FLUOR ENGINEERS AND CONSTRUCTORS, INC. Contract 836504

12.0 WATER MANAGEMENT

12.1 PROCESS DESCRIPTION

12.1.1 Raw Water Treatment

Raw water from the Ohio River is pumped through a clarifier for removal of suspended solids. 10,447 gpm of raw water is taken from the Ohio River and chlorinated during clarification to remove organic material. The clarified water is collected at the clarified water reservoir for use as cooling tower makeup, utility water, process water, firewater, and solids handling systems makeup water. Clarified water is diverted prior to the clarified water reservoir and pumped through a sand filter for use as boiler feedwater make-up.

Consideration was also given to using a Ranney Well System instead of an intake structure. If a Ranney Well is used, a clarifier won't be needed but there will be more treatment required for the boiler feed water due to the higher TDS level of the Ranney Well water. Refer to Section 12.1.3 for a description of the boiler feedwater treatment scheme for a Ranney Well System. A full description, including an overall block flow diagram, unit block flow diagrams and material balances are contained in this section.

A full description, including an overall block flow diagram and ionic balance, and unit block flow diagram and corresponding ionic balances, is contained in Appendix 1 for case 15. The system design given in this Appendix is based on a ground water system with a flow rate of 19,553 gpm from the Ranney Well structure.

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12.1.2 Potable Water

The potable water supply, 31 gpm, is drawn upstream from the raw water clarifier and treated in the following manner prior to distribution within the plant:

- 1. Chlorination
- 2. Clarification
- 3. Filtration (sand)
- 4. Filtration (activated carbon)
- 5. Post Chlorination

If a Ranney Well system is used, a potable water clarifier will not be required.

12.1.3 Boiler Feed Makeup Treatment System (Demineralizer)

The boiler feedwater treatment system encompasses activated carbon filtration and demineralization by ion exchange. The treatment system for raw water from an intake structure proceeds as follows:

- 1. Filtration (activated carbon)
- 2. Cation Exchange
- 3. Anion Exchange
- 4. Mixed bed polishing

The ion exchange demineralizers consist of strong acid cation exchangers, and strong basic anion exchangers which contain a mixture of strong acid cation resin and strong basic anion resin. The water initially flows through activated carbon filters before flowing through the cation exchanger where the cations are exchanged with hydrogen ions. The water is then pumped through the anion unit for the removal of the anions, CO_2 , and silica in exchange for hydroxyl ions. A mixed bed polisher is then provided for removal of remaining cations and anions. The demineralized water is then routed to a storage tank for boiler makeup.

The treatment system for ground water from a Ranney Well system includes a reverse osmosis unit (R.O.) prior to the cation exchanger. The reverse osmosis units reduces the cation and anion concentrations of the water. Polishing would not be required with this system that utilizes the R.O. unit.

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12.1.4 Condensate Mixed Bed Polishing

Condensate will be treated in a mixed bed ion exchange polishing system. The condensate polishing system is a high flow rate, externally regenerated mixed bed design sized to treat 100 percent of the boiler condensate flow on a continuous basis. Each mixed bed polisher contains a mixture of strong acid cation resin and strong basic anion resin. The condensate polishers perform a dual function:

- By retaining suspended material, mainly corrosion products,
- o By exchanging dissolved ions against hydrogen and hydroxyl ions, they function as demineralizers.

The condensate is then sent to the treated condensate storage tank for reuse as boiler makeup.

12.1.5 Raw Water/Condensate Treatment System Wastes

Various wastes are generated in operating the raw water and condensate treatment systems. The individual wastes from these systems are handled with a sludge thickener that is provided to receive the blowdown, backwash, the rinse water from the potable and raw water clarifiers, activated carbon filters, and sand filters. The clean supernatant from the thickener is routed to the supernatant collection sump where it is pumped back to the clarification system. Regeneration and waste from the demineralizer and mixed bed polishers are combined and neutralized in a neutralization sump. This neutralized water is then pumped to a monitoring pond before discharging to the Ohio River.

Sludge resulting from the operation of the potable water treatment system and the raw water pretreatment system is classified non-hazardous and will be sent to landfill following dewatering.

12.1.6 Utility and Process Cooling Tower Makeup

Makeup requirement for the cooling towers is clarified river water and is drawn from the clarified water reservoir. Blowdown from the cooling tower is discharged to the effluent treatment pond or, for minor leakage from the process side, the blowdown will be diverted directly to the oily water sewer. For major leakage, cooling water return will be diverted to the oily water sewer.

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12.1.7 Effluent Treatment

Oily water is initially treated in API separators and dissolved Air Flotation Units (DAF) to remove free oil and suspended material. The DAF effluent is then combined with the other process plant waste streams in the equalization pond. Other process plant waste streams include stripped gas liquor, Mobil MTG reaction water, Lurgi liquid POX water, SNG condensate, and miscellaneous plant process effluents such as equipment leakage, wash water, etc. The pond serves to absorb any flow and chemical composition swings prior to the biological treatment. Off-spec waste is diverted to the off-spec collection pond and recycled back to the equalization pond. The collected water is initially pumped to a primary aeration basin, where Biological Oxygen Demand of the water is _educed. Phosphate is added at the aeration tank, then the aerated waste is pumped into a clarifier where suspended solids are collected. Next, the partially treated waste flows to a secondary aeration basin, where it will be further aerated. After extended aeration/contact time, the water is pumped to a secondary clarifier unit, where polyelectrolyte is added to obtain more efficient suspended solids removal.

Effluent from the second clarifier will be recycled to the off-spec collection pond if it is not acceptable. The clarifier overflow is routed through sand filters for the removal of residual suspended material. The waste water is then treated by activated carbon filters for further removal of residual organics and toxic substances. Then it is pumped to the effluent treatment pond where it is blended with other plant wastes for ph adjustment, monitoring, and eventual discharge to the Ohio River. Part of the biosludge from the primary and secondary clarifier is routed to a sludge thickener for concentration. The balance of the sludge is recycled to the aeration tanks.

12.1.8 Sanitary Wastewater Treatment

The sanitary sewage, comprised primarily of human liquid and solid wastes, is discharged to a biological treatment facility. The solid wastes are ground to pulp prior to biological aeration in which the sewage is decomposed by an aerobic digestion process. A portion of the biological sludge is dewatered and incinerated and the remaining is recycled. The final effluent after sand filtration and post chlorination is discharged to the effluent treatment pond.

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12.1.9 Unit 46 - FGD Unit

The sludge from the Flue Gas Desulfurization system is concentrated in the thickener. The slurry from the bottom of the thickener is pumped to vacuum filters where a filter cake of approximately 60 percent solids is discharged to the Fly Ash Stabilization system of the Boiler Ash Handling Unit (Unit 03).

Blowdown from the scrubbing water surge tank is combined with the blowdown from ash handling and discharged to the front end of the biological treatment unit.

