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# SLURRY FISCHER-TROPSCH/MOBIL TWO-STAGE PROCESS OF CONVERTING SYNGAS TO HIGH-OCTANE GASOLINE. QUARTERLY REPORT, 1 OCTOBER-31 DECEMBER 1982

MOBIL RESEARCH AND DEVELOPMENT CORP. PAULSBORO, NJ

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SLURRY FISCHER-TROPSCH/MOBIL TWO-STAGE PROCESS OF CONVERTING SYNGAS TO HIGH-OCTANE GASOLINE

> QUARTERLY REPORT FOR THE PERIOD 1 OCTOBER - 31 DECEMBER, 1982

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#### I. Abstract

Long-term stability of a Fe/Cu/K2CO3 Fischer-Tropsch catalyst, designated as I-B, was successfully established using the two-stage bench-scale pilot plant. Eighty-six days on stream operation with high gas throughput and conversion have been demonstrated. A total hydrocarbon production of 815 g/g-Fe was obtained. This is substantially greater than the productions reported in the literature. Process variable studies were carried out during the latter part of the run including pressure, superficial feed-gas velocity, feed  $H_2/CO$  ratio, and addition of a potassium-salt. Low  $H_2/CO$  feed gas (0.6 instead of 0.7), which resulted in better usage of the synthesis gas, was used for twenty-eight days. Higher operation pressure (2.5 MPa instead of 1.48 MPa), resulting in even higher gas throughput, was carried out for twenty-one days. No substantial change in the Fischer-Tropsch catalyst aging rate during these studies was observed. The addition of a potassium-salt resulted in an immediate reduction of methane + ethane yield. A second-stage ZSM-5 catalyst (II-B) has accumulated two regenerations and a total on-stream time of eighty-seven days. No apparent loss of activity was observed after regeneration.

The two-stage bench-scale pilot plant was modified to include devices for improving separation of the reactor-wax (heavy hydrocarbons retained in the slurry reactor under the operating conditions) and the Fischer-Tropsch catalyst. These devices are ready for testing.

Detailed analytical procedures and supporting tests for analyzing the hydrogen, the carbon monoxide, the feed-gas, the first-stage Fischer-Tropsch products, and the second-stage ZSM-5 products are summarized. Due to the wide distribution and diversity of the Fischer-Tropsch products, comprehensive analytical schemes were developed with great effort.

Gasoline stability tests for gum formation and oxidation showed that existent (heptane washed) gum contents were all within acceptable limits for conventional gasoline, as were the oxidation stabilities indicated by the induction period method. However, total residues on evaporation (unwashed gum) were generally high, probably due to the presence of a small amount of high-boiling, heptane-soluble hydrocarbons.

#### II. Objective and Scope of the Project

The overall objective of the contract is to develop a two-stage slurry Fischer-Tropsch/ZSM-5 process for direct conversion of synthesis gas, of the type produced in a coal gasification system, to high octane gasoline. The specific objective is to design, construct, and operate a bench-scale pilot plant so that the economic potential of this process concept can be evaluated. To accomplish these objectives, the following specific tasks will be undertaken:

#### Task 1 - Design of Bench-Scale Pilot Plant

A two-stage slurry F-T/ZSM-5 bench-scale pilot plant will be designed for conversion of synthesis gas to high octane gasoline. The slurry F-T reactor will be 5.1 cm diameter and 762 cm high. The fixed-bed ZSM-5 reactor will be 5.1 cm diameter and 10-46 cm high. A distillation column will be designed to obtain stabilized gasoline products.

#### Task 2 - Construction and Shakedown of Pilot Plant

The pilot plant will be constructed in MRDC Paulsboro Laboratory. The unit will be shaken down when completed.

#### Task 3 - Operation of Pilot Plant

At least three slurry F-T catalysts will be tested in the bench-scale pilot plant. One of these catalysts may be provided by DOE's alternate catalyst development projects. The best first-stage catalyst together with a ZSM-5 class zeolite catalyst will be used for process variable studies and catalyst aging tests in the bench-scale unit. Products obtained from the unit will be evaluated to define their gualities.

#### Task 4 - Conceptual Design Study

A preliminary conceptual design of the process will be developed for a commercial size plant for the conversion of synthesis gas to high octane gasoline. Scoping costs of the plant will be estimated.

#### III. Summary of Progress to Date

The third run of the two-stage bench-scale pilot plant, designed as Run CT-256-3, was concluded after eighty-six days on stream. The highlights of the run were:

- o The evaluation of the second Fischer-Tropsch catalyst, a  $Fe/Cu/K_2CO_3$  catalyst designated as I-B, was completed. The catalyst showed excellent stability and high production of hydrocarbons (815 g/gFe).
- o The ranges of the operating conditions for the first-stage slurry Fischer-Tropsch reactor were:

Temperature, °C	259-267
Pressure, MPa	1.13-2.51
H <sub>2</sub> /CO Feed Ratio, Molar	0.6-1.0
Superficial Feed-Gas Velocity, cm/s	1.2-4.4
Space Velocity, NL/gFe-hr	1.3-3.4
Catalyst Loading, wt % (nominal)	11-20

The  $H_2$  + CO conversion ranged from 54 to 93 mol % and the methane + ethane yield from 6 to 18 wt % of the total hydrocarbons produced.

 The second-stage ZSM-5 catalyst (II-B) has accumulated two regenerations and a total on-stream time of eighty-seven days. No apparent loss of activity was observed after each regeneration. The ranges of the operating conditions for the second-stage reactor during this run were:

Temperature, Inlet, °C 288-466 GHSV, 1/hr 1,350-4,580

An average of 3.3 °C/day increase on the inlet temperature was used to maintain a constant operating severity. Maximum gasoline yield was obtained at an operating severity index, expressed as the molar ratio of  $i-C_4/(C_3^{\pm}+C_4^{\pm})$  in the product, of 0.5-1.0. Peak research octane numbers of 90-94 were achieved at severity indexes of 0.3-2.0.

 During a major interruption in Fischer-Tropsch synthesis operation, the Fischer-Tropsch catalyst slurry was unloaded and then reloaded into the slurry reactor. Substantial deterioration of the catalyst activity and substantial increase in methane + ethane yield were observed, probably due to catalyst damage incurred when it was exposed to air.

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- The Fischer-Tropsch reactor-wax yield increased significantly with decreasing methane + ethane yield.
- o Process variable studies were conducted during the latter part of the run. The results are:
  - Use of 0.6  $H_2/CO$  feed ratio synthesis gas instead of 0.7 lowered the methane + ethane yield. It also gave better usage of the synthesis gas since the  $H_2/CO$  usage ratio was nearly 0.6
  - Higher operating pressure at the same superficial feed-gas velocity in the slurry reactor resulted in slightly lower H<sub>2</sub>/CO conversion, but a significantly lower methane + ethane yield. It also increased the oxygenate yield significantly.
  - Addition of a potassium-salt to the slurry reactor drastically decreased the methane + ethane yield.
- o A "hydrodynamic upset" of the slurry reactor was observed at the end of the run, probably due to catalyst settling. This resulted in low  $H_2$ +CO conversion and lower temperature at the upper portion of the reactor. The upset disappeared after eight hours of high gas velocity operation, but reappeared after the velocity was reduced. Low velocity operation of the slurry reactor may not be desirable.

After the end of Run CT-256-3, the two-stage bench-scale pilot plant was shut down for modifications. The major modification was addition of two external filter assemblies to withdraw reactor-wax from the first-stage slurry reactor. Other modifications included installation and replacement of many filter elements, installation of pressure transducers and heating tape, and replacement of the flange gaskets for the first-stage reactor.

Analytical procedures and supporting tests for the current pilot plant operation are summarized. The streams that required analysis include carbon monoxide, hydrogen, the combined feed gas, the first-stage Fischer-Tropsch products (gaseous, aqueous, liquid hydrocarbon, and reactor-wax phases), and the second-stage ZSM-5 products (gaseous, aqueous, and liquid hydrocarbon phases). The gaseous streams are analyzed on-line using a Mobil-developed automated gas chromatographic (GC) system. The non-acid and acid oxygenates in the Fischer-Tropsch aquects phase are determined by fused silica capillary column GC (FS-GC) and ion chromatography, respectively. Oxygenates in the Fischer-Tropsch liquid hydrocarbon phase are determined using GC analysis. The Fischer-Tropsch liquid hydrocarbon phase is analyzed using various GC and liquid chromatography (LC) techniques, while its carbon number distribution is routinely obtained using an FS-GC capillary column. More detailed analysis of the Fischer-Tropsch liquid hydrocarbon is performed on selected cases using distillation, oxygenate separation, and a GC equipped with an olefin scrubber and flame ionization detectors. The analysis of Fischer-Tropsch reactor-wax is done using FS-GC, LC, and solvent extraction. The second-stage liquid hydrocarbon product is analyzed using a GC system and an olefin scrubber. Various supporting tests are also employed, including acid number, bromine number, hydroxyl number, simulated distillation, kinematic viscosity, surface tension, vacuum distillation, and octane numbers.

Gasoline stability tests for gum formation and oxidation showed that existent (heptane washed) gum contents were all within acceptable limits, as were the oxidation stabilities. However, total residues on evaporation (unwashed gum) were generally high, probably due to the presence of a small amount of high-boiling, heptane-soluble hydrocarbons.

The task on "Conceptual Design Study" has been initiated. The process design basis was constructed and the conceptual design work is in progress.

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#### IV. Detailed Description of Technical Progress

A. <u>Task 3 - Operation of the</u> Pilot Plant

#### 1. Run CT-256-3 - Conclusion

The third BSU run, designated as Run CT-256-3, using Catalyst I-B (containing Fe/Cu/K<sub>2</sub>CO<sub>3</sub>) in the first-stage bubble-column F-T reactor and Catalyst II-B (a ZSM-5 class catalyst) in a second-stage fixed-bed reactor, was smoothly started up on July 27, 1982. The major objectives of the runs were to evaluate the performance of the F-T catalyst I-B over an extended period of time and then to perform process variable studies.

Major highlights from this run were:

 Smooth operation of the slurry F-T reactor with high catalyst loading (19.5 wt % initially), high synthesis gas throughput, and high conversion, was demonstrated over a period of eighty-six days.

The ranges of the first-stage slurry F-T reactor operating conditions and performance were:

H <sub>2</sub> +CO flow rate, Nm <sup>3</sup> /hr	1.0-2.6
Temperature, °C	259-267
Pressure, MPa	1.13-2.51
H <sub>2</sub> /CO feed molar ratio	0.6-1.0
Superficial feed-gas velocity, cm/s	1.2-4.4
Space velocity, NL/gFe-hr	1.3-3.4
Catalyst Loading, wt % (nominal)	11-20
•	

H2+CO conversion, mol %54-93Methane + ethane yield, wt % HC6-18Hydrocarbon production, gHC/gFe815

- The Catalyst I-B is an excellent catalyst from a stability point of view.
- o The ranges of the second-stage fixed-bed reactor operating conditions were:

 Temperature inlet, °C
 288-466

 GHSV, l/hr
 1,350-4,580

This catalyst performed satisfactorily in converting the first-stage F-T products into high octane gasoline.

- o There were two small and one large interruption in the Fischer-Tropsch synthesis operation. A slight loss in Fischer-Tropsch catalyst activity and a slight increase in methane yield were observed during the small interruptions. The major interruption took place at sixty-one DOS due to a leak at the bottom flange of the slurry reactor. The slurry was unloaded and reloaded into the reactor after a new gasket was installed. Substantial deterioration of the catalyst activity and substantial increase in the methane + ethane yield were observed. The F-T catalyst seems to be very sensitive to exposure to the air.
- The reactor-wax yield increased significantly with decreasing methane + ethane yield.
- A  $H_2/C0$  feed ratio of 0.6 (instead of 0.7) was used for twenty-six days with no significant effect on the F-T catalyst stability. Lower methane + ethane yield was observed during this time. The  $H_2/C0$  usage ratio is very close to 0.6 as indicated by the fact the exit  $H_2/C0$  ratio remained at nearly 0.6 over a wide range of conversion. The usage of the synthesis gas is better at 0.6 feed-gas  $H_2/C0$  ratio.
- Higher operating pressure with constant superficial feed-gas velocity in the slurry F-T reactor resulted in a slightly lower H<sub>2</sub>+CO conversion, but a significant decrease in methane + ethane yield (from 11 wt % to 9% when pressure increased from 1.48 MPa to 2.51 MPa). The oxygenate yield also increased significantly over the same pressure range.
- Addition of a potassium-salt to the slurry reactor drastically decreased the methane + ethane yield from 13 wt % to 8% with little change on synthesis gas conversion. Unfortunately, no conclusions on catalyst stability could be drawn.
- o The gas holdup data in the slurry bubble-column were estimated using a system of DP-cell legs. There was no significant change of gas holdup profiles in the first seventy-five DOS. Catalyst concentration profiles along the bubble-column were also measured and found to follow profiles predicted by a published mathematical model on slurry settling.
- A "hydrodynamic upset" of the slurry reactor occurred at eighty-two DOS, probably due to catalyst settling, resulting in a low H<sub>2</sub>+CO conversion and a 5°C lower temperature at the upper portion of the reactor. The

upset disappeared after eight hours of high gas velocity operation, but reappeared after the velocity was lowered.

o A second-stage ZSM-5 reactor operating severity index, expressed as the  $i-C_4/(C_3^- + C_4^-)$  molar ratio in the product, of 0.5-1.0 gave maximum gasoline yield. Higher pressure operation had no significant effect on the second-stage operation and yield. Peak research octane numbers of 90-94 were obtained for the raw gasoline at severity indexes of 0.3-2.0.

a. First-Stage Fischer-Tropsch Reactor Operation

The last Quarterly Report (July-September, 1982) contained details of the start-up and pretreatment of the third BSU run, designated as Run CT-256-3. These will not be repeated here. A brief discussion of the synthesis operation was also given. A detailed description and discussion of the entire run is given here. The run lasted eighty-six days and was very successful, providing high conversion at high synthesis gas throughput for a long period of time, as well as a great deal of new information on slurry F-T reactor performance. The F-T catalyst used in this run was Catalyst I-B, the same Fe/Cu/K<sub>2</sub>CO<sub>3</sub> catalyst used in run CT-256-2.

Immediately after the catalyst pretreatment, at which point the carbon monoxide conversion had reached 82%, the slurry reactor temperature was lowered to  $260\,^\circ$ C in steps of  $3\,^\circ$ C over a thirty-seven hour period. With each drop in temperature, the conversion first declined, then gradually increased back to the original conversion level. This policy of temperature reduction kept the conversion high during this transition period. The pressure was increased to 1.48 MPa at this time, establishing the conditions which were used for the majority of the run. This brought the carbon monoxide conversion to about 90%, which was one of the objectives of the run. Other priorities were long-term stability and process variable studies. These were done, including operating at 2.52 MPa (350 psig) and also using a H<sub>2</sub>/CO molar feed ratio of 0.6 for extended periods.

Figure 1 shows the conversion and methane + ethane selectivities, as well as the temperature pressure, and superficial gas velocity for the entire run. The range of synthesis conditions and performance of the first stage F-T reactor were:

$H_2$ + CO flow rate, Nm <sup>3</sup> /hr	1.0-2.6
H <sub>2</sub> /CO feed ratio, molar	0.6-1.0
Superficial feed-gas vel., cm/s	1.2-4.4
Space velocity, NL/gFe-hr	1.3-3.4

Temperature, °C	259-263
Pressure, MPa	1.13-2.5
H <sub>2</sub> + CO conversion, mol %	54-93
Methane + Ethane yield, wt 💈	6-18

The run can be roughly divided into two parts. In the first part, a long-term aging study on the Catalyst I-B was carried out. After the long-term stability of the synthesis operation was well established, a period of process variable studies was commenced at sixty-one days on-stream. The process variables examined included:

- o Superficial feed-gas velocity
- o Reactor pressure
- o Feed H<sub>2</sub>/CO ratio
- o Addition of a potassium-salt

The range of synthesis conditions mentioned above includes the period of process variable studies.

The synthesis operation was interrupted three times during the run. The first interruption was due to a false alarm and lasted nine hours, during which time the unit was purged with nitrogen. The second interruption was caused by a small slurry leak at the 305 cm flange. Tightening the flange stopped the leak, and synthesis gas flow was restored after thirty-six hours. Both of these interruptions caused a slight decrease in conversion and a corresponding increase in methane + ethane selectivity. This type of behavior has been observed before. The third interruption was another slurry leak, this time at the bottom flange. The leak could not be stopped by merely tightening the flange, so the slurry had to be removed while the gasket was replaced. The slurry was then reloaded after fifty hours and synthesis continued. The result was a substantial increase in the methane + ethane selectivity (from 13 to 19%) and a substantial decrease in  $H_2$  + CO conversion (from 87 to 70%) which rose over a five-day period to 80%, aided by a 2°C temperature increase. It appears that exposure to air is detrimental to catalyst performance.

At eighty-one DOS, an amount of potassium-salt was added to the slurry reactor through the catalyst slurry loading pot in an effort to lower the methane + ethane yield. It is clear from Figure 1 that this did occur, the methane + ethane selectivity dropping from 13 to 8% with little change on synthesis gas conversion. Unfortunately, however, an unusual upset which occurred about twelve hours after the potassium-salt

addition negated any improvement that addition of the potassium-salt might have imparted to the  $H_2$  + CO conversion. Addition of a potassium salt to a slurry F-T reactor has been reported by Koelbel and Ralek (1980) previously.

This "Hydrodynamic Upset" took place following a process-variable study, the variation of the superficial gas velocity. The velocity had been brought down as low as 1.1 cm/s for the study, after which it was reestablished at 2.6 cm/s. The potassium-salt was then added, and the conversion then dropped rapidly from over 80% to 55%, while the reactor temperature above the 305 cm level was 5°C lower than that below 305 cm. It was suspected that the catalyst had settled during the low-velocity operation. Therefore, at eighty-two DOS, the reactor pressure was dropped in stages to 1.48 MPa (200 psig) in order to increase the linear superficial gas velocity in the slurry reactor. This higher gas velocity reestablished uniform reactor temperature after three days. At that time, the synthesis gas conversion also rose to 78%. The superficial feed-gas velocity at this point was 4.2 cm/s. However, when the pressure was increased back to 2.51 MPa (350 psig) the next day, the temperature discrepancies returned with conversion slowly dropping back to the 55% level. It was this failure to reestablish the high conversion after this upset which caused the ending of run CT-256-3. The total hydrocarbon production for the run was 815 gHC/gFe, a significantly high figure.

Material balances were performed daily. Tables A-1 and A-2 summarize the operating conditions and results for this run. Detailed analytical breakdowns of the first-stage product were also performed. Table A-3 shows the composition of the hydrocarbon products produced by the Fischer-Tropsch catalyst for several balances. The oxygenated products were broken down separately and are displayed in Table A-4. In addition, the oxygenates contained in the aqueous phase were analyzed, as shown in Table A-5. The reactor-wax, which was removed by filtration, was broken down on the basis of carbon number. Table A-6 tabulates the results of these studies, while Figures 2 and 3 are It is graphical representations of some of these distributions. interestsing to note that between six and twenty-one DOS the distribution had reached a steady-state at 1.48 MPa reactor pressure with peak carbon numbers of 27-28 and similar average carbon numbers. The initial wax medium obviously contained some heavier components as indicated by a peak carbon number of 35 at six DOS. At higher reactor pressures (2.17 MPa and higher), however, slightly heavier hydrocarbons are retained in the slurry reactor as shown in Figure 3. The shift is small with a peak carbon number about 30-31 and an average carbon-numbers of about 28-29.

Reactor-wax was removed regularly to keep the level in the slurry reactor at 610-670 cm. By using DP-cell readings to determine the slurry inventory in the reactor at any given time, a cumulative reactor-wax production plot was constructed This plot is very smooth up to sixty-one DOS as (Figure 4). indicated by the least-squares-fitted curve included in Figure 4. The reactor-wax production rate at any given time was estimated using the corresponding slope of the curve at the given time. In this way, the reactor-wax yields as a percentage of the total hydrocarbon yield were calculated and included in Tables A-1 and A-2. The reactor-wax production data between sixty-one and seventy-four DOS were out-of-line because the major operational upset occurred at sixty-one DOS. During that time, the slurry was unloaded from and reloaded into the reactor. The reactor-wax yields up to sixty-one DOS are plotted against the methane + ethane selectivities in Figure 5. It is obvious from this plot that at lower methane + ethane yields, the reactor-wax yield increases significantly.

