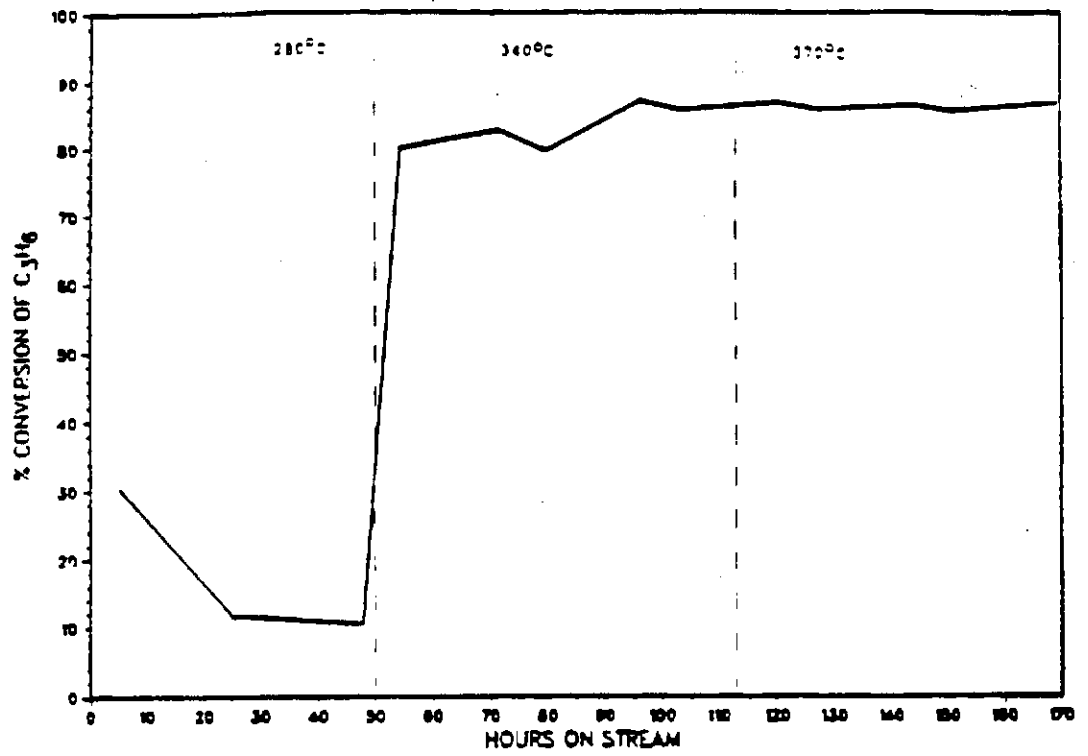
RUN NO. 9972-15

Fig. 48



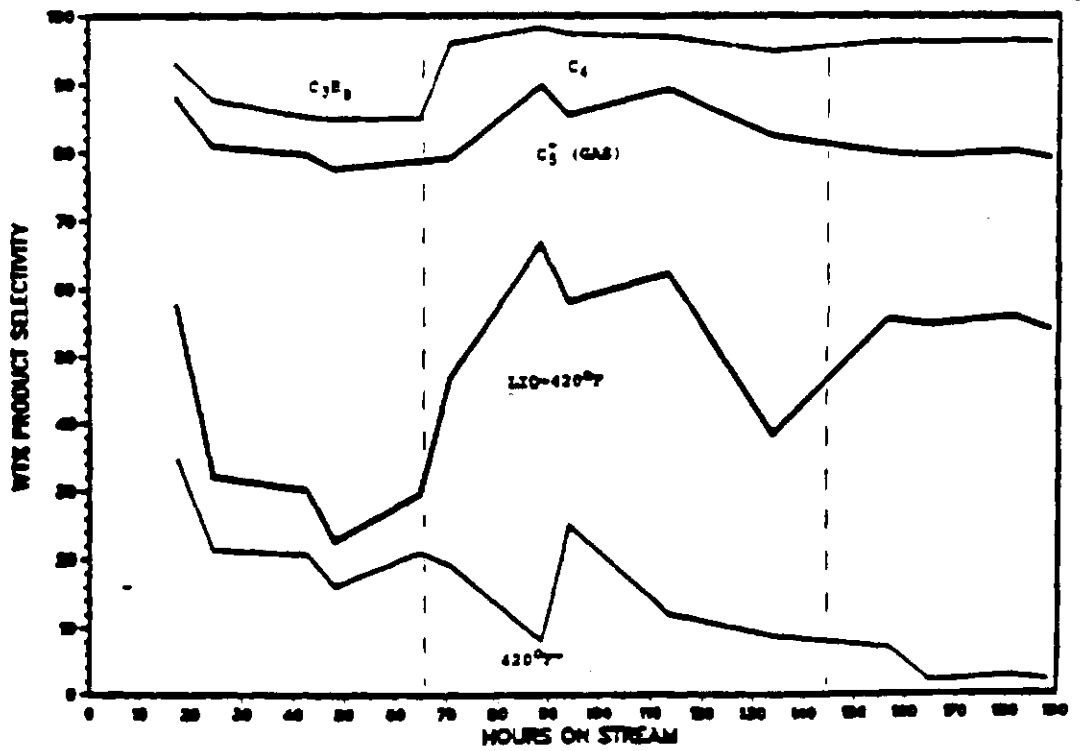
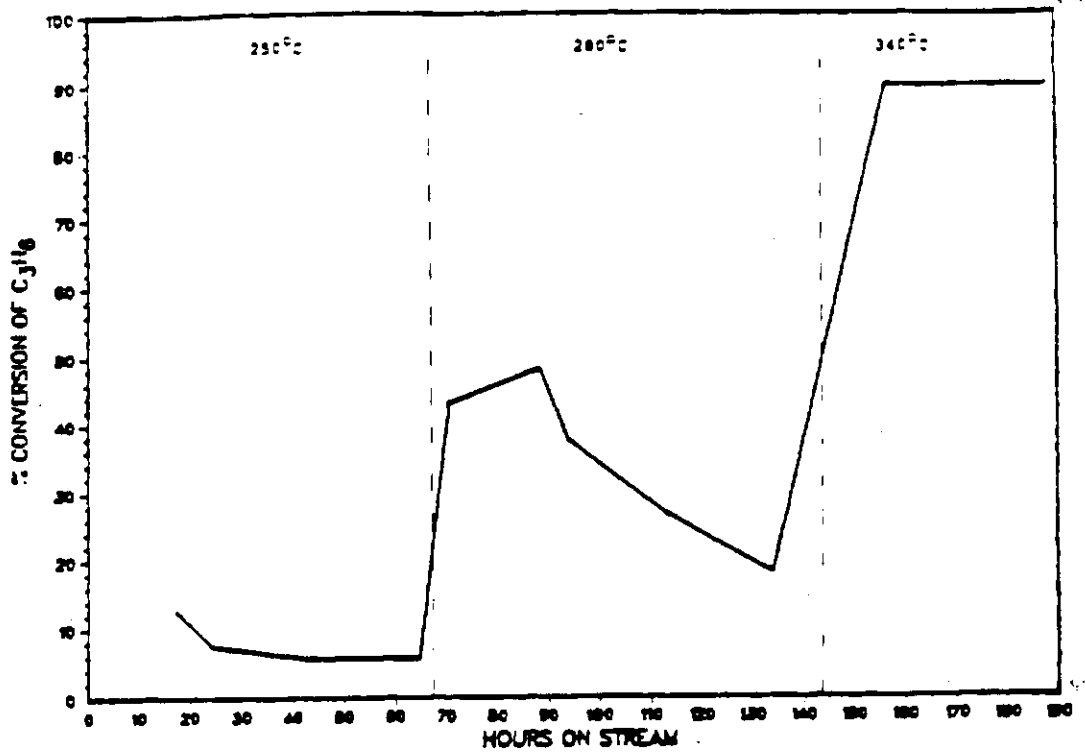
RUN NO. 9972-16

