

Appendix A Computer Program for the Calculation of Gas Solubility

A computer program has been prepared to calculate gas solubility in a n-paraffin wax based on the correlation of this work. Solubilities are calculated for any one of the gases H_2 , CO , CH_4 , CO_2 , C_2H_4 , C_2H_6 or their mixtures at a specified temperature, pressure, and wax-free mole fractions of the gases in the gas phase. The molecular weight of the wax is specified. The calculation gives the solubilities in mols per kg of wax and in mole fractions in the liquid. The equilibrium mole fraction of wax in the vapor is also calculated. The program is in FORTRAN.

The input for the program is illustrated below.

Card no.	Format	Variables
1	2F10.3	T in K and P in atm of the system
2	I1	number of solutes in the system
3	F10.3	molecular weight of the wax
4 and after	I1,f10.3	gas identification and vapor phase composition on wax-free basis

Gas identification is an interger, which represents a gas species studied in this work.

Gas Identification	Gas species
1	hydrogen
2	carbon monoxide
3	methane
4	carbon dioxide
5	ethane

The output lists the calculated liquid phase composition of each solute and the wax mol fraction in the vapor phase.

Program List

```

35005,b2u,t80,ll,pr
mnfftn(t,u)
#eor
  program gas(input,output,tape5 = input,
&tape6 = output)
  dimension x(5),f(5)
  common t,p,y(5),ngas,ncomt,vcal(2),
&r,rmw,rn,tc(6),pc(6),omeg(6),
&a(6),b(6),am,bm,alpha(6),
&phi(6),chi(6,6),q(6)
  common con(2,6),xx(6)
  common /was/ ywax

c
c  t: temperature, K
c  p: pressure, atm
c  ngas: number of gas species
c  ncomt: number of total species including wax
c  vcal: calculated gas or liquid volume
c  x: liquid composition
c  y: vapor composition
c  r: gas constant
c  rmw: molecular weight of the wax
c  rn: equivalent carbon number of the wax
c  a,b, alpha: RKS equation parameters
c  tc,pc,omeg: physical properties of species
c  phi: volume fraction of a species
c  chi: adjustable parameter Aij
c  q: van der Waal's volume of a species
c  parm,xjac,xjtj,work,con: working variables
c
  r=82.06
  read(5,2)t,p
  2 format(2f9.3)
  read(5,3)ngas
  3 format(i1)
  ncomt=ngas+1
  read(5,12)rmw
  12 format(f10.3)
  call wax
  write(6,14)t,p
  14 format(1x,'system temperature in k=',f10.2,/,
&8x,'pressure in atm =',f10.2,/)
  write(6,15)rmw
  15 format(1x,'wax molecular weight=',f10.2,/)
c
c  read in the the gas id, and its vapor-
c  phase composition

```

```

c
  do 20 n1=1,ncomt
    do 20 n2=1,ncomt
20  chi(n1,n2)=0.0
    write(6,22)
22  format(/,1x,'species no.',5x,'species',12x,
    &'y on wax-free basis')
    do 39 j1=1,ngas
      read(5,25)id,y(j1)
25  format(i1,2f10.4)
      x(j1)=y(j1)/10.
      call gprop(id,j1)
39  continue
      ywax=0.0
      write(6,49)ncomt,ywax
49  format(4x,i1,12x,'wax ',12x,f12.4)
      do 50 n1=1,ncomt
50  chi(ncomt,n1)=chi(n1,ncomt)
c
c call subroutine to find the liquid composition.
c
  do 100 icon=1,10
    call obj(x,f)
    grad=0.0
    do 60 n1=1,ngas
60  grad=grad+abs(f(n1)/x(n1)-1.0)
      if(grad-5.0e-5)110,90,90
90  do 95 n1=1,ngas
95  x(n1)=f(n1)
100 continue
110 write(6,118)
118 format(//,'species no.',5x,'calc. y',5x,
    &'calc. x',7x,'solubility',/,
    &42x,'(mol per kg wax)',/)
    dumx=0.0
    dummy=0.0
    do 150 j1=1,ngas
      sol=x(j1)/(1.0-x(j1))/rmw*1000.0
      dumx=dumx+con(1,j1)
      dummy=dummy+con(2,j1)
      write(6,120)j1,con(2,j1),x(j1),sol
120 format(1x,i3,4x,f12.4,3x,e12.3,4x,e12.3)
150 continue
      dumx=dumx+con(1,ncomt)
      dummy=dummy+con(2,ncomt)
      write(6,130)ncomt,con(2,ncomt),con(1,ncomt)
130 format(1x,i3,4x,f12.4,3x,e12.3,12x,'-')