12.1.10 Coal Wet Screening and Wash Water Handling (Unit 20)

Several wet screens are employed to wash down any fines (minus 1/4") from the coarse coal (2" x 1/4") which is fed directly to Lurgi gasifiers.

The minus 1/4" size fines are dewatered in a centrifuge. The coal fines, after being dried, are stored in the fines silo. The fines can either be used for power and steam generation or sold to customers.

Overflow from the centrifuge is collected in clarified water head tank for recirculation. Makeup water is added to the system to compensate for water retained with the coal particulates and blowdown.

12.1.11 Ash Handling Effluent Treatment

12.1.11.1 <u>Gasifier Effluent Treatment</u> - The quenched gasifier ash and its sluiced water are screened to remove large ash particles and the remainder is collected in a holding sump. The coarse ash is directly conveyed to landfill.

The slurry is pumped to dewatering cyclones and then to plate thickeners for further clarification. The thickened sludge is filtered and discharged to landfill. The clarified water from the plate clarifier and dewatering cyclone is recirculated. Make-up water is added for water losses due to evaporation by quenching and for retained water in sludge and blowdown.

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12.1.11.2 <u>Boiler Effluent Treatment</u> - The fly ash from the boilers is discharged dry to stabilize the sludge from the FGD unit before disposing to a landfill.

The bottom ash is pumped to the ash dewatering plant and combined with the gasifier ash slurry.

The effluent water from the clarifier and dewatering press is recycled back to the system. Make-up water is continuously added due to water losses and blowdown.

12.1.12 Utility Water

The utility water requirements for the plant are drawn from the clarified water reservoir. Utility water is primarily required for area washdown water. Utility water so used will be collected by the oily water runoff system and eventually returned to the biological treatment system.

12.1.13 Storm and Oily Water Separation System

The plot is divided into clean and contaminated drainage areas for the purpose of collecting and segregating clean and contaminated storm runoff.

The clean storm sewer (OWS) collects oily storm runoff and washdown from contaminated process areas and tankage. The oily water process sewer (OPS) collects equipment drains and washdown. Oily water from both sewers flows to the oily water separation system where free oil is skimmed off and sludge is collected and removed. The overflow from the oily water separator is routed through an air flotation unit for final oil removal prior to being pumped to biological treatment. Two oily water ponds are provided to impound oily water during a storm event. The ponds are sized to impound the runoff resulting from a 25-year frequency, one day duration, storm event equivalent to 5.7 inches total rainfall collected over the surface of the oily water storm sewer watershed plus 30 minutes flush from the clean storm sewer for the same storm event.

Normally, under dry weather conditions, the OWS flow is to the oily water separator. During a storm event, the initial flush of the CSS is automatically diverted to the oily water ponds. During an extended storm event, the OWS may be manually diverted to the clean storm ponds. Oily water impounded in the oily water ponds during a storm event is pumped to biotreating through the oily water separator and air flotation unit.

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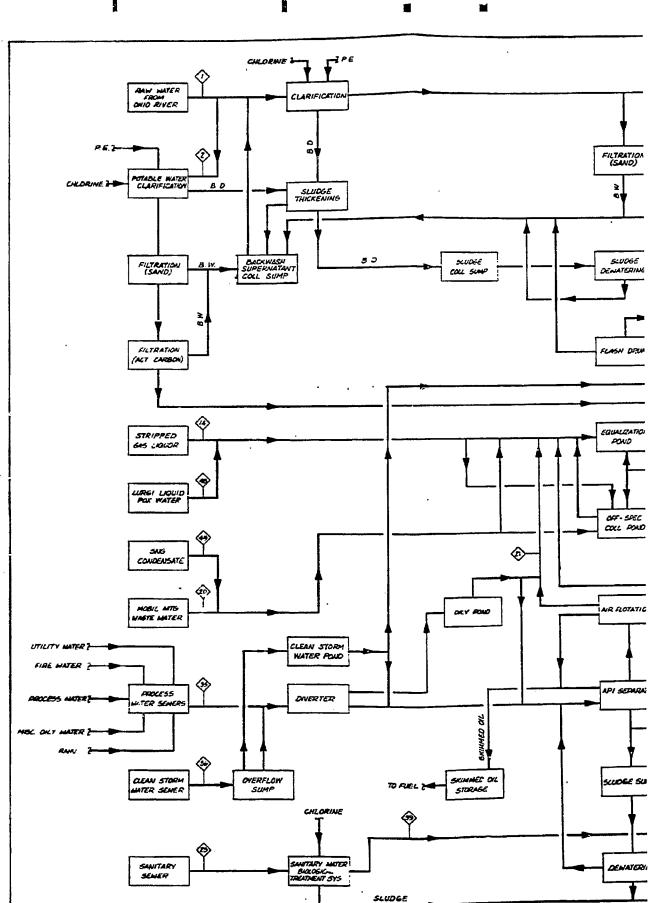
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FLUOR ENGINEERS AND CONSTRUCTORS, INC. Contract 835504

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12.2 FLOWSHEETS

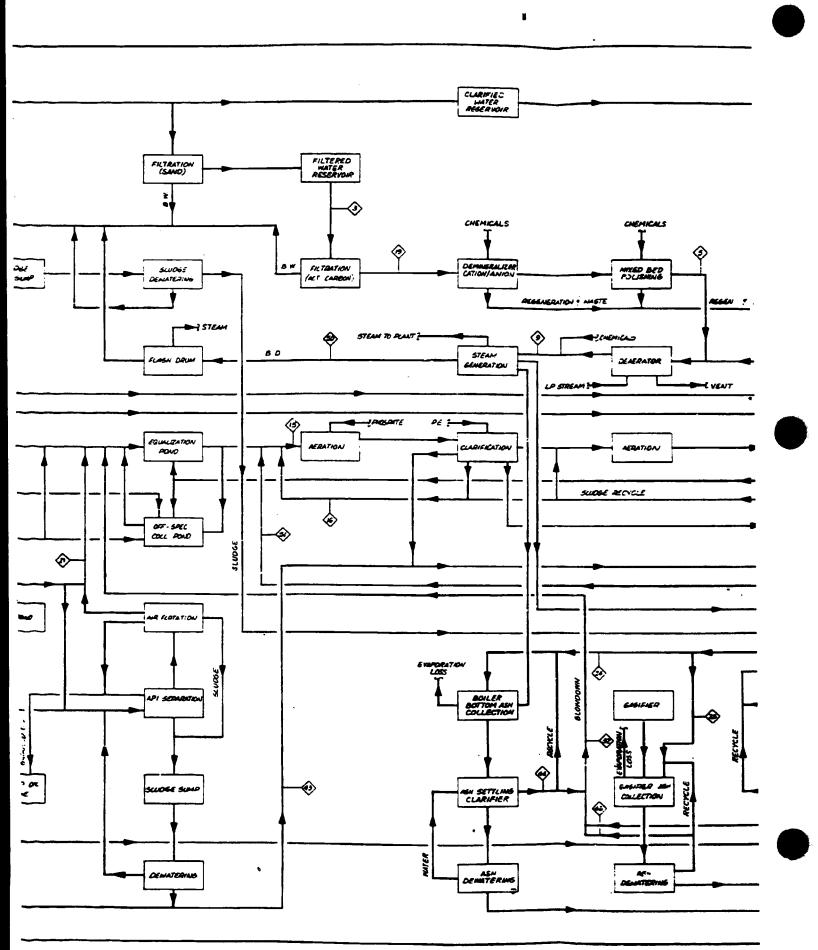
Attached is Drawing No. 835504-00-12-043 Rev. B - Block Flow Diagram - Water and Solid Waste Management.