Fig. 49



RUN NO. 9972-17

Fig. 50



997218

0.60 WHSV
150 PSIG
280°C

340°C

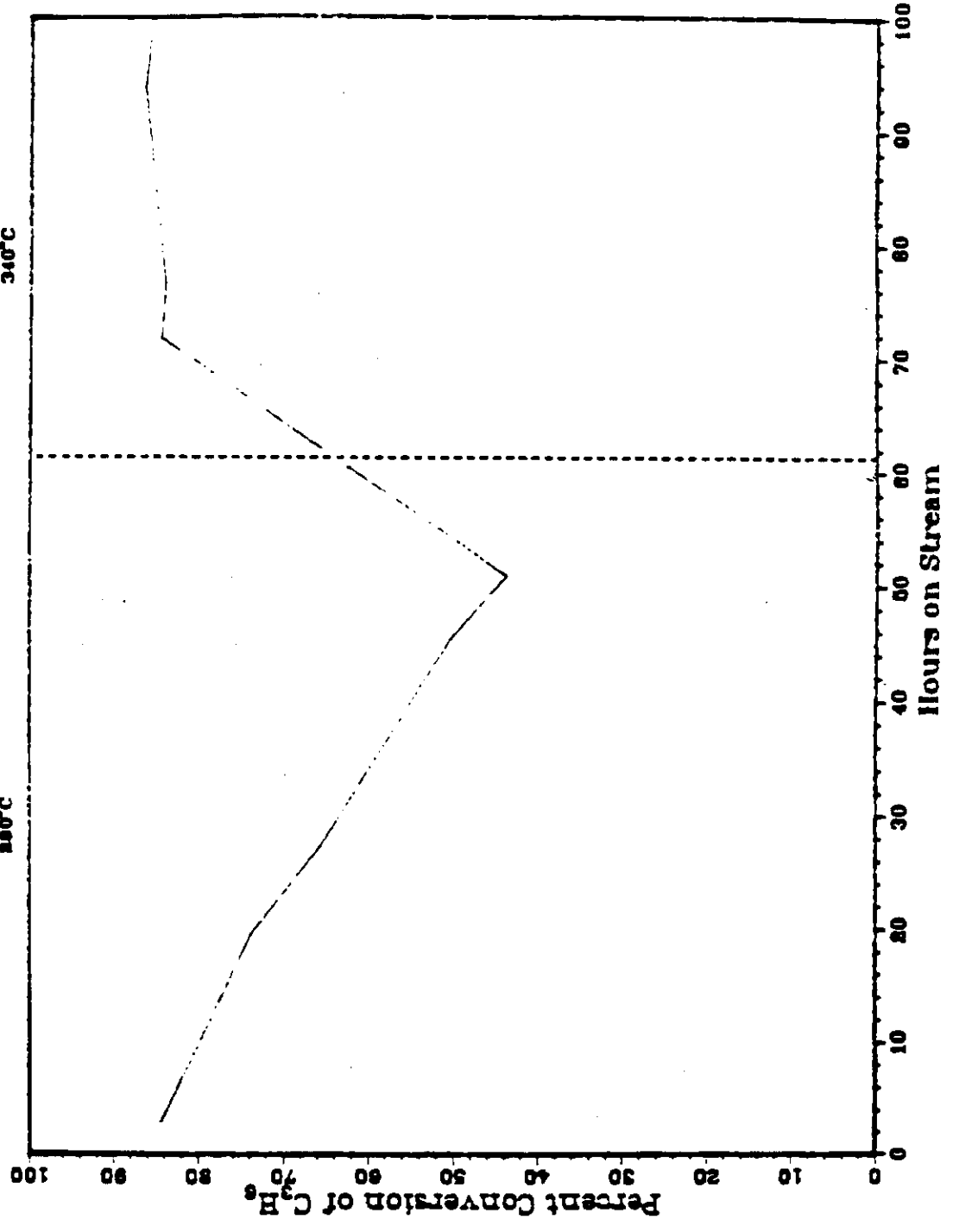


Fig. 51

Fig. 52

UNION

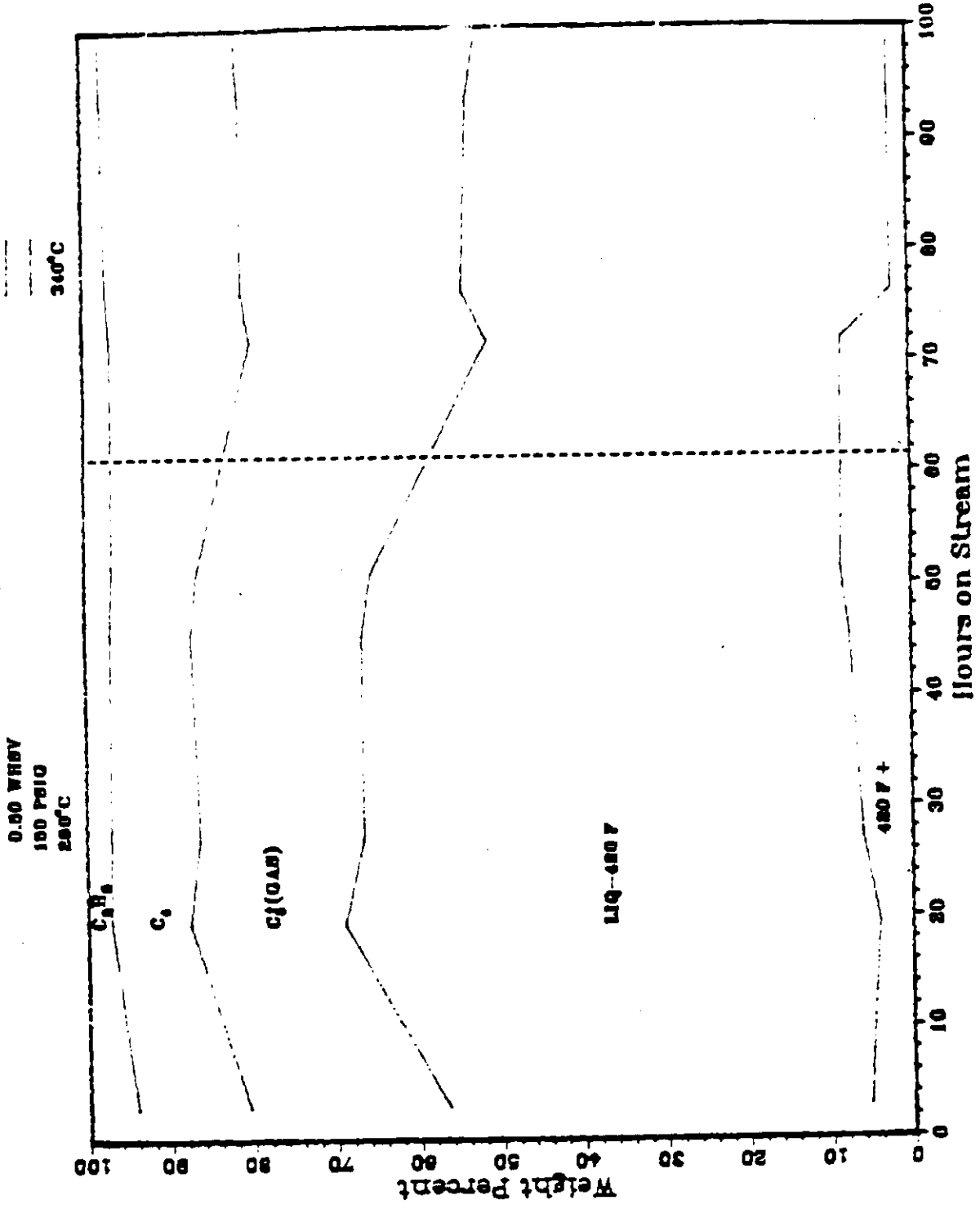
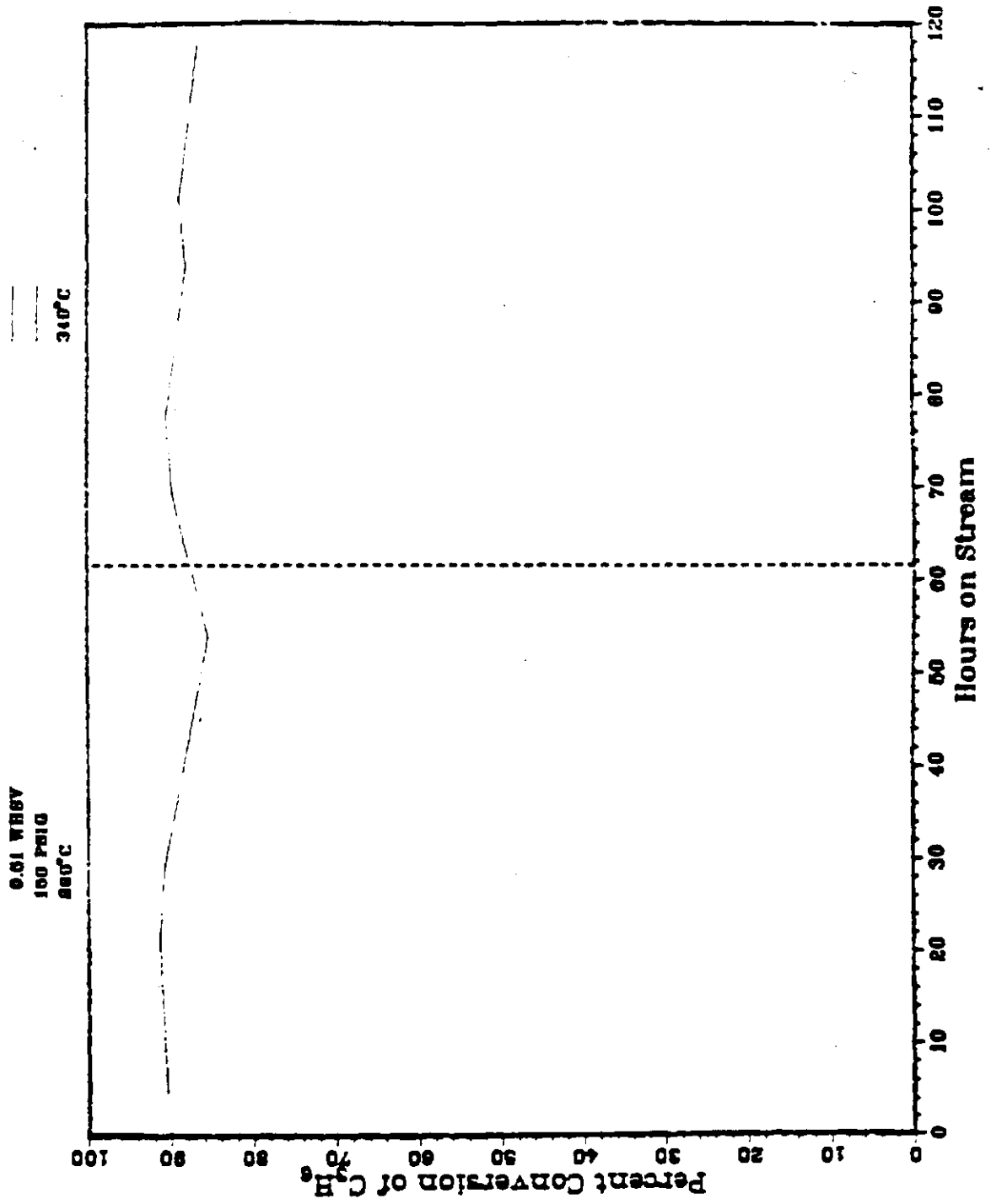