```

```

    write(6,140)dumy,dumx
140 format(/,1x,'total',3x,f12.4,1x,f12.4,/)
    stop
    end

```

```

    subroutine obj(x,f)

```

```

c
c this subroutine set the objective function to be
c minimized by the levenberg-marquardt algorithm
c

```

```

    dimension coef(4),fu(2,6),x(5),f(5)
    common t,p,y(5),ngas,ncomt,vcal(2),
    &r,rmw,rm,tc(6),pc(6),omeg(6),
    &a(6),b(6),am,bm,alpha(6),
    &phi(6),chi(6,6),q(6)
    common con(2,6),xx(6)
    common /was/ ywax
    complex z(3)
    con(1,ncomt)=1.0
    con(2,ncomt)=ywax
    do 10 i0=1,ngas
    con(1,i0)=x(i0)
    con(2,i0)=(1.0-ywax)*y(i0)
    con(1,ncomt)=con(1,ncomt)-con(1,i0)
10 continue
    con(1,ncomt)=abs(con(1,ncomt))
    do 50 j1=1,ncomt
    tr=t/tc(j1)
    call calal(j1,tr)
    a(j1)=0.42747*alpha(j1)*r*r*tc(j1)*tc(j1)/pc(j1)
    b(j1)=0.08664*r*tc(j1)/pc(j1)
50 continue
    do 90 i1=1,2
    call mix(i1)
    call mixb(i1)
    call mixa(i1)
    rtop=r*t/p
    coef(1)=1.0
    coef(2)=-rtop
    coef(3)=-bm*rtop-bm*bm+am/p
    coef(4)=-bm*am/p
    call zrpolynomial(coef,3,z,ier)
    vcal(1)=real(z(1))
    vcal(2)=real(z(3))
    do 90 j1=1,ncomt
    call rabrn(i1,j1,dadxi,dbdxi)
    call fugac(fu,vcal(i1),i1,j1,dadxi,dbdxi)

```

```

90 continue
  do 100 i1=1,ngas
    f(i1)=con(2,i1)/(fu(1,i1)/fu(2,i1))
100 continue
  xwax=1.0
  do 110 n1=1,ngas
110 xwax=xwax-f(n1)
  ywax=fu(1,ncomt)/fu(2,ncomt)*xwax
  return
  end

```

subroutine wax

```

c
c given the molecular weight of the wax, this
c subroutine generates the necessary wax
c information for further calculations
c

```

```

  dimension coef(4,4)
  common t,p,y(5),ngas,ncomt,vcal(2),
&r,rmw,rn,tc(6),pc(6),omeg(6),
&a(6),b(6),am,bm,alpha(6),
&phi(6),chi(6,6),q(6)
  common con(2,6),xx(6)
  rn=(rmw-2.0158)/14.0268
  q(ncomt)=13.67*2.0+(rn-2.0)*10.23
  coef(1,1)=0.31391e04
  coef(2,1)=-0.76229e00
  coef(3,1)=0.11165e03
  coef(4,1)=0.16693e-01
  coef(1,2)=0.36175e-01
  coef(2,2)=-0.47775e00
  coef(3,2)=0.58524e00
  coef(4,2)=0.43358e-01
  coef(1,3)=0.80936e01
  coef(2,3)=0.99288e00
  coef(3,3)=0.54555e02
  coef(4,3)=0.80034e01
  coef(1,4)=0.73697e-01
  coef(2,4)=0.14880e00
  coef(3,4)=0.57077e-1
  coef(4,4)=0.44621e-1
  do 10 i=1,4
  c1=coef(1,i)
  c2=coef(2,i)
  c3=coef(3,i)
  c4=coef(4,i)
  dum1=c1/c4

```

```

dum2=1.0-c2
term2=dum1-c3**dum2
term3=exp(-c4*(rn-1.0)*dum2)
dum=(dum1-term2*term3)**(1.0/dum2)
goto (3,4,6,8),i
3 haha=3.03191-0.049901*rn**(2.0/3.0)
tb=1078.-10.0**haha
goto 10
4 tc(ncomt)=tb/dum
goto 10
6 pc(ncomt)=dum
goto 10
8 omeg(ncomt)=dum
10 continue
return
end

```

```

subroutine gprop(id,j1)
common t,p,y(5),ngas,ncomt,vcal(2),
&r,rmw,rn,tc(6),pc(6),omeg(6),
&a(6),b(6),am,bm,alpha(6),
&phi(6),chi(6,6),q(6)
common con(2,6),xx(6)

