

PROCESS DEVELOPMENT STUDY NO. 17

CO₂ COMPRESSION FOR ENHANCED OIL RECOVERY

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PROCESS DEVELOPMENT STUDY NO. 17

CO₂ COMPRESSION FOR ENHANCED OIL RECOVERY

1.0 INTRODUCTION

Tri-State Synfuels Company proposes to construct and operate an indirect coal liquefaction plant in Western Kentucky based on Sasol-type technology. The original Feasibility Study identified the availability of large quantities of carbon dioxide from the Stretford and Synthol units in the liquefaction plant. The possibility of selling this carbon dioxide for enhanced oil recovery (EOR) is now under investigation.

The purpose of this process study is to develop the data required to perform an economic evaluation of this possibility. The scope of work for this study (Appendix 1) included the following:

- a. A determination of the preferred method of pressurizing the gas and moving it to the plant "battery limits". This entailed a comparative evaluation of three possibilities:
 1. Straight compression to 2200 psia.
 2. Compression, liquefaction with ammonia absorption refrigeration, pumping to 2200 psia and revaporization. This case is referred to as the ammonia absorption refrigeration scheme.
 3. Compression, liquefaction with propylene refrigeration, pumping to 2200 psia and revaporization. This case is referred to as the propylene refrigeration scheme.
- b. A process design for compressing and drying the gas to desired battery limit conditions via the method determined above.
- c. A determination of the pressure, temperature, and dew-point requirements for the CO₂ stream.
- d. An evaluation of the effect on CO₂ compressor plant operations when CO₂ is either not available or demand for it slackens.
- e. An estimate of capital and operating costs for the "battery limits" installation.

This work has been completed and this report presents the pertinent findings.

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2.0 SUMMARY

- 2.1 The capital costs given in this report are the costs associated with pressurizing the CO₂ and delivering it to plant battery limits at the following conditions:

Pressure	2200 psia
Temperature	100 °F
Dew Point at 2200 psia	-20 °F

- 2.2 Straight compression was selected as the preferred method of pressurizing the CO₂ gas and moving it to the plant battery limits. This selection was based on comparative analysis of compression versus liquefaction, pumping and re-vaporization using either propylene refrigeration or ammonia absorption refrigeration to liquify the CO₂.
- 2.3 The recommended design for compressing and drying the gas consists of a 5 stage compressor that increases CO₂ pressure from 15 psia to 2205 psia. Intercoolers and condensate separators are provided between stages and an aftercooler is also included. Final gas drying requirements are met by operating an ethylene glycol dryer on gas flowing between the 4th and 5th compressor stages. Six parallel compressor trains are provided to accommodate design flow rates.
- 2.4 The feed rate may be reduced because CO₂ gas is either not available or demand for it slackens. This change in operations will be accomplished by unloading and shutting down individual compressor trains until the desired flow rate is reached. Lack of demand will require venting of any excess CO₂ gas. Incineration prior to venting will be required because of its hydrocarbon content. The required incineration and waste heat recovery equipment were included in the initial Feasibility Study where all of this gas was vented continuously. This equipment must be retained in this arrangement to accommodate periodic venting operations.
- 2.5 CO₂ sales for EOR will increase coal consumption in the main steam plant. The steam balance in the Feasibility Study utilized waste heat recovered from CO₂ incineration prior to venting. CO₂ sales here would eliminate incineration and waste heat recovery from this gas. To maintain the present steam balance, an additional 572 MM BTU (HHV)/hr of coal must be burned in the coal burning steam plant boilers.

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2.6 Production data for the battery limit CO₂ compression plant are estimated as follows:

Maximum Operation	340 days/yr
CO ₂ Sales Gas Production Rate	429.87 MMSCFD
HHV Sales Gas	572 MM BTU/hr @ 31.9 BTU/SCF

2.7 The capital cost of the battery limit CO₂ compression plant on a January 1981 instantaneous cost basis with 10 percent contingency and no sales tax is estimated to be \$106.58 MM.