Inclusion of all hydrocarbons and oxygenates from a total material balance allows the construction of a Schultz-Flory type plot (Flory, 1967). One of these, from eleven DOS, is shown in Figure 6. There is a distinct change in the slope of the distribution from  $\alpha$  (probability of chain growth) = 0.79 to 0.88 at carbon number twenty-two, coinciding approximately with the inclusion of the reactor-wax. This trend resembles that of a similar plot given for run CT-256-1 (see Figure 2 of April-June 1982 Quarterly Report). As explained earlier, this phenomena may be due to the fact that in a slurry system large molecules can re-entrain themselves onto active cr alyst sites, allowing for further chain growth.

As mentioned earlier, process variable studies were carried out in the slurry reactor during the run. Variables studied included superficial gas velocity, pressure, feed  $H_2/CO$ ratio, and addition of a potassium-salt. The results of the first three studies wil now be presented.

As the superficial gas velocity is lowered, the residence time of the synthesis gas is increased, which should obviously lead to an increased synthesis gas conversion. Table 3 illustrates this point, showing three different space velocities and the results. It is seen that in addition to the conversion, the methane yield and the exit  $H_2/CO$  ratio show definite trends. The methane yield goes down slightly. At higher conversions more water is produced by the Fischer-Tropsch reaction, which in turn causes the water-gas shift reaction to produce more hydrogen. This results in a higher  $H_2/CO$  ratio in the exit gas.

In another study, the  $H_2/CO$  ratio in the feed gas was changed from 0.7 to 0.6 to observe its effect on slurry reactor This was done at sixty-five DOS and continued until performance. the end of the run. At the time of the switch, however, the catalyst was in the process of recovering from the third upset, which had occurred at sixty-one DOS. It has already been seen that following an upset such as this, the conversion is initially low, then climbs steadily to a point slightly lower than that before the upset. Similarly, the methane + ethane selectivity is initially high, but then declines gradually to a level somewhat above the previous one. This seems to correspond to a "reactivation" of the catalyst. Looking at Figure 1, this same trend is evident following the upset, but it appears that the methane + ethane selectivity declines to the same level it was before the upset. It is logical to conclude that the lower  $H_2/CO$ ratio in the feed enabled this to occur. By decreasing the amount of available hydrogen, the yields of hydrogen-rich components, such as methane and ethane, were decreased.

Also interesting is the effect of the feed  $H_2/CO$  ratio on the  $H_2/CO$  ratio in the exit gas shown in Figure 7. Data from the whole run are used to construct this plot. Consequently, they include wide ranges of operational variables and result in a large spread of the data. The lines shown on the figure are least-squares-fitted lines. Here it is seen that a feed ratio of 0.6 causes the exit ratio to remain at nearly that same ratio over a wide range of conversions. At 0.7  $H_2/CO$  feed ratio, however, an excess of hydrogen appears at all times, particularly at the higher conversions. This figure indicates that the  $H_2/CO$ usage ratio is very close to 0.6. When a feed  $H_2/CO$  ratio higher than the usage ratio is used, the excess hydrogen is reflected as higher  $H_2/CO$  in the exit gas. The exit  $H_2/CO$  ratio increases with increasing synthesis gas conversion because the water-gas shift reaction favors the formation of hydrogen according to thermodynamic equilibrium. It appears, then, that there are distinct advantages to operate the synthesis at a feed  $H_2/CO$ ratio close to the usage ratio, i.e., lower methane + ethane yield and better usage of the synthesis gas. However, the long-term effects on catalyst aging have yet to be determined. Also, most of the 0.6  $H_2/CO$  data were taken when the operating pressure was higher than it was for the 0.7 data, so further study is needed.

Pressure effects were studied at a feed  $H_2/CO$  ratio of 0.6 and a temperature of 267°C. As the pressure was changed, the synthesis gas flow rate was altered so that the same superficial feed-gas velocity was maintained, i.e., 2.5 cm/s. Table 4 shows the results of this process variable study. The pressure was varied from 1.48 to 2.51 MPa (200 to 350 psig) over an eight-day period. It is seen that the methane and ethane yields decrease as the pressure is raised. This is due to the higher probability

of the chain-growth under higher pressure. The effect on synthesis gas conversion, though, is less clear. The conversion drops slightly with increasing pressure although the superficial feed-gas velocity is maintained constant. The 2% drop in conversion between 2.17 and 2.51 MPa may be mainly due to experimental data scattering. However, the space velocities were greatly increased due to higher pressure operation. This increase in the space velocity is more than sufficient to compensate for the slightly lower conversion at the same feed-gas velocity. Of course, the other major advantage is the significant drop in the methane yield.

The effect of pressure on the yield of oxygenated products was also noted, as shown in Table 5. This shows that as the pressure is increased, so is the oxygenate yield. The magnitude of the increase is not clear from this table, however, because the other process variables were also changed  $(H_2/CO)$ ratio, superficial velocity, and temperature).

Also studied during this run was the gas holdup in the slurry reactor. This was done by using a system of DP-cell legs spaced along the bubble-column reactor. (See April-June 1981 Quarterly Report for details on this setup.) This resulted in estimated holdups for sections of the column between any two legs. Table 6 summarizes overall gas holdups taken from different times on-stream. It can be seen from this table that the holdup did not change very much over the first seventy-five days on-stream. However, the data from seventy-nine to eighty-one DOS show a hysteresis effect of the gas holdup. That is, once the velocity was dropped to 1.1 cm/s, the gas holdup did not respond instantly with raising the velocity. This may have been due to catalyst settling at the low velocity as described earlier, and difficulty in re-entraining the catalyst at the higher velocity. Figure 8 illustrates the typical gas holdup profiles along the bubble-column reactor. This profile is similar to that reported by Langemann and Koelbel (1967) in cold-flow bubble-columns.

The gas holdup near the top of the column is high, due to the fact that the bubbles have to disengage from the slurry, i.e., an end effect. At the bottom of the column there is a short zone where the gas holdup changes dynamically with distance. This arises from the bubble dispersion, formation, and coalescence. After that the holdup decreases, probably due to the fact that the gas volume contracts as the reaction proceeds. Two separate profiles show that the gas holdup is similar at the beginning and the end of the run, with absolute differences due to the change in the gas velocity. Finally, catalyst concentration profiles were obtained by taking slurry samples from several fixed locations of the bubble-column reactor and then by burning off the wax from the samples. The solids concentration was then plotted in semi-log fashion in Figure 9 for different days on-stream, corresponding to different gas velocities. The straight lines shown by this plot indicates that the trend of the catalyst concentration profile follows very well with the established solid settling mathematical model in bubble-columns (Kato, et al., 1982). Highlights of this figure are:

- Increased velocity decreases the catalyst settling, so a flatter profile is achieved.
- The profile is steeper during the hydrodynamics upset at eighty-two DOS, indicating increased catalyst settling. The profile after the upset is slightly flatter.

Unfortunately, no meaningful catalyst concentration profile data were available between ten and eighty-two DOS. It also is not clear if the larger concentration profile at the end of the run could be completely attributed to lower gas velocity.

#### b. Second-Stage Fixed-Bed ZSM-5 Reactor Operation

In the last Quarterly Report, it was reported that a second-stage reactor, containing 215 g of II-B ZSM-5 catalyst, was brought on-stream one hour after the end of the F-T catalyst pretreatment. The material balances performed over the sixty-seven days of second-stage operation are summarized in Table A-7 of Appendix A. The properties of the raw liquid hydrocarbons collected from the ambient and chilled condensers are reported in Table A-8, while Table A-9 gives the detailed product hydrocarbon compositions.

Also, as mentioned in previous Quarterly Reports, the severity of the second-stage operation was guided by the  $i-C_4/(C_3^-+C_4^-)$  molar ratio (the severity index) in the combined gas stream after the second-stage reactor. To achieve a severity index of 0.8-1.0, the initial inlet temperature of the second-stage reactor was set to 343°C in continuation of a cycle started in run CT-256-2 (see July-September, 1982 Quarterly Report). In the current run, the catalyst was regenerated twice with no apparent loss of activity. During the twenty-five day second cycle and the unfinished thirty-two day third cycle operation, the second-stage inlet temperature had to be increased at about 3.3 °C/day to maintain the target severity. The fixed-bed inlet temperature and the temperature rise across the catalyst bed are reported in Figure 10.

Figure 11 shows the effect of second-stage operating severity index on the hydrocarbon yield. As in the July-September, 1982 Quarterly Report, the alkylate yield is estimated by alkylating first butenes and then propylene with i-butanes. If there is an excess of light olefins, they are converted to "Cat-Poly gasoline" using conventional catalytic polymerization process. In making this plot the product yields are renormalized after excluding the components that either bypass or are nonreactive to the ZSM-5 catalyst, i.e., reactor-wax and  $C_4$  paraffins. Peak  $C_5^+$  gasoline yields of 85-90 wt % could be achieved when a severity index was maintained at 0.5-1. High pressure operation (2.51 MPa versus 1.48 MPa) had no significant effect on the second-stage operation and yield.

Figure 12 shows the Research Octane Number, and the aromatic and olefin content of the raw liquid hydrocarbon product collected in the cold and chilled condenser of the pilot plant, as a function of the second-stage operating severity index. Peak octane number of 90-94 is obtained for severity indexes of 0.3-2.0. The corresponding aromatics content is 30-50 wt % and olefins content is 5-25 wt %. Too high aromatics content is equivalent to high severity or reduced gasoline yield. Too low aromatics content, however, also results in a low gasoline yield. Hence, optimal gasoline yield is restricted to a severity index of 0.5-1.0. Motor octane number for the raw liquid hydrocarbon products are summarized in Table A-8.

Another indication of the second-stage catalyst performance is the acid number of the raw liquid hydrocarbon products, also reported in Table A-8. Acid numbers of 0.04-0.4 show the ability of the ZSM-5 catalyst to convert organic acids. For comparison, first-stage F-T liquid hydrocarbon products have acid number of 1.1-3.0. Finally, the ASTM distillation properties of the raw hydrocarbon products did not vary much with catalyst aging or changing operating severity.

#### 2. Bench-Scale Unit Modifications

After the end of the third run the BSU was shut down for modifications. The major modification was addition of two external filter assemblies to withdraw reactor-wax from the first-stage reactor. The assemblies were installed to withdraw wax from 157 and 762 cm above the distributor.

A schematic of the external filter assembly is shown in Figure 13. The slurry from the reactor is brought into the filter vessel, maintained at 204-260°C, via valve V-1. The catalyst settling in the vessel is prevented by continuous

agitation provided by the stirrer. The slurry can be purged with hydrogen to remove dissolved carbon monoxide, carbon dioxide, and water from the slurry. The filtered reactor-wax can be collected in the wax receiver maintained at a certain pressure depending upon the differential pressure desired across the filter. The concentrated slurry deposited in the filter vessel can be frequently flushed back into the reactor by pressuring the filter vessel through the wax receiver. If necessary, the filter vessel and filter surface can be flushed with hot solvent using the solvent pot and the contents can then be pushed back into the reactor. The size of the filter is 1.59 cm OD x 12 cm long with ten micron pore size filter element.

A trial operation of the filter assembly was, however, unsuccessful. The reactor-wax withdrawal rate was lower than expected and the filter element got plugged after withdrawing about 200 g of reactor-wax. The solvent cleaning of the filter or the filter blowback did not significantly improve the operation. The filter vessel was then modified to carry out separation of the catalyst from the slurry by catalyst settling and was very successful. This method will be described in full details in the next Quarterly Report.

Other minor modifications included:

- The ten micron filter element at 305 cm level of the first-stage reactor was replaced with a new element. The older element was in use for 106 days (Runs CT-256-2 and -3).
- Another identical filter (ten micron, 1.59 cm OD x 30.5 cm long) was installed inside the slurry reactor at 762 cm level to provide additional wax withdrawal capability.
- 3. The two micron filter (1.25 cm OD x 12.7 cm long), inserted into the slurry reactor from the side tube at 458 cm level, was replaced with a five micron filter to improve the filtration rate.
- A new ten micron filter (1.25 cm OD x 12.7 cm long) was inserted into the slurry reactor from the side tube at 610 cm level.
- 5. All gaskets between flanges of the first-stage reactor were replaced with new "Graphoil"(1) (0.3175 cm thick) gaskets. The old gaskets were made of "Bimetallic" (1) material and were found to split and leak during a run.

(1)A registered trade mark.

Page 16

- 6. Ten pressure transducers were installed to record unit pressures on the datalogger computer.
- 7. Additional heating tape was added on the conical part of the disengager bottom and the flange at 762 cm level of the slurry reactor. Previously only one tape was used for this section. Additional heating of this section should minimize heat loss.

#### 3. Analytical Procedures and Supporting Tests

The streams that require analysis include carbon monoxide, hydrogen, the combined feed gas, the first-stage F-T products (gaseous, aqueous, liquid hydrocarbon, and reactor-wax phases), and the second-stage products (gaseous, aqueous, and liquid hydrocarbon phases). The analytical procedures and supporting tests for these streams are summarized in this subsection.

#### a. Carbon Monoxide, Hydrogen, and Combined Feed-Gas

These streams are analyzed on-line using a Mobil-developed automated gas chromatographic (GC) system. The same system is used for on-line analysis of the gaseous product streams from both the first-stage and the second-stage reactors.

b. First-Stage Fischer-Tropsch Products

The analysis of the F-T products are very complicated because of the wide boiling range and the diversity of the product components. Figure 14 summarizes the analytical scheme.

The combined gas stream from the cold and chilled condenser are analyzed on-line using the on-line GC system. The stream contains N<sub>2</sub>, H<sub>2</sub>, CO, CO<sub>2</sub>, H<sub>2</sub>O, and hydrocarbons. The amount of hydrocarbons heavier than C<sub>8</sub> is insignificant in this stream. A typical GC plot for such a sample is given in (A) of Figure 15.

Non-acidic oxygenates in the aqueous phase are determined by fused silica capillary column GC (FS-GC) as indicated in Figure 16. Acids are determined by ion chromatography. The major oxygenates in the aqueous phase identified by gas chromatography-mass spectrometry (GC-MS) are summarized in Table 7. Normally, the  $C_1-C_8$  linear alcohols are the major components with smaller amounts of ketones, acids, and mixed esters.

The  $C_5^+$  oxygenates present in the liquid-hydrocarbon phase are determined as previously reported by Di Sanzo (1981). Recently the gas chromatographic analysis has been improved by employing fused silica capillary columns and cool on-column splitless injections. Figures 17 and 18 represent typical gas chromatograms of  $C_6^+$  alcohols and acids, and  $C_5^+$  ketones and esters, respectively, isolated from the hydrocarbon phase by liquid chromatography (LC). Normally, linear alcohols are the major oxygenates, followed by methylketones. Mixed esters as a result of secondary reactions between the various acids (acetic, propanoic, butyric, etc.) and alcohols are also present. The  $C_5^+$ acid content has been determined to be generally low (-0.1% wt of hydrocarbon phase). With these methods, oxygenates up to  $C_{40}$  can readily be determined.

The  $C_1$  to  $C_5$  oxygenates present in hydrocarbon phase are determined by aqueous extraction of the hydrocarbon phase followed by gas chromatographic analysis of the aqueous extract.

An optional analysis for the aqueous F-T product is Acid Number.

The liquid hydrocarbon phase contains mainly olefins and paraffins. The olefinic and paraffinic compositional analysis is carried out by various GC and LC techniques. Carbon number distribution is readily obtained (<1 hour) by a single injection onto an FS-GC capillary column. A typical chromatogram is given in Figure 19. This simple technique will indicate how the major olefins, i.e., linear  $\alpha$ -olefins and cis, trans-2 linear olefins, vary with respect to the normal paraffins. Significant changes in the reactor hydrocarbon composition is thus readily detected.

A more detailed analysis is also performed on selected material balances. A liquid hydrocarbon sample is separated into two fractions distilled at 196°C B.P. (about  $C_{11}$ ) by distillation. The  $C_{11}$  fraction after removing the oxygenates by passage through a silica gel SepPak<sup>(1)</sup> (Waters Associates, Milford, Mass.) is injected into a gas chromatograph equipped with an olefin scrubber and two flame ionization detectors. By comparing detector signals total olefins/paraffins can be determined. In addition, major components are identified. Figure 20 shows typical chromatograms of these analyses. The  $C_{12}$  olefins/paraffins are determined as described by Di Sanzo (1981).

(1)A registered trade mark.

Other analyses for the first-stage liquid hydrocarbon product include Acid Number, Bromine Number, Hydroxyl Number, and Simulated Distillation.

The reactor-wax withdrawn from the slurry reactor consists of components with a carbon number distribution from about  $C_8$  to  $C_{70}^{-1}$ . A gas chromatographic technique has been developed employing a short (8 m) fused silica capillary column. The reactor-wax (catalyst-free) after being dissolved in hot toluene is injected into the capillary column by the cool on-column injection technique. The latter injection technique minimizes discrimination for the high boiling components. A sample chromatogram is given in Figure 21 for a F-T wax sample employed as a start-up medium in the operation of the two-stage bench-scale pilot plant.

A LC method has also been developed for the determination of oxygenates in the reactor-wax. Total oxygenates is obtained by weighing the isolated oxygenate fraction after solvent evaporation. Carbon number distribution of the oxygenates is then obtained by FS-GC with cool on-column injection. Alcohols and ketones (major oxygenates) can be distinguished chromatographically.

Other analyses for the F-T reactor-wax are Kinematic Viscosity, Surface Tension, Vacuum and Analytical Distillation, and Specific Gravity.

#### c. Second-Stage ZSM-5 Products

The products from the second-stage ZSM-5 reactor are separated into three streams, i.e., gaseous, aqueous, and liquid hydrocarbons. Analysis of the gaseous product is similar to that of the first-stage gaseous product using an on-line GC system. A typical chromatogram is included as (B) of Figure 15. The aqueous stream from the ZSM-5 reactor contains insignificant amounts of oxygenates and only its pH values are occasionally measured.

Analyses of the liquid hydrocarbon product from the ZSM-5 reactor is more complex. A three-column GC system and an olefan scrubber are used. This setup is similar to a system employed for the Methanol-to-Gasoline process (Bloch, et al., 1977). The system is highly automated and can identify individual components up to approximately  $C_{10}$ .

The small amount of components boiling above 204°C are determined by capillary column GC and identified by GC-MS. Finally, a LC method has been developed for the determination of trace oxygenates which may be present.

Other analyses employed for the second-stage liquid hydrocarbon product include Research and Motor Octane Numbers, and Acid Number.

#### 4. Product Evaluation

Two raw gasoline product samples, taken at sixty-six and seventy-eight DOS from Run CT-256-3, were tested for existent and total gums (ASTM D381) and oxidation stability (ASTM D525).

Metal deactivator at 0.5 lb per 1000 bbl. and antioxidant at 2.5 lb per 1000 bbl. were used in one portion and the antioxidant level was increased to 15 lb per 1000 bbl. in a second portion. As shown in Table 8, existent (heptane-washed) gum contents of 1 to 4 mg/100 ml were found in all samples, thus meeting the 5 mg/100 ml maximum specification of ASTM D439 for automotive gasolines. However, total residues on evaporation were very high (ranging up to 119 mg/100 ml) in several of the tests, indicating the presence of high-boiling, heptane-soluble materials. The higher antioxidant usage rates were only partially effective for reducing the total residue levels. The high-boiling materials were confirmed by subsequent ASTM D86 distillations, in which end points up to 249°C were measured (225°C is a typical end-point specification maximum for U.S. gasolines). The drastic difference in the total gums for the two samples may be due to operating condition differences in the second-stage reactor.

The ASTM D525 procedure provides an indication of gasoline tendencies to react with oxygen to form gum during storage. ASTM D439 specifications require 240 minutes or more for the stability period in this test; time periods of 620 to 825 minutes were obtained for these samples, indicating acceptable performance.

Standard N.A.C.E. (National Association of Corrosion Engineers) corrosion tests were conducted on a water-washed composite raw gasoline sample (ASTM D974 total acid number of 0.05 mg KOH/g) of Run CT-256-3 to determine the corrosion tendencies. The N.A.C.E. test method involves contacting a cylindrical steel specimen with a constantly stirred mixture of 91% distilled water (maintained at 38°C) for a period of 3-1/2 hours. Performance is expressed by a scale dependent on a visual observation of the rust on the steel specimen surface.

Metal deactivator at 1 lb. per 1000 bbl. and two different antioxidants at 10 lb. per 1000 bbl. were added to two separate portions of the sample. The N.A.C.E. corrosion tests were conducted on duplicate samples from these two portions and on an additized petroleum-sourced unleaded gasoline. Tests also

were run on these fuels with a commercial corrosion inhibitor at a conventional level of 2.1 lb. per 1000 bbl. The steel specimen surfaces were severely rusted in tests of fuel samples without corrosion inhibitor. Testing of the fuels containing the corrosion inhibitor, on the other hand, indicated satisfactory protection, showing zero to less than 0.1% (2 or 3 spots of no more than 1 mm diameter) of the surface rusted. Based on these results, it is concluded that the use of a commercial corrosion inhibitor will satisfactorily control the fuel's corrosion tendencies.