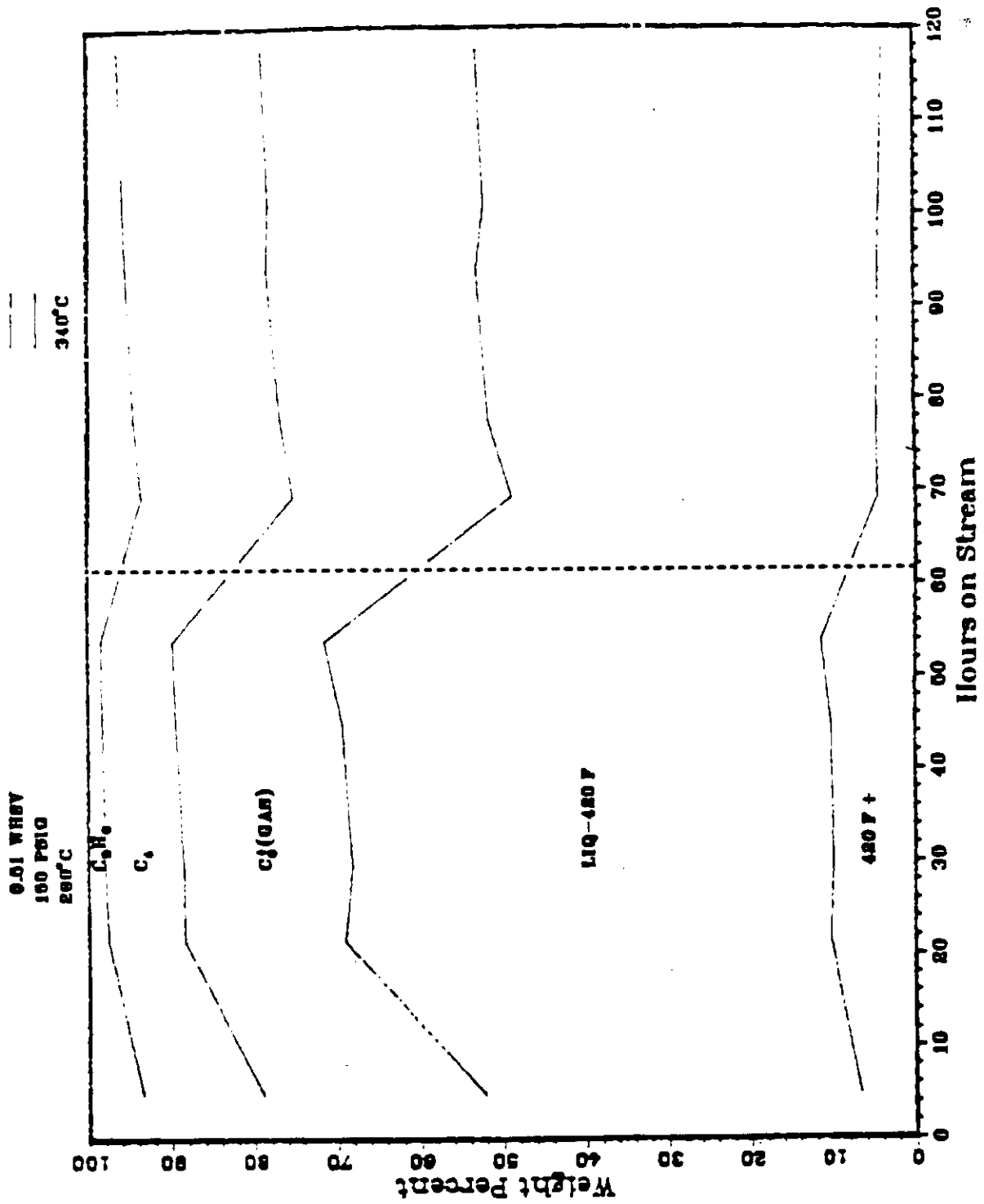
Fig. 53

997220



997220

Fig. 54



997219

0.01 WUBV
100 PBD
300°C

300°C

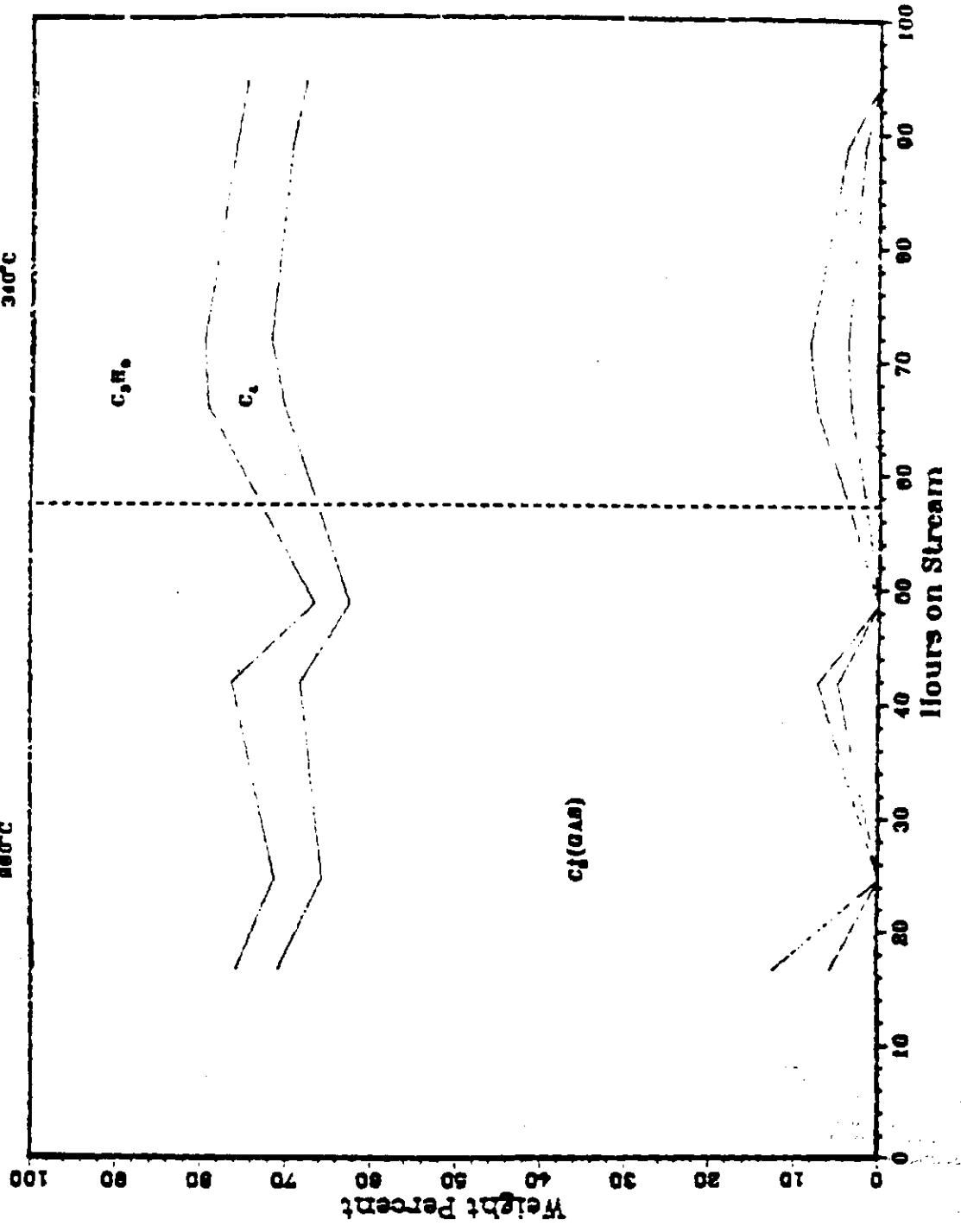


Fig. 55

Fig. 56

997233

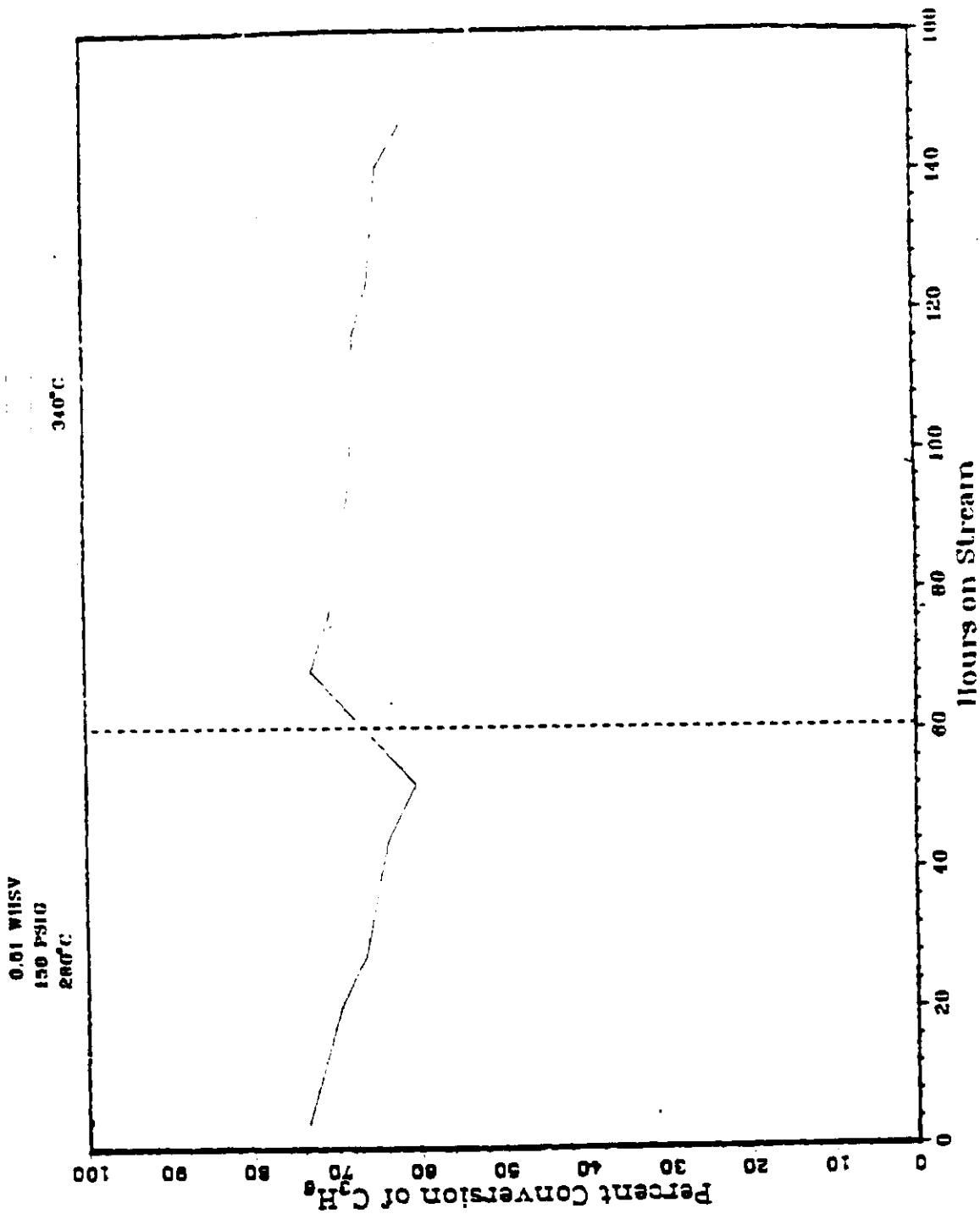