2.8 The direct operating requirements for the battery limit CO₂ compression plant are estimated as follows:

Electricity: 114,548 kw @ \$0.055/kwh	\$51.41 MM/yr
Cooling Water: 51,900 GPM @ \$0.053/1000 gal	1.35
Make-up Water: (112.1 GPM) @ \$0.50/1000 gal	(.03)
Coal: 572 MM BTU (HHV)/hr @ \$2.00/Million BTU	9.34
Ethylene Glycol: 0.10 gal/MM SCF @ \$3.35/gal	.05
Operating Labor: 3 men/shift @ \$29,000/man yr	.35
Maintenance Labor: 40% x 3.5% x capital cost	1.49
Maintenance Material: 60% x 3.5% x capital cost	2.24
Total Direct Operating Costs	\$66.20 MM/yr
	\$ 0.45/MSCF

NOTE: Utility costs above are those currently proposed by Fluor and are given in the Process Design Criteria manual.

3.0 PROCESS DESCRIPTIONS

Descriptions of the alternative designs are presented in the following sections.

3.1 Straight Compression Scheme

The configuration used for the design of this scheme is graphically described by the Process Flow Diagram on Fig. 6-1. Here the combined CO₂ gas from the Sulfur Recovery and Synthol units is pressurized from 15 to 2205 psia in a five stage compressor (28-1-C-1). Intercoolers with moisture separators (28-1-E-1, 28-1-E-2, 28-1-E-3, and 28-1-E-4) are provided interstage to minimize compression work and moisture load on the dryer. Sales gas moisture requirements are met by operating an ethylene glycol dryer (28-1-D-1) on the gas flowing between the 4th and 5th compressor stages. Ethylene glycol drying was selected in this case instead of molecular sieves because the pressure is sufficiently high to justify its use. Operating conditions at this point preclude the possible formation of

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hydrates. An aftercooler (28-1-E-5) is used to provide a moderate sales gas temperature. Recovered water is recycled to the Stretford unit after flash-degassing (28-1-V-1).

Six parallel equipment trains are provided to accommodate the design flow rate.

3.2 Ammonia Absorption Refrigeration Scheme

Compression of carbon dioxide in this arrangement is graphically described by the Process Flow Diagram on Fig. 6-2. Here, feed gases from the Sulfur Recovery and Synthol units are combined and compressed (28-1-C-1) in three stages to 315 psia. Intercoolers (28-1-E-1 and 28-1-E-2) including water knock-out facilities are provided between stages to reduce work requirements. A molecular-sieve dryer (28-1-D-1) is also included between the second and third compressor stages to produce a dry gas that meets product moisture requirements.

Dry gas at 315 psia is then cooled to 20°F in a feed-effluent exchanger (28-1-E-3) which utilizes the transferred energy to superheat pressurized CO₂ to about 127°F. The CO₂ gas at 20°F is further cooled and condensed to a liquid in exchanger 28-1-E-4 which removes the necessary heat by vaporizing liquid NH₃. The resultant liquid CO₂ is separated (28-1-V-2) from non-condensable gases and pumped (28-1-P-1) to the desired supercritical pressure. Refrigeration available in the supercritical CO₂ is recovered by inter-change (28-1-E-3) as mentioned above.

Available refrigeration associated with non-condensable gases leaving separator 28-1-V-2 is recovered by exchange (28-1-E-5) with liquid NH₃. These gases are then compressed from 300 psia to 2205 psia in a two stage reciprocating compressor (28-1-C-2) and combined with the main gas stream as part of the sales gas. An intercooler (28-1-E-6) and aftercooler (28-1-E-7) are provided here.

Six parallel equipment trains as shown on Fig. 6-2 are provided to accommodate the design flow rate.

The NH₃ absorption refrigeration system required to sustain these compressor trains is graphically depicted by the Process Flow Diagram on Fig. 6-3. In this arrangement, recycle strong ammonia liquors are fed into a fractionating column (28-1-V-4) which separates them into a liquid overhead that is essentially pure NH₃ (99.9 mol percent) and aqueous bottoms that contain 19 mol percent NH₃. Part of the liquid NH₃ is returned to the fractionating column as reflux. The remainder is sub-cooled (28-1-E-5 and 28-1-E-10) prior to Joule-Thompson expansion in order to reduce the amount of gas flashed in this

auto-refrigeration step. Low temperature flash gas is separated (28-1-V-6) from the remaining liquid which is then vaporized in the CO₂ condenser (28-1-E-4). A small liquid purge stream is withdrawn (28-1-P-5) to prevent trace H₂O impurity buildup in the boiling liquid NH₃. Vaporized NH₃ from the condenser is combined with cold flash gas from the separator and the liquid purge stream. Refrigeration available in this mixed stream is recovered by interchange (28-1-E-10).