Long-term 43°C storage stability testing of this additized, water-washed composite product is in progress. Results will be reported in our next quarterly summary.

#### 5. Conclusions

The third run of the two-stage bench-scale pilot plant, designed as Run CT-256-3, was concluded after eighty-six days on stream. The highlights of the run were:

- o The evaluation of the second Fischer-Tropsch catalyst, a  $Fe/Cu/K_2CO_3$  catalyst designated as I-B, was completed. The catalyst showed excellent stability and high production of hydrocarbons (815 g/gFe).
- The ranges of the operating conditions for the first-stage slurry Fischer-Tropsch reactor were:.

Temperature, °C	259-267
Pressure MPa	1.13-2.51
H <sub>2</sub> /CO Feed Ratio, Molar	0.6-1.0
Superficial Feed-Gas Velocity, cm/s	1.2-4.4
Space Velocity, NL/gFe-hr	1.3-3.4
Catalyst Loading, wt % (nominal)	11-20

The  $H_2+CO$  conversion ranged from 54 to 93 mol % and the methane + ethane yield from 6 to 18 wt % of the total hydrocarbons produced.

 The second-stage ZSM-5 catalyst (II-B) has accumulated two regenerations and a total on-stream time of eighty-seven days. No apparent loss of activity was observed after each regeneration. The range of the operating conditions for the second-stage reactor during this run were:

Temperature,	Inlet,	°C	288-466
GHSV, l/hr			1,350-4,580

An average of 3.3 °C/day increase on the inlet temperature was used to maintain a constant operating severity. Maximum gasoline yield was obtained at an operating severity index, expressed as the molar ratio of  $i-C_4/(C_3^- + C_4^-)$  in the product, of 0.5-1.0. Peak Research Octane Numbers of 90-94 were achieved at severity index of 0.3-2.0.

- Two small and one large interruption in the F-T synthesis operation occurred. A slight loss in F-T catalyst activity and a slight increase in the methane yield were observed during the small interruption. The F-T catalyst slurry were unloaded and then reloaded into the slurry reactor during the major interruption. Substantial deterioration of the F-T catalyst activity and a substantial increase of the methane + ethane yield were observed, probably due to catalyst damage incurred when it was exposed to air.
- The F-T reactor-wax yield increased significantly with decreasing methane + ethane yield.
- o Process variable studies were conducted during the latter part of the run. The results are:
  - Use of 0.6  $H_2/CO$  feed ratio synthesis gas instead of 0.7 lowered the methane yield. It also gave better usage of the synthesis gas since the  $H_2/CO$  usage ratio was nearly 0.6.
  - Higher operating pressure at the same superficial feed-gas velocity in the slurry reactor resulted in slightly lower H<sub>2</sub>+CO conversion, but a significantly lower methane + ethane yield. It also increased the oxygenate yield significantly.
  - Addition of a potassium-salt to the slurry reactor drastically decreased the methane + ethane yield.
- c A "hydrodynamic upset" of the slurry reactor was observed at the end of the run, probably due to catalyst settling resulting in low  $H_2+CO$  conversion and lower temperatures at the upper portion of the reactor. The upset disappeared after eight hours of high gas velocity operation, but re-appeared after the velocity was reduced.

After the end of Run CT-256-3, the two-stage bench-scale pilot plant was shut down for modifications. The major modification was addition of two external filter assemblies to withdraw reactor-wax from the first-stage slurry reactor. Other modifications included installation and replacement of many filter elements, installation of pressure transducers and heating tape, and replacement of the flange gaskets for the first-stage reactor.

Analytical procedures and supporting tests for the current pilot plant operation are summarized. The streams that required analysis include carbon monoxide, hydrogen, the combined feed gas, the first-stage F-T products (gaseous, aqueous, liquid hydrocarbon, and reactor-wax phases), and the second-stage ZSM-5 products (gaseous, aqueous, and liquid hydrocarbon phases). The gaseous streams are analyzed on-line using a Mobil-developed automated gas chromatographic (GC) system. The non-acid and acid oxygenates in the F-T aqueous phase are determined by fused silica capillary column GC (FS-GC) and ion chromatography respectively. Oxygenates in the F-T liquid hydrocarbon phase are determined using GC analysis. The F-T liquid hydrocarbon phase is analyzed using various GC and liquid chromatography (LC) techniques, while its carbon number distribution is routinely obtained using a FS-GC capillary column. More detailed analysis of the F-T liquid hydrocarbon is performed on selected cases using distillation, oxygenate separation, and a GC equipped with a olefin scrubber and flame ionization detectors. The analysis of F-T reactor-wax is done using FS-GC, LC, and solvent extraction. The second-stage liquid hydrocarbon product is analyzed using a GC system and an olefin scrubber. Various supporting tests are also employed, including acid number, bromine number, hydroxyl number, simulated distillation, kinematic viscosity, surface tension, vacuum distillation, and octane numbers.

Gasoline stability tests for gum formation and oxidation showed that existent (heptane washed) gum contents were all within acceptable limits, as were the oxidation stabilities. However, total residues on evaporation (unwashed gum) were generally high, probably due to the presence oir a small amount of high-boiling, heptane-soluable hydrocarbons.

6. Future Work

 The two-stage bench-scale pilot plant will be operated to test new devices for F-T reactor-wax/catalyst separation.

- The two-stage bench-scale pilot plant will be operated to evaluate a new F-T catalyst.
- o The evaluation of the raw gasoline product from the bench-scale pilot plant will be continued.
- B. Task 4 Conceptual Design Study

#### 1. Task Status

This task was initiated. The process design basis, including the material balance and operating conditions, was constructed based on the data obtained from the two-stage bench-scale pilot plant. The design basis data were forwarded to Mobil's Engineering Department at Princeton, New Jersey, and the conceptual design work is now in progress.

#### 2. Future Plan

Additional work on this task include process design, equipment sizing, and cost estimation. The task is expected to be complete by March 1983. The results of the study will be included in the Final Report.

- V. NOMENCLATURE
- P Pressure, (MPa)
- T Temperature, (°C)
- u Superficial velocity, (cm/s)
- w Weight fraction of catalyst in slurry, (gCat/g slurry)

### Greek Letters

α Probability of the chain-growth
 ε<sub>q</sub> Gas holdup, (mL gas/mL expanded slurry)

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### Superscripts

i At reactor inlet

### Subscripts

c Catalyst

g Gas

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# <u>Table l</u>

# Major Events in Run CT-256-3

(Excluding Rx-Wax & Slurry Inventory)

DOS	Major Events
-0.4-0.0	Pretreatment: 1st-Stage: 282°C; 1.14 MPa; 4.2 cm/s.
0.0-8.0	Syntheses Started: 1st-Stage: 282-260 <sup>0</sup> C, 1.48 MPa; 4.2-3.7 cm/s.
8.0-16.0	2nd-Stage: on: 329-385°C.
16.3-17.9	2nd-Stage Regeneration Upset: Unit under nitrogen nine hrs.
18.0-29.4	lst-Stage: 260-261 <sup>0</sup> C; 3.7-3.5 cm/s. 2nd-Stage: 343-304-346 <sup>0</sup> C.
29.7	Upset: Leak at 305 cm level flange; unit under nitrogen purge 36.5 hrs.
29.7-45.9	lst-Stage: 3.45-3.0 cm/s. 2nd-Stage: 346-466°C.
46.5-47.8	2nd-Stage Regeneration
50.0-59.7	lst-Stage: 261-266 <sup>0</sup> C. 2nd-Stage: 304-318 <sup>0</sup> C.
59.8-60.8	1.48-2.17 Mpa Upset: Leak at 0 cm level flange. Slurry unloaded, then reloaded after fifty hrs.
60.9-67.9	lst-Stage: 267 <sup>°</sup> C; 1.48 MPa; 3.1-2.5 cm/s. H <sub>2</sub> /CO in feed: 0.7-0.6 2nd-Stage: 323-349 <sup>°</sup> C.
67.9-80.0	1st-Stage: 1.48-2.5 MPa: 2.5-3.7-1.0-2.8-2.5 cm/1 2nd-Stage: 349-408 C.
80.8	Addition of a potassium-salt.
81.3-85.9	Hydrodynamic Upset 1st-Stage: 2.51-1.48-2.51 MPa. 2nd-Stage: 408
86.0	End Synthesis

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### Ranges of Operation Results (Run CT-256-3)

First Stage	•	Range of Results
H2+CO Conv., mol % Methane + Ethane Yield, wt % HC Reactor-Wax Yield, wt % HC		54-93 6-18 3-13

## Second-Stage Hydrocarbon Yield, Wt %

		Before Alkylation	After Alkylation
C1+C2	ະ	9-20	9-20
$C_3 - C_4$		14-38	12-28
$C_{5} - C_{11}$	•	32-55	46-68
$C_{12}^+$ (excel.	reactor-wax)	1-9	1-9

# Properties of Raw Liquid Hydrocarbons (1)

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Aromatics, Wt 3		12-41
Acid No., mgKOH/gHC	(unwashed)	0.04-0.4
Octane No., R+O M+O		82-94 74-84
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(1) Collected in ambient and chilled condensers.

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Effect of Superficial Fischer-Tropsch F	Feed-Gas Veloc Reactor Perform	ity on Slurry ance (1)	
(Run	CT-256-3)		
DOS	75.4	76.5	77.4
Gas Superficial Velocity, cm/s	2.5	2.1	1.6
Space Velocity, NL/gFe-hr	3.12	2.53	1.95
H <sub>2</sub> +CO Conv., mol %	77.1	87.2	93.1
Methane, wt % HC	8.8	8.5	7.8
Methane + Ethane, wt % HC	12.7	12.6	12.2

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<sup>(1)</sup>0.6 H<sub>2</sub>/CO, 267<sup>°</sup>C, 2.51 MPa

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# Table 4

Effect of Pressu Rea	ure on Slu actor Perf			
DOS	66.8	58.8	72.4	74.8
Pressure, MPa .	1.48	1.82	2.17	2.51
Space Velocity, NL/gFe-hr	1.95	2.32	2.77	3.12
H <sub>2</sub> +CO Conv., mol %	81.2	81.7	79.5	77.5
Methane, wt % HC	10.8	10.1	9.4	8.7
Methane + Ethane, wt % Hc	14.8	14.4	13.6	12.6

(1)0.6 H<sub>2</sub>/CO, 267°C, 2.6 cm/s superficial feed-gas velocity.

Table 5

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Effect	of	Press	ure (	on (	Oxygen	ates	Yield	from
Fij	st-	Stage	Fise	cher	r-Trop	sch B	Reactor	

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DOS	50.6	74.5
Pressure, MPa	1.48	2.51
Temperature, <sup>°</sup> C	263	267
Feed H <sub>2</sub> /CO, molar	0.7	0.6
Superficial Feed-Gas Velocity, cm/s	3.1	2.6
Space Velocity, NL/gFe-hr	2.18	3.15
H <sub>2</sub> +CO Conversion, mol %	85.2	7.7.5
Oxygenates, wt % of HC	8.6	12.0

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<u>Table 6</u>

	Summary of Est	imated Gas	Holdup ;	om DP-Cell	Data
DOS	9.2	75.5	78.6	78.8	80.8
ug, cm/s	3.9	2.6	- 1.1	2.6	2.6
I, C	260	267	260-267	260-267	267
P, MPa	1.48	2.51	2.51	2.51	2.51
w <sub>c</sub> , wt %	14.3	13.9	11.7	12.0	12.0
<sup>E</sup> g, vol	% 26.6	19.7	6.8	9.3	19.8
### <u>Table 7</u>

### Identities of Major Aqueous Phase Oxygenates

<u>Peak No. (1)</u>	Oxygenates
Peak No. (1)  1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22	Oxygenates methanol ethanol acetone isopropanol t-butyl alcohol 1-propanol butanal methylethyl ketone 2-butanol ethyl acetate isobutanol 2-methyl-2-butanone acetic acid 1-butanol 2-pentanone pentanal+3-pentanone 2-pentanol ethyl propanoate propyl acetate methyl butanoate propanoic acid 3-methyl-1-butanol
24	1-pentanol
25	2-hexanone
26	butyric acid
27	other C <sub>6</sub> oxygenates
28	l-hexanol
29	other C <sub>7</sub> oxygenates
30	l-heptanol
31	other C8 oxygenates
32	l-octanol

(1) Identified by gas chromatography-mass spectrometry.

### Table 8

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### Raw Gasoline Product Evaluations

			ASIM D3 mg/100 1	al Guns,	ASIM D525
Material Balances	Days en Stream	Additive Pkg. No.*	Existent	(Unwashed)	Oxidation Stab., mins.
CI-256-3-57	66.5	l	4	119	620
<b>CT-256-3-57</b>	56.5	2	4	81	710
CI-256-3-63	78.5	1	l	37	660
<b>CI-256-3-63</b>	78,5	2	2	12	825

\*Additive Package No. 1 -- 0.5 lb/1000 bbl. metal deactivator + 2.5 lb/1000 bbl. antioxidant.

*Additive	Package	No.	2	 0.5	16/1000	bbl.	metal deactivato	r	÷
				15	16/1000	bbl.	antioxidant.		

### SYNTHESIS GAS CONVERSION AND METHANE & ETHANE YIELD

(RUN CT-256-3 1ST-STAGE CATALYST I-B:PPTD Fe/Cu/K2CO3; 2ND-STAGE CATALYST II-B:ZSM-5)





### **REACTOR-WAX CARBON-NUMBER DISTRIBUTION**



### EFFECT OF REACTOR PRESSURE ON REACTOR-WAX CARBON-NUMBER DISTRIBUTION





### **REACTOR-WAX YIELD**



### SCHULZ-FLORY DISTRIBUTION FOR FIRST-STAGE FISCHER-TROPSCH PRODUCTS



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### EXIT H<sub>2</sub>/CO RATIO OF FIRST-STAGE SLURRY FISCHER-TROPSCH REACTOR



### SLURRY FISCHER-TROPSCH BUBBLE-COLUMN GAS HOLDUP PROFILES



### SLURRY FISCHER-TROPSCH BUBBLE-COLUMN CATALYST CONCENTRATION PROFILES

(Run CT-256-3)



\*\*After hydrodynamic upset was corrected

### SECOND-STAGE FIXED-BED ZSM-5 REACTOR INLET AND OUTLET TEMPERATURES



### PRODUCT YIELDS VERSUS SECOND-STAGE OPERATING SEVERITY

(Run CT-256-3)



\*Excluding  $C_4^-$  paraffins in feed and reactor-wax

### SECOND-STAGE RAW LIQUID HYDROCARBON PROPERTIES

### (Run CT-256-3)



I-Butane/(Propylene + Butenes), molar

SCHEMATIC OF THE EXTERNAL WAX FILTER ASSEMBLY



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# ANALTYICAL SCHEME OF FISCHER-TROPSCH PRODUCTS



### GC PLOT OF TYPICAL COMBINED GAS SAMPLES



### ANALYTICAL SCHEME OF FISCHER-TROPSCH OXYGENATES



• FS-GC = Fused Silica Capillary Gas Chromatography.

### GAS CHROMATOGRAM OF $C_6^+$ ORGANIC PHASE ALCOHOLS AND ACIDS

 $C_x = N$ -Alcohols



# GAS CHROMATOGRAM OF Cs<sup>+</sup> ORGANIC PHASE KETONES AND ESTERS





# TYPICAL GAS CHROMATOGRAM OF TOTAL LIQUID HYDROCARBON PHASE

### FIGURE 19

### TYPICAL CHROMAGRAMS OF C<sub>11</sub><sup>-</sup> FROM LIQUID HYDROCARBON PHASE





APPENDIX A

### MATERIAL BALANCE DATA FROM CT-256

# Table A-1 First Stage Fischer-Tropsch Slurry Reactor Operating Conditions and Material Balances Second-Stage Not-Operative Run CT-256-3

Total	Slurry Rx. Wax	Heavy Hydrocarbons (4)	Light Hydrocarbons (3)	C5 - C11	n-Butane	i-Butane	Butenes	Fropane	Propene	Ethane	Ethene	Methane	Selectivities, Wt % of HC :	g HC/Nm3 (H2+CO) convtd.:	Bal Recovery, Wt % of Charge:	Total	00	H2	H20 (1)	CO2	Hydrocarbons (1)	Yields, Wt % of Froducts :	H2+C0	28	H2	Conversions, Mol % :	N2 in Feed, Mol %	Space Vel., NL/gFe-hr	Feed Sup. Vel., cm/s	Fressure, MPa	Temperature, oC	Charge H2/CO (Molar)	pays un-stream First-Stage Conditions:	1. D. NG.	(Nitrogen-Free Basis)
100	12.76	23.82	21.15	14.79	1.44	0.08	6.10	1.38	7.64	2.75	2.29	5.69		217	106.03	100	8.66	1.11	1.07	67.92	21.24		84.14	90.29	75.36		о. 5	2.373	3.652	1.480	257	0.701	3.4	ی 1 ۱	3
100	12.34	24.17	20.42	15.71	1.47	0.07	6.00	1.39	7.54	2.61	2.16	5.85		218	102.57	001	7-64	0.94	0.82	68.10	22.50		86.69	Ý1.75	79.19		ć.7	2.402	3.713	1.480	259	0.676	4.4	4 1	) )
100	11.89	27.54	19.89	15.34	1.48	0.07	6.06	1.50	7.56	2.60	2.12	5.85		222	106.38	100	8.30	0.93	0.94	68.02	21.81		85.79	90.68	78.59		6.6	2.394	3.696	1.480	259	0.677	и. 4		\$ \$
100	11.75	25.85	20.62	14.49	1.48	0.07	5.86	1.51	7.32	2.61	1.91	5.86		223	107.80	100	7.90	0.89	0.79	68.54	21.87		86.30	50.98	79.53		6.7	2.404	3.719	1.480	259	0.689	ó. 4	31	•
100	11.14	25.69	20.60	14.52	1.57	0.07	6.03	1.59	7.54	2.74	1.79	6.15		216	103.71	100	7.70	0.92	0.86	68.43	22.10		86.72	65-16	79.53		ć.4	2.541	3.918	1.480	N09	0.677	7.4	ິ ເ	) 4
100	6.42	22.23	33.34	8.63	1.85	0.07	5.87	1.68	7.84	2.65	1.40	7.59		227	106.01	100	9.07	0.85	0.71	66.64	22.73		86.03	53 68	80.37		с.	2.101	3.210	1.480	261	0.675	48.5	40	)
100	6.18	23.29	31.02	9.12	1.92	0.07	6.04	1.88	<b>ğ.</b> 18	2.95	1.49	7.86		227	107.55	100	06-6	0.91	0.78	66.38	22.03		84.73	88.74	78.82		6.4	2.068	3.131	1.494	261	089-0	49.5	3-4/	)
100	5.97	23.73	30.12	8.72	1.95	0.07	6.21	2.22	8.13	3.21	1.52	7.99		210.	103.73	100	8.76	0.89	0.65	68.17	21.52		86.22	90.42	60.02		6-4	2.088	3.149	1.501	261	0.678	50, 5	3- 48	) 5
100	5,85	23.62	29.01	9.14	2,00	0.07	6.39	2.27	8.41	3,33	1.53	8.20		214	105.69	100	8.84	0.95	0.72	68.05	21.44		85.43	90.13	78.53		6.5	2.082	3.176	1.487	262	0.681	51.5	3-45	)
100	5.76	22.75	31.91	8.78	1.91	0.07	5.99	1.93	8.17	3.11	1.51	7.97		225	104.97	100	10.00	1.01	0.87	65,65	22.47		84.27	88.92	77.56		6.4	2.120	3.230	1.487	262	0.693	52.5	3- 50	) ]

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Including Oxygenates
 In Gas Phase Only
 Collected in Chilled and Ambient Condensers
 Collected in Hot Condenser

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First Stage Fischer-Tropsch Slurry Reactor **Operating Conditions and Material Balances** Based on Inter-Reactor Sample Run CT-256-3 Table A-2