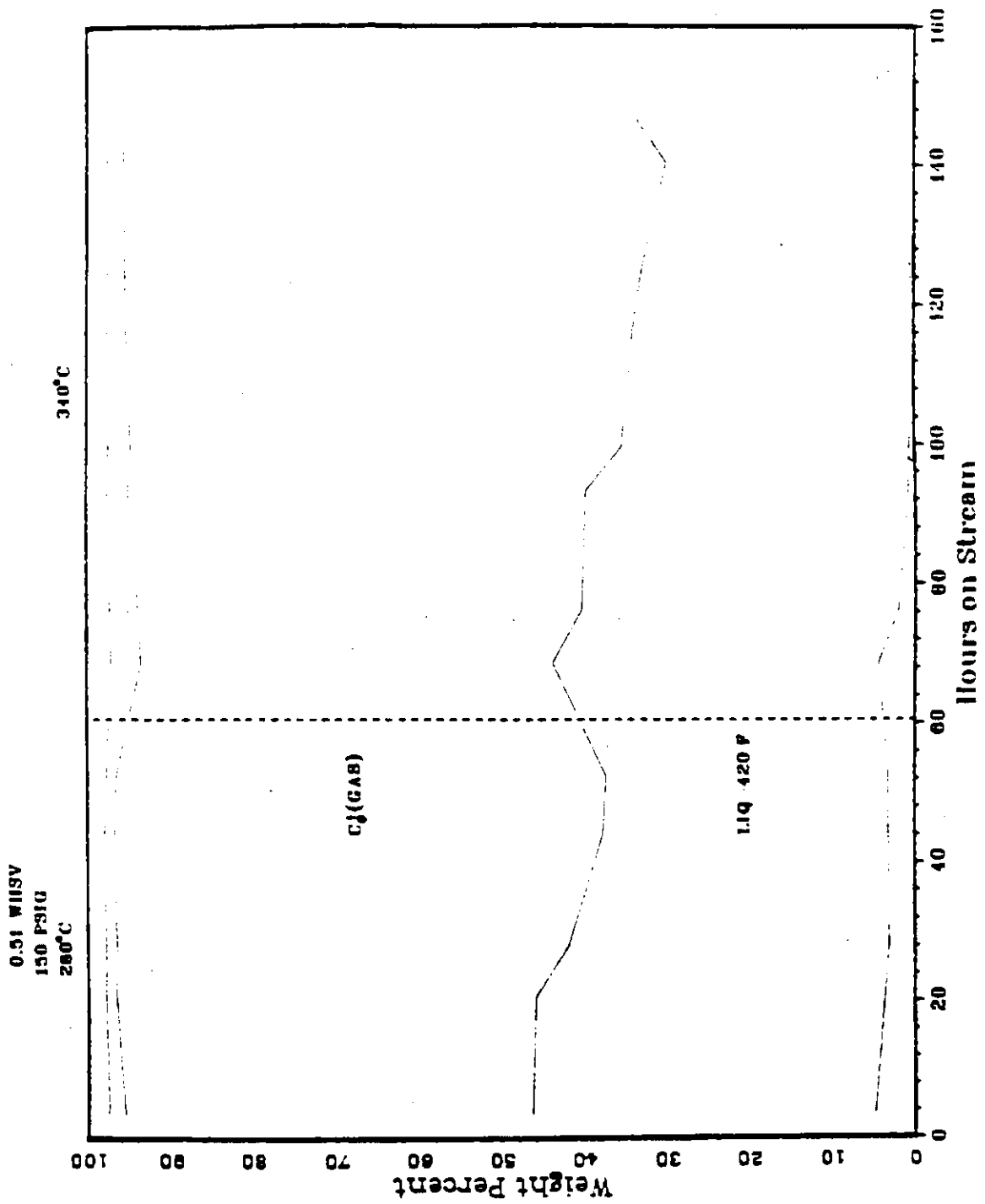


Fig. 57

097222



RUN NO. 10011-06
Fig. 58

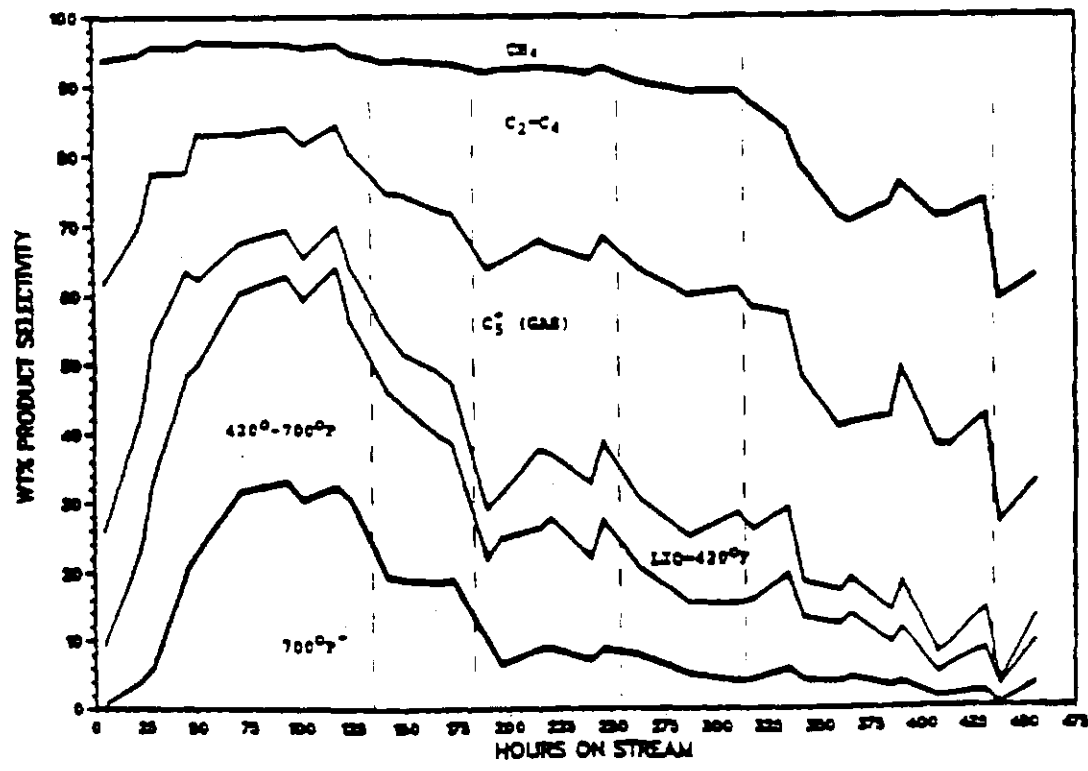
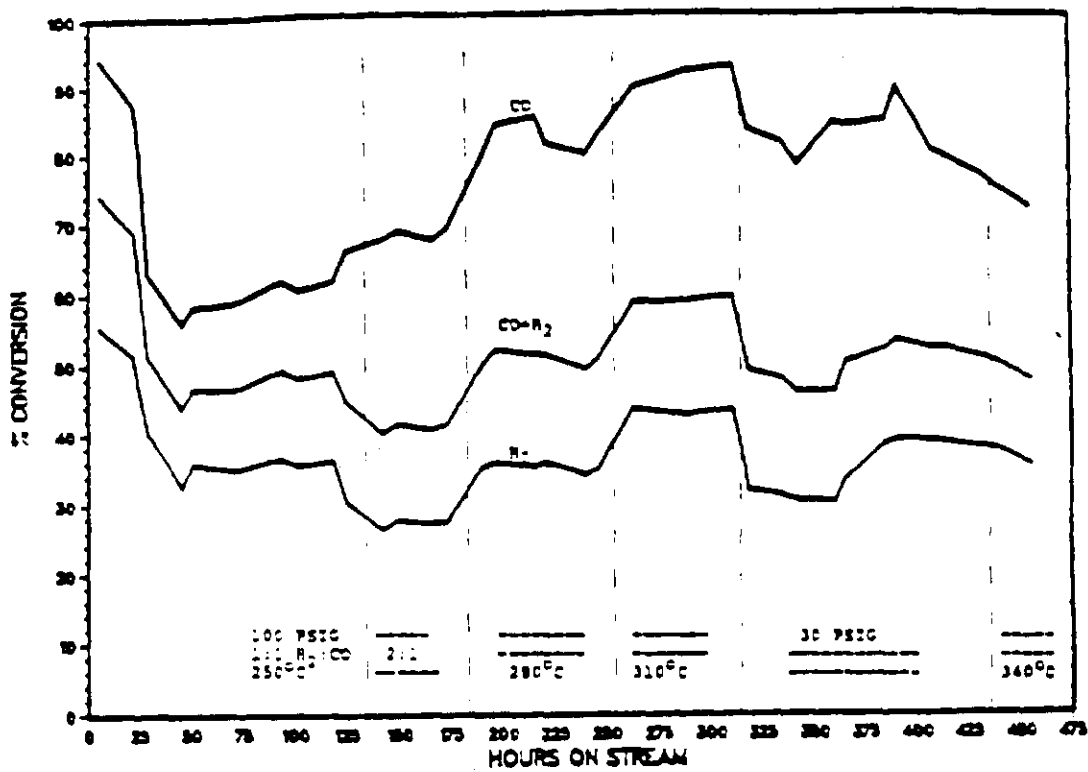
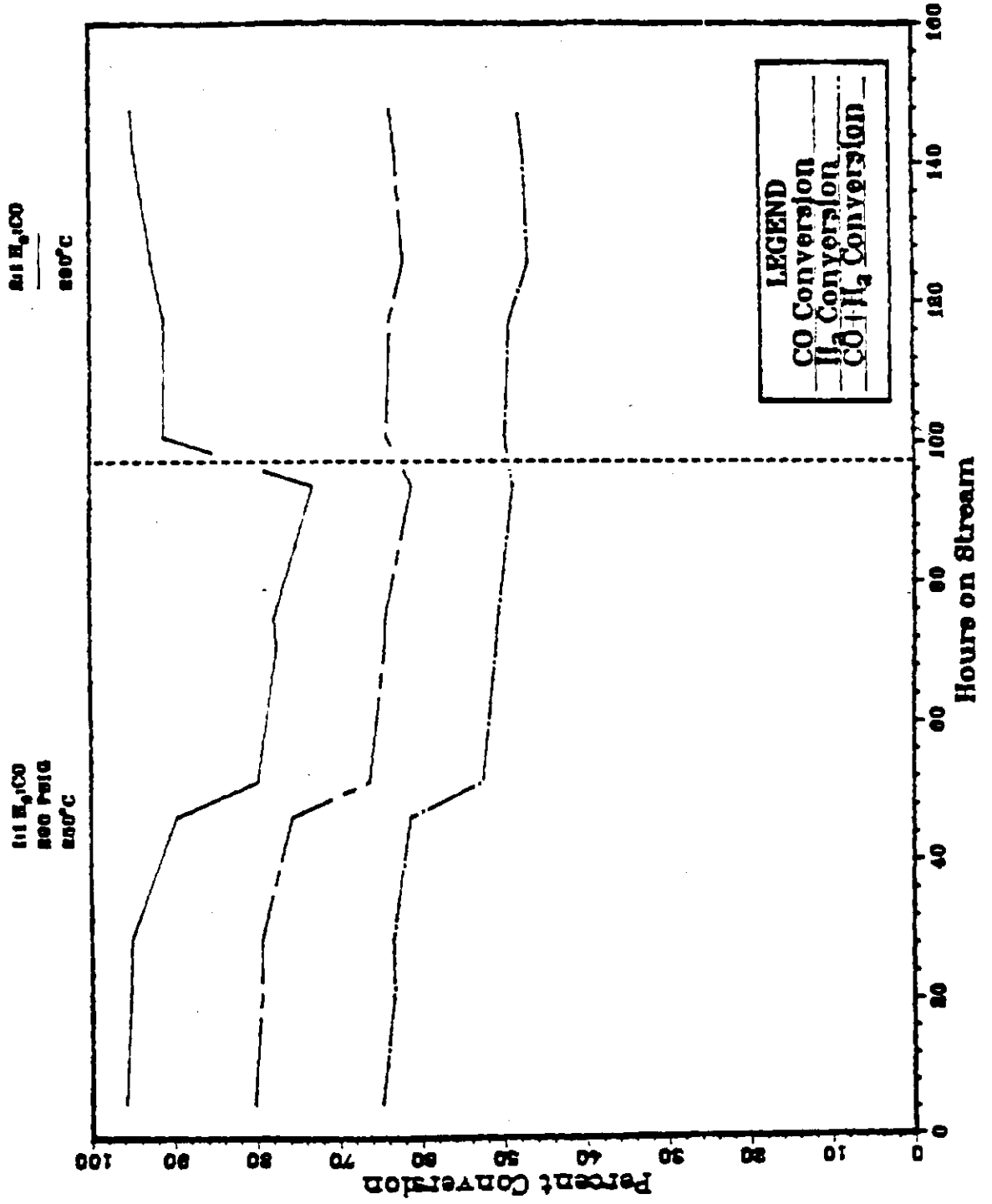