Bottoms from fractionating tower 28-1-V-4 are stripped in a separate tower (28-1-V-8) which operates at about 135 psia to produce a 10 mol percent NH₃ bottoms product and a nominal 72 mol percent NH₃ overhead product. These bottoms are cooled by interchange (28-1-E-13) from 300°F to 105°F and used to reconstitute a 25 mol percent NH₃ strong liquor by reabsorption of the NH₃ vapors from 28-1-E-10. The resultant heat of absorption is removed from the system by a water cooled exchanger (28-1-E-12) that also serves as the absorber. The reconstituted strong liquor is then pumped (28-1-P-2) back to the ammonia stripper as feed after preheating in exchanger (28-1-E-13).

Overheads from the bottoms stripper (28-1-V-8) are totally condensed and pumped back (28-1-P-4) to fractionating tower 28-1-V-4 as a second strong liquor feed. Feed heater 28-1-E-11 partially vaporizes this liquor by transferring heat from the steam condensate leaving reboilers 28-1-E-8 and 28-1-E-15.

Two trains as shown on Fig. 6-3 are required for the total system.

3.3 Propylene Refrigeration Scheme

Compression of carbon dioxide in this arrangement is graphically described by the Process Flow Diagram on Fig. 6-4. Here, feed gases from the Stretford and Synthol units are combined and compressed (28-1-C-1) in three stages to 180 psia. Intercoolers (28-1-E-1 and 28-1-E-2) with water knock-outs are provided between stages to reduce work requirements. A molecular-sieve dryer (28-1-D-1) is also included between the second and third compressor stages to produce a dry gas that meets product moisture requirements.

Dry gas at 180 psia is then cooled to -20°F in a feed-effluent exchanger (28-1-E-3) which utilizes the transferred energy to superheat pressurized CO₂ to about 106°F. The CO₂ gas at -20°F is further cooled and condensed to a liquid in exchanger 28-1-E-4 which removes the necessary heat by vaporizing propylene. The resultant liquid CO₂ is separated (28-1-V-2) from non-condensable gases and pumped (28-1-P-1) to the desired supercritical pressure. Refrigeration available in the supercritical CO₂ is recovered in exchanger 28-1-E-3 as mentioned above.

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Available refrigeration in the non-condensable gases leaving separator 28-1-V-2 is recovered by exchange (28-1-E-5) with liquid propylene. These gases are then compressed from 166 psia to 2205 psia in a reciprocating compressor (28-1-C-2) and combined with the main gas stream as part of the sales gas. An intercooler (28-1-E-6) and aftercooler (28-1-E-7) are also provided.

Six parallel propylene refrigeration trains are provided to accommodate the design flow rate.

The propylene refrigeration system required to sustain these compressor trains is graphically depicted by the Process Flow Diagram on Fig. 6-5. In this arrangement, recycled propylene gas is liquified by two stage compression (28-1-C-3) and condensation (28-1-E-9). An interstage cooler (28-1-E-8) is included to minimize compressor work. Liquid propylene is then subcooled (28-1-E-5 and 28-1-E-10) prior to Joule-Thompson expansion in order to reduce the amount of gas flashed in this auto-refrigeration step. Low temperature flash gas is separated (28-1-V-4) from the remaining liquid which is then vaporized in the CO₂ condenser (28-1-E-4). Revaporized propylene from the CO₂ condenser is combined with flash gas from the separation and recycled to the compressor. Refrigeration available in this mixed stream is recovered by interchange in 28-1-E-10 prior to recompression.

4.0 DISCUSSION AND CONCLUSION

The Capital Cost reported in Section 2.0 of this report does not represent the total capital required. Tri-State's economic model should add items such as:

- Pre-Production Costs
- Working Capital
- Construction Loan Interest
- Capital Cost Escalation
- Land

About 12 acres of land would be needed to accommodate this compressor installation. This area is in addition to the plant site requirements identified in the Feasibility Study.

Licensor fees and royalties will not be required to construct and operate this compressor plant.

The comparative evaluation of the three processing methods considered in this study utilized Fluor's proprietary, computerized process simulation system to prepare complete heat and material balances for each alternate. This method provided a consistent and reliable basis for defining the energy requirements for each case. The magnitude of the differences identified by this method was adequate to delete the propylene refrigeration scheme from further consideration without additional engineering work.

Table 4-1 summarizes the mechanical work requirements for all cases. By inspection of this table and the appropriate process flow diagrams, it is apparent that the propylene refrigeration scheme is not competitive with straight gas compression for two reasons. First, it requires about 12½ percent more energy. Second, it is more complex mechanically and there are no obvious advantages.

Comparing the NH₃ absorption refrigeration system with straight gas compression is more difficult. The NH₃ absorption refrigeration system has a disadvantage of being more complicated mechanically but has the advantage of using less work. The latter is inherent in the NH₃ absorption refrigeration process which substitutes steam consumption in reboilers for mechanical work in a compressor. To determine if the advantages outweigh the disadvantages, it is necessary to consider overall plant operations.

The NH₃ absorption refrigeration system reduces total work requirements by 39,568 HP at the expense of using 807,903 lbs/hr of 120 psig saturated steam. This steam was used to generate electrical power in the feasibility study. It also increases system cooling water requirements nearly threefold as indicated in Table 4-2.

Comparative operating requirements for these two cases are presented in Table 4-3.

The Capital Costs of the two plants on a January 1981 instantaneous cost basis with 10 percent contingency and no sales tax have been estimated as follows:

Straight Compression Scheme	\$106.58 MM
NH ₃ Absorption Refrigeration Scheme	\$125.45 MM

Direct operating costs were estimated for both cases from the above data and the results are presented in Table 4-4. It should be noted that the "bottom line" here is dominated by the steam and power charges. The unit costs used to compute the operating costs are relative to both coal costs and the conversion costs of generating steam and power. Subsequent changes in the cost of one of these cases should produce proportionate changes in the other. Consequently, the selection of the straight gas compression route as the preferred method is expected to remain valid for reasonable changes in these unit costs.

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There are no major problems associated with the compression of large quantities of CO₂ to pipeline pressure. The compressor train shown in this report probably represents the maximum desirable size.

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TABLE 4-1
SUMMARY OF TOTAL
WORK INPUT (BHP)

<u>User Equipment Number</u>		<u>Propylene Refrigeration</u>	<u>Gas Compression</u>	<u>NH₃ Absorption Refrigeration</u>
Compression System				
28-1-C-1:	Stage 1	30,240	35,742	35,742
	Stage 2	28,344	31,326	34,590
	Stage 3	27,348	31,680	32,952
	Stage 4	-	29,550	-
28-1-C-2:	Stage 1	5,148	-	2,790
	Stage 2	4,722	-	2,478
28-1-P-1		4,380	-	4,512
	Sub-total	100,182	153,612	113,064
Refrigeration System				
28-1-C-3:	Stage 1	37,411	-	-
	Stage 2	35,293	-	-
28-1-P-2		2	-	904
28-1-P-3		-	-	3
28-1-P-4		-	-	70
28-1-P-5		-	-	3
	Sub-total	72,706	0	980
	Total	172,888	153,612	114,044

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TABLE 4-2

SUMMARY OF TOTAL COOLING

WATER REQUIREMENTS (lbs/hr)

<u>User Exchanger Number</u>	<u>Straight CO₂ Compression</u>	<u>NH₃ Absorption Refrigeration</u>
Compression System		
28-1-E-1	5,150,880	5,150,880
28-1-E-2	3,867,600	4,252,080
28-1-E-3	3,854,160	-
28-1-E-4	4,548,240	-
28-1-E-5	8,565,600	-
28-1-E-6	-	380,880
28-1-E-7	-	686,400
Sub-total	25,986,480	10,470,240
Refrigeration System		
28-1-E-9	-	38,957,000
28-1-E-12	-	9,141,924
28-1-E-14	-	13,696,716
Sub-total	0	61,795,640
Total	25,986,480	72,265,880

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TABLE 4-3

SUMMARY OF PROCESS

OPERATING REQUIREMENTS

<u>Expended Item and Measuring Unit</u>	<u>Gas Compression</u>	<u>NH₃ Absorption Refrigeration</u>
Electricity - kw	114,548	85,043
Cooling Water - GPM	51,900	144,400
Make-up Water - GPM	(112.1)	(98.3)
120 psig Steam - lbs/hr	0	807,903
Coal - MM BTU (HHV)/hr	972	972
Ethylene Glycol - gal/MMSCF	0.10	0
Operating Labor - men/shift	3	4

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TABLE 4-4

SUMMARY OF DIRECT

OPERATING COSTS

<u>Expended Item and Unit Cost</u>	<u>Gas Compression</u>	<u>NH₃ Absorption Refrigeration</u>
Electricity @ 0.055/kwh	\$51.41 MM/yr	\$38.17 MM/yr
120 psig steam @ \$2.45/M lbs	0	16.15
Cooling water @ \$0.053/M gal	1.35	3.75
Make-up water @ \$0.50/M gal	(.03)	(.02)
Coal @ \$2.00/MM BTU (HHV)	9.34	9.34
Ethylene Glycol @ \$3.35/gal	.05	0
Operating Labor @ \$29,000/man-yr	.35	.47
Maintenance Labor @ 40% x 3.5% x capital cost	1.49	1.76
Maintenance Material @ 60% x 3.5 x capital cost	2.24	2.63
	<hr/>	<hr/>
Total Direct Operating Cost:	\$66.20 MM/yr	\$72.25 MM/yr
	\$ 0.45/MSCF	\$ 0.49/MSCF

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5.0 DESIGN BASIS

5.1 Battery Limit Conditions

In all cases, the CO₂ battery limit pressure was set at 2200 psia to allow for some transport pressure loss before dropping below the critical pressure for this gas. Historically, this is consistent with the 2200 psia used by Shell/Texaco/Arco in their Crossett field (West Texas) which has an injection pressure of 1700 psi.

The battery limit temperature was arbitrarily set at 100°F to minimize pipeline expansion problems.

The dew point at battery limit pressure was set at -20°F to eliminate the possibility of water condensation in the pipeline.

Specific sulfur limits were set at 50 ppm for H₂S and no limit on other sulfur compounds. Tri-State should verify the acceptability of these limits. Fluor is not aware of any legal restrictions with pipelining CO₂. Precedents are being established in other EOR operations using CO₂ gas with a higher sulfur content than that considered here. (Arco's Wesson field is an intra-state operation that will use an 89% CO₂ gas containing 5% H₂S. This compares with a total H₂S + COS content of 0.07 percent for this application).

If unforeseen future events require this gas to meet natural gas sulfur requirements additional capital expenditure will be required. In this case the COS would be hydrolyzed to CO₂ and H₂S, and the H₂S selectively removed and converted to elemental sulfur. This study does not include this additional processing.

5.2 Liquefaction Conditions

In both refrigeration cases, CO₂ is condensed in a heat exchanger by transferring heat to a boiling refrigerant. The CO₂ condensing pressure is determined by the respective refrigerant evaporation temperature. Good design practice precludes operation at less than atmospheric pressure anywhere in the system in order to eliminate the risk of drawing air into the system. For this study this was accomplished by setting the system pressure for evaporating the refrigerant at nominally 18 psia. The corresponding saturation temperatures for propylene and ammonia are about -46°F and -20°F, respectively. Practical considerations in heat exchanger design translates these temperatures to CO₂ condensation temperatures of about -35°F and -10°F respectively. The corresponding pressures required for CO₂ condensation are nominally 170 psia and 305 psia.

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5.3 Gas Drying Methods

Only in the straight gas compression scheme is it possible to achieve a gas drying pressure above 700 psia. This is the nominal lower limit for using an ethylene glycol dryer to achieve the required dew point. The benefits derived from absorption drying accrued specifically to this scheme.

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6.0 FLOW DIAGRAMS

The flow diagrams depicting the three processes which were examined in this study are included here as follows:

<u>Figure</u>	<u>Title</u>
6-1	Straight CO ₂ Compression
6-2	CO ₂ Compression System with NH ₃ Absorption Refrigeration
6-3	NH ₃ Absorption Refrigeration System
6-4	CO ₂ Compression System with Propylene Refrigeration
6-5	Propylene Refrigeration System

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Indirect Coal Liquefaction Plant
Western Kentucky

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Contract 835504

APPENDIX 1

PROCESS DEVELOPMENT STUDY NO. 17

CO₂ COMPRESSION FOR ENHANCED OIL RECOVERY

SCOPE OF STUDY

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PROCESS STUDY

CO₂ COMPRESSION STUDY

1.0 GENERAL

This study will provide the capital and operation costs for drying and compressing CO₂ to be sold for use in enhanced oil fields recovery operations.

2.0 WORK DEFINITION

2.1 Composition of the feed gas is proprietary with the process developer.

2.2 A study will be made to determine the method for pressurizing the carbon dioxide and moving it to the plant battery limits. Two possible routes are: 1) compression, 2) liquefaction and pumping through a vaporizer to the desired pressure. For the liquefaction of CO₂, the use of ammonia absorption refrigeration and propylene refrigeration will be investigated.

2.3 The pipeline from battery limits to the CO₂ purchaser is not included in this study.

2.4 Determine the pressure, temperature and dew point requirement for the CO₂ stream. Condensation must be avoided to prevent excessive corrosion to carbon steel.

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PROCESS STUDY (Continued)

- 2.5 Evaluate the effect of partial feed on plant operations when CO₂ is either not available or demand for CO₂ slackens.
- 2.6 Estimate capital cost for the selected process.
- 2.7 Estimate operating cost for the system selected.

3.0 DELIVERABLE TO TRI-STATE

A formal report that contains the following:

- 3.1 Capital cost estimate.
- 3.2 Operating cost estimate.
- 3.3 Block Flow diagrams, process description and material balances.

4.0 SCHEDULE

It is estimated that the above work will be completed 10 weeks after the work is started by Fluor.

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APPENDIX 2

PROCESS DEVELOPMENT STUDY NO. 17

CO₂ COMPRESSION FOR ENHANCED OIL RECOVERY

ESTIMATE DETAILS

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COST ESTIMATE

CLIENT TRI-STATE SYNFUELS DESCRIPTION "CO₂ COMPRESSION STUDY" PROP. NO. _____
 LOCATION WESTERN KENTUCKY GAS COMPRESSION SCHEME W.O. NO. _____
 PROJECT TRI-STATE SYNFUELS JAN 1980 INSTANTANEOUS CONT. NO. 835504
 MADE BY _____ APPROVED _____

A/C NO.	ITEM & DESCRIPTION	MANHOURS x 10 ³	ESTIMATED COST U.S. \$ x 10 ³						
			LABOR	SUB-CONTRACTS	MATERIALS	TOTAL			
00-00	Excavation								
10-00	Concrete								
20-00	Structural Steel								
30-00	Buildings								
40-00	Machinery & Equipment								
50-00	Piping								
60-00	Electrical								
70-00	Instruments								
80-00	Painting & Scaffolding								
85-00	Insulation								
	DIRECT FIELD COSTS	818	9	816	56	056	65	872	
90-00	International Expense								
91-00	Temporary Construction Facilities								
92-00	Constr. Services, Supplies & Expense								
93-00	Field Staff, Subsistence & Expense								
94-00	Craft Benefits, Payroll Burdens & Insur.								
95-10	Equipment Rental								
95-50	Small Tools								
99-40	Field Staff Overhead Costs								
	INDIRECT FIELD COSTS							19	113
	TOTAL FIELD COSTS							84	985
96-00	Home Office Construction								
	Project Engineering								
	Process Engineering								
	Design								
	Purchasing								
	Business Services								
97-00	Office Expense								
98-00	Office Payroll Burdens								
99-50	Office Overhead Costs								
	TOTAL OFFICE COSTS							12	117
	TOTAL FIELD & OFFICE COSTS							97	102
99-30	Sales Tax								
99-10	Escalation								
99-20	Contingency							9	473
	TOTAL								
99-60	Fee								
	TOTAL PROJECT							106	575

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COST ESTIMATE

CLIENT TRI-STATE SYNFUELS DESCRIPTION "CO2 COMPRESSION
STUDY" PROP. NO. _____
 LOCATION WESTERN KENTUCKY NH₃ ABSORPTION SCHEME W.O. NO. _____
 PROJECT TRI-STATE SYNFUELS JAN 1980 INSTANTANEOUS CONT. NO. 835504
 MADE BY _____ APPROVED _____

A/C NO.	ITEM & DESCRIPTION	MANHOURS x 10 ³	ESTIMATED COST U.S.\$ x 10 ³			
			LABOR	SUB-CONTRACTS	MATERIALS	TOTAL
00-00	Excavation					
10-00	Concrete					
20-00	Structural Steel					
30-00	Buildings					
40-00	Machinery & Equipment					
50-00	Piping					
60-00	Electrical					
70-00	Instruments					
80-00	Painting & Scaffolding					
85-00	Insulation					
	DIRECT FIELD COSTS	1015	12 180		64 401	76 581
90-00	International Expense					
91-00	Temporary Construction Facilities					
92-00	Constr. Services, Supplies & Expense					
93-00	Field Staff, Subsistence & Expense					
94-00	Craft Benefits, Payroll Burdens & Insur.					
95-10	Equipment Rental					
95-50	Small Tools					
99-40	Field Staff Overhead Costs					
	INDIRECT FIELD COSTS					23 659
	TOTAL FIELD COSTS					100 240
96-00	Home Office Construction					
	Project Engineering					
	Process Engineering					
	Design					
	Purchasing					
	Business Services					
97-00	Office Expense					
98-00	Office Payroll Burdens					
99-50	Office Overhead Costs					
	TOTAL OFFICE COSTS					14 054
	TOTAL FIELD & OFFICE COSTS					114 294
99-30	Sales Tax					
99-10	Escalation					
99-20	Contingency					11 152
	TOTAL					
99-60	Fee					
	TOTAL PROJECT					125 446

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COST ESTIMATE

CLIENT TRI-STATE SYNFUELS DESCRIPTION "CO₂ COMPRESSION PROP. NO. _____
STUDY" W.O. NO. _____
 LOCATION WESTERN KENTUCKY GAS COMPRESSION SCHEME CONT. NO. 835504
 PROJECT TRI-STATE SYNFUELS JAN 1980 INSTANTANEOUS MADE BY _____
 APPROVED _____

A/C NO.	ITEM & DESCRIPTION	MANHOURS x 10 ³	ESTIMATED COST U.S.\$ x 10 ³			
			LABOR	SUB- CONTRACTS	MATERIALS	TOTAL
00-00	Excavation					
10-00	Concrete					
20-00	Structural Steel					
30-00	Buildings					
40-00	Machinery & Equipment					
50-00	Piping					
60-00	Electrical					
70-00	Instruments					
80-00	Painting & Scaffolding					
85-00	Insulation					
	DIRECT FIELD COSTS	818	9	816	56	056
						65 872
90-00	International Expense					
91-00	Temporary Construction Facilities					
92-00	Constr. Services, Supplies & Expense					
93-00	Field Staff, Subsistence & Expense					
94-00	Craft Benefits, Payroll Burdens & Insur.					
95-10	Equipment Rental					
95-50	Small Tools					
99-40	Field Staff Overhead Costs					
	INDIRECT FIELD COSTS					19 113
	TOTAL FIELD COSTS					84 985
96-00	Home Office Construction					
	Project Engineering					
	Process Engineering					
	Design					
	Purchasing					
	Business Services					
97-00	Office Expense					
98-00	Office Payroll Burdens					
99-50	Office Overhead Costs					
	TOTAL OFFICE COSTS					12 117
	TOTAL FIELD & OFFICE COSTS					97 102
99-30	Sales Tax					
99-10	Escalation					
99-20	Contingency					5 473
	TOTAL					
99-60	Fee					
	TOTAL PROJECT					106 575

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