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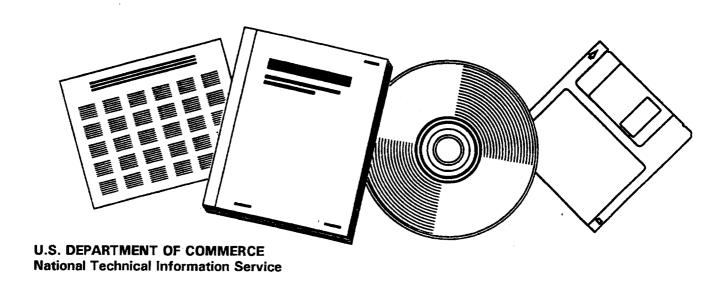
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LIQUID-PHASE METHANOL. FINAL REPORT

CHEM SYSTEMS, INC. FAIRFIELD, NJ

DEC 1979



Liquid-Phase Methan

AF-1291 Research Project 317-2

Final Report, December 1979

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ABSTRACT

This report is the final report in the RP317 series. This project dealt with the development of the Liquid-Phase Methanol Reactor. The concept of this reactor involves carrying out the synthesis of methanol in an ebullated bed of catalyst in the presence of an inert liquid heat carrier. Work was carried out on laboratory, continuous bench-scale, and a process development unit.

Previous reports on this project included AF202 and AF693.

TPRI PERSPECTIVE

PROJECT DESCRIPTION

This is the final comprehensive report in a series of reports, including EPRI Annual Report AF-202 and EPRI Interim Report AF-693, that have been published under EPRI Research Project (RP) 317. This project involved bench-scale and process development unit (PDU) testing of a new liquid-phase reactor, conceived by Chem Systems Incorporated for producing methanol from synthesis gas. Data were collected at two levels of reactor size to determine scale-up effects in a liquid-phase ebullated catalyst bed. Compositions of synthesis gas (a mixture of carbon monoxide and hydrogen) that closely simulated gases which would be obtained from either a Koppers-Totzek or a Lurgi coal gasifier were tested. An ebullated-bed reactor has the adventage that it allows for the operation of an isothermal reactor with the heat of relation being removed essentially by an inert oil, thereby reducing the quantity of recycle gas required for reactor cooling. For this reason and others, the technology proposed by Chem Systems, while still at a developmental stage, has some potential for improvements over the commercially available methanol synthesis processes.

PROJECT OBJECTIVE

The underlying objective is to develop lower cost technology by which methanol can be produced from coal gasification products. The technology for converting synthesis gas derived from natural gas to methanol has been highly developed commercially. Processes licensed by Lurgi and Imperial Chemical Industries (ICI) have dominated the field in recent years. In the production of methanol from coal, the actual preparation of methanol from synthesis gas represents less than IOX of the total plant investment.

However, the Chem Systems reactor has the potential for energy savings as well as capital savings. This process is also of particular interest to the electric utilities because methanol has excellent combustion properties and because it is conceptionally possible to integrate this process to co-produce some amount of clean peaking fuel with a gasification-combined-cycle for electric power generation.

EPRI has helped support a large-scale combustion turbine test at Southern California Edison Company to demonstrate the use of fuel (RP998-1). EPRI has also conducted several engineering evaluations regarding methanol production (e.g., RP1715, RP832-1, and RP832-3). The Chem Systems' reactor is the only major experimental effort supported by EPRI to improve methanol fuel production economics. Lehigh University (RP779-12) performed some catalyst deactivation analyses in support of the development.

PROJECT RESULTS

The project was divided sequentially into two parts. The first part, reported in EPRI AF-202, demonstrated the basic process concept without particular concern to datalyst life. In the second part of the program, an attempt was made to develop a catalyst that would be suitable for the process and to test that catalyst in a process development unit producing about one gallon of methanol per hour. Here problems were encountered. Attempts to produce an active, attrition-resistant catalyst were not completely successful. This work has been reported in EPRI Interim Report AF-693. The best catalyst formulation available was 3/32-inch minitablets. A 30-day side-by-side test run was conducted using both the bench-scale and process development units. These results discussed in this report showed not only a catalyst attrition problem but also a catalyst deactivation problem.