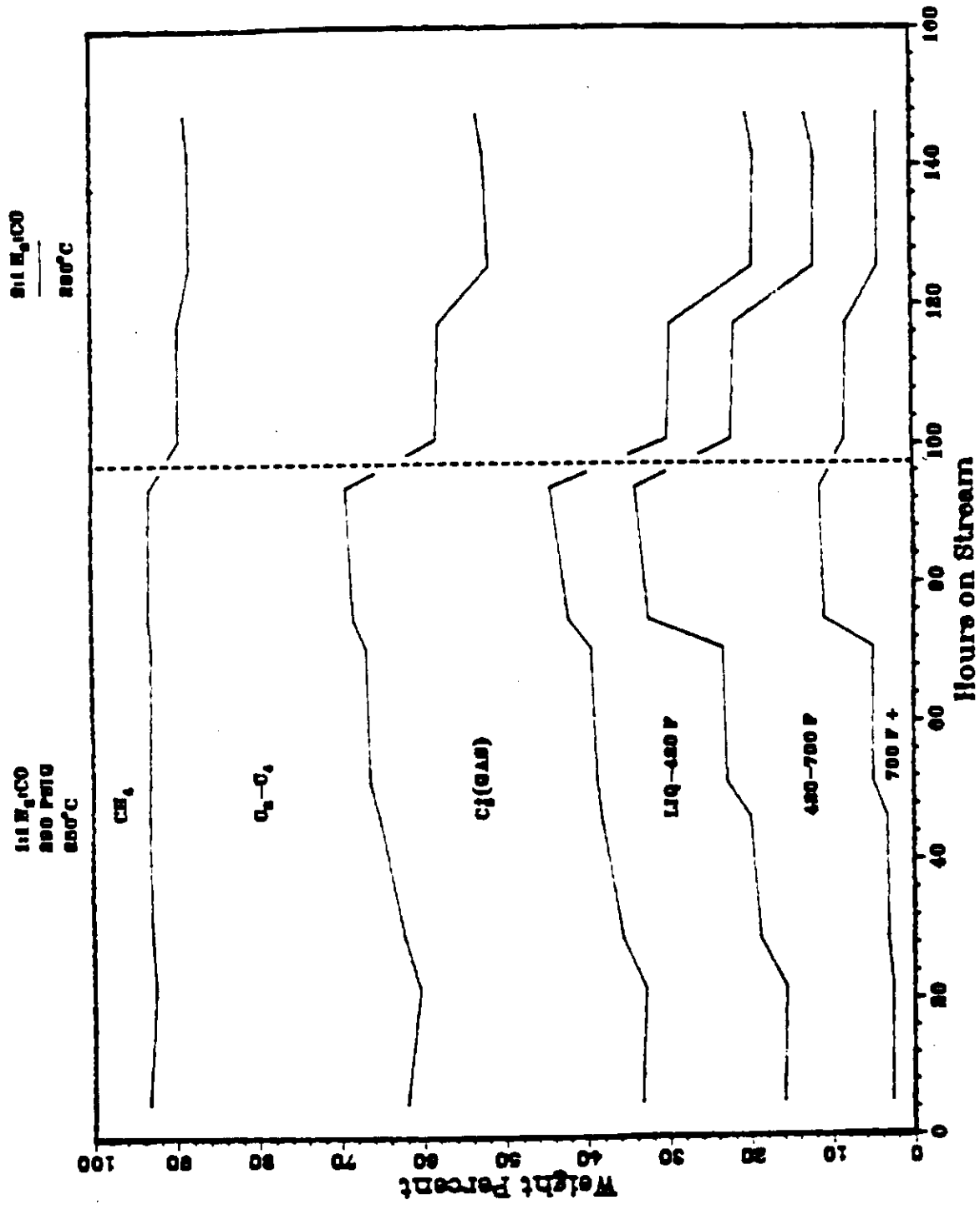
Fig. 59

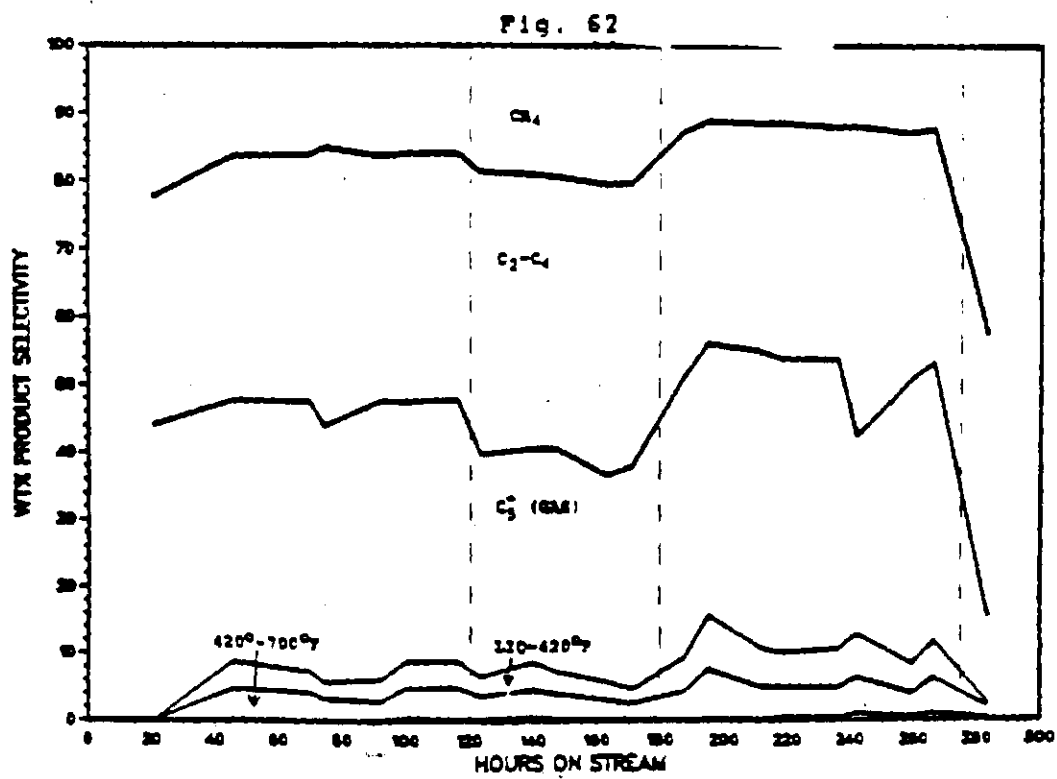
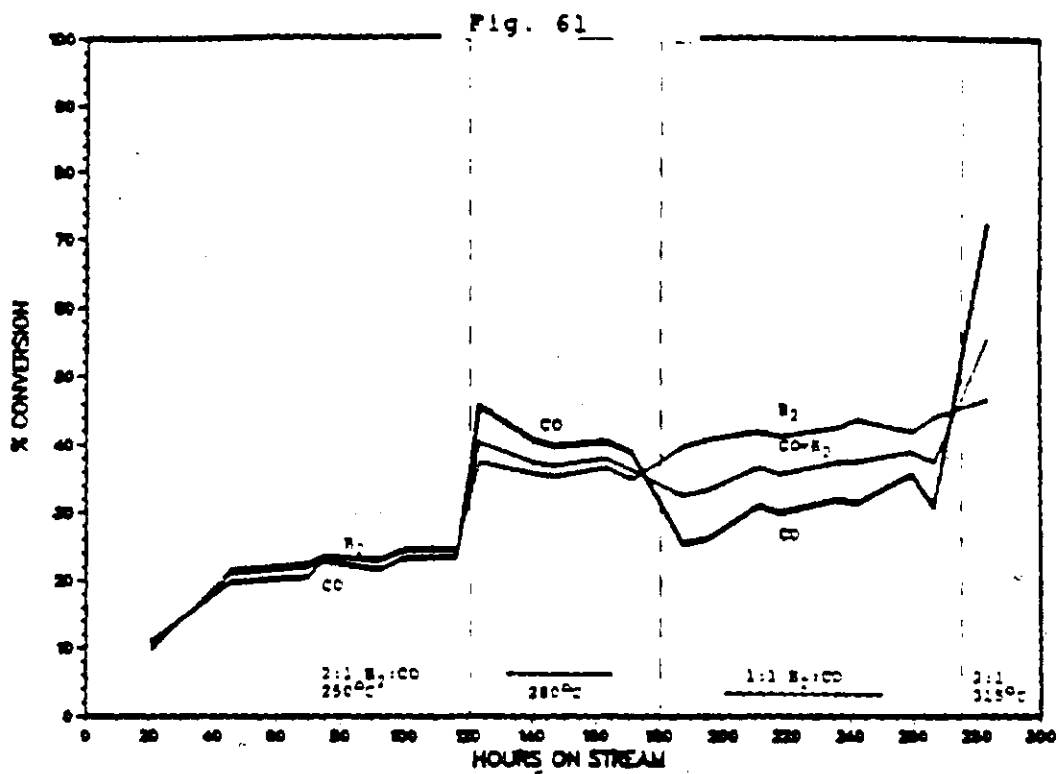
1001115



1001115

Fig. 60





RUN NO. 10011-08

Fig. 63

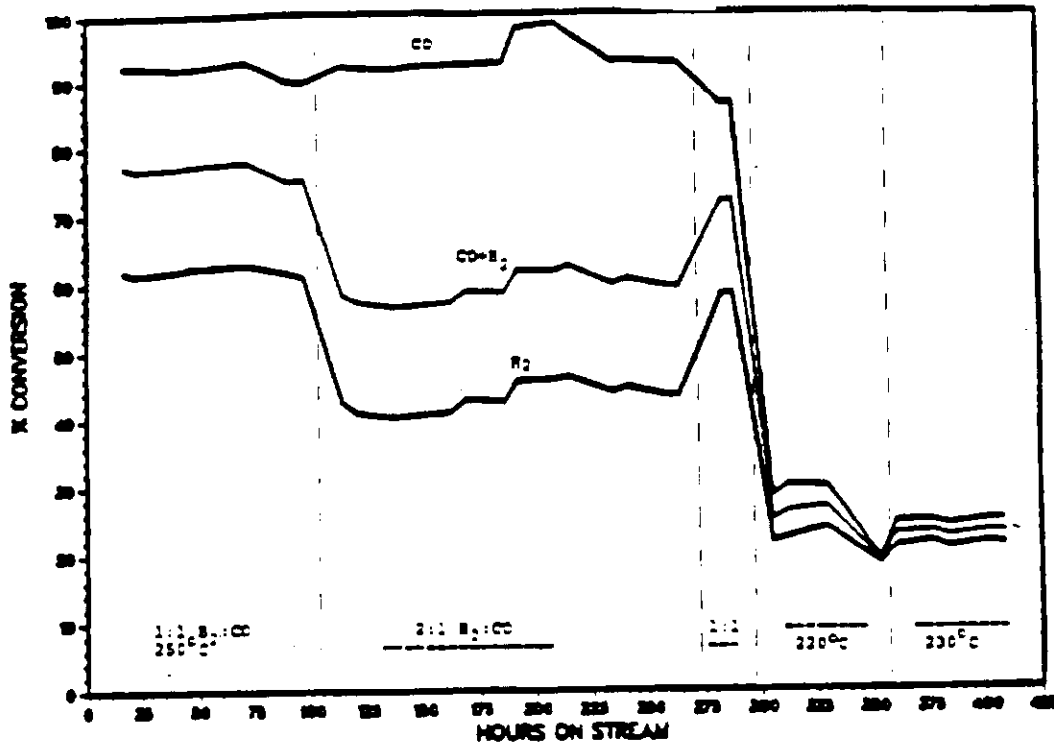
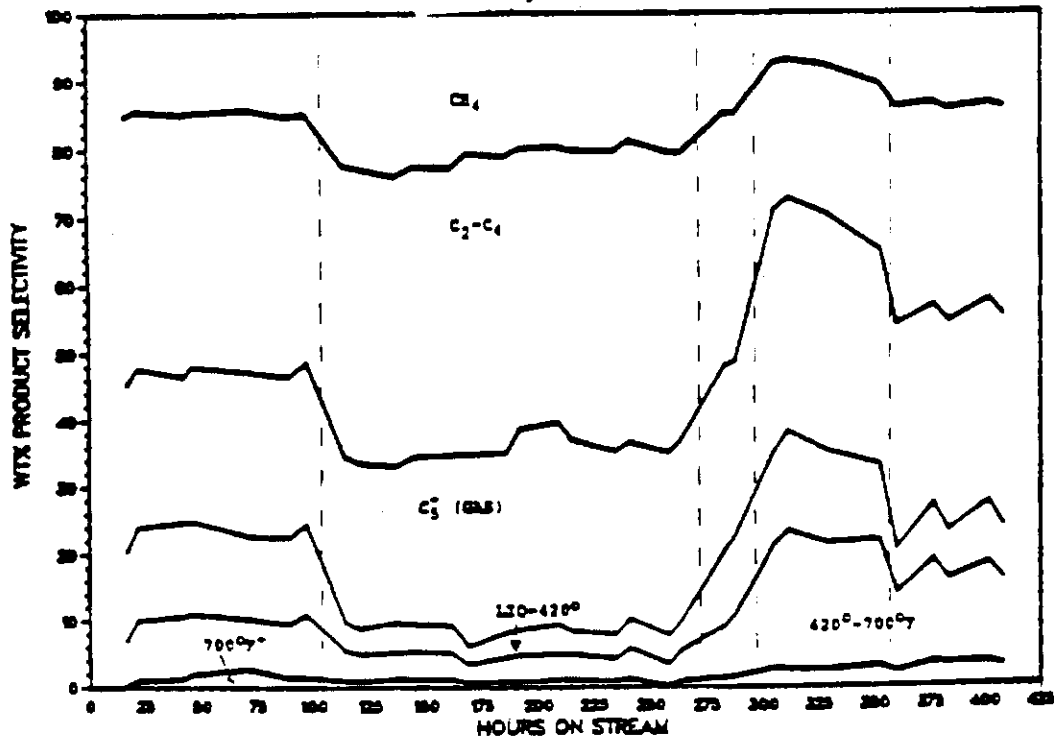
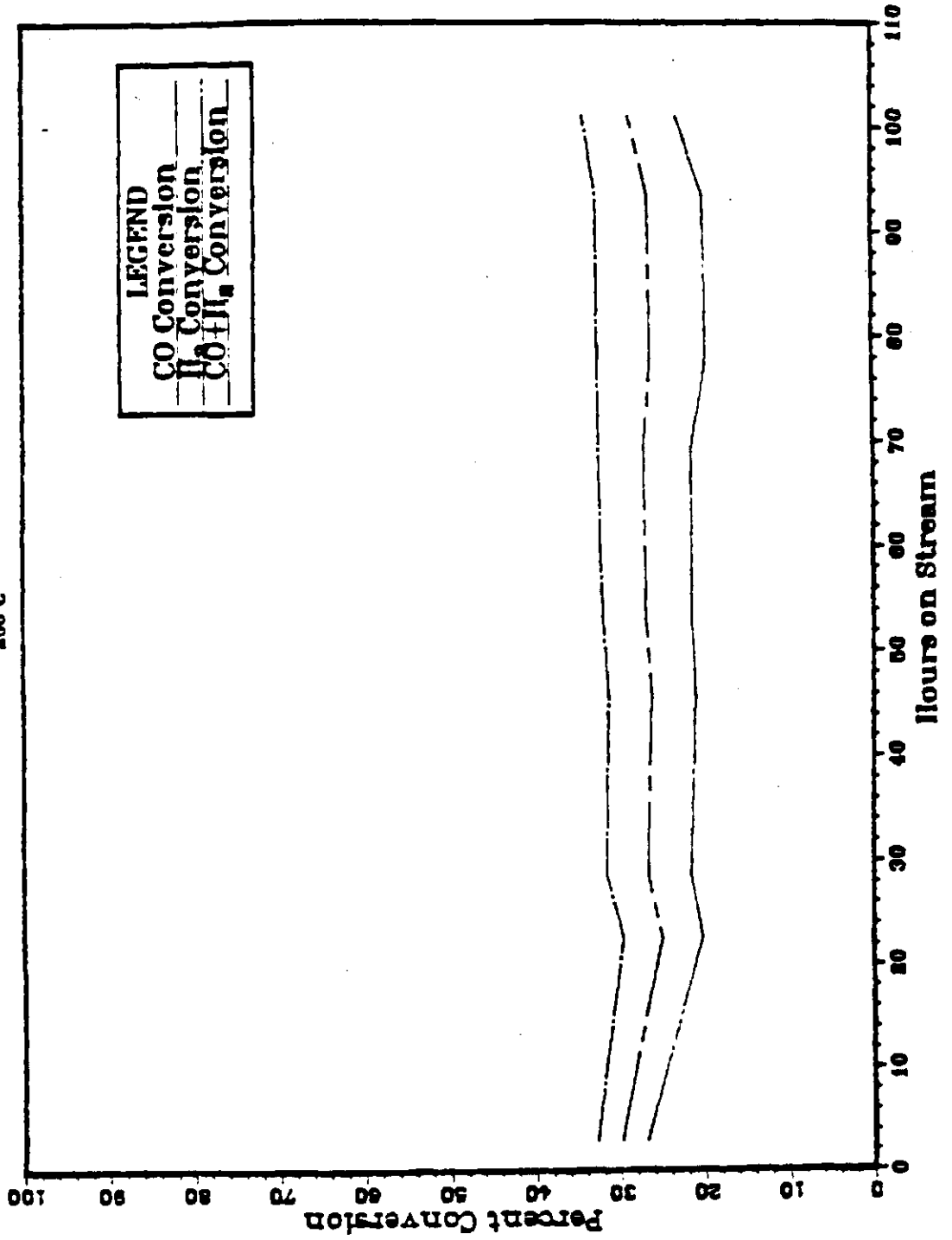