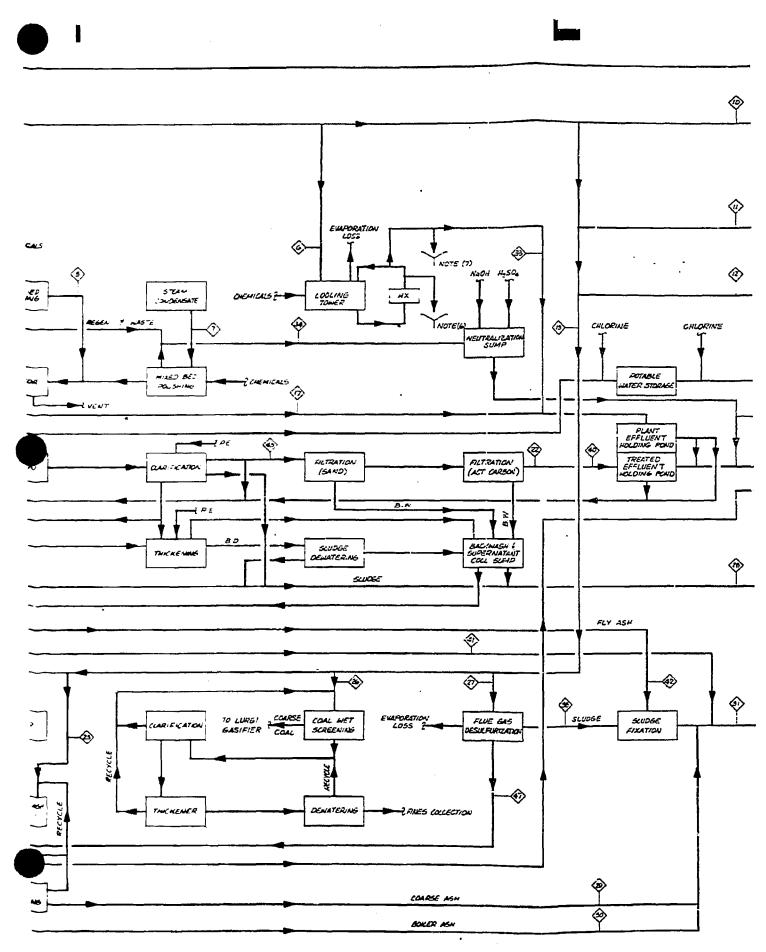


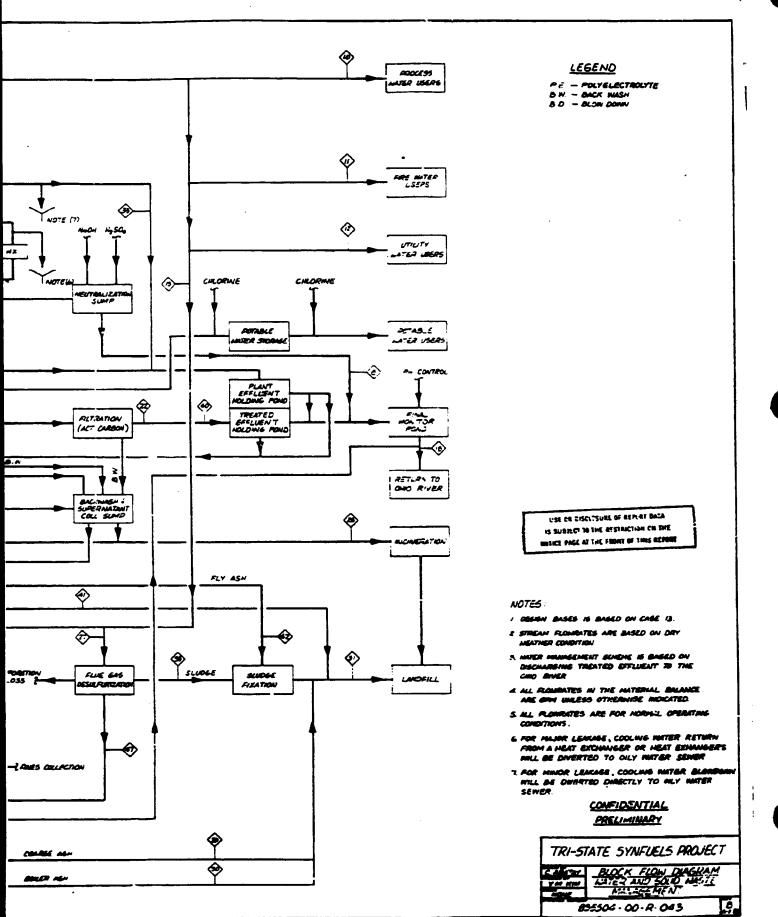
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12.3 MATERIAL BALANCE

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Attached is a Material Balance for Water Management.

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	UNITOZ - COAL WASH HANDLING UNIT	
s	2 - COAL FINES	$\frac{Date 3/24/82}{Cont. No. 83557}$
TR	3 - COARE COAL 4 - RECYCLE WATER	By P. Hunky. CALE 13
E A	5- MAKE-UP WATER	
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	PPM	STREAM NO.					
CONSTITUENT	PPN	2	3	4	5		
Total Dissolved Solids		1 1		607	-143	•	
Total Alkalinity	CaCO 3	1		54	54		
Total Hardness	CaCO3			332	332	•	
PH				la b	6.6		
Total Suspended Solics		<u> </u>			-		
Calcium (Ca++)	CaCOn			216	216		
Magnesium (Mg++)	CaCO3			1/6	116		
Sodium (Na+)	CaCO3			143			
Hydrogen (H+)	CaCO3			-			
Potasium (K+)	CaCO3			60	-	-	
Ammonia (NH4+)	CaCO ₃				-		
TOTAL CATIONS	CaCO3			475	332 .		
			1				
Bicarbonate (HCO ₃)	CaCO3			54	54		
Carbonate (CO-3)	CaCO3						
Hydroxide (OH-)	CaCO3			-	-		
Chloride (Cl-)	CaCO3		!	161	21		
Sulfate (SO 4)	CaCO ₃		<u> :</u>	200	254		
Nitrate (NO3)	CaCOg						
Fluoride (F-)	CaCOz			1.9	1.9		
Phosphate (PO ₄ ⁻)	CaCO3			2.14	0,14		
Sulfite (SO3)	CaCO3			201			
Thiocyanide (CNS ⁻)	CaCO3						
ORGANIC ACIDS	CaC03			1.8	-		
TOTAL ANIONS	Cacoz		ļ	475	3:2		
Iron Total	Fe			0	G		
Carbon Dioxide, Free	C02			25	<u>C</u> 2:-		
Silica	SiO2			2.1	21		
Residual Chlorine	Cl ₂				1.5		
Organic CHEMICALS				0.2			
Manganese	Mn			20.15	<0,15		
Copper	Cu			< 0.01	<0,01		
Hydrogen Sulfide	H ₂ S			(—			
Other Trace Metals				26.28	-		
Phenol	C6H6			600	-		
Sulfur	S			110	-		
TKN	N			120	-		
SLUDGE	LBS. /HR.		1		<u> </u>		
COAL	LBS./HR.	41670	517585		-		
ASH	LBS./HR.	•	1				
COD				<u>+2</u> 30	-		
BOD			,				
Flow Rate	GPM	60	75	1810	135		