This problem has shown that methanol from coal-derived synthesis gar can be produced via Chr Systems' process. However, development of an abrasion-resistant, active, low-cost catalyst still remains one of the unsolved problems. In an effort to develop such a catalyst, EPRI has contracted with United Catalyst (RP1656). That work should be completed by mid-1980. Other suggestions have been made to circumvent the catalyst attrition problem. The most significant one is to revert to a slurry catalyst system in place of the particulate catalyst system. Still, additional research is needed to resolve the catalyst deactivation problem.

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CONTENTS

Section		Page
1	INTRODUCTION AND BACKGROUND	1-1
	PROCESS DESCRIPTION	1-1
	PROCESS APPLICATIONS	1-3
	ACCOMPLISHMENTS	1-4
2	SUMMARY	
	DEMONSTRATION OF CONCEPT FEASIBILITY	2-1
	THREE-PHASE BENCH SCALE UNIT	2-2
	CATALYST DEVELOPMENT	23
	PROCESS VARIABLE STUDIES	2-4
	SCALE-UP IN A LARGER PROCESS DEVELOPMENT UNIT	2-6
	LONG TERM CONTINUOUS RUNS	2-7
	ENGINEERING STUDIES	2-8
3	DEMONSTRATION OF THE BASIC CONCEPT	3-1
	PROCESS THERMODYNAMICS	3-i
	FIXED BED REACTION TESTS	3-15
	Gas Feed Only and Gas-Liquid Co-Feed	3-15
	Recirculating Liquid	3-25
4	BENCH SCALE UNIT DEVELOPMENT PROGRAM	4-1
	DESCRIPTION OF THE BEHICH SCALE UNIT	4-1
	INITIAL OPERATIONS	4-3
	PROCESS VARIABLE SCANG - CRUSHED COMMERCIAL CATALYSTS	4_7
	Effect A Feed Gas Composition	4-9
	Effect of Temperature	4 -13
	Effect of Fressure	4-28
	Effect of Particle Size	4-28
	Effect of Process Liquid and Catalyst Composition	4-32

Section		Page
	CATALYST DEVELOPMENT	4-44
	Catalyst Forming Tests	4-44
	Sevelopment of a Mini-Tablet	4-44
	Development of an Extrudate	4-52
	Modified Tablet Composition	4-70
	Tests at Lehigh University	4-71
	Contaminant Metals Analysis	4-76
5	TESTING OF PROCESS VARIATIONS	5 - 1
	SLURRY PHASE OPERATION	5-1
	COMBINED WATER-GAS SHIFT/METHANOL SYNTHESIS REACTIONS	5-3
	CO ₂ REACTIVITY	5-3
	MAXIMIZATION OF CO CONVERSION TO LIQUIDS	5-14
	Chamical Means	5-14
	Catalyst Modifications to Produce Dimethyl Ether	5-14
	Catalyst Modifications to Produce Higher Alcohols	5-17
	Physical Means	5-22
6	LONG-TERM ACTIVITY TESTS	6-1
	LURGI GAS	6-1
	Bench Scale Unit Tests	61
	Discontinuous Tests	6-1
	Cuntinuous Tests	6-4
	Process Development Unit Tests	6-7
	Description of the Process Development Unit	6-7
	Discontinuous Tests	6-8
	Continuous Tests	6-9
	KOPPERS-TOTZEK GAS	6-15
	Continuous Bench Scale Unit and Process Development Unit Tests	6-15
	Synopsis of BSU Run	6-35
	Synopsis of PDU Run	6-15
7	POTSONING STUDIES	
	Reactor A - Control Run	7-1 7-4
	Reactor B - Ethylene	7-4 7-4
	Reactor B NO.	7-4 7-4
	Reactor B - Sulfur (AS H ₂ S)	7-21
	Reactor A - Chlorine (AS CH ₃ C1)	7-22
	- 4.10 31.30 7	· -