Fig. 64



1001111

1:1 H₂:CO
200 PSIG
200°C



LEGEND
CO Conversion
H₂ Conversion
CO+H₂ Conversion

.Fig. 65

1001111

111 B₂CO
300 PmlO
201°C

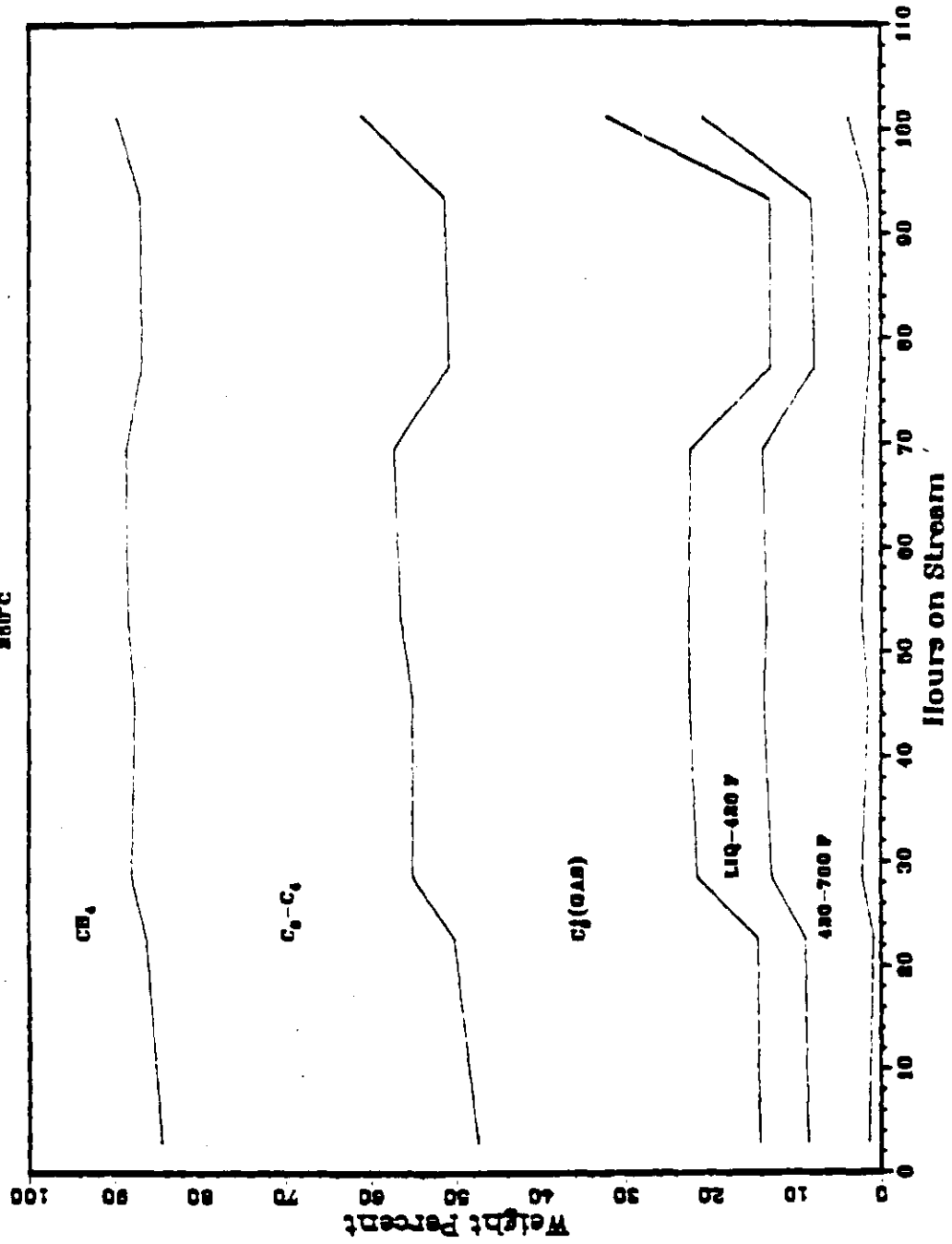


Fig. 66

1001110

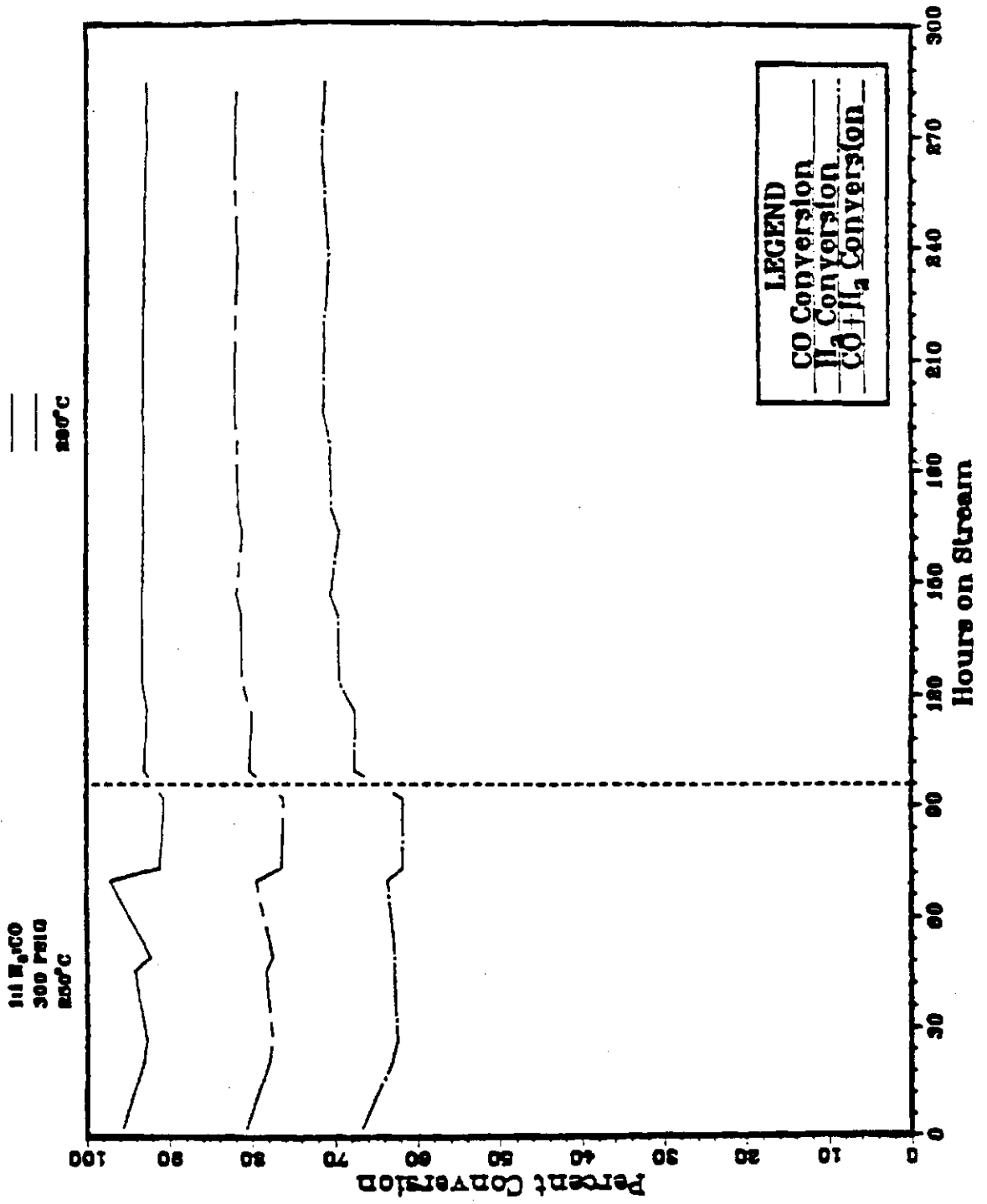


Fig. 67

1001110

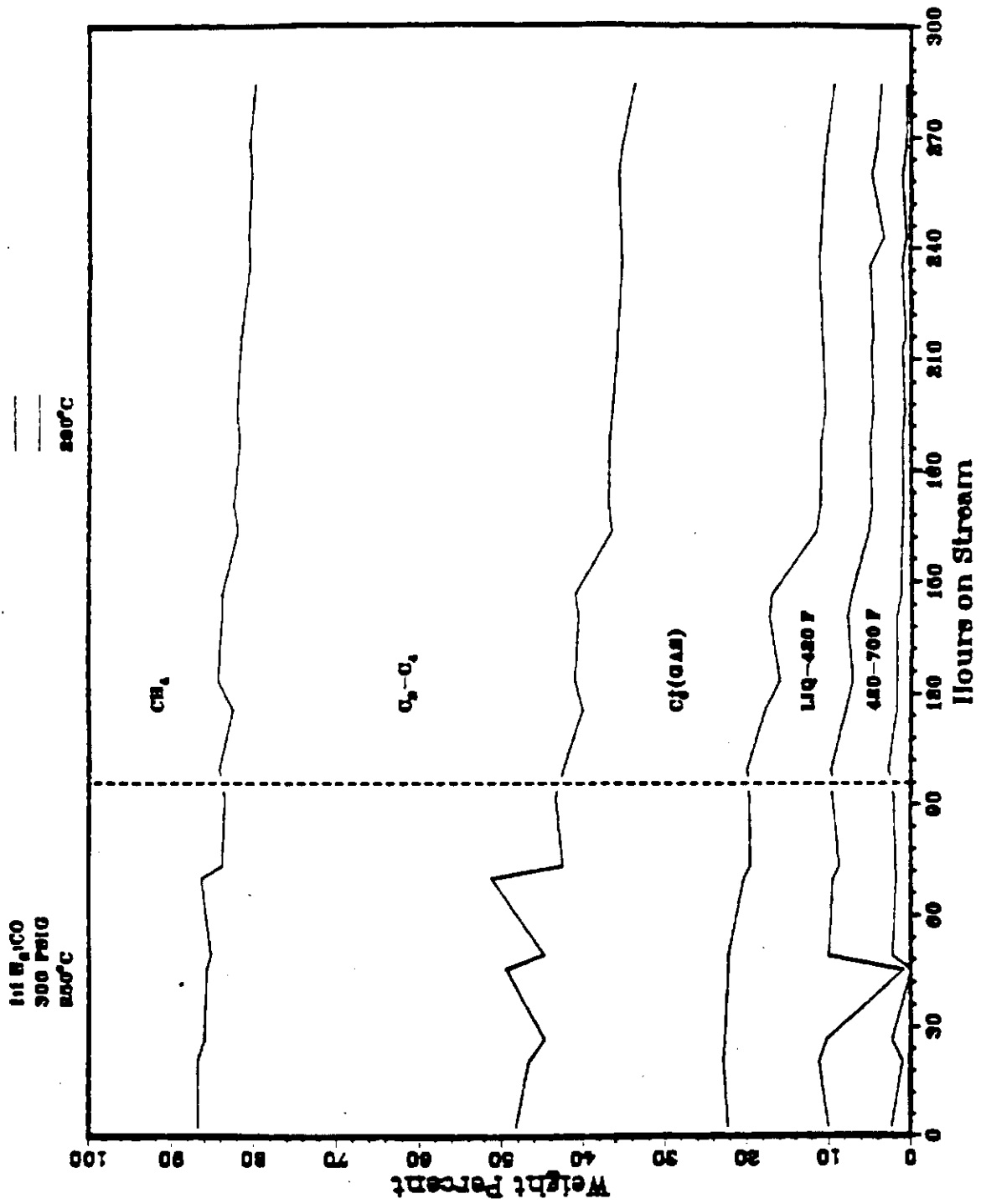


Fig. 69

RUN 11677-07

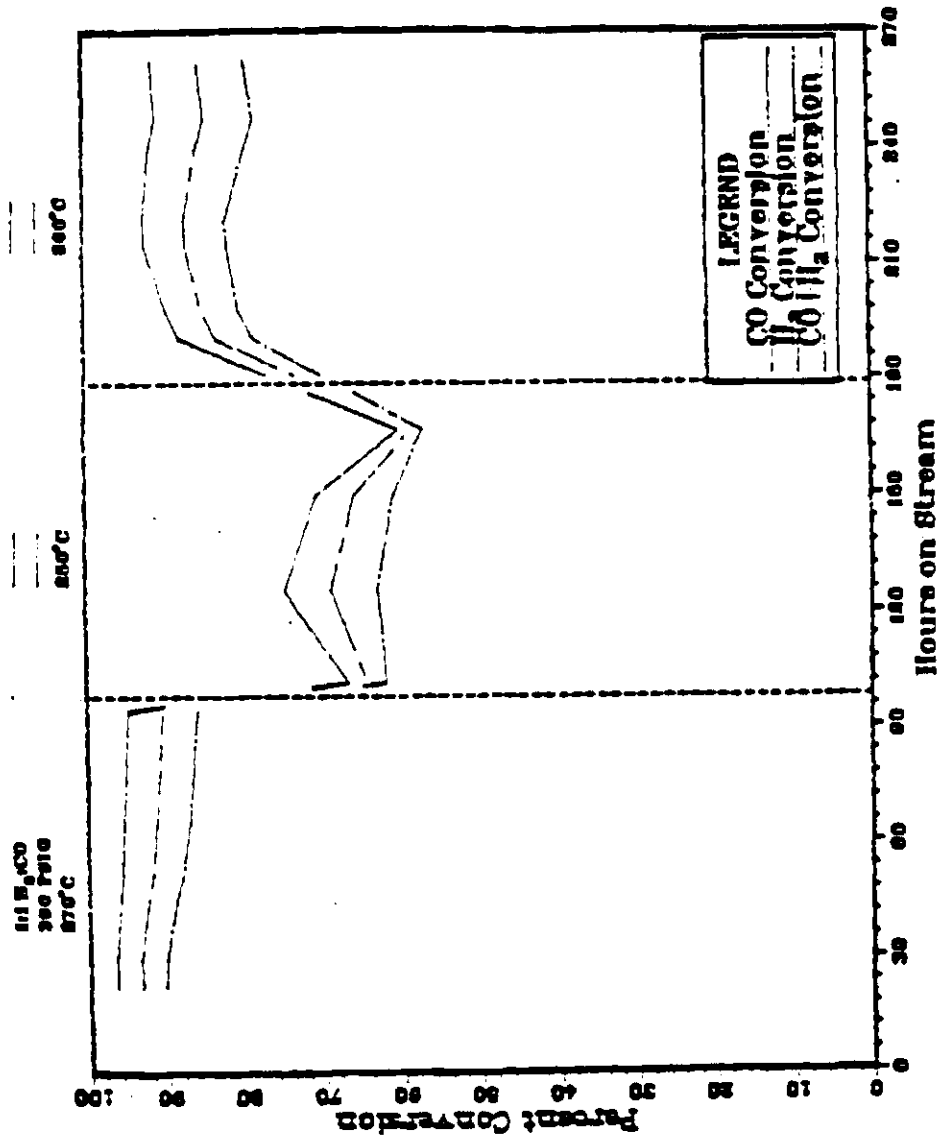
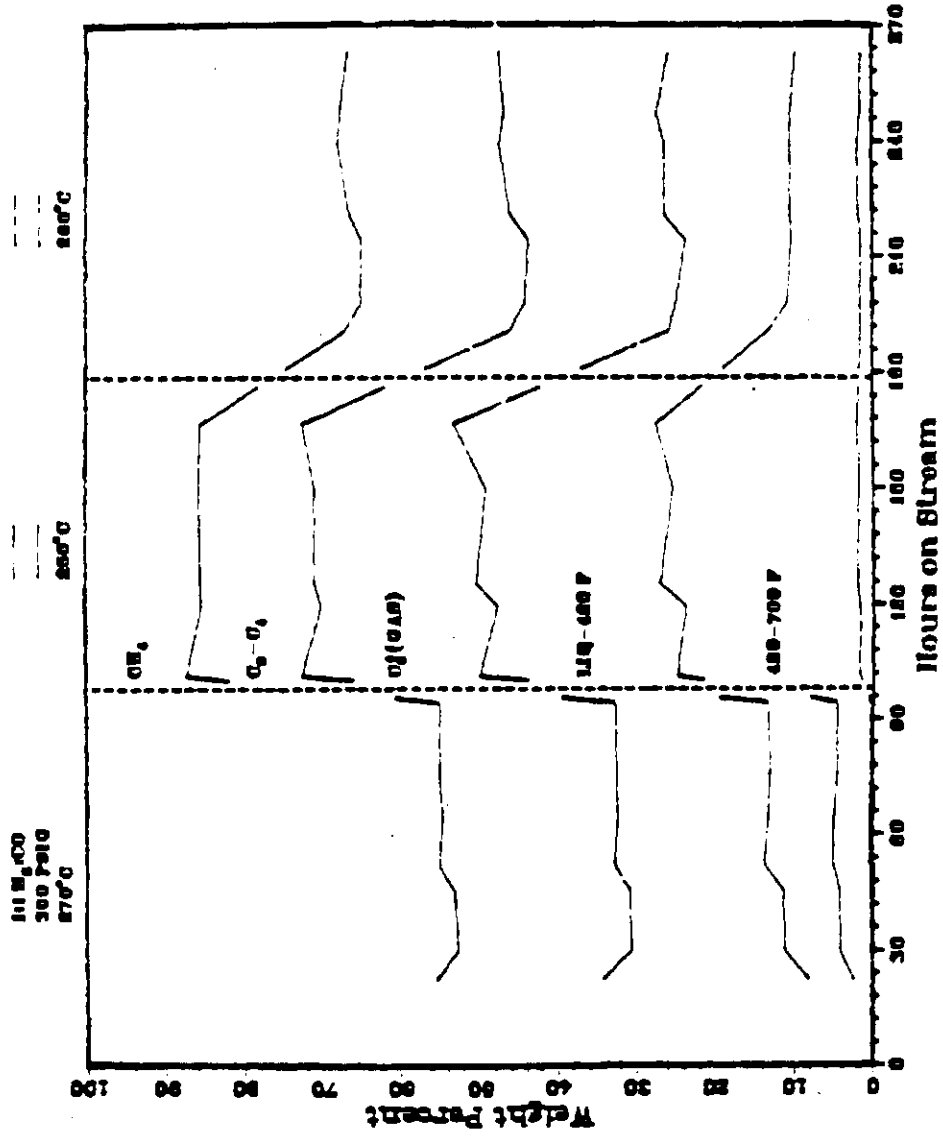