3- 23 24.0 1.487 3.736 2.516 4.5 \$0.77 1.6915.6y 19.27 **65.**90 0.36 6.02 2.52 2.52 5.50 0.05 259 86.36 23.59 0.89 8.77 100 100.18 220 ć. 64 1.77 0.637 81.17 29.95 8: 4:3 100 3- 22 23.0 3.829 2.507 6.6 78.36 89.37 84.89 19.43 69.41 0.78 0.95 9.43 100 260 0.685 1.480 198 20.13 26.93 8.91 100 106.61 3- 21 22.0 1.480 3.734 2.502 4.5 79.12 6ć. 12 21.77 67.98 30. 62 9. 00 100 103.05 0.061.85 260 \$0.51 0.92 0.95 8.38 100 6.98 1.47 7.54 1.81 5.99 18.50 0.683 211 3- 20 21.0 1.480 3.701 2.498 3.9 9.42 3.75 1.91 10.02 92.31 88.24 29.09 61.65 0.81 0.87 7.58 0.08 502 82.33 100 2.46 2.46 13.81 20.87 9.08 100 0.685 96.81 0000 8.07 3- 19 20.0 90.88 85.97 1.460 3.729 2.494 106.08 0. 689 259 78.85 21.85 68.20 2100 7.12 1.47 2.86 7.82 1.90 0.07 1.96 14.30 17.29 28.62 28.62 28.62 100 4.7 0.94 8.15 100 6.40 0.87 3- 13 15.4 77.24 89.90 84.76 1.460 3.687 2.598 4.7 20.48 68.79 0.99 8.89 100 107.55 211 260 0.85 6.95 1.41 7.96 1.86 6. GG .0.07 1.85 11.56 18.18 31.39 9.60 100 0.684 1.480 3.906 2.578 5.9 3- 12 14.4 76.22 89.54 21.01 68.64 259 0.98 8.66 100 1.92 25.00 9.80 100 0.638 0.71 6. % 6. % 6. % 6. % 7. % 7. % 7. % 0.07 84.11 113.65 21.10 12.54 231 3- 11 13.4 1.480 3.926 2.600 2.6 77.74 91.15 85.72 21.52 68.53 0.69 1.02 8.24 18.93 14.16 27.33 9.88 100 260 100 000 000 000 6. 60 7. 60 1. 64 6. NU 0.06 1.80 02.11 0.681 ۳. 11. % 0.679 90.70 85.53 1.480 21406 68.28 1.01 8.65 205 260 4.9 14.19 17.23 29.27 10.26 10.26 3.851 100 1.75 2.569 77.91 1.01 102.30 0.07 ĥ 1.480 3.850 2.580 ມ 10.08 78.40 91.35 86.11 21.62 68.26 100.38 205 200 4.5 0. 50 6.37 2.65 7.46 7.46 7.46 7.46 6.03 6.03 0.00 1.68 15.68 17.13 28.56 28.56 1.00 <u>8</u> 8.21 100 0.680 ĥ Bal Recovery, Wt % of Charge: .. • # g HC/Nm3 (H2+CQ) conv.: Selectivities, Wt X of HC Yields, Wt X of Products Light Hydrocarbons (3) Heavy Hydrocarbons (4) Space Vel., NL/gFe-hr N2 in Feed, Mol X First-Stage Conditions: Feed Sup. Vel., cm/s Charge H2/CO (Molar) (Nitrogen-Free Basis) Conversions, Mol % : Hydrocarbons (1) Temperature, oC Ŭays Ŭn−stream Slurry Kx. Wax Pressure, MFa C5 - C11 (2)i-Gutane n~©utane M.B. No. H20 (1) Fropene Propane Butenes Methane Ethene Ethane H2+C0 Total lota] 000 N H 00 88

Including Cargenates
 In Gas Phase Only
 Collected in Chilled
 Collected in Chilled

In Gas Phase Only Collected in Chilled and Ambient Condensers Collected in 25. Condenser

First Stage Fischer-Tropsch Slurry Reactor Operating Conditions and Material Balances Based on Inter-Reactor Sample Table A-2 (Contd.) Kun CT-256-3 3.375 2.267 5.0

3.409 2.273 5.0

0.470 2.253 5.2

3.426 2.279 5.1

3.428 2.282 5.1

79.86 91.44

79.07 86.63

79.50

79.45

77.80 89.61 84.83

86. 63

86.33

86.73

22.99

23.56 67.28

22.37 68.27

22.63 67.54 0.78 0.95 8.10 100

21.96 66.85 0.78 0.99

67.42

0.34 0. \$1 7.84

0.82 0.96 7.33 100

0.81 7.61

100

103.74 222

103.32

104.61

103.66 217

104.61

224

100 219

100

<u></u> Ф. 43

100

0. 93

225

216

7.23 1.34 2.757 1.63

7.66

6.01 0.06 1.93

6.34 0.07 2.01

6.42 0.07 2.03 15.87 17.94

7.50 1.29 2.77 2.77 7.86 7.86 6.07 6.07 6.07 1.94

2.02 2.02 2.02

7.57 1.41 2.92 7.68 1.98 1.98 6.25 6.25 6.25

а. 09 2. 09 2. 09

1.54

7.71

12,55

18.50

12.68 19.58 29.97

14.28 17.85 28.44 7.97

14.35 18.28 27.38 8.99 100

25.95 8.30 100

31.73 7.63 100

7.66

1.454

1.480

0. 635 260

0.695 259

0.696 1.446

0.702 260 1.450

0. 679 260 1.480

260

3- 35 36.5

35.5

3- 33 34.5

ର- ସମ ସ**ସ**- ଅମ

32.5

31.5

3- SI

8

9- 04 8-

1.487 a.546 2.373 5.0 1.09 12.43 0. 686 260 75.58 86.21 81.88 61.26 0.57 21.66 2.96 8.11 1.99 15.29 18.93 24.76 8.84 100 7.79 0.07 2.00 6.41 105.26 å 0.00 0.00 0.00 0.00 71.19 80.15 30**.**5 20.39 60.55 0.85 1.22 17.50 261 1.480 3- 29 0.665 100 13.88 21.57 22.79 8.45 8.45 76.57 07.54 234 0- 58 20-08 0.590 2.938 4.9 1.480 78.16 89.80 21.94 67.22 0.75 0.93 2.87 7.94 1.94 0.07 35.03 0.694 100 06.10 220 7.36 13.81 18.50 28.35 8.46 100 261 9.11 3- 25 26.0 3.727 2.489 4.7 260 77.47 89.04 21.54 66.58 1.480 0.79 1.03 100 214 6.84 1.47 2.80 7.42 1.93 6.04 0.06 1.87 13.28 13.56 30.44 8.54 100 64.32 0.689 10.06 103.47 3- 24 25.0 78.35 90.33 85.46 101.76 207 260 1.480 3.732 2.497 4.5 21.48 1.00 0.04 67.71 0.77 100 7.19 13.70 18.37 28.86 8.62 9.62 0.685 Bal Recovery, Wt % of Charge: .. Vields, Wt % of Products : g HC/Nm3 (H2+CO) conv.: Selectivities, Wt % of HC Light Hydrocarbons (3) Heavy Hydrocarbons (4) First-Stage Conditions: Charge H2/C0 (Molar) Feed Sup. Vel., cm/5 Space Vel., NL/9Fe-hr N2 in Feed, Mol % (Nitrogen-Free Basis) Conversions, Mol X : Hydrocarbons (1) Temperature, oC Days Ön-stream Slurry Kx. Wax Fressure, MPa C5 - C11 (2)M. B. No. i-Butane n-Butane 002 H20 (1) Ethane Fropene Methane Fropane Butenes Ethene H2+C0 Total Total Ŷ 3 앞 8

Including Oxygenates

909

In Gas Phase Only Collected in Chilled and Ambient Condensers Collected in Hot Condenser

First Stage Fischer-Tropsch Slurry Reactor Operating Conditions and Material Balances Based on Inter-Reactor Sample Table A-2 (Contd.) Run CT-256-3

0.130 2.035 5.8 21.78 66.32 0.54 0.9617.24 27.68 4.33 100 0.675 1.480 8.7% 2.31 6.94 2.15 2.15 15.54 57.5 78.35 10.40 8.30 1.64 3.47 264 88.61 64.47 100 103.97 217 ය ස 3.270 2.217 4.1 87.58 21.32 64.86 0.74 45.5 1.494 82.62 1.17 100 1.51 7.91 1.97 6.33 0.07 2.0215.60 17.13 29.24 6.59 44 360 75.43 95-34 206 7.65 11.91 0.689 ဂို  $c. 92 \\ c. 92 \\ c. 92 \\ c. 93 \\ c. 9$ 77.52 89.06 0.030 4.1 22.57 64.85 0.89 1.08 44.5 0.687 260 1.454 84.36 98. 36 212 9- 49 9-10.60 100 ର- 42 43**.**ଅ 3.267 2.228 4.5 83, 23 19.51 66.67 2.04 1.07 10.71 100 105.10 198 1.58 3.12 8.45 260 1.508 88.13 S. 14 2.17 ć. 87 0.693 76.16 7.35 2.767 7.62 7.62 7.62 1.87 1.87 1.98 1.98 1.98 3.452 2.277 5.8 79.38 66.63 0.75 42.5 0.689 260 1.450 85.73 22.10 0.99 9.53 100 205 17.81 28.37 6.82 6.82 98.98 <u>0-</u> 41 3-40 41.5 1.480 3.379 2.269 79.08 89.38 2.299.996.68 260 4.0 \$5.18 20.28 100 0.07 2.14 0.689 éé.46 194 7.80 1.54 2.94 0.29 2.06 22.77 20.11 6.84 101.16 6.34 6.39 1.15.96 6.48 27.88 6.98 1.17.98 6.98 1.17.98 6.98 1.00 100 1.480 3.407 2.288 40.5 80.25 90.97 0.693 260 21.64 68.13 0.90 0.50 86.59 0.74 7.63 1.45 2.83 8.02 4.1 100 100.40 3- 39 201 0.679 1.494 3.370 2.282 4.3 22.67 67.53 0.73 39.5 87.03 21.12 28.04 7.22 100 260 91.16 0.87 8.20 100 216 89 - 88 1- 88 SO. 94 102.52 3.402 2.273 4.8 0. 695 259 37 3**8. 6** 80.74 87.06 0.85 0.50 7.95 102.292261.480 **91.44** 23. 91 66.39 100 \$ 0.674 260 7.62 37.5 3.391 2.275 4.7 67.47 0.70 1.89 1.95 1.487 80.75 87.06 22.52 0.%0 8.40 100 98. ċć 7.16 1.39 2.73 é. 14 0. 0é 18.12 28.62 7.47 100 3- 36 91.32 207 Bal Recovery, Wt X of Charge: .. g HC/Nm3 (H2+CO) conv.: Selectivities, Wt X of HC Yields, Wt % of Froducts Light Hydrocarbons (3) Heavy Hydrocarbons (4) First-Stage Conditions: Feed Sup. Vel., cm/s Space Vel., NL/9Fe-hr N2 in Feed, Mol X Charge H2/C0 (Molar) (Nitrogen-Free Basis) Conversions, Mol X : Hydrocarbons (1) Temperature, oC Days On-stream Slurry Km. Wam Total Fressure, MPa CS = C11 (2) 1-Butane M. B. No. n-Butane 002 H20 (1) Eutenes Fropane Me thane Propene Ethene Ethane H2+C0 Total 8 연 Ě 8

Including *Oxygenates* (1)

In Gas Phase Only

Collected in Chilled and Ambient Condensers 9 8 F

Collected in Hot Condenser

First Stage Fischer-Tropsch Slurry Reactor Operating Conditions and Material Balances Based on Inter-Reactor Sample Table A-2 (Contd.) Run CT-256-3

> Nitrogen-Free Basis) Days On-stream M.B. No.

3- 63 78.5

3- 62 76.5

3- 61 74.5

3- 60 72.5

70.5

68.5

66.5

3-58

3-57

. Э- 59

0.634 264 2.515

0.600

0.596

266 2.515

265 0.607

> 266 2.170

0.601

0.601 265

265 1.535

0.693

1.825 2.665 2.186

2.170

1.402

2.515 266

2.204 2.612 3.2

2.567 3.014 4.0

2.591 2.616 4.5

2.587 4. ć

2.773 1.923 7.3

2,568

6.9

91.02 92.11 91.63

68.61 87.60 87.58

79.77 78.68 79.09

80.25 79.23 79.62

80.35 79.36 79.73

83.88 83.27 83.50

84.02 80.57 81.98 73.96 0.09 7.04 100

66.64

19.62 58.41

20.04 59.90 0.08 0.80

> 0.09 60.49

62.61 0.04 21.17

19.75

20.16 61.47

60.42

20.15 0.16

20.81

18.54

0.37

**0.05** 0.48

0.05 0.80 19.19 100 219

173

197

106.33

98.79

102.78

100

9.30

4.75 0.00 9.27 0.16 24.40

7.79 2.05 3.55 3.55 8.74 6.70 0.12

7.90 2.36 3.47 8.62 8.62 8.62 8.63 8.63 8.63 8.63 8.63 8.63 8.63

13.61 19.36 12.23 100

2.30 15.88 22.89 20.74 5.62 100

2.23 115.11 222.74 222.74 4.59

19.43 4.59 100

20.14 18.92 4.04 100

2.71

0.13

3- 56 64.5 71.87 78.73 75.93 266 1.577 2.857 2.036 7.1 0.651 First-Stage Conditions: Space Vel., NL/9Fe-hr N2 in Feed, Mol X Conversions, Mol X : Charge H2/C0 (Molar) Feed Sup. Vel., cm/s Temperature, oC Fressure, MPa H2+C0 Сі Т 30

• • Vields, Wt % of Products Hydrocarbons (1) H20 (1) C02

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100 218 19.19 103.37 0.78 18.69 8.49 3.836 9.60 8.57 7.18 8.44 8.44 100 218 04.33 9,02 35,24 35,24 36,29 37,21 3 0.65 15.54 100 102.78 220 8.45 2.25 3.51 0.05 0.72 17.59 104.53 001 1.1918.08 10.77 2.84 2.84 11.40 2.90 2.90 2.90 2.90 3.25 14.02 3.25 14.02 3.25 100 100 100 110.68 236 Bal Recovery, Wt % of Charge: .. g HC/Nm3 (H2+CO) conv.: Selectivities, Wt X of HC i-Butane Methane Propene Propane Butenes Ethene Ethane Total 3

(1) Including Uxygenates

Light Hydrocarbons (3) Heavy Hydrocarbons (4)

C5 - C11 (2)

n-Butane

Slurry Rx. Wax

Tota)

In Gas Phase Unly 999

Collected in Chilled and Ambient Condensers

Collected in Hot Condenser

Table A-3 Composition of Hydrocarbon Products from First-Stage Slurry F-T Reactor Run CT-256-3

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						(1)	(1)	(1)	(1)	(1)
M. B. No.	- M	ମ - ୯ ୯	0 1 10	0-4-0	ທ - ຕ	で う つ	シーご	0-11 0-11	8-19 8-19	8-13
Days On Stream	3.4	4-4	5.4	ć.4	7.4	10.4	11.4	18-4	14.4	15.4
METHANE	5.69	5,85	5.65	5.86	ć. 15	6.37	6.60	<b>6.</b> 60	96.9	ć. 95
ETHENE	010	9  	010 1 0 0 0	1.91	0,79	1.65	1.00	1.61 0 27	1.66	1.41
FROFENE	7.64	7.54	7.56	7.52	7.54	7.46	7.64	7.64	80.8	7.86
FROPANE	1.38	1.39	1.50	1.51	1.5%	1.78	1.85	1.64	1.83	1.61
	0.08	0.0	0.07	0.07	0.07	0.00	0.07	0.06	0.07	0.07
L-BUTENE+2-METHYLFROPENE	00° 20°	58	5.00	5. 46. 9	5. ć1	5.61	5.78	5 5 7	ć. 16	5. YU
N BUTANE TRANC-5-5117545	1.44	- <del>4</del> - 4	1.43	1 • 4B	1.5/	1.68	1.75	1.80	1.92	1.85
LIANS-2-BUTENE CIS-2-BUTENE	0,10	0.00		0.14	0.10	0.10		0.16	0.16	0.14
S-METHYL-1-GUTENE	0.30	0.34								
I-FENTANE	0.24	0.36	0.22	0.18	0:12	, 8 <u>9</u> . 0	0.16	40.4	0.17	0.17
1-PENTENE	4.21	4.80	4.32	4.11	4.19	4.14	4.27	1.51	4.84	0.04
2-METHYL-1-BUTENE	0.28	0.19	0.22	0.17	0.17	0.19	ù. 15	0.15	0.19	4.40
N-FENTANE	1.13	1.17	1.20	1.20	1.25	1.36	1.37	0.16	1.ć4	1.39
IRANG-W-PENTENE	0.11	0.12	0.12	0.12	0.13	0.13	<u>0.13</u>	0.01	0.15	0.06
GIG-Z-PENTENE Q-METUVI - C. THITCH	0.15	0.15	0.14	0.14	0.15	0.15	0.15	0.00	0.16	0.13
がつけに I T L ー メージリー にNE グング・ ふりらい T ANG	<b>6.</b> 0		00°0	5.0 0	0.0	0.00 0.00	0.01	0.00	0.01	0.27
HEXENES + ISO-HEXANES	् २ २ २ २	0.67	0.74	0.00	0. 40 2. 5 2. 5					
2, 3-DIMETHYLBUTANE	0.00	0.00	0000	0.00	0.00	0.00	00.00	00 0	33	
2-METHVLPENTANE	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.0	0.00	3.17
3-METHYLPENTANE	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	1.15
I-HEXENE	2.61	0. 0. 0.	ທີ່ ທີ່ ທີ	2.70	2.67	2.57	2.64	З. 08	3.46	0.11
N-HEXANE Virganino - 400 virganino	2.0	0.93	0.0	0.87	0.00	0. %	0.95	1.17	1.28	0.10
HEFIENEG + JSC-HEFIANEG	- () 	0. 64	0.53	0.57	0.57	1.06	0.51	0.85	0.45	0.00
L-AFFART L-TGANG G-FAMPTING NG	9 		1. 5H	-1-4C	38.1	1.17	1.19	1.83	រ រ រ រ	0.00
L - I NANG-G-UINE I N L-WO W-HFFTANF		32		0.00			0.16	0°0	33	000 000
C8-OLEFINS + ISO-P	0.45	0.33	0.00		0.31	0. AQ				
1-OCTENE ·	0.60	0. 63	0.55	0.53	0.52	0.40	0.35	0.98	8.1	0.00
N-OCTANE	0.23	0.29	0.24	0.27	ù. 25	0.21	0.15	0.53	0.74	0.00
C9-OLEFINS + ISO-P	0.26	<u>0.1ć</u>	0.15	0.11	0.16	0.84	0.26	0.78	0.73	0.00
C9-ULEFINS	0.19	0.17	0.13	0.12	0.11	0.10	0.07	0.36	0.51	0.00
N-NUNANE A OFT-ONE	0°0	0.10	0.07	0.00	G. 08	0.07	0.05	0.37	0.47	0.00
200-000 1-666660000	•0•0 0•0		0.0 0 0	0.44	0.34	0.17	0.5%	0.89	0.48	0.00
JTTAUF ANOL					0.24	0.20	0 () 0 ()	0.15	0.36	0.00
R BOTHNONG LINKNAMN I I TE HVARA-CARR I TO 101	00.00 10 10				00°0	8 9 9 7	3	0.00	0.66	0.00
UNKNOWN HVY, HYDRO-CARE LIG (3)	00. 00. 00.	24.17	00. 54	າ ເມື່ອ ເມື່ອ ເມື່ອ ເມື່ອ	200 201 201 201	9	11.20	14. TO	1 v. 01	10.10 *1 20
SLURRY REACTOR WAX	12.76	12.34	11.89	11.75	11.19	10.43	10.26	) (2) (2) (3) (3) (4) (4) (4) (4) (4) (4) (4) (4) (4) (4	200 200 200 200 200	5. ćÙ
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Based on Inter-Keactor Sample
 Collected in Ambient and Chilled Condensers
 Collected in Hot Condenser

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Composition of Hydrocarbon Products from First-Stage Slurry F-T Reactor Run CT-256-3 Table A-3 (Contd.)