Section

8	SUPPORTING STUDIES	Page
	REACTION MGDELLING	8-1
	Kinetic Model	9-1
	Empirical Model	8-1
	K-T Type Feed Gas	8-7
	METHANOL PRODUCT ANALYSIS	8-15
	Lurgi Feed Gas	8-16
	Koppers-Totzek	8-16
	LIQUID PHASE ACTIVITY COEFFICIENTS	8-23
	SOLUBILITY OF MINERAL OIL IN METHANOL -WATER	8-23
	GAS SOLUBILITY IN MINERAL OIL	8-25
	M. Wherear Off	8-27

ILLUSTRATIONS

Figure		Page
1-1	Methanol Reaction System	1-2
3-1	Equilibrium Conversion - KT Gas	3-8
3-?	Equilibrium Conversion - Lurgi Gas	3-9
3-3	Equilibrium Conversion Lurgi Gas After CO2 Removal	3-10
3-4	Equilibrium Conversion - Lurgi Balanced Gas	3-11
3-5	Equilibrium Conversion Case Summary at 30 Atm.	3-12
3-6	Equilibrium Conversion Case Summary at 50 Atm.	3-12.
3-7	Equilibrium Conversion Case Summary at 250°C	3-13
3-8	Theoretical Equilibrium CO Conversion to Methanol for Lurgi Gas vs. Temperature	3-19
3-9	Theoretical Equilibrium CO Conversion to Methanol for Lurgi Gas vs. Pressure	3-20
3-10	Fixed Bed Reactor Conversion (Catalyst A/Mitco 40) for Lurgi Gas	3-21
3-11	Fixed Bed Reactor Conversion (Catalyst B/Witco 40) for Lurgi and KT Gases	3-23
3-12	Comparison of Catalysts A and B in Fixed Red Reactor	3-24
3-13	Recirculating Liquid Reactor Conversion (Catalyst 8/	3-24
	Witco 40) for Lurgi Gas	3-27
4-1	Bench Scale Unit	4-2
4-2	Comparison of Fixed and Fluidized Bed Reaction Results	4-6
4-3	Effect of Pressure on Methanol Conversion	4-8
1-4	Conversion With Gatalyst A (-12+16 Mesh)/Witco 40 System and Lungi Gas	4-14
4-5	Conversion With Catalyst A (=12+16 Mesh)/Witco 40 System and KT Gas	4-15
4-6	Conversion With Catalyst A (~16+20 Mesh)/Witco 40 System and Lungi Gas	4- 16
4-7	Conversion With Catalyst A (=16+20 Mesh)/Witco 40 System and KT Gas	4-17
1-8	Conversion With Catalyst A (-15+20 Mesh)/Witco 40 System and 2H ₂ /1CO/0.1 CO ₂ Gas	4-18

Figur	<u>re</u>	Page
4-9	Conversion With Catalyst A (-16+20 Mesh)/Witco 40 System and 2H ₂ /CO Gas	4-19
4-10	Methamol Productivity for Various Synthesis Gases at -12+16 Mesh	4-20
4-11	Methanol Productivity for Various Synthesis Gases at -16+20 Mesh	4-21
4-12	Effect of Temperature With Catalyst A (-12+16 Mesh) and Lurgi Gas	4-22
4-13	Effect of Temperature With Catalyst A (-16+20 Mesh) and KT Gas	4-23
4-14	Effect of Temperature With Catalyst A (-16+20 Mesh) and Lurgi Gas	4-24
4-15	Effect of Temperature With Catalyst A (-16+20 Mesh) and KT Gas	4-25
4-16	Effect of Temperature With 2H ₂ /1CO/0.1 CO ₂ Feed Gas	4-26
4-17	Effect of Temperature With 2H ₂ /CO Feed Gas	4-27
4-18	Effect of Pressure With Catalyst A (-12+16 Mesh)	4-29
4-19	Effect of Pressure With Catalyst A (-16+20 Mesh)	4-30
4-20	Effect of Particle Size With Catalyst A.	4-31
4-21	CO Conversion vs. Space Velocity For Catalyst A (-16+20 Mesh)/Exxon 150 System	4-38
4-22	CO Conversion vs. Space Velocity For Catalyst B (-16+20 Mesh)/Witco 40 System	4-39
4-23	MeOH Productivity vs. Temperature For Catalyst A/ Exxon Aromatic 150	4-40
4-24	MeOH Productivity vs. Temperature For Catalyst B/ Witco 40 Mineral Oil	4-41
4-25	Methanol Productivity For Various Catalyst/Liquid Pairs	4-42
4-26	Effect of Crush Strength	4-50.
4-27	Bench Scale Methanol: Catalyst A-Witco 40 System and K-T Gas/1000 PSIG	4-53
4-28	Conversion versus On-Stream Time Catalyst A: 3/16" Ø Tablets	4-56
4-29	Conversion versus On-Stream Time W. R. Grace Catalyst #1	4-59
4 - 30	Conversion versus On-Stream Time W. R. Grace Catalyst #2	4-60
4-31	Conversion versus On-Stream Time W. R. Grace Catalyst #3	4-61
4-32	Mixture Consistencies for the Catalyst A: SECAR 80: Water System	4-72