Fig. 70

RUN 11677-07



RUN 10225-08

111 H₂,CO
300 FBID
87°C

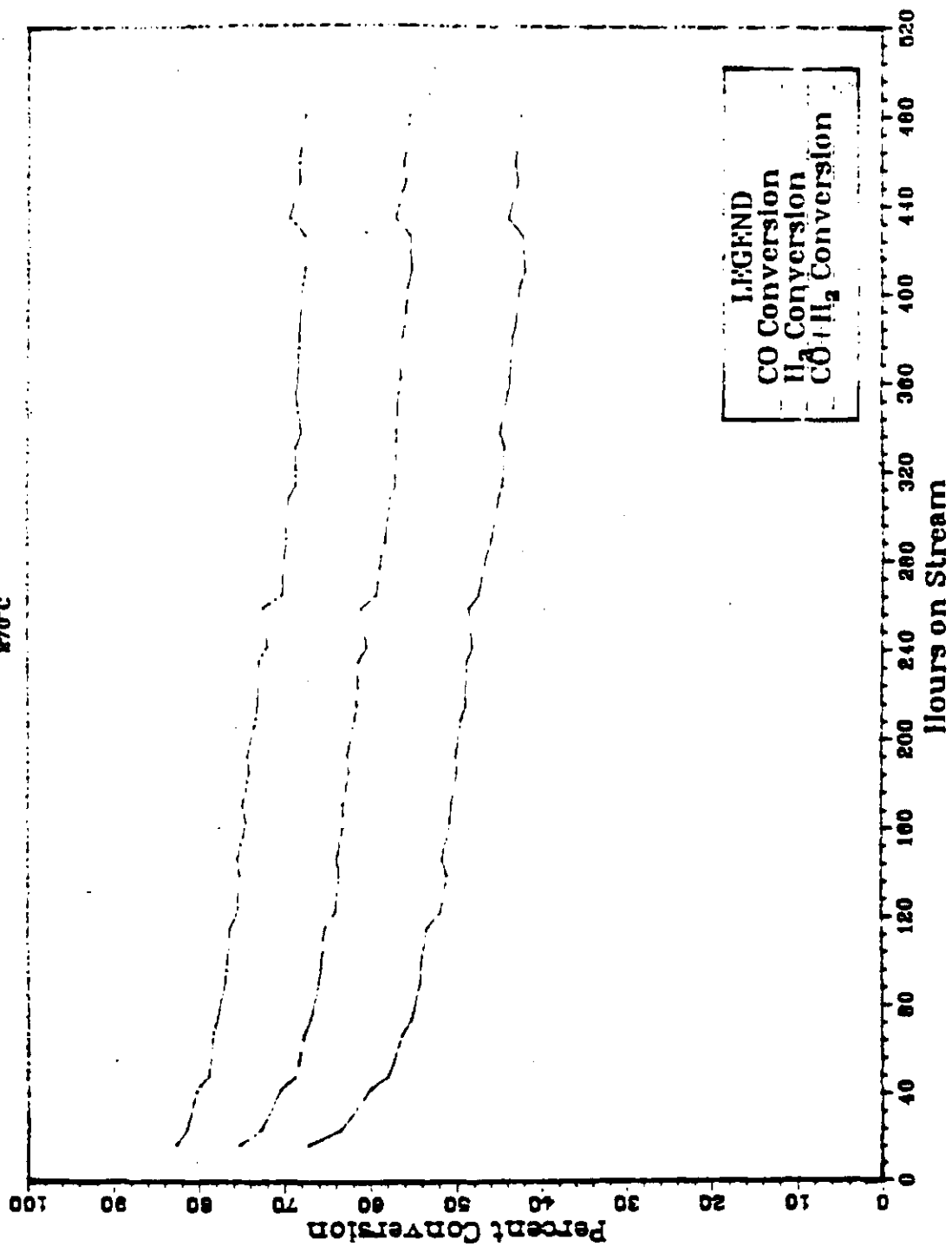


Fig. 71

RUN 10225 08

1:1 H₂O
300 PSIG
270°C

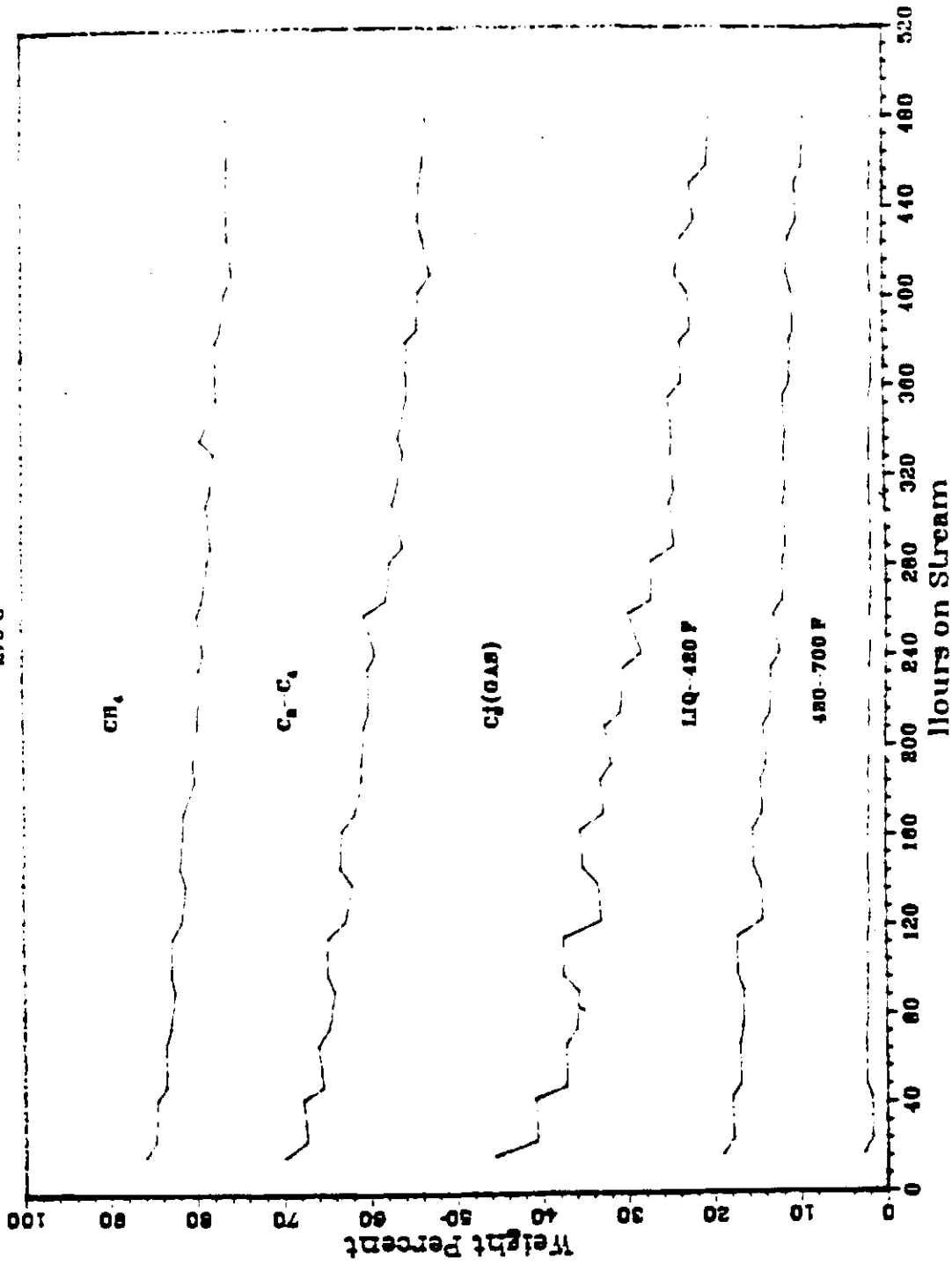
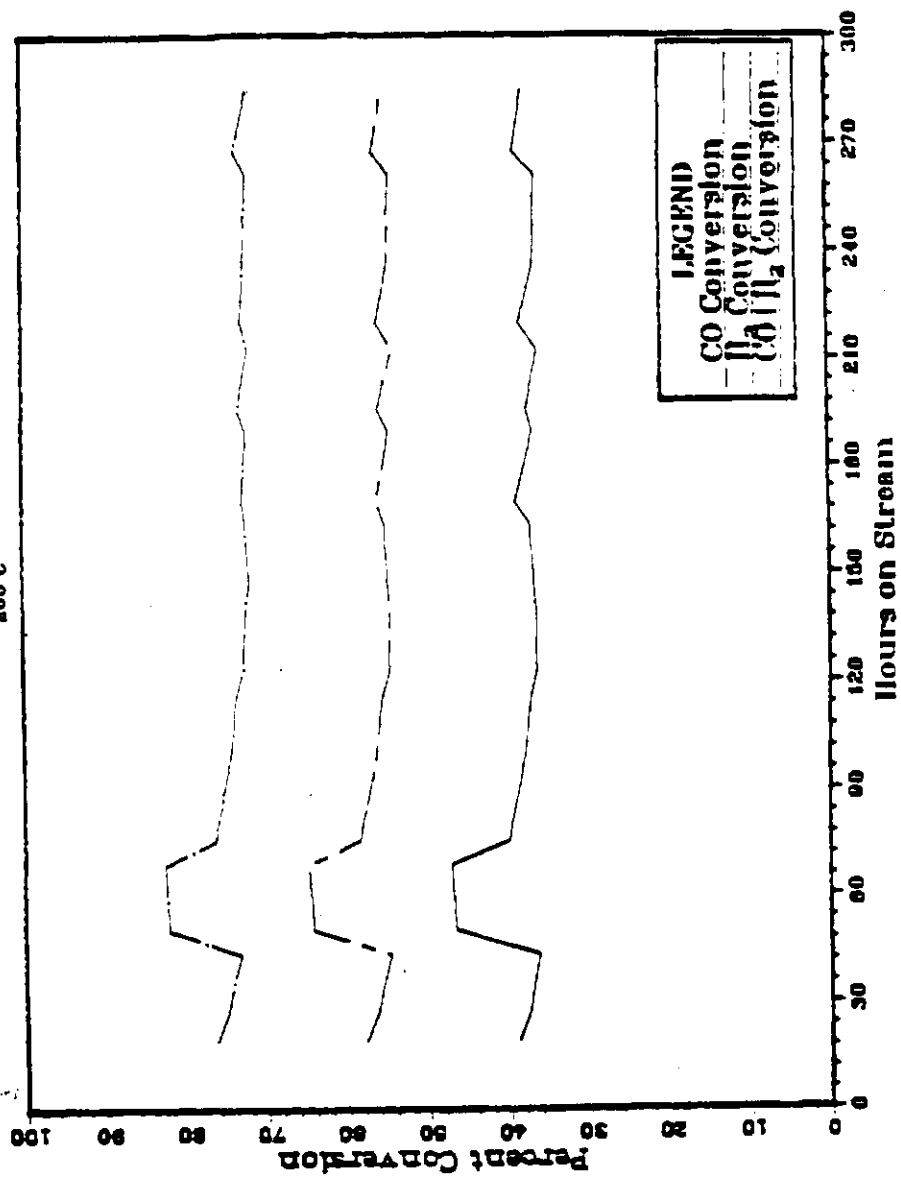


Fig. 72

Fig. 73

RUN 11677-11

111 W₂O
300 P₁₀
200°C



RUN 11677-11

611 P₂O₅
300 P₂O₅
880°C

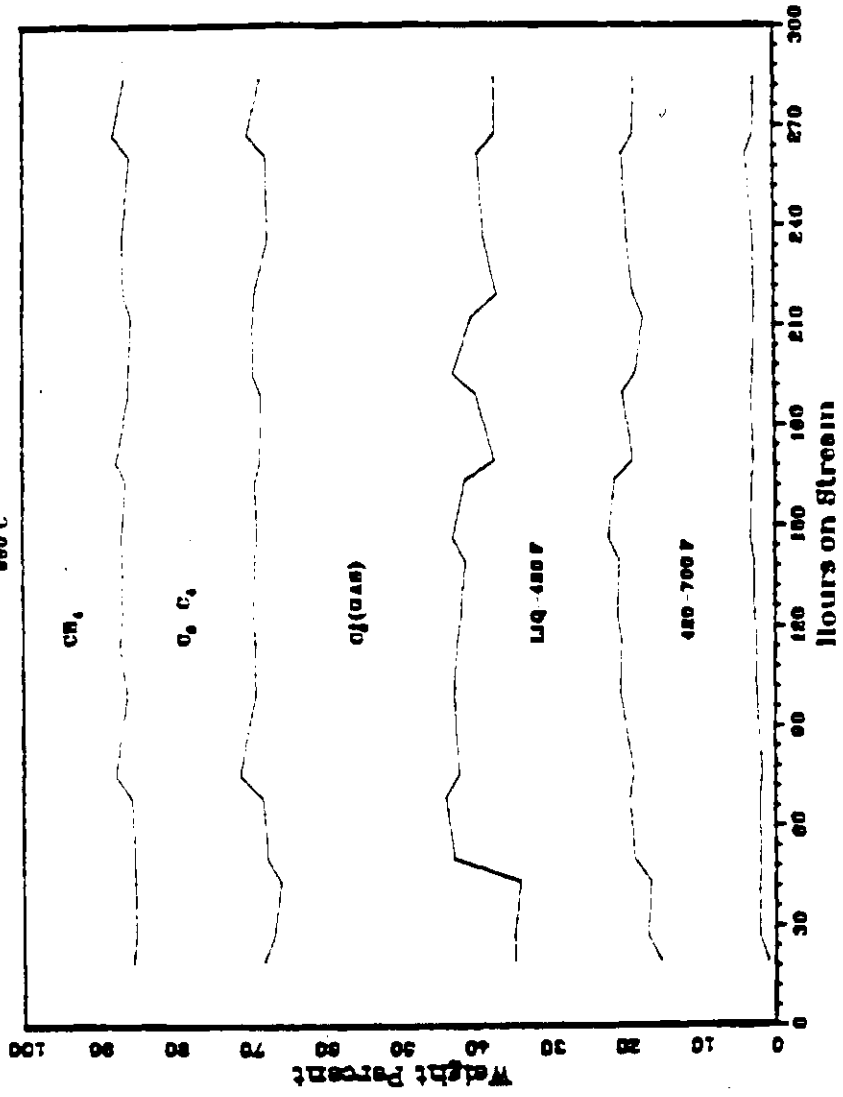


Fig. 74

Fig. 75

RUN 11677-11

1:1 N₂:CO
300 Torr
800°C

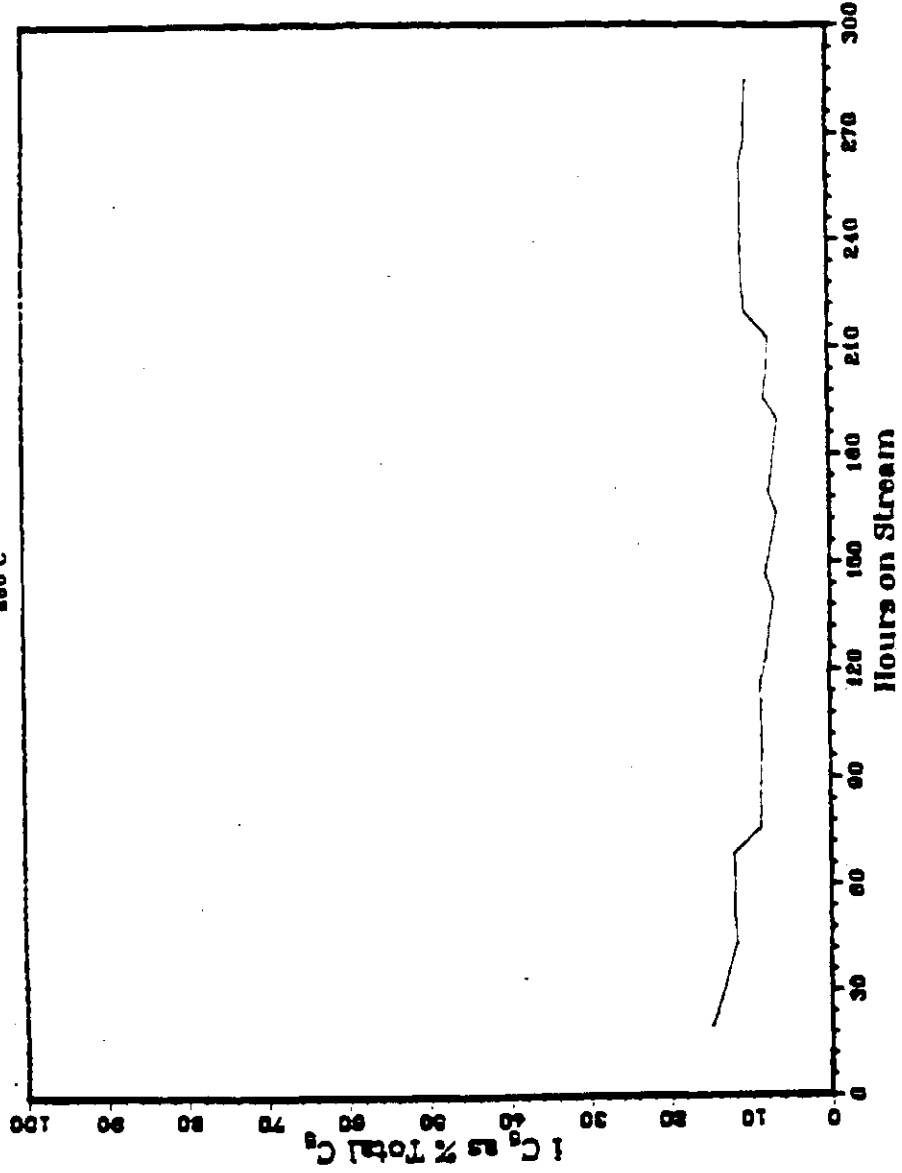


Fig. 76

RUN 11677-11

11 H₂O
300 Pa10
850°C

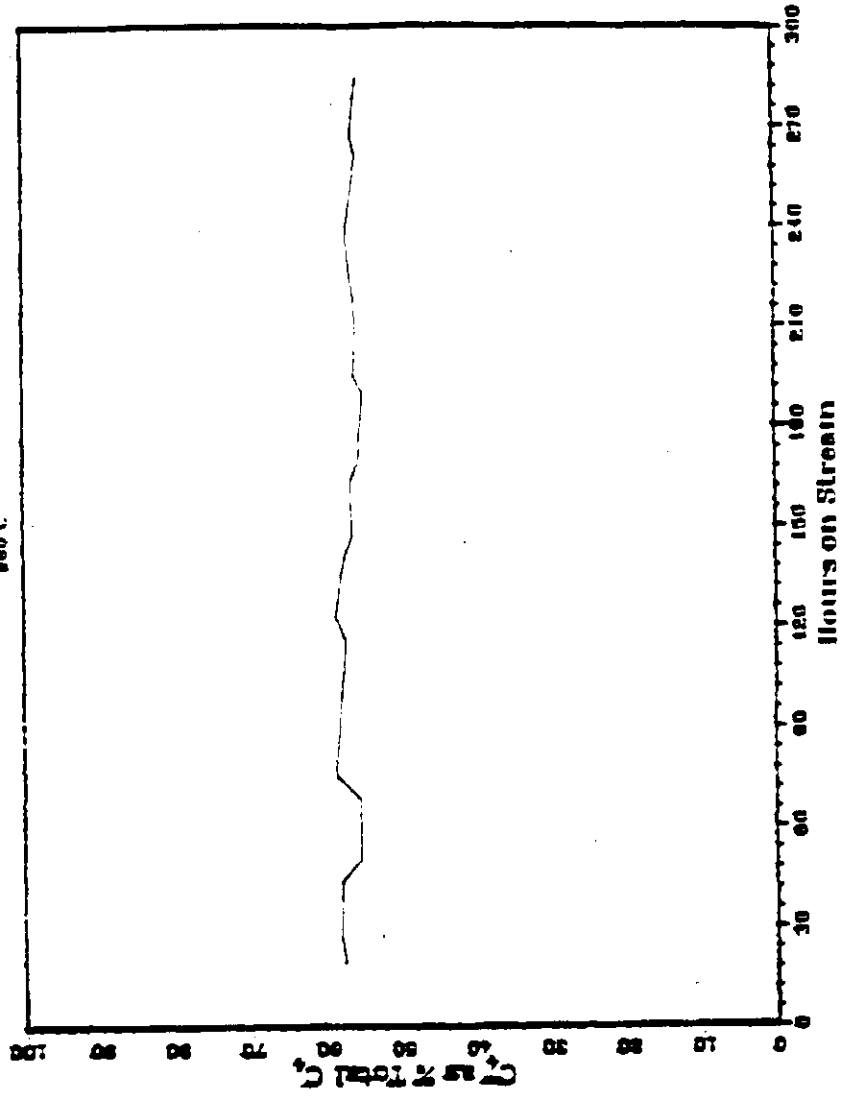


Fig. 77

SCHMATIC DIAGRAM OF UNION CARBIDE
TUBULAR REACTOR SIMULATION PROGRAMS
INPUT/OUTPUT RELATIONS

THE BERTY REACTOR, a CSTR (CONTINUOUS-FEED STIRRED TANK REACTOR) operates under steady state conditions with a high internal recycle rate, causing the catalyst to be exposed to a known and unvarying gas phase composition