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E			CASE	- 13
		STREAM NO.		
RACE METAL PPM	4			
Aluminum	20.05			
Antimony	0.001			
Arsenic	0 0 21-	<u> </u>		
Barium Bervllium	0.22			
Boron	11			
Cadmium	20.002			
Chromium	3.4			
Cobalt				
<u>Copper</u> Lead	0.019			
	20.002			
Lithium				
Molybdenum	20,002			
Mercury	0.0071			
Nickel	0.004			
Phosphorous	20.2			
Selenium	0.14			
Silicon	8.6			
Silver Strontium	<0.002-			
Thallium				
Titanium	<u>< 0.002</u> C. 27%			
Vanadium	0.18			
Yttrium	<0.002			
Zinc	<0.003			
Total Trace Metal	26.28			
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A <u>4 - MAKE-C</u> M <u>5 - BLOWDO</u>	sund					
	PPM		STREAM NO.			
CONSTITUENT	PPM		2	3	4	5
Total Dissolved Solids			2654	,	443	2454
Total Alkalinity	CaCO ₃	+	54		54	2654 54
Total Hardness	CaCO3	1 1	1600	1	332	1600
РН			6,6	1	6.6	6.6
Total Suspended Solics	l	1				
Calcium (Ca++)	CaCO3	1/	880		216	880
Magnesium (Mg++)	CaCO ₃		150 338		116	550
Sodium (Na+)	CaCO3		538			32/
Hydrogen (H+)	CaCO3					110
Potasium (K+)	CaCO ₃		110			110
Ammonia (NP.4+)	CaCO3	<u>i</u>				
TOTAL CATIONS	CaCO3		1978		332	1972
				1	54	54
Bicarbonate (HCO3)			54			
Carbonate (CO-3)	CaCO3 CaCO3	· · · · · · · · · · · · · · · · · · ·				
Hydroxide (OH-) Chloride (Cl-)	CaCO3	<u> </u>	21		21	21
Sulfate (SO 4)	CaCO1		1900		21 254	1900
Nitrate (NO ₃)	CaCO ₂	i	1.0		1 .	1
Fluoride (F-)	CaCO		1.9	1	1.9	1 1.9 0,14
Phosphate (PG4-)	CaCO ₃	1	0.14		0.16.	0.14
Sulfite (SO3)	CaCO3					
Thiocyanide (CNS ⁻)	CaCO3		-	1		
TOTAL ANIONS	CaCO3		1778		332	1978
	<u> </u>		++	:		
Iron Total	Fe		-		-	
Carbon Dioxide, Free	CC2		25		25	25
Silica	Si02		70	1	2.1	70
Residual Chlorine	C12	+			1.5	_
Organic Acids				i i		
Manganese	Mn		20.15	1	20.15	20.15
Copper	Cu		< 0.01	•	20.01	20,01
Hydrogen Sulfide	H ₂ S	1				
Other Trace Metals			5.5	1		5.5
Phenol	C6H6					↓;
Sulfur	S					
TKN	N		↓/ ↓	1		ļ
SLUDGE	LBS. / HR.		↓↓			<u> </u>
COAL	LBS./HR.		↓ ↓			
ASH	LBS./HR.	81910	┼ ──┤			
	↓	/	╁━╾┟╌╍──╂			<u> </u>
BOD		├ ──/──	2574	15	200	115
Flow Rate	GPM	1	6714		200	115

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CONSTITUENT	PPM		STREAM NO	- -	1	·
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Total Dissolved Solids	3	,				11
Total Alkalinity	CaCO ₃					
Total Hardness	CaCO3		1	1		·
PH						,
Total Suspended Solids	5			1		
Calcium (Ca++)	CaCO ₃					
Magnesium (Mg++)	CaCO	1				
Sodium (Na+)	CaCO3	1 1	1	I		
Hydrogen (H+)	CaCO ₃	1				
Potasium (K+)	CaCO ₃					
Ammonia (NH4+)	CaCO					
TOTAL CATIONS	CaCO ₃	1 1			1	
			<u> </u>			·····
Bicarbonate (HCOq)	CaCO3			i		
Carbonate (CO-3)	CaCO ₃					
Hydroxide (OH-)	CaCO ₃			1		
Chloride (Cl-)	CaCO3	1				
Sulfate (SO 4)	CaCOa	1 1				
Nitrate (NO3)	CaCO2	1				
Fluoride (F-)	CaCO2	1				
Phosphate (PO ₄ ⁻)	CaCO ₃	1 1	1			
Sulfite (SOT)	CaCO ₂	1 1				
Thiocyanide (CNS ⁻)	CaCO3	1				
TOTAL ANIONS	CaCO3					
						- <u> </u> i
						i
Iron Total	Fe	1 .				
Carbon Dioxide, Free	C02					
Silica	SiO ₂	1				
Residual Chlorine	Cl ₂	1	· ·			
Organic Acids			1			
Manganese	Mn					
Copper	Cu	1 !	1			
Hydrogen Sulfide	H ₂ S	11				
Other Trace Metals	1					
Phenol	CERE	1 1				
Sulfur	S					
TKN	N		1			
SLUDGE	LBS. /HR.				-	
COAL	LBS./HR.					
ASH	185./HR.	81910	1			
COD	1	1 1				
BOD						
Flow Rate	GPM	70				