Figu	<u>ire</u>	Page
5-1	Effect of Direct Steam Addition on Methanol Productivity	5-6
5-2	Process Variable Scan Catalyst A/Witco 40 Mineral Oil Feed Gas: 9.2% CO, 7.3% CO ₂ , 7.6% Inerts and Balance H ₂	5-11
5-3	Process Variable Scan Catalyst A/Witco 40 Mineral Oil Feed Gas: 14.3% cO, 9.5% cO ₂ . 4.1% Incrts and Balance H ₂	5-12
5-4	Process Variable Scan Catalyst A/Witco 40 Mineral Oil Feed Gas: 2.7% CO, 13.9% CO ₂ , 5.6% Inerts and Balance H ₂	5-13
5-5	High Pressure Autoclave Unit	•
5-6	BSU Modifications for Flash Vaporization Scheme	5-15
57	Effect of Removal of Methanol from the Circulating Process Liquid by Flash Vaporization	5-23 5-26
5-8	Effect of Flash Vaporization Pressure: Catalyst A/Witco 40/Lurgi Gas at PReactor = 1000 prig	5-33
5-1	Discontinuous Activity Maintenance Test Catalyst A/Witco 40/Lurgi Feed Gas	6-3
5-2	BSU Continuous Run - Lurgi Gas	6-6
5-3	PDU Continuous Run - Lurgi Gas	5-12
6-4 5-5	BSU Continuous Run - KT Gas PDU Continuous Run - KT Gas	6-16 6-17
7-1	Poisoning Reactor Flow Diagram	7-2
7-2	Cumulative Effect of H ₂ S Poisoning Commercial Catalyst A (3/16" Ø x 3/16") Mixed Lurgi - Nr Feed (2/3, 1/3) 450 psia - Salt Bath at 220°C	
7-3	Cumulative Effect of CH ₃ Cl Poisoning Commercial Catalyst A (3/6" Ø 3/16") Mixed Lurgi = N ₂ Feed (2/3, 1/3) 450 psia = Salt Bath at 225°C	7-23
3-1		7-24
3-1 3-2	Equilibrium Conversion for a Lurgi Feed Gas at 1000 psir	8-8
	CO Conversion versus Space Velocity for a Lurgi Feed Gas at 1000 psig with 12-16 Mesh Catalyst A	8-9
3-3	Approach to Equilibrium versus Space Velocity for a Lurgi Feed Gas at 1000 psig with 12-16 Mesh Catalyst A	0 10

Figure	1	Page
8-4	CO Conversion Versus Space Velocity for a Lurgi Feed Gas at 1000 psig with 16-20 Mesh Catalyst A	8-11
8-5	Approach to Equilibrium versus Space Velocity for a Lung: Feed Gas at 1000 psig with 16-20 Mesh Catalyst A	8-12
8-6	CO Conversion versus Space Velocity for a Lungi Feed Gas at 1000 psig with 3/32" Tablets Catalyst A	8-13
8-7	Approach to Equilibrium versus Space 2 locity for a Lurgi Feed Gas at 1000 psig with 3/5 Tablets Catalyst A	8-14
8-8	Approach to Equilibrium versus Space Velocity for a Lurgi Feed Gas at 1000 psig for Three Catalyst Sizes	8-15
8-9	Equilibrium for a K-T Feed Gas at 1000 psig	8-17
8-10	CO Conversion versu: Space Velocity for a K-T Feed Gas at 1000 psig with 3/32" Catalyst A Tablets	8-18
8-11	Approach to Equilibrium versus Space Velocity for a K-T Feed Gas at 1000 psig with 3/32" Catalyst A Tablets	8-19
8-12	CO Conversion versus Space Velocity for a K-T Feed Gas at 1000 psig with 12-16 Mesh Catalyst A	8-20
8-13	Approach to Equilibrium versus Space Velocity for a K-T Feed Gas at 1000 psig with 12-16 Mcsh Catalyst A	8-21
8-14	Oil Solubility in Methanol	3-29