↓
YIELDS DATA



DATA CORRELATION by MULTIPLE REGRESSION FOR:

1. CO Conversion Rate as a function of partial pressures of H₂, CO and Temp. in the Power Law form
2. Wt%CH₄ as a function of H₂/CO ratio and Temp.
3. Schulz-Flory alpha as a function of H₂/CO ratio and Temp.

is put ↓ into

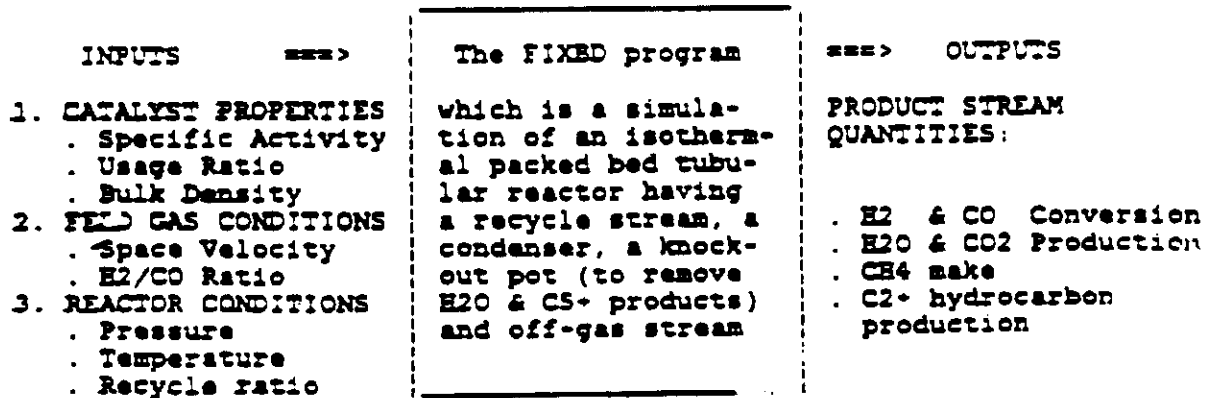
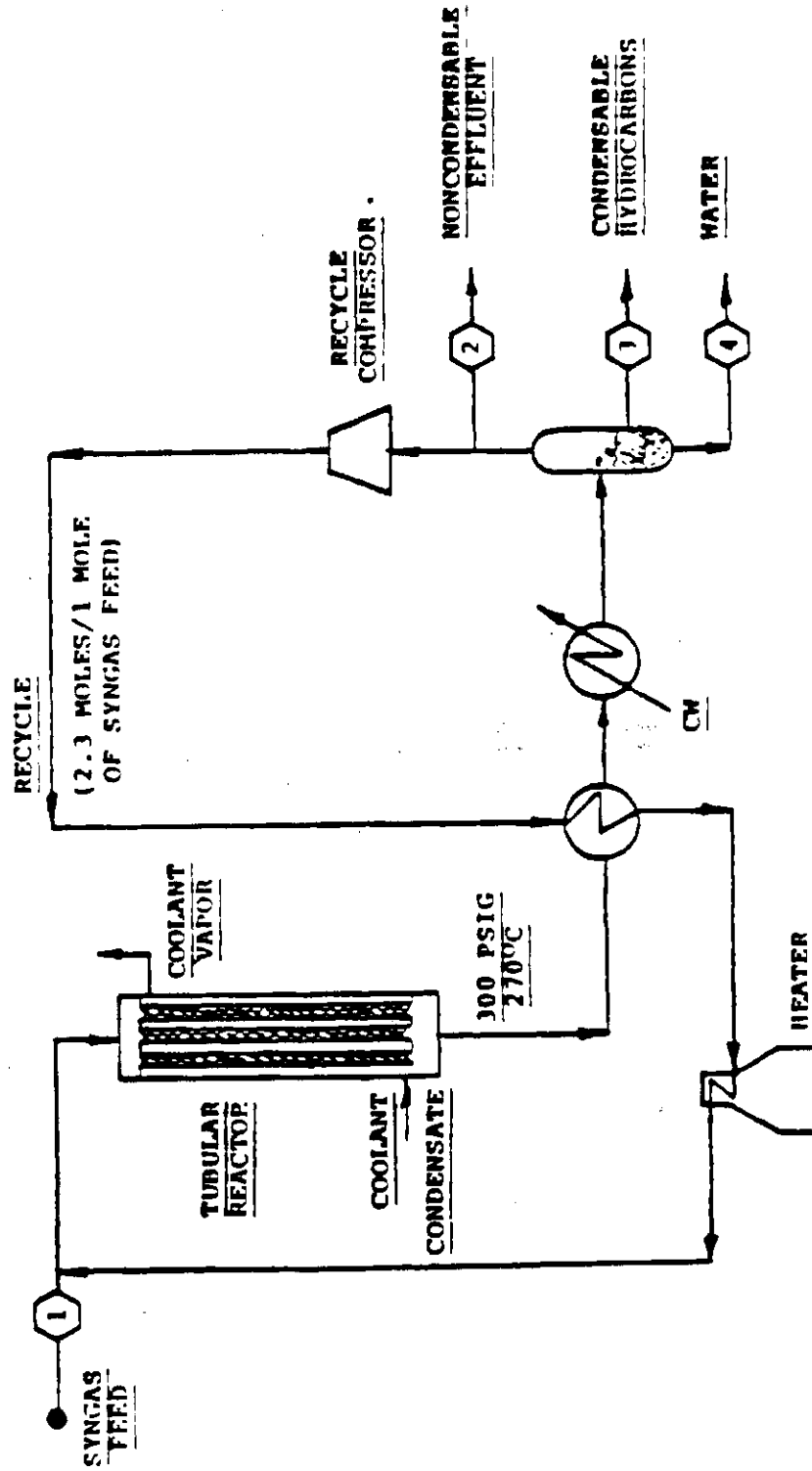
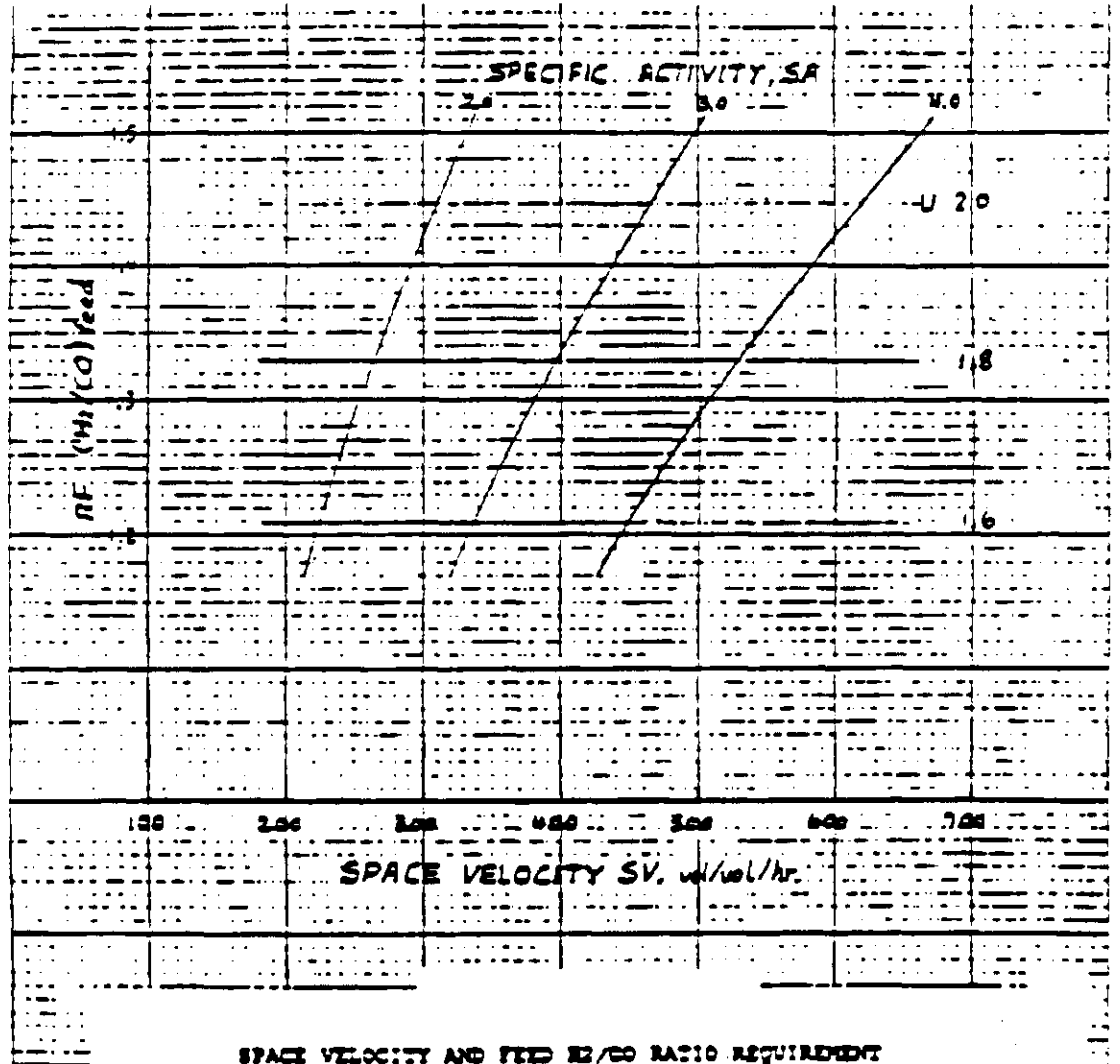
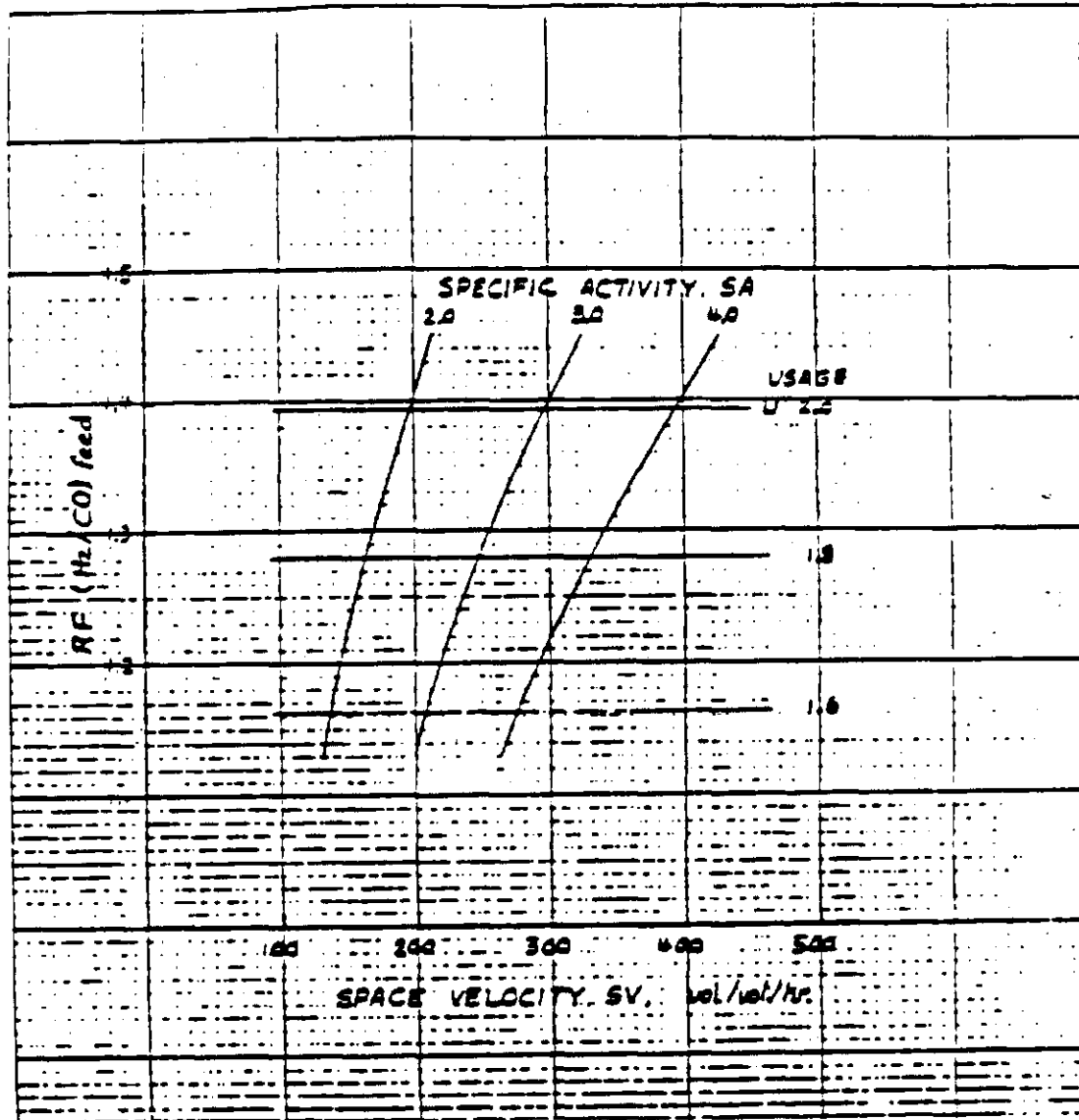


Fig. 78 Schematic diagram of the F-T reactor system.





SPACE VELOCITY AND FEED R2/CO RATIO REQUIREMENT
 FOR TUBULAR REACTOR AT 270 C 300 PSIG 2.3 RECYCLE RATIO
 WITH A CATALYST CAPABLE OF CERTAIN USAGE RATIO U
 SPECIFIC ACTIVITY SA, AND BULK DENSITY 0.6 GR/CC
 FOR 85 % SYNGAS CONVERSION AND 12.3 WT%₄ PRODUCT SELECTIVITY



SPACE VELOCITY AND FEED H₂/CO RATIO REQUIREMENT
 FOR TUBULAR REACTOR AT 270 C 300 PSIG 2.3 RECYCLE RATIO
 WITH A CATALYST CAPABLE OF CERTAIN USAGE RATIO U
 SPECIFIC ACTIVITY SA, AND BULK DENSITY 0.6 GM/CC
 FOR 85 % SYNGAS CONVERSION AND 7.8 WT%₄ PRODUCT SELECTIVITY