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UNIT OJA - GASIFIER ASH HANDLING UNIT RECYCLE WATER -

BLOWDOWN

3/29/22 504 CASE 13

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TRACE METAL PPM	2	5	STREAM N	<u>.</u>		
	+ E			1		
Aluminum	-+	L		├		
Antimony	+			+	<u> </u>	
Arsonic	2:5	2.5				
Barium		0.4		<u> </u>		
Beryllium	0.4	1.42		<u> </u>		
Boron	1,42	0.82		<u> </u>		
Cadmium Chromium	0.08	0.00		<u> </u>		
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Cobalt	·			ļ		
Copper Lead	╶╅─┈┼───					
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Lithium		╂╼╍╌┼───┤	<u> </u>	<u>↓</u>		<u> </u>
Molybdenum				L		L
Mercury						
Nickel						
Phosphorous						
Selenium						
Silicon	0.1	0.1				
Silver						
Strontium					1	
Thallium					I	
Titanium						
Vanadium		í		I	1	
Yttrium						
Zinc	1.0	1.0				
	-			L	ļ	
Tot:1 Trace Metal	5.5	5.5		<u> </u>	ļ	
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STREAM

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UNITO38- B	OILER 6	SH HAN	NDLING (<u>UNN</u>	Date 3	124/82: 835504 Hung
$S = \frac{1 - FLY}{S} \frac{AS}{S}$ $T = \frac{S - FGD}{S} \frac{S}{S}$ $E = \frac{E}{S} - TOTAL$ $A = \frac{7 - BOILER}{S}$ $M = \frac{E}{E} - EVAPOR$	<u>I DISCI</u>	HARGED) 		Cont. No	. 835504
T 5- EGD 5	LUDGE				<u>Ву Р.</u>	HUNG.
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E	2145/212	<u>er 320</u>	DGE		-1	, ·
A 7 - BOILER.	EOTTON	1 ASH	DISCHAR	GED)
M & - EVAPOR	PATION L	OSS (BO	TOM ASHI	QUENCHING	7	•
	PPM		STREAM NO	•		
CONSTITUENT	PPM		5	6	7	8
Total Dissolved Solids		1 1	1	1		
Total Alkalinity	CaCO ₂					
Total Hardness	CaCO3	+		1 1		
PH						
Total Suspended Solids						
Calcium (Ca++)	CaCO ₃					
Magnesium (Mg++)	CaCO ₃					
Sodium (Na+)	CaCO3					
Hydrogen (H+)	CaCO3			<u> </u>		
Potasium (K+)	CaCO ₃	<u></u>		<u> </u>		
Ammonia (NH4+)	CaCO3					
TOTAL CATIONS	CaCO ₃					
		i	1			
	0.00	1		!		
Bicarbonate (HCO ₃)	CaCO3 CaCO3			<u> </u>		
Carbonate (CO-3) Hydroxide (OH-)	CaCO3	+	+			
Chloride (Cl-)	CaCO ₂	+	+		1	
Sulfate (SO 4)	CaCO			<u> </u>		
Nitrate (NO ₇)	CaCO	+				
fluoride (F-)	CaCO					
Phosphate (PG4)	CaCOn	+ +				
Sulfite (SOE)	CaCO ₂	1 1				1
Thiocyanide (CNS ⁻)	CaCO ₃	1				
TOTAL ANIONS	CaCO ₃					
	····	+ +				
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Iron Total	Fe	<u> </u>	+	┝───┤────┤		<u></u>
Carbon Dioxide, Free	C02	+		┟┠┠		
Silica	Si0 ₂	╉╼╌┼╾╌┈		┟┊╾╾╌┥	1	
Residual Chlorine Drganic Acids	C12	+	┼┈┼──╴╴	┟───┤────┤		
langanese	tin	++	+	<u>├</u>		_
Copper	Cu		+	┝━━━┼━━━┤		
lydrogen Suliide		·}		┝}		
Other Trace Metals	H ₂ S		++	┝/	{{	—— <u> </u> i
Phenol	C6H6	╂────┼────		╞━━━┼╼━╍╌┨		
Sulfur	S	+				
TKN	N	<u> </u>	<u>+</u> }			
SLUDGE	LBS. /HR.	<u>+</u>	27728	58690		
JOAL	LBS./HR.					
SH	185./HR.	30962			7740	
OD		/			1	
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UNIT 03B - BOILER ASH HANDLING	UNIT	•
9- MAKE - UP WATER		

- 10 RECYCLE WAICK 11 BLOWDOWN 12 BOTTOM ASH DISPOSED

3/24/82 Date 3/24/82, Cont. No. 835504 By P. HUNG CASE 13 r

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			STREAM NO	•		
CONSTITUENT	PPM	9	10	11	12	
Total Dissolved Solids		443	2654	2654		•
Total Alkalinity	CaCO3	54	54	54		
Total Hardness	CaCO3	332	1530	1530		•
PH		6.6	6.6	6.6		
Total Suspended Solics		<u> </u>	1			
Calcium (Ca++)	CaCO3	216	880	880		
Magnesium (Mg++)	CaCO3	116	650	650		
Sodium (Na+)	CaCO3		338	33.0		
Hydrogen (H+)	CaCO3					
Potasium (K+)	CaC03	<u> </u>	110	110		
Ammonia (NH4+)	CaCO3					
TOTAL CATIONS	CaC03	332	1978.	1978		•
Bicarbonate (HCO ₇)	CaCO ₃	54	54	54		
Carbonate (CO-3)	CaCO ₃					
Hydroxide (OH-)	CaCO3					
Chloride (Cl-)	CaCO ₃	21	21	21		
Sulfate (SO_4^-)	CaCO3	254	21 1900	1900		
	CaCO3	754	1100	1400		
Nitrate (NO ₃) Fluoride (F-)		1.9	1.9	1.9		
Phosphate (PO ₄ ⁻)	CaCO3	0.14	0.14	0.14		
Sulfite (SOZ)	CaCO ₃	0.74				
Thiocyanide (CNS ⁻)	CaCO ₃					
TOTAL ANIONS	CaCO3	220 -	1070	10-0		
		332	1978_	1978		
Iron Total	Fe			_		
Carbon Dioxide, Free	C02	25	25	25		
Silica	SiO ₂	211	70	-70		
Residual Chlorine	Cl ₂	1.5				
Organic Acids	<u> </u>					
Manganese	Mn	20.15	20.15	20.15		
Copper	Cu	<0.01	0.01	0.01	i	
Hydrogen Sulfide	H2S					
Other Trace Metals	173		5.5	5.5		
Phenol	C6H6					
Sulfur	<u> </u>		<i> </i>	<u>├/</u>		
TKN	N		├── / ──	<u> </u>		<u></u>
SLUDGE	LBS. /HR.		├──┤───			
	LBS./HR.		<u>├──</u> /──			
COAL	185./HR.		├ <i> </i>	<u>├ /</u>	7140	
COD	AND THE			<u>├</u>		
BOD	<u> </u>			;		
Flow Rate	GPM	22	250	13		

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	mound				Cont. No.	8355
				<u> </u>	By Cash	1-2
E					case	15
A]	
M						
· · · · · · · · · · · · · · · · · · ·					-	
		~			1	
	PPM		STREAM N	<u>vo.</u>		
TRACE METAL CONSTITUENT		10	11			+
Aluminum						
Antimony						<u> </u>
Arsenic	L	215	2.5			
Barium						<u> </u>
Beryllium		0.4	0.4			+
Boron	<u> </u>	1.42	1.42			<u> </u>
Cadmium		12:08	0.08			<u> </u>
Chromium Cobalt	<u>-</u>		/			
	- -	/				<u> </u>
Copper Lead		- / -				<u> </u>
		+				
Lithium Molybdenum	······································	+	<u>├</u>			
	•					<u> </u>
Mercury					1	1
Nickel				1	1	
Phosphorous					1	
Selenium			1		1	1
Silicon	-	0.1	0.1			
Silver	ч	1	1			
Strontium			1			
Thallium			/			
Titanium			<u> </u>			
Vanadium						
Yttrium		<u> </u>	1	ļ	1	<u> </u>
Zinc		1.0	1.0		1	
		·				
TOTAL TRACE METAL		5.5	5.5	1		
		1				i
						<u> </u>
	······				<u> </u>	
					<u> </u>	
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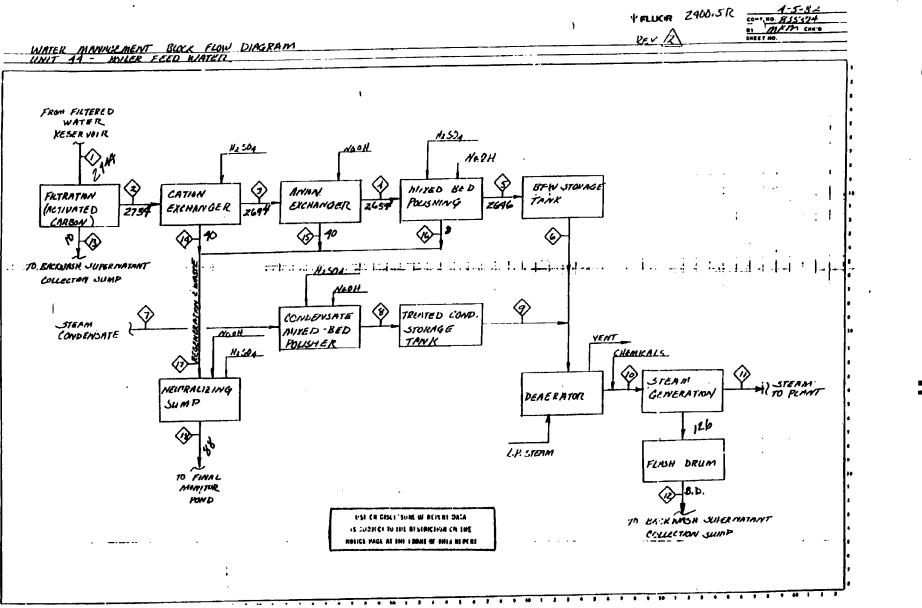
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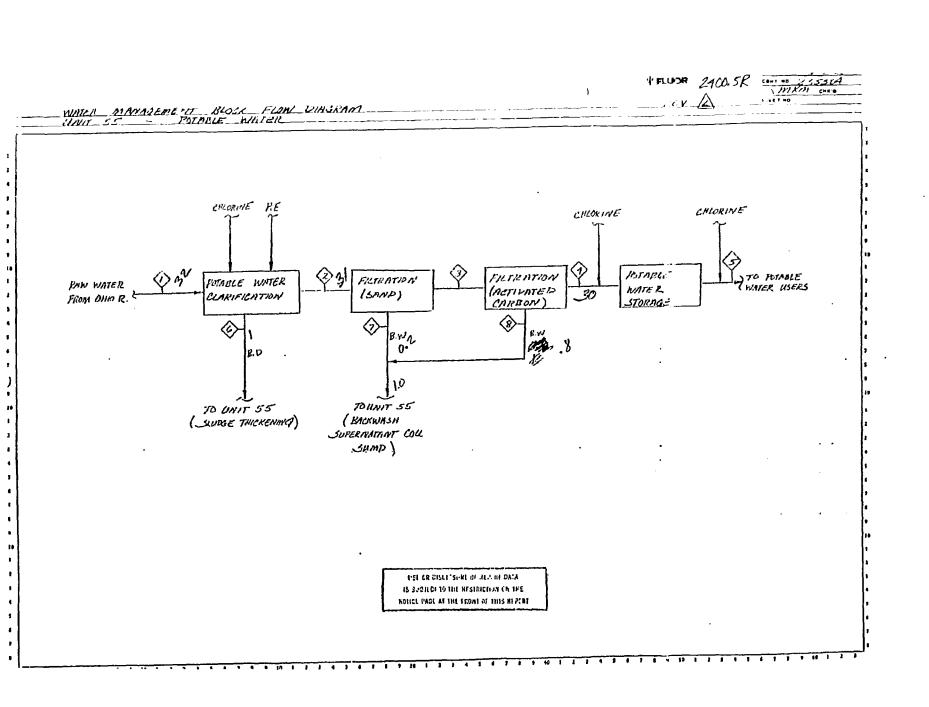
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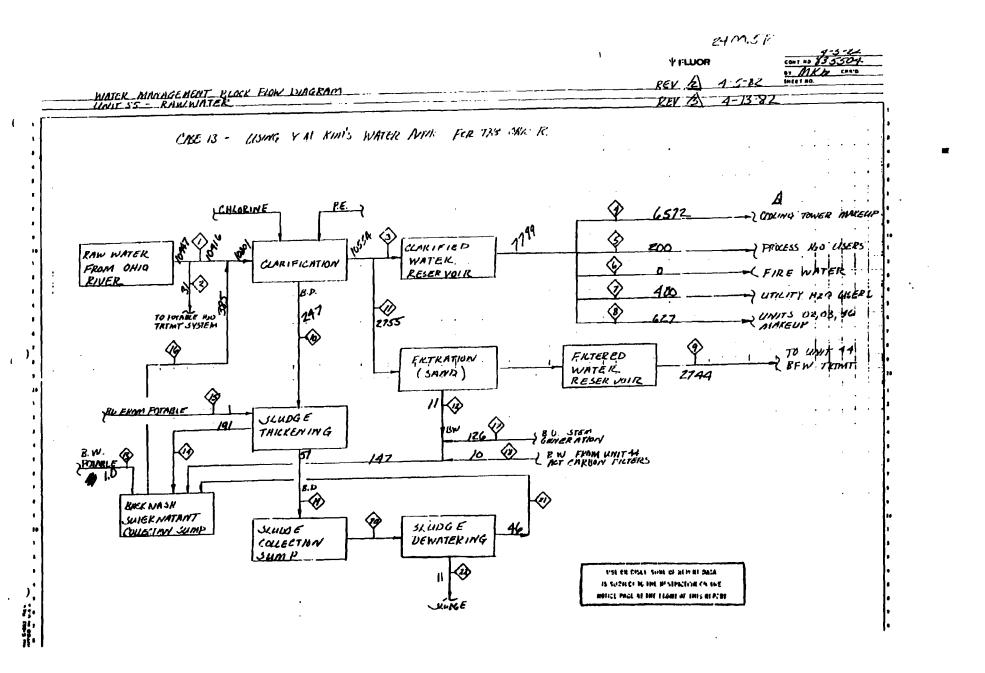
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UNIT 52 - DISS	OLVËD S	OLIDS	AND SLUD	GE DISP	DSAL	3/10/0
S I - GASIFIER	ASH RI	OWDOW	V	40		
T 2- BOILER B	By P.	Cont. No. 835304 By P. Huller				
R Z COD		E 13				
E <u>3-FGD</u> BL						
A 4- TOTAL BL	owdown	/				•
M 5-WET CO	ARSE ASH	1			_{	•
						·
CONSTITUENT	PPM	· · · · · · · · · · · · · · · · · · ·	STREAM NO			·
			2	3	.4	5
Total Dissolved Solids		2654	2654	1765	2579	
Total Alkalinity	CaCO ₃	54	54	216	68	
Total Hardness	CaCO3	1530	1530	1148	1498	
PH Total Suspended Solids		6.6	6.6	8,5	6.8	<u>+</u> ↓
Calcium (Ca++)	CaCOz	880	880	774	871	+
Magnesium (Mg++)	CaCO	650	650	374	627	+
Sodium (Na+)	CaCO3	338	338	180	325	
Hydrogen (H+)	CaCO3					
Potasium (K+)	CaCO3	110	110		100	
Ammonia (NH ₄ +)	CaCO ₃	<u> </u>				
TOTAL CATIONS	CaCO ₃	1978	1978	1328	1923	
Bicarbonate (HCO-)	CaCO ₃	54	54	216	68	
Carbonate (CO-5)	CaCO ₃		+24	-	- 60	
Hydroxide (OH-)	CaCO ₃			-		
Chloride (C1-)	CaCOa	21	21	180	34	
Sulfate (SO_{4}^{-})	CaCO ₃	1900	1900	920	1817	
Nitrate (NO3)	CaCOg			4	1.5	<u></u>
Fluoride (F-)	CaCO ₃	1.9	1.9	7.6	2.5	<u>↓ </u>
Phosphate (PO_4^-) Sulfite (SO_5^-)	CaCO3 CaCO3	0.14	0.14	0.56	12.18	<u> </u>
Thiocyanide (CNS ⁻)	CaCO ₃	<u>+ </u>				<u>├──</u> ┤
TOTAL ANIONS	CaCO ₃	1070	1978	1270	1.0.2.2	<u> </u>
·		1978	1718	1328	1923	<u>├──</u> └
Iron Total	Fe					<u> </u>
Carbon Dioxide, Free	C02	25	25			<u>├──</u> ┤ ───┤
Silica	Si02	70	70	. 10	23	1 - 1
Residual Chlorine	Cl ₂	-			65	
Organic Acids				-		
Manganese	Mn	< 0.15	20.15	20.15	K0-15	
Copper	Cu	< 0.01	20.01	<0.01	20.01	
Hydrogen Sulfide	H ₂ S			·		└── └──
Other Trace Metals Phenol	Call	5.5	5.5		5.04	
Sulfur	C6H6 S	<u> </u>			===	├─∤ ────┤
TKN	N					┝ <u></u>
SLUDGE	LBS. /HR.					
COAL	L3S./HR.		-		_	1
ASH	185./HR.		-			81,910
COD			11			
BOD	6701/					
Flow Rate	GPM	115	13	12	140	70

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UNIT 52-DISSOLVED SOLIDS AND SLUDGE DISPOSAL WET BOILER BOTTOM ASH 6 -

S T R E A M

1 - TOTAL ASH TO LANDFILL 8 - DRY FLY ASH DISCHARGED 9 - FGD SLUDGE 10 - TOTAL STABILIZED SLUDGE

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			STREAM NO.			
CONSTITUENT	PPM	6	7	3	9	10
Total Dissolved Solids				ì	1 I	١.
Total Alkalinity	CaCO ₃					
Total Hardness	CaC03					
PH						
Total Suspended Solics						
Calcium (Ca++)	CaCO ₃					
Magnesium (Mg++)	CaCO ₃					
Sodium (Na+)	CaCO3					
Hydrogen (H+)	CaCO3					
Potasium (K+)	Ca003					
Amponia (NH ₄ +)	CaCO ₂					
TOTAL CATIONS	CaCO3					
					<u> </u>	<u> </u>
					<u>├</u>	┞───┠───
Bicarbonate (HCO3)	CaCO3	1				
Carbonate (CO-3)	CaCO ₃					
Hydroxide (OH-)	CaCO ₃				1	
	CaCO ₃					
Chloride (Cl-)	CaCO ₃				1	
Sulfate (SO 4.)						
Nitrate (NO3)	CaCC ₃					
Fluoride (F-)						
Phosphate (PO4 ⁻)	CaCO3				i	l
Sulfite (305)	CaCO ₃				<u>├</u>	
Thiocyanide (CNS ⁻)					<u> </u>	<u> </u>
TOTAL ANIONS	CaCO3				 	
					┟╌╌╌╋╼╌╌╍	<u>├</u>
Iron Total	Fe					
Carbon Dioxide, Free	CO2					
Silica	Si0 ₂			<u> </u>	<u></u>	<u> </u>
Residual Chlorine	C12					<u>}</u> }_──
Organic Acids			L		i	<u> </u>
Manganese	Mn					↓
Copper	Cu		L		╏╸╺╌┠╼───	<u> </u>
Hydrogen Sulfide	H ₂ S	· ·		·		↓
Other Trace Metals					L	
Phenol	C6H6					
Sulfur	S					ļ
TKN	N					
SLUDGE	L85. / HR.				27728	58690
COAL	LBS./HR.			1		
ASH	185./HR.	7140	89650	30962		
COD			· ·			
BOD			-			
Flow Rate	GPM	17	77		43	43