```

c
c this subroutine generates the gas information needed
c for calculations

```

c
goto (10,20,30,40,50),id
10 write(6,11)j1,y(j1)
11 format(4x,i1,12x,'hydrogen ',9x,f12.4)
tc(j1)=33.2
pc(j1)=12.8
omeg(j1)=-0.22
q(j1)=10.51
a1=0.2456
a2=0.2406
rmw0=127.30
goto 100
20 write(6,21)j1,y(j1)
21 format(4x,i1,12x,'carbon monoxide',3x,f12.4)
tc(j1)=132.9
pc(j1)=34.5
omeg(j1)=0.049
q(j1)=13.01
a1=0.1723
a2=0.1707
rmw0=140.75

```

```

      goto 100
30 write(6,31)j1,y(j1)
31 format(4x,i1,12x,'methane ',9x,f12.4)
      tc(j1)=190.6
      pc(j1)=45.4
      omeg(j1)=0.008
      q(j1)=17.09
      a1=0.1009
      a2=0.1050
      rmw0=146.06
      goto 100
40 write(6,41)j1,y(j1)
41 format(4x,i1,12x,'carbon dioxide ',3x,f12.4)
      tc(j1)=304.2
      pc(j1)=72.8
      omeg(j1)=0.225
      q(j1)=19.68
      a1=0.09710
      a2=0.09240
      rmw0=137.02
      goto 100
50 write(6,51)j1,y(j1)
51 format(4x,i1,12x,'ethane ',9x,f12.4)
      tc(j1)=305.4
      pc(j1)=48.2
      omeg(j1)=0.098
      q(j1)=27.34
      a1=0.04967
      a2=0.05277
      rmw0=188.16
100 factor=1.0-exp(-rmw/rmw0)
      chi(j1,ncomt)=(a1+(a2-a1)*(t-473.2)/100.0)*factor
      return
      end

```

```

      subroutine calal(j1,tr)

```

```

c
c this subroutine calculates equation parameter
c alpha for each species
c

```

```

      common t,p,y(5),ngas,ncomt,vcal(2),
&r,rmw,rn,tc(6),pc(6),omeg(6),
&a(6),b(6),am,bm,alpha(6),
&phi(6),chi(6,6),q(6)
      common con(2,6),xx(6)
      if(tc(j1)-550.)10,10,40
10 dum=omeg(j1)

```

```

      haha=0.480+1.574*dum-0.176*dum*dum
      haha=1.0+haha*(1.0-sqrt(tr))
      alpha(j1)=haha*haha
      goto 120
40 if(rmw-268.529)50,50,60
50 power=0.5
      goto 70
60 power=0.5685-2.552e-4*rmw
70 dum=rmw
      ff=0.47135+0.63325e-2*dum-0.78307e-5*dum*dum
      &+0.44238e-8*dum**3-0.97365e-12*dum**4
      alpha(j1)=(1.0+ff*(1.0-tr**power))**(1.0/power)
120 return
      end

```

```

      subroutine mix(i1)

```

```

c
c   this subroutine sets the parameters needed
c   for executing the mixing rules.

```

```

c
      common t,p,y(5),ngas,ncomt,vcal(2),
      &r,rmw,rn,tc(6),pc(6),omeg(6),
      &a(6),b(6),am,bm,alpha(6),
      &phi(6),chi(6,6),q(6)
      common con(2,6),xx(6)
      dum=0.0
      do 10 n1=1,ncomt
10 dum=dum+con(i1,n1)*q(n1)
      do 20 n1=1,ncomt
20 phi(n1)=con(i1,n1)*q(n1)/dum
      dum1=0.0
      do 30 n1=1,ncomt
30 dum1=dum1+con(i1,n1)
      do 40 n1=1,ncomt
40 xx(n1)=dum1*q(n1)/dum
      return
      end

```

```

      subroutine mixa(i1)

```

```

c
c   this subroutine executes the mixing rule
c   for a

```

```

c
      common t,p,y(5),ngas,ncomt,vcal(2),
      &r,rmw,rn,tc(6),pc(6),omeg(6),
      &a(6),b(6),am,bm,alpha(6),
      &phi(6),chi(6,6),q(6)

```



```

common con(2,6),xx(6)
rlam = alog(2.0)
sum1 = 0.0
do 100 n1 = 1, ncomt
100 sum1 = sum1 + con(i1, n1) * q(n1)
sum0 = 0.0
do 150 n1 = 1, ncomt
do 150 n2 = 1, ncomt
150 sum0 = sum0 + chi(n1, n2) * phi(n1) * phi(n2)
sum1 = 0.5 * sum1 * sum0
sum2 = 0.0
do 200 n1 = 1, ncomt
200 sum2 = sum2 + con(i1, n1) * alog(xx(n1))
ge = (sum1 + sum2) * r * t
sum3 = 0.0
do 400 n1 = 1, ncomt
400 sum3 = sum3 + con(i1, n1) * a(n1) / b(n1)
am = bm * (sum3 - ge / rlam)
return
end

```

```

subroutine mixb(i1)

```

```

c
c this subroutine executes mixing rule
c for b

```

```

c
common t, p, y(5), ngas, ncomt, vcal(2),
&r, rmw, rn, tc(6), pc(6), omeg(6),
&a(6), b(6), am, bm, alpha(6),
&phi(6), chi(6,6), q(6)
common con(2,6), xx(6)
bm = 0.0
do 100 n1 = 1, ncomt
bm = bm + con(i1, n1) * b(n1)
100 continue
return
end

```

```

subroutine rabrn(i1, j1, dadxi, dbdxi)

```

```

c
c this subroutine calculates the partial
c derivatives for a and b

```

```

c
common t, p, y(5), ngas, ncomt, vcal(2),
&r, rmw, rn, tc(6), pc(6), omeg(6),
&a(6), b(6), am, bm, alpha(6),
&phi(6), chi(6,6), q(6)

```

```

common con(2,6),xx(6)
c
c calculate partial b
c
dbdxi = b(j1)
c
c calculate prtial a
c
rlam = alog(2.0)
sum0 = 0.0
do 100 n1 = 1, ncomt
100 sum0 = sum0 + con(i1, n1) * q(n1)
sum1 = 0.0
do 150 n1 = 1, ncomt
do 150 n2 = 1, ncomt
150 sum1 = sum1 + chi(n1, n2) * phi(n1) * phi(n2)
tum1 = 0.5 * sum0 * sum1
sum2 = 0.0
do 200 n1 = 1, ncomt
200 sum2 = sum2 + con(i1, n1) * alog(xx(n1))
tum2 = sum2
ge = (tum1 + tum2) * r * t
sum3 = 0.0
do 400 n1 = 1, ncomt
400 sum3 = sum3 + con(i1, n1) * a(n1) / b(n1)
term1 = b(j1) * (sum3 - ge / rlam)
term21 = 0.5 * sum1 * q(j1)
sum4 = 0.0
do 450 n1 = 1, ncomt
450 sum4 = sum4 + chi(n1, j1) * xx(j1) * phi(n1)
sum5 = 0.0
do 470 n1 = 1, ncomt
do 470 n2 = 1, ncomt
470 sum5 = sum5 + chi(n1, n2) * phi(n1) * phi(n2) * xx(j1)
term22 = sum0 * (sum4 - sum5)
term23 = alog(xx(j1))
sum6 = 0.0
do 490 n1 = 1, ncomt
490 sum6 = sum6 + con(i1, n1) * xx(j1)
term24 = 1.0 - sum6
dum2 = term21 + term22 + term23 + term24
term2 = bm * (a(j1) / b(j1) - r * t / rlam * dum2)
dadxi = term1 + term2
return
end

subroutine fugac(fu, v, i1, j1, dadxi, dbdxi)

```

```

dimension fu(2,6)
c
c this subroutine calculates the fugacities
c
common t,p,y(5),ngas,ncomt,vcal(2),
&r,rmw,rn,tc(6),pc(6),omeg(6),
&a(6),b(6),am,bm,alpha(6),
&phi(6),chi(6,6),q(6)
common con(2,6),xx(6)
z=abs(p*v/r/t)
vdum=abs(v)
vpb=abs(v+bm)
vmb=abs(v-bm)
term1=log(vdum/vmb)
term2=(dadxi/bm-am*dbdxi/bm/bm)*alog(vdum/vpb)/r/t
term3=dbdxi/vmb
term4=am/bm*dbdxi/vpb/r/t
philn=term1+term2+term3-term4-alog(z)
fu(i1,j1)=exp(philn)
return
end

```

```
#eor
```

```
-----Sample Input-----
```

```

573.2  20.00  t(K) and p(atm) of the system; 2f10.3
5      number of gases; i1
394.77      molecular weight of wax;f10.3
1 0.3000    gas id. and y in mol fraction;i1,2f10.3
2 0.3000
3 0.0500
4 0.3000
5 0.0500
#eof

```

```
----- Sample Output -----
```

```

system temperature in k= 573.20
pressure in atm = 20.00

```

```
wax molecular weight= 394.77
```

```

species 1 gas id= 1
species 2 gas id= 2
species 3 gas id= 3
species 4 gas id= 4
species 5 gas id= 5

```

species	y	calc. x	solubility (mol per kg wax)
1	.3000	.173e-01	.445e-01
2	.3000	.204e-01	.527e-01
3	.0500	.520e-02	.132e-01
4	.3000	.380e-01	.100e+00
5	.0500	.870e-02	.222e-01

wax mole fraction in vapor phase= .165e-02