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3-30 (1) 9.42 3-29 3 882. 862. (1) 3-28 29.0 (1) 3-25 27.0 (1) 3-24 26.0 (1) 3-23 24.0 3-22 7.55 8.18 8.18 9.18 7.55 0.18 0.18 0.18 0.18 Ξ (1) 3-21 22.0 (1) 3-20 21.0 (1) 3-19 20.0 unknown Lite Hydro-Carb Lig (2) Unknown Hvy Hydro-Carb Lig (3) Slurry reactor Wax I-BUTENE+2-METHYLPROPENE HEPTENES + ISO-HEPTANES HEXENES + ISO-HEXANES 2, 3-DIMETHYLBUTANE 2-METHYL-2-DUTENE 2, 2-DIMETHYLEUTANE CYCLOPENTANE 4-031 + C9-OLEFINS + ISO-P 1-NÜNENE 2-METHYL-1-BUTENE 3-METHYL-1-BUTENF TPANS-2-PENTENE CIS-2-PENTENE Days On Stream IRANS-2-BUTENE CIS-2-BUTENE N-HEPTANE C8-CLEFINS C9-OLEFINS N-FENTANE I-FROPANCL N-BUTANONE I-PENTENE **I-PENTANE** 1-HEPTENE I-BUTANE N-BUTANE ethene Ethane Propene Propane N-HEXANE N-NONANE I-HEXENE 1-00TENE N-OCTANE METHANE 4. B. No. ACETONE

Eased on Inter-Feactor Sample
 Collected in Ambient and Chilled Condensers
 Collected in Hot Condenser

Table A-3 (Contd.) Composition of Hydrocarbon Froducts from First-Stage Slurry F-T Reactor Run CT-256-3

(1) 3-40 41.5 0.40 0.42 0.31 0.23 0.15 0.37 22.77 20.11 6.84 0.14 821-24 822-6-24 1.11 0.46 0.58 0.28 0.28 0.28 0.17 0.17 0.10 0.10 0.33 0.35 17.98 27.63 6.96 (1) 8-89 40.5 0.40 (1) 3-38 39.5 0.16 0.10 0.33 0.33 21.12 23.04 7.22 0.11 0.31 (1) 3-37 38.5 56.71 27.55.71 75.57 75.54 76.71 76.75 76.75 76.71 76.75 77.75 76.75 777 1.13 0.51 1.59 0.66 0.33 0.367 0.367 0.36 0.35 0.35 0.13 0.13 0.43 0.40 0.42 18.99 29.05 7.31 0.12 0.33 0.38 0.38 0.35 0.35 18.12 18.12 7.47 (1) 3-36 37.5 0.11 (1) 3-35 36.5 1.25(1) 3-34 35,5 (1)
3-33
34.5 0.11 0.27 0.37 0.37 13.28 13.28 27.38 27.38 27.38 27.38 (1) 3-32 33**.**5 0.11 (1) 3-31 32.5 UNKNOWN LITE HYDRO-CARB LIQ (2) unknown hvy hydro-carb lig (3) Slurry reactor Wax I-BUTENE+2-METHYLPROPENE HEPTENES + ISO-HEPTANES HEXENES + ISO-HEXANES 1-HEXENE CS-OLEFINS + ISO-P N-OCTANE C9-OLEFINS + ISO-P C9-OLEFINS 2-METHYL-2-BUTENE 3-METHYL-1-BUTENE 2-METHYL-1-BUTENE TRANS-2-PENTENE CIS-2-PENTENE FRANS-2-BUTENE **Days On Stream** CIS-2-BUTENE N-BUTANONE I-PENTANE L-PENTENE I-PROPANOL N-PENTANE -HEPTENE N-DECENES N-HEPTANE N-BUTANE -OCTENE I-BUTANE N-NONANE **Z-HEXANE** N-DECANE METHANE ETHENE ETHANE PROPENE ACETONE FROPANE M.B.No.

Based on Inter-Reactor Sample . Collected in Ambient and Chilled Condensers Collected in Hot Condenser 388

Composition of Nydrocarbon Froducts from First-Stage Slurry F-T Reactor Table A-3 (Contd.) Run CT-256-3 (1) 3-53 57.5 0.35 0.15 0.00 0.00 0.11 0.28 0.33 0.41 0.41 27.63 4.83 3-50 3-49 3-48 50.5 7.9% 8.21 5.82 5.82 5.87 5.87 5.87 3-47 3-46 48.5 (1) 3-44 45.5 0.17 0.22 0.10 (1) 3-43 44.5 (1) 3-42 43.5 0.11 0.11 0.38 0.38 0.38 0.44 0.44 5.74 5.74 5.74 5.74 5.74 5.74 (1) 3-41 42.5 UNENDAN LITE HYDRO-CARB LIG (2) UNI:NOWN HVY HYDRO-CARB LIG (3) SLURRY REACTOR WAX **I-BUTENE+2-METHYLFROPENE** IEPTENES + ISU-HEPTANES UNKNOWN CS-MONOOLEFINS HEXENES + 1SO-HEXANES 2, 2-DIMETHYLBUTANE 2, 3-DI METHYLBUTANE 2-METHYL&ENTANE 3-METHYLFENTANE CB-CLEFINS + ISO-F C9-OLEFINS + 180-F C9-OLEFINS N-NONANE 3-METHYL-1-BUTENE 2-METHVL-1-BUTENE 2-METHYL-2-BUTENE RANS-2-PENTENE M.B.No. Days On Stream **FRANS-2-BUTENE** CIS-2-PENTENE CIS-2-BUTENE CYCLOPENTANE S0-C6-F40 N-PENTANE -PROPANOL N-DECANE N-DECENES 4-BUTANONE -PENTANE -PENTENE -HEF TENE V-HEPTANE -BUTANE N-BUTANE I-HEXENE -0CTENE N-HEXANE N-OCTANE METHANE PROPENE FROPANE ACE TUNE ETHANE

Eased on Inter-Reactor Sumple
 Collected in Ambient and Chilled Condensers
 Collected in Not Condenser

Tablé A-3 (Contd.) Composition of Hydrocarbon Products from First-Stage Slurry F-T Reactor Run CT-256-3

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M. B. No.		(1) (1)		(1) (1) (1) (1) (1) (1) (1) (1) (1) (1)	(1) 3-40	(1) 2-21	(1) (1)	
Days Cin Stream	é4.5	čé. 5	68°.5	70.5	72.5	74.5	76.5	78.5
METHANE	10.77	9. OB	8.45	8.49	B. ćć	7.90	7.79	9.30
ETHENE	2.64	N. 69	8: S	ល់ លំ	2.37	2.30	2.06	2.41
	62 H	20.00		9°83	3.67	3.47	a. 55	4.75
FRUPENE	11.40	9.61	9.02	9.60	9.55	<b>B.</b> 62	<b>B.74</b>	00.00
FRUPANE	202	50.02	N. N	2.57	2.57	N 40	ନ <b>୍</b> ଯ	0.0
I -GUTANE	0.14	0.03	0.0%	0.13	0.12	0.13	0:12	0.16
1-BUTENE+2-METHYLPROPENE	7.84	ó. 45	ć. 43	6.50	6. 93	6.22	č. 38	6.50
N-BUTANE	2.50	0.51	0.07	0. 96 96	0, 92 0	2.23	2.30	2.71
TRANS-2-BUTENE	ù.16	0.14	0.0 0	0.09	0. IO	0.00	0.11	0.30
CIS-2-BUTENE	0.32	0.11	0.16	0.19	0.20	<b>0.1</b> 8	0.20	Ú. 48
3-METHYL-1-BUTENE	0.50	0.19	0.32	0.44	0.40	0.35	C. GA	0.53
I-FENTANE	0.25	0.28	0.17	0 20 20	0.26	0.20	0.18	0.25
	0 0 0	0.34	0.00	4.60	4.87	4.30	4.42	6.07
Z-METHYL-1-BUTENE	0.20	4.95	0.00	Ú. 16	Ú.1ć	<u>0.14</u>	0.14	0.26
	1.65	0.10	0.00 0.00	1. ċó	1.71	1.53	1.65	2.12
IKANS-Z-PENTENE	0.10	1.54	0.00	0.07	0.07	0.06	0.08	0.21
CIS-2-PENTENE	0.15	0.09	0.00	0.09	0.11	0.03	0.10	0.24
	0000	0.10	0. EÖ	0.00	0.12	0.02	0.02	0.00
HEXENES + 10U-HEXANES	0.46	0.28	0.00.	0.08	0.37	Ф. 2V	ତ <b>.</b> ପ୍ର	ú. 58
1-HEXENE	3.51	3. S	4.36	3.10	3.23	2.63	2.68	4.04
	2.	0.58	1.43	1.16	1.19	1.04	1.08	1.53
HEFTENES + ISU-HEFTANES	5.5	0.41	0.83	0.71	Ú. 65	0.52	0.51	1.21
I THEFTENE	2.07	1.95	2.46	1. EO	1.75	1.34	1.35	2.18
NTHEFIANE COLOI FRING A 100-R	0.67	0 4 4	0.85	0.73	0.74	0.57	0.64	1.04
CO-CLETING + IGU-T + JOTENE	9. / R	0.81	N	0.20	0.47	0. NV	0.30	0.71
A TOO TENSE Nijet Anst			94°0	0.80	0.83	C. 56	0.57	1.01
C9-01 EGINS + 150-5			N	5 0	0.42	0.19	0.33	0.61
Co-creations 130-r				0	0.43	0. N 0	0.37	0.62
AL ANTANE		0.00	0. NG	0.35	N	0.01	0.27	0.41
	ກ ວິ ວິ	9. •	0.13	0.0	0.21	0.13	0.18	0.31
	0. KG	00.00	0.07	6-17	0.16	0.10	0.14	0.19
N - FECENCO Afetane		0.00	0.04	0.14	0.10	Ú. ŬB	0.11	0.19
	N ( 0 ) 0 (	0.46	0.00	0.67	0.62	0.40	0.28	0.41
A TOOL ANGUE MERIET ANGUE	02.0	0.00	0.26	0.38	0.44	0.30	0.100	0.53
N-DULANUNG Linikasini 1777 sasata sasata sasata	1.06	0.44	00.00	0.79	0.¢1	0.50	<b>0.5</b> 2	0.86
UNKNUMN LITE HYDRU-CARB LIU (2) HARANIN HINY UVERS SAFE (3) (3)	14.22	19.75	23.46	20.14	19.15	22.74	22, 89	13.61
UNKNUMN AVI MILMA-LAMB LIG (3) Singsv ssafts nav	14.07	23. 41 23. 41	22°21	10.52	19.43	22.74	20.74	19.36
SCUTAT NEALER WAS	6-1-3-	4.13	З.76	4.04	4.59	4.5%	5.02	12.23

Based on Inter-Reactor Sample
 Collected in Ambient and Chilled Condensers
 Collected in Hot Condenser

### Table A-4 Composition of Pischer-Tropsch Hydrocarbon Phase Oxygenates Run CT-256-3

M.B.No. Days on Stream Component	3-4 6.4 Weight	3-48 50.5 % of	3-61 74.5 Hydrocarbon	3-64 80.5 Phase
METHANOL	0.350	0.650	1 800	1 330
FORMIC ACID	0.033	0.090	0.040	0 120
ETHANOL	1.550	2.310	5.270	4.550
ACETIC ACID	0.140	0.290	0.340	0.510
ACETONE	0.170	0.420	0.540	0.370
PROPANOLS	1.520	2.250	3.940	3.410
PROPANOIC ACIDS	0.058	0.130	0.150	0.280
C4-ESTERS + I-KETONE	0.210	0.460	0.570	0.500
BUTANOLS	0.910	1.390	2.190	1.970
BUTANUIC ACIDS	0.037	0.081	0.070	0.140
CJ-LSTERS + 1-PENTANONE	0.190	0.360	0.410	0.380
renimulo Cé-retere + i-vemoner	U.843	0.842	1.132	0.940
HEYANOLG	0.000	0.000	0.000	0.204
7-FSTFDS + 1-FTTONEC	0.151	0.334	0.638	0.000
UFDWANOIS	0.001	0.001	0.039	0.568
C2-FETERC : I TETRONTO	0.427	0.627	1.117	0.025
CO-LSIERS + I-RETUNES	0.133	0.064	0.234	0.681
	0.583	0.781	1.126	0.416
VONTAGE C	0.356	0.347	0.428	0.669
	0.532	0.667	0.941	1.501
CLU-ESTERS + I-KETONES	0.432	0.447	0.533	0.621
DECANOLS	0.418	0.515	0.747	1.388
CIL-ESTERS + I-KETONES	0.402	0.460	0.517	0.547
C11-NOLS	0.283	0.380	0.558	1.105
C12 PLUS ESTERS + I-KETONES	1.376	1.781	2.249	2.111
C12 PLUS ALKANOLS	0.503	0.883	1.370	2.765
Total, Wt %	11.608	16.561	26.900	27.100
Yield per HC Produced,g/100g	4.793	8.922	12.293	

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### Table A-5 Composition of Fischer-Tropsch Aqueous Phase Organic Oxygenates Run CT-256-3

M.B.No. Days On Stream	3-4 6.4	3-10 12.4	3-25 26.0	3-32 33.5	3-48 50.5
Component		Weight %	of Aque	ous Phase	2
METHANOL ETHANOL ETHANOL ACETIC ACID ACETONE N-PROPANOL I-PROPANOL PROPANOIC ACIDS C4-ESTERS + I-KETONE N-BUTANOL N-2-BUTANOL OTHER BUTANOLS BUTANOIC ACIDS C5-N-METHYL KETONE C5-ESTERS + I-PENTANONE N-1-PENTANOL N-2-PENTANOL OTHER PENTANOLS C5-N-METHYL KETONE N-1-HEXANOL N-1-HEPTANOL N-1-OCTANOL C9+ ALYANOLS	5.01 11.49 0.59 0.67 3.07 0.43 0.16 0.23 1.02 0.11 0.03 0.07 0.00 0.08 0.35 0.00 0.02 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.03 0.00 0.00 0.03 0.00 0.00 0.00 0.03 0.00 0.00 0.00 0.03 0.00 0.00 0.03 0.00 0.03 0.00 0.00 0.00 0.03 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.000 0.0000 0.0000 0.0000 0.0000 0.00000 0.00000000	7.46 15.45 0.39 1.22 4.80 0.87 0.15 0.21 2.32 0.09 0.03 0.08 0.00 0.25 0.72 0.07 0.06 0.10 0.22 0.06 0.01	7.68 14.85 0.15 1.50 5.45 1.01 0.14 0.73 2.34 0.36 0.02 0.15 0.00 0.40 1.11 0.10 0.08 0.19 0.41 0.13 0.03 0.29	7.16 14.26 0.20 1.45 5.27 1.02 0.17 0.68 2.12 0.36 0.02 0.18 0.00 0.37 0.95 0.10 0.37 0.95 0.10 0.33 0.11 0.33 0.11 0.03	10.40 17.45 0.10 2.27 5.10 1.28 0.10 1.09 1.51 0.31 0.03 0.12 0.00 0.46 0.45 0.04 0.05 0.12 0.02 0.12 0.02 0.12 0.25 0.12 0.25 0.12 0.25 0.12 0.25 0.12 0.25 0.12 0.25 0.55 0.25 0.55 0.25 0.55 0.55 0.55 0.55 0.55 0.55 0.55 0.55 0.55 0.55
Total, Wt %	23.47	34.66	37.12	35.24	41.02
Yield per HC Produced,g/100g	0.84		1.34	1.20	1.22
Table A-6 Composition of Fischer-Tropsch Keactor Wax Run CT-256-3

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Oays On-Stream	Ŷ	20. B	23.8	35	42.4	51.5	60.8	68.8	71.7	83.3
Press., MPa	1.48	1.48	1.48	1.48	1.48	1.48	1.82	1.48	2.17	2.51
Tamp., oC	260	260	260	260	260	262	266	267	267	267
Carbon No.					Weighl	%		•.		-
13-20	5.39	10.75	10.34	9.20	12.23	11.06	10.51	8.51	9.92	9.54
21-25	18.14	22.27	21.22 .	19.47	23.01	19.86	17.80	13.69	14.73	18.27
26-30	25.25	29.23	27.92	27.95	25.42	26.08	28.10	26.75	24.72	25.61
31-35	22.84	20.19	21.90	21.81	18.63	20.79	22.44	25.24	23.87	21.18
36-40	12.45	8,86	11.21	12.43	10.31	11.58	13.20	14.26	14.74	13.33
41-45	8.14	3.43	4.86	5.43	4.86	4.96	5.09	7.45	7.36	7.35
46-50	4.15	2.41	1.82	2.45	2.83	2.70	2.11	3.60	3.99	3.72
5155	1.50	2.01	0.73	1.06	1.69	1.59	0.76	0.50	0.66	1.00
56-60	0.79	0.61	0.00	0.60	1.01	0.97	0.00	0.00	0.00	0.00
61-67	1.36	0.00	0.00	0.61	00.00	0,00	0.00	0.00	0.00	0.00
1ol Avg C No. °eak C No.	30.0 35	27.4	27.6 28	28 <b>.</b> 3 28	27 <b>.</b> 3 26	27.8 28	27.8 28	29. 1 30	28.7 31	28 <b>.4</b> 30.4

Second-Stage Fixed-Bed ZSM-5 Reactor Operating Conditions and Material Balances Run CT-256-3 Table A-7

3- 19 20-0 1.446 3096 80.64 91.62 87.14 22.59 67.73 67.73 1.37 0.85 7.45 100 100 100 223 2.5% 5.60 7.68 0.00 48. 63 6. 61 7. 14 37. 57 0. 689 2559 1. 480 3. 797 2. 494 2. 494 6.00 3- 13 15.4 0. 684 260 1. 480 3. 956 2. 598 2. 598 79.50 90.44 86.00 23.04 66.44 7.01 8.51 00.24 005.34 200 230 368 424 1.439 3214 3214 0.57 1.63 13.17 3.77 39.41 20.73 8.28 31.58 0.688 259 3.946 2.578 2.578 2.578 115.45 3- 12 14.4 20.16 69.69 1.08 8.13 8.13 0.31 0.85 0.85 11.14 10.10 369 401 1.446 **ć.** 4 76.47 89.59 84.48 3611 ରିତିରିତି 0.681 260 1.460 3.948 2.600 2.600 3~ 11 13.4 73.40 91.56 86.23 20.21 69.90 1.16 0.97 0.97 7.74 7.74 100 03.53 0.16 0.75 5.40 2.74 35.06 52.73 1.66 10.54 4516 949 949 949 5.4 0. 681 260 1. 480 3. 943 2. 610 2. 610 3- 10 12.4 1.446 2833 80.96 92.49 87.82 23.19 67.20 0.93 0.93 7.43 7.43 7.43 7.652 208 0.02 0.66 0.65 0.05 329 346 55.52 62.42 0.21 1.65 4.4 41.70 44.55 1.78 11.58 3- 9 11.3 0.679 260 1.480 3.920 2.569 2.569 302 332 332 332 332 3160 3160 3.4 79.10 91.19 86.30 20.40 69.62 0.98 0.94 8.07 103.60 199 6, 50 10, 56 0.23 1.02 6.36 9.22 ř 3- 8 10.3 0.680 259 1.480 3.923 2.580 2.580 6.3 316 357 1.446 3143 2.4 78.78 91.59 66.40 20.53 69.37 1.34 0.96 7.80 100 102.52 198 6. %8 9. 82 9. 13 9. 13 9. 13 10. 45 0.42 1.27 10.09 5.74 46.36 32.59 3.82 17.23 ~ °.° 0.682 260 1.460 3.883 2.535 2.535 5.8 80.59 92.49 87.67 21.83 68.97 0.99 0.93 7.28 100 98.33 98.33 332 378 378 1.432 2928 2928 0.76 ରିଚିତିତି ĥ 18.72 72.07 1.29 1.29 0.94 6.97 111.93 111.93 3-8,3 8,3 0.685 259 1.480 2.563 2.563 7.1 77.30 91.70 85.84 0.81 2.03 14.53 1.34 330 330 330 3302 3502 0.4 (1) ରିରିରିରି Bal Recovery. Wt % of Charge: 9 HC/Nm3 (H2+CO) conv.: Selectivities, Wt % of HC : , Nol % : U H2+CO Yields, Wt % of Products : Hydrocarbons C02 H20 H20 C0 ... Charge H2/CO (Molar) Temperature, cC Pressure, MPa Feed Sup. Vel., cm/s Space Vel., NL/9Fe-hr N2 in Feed, Mol X Second-Stage Conditions: First-Stage Conditions: 12+ (Excl. Rx. Wax) i-C4/(C3= + C4=) Molar (C3/C3=) Molar Ratio Alkylate, Wt % of HC Cat-Poly,Wt % of HC C5 - C11 PONA, Wt % : Paraffins Nitrogen-Free Basis) Outlet, oć Temp., Inlet, oC Cl2+ (Ежс]. Кж. Slurry Кж. Wax Total GHSV, hr D<sub>-</sub>ys On-stream Days On-stream Pressure, MPa Olefins Naphthenes Aromatics i-Butane n-Butane M.E. No. C5 - C11Ethene Ethane Propene Butenes Methane Total

0

347 385

All MB's except MB H & are adjusted for inter-reactor sampling Not Available 39

Table A-7 (Contd.) Second-Stage Fixed-Bed ZSM-5 Reactor Operating Conditions and Material Balances Run CT-256-3

3- 28 29.0 0.694 261 3.663 2.3883 2.3883 2.3883 6.3 334 379 1.398 2976 11.2 1.15 0.90 8.37 100 105.72 224 79.83 90.64 22.65 66.92 86.21 0.72 13.90 2.69 45.96 23.88 5.59 24.57 28.0 0. 692 261 1.480 3.683 2.409 329 2845 80.48 91.25 86.85 5 ć.5 22.26 67.37 1.13 0.92 8.31 100.19 1.411 206 0.65 1.53 2.59 46.66 24.11 5.85 23.38 ĥ 1.411 2978 9.2 3- 26 27.0 0.685 260 1.480 3.800 3.800 79.12 89.63 65.36 321 366 19.79. 67.97 6.7.97 1.30 0.99 9.96 9.96 9.26 99.26 8.00 8.00 9.25 10.45 10.45 10.05 100 6.3 0.51 1.42 11.94 11.94 44.19 29.17 3.86 22.78 3- 25 314 358 3110 79.06 89.53 85.26 0.669 260 1.480 3.804 2.489 20.29 67.32 1.99 0.94 9.46 100 100 100 202 6.6 1.411 0.51 1.53 11.50 47.36 31.83 4.41 16.39 3- 24 25.0 0.685 260 1.480 3.817 2.497 2.497 312 356 356 1.446 3096 7.2 78.71 90.33 65.61 19.98 69.03 1.17 0.96 8.86 100 100 195 45.66 28.66 4.87 20.81 0.54 1.52 11.33 3.62 21.59 68.09 1.02 0.92 8.39 103.47 103.47 209 3- 23 24.0 0.687 259 1.487 3.810 2.516 6.3 308 353 79.82 90.86 86.36 1.446 3051 6. 2 51.02 26.08 3.44 19.46 0.64 1.85 11.64 2.39 6.77 0.95 1.96 5.60 5.74 6.74 5.28 5.28 8.91 100 3- 22 23.0 0.685 260 3.820 2.507 2.507 306 356 1.446 3101 5.2 20.95 67.90 1.27 0.91 8.97 105.02 105.02 79.59 90.06 85.80 0.92 2.73 12.72 0.41 21.46 4.13 23.32 51.09 3- 21 22.0 0.683 260 3.807 2.502 2.502 6.3 315 60.14 91.40 86.63 21.21 68.75 68.75 1.38 0.88 7.77 1.00 100 3 1.446 3077 207 11.09 4.2 1.50 3.39 0.00 52.97 10.04 6.00 31.00 3- 20 0.689 259 1.480 3.803 2.498 2.498 1.446 3108 3.2 332 79.78 90.95 86.39 20.42 69.26 1.34 0.90 8.08 100 105.87 2.03 9.22 0.00 47.13 7.41 7.44 38.02 Bal Recovery, Wt % of Charge! ... Temperature, oc Fressure, MPa Freed Sup. Vel., cm/s Space Vel., NL/gFe-hr N2 in Feed, Mol X Second-Stage Conditions: Temp., Inlet, oC •• of HC Yields, Ht % of Products First-Stage Conditions: Charge H2/CO (Molar) 1-C4/(C3= + C4=) Molar : (C3/C3=) Molar Ratio : Alkylate, 44 % of HC : Cat-Poly, W1 % of HC : C5 - C11 PONA, W1 % : .... conv. Cl2+ (Excl. Kx. Wax) Slurry Rx. Wax Nitrogen-Free Basis) ... g HC/Nm3 (H2+CO) cor Selectivities, Wt 2 Outlet, oC 2 Conversions, Mol Days On-stream Days On-stream Pressure, MPa GHSV, hr Hydrocarbons Paraffins Butenes 1-Butane n-Butane M.B. No. C5 - C11 Ethene Ethane Propeng Methane Olefins Propane H2+C0 H20 H20 C0 C0 Total [ota] 20

6 are adjusted for inter-reactor sampling (I) All MB's except MB #

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Aromatics

. Table A-7 (Contd.) Sacond-Stage Fixed-Bed 2SM-5 Reactor Operating Conditions and Material Balances Run CT-256-3

3- 37 38.5 1.480 3.480 2.273 2.273 373 415 415 2845 2845 20.6 61.25 91.53 87.32 7.54 8.24 7.55 7.38 7.38 7.38 7.38 7.38 7.38 7.31 7.31 22.13 67.91 1.43 0.84 7.55 100 100 100 215 215 0.695 259 42.46 14.27 8.35 34.92 0.79 2.01 16.57 1.63 3- 36 37.5 0.674 260 1.487 3.468 2.275 6.9 383 418 1.377 2822 19.6 80.80 91.39 87.13 20.29 69.60 1.24 0.87 0.87 8.00 102.31 193 8. 44 3. 52 5. 53 5. 50 5. 53 5. 50 5. 53 5. 53 5. 53 5. 53 5. 53 5. 53 5. 53 5. 53 5. 53 5. 54 5. 555 0.83 2.11 17.85 1.41 40.50 11.71 8.68 39.12 3- 35 36.5 0.685 260 1.494 3.448 2.267 2.267 81.94 52.07 87.95 22.98 67.85 1.25 0.60 7.12 7.12 105.42 223 378 2817 18.6 0.89 16.69 0.78 0.78 43.90 13.10 8.86 34.14 115.1 0.655 259 1.480 3.481 2.273 7.0 3- 34 35.5 373 2840 17.6 80.68 92.24 87.50 1.411 23.00 67.74 1.43 0.87 6.96 100 105.40 0.23 44.93 13.11 9.15 32.80 8.08 1.10 3.45 3.59 3.59 4.22 4.22 7.45 7.97 7.97 7.97 100 21.82 68.82 1.25 0.85 7.25 100 3- 33 34.5 0.696 260 1.446 3.549 2.253 2.253 2.253 368 412 0.101 2812 16.6 81.11 91.91 87.48 05.46 2.52 0.00 42.46 12.49 9.27 35.79 211 3- 32 33.5 0.702 260 1.480 3.499 2.279 7.0 363 1.411 2631 15.6 80.81 91.63 87.17 22.73 67.52 67.52 1.31 0.68 7.56 100 100 218 1.07 2.53 16.19 0.00 44.13 12.59 8.76 34.12 3- 31 32.5 0.679 260 3.489 2.282 2.282 1.446 2690 14.6 361 405 22.10 66.93 64.93 1.14 0.91 8.92 8.92 100 105.18 79.40 90.09 85.77 1.04 0.05 0.05 44.62 14.23 8.55 32.40 3- 30 31.5 0.686 260 1.487 3.614 2.373 6.8 1.467 3109 13.6 352 356 76.71 86.59 82.81 64.47 1.13 1.03 1.03 11.66 105.67 0.95 2.18 6.37 0.31 45.77 15.26 8.21 30.76 21.71 0.665 261 1.480 3.554 2.343 2.343 3- 29 30.5 347 3246 72.84 60.66 77.67 21.78 58.94 1.16 105.89 243 1.460 0.84 1.84 15.65 1.12 47.65 17.64 7.75 26.97 Bal Recovery, Wt % of Charge: 9 HC/Nn3 (H2+EG) conv.: Selectivities, Ht % of HC ; Yields, Wt % of Froducts : Hydrocarbons CO2 H20 H20 H2 CO Total Second-Stage Conditions: .. .. i-C4/(C3m + C4=) Molar: (C3/C3=) Molar Ratio Alkylate, Ht X of HC Cat-Poly, Mt X of HC C5 - C11 FGNA, Mt X : First-Stage Conditions: Charge H2/CO (Molar) Temperature, oC Pressure, MPa Feed Sup. Vel., cm/5 Space Vel., NL/9Fa-hr N2 in Feed, Mol 2 C12+ (Excl. Rx. Wax) Slurry Rx. Wax Total Nitrogen-Free Basis) Temp., Inlet, oC Gutlet, oC N Conversions, Mol Days On-stream GHSV, hr Days On-stream Pressure, MPa Naphthenes Paral fins Aromatics n-Butane CS - C11 M.B. No. i-Bulane Olefins Methane Fropena Fropane Ethene Ethane Butenes 엎

All MB's except MB # 6 are adjusted for inter-reactor sampling

Table A-7 (Conld.) Second-Stage Fixed-Bed 25M-5 Reactor Operating Conditions and Material Balances Run CT-256-3

0.90 0.72 0.72 17.23 100 104.40 203 66.5 0.6%3 265 1.535 2.773 2.773 1.923 1.923 1.923 344 355 355 355 1.342 2386 13.7 84.21 51.00 82.31 43.71 16.36 7.97 31.96 20.02 61.14 0.52 2.06 16.53 0.57 3- 57 1.41 1.06 16.48 160 100 105.87 236 3- 56 64.5 266 2.519 2.036 1.308 2772 11.7 397 76.12 81.51 79.31 22.02 59.04 0.63 1.93 14.64 1.14 43.11 18.02 7.86 31.02 <u>°</u> 0.691 3- 55 62.5 1.64 1.24 21.82 100 327 381 1.308 2907 2907 72.38 76.28 74.70 19.75 55.56 03.14 0.675 264 1.591 2.917 2.917 2.067 8.7 10.98 4.76 4.76 5.29 5.52 5.52 5.52 5.52 5.98 7.53 2.41 2.41 2.41 0.69 1.59 14.38 2.46 44.67 22.21 6.70 26.42 20.29 67.90 1.42 9.47 9.47 101.59 8- 54 59.5 0.687 264 1.508 3.171 2.063 2.063 313 361 1.294 1.294 2604 2604 80.04 89.87 85.87 0.75 1.83 13.74 1.76 39.93 17.26 6.79 36.02 ė 304 353 353 1.301 2648 2648 3- 53 0.675 264 3.247 3.247 2.095 7.8 21.88 66.50 1.16 0.90 9.56 9.56 100 103.48 214 79.94 89.57 85.69 0.68 2.33 12.94 0.65 50.78 20.57 5.55 23.10 ĥ 3- 52 55**.**5 0.686 263 3.166 2.094 2.094 303 357 1.322 2659 2.6 20.57 67.46 1.32 0.93 9.72 103.01 199 49.38 12.78 6.44 31.40 79.62 89.44 85.45 1.37 3.12 11.53 0.00 0.687 262 1.508 3.163 2.101 2.61 3- 51 53.5 317 362 1.322 49.96 8.70 8.13 33.22 2636 0.6 79.29 68.86 84.96 105.62 22.58 64.88 1.63 0.92 9.99 4.07 9.42 4.85 0.00 ឆ 3- 39 40.5 0.693 260 3.462 2.238 2.238 21.36 68.48 68.48 1.33 0.87 7.96 100 101.94 101.94 200 22.6 394 419 1.377 81.41 91.46 87.35 0.68 1.76 6.79 3.06 43.15 15.58 8.60 32.28 ĥ 3- 38 39**-5** 0.679 2.260 2.282 392 418 1.398 22.25 68.11 5.11 1.15 0.82 7.63 100 100 100 215 7.30 4.46 4.46 4.46 4.72 45.27 45.27 45.27 45.23 45.23 150 150 81.60 91.61 87.57 2764 0.75 1.93 16.36 2.09 42.89 16.38 7.63 33.10 Bal Recovery, Wt % of Charge: g HC/Nm3 (H2+C0) conv.: Selectivities, Wt % of HC : H2+CO Vields, Wl % of Products : Hydrocarbons CO2 H2O H2 CO Second-Stage Conditions: First-Stage Conditions: Charge H2/CO (Molar) Temperature, oC Pressure, MPa Feed Sup. Vel., cm/s Space Vel., NL/9Fe-hr N2 in Feed, Mol X (C3/C3=) Molar Ratio : Alkylate, Wt X of HC : [2+ (Excl. Rx. Wax) [-C4/(C3= + C4=) Molar Nitrogen-Free Basis) ..... Cat-Poly, Wt X of HC C5 - C11 PONA, Wt X Temp., Inlet, aC Outlet, aC Conversions, Mol 2 **Days On-stream** Days Un-stream Slurry Rk. Wax Pressure, MPa Olefins Naphthenes Paraffins M.E. No. Aromatics GHSV, hr Ethane Propene Propane Butenes i-Butane n-Butane C5 - C11 Methane Ethene Total Total Ĥ 8 ū

(1) All MB's except MB # 6 are adjusted for inter-reactor sampling

# Table A-7 (Contd.) Second-Stage Fixed-bed 2SM-5 Reactor Operating Conditions and Material Balances Run CT-256-3

2.515 3707 27.7 75.18 76.52 76.02 3- 64 80.5 0. 603 265 2.515 2.58 3.028 3.028 2.9 408 442 16.60 58.16 1.51 1.51 1.04 1.04 22.68 22.68 22.68 100 99.16 7.72 2.77 9.25 9.46 7.44 1.72 0.20 0.75 9.71 4.61 40.36 6.31 100 32.14 32.04 5.87 29.95 1.41 6.41 3- ć3 78.5 0.634 264 1.365 1.602 1.602 1.602 399 442 2.370 1665 25.7 21.41 68.38 0.90 0.37 0.37 0.37 0.37 0.37 0.37 0.37 100 100 7.70 4.08 9.60 7.34 7.35 6.65 7.34 8.65 7.34 8.37 9.37 9.37 91.38 90.55 90.87 0.70 13.81 19.89 39.77 12.19 7.49 40.55 12.23 å 3- 62 0.600 266 2.515 2.617 2.612 3.3 392 439 2.384 23.761 23.7 83.78 87.41 87.92 19-94 66-50 0-92 0-47 12-17 12-17 99-13 0.78 2.43 15.15 1.67 37.95 15.10 7.62 39.33 189 i N 0.5% 266 2.515 2.537 3.014 2.8 424 2.390 3677 21.7 3- 61 74.5 78.83 76.54 77.40 18.31 58.02 1.08 0.84 21.74 100 103.23 206 0.69 1.90 15.33 2.62 377 38.62 16.53 7.49 37.36 9, 14 1, 16 4, 311 3- 60 72.5 2.5%3 2.5%3 2.616 4.5 0. 607 265 370 418 .032 3216 19.01 59.69 1.28 0.80 0.80 19.22 19.22 19.22 103.10 39.06 14.31 7.98 38.65 80.32 79.26 79.66 0.83 2.22 1.25 3- 59 70.5 0.601 266 2.170 2.564 2.564 2.567 2.567 412 100 101.50 218 2.032 3061 17.7 81.91 60.54 81.05 20.70 59.08 1.13 0.74 18.35 0.98 2.48 5.50 0.15 40.40 17.85 7.50 34.25 8.39 0.91 3.97 3.97 8.41 8.41 8.41 8.61 7.17 7.17 2.01 4.04 100 361 1.10 0.67 15.97 15.97 100 104.87 209 3- 58 68.5 0.601 265 1.825 2.605 2.186 2.186 354 402 1.646 2717 15.7 82.99 82.45 82.65 19.55 9. 46 0. 92 8. 71 8. 71 9. 01 9. 01 9. 01 10 0. 88 0. 88 0. 88 0. 88 1.07 2.76 5.60 0.00 42.84 13.34 7.97 35.66 of Charge: .. •• Pressure, MPa Feed Sup. Vel., cm/s Spaca Vel., NL/gFe-hr N2 in Feed, Mol % Second-Stage Conditions: of HC Vields, Wt % of Products Hydrocarbons CO2 H2O canv.: First-Stage Conditions: Charge H2/CO (Molar) C124 (Exel. Rx. Wax) i-C4/(C3= + C4=) Molar (C3/C3=) Molar Ratio Alkylate, Ut % of HC Cat-Poly, Ut % of HC C5 - C11 PONA, Wt % 3 (Nitrogen-Free Basis) M.B. No. g HC/Nm3 (Hztuu, Lu.) Selectivities, Ht 2 Outlet, oC GHSV, hr Days On-stream Conversions, Mc1 2 N Temp., Inlet, oC Temperature, oC Bal Recovery, Wt g HC/Nm3 (H2+CO) Days On-stream Slurry Rx. Wax Total Pressure, MPa Naphthenes Paraffins Aromalics i-Butane n-Butane Ethene Ethane Propene Butenes C5 - C11 Methane Olefins H2+C0 **Total** NS 19 Ĥ

sampling are \* (1) All M6's axcept MB

for inter-reactor adjusted Ŷ

TABLE A-8

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SECOND-STAGE ZSH-5 REACTOR RAW GASOLINE<sup>(1)</sup> PROPERTIES

59.5 5 57.5 G 33.5 2 26 27 2 23 (Run CT-256-3) 21 33 20 33 5 8 13.4 Ħ 11.3 ი

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3

65

5

5

0.20 0.75 .702 0.11 22.4 32.0 6.0 35.8 91.1 80.7 32 123 183 183 238 238 238 1.0 80.5 0.83 2.32 .769 0.13 33 123 183 236 236 238 2.5 72.5 30.2 13.9 10.8 45.1 92.0 02.8 0.98 2.48 .770 0.05 31 124 105 105 237 241 0.9 3.1 70.5 28.6 19.1 10.1 43.2 92.7 82.7 1.07 2.77 .781 0.04 37 128 184 220 220 2.0 2.0 92.3 82.5 1 1 1.1 68.5 0.92 2.06 .784 0.1 42 133 184 222 249 0.9 66.5 29.1 14.1 11.8 45.0 94.0 81.8 0.75 1.83 .765 0.4 37 128 185 219 219 242 1.3 25.9 16.8 12.2 45.1 87.9 79.5 0.88 2.33 .767 0.15 39 130 186 218 218 246 1.4 33.2 21.2 13.1 87.6 79.7 1.07 2.53 .782 0.04 27.6 11.9 12.2 48.3 36 130 189 236 255 1.3 93.9 03.5 0.51 1.42 .759 -35 128 187 212 248 1.0 2.0 87.4 79.0 32.5 28.0 11.0 28.5 0.92 2.72 .763 86.9 70.4 39 127 184 212 212 245 1.1 36.8 21.5 11.7 30.0 1.5 3.39 .771 -37 128 185 224 223 2.23 2.23 2.23 10.3 10.3 11.4 11.4 80.4 80.4 2.03 4.64 .769 -32 125 184 2318 251 251 2,3 34.5 7.8 11.0 46.3 92.0 82.8 2.56 5.61 .726 ... 36 130 192 245 245 263 2.6 30.4 4.6 10.2 54.8 09.6 83.4 0.16 0.75 .729 -33 120 189 232 247 1.1 85.1 83.4 24.4 50.2 10.9 14.5 0.23 85.4 75.6 39 126 187 231 231 250 1.1 27.5 43.8 12.8 15.9 0.42 1.27 .757 0.15 89.J 33.2 32.3 11.6 22.9 10.3 35 182 182 239 239 3 3 0.76 1.91 .760 0.15 9.3 90.5 80.8 1111 36 184 184 240 240 Second-Stage Severity: ASTH Distillation, <sup>°</sup>C: IBP 50 Vol **1** 95 Vol **1** 95 Vol **1** ReaidDe, Vol **1** Loss, Ht **1** Octane Numbers: Paraffins Olefins Naphthenes Aromatics RONA, WE NE H.B. No. R+0 RIG SOS

(1) Collected in ambient and chilled condensers. Hydrocarbons collected in hot condenser was very small.

# Table A-9 Composition of Hydrocarbon Products from Two-Stage Slurry F-T/25M-5 Syngas Conversion Fun CT-250-3

M.B.No. Days ûn Stream	3-0 3.3	3-7 9.3	3-8 10.3	3-0 11.3	3-10 12.4	2-11 13.4	3-12 14.4	3-13 15.4	3-19 20.0
METHANE	S.10	s.57	o. 98	0.90	5.36	7.13	7.44	0.05	7.13
ETHENE	0.90	0.75	0.32	1.07	2.13	1.45	1.50	1.15	0.55
ETHANE	3.70	3.07	3.19	3.17	2.03	3.30	3.32	2.37	3.09
FROPENE	3.32	2.39	3.40	3.53	2.78	4.72	3.34	4.32	1.54
FRUPANG LEDHTANG	7 73	5.75	4.00	2.7*	1.73	3.67	5.83	1.38	9.05
I-BUTENE+2-METHYL PROPENE	2.99	2.83	14.41	5.40	3.08	2.70	5.0/	7 47	10.00
N-BUTANE	0.10	5.00	5.29	4.18	2.04	3.76	5.23	5.81	3.22
TRANS-2-BUTENE	1.10	1.07	1.58	2.04	2.33	2.49	2.14	1.33	Q.47
	0.74	0.73	1.15	1.37	1.50	1.63	1.49	0.98	0.32
I-PENTANE	4.35	4.08	4.22	3.02	0.40	2.35	2.49	4.76	7.52
1-PENTENE	0.09	0.10	0.20	0.27	0.27	0.33	0.23	0.16	0.04
2-RETHYL-1-BUTENE	0.45	0,00	1.13	1.74	1.24	2.10	1.20	0.73	0.22
N-PENTANE TRANS- 2-DENTENE	2.00	3,20	4.12	3.37	1.28	2.39	2.32	3.49	4.44
CLORD-PENTENE	0.13	0.00	0.77	0.55	6.38	1.+1	0.32	0.57	0.15
2-METHYL-2-BUTENE	0.99	1.32	3.05	4.77	7.74	5.09	2 20	1 37	0.07
CS-DIOLEFINS (DIENES)	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
2, 2-DIMETHYLBUTANE	0.00	ù. U4	0.04	0.03	<b>0.00</b>	0.03	0.07	0.06	0.02
CYCLOPENTANE	0.04	0.00	0.08	0.03	0.00	0.12	0.07	Ú. 26	9.19
2. 2-RIMETHYL BUTANE	0.07	0.07	0.04	0.26	0.00	0.00	0.24	0.06	0.00
2-METHYLPENTANE	0.91	1.00	2.65	1.97	0.42	1.33	1.00	1.30	2.73
3-METHYLPENTANE	0.31	0,60	0.52	0.63	0.17	0.43	0.29	0.85	1.17
HEVENES	0.00	0.00	1.53	2.45	4.07	3.05	0.00	1.01	0.19
1-HEIENE N-LEYANE	0.05	0.09	0.56	.0.30	0.45	0.26	0.22	0.09	0.02
2. 4-DIMETHYLPENTANE	0.07	0.00	0.01	0.00	0.00	0.00	0.90	0.01	1.31
METHYLCYCLOPENTANE	0.14	0,33	0.22	0.07	0.03	0.07	0.00	1.19	0.39
3. 3-DIMETHYLPENTANE	0.00	Ú.ÚÚ	<b>v.</b> 00	0.00	0.00	0.00	0.00	0.00	0.00
CYCLUHEIANE	0.00	Ú.00	0.02	0.00	0.00	0.00	0.00	0.03	0.02
HEPTENES + ISU-HEPTANES	0.20	0.68	0.00	0.00	4.36	0.00	0.00	0.68	0.20
2-METHYLHEXANE	0.09	0.40	0.79	0.55	0.25	0.44	0.00	0.00	0,00
2, 3-DIHETHYLPENTANE	0.00	0.03	0.08	0.04	0.01	0.04	0.00	0.19	0.13
3-METHYLHEXANE	0.07	0.39	0.70	Ú. 10	0.23	Ŭ, 37	0.00	0.58	0.94
1-CIS-3-DIMETHYL-N5	0.02	0.10	0.14	0.05	0.00	0.05	0.00	0.37	0.29
	0.00	0.00	0.19	0.05	0.01	0.05	0.00	0.26	0.21
S-ETHYL-PENTANE	0.00	Ú. 55	0.00	0.00	0.00	0.00	0.00	0.00	0.00
N-HEPTANE	0.11	0.59	1.35	1.27	1.03	1.20	0,00	1,34	0.03
C7-OLEFINS	0.00	0.00	2.21	3.08	5.12	3.93	0.00	1.14	0.24
METHYLCYCLDHEXANE	0.00	0.07	0.15	0.07	0.01	0.06	0.00	0.19	0.15
L-OCTENE	0.04	0.00	0.00	0.00	0.00	0.00	0.00	0.31	0.00
MONOMETHYL-ISO-C3-P	0.00	0.00	1.22	0.97	0.35	0.72	0.00	0.49	0.67
OTHER ISO-C3-P	0.00	0.00	0.10	0.05	0.01	0.04	0.00	0.15	0.12
C3-OLEFINS	0.00	0.00	4.20	4.70	4.01	5.47	0.00	1.00	0.33
US-NAPHTHENES (NS+N6)	0.00	0.00	0.74	0.30	0.03	0.28	0.00	1.12	1.01
C9-01.EFINS + 130-2	0.03	0.13	0.00	1.25	1.30	1.29	0.00	0.72	0.22
MONOMETHYL-120-C7-P	0.00	0.00	0.76	0.00	0.32	0.48	0.00	0.33	0.30
OTHER ISO-C7-P	0.00	0.00	0.15	0.02	0.02	0.08	0.00	0.12	0.13
C9-OLEFINS	0.00	0.00	1.?7	2.57	2.02	2.93	0.00	0.03	0.13
N-NONANE	0.00	0.00	0.39	0.25	0.02	0.15	0.00	0,31	0.26
150-C10-P + 0 + N5 + N6	0.00	Ú. 00	2.49	3.01	7.33	3.24	0.00	1.11	0.03
BENZENE	0.09	0.25	0.31	0.27	0.21	0.33	0.00	0.80	ea,0
TOLUENE	0.07	0.38	0.99	0.42	0.12	0,42	0.00	2.90	3.74
ETHYLBENZENE	0.00	0.44	0.30	0.11	0.07	0.09	0.00	1.10	1.52
M-IYLENE	0.00	0.00	1.03	0.13	0.02	0.10	0.00	2 10	0.00
0-AYLENE	0.00	0.00	0.40	0.46	0.07	0.73	0.00	1.03	1.01
ISOPROPYLBENZENE	0.00	0.00	0.00	0.00	0.00	0.15	0.00	0.00	0.00
N-PROPYLBENZENE	0.00	0.00	0.18	0.12	0.04	0.10	0.00	0.16	0.11
1-DETHYL-3-ETHYL-BENZENE	0.00	0.00	1.59	0.38	0.02	0.62	0.00	2.21	2.31
1-METHYL-2-ETHYLBENZENE	0.00	0.00	0.05	0.10	0.00	0.14	0.00	0.08	0.03
130-C4-SENZENE	0.00	Ů.00	0.00	0.05	0.00	0.04	0.00	0.00	0.20
SEC-C4-BENZENE	0.00	0.00	0.00	0.17	0.00	0.20	0.00	0.00	0.00
1. 2. 4-TRIMETHYLBENZENE	0.00	0.00	0.96	0.48	0.07	0.71	0.00	1.53	1.52
1.3-DIETHYLBENZENE	0.00	0.00	0.00	0.00	0.00	0.03	0.00	0.03	0.07
I-METHYL-3-N-C3-BENZENE	0.00	0.00	0.52	0.33	0.00	0.00	0.00	10.11	0.41
N-C4-BENZENE	0.00	0.00	0.19	0.16	0.00	0.00	0.00	0.11	0.00
I, 2, 3-TRIMETHYLBENZENE	0.00	0.00	0.02	0.04	0.00	0.01	0.90	0.08	0.13
LANETHYLADANACOARD	0.00	0.00	0.00	0.00	0.00	0.06	0.00	0.00	0.11
C10-ALKYLBENZENES	0.00	0.00	0.70	0,02	0.00	0.00	0.00	0.00	0.00
1. 2. 4. 5-TETRAMETHYLBENZENE	0.00	0.00	0.05	0.01	0.01	0.03	0.00	0.07	0.09
L. 2. 3. 5-TETRAMETHYLEENZENE	0.00	0.00	9.00	0.00	0.00	0.00	0.00	0,04	0.07
1. 2. 3. 4-TETRAMETHYLBENZENE	0.00	0.00	0.00	0.03	0.00	0.01	0.00	0.01	0.02
UI 1-ALKYLBENZENES	0.00	0.00	0.30	1.09	0.19	0.05	0.00	0.88	1.07
UNKNOWNS (HC AROMATICS)	0.00	0,00	0.00	0.00	0.00	0.01	0.00	0,00	0.00
METHANOL	0.00	0.00	0.00	0.00	0.02	0.00	0,00	0.00	0.00
DIMETHYL ETHER	0.00	0.00	0.00	ö. 🦾	0.00	0.00	0,00	0.00	0.00
I-PROPANOL	0.00	0.00	0.00	0.00	0.12	0.00	0.00	0.00	0.00
UNKNUWN LITE MYDRO-CARB LIQ ()	0 33.01	32.31	0.00	0.00	0.00	0.00	28.94	0.00	0.00
ILURRY REACTOR WAX	10-69	10.43	10,43	10.20	13.31	3.11 -2,29	9,80	9.00	2.72

(1) Collected in Amoient and Chilled Condensers

# Table A-° (Contd.) Composition of Hydrocarbon Products from Two-Stage Slurry F-T/ZSM-5 Syngas Conversion Run CT-256-2

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M.B.No. Days On Stream	3-20 21	3-21 22	3-22 23	3-23 24	3-24 25	3-25 26	3-26 27	3-27 28	3-29 29
METHANE	7.83	7.46	6.77	7,27	8.20	7.37	s.00	7.33	7.21
ETHENE	0.53	0.55	0.95	0.60	0.74	0.71	0.86	0.81	0.81
PROPENE	3.34	3.11 1.99	2.99	2.97	3.27	3.17	3.29	3.01	3.00
PROPANE	8.55	7.04	5.00	4.92	4.93	5.01	5.29	5.27	5.72
I-BUTANE 1-BUTENE+2-METHVI PROPENE	10,12	9.96	6.74 7.70	5.95	5.76	5.85	6.07	6.75	7.29
N-BUTANE	7.83	7.37	6.28	5.81	3.70	4.20	5.43	3.40	5.07
TRANS-2-BUTENE	0.58	0.73	1.03	1.29	1.40	1.61	1.53	1.31	1.27
CIS-2-BUTENE	0.39	0.50	0.70	0.88	0.93	1.08	1.03	0.89	0.97
I-PENTANE	6.64	7.00	5.87	5.27	4.34	4.92	4.37	N. 11	5.44
1-PENTENE	0.04	0.06	0.11	0.16	0,15	0.18	0.15	0.14	0.14
2-METHYL-1-BUTENE	0.26	0.44	0.71	0.97	0.92	1.13	0.89	0.81	0.79
TRANS-2-PENTENE	4.12	5.10	4.86	4.71	3.71	4.20	3.45	4.11	4.21
CIS-2-PENTENE .	0.09	0.14	0.23	0.32	0.28	0.37	0.28	0.27	0.26
	0.67	1.13	1.87	2.65	2.34	2.95	2.21	2.07	2.00
CYCLOPENTANE	0.14	0.03	0.03	0.03	0.02	0.03	0.05	0.04	0.09
HEXENES + 1SO-HEXANES	0.00	0.03	0.08	0.13	0.04	0.04	0.08	0.08	0.03
2.3-DIMETHYLBUTANE	0.13	0.15	0.09	0.11	0.09	0.10	0.08	0.09	0.15
3-METHYLPENTANE	1.01	1.27	1.11	1.02	2.08	2.34	1.70	2.30	2.33
HEXENES	0.35	0.42	0.89	1.18	1.42	2.07	1.20	0.90	0.30
	0.02	0.05	0.08	0.15	0.07	0.08	0.06	0.10	0.10
2.4-DIMETHYLPENTANE	1.85	2.76	3,10	3.26	2.27	2.79	2.05	2.59	2.55
METHYLCYCLOPENTANE	0.58	0.41	0.26	0.20	0.27	0.27	0.19	0.56	0.62
3, 3-DIMETHYLPENTANE	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
NEPTENES + ISD-HEPTANES	0.03	0.03	0.02	0.02	0.02	0.02	0.02	0.02	0.02
2-METHYLHEXANE	0.34	0.92	0.92	0.97	0.79	0.98	0.82	1,00	0.96
2. 3-DIMETHYLPENTANE	0.17	0.13	0.09	0.08	0.08	0.09	0.09	0.12	0.13
3-ALINTLALXANL 1-CIS-3-DIMETHVI-N5	0.81	0.76	0.78	0.72	0.80	0.80	0.73	0.85	0.24
1-TRANS-3-DIMETHYL-NS	0.27	0.21	0.15	0.09	0.13	0.20	0.11	0.18	0.17
1-TRANS-2-DIMETHYL-N5	0,20	0.15	0.10	0.00	0.09	0.00	0.01	0.07	0.21
CT-OLEEINS	0.82	0.94	1.31	1.38	1.61	1.55	1.44	1.64	1.59
METHYLCYCLOHEXANE	0.21	0.19	0.15	0.16	0.16	0.16	0.17	0.22	0.22
CB-OLEFINS + ISO-P	0.00	0.00	0.51	0.00	0.04	0.00	0.12	0.85	0.69
NTHER ISA-C8-P	1.01	1.09	1.33	1.29	0.79	1.14	1.11	0.52	0.83
C8-OLEFINS	1.00	1.07	2.89	3.56	4,18	4.01	3.73	2.63	3.02
CB-NAPHTHENES (NS+N6)	1.32	1.19	0.85	0.84	1.27	0.66	0.23	1.24	0.95
C9-OLEFINS + ISO-P	0,00	0.52	0.93	1.08	1.17	1.02	1.00	0.92	0,17
MONOMETHYL-ISD-C9-P	0,48	0.58	0.81	0.81	0.81	0.19	0.66	0.51	0.46
OTHER ISO-C9-P	0.17	0.17	0.16	0.08	0.15	0.12	0.20	0.14	0.16
C7-JLEFINS C9-NAPHTHENES (N5+NK)	0.33	0.73	2.11	2.09	2.19	1.87	2.03	1.51	1.29
N-NONANE	0.10	0.17	0.07	0.60	0.67	0.55	0.55	0.39	0.36
ISO-C10-P + 0 + N5 + N6	0.83	1.11	2.17	2.69	3.00	2,19	2.47	1.84	1.37
BENZENE	0.44	0.31	0.41	0.26	0.39	0.40	0.40	0.52	0.51
ETHYLBENZENE	1.04	0.32	0.51	0.37	0.38	0.30	. 0.59	0.93	0.97
P-XYLENE	1.05	0.85	0.57	0.44	0.42	0.36	0.55	0.56	0.62
	2.62	2.15	1.59	1.30	1.33	1.04	1.38	1.47	1.53
ISOPROPYLBENZENE	0.00	0.00	0.00	0.00	0.00	0.50	0.34	0.66	0.69
N-PROPYLBENZENE	0.17	0.22	0.26	0.26	0.24	0.19	0.25	0.24	0.22
1-METHYL-3-ETHYL-BENZENE	3.00	2.91	2.53	2.02	2.02	1.53	2.18	2.10	2.22
1-METHYL-2-ETHYLBENZENE	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.02
ISO-C4-BENZENE	0.04	0.05	0.00	0.06	0.06	0.04	0.04	0.05	0.05
1.2.4-TRIMETHYLBENZENE	0.00	0.00	0.00	0.00	0.00	0.00	0.04	0.00	0.00
1-METHYL-2-ISO-C3-BENZENE	0.06	0.04	0.05	0.05	0.05	0.02	0.15	0.19	0.17
1, 3-DIETHYLBENZENE	0.00	0.00	0.00	0.59	0.70	0.50	0.00	0.00	0.00
1-MLINYL-3-N-CJ-BENZENE N-CA-BENZENE	0.37	0.47	0.76	0.24	0.25	0.19	0.66	0.59	0.57
1, 2, 3-TRIMETHYLBENZENE	0.00	0.05	0.04	0.03	0.03	0.01	0.02	0.18	0.17
1, 2-DIETHYLBENZENE	0.10	0.14	0.23	0.00	0.00	0.00	0.00	0.00	0.00
1-METHYL-2-N-C3-BENZENE C10-ALKYLBENZENES	0.12	0.04	0.04	0.03	0.03	0.02	0.03	0.07	0.06
1, 2, 4, 5-TETRAMETHYLBENZENE	0.10	0.09	0.12	0.07	0.07	0.05	0.10	0.10	0.13
1, 2, 3, 5-TETRAMETHYLBENZENE	0.00	0.00	0.00	0.02	0.02	0.00	0.05	0.00	0.00
1, 2, 3, 4-TETRAMETHYLBENZENE C11-ALKYLBENZENES	0.13	0.10	0.08	0.01	0.00	0.06	0.15	0.06	0.07
NAPHTHALENE	0.00	0.00	0.03	0.00	0.02	0.02	0.00	0.01	0.00
METHYL-NAPHTHALENES	0.00	0.00	0.00	0.00	0.00	0.00	0.02	0.00	0.00
UNKNOWN C12+	0.06	0.01	0,00 3, <b>47</b>	0.00	2,50	0.00	0.04	0.00	0.00
SLURRY REACTOR WAX	9.08	ā. 00	9.91	8.63	8.62	3.54	3.47	8.32	8.44

# Tably A-9 (Contd.) Composition of Hydrocarbon Products from Two-Stage Slummy F-T/ZSM-5 Syngas Conversion Run CT-256-3

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M.B.No.	3-29	3-30	3-31	3-32	3-33	3-34	3-35	3-36	3-37
Days Un Stream	30.5	51.3	32.3	€ • 5ک	34.5	33.3	36.5	37.5	33.0
METHANE	7.73	7.93	7.91	7.69	8.08	7.09	7.43	8.44	7.54
ETHENE	0.94	0.93	Ú.97	1.01	1.10	1.06	1.15	1.39	1.24
ETHANE	3.05	3.30	3.34	3.31	3.45	3.25	3.21	3.62	3.21
PROPENE	5.64	7.62	9.39	3.33	9.46	9.03	9.08	10.23	9.08
I-BUTANE	8.36	8.94	9,32	9.45	9.53	9.26	9.02	9.63	9.97
1-BUTENE+2-METHYLPROPENE	2.96	2.71	2.54	2.42	2.54	2.62	2.71	3.03	3.04
N-BUTANE	6.59	6.94	7.35	7.58	7.59	7.01	7.46	7.73	7.33
TRANS-2-BUTENE	1.18	0.73	1.00	0.95	1.00	0.72	1.07	1.17	0.83
3-HETHYL-1-BUTENE	0.05	0.08	0.07	0.07	0.07	0.07	0.08	0.08	0.03
I-PENTANE	6.76	6.16	6,29	6.28	5.72	ó. 12	ó.02	5.28	5.61
	0.13	0.10	0.10	0.09	0.09.	0.10	0.11	0.10	0.11
N-PENTANE	4.40	3.95	4.03	3.99	3.77	4.02	4.01	3.30	3.74
TRANS-2-PENTENE	0.49	0.39	0.35	0.32	0.33	0.36	0.38	0.33	0.40
CIS-2-PENTENE	0.24	0.18	0.17	0.16	0.16	0.18	0.19	0.16	0,17
CS-BIOLEFINS (DIENES)	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
2, 2-DIMETHYLBUTANE	0.06	0.04	0.04	0.04	0.03	0.04	0.04	0.03	0.04
CYCLOPENTANE	0.20	0.19	0.22	0.24	0.25	0.29	0.29	0.15	0.17
HEXENES + ISO-HEXANES	0.07	0.02	0.02	0.02	0.00	0.05	0.02	0.00	0.05
2-METHYLPENTANE	2.69	2.23	2.14	2.09	1.91	2.04	2.06	1.40	1.91
3-METHYLPENTANE	1.14	0.96	0.94	0.96	0.90	0.74	0.96	0.66	0.34
HEXENES	0.45	0.48	0.46	0.42	0.67	0.43	0.44	0.52	0.58
	0.11	0.06	0.05	0.05	0.02	0.05	0.07	0.03	0.06
2.4-DIMETHYLPENTANE	2.31	0.01	0.01	0.01	0.01	0.01	0.00	0.00	0.00
METHYLCYCLOPENTANE	1.02	0.95	1.00	1.12	1.08	1.20	1,23	0.88	1,11
3. 3-DIMETHYLPENTANE	0.00	0.00	0.00	0.00	0.01	0.03	0.00	0.00	0.00
HEPTENES + ISO-HEPTANES	0.02	0.02	0.03	0.03	0.04	0.02	0.02	0.03	0.00
2-METHYLHEXANE	0.77	0.75	0.70	0.69	0.61	0.67	0.62	0.43	0.57
2. 3-DIMETHYLPENTANE	0.17	0.18	0.19	0.19	0.20	0.20	0.19	0.15	0, 13
	0.73	0.72	0.69	0.57	0.62	0.67	0.61	0.44	0.57
1-TRANS-3-DIMETHYL-NS	0.32	0.33	0.33	0.35	0.34	0.38	0.25	0.28	0.34
1-TRANS-2-DIMETHYL-NS	0.25	0.26	0.27	0.28	0.27	0.30	0.28	0.22	0.27
N-HEPTANE	1.08	1.10	1.01	0.98	0.37	1.09	0.97	0.76	1.02
METHYLCYCLOHEXANE	0.24	0.25	0.23	0.24	0.19	0.25	0.19	0.16	0.03
C8-OLEFINS + ISO-P	0.69	0.52	0.41	0.40	0.01	0.50	0.40	0.01	0,42
MONOMETHYL-ISO-C8-P	0.57	0.61	0.56	0.51	0.57	0.44	0.43	0,46	0.43
C8-OLEFINS	1.11	1.10	1.30	1.17	1.27	0.83	0.58	0.31	0.94
C8-NAPHTHENES (N5+N6)	1.03	1.10	1.11	1.05	1.26	0.99	0.97	1.10	0.87
N-OCTANE	0.52	0.51	0.06	0.05	0.00	0.42	0.42	0.42	0.49
MONOMETHYL-ISD-C9-P	0.29	0.32	0.29	0.25	0.27	0.23	0.22	0.26	0.29
OTHER ISO-C9-P	0.13	0.14	0.13	0.12	0.13	0.11	0,02	0.09	0.14
C9-OLEFINS	0.68	0.67	0.54	0.41	0.45	0.40	0.54	0.26	0.33
N-NONANE	0.17	0.17	0.32	0.14	0.15	0.14	0.47	0.37	0.37
ISO-C10-P + 0 + N5 + N6	0.58	0.85	0.71	0.70	0.72	0.52	0.53	0.61	0,55
BENZENE	0.64	0.64	0.48	0.76	0.75	0.91	ə.32	0.61	0.74
TOLUENE	2.49	2.97	3.10	3.33	3.24	3.02	3.52	3.37	3.44
P-XYLENE	0.62	0.73	0.71	0 <u>.88</u>	0.99	2.90	2.91	0.35	0.79
M-XYLENE	1.61	1.31	2.06	2.11	2.43	0.00	0.00	2.43	2.12
	0.70	0.79	0.88	0.93	1.07	0.92	0.90	1.01	0.71
1-METHYL-3-ETHYL-BENZENE	1.98	2.14	2.13	2.14	2.31	1.78	2.00	2.23	2.05
1, 3, 5-TRIMETHYL-BENZENE	0.06	0.06	0.06	0.06	0.06	0.06	0.06	0.07	0.06
1-METHYL-2-ETHYLBENZENE	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.05
1. 2. A-TRISETAVI RENZENE	1-21	1.32	1-37	1.46	1.60	1.40	1 43	0.03	0.03
1-METHYL-2-ISO-C3-BENZENE	0.11	0.11	0.10	0.04	0,04	0.12	0.03	0.04	0.03
1, 3-DIETHYLBENZENE	0.00	0.00	0.00	0.00	0.40	0.00	0.36	0.39	0.37
N-C4-BENZENE	0.09	0.43	0.00	0.09	0.10	0.34	0.07	0.08	0.08
1, 2, 3-TRIMETHYLBENZENE	0.04	0.03	0.03	0.00	0.06	0.00	0.08	0.09	0.05
1, 2-DIETHYLBENZENE	0.00	0.00	0.08	0.08	0.00	0.07	0.00	0.00	0.00
C10-ALKYLBENZENES	0.61	0.4	0.61	0.68	0.03	0.15	0.57	0.00	0,54
1, 2, 4, 5-TETRAMETHYLBENZENE	0.06	2.07	0.07	0.07	0.08	0.07	0.07	0.08	0.07
1. 2, 3, 5-TETRAMETHYLBENZENE	0.00	0.00	0.00	0.05	ù.05	0.00	0.04	0.05	0.04
TI AN SI ATTETRATETATETATENCENCENE	0.09	0.09 0.74	0.09	0.01	0.11	0.10	0.10	0.11	0.10
NAPHTHALENE	0.05	0.07	0.00	Ú.07	0.07	0.07	0.75	0.09	0.00
METHYL-NAPHTHALENES	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00
UNKNOWNS (HC AROMATICS)	0.00	0.00	0.01	0.02	0.00	0.00	0.06	0.01	0.07
SLURRY REACTOR WAX	S. 45	9.34	3.30	3.93	7.97	7.60	7.63	7,47	7,31

# Table A-9 (Contd.) Composition of Hydrocarbon Products from Two-Stage Slurry F-7/25M-5 Syngas Conversion Kun CT-256-3

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M. 9. No.	3-38	3-39	3-51	3-52	3-53	3-54	3-35	3-54	3-57
Days Un Stream	39.5	40.5	53.5	55.5	57.5	59.S	62.5	04-5	46.5
					•••••		0210	0.10	0010
METHANE	7.30	7.97	7.97	9.51	6 71	0 4 9	10 99	ê 30	6 24
ETHENE	1.29	1.43	0.96	0.51	0 54	0.73	1 65	0.04	1 04
ETHANE	3.15	3.35	3.27	3-81	3.67	4.04	4.74	4 60	2.00
PROPENE	4.46	5.10	0.87	2.08	2.28	3.07	2.76	3 70	3.33
PROPANE	9.02	9.40	8.54	6.81	5.57	5.91	A. 29	5.27	7 70
I-SUTANE	8.73	8.99	10.64	8.51	6.86	7.22	7.54	7.91	2 05
1-BUTENE+2-METHYLPROPENE	3.15	3.52	0.85	1.99	2.73	3.21	3.55	2 80	3 79
N-BUTANE	7.41	7.59	8.35	7.19	A- 24	6 20	5 04	6 22	4 90
TRANS-2-BUTENE	1.24	1.37	0.31	0.74	1.03	1 20	1 30	4 4 4	1 03
CIS-2-BUTENE	0.88	0.96	0.21	0 50	0.00	0.21	1.30	A*44	1.00
3-METHYL-1-PUTENE	0.09	0.10	0.02	0.05	0.00	0.00	0.00	0.00	0.70
I-PENTANC	<b>K</b> 20	<b>F</b> /0	0.03	0.03	0.07	0.09	0.10	0.08	0.08
-PENTENE	0.12	3.02	6.73	0.14	2.90	5.18	5.90	5.96	6.21
2-METHYL-1-BUTENE	0.13	0,14	0.33	0.08	0.11	0.11	0.13	0.12	0.11
N-PENTANE	4 07	2 27	4 74	0.40	0.08	0.00	0.74	0.63	0.55
TRANS-2-PENTENE	0 44	0.44	2.10			5-87	3.88	3.85	3.98
CIS-2-PENTENE	0.23	0.33	0.04	0.17	0.45	0.72	0.50	0.27	0.39
2-METRAL -2-BUTENE	1 50	1 40	0.00	0.13	1.00	0.20	0.24	0.25	0.18
LINKNOUN CS-RONOOLETING	1.00	A. 47	0.41	1.04	1.85	1.03	1.75	1.60	1.39
2. 2-RIMETUVI BUTANE	0.05	0.00	0.00	0.00	0.00	0.00	0.00	0.14	0.00
CYCL DEENTANE	0.03	0.05	0.04	0.02	0.03	0.03	0.05	0.04	0.04
WEYELES A TOD-WEYANES	0.22	0.20	0.04	0.02	0.05	0.06	0.12	0.12	0.14
	0.00	0.05	0.11	0.03	0.0/	0.04	0.09	0.05	0.02
	0.13	0.12	0.16	0,10	0.09	0.06	0.12	0.12	0.13
	1.98	1.75	2.98	2.58	3.10	1.74	2.65	2.45	2.29
JTHE HTLPEN I ANE	0.89	0.81	1.19	0.93	1.04	0.75	1.05	1.04	D.94
TELENES	0.58	0.04	0,38	0.59	C.79	0.81	0.58	0.82	Q.40
	0.07	0.07	0.65	0.03	0.09	0.05	0.13	0.07	0.06
	2.00	1.87	2.20	2.25	2.99	1.70	2.33	2.23	2.01
2. 4-ULIEINTLYENTANE	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.01	0.01
TETHTLCYCLOPENTANE	1.15	1.10	0.69	0.48	0.46	0.58	0.75	0.97	0.33
3. 3-UIMETHYLPENTANE	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0,00
LYCLOMEIANE	0.00	0.02	0.02	0.03	0.02	0.03	0.02	0.03	0.03
MEPTENES + ISO-HEPTANES	0.42	C.39	0.19	0.14	0.53	0.23	0.80	0.39	0.35
2-METHYLHEXANE	0.55	0.49	0,97	1.12	1.34	0.74	0.99	0.90	0.79
2. 3-DIMETHYLPENTANE	0.19	C,17	0.15	0.12	0.11	<b>0.15</b>	0.17	0.21	0.18
3-RETHYLHEXANE	0.55	0.30	0.91	0.97	1.09	0.70	0.86	0.85	0.73
1-CIS-3-DIMETHYL-NS	0.34	0.32	0.26	0.23	0.21	0.27	0.29	0.36	0.36
1-TRANS-3-DIMETHYL-NS	0.34	0.32	0.75	0.24	0.17	0.26	0.30	0.35	0.32
1-TRANS-2-DINETHYL-NS	0.28	9.26	0.51	0.23	0.06	0.19	0.25	0.28	0.23
N-HEPTANE ·	1.06	1-01	0.66	1.23	1.82	1.03	1.32	1.26	0.99
C7-OLEFINS	0.64	0.67	0.48	0.81	1.37	1.14	0.85	0.95	0.73
METHYLCYCLOHEXANE	0.23	0.23	0.25	0.20	0.22	0.32	0.34	0.33	0.36
C9-GLEFINS - ISO-P	0.52	0.43	0.55	0.01	0.25	0.01	0.87	0.70	0.56
MONOMETHYL-ISO-C8-F	0.25	0.38	0.89	1.41	0.76	0.94	0.69	0.72	3.63
OTHER ISO-C9-P	0.19	0.12	0.13	0.14	0.08	0.16	0.11	0.15	0.13
C8-OLEFINS	1.34	0.97	0.55	1.88	2.96	2.24	1.95	1.36	1.13
C8-NAPHTHENES (N5+N6)	0.61	0.91	1.10	1.25	1.48	1.24	0.94	1.12	0.99
N-OCTANE	0.51	0.49	0.45	0.75	1.02	0.60	0.54	11 55	0 44
CP-OLEFINS + ISD-P	0.33	0.32	0.20	0.00	0.55	0.00	1.94	6.05	0.00
MONOMETHYL-ISO-C9-P	0.22	0.26	0.42	6.79	0.77	0.00	1.30	0.35	0.74
OTHER ISO-C9-P	0.03	0.13	0.12	0.00	0.14	0.01	0.37	0.13	0.33
C9-OLEFINS	0.47	0.20	0.21	0.00	1 27			0.20	
CS-NAPHTHENES (NS+NA)	0.20	0.00	0.20	0.54	0.34	1. <u>2</u>	0.3/	0.90	0.67
X-NONANE	0.22	0.34	0.37	0.30	0.30	0.51	0.42	0.41	0.32
ISD-C10-P + 0 + N5 + N6	0.23	0.49	0.08	0.23	0.45	0.20	0.12	0.13	0,15
BENZENE	0.75	0.71	0.74	1.37	2.20	1.03	0.85	0.92	0.73
TOLUENE	3 44	3 33	0.40 7.18	0.30	0.43	0.48	0.01	0.69	0.69
ETHYLBENZENE	1,20	1 17	4-10	4.07	1.70	20 eta	4.27	3.06	3.43
P-XYLENE	0.74	0 70	A 01	0.74	0.12	0.39	1.17	1.55	1.63
M-XYLENE	2.70	1.73	0.81	0.76	0.54	1-00	0.57	0.79	0.79
0-XYLENE	0.07	0.00	6. ZV	2.07	1.51	2.34	1.52	1.91	2.27
N-PROPYI SENTENE	0.3/	0.82	0.9/	0.78	0.35	0.98	0.60	9,80	0.93
	1.07	0.12	0.16	0.30	0.29	0.29	0.19	0.17	0.19
L 3. S-TOTHSTUVI -DENTENE	1.70	1.83	2./1	3.13	2.37	3.26	2.14	2.41	2.57
	0.05	0.05	0.10	0.09	0.10	0.11	0.06	0.06	0.07
ISO-CA-BENTENE	0.04	0.00	0.00	0.03	0.03	0.03	0.10	0.10	0.13
1. 2. A-TRINETHY RENTING	0.03	0.02	0.04	0.00	0.00	0.00	0.01	0.02	0.00
1-METHVI _3-ICA_CO_PENTENT	1140	1.00	1.52	1.53	1.17	1.93	1.23	1.41	1.29
1. GENTETUVI BENTENC	0.03	0.03	0.07	0.06	0.05	0.09	0.08	0.03	0.03
INMETHYI SANNORADENE	0.35	0.00	0.00	0.00	0.00	0,00	0.00	0.00	0.43
N-CA-SFNTCHE	0.07	0.00	0.54	0.84	0.76	0.80	0.53	0.51	0.09
1 9 9-701WETLYI DENJENS	0.09	0.08	0.13	0.22	0.19	0.23	0.14	0.13	0.12
TTATETATETATEBERG	0.08	0.07	0.12	<b>V.11</b>	0.04	0.17	0.04	0.05	0.06
	0.00	0.07	0.09	0.19	0.24	0.18	6.12	0.11	0.00
CIOLAL VYLDENTENEC	0.05	v.00	0.00	0.00	0.03	0.00	0.00	0.00	0.00
1. 2. 4. SUTETRANETUVI DENTENE	0.51	0.53	0.83	1.04	0.39	1.23	0.77	0.77	0.19
1 7 3 STETRAREINIUSCNLENE	0.0/	0.07	0.11	0.11	0.08	0.17	0.09	0.09	0.02
1. 2. 3. ALTETRANETUV DENJENIE	0.04	0.04	0.10	0.06	0.03	0.08	0.03	0.05	0.01
CITAN AND DENTENCE	0.09	0.08	0.11	0.0	0.08	0.12	0.08	0.10	0.02
NAPHTWAI ENF	0.71	0.63	1.04	1.08	0.89	1.29	0.90	0.97	0.20
UNKNOWNS (HC AROMATICS)	0.00	0.00	0.07	0.00	0.00	0.00	0.00	0.00	0.00
ENCHOWN (124	0.05	0.05	0.00	0.00	0.00	0.00	0.00	0.00	0.00
STUDDY DEACTOD HAY	7	V. D/	3.33	2.94	2.17	2.41	1.73	1.91	2.00
	1.0	0.40	2,30	3.17	e.83	4.52	2.41	3.24	4.13

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### Table A-9 (Contd.) Composition of Hydrocarbon Preducts from Two-Stage Slurry F-T/ZSM-5 Syngas Conversion Run CT-256-3

M.B.No.	3-50	3-59	3-60	3-01	32	3-53	5-54
Dars on Straam	00.0	70.5	12.5	/4.5	/9.3	/9.5	30.5
METHANE	7.46	8.29	9.14	8.81	9.51	7.70	7.72
ETHENE	0.92	0.91 2 37	1.16	1.27	1.26	1.29	2.77
PROPENE	2.99	3.24	3.92	4,15	4.01	4.17	2.46
PROPANE	9.71	9,41	9.11	8.90	10.22	9.60	7.44
	9.01	8.01	8.56	3.11	8.11	7.34	4.94
N-RUTANE	2.51	2.51	2.82	3.22	2.79	2.76	6.77
TRANS-2-BUTENE	0.98	0.99	1.10	1.27	1.12	1.05	2.36
CIS-2-BUTENE	0.68	0.70	0.77	0.90	0.30	Ú.76	2.10
3-RETHYL-I-BUTENE I-BENTANE	0.07	0.08	0.08	0.10	0.09	0.07	0.21
L-PENTENE	0.10	0.10	0.11	0.13	0.13	0.11	0.23
2-METHYL-1-BUTENE	0.50	0.53	0.53	0.61	0.57	0.40	1,23
N-PENTANE TDANG_9_DENTENE	3.89	4,01	3.46	3.38	2.38	3.01	2.14
CIS-2-PENTENE	0.17	0.19	0.13	0.22	0.21	0.33	0.37
2-METHYL-2-BUTENE	1.25	1.26	1.23	1.35	1.24	0.95	2.55
2, 2-DIMETHYLBUTANE	0.04	0.03	0.03	0.03	0.03	0.02	0.05
HEYENES + ISO-HEYANES	0.19	0,13	0.10	0.12	0.13	0.11	0.15
2. 3-DIMETHYLBUTANE	0.10	0.13	0.12	0.11	0.11	0.03	0.06
2-METHYLPENTANE	2.17	2,19	1.70	1.62	1.52	1.13	0.73
3-METHYLPENTANE	0.96	0.42	0.31	0.77	0.77	0.60	0.47
MEXENES	0.54	2.75	0.87	0.94	0.95	0.67	1.61
N-HEXANE	1.28	1.93	1.61	1.02	1.45	0.03	0.09
2, 4-DIMETHYLPENTANE	0.01	0.01	0.01	0.01	0.01	0.00	0.01
METHYLCYCLOPENTANE	1.00	1.03	0,91	0.96	0.76	0.70	0.43
3, 3-DIMETHYLPENTANE	0.00	0.00	0.00	0.00	0.00	0.00	0.00
HEPTENES + ISO-HEPTANES	0.03	0.03	0.03	0.00	0.03	0.02	0.02
2-METHYLHEXANE	0.75	0.69	0.56	0.55	0.48	0.30	0.28
2.3-DIMETHYLPENTANE	0.19	0,19	0.17	0.16	0.17	0.12	0.11
3-METHYLHEXANE	0.73	0.63	0.58	0.57	0.50	0.34	0.31
1-CIS-3-DIMETHYL-N5 1-TRANS-3-DIMETHYL-N5	0.35	0.33	0.30	0.28	0.28	0.20	0.20
I-TRANS-2-DIMETHYL-N5	0.35	0.26	0.25	0.23	0.29	0.17	0.19
N-HEPTANE	1.04	0.91	0.83	0.39	0.99	0.51	0.91
C7-OLEFINS	0.65	0.65	0.30	0,95	0.77	0.44	1.74
C8-OLEFINS + ISO-P	0.2/	0.21	0.04	0.19	0.17	0.09	0.10
MONOHETHYL-ISO-C8-P	0.64	0.57	0.57	0.57	0.42	0.00	0.34
OTHER ISO-CO-P	0.14	0.14	0.15	0.15	0.13	0.10	0.11
CB-OLEFINS	1.05	1.00	1.26	1.23	0.38	0.40	2.61
N-OCTANE	1.11	1.06	1.11	1.12	0.78	0.71	0.77
C9-OLEFINS + ISO-P	0.36	0.56	0.30	0.55	0.77	0.06	0.03
MONOMETHYL-ISO-C9-P	0.30	0.28	0.29	0.29	0.21	0.13	0.21
OTHER ISO-C9-P	0.13	0.12	0.04	0.12	0.10	0.02	0.11
CY-ULEFINS C9-NAPHTHENES (NS+NA) /	0.6/	0.42	0.53	0.70	0.21	0.13	0.97
N-NONANE	0.15	0.14	0.18	0.22	0.16	0.09	0.46
150-C10-P + 0 + N5 + N6	1.13	0.69	0.79	0.75	0.66	0.41	1.08
	0.71	0.72	0.69	0.65	0.78	0.71	0.59
FTHVI RENZENE	3.37	3.44	3.30	3.10	3.80	3.38	1.30
P-XYLENE	0.87	0.97	1.02	0.99	1.45	0.72	0.39
M-XYLENE	2.37	2.30	2.57	2,42	2.51	2.20	1.60
	1.01	0.97	1.10	1.07	1.09	0.93	0.75
NEFRUFTLBENGENE	0.18	0.15	0.18	0.19	0.14	0.08	0.17
1, 3, 5-TRIMETHYL-BENZENE	0.07	0.06	0.07	0.07	2.19	0.06	1.93
1-METHYL-2-ETHYLBENZENE	0.14	0.00	0.02	0.02	0.01	0.01	0.03
ISO-C4-BENZENE	0.02	0.03	0.04	0.04	0.03	0.02	0.10
1, 2, 4-TRIMETHYLSENZENE	1.67	1.63	1.81	1.74	1.64	1.19	1.27
1. 3-DIETHYLBENZENE	0.62	0.00	0.49	0.03	0.04	0.03	0.02
1-METHYL-3-N-C3-BENZENE	0.00	0.46	0.11	0.48	0.08	0.22	0.39
	0.02	0.12	C'13	0.12	0.09	0.05	0.09
1.2-BIETHYLBENZENE	0.14	0.09	0.00	0.11	0.09	0.07	0.06
1-METHYL-2-N-C3-BENZENE	0.00	0.00	0.00	0.00	0.00	0.00	0.03
C10-ALKYLBENZENES	0.74	0.75	0.91	0.30	0.67	0.43	0.62
1, 2, 4, 5-TETRARETHYLBENZENE	0.13	0.10	0.10	0.10	0.09	0.04	0.03
1. 2. 2. 4-TETRANETHVI AFN7ENE	0.02	0.05	0.05	0.05	0.05	0.01	0.04
C11-ALKYLBENZENES	1.14	0.76	0.90	0.79	0.47	0.55	0.48
NAPHTHALENE	Ŭ.02	0.00	0.00	0.00	0.00	0.01	0.01
METHYL-NAPHTHALENES	0.00	9.00	0.00	0.00	0.02	0.02	0.00
DIMETHYL ETHER	0.00	0.00	0.00	0.00	0.00	0.06	0.00
UNKNOWN C12+	0.28	2.01	2.05	2.20	2.44	9,37	1.41
SLURRY REACTOR WAX	3.70	4.04	4.59	4.50		17.23	