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Date

Cont. No. 855

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By P. A CASE

UNIT 52 - DISSOLV	E Der SOLL	DS AND	SLUDGE	DISPOSAL-		s lealos.
S 11- SLUDGE	Date 2/30/82: Cont. No. 835504 By P. HUNG CASE 13					
$\frac{S}{R} \frac{11 - SLUDGE}{12 - WASTE}$						
R TI NASIE						
E 13 - 014YSL	UDGE FR	ON OIL	WATER IN	CERIMENT		
A 14 - SLUDGE	ł					
M 15 - TOTAL						•
101110	200190					
CONSTITUENT	PPM	STREAM NO.				
		11	12	13	14	15
Total Dissolved Solids					1	
Total Alkalinity	CaCO	+	+			+
Total Hardness	CaCO3					
PH		1				
Total Suspended Solius		1 1				
Calcium (Ca++)	CaCO ₃					
Magnesium (Mg++)	CaCO ₃		·			
Sodium (Na+)	CaCO3					I
Hydrogen (H+)	CaCO3	L				<u> </u>
Potasium (K+)	CaCO ₃	. <u> </u>		╏╌╴╌┠╺╍╍╂╸		
Ammonia (NH ₄ +)	CaCO ₃	<u> </u>	- /	┞╾╼┾╼╇		<u> </u>
TOTAL CATIONS	CaCO ₃	1 1				ŀ
	C2C0-					
Bicarbonate (HCO ₃) Carbonate (CO-3)	CaCO3 CaCO3		+	<u> </u>		
Hydroxide (OH-)	CaCO3		+			} }
Chloride (C1-)	CaCO2	<u> </u>	+			
Sulfate (SO 4)	CaCO2	1	++			1 1
Nitrate (NOy)	CaCO ₂		+			
Fluoride (F-)	CaCO					1. 1
Phosphate (PO4)	CaCO ₃					
Sulfite (SOE)	CaCO ₃					
Thiocyanide (CNS ⁻)	CaCO3					<u> </u>
TOTAL ANIONS	CaCO ₃					
······	ú					
T						
Iron Total	Fe	<u> </u>	- 	┟───┼╶──┼		┼──┼────
Carbon Dioxide, Free	<u> </u>	├	┼ <u></u> ─┼────	├ <u>├</u> - <u>├</u>		<u>}</u> }
Silica Residual Chlorine	SiO ₂		╋	<u>├</u> ──- <u>├</u> ── <u>┼</u> -		+
Organic Acids	Cl2		+	<u>├/-</u>		<u> </u>
Manganese	Mn	<u> </u>	<u> </u>	<u> </u>		1
Copper	Cu					
Hydrogen Sulfide	H2S					
Other Trace Metals	£					
Phenol	C6H6					
Sulfur	S					
TKN	N					
SLUDGE	LBS. /HR.	0	10	85	800	845
COAL	LBS./HR.	I		/		
	LBS./HR.					
COD			├──/ ──-			¦
BOD		۱.	0.5	2	14.5	·····

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12.4 ACCOMPLISHMENTS AND DECISIONS MADE AND FINALIZED

A preliminary decision was made to base the water management conceptual design on river water, then Ranney Well water, and finally again to river water due to high TDS levels in the Ranney Well water. This decision was based on the Ranney Well Water Analysis contained in Ranney's Preliminary Report, Fluor Contract 835504-0-K003. This water analysis misrepresented the water analysis data contained in the final Ranney Report. The final report was not received and evaluated until after the water management work contained in this report was completed. Comparison of a Ranney System (Section 2.6, Volume 11A) with a river water system showed the Ranney system was superior to a river water system in both cost and water quality. The decision to use a Ranney well will require that the water management design be revised to reflect the Ranney well water analysis. This should be incorporated into project restart.

Case 15 was the initial design case then the basis was changed to Case 13 for water management work. Major work has been completed on the Case 13 Water Management Block Flow Diagram and corresponding Material Balance. The ionic balance needs updating when the information for the final design water analysis is available. The current ionic balance (in Appendix 1) is based on Case 15 and Ranney Well water analysis.

Other decisions made concerning water management include changing the conceptual design basis from zero discharge to discharge to the Ohio River. It was also decided to eliminate the reverse osmossis unit in the boiler feed water system, the TDS concentrator, and two buffer tanks included in the effluent treatment system. In the final design, it was agreed to include tow landfill, hazardous for residue containing heavy metals, and nonhazardous for the remaining residue. A final design configuration for the oily water diverter system was also decided on. A detailed description of this system is included in Section 12.1.2.

FLUOR ENGINEERS AND CONSTRUCTORS, INC. Contract 835504 ì

12.5 WATER MANAGEMENT STATUS

Overall water and waste water treatment schemes for Case 13 have been established and incorporated in "Block Flow Diagram - Water and Solid Waste Management" (Drawing Number 835504-00-R-043 Rev. B, Section 12.2). The material balance for Case 13 is updated and complete, including overall raw water requirements. The current status of Case 13 water management does not include an ionic balance which needs to be updated when a design raw water analysis is available. Ionic and material balances have been prepared for Case 15. This information is included in Appendix 1.

Input information was prepared for a cost comparison estimate of a Ranney Well versus an intake structure. A copy of the rough installed cost and operating cost estimate is included in Volume 11A, Section 2.6.

12.6 LICENSORS AND EVALUATIONS

A detailed study was conducted comparing the use of ground water from a Ranney Well versus raw water from the intake structure. This study includes a cost estimate for treatment of Ohio River Water versus Well water, cost information from Ranney for the intake and well structures, and water analysis for river water and well water. Refer to Volume 11A, Section 2.6 for this information. UNIT 52 - DISSOLVED SOLLOS AND SLUDGE DISPOSAL

Date 3/30/82 Cont. No. 835104 By P, HENG CASE 13

** .	PPM	STREAM NO.				
CONSTITUENT		16		•		
Total Dissolved Solids						
Total Alkalinity	CaCO ₃	1			1	
Total Hardness	CaCO3	+			· · · ·	
PH				1		
Total Suspended Solics	1			1		
Calcium (Ca++)	CaCO3					
Magnesium (Mg++)	Ca003	1	· · · ·			
Sodium (Na+)	CaC03					
Hydrogen (H+)	CaCO ₃					
Potasium (K+)	CaCO3		1.			
Ammonia (NH4+)	CaCO ₃			;		
TOTAL CATIONS	CaCO ₃					
		<u> </u>	_ 		- †	
	<u> </u>					
Bicarbonate (HCO ₇)	CaCO3				1	
Carbonate (CO-3)	CaCO3				T	
Hydroxide (OH-)	CaCO3					
Chloride (Cl-)	CaCO3					
Sulfate $(SO_{\overline{4}})$	CaCO3					
Nitrate (ND7)	CaCO3					
Fluoride (F-)	CaCO2					
Phosphate (PO_4^{-})	CaCO3					
Sulfite (SO3)	CaCO3					
Thiocyanide (CNS ⁻)	CaC03					
TOTAL ANIONS	CaCO3					
	<u> </u>	+		· ·	1	
	<u> </u>	++				
Iron Total	Fe				<u></u>	
Carbon Dioxide, Free	C02	I	<u>_</u>			
Silica	Si02					
Residual Chlorine	C12			:		
Organic Acids						
Manganese	Mn					
Copper	Cu				ļ	
Hydrogen Sulfide	H ₂ S			· · · · · · · · · · · · · · · · · · ·		
Other Trace Metals			<u> </u>		1	
Phenol	C6H6					
Sulfur	S					
TKN:	N	L			<u> </u>	
SLUDGE	LBS. / NR.		<u> </u>		ļ	
COAL	LBS-/HR.				Į	
ASH	LBS./HR.	100			<u> </u>	
COD			<u></u>			
BOD			<u></u>		Į	
Flow Rate	GPM				1	

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