TABLES

Tab1	<u>e</u>	Page
2-1	Produc MeOH Composition	2-5
3-1	Equilibrium Compositions	
3-2	Synthesis Gas Compositions	3-2
3-3	Methanol Synthesis Feed Gas - Cases 1 and 2	3-4
3-4	Mathaniol Synthesis Feed Gas - Cases 3 and 4	3-5
3-5	Mechanol Synthesis in a Fixed Bed With CCI Catalyst	3-6
3-6	Methanol Synthesis in a Fixed Bed With Catalyst A	3-16
5 5	(-5+12 Mesh)/Witco 40	3-17
3-7	Methanol Synthesis in a Fixed Bed with Lungi and KT Gases and Catalyst B (-5*12 Mesh)/Witco 40	3-17
3-8	Methanol Synthesis in a Recirculating Liquid Bed With Lungi Gas and Catalyst B	3-26
4-1	Process Variable Scan For Catalyst A (~12+10 Mesh) and OH ₂ /CO Gas	
4-2	Process Variable Scan For Catalyst A (-12+16 Mesh)/ Witco 40 System	4-5 4-10
4-3	Process Variable Scan For Catalyst A (-16+20 Mesh)/ Witco 40 System	4-10
4-4	Process Variable Scan For Catalyst A (-16+20 Mesh/ Witco 40 System	4-17
4-5	Physical Properties of Liquids Used	4-33
4-6	Process Variable Scan For Catalyst A (-16+20 Mesh)/ Exxon Aromatic 150 System	4-33
4-7	Process Variable Scan For Catalyst B (-16+20 Mesh)/ Witco 40 Mineral Oil System	4-35
4-8	Process Variable Scan For Catalyst B (-16+20 Mesh)/ Hitco 40 System	4-36
4-9	Physical Properties of Catalyst Tablet Samples Prepared from Catalyst A Calcined Powder	435
4-10	Screening Tests with Mini-Tablets	4-47
4-11	BSJ Process Variable Scan for Catalyst A (3/32" @)/ Witco 40 Mineral Oil/KT Gas	4-47
4-12	Alcohol Product Composition - Effect of Feed Gas	4-51 4-54
4-13	Alcohol Product Composition - Effect of Catalyst Formulation	4-54
		4-7/

Table		Page
4-14	Composition and Physical Properties of Methanol Catalyst Formulations	4-58
4-15	Vapor Phase Screening Tests - Effect of Catalyst Composition	4-63
4-16	BSU Process Variable Scan for Cataly: t A (3/32" 0)/ Witco 40 Mineral Oil/Lurgi Gas	4-65
4-17	BSU Process Variable Scan for Catalyst A (3/32" 9)/ Witco 40 Mineral Dil/Lurgi Gas	4-66
4-18	BSU Process Variable Scan for Catalyst A/ a-Alumina (3/32" @)/Witco 40 Mineral Oil/Eurgi Gas	4-67
4-19	Comparison of Vapor Phase and Liquid Phase Screening Tests	4-68
4-20	BSU Process Variable Scan for WRG #2/ α -Alumina (1/16" extrudates)/Witco 40 Mineral Oil/Lurgi Gas	4-69
4-21	Catalyst-Water-Hydraulic Cement Formulations	4-73
4-22	Surface Areas of Catalyst Samples	4-75
4-23	Catalyst Contaminant Levels	4-77
5-1	Simulated Slurry Operation in the Bench Scale Unit Commercial Catalyst A (50 x 70 Mesh)/Witco 40/ Lurgi Feed Gas	5-2
5-2	Comparison of Methanol Productivity for 16 x 20 and 50 x 70 Mesh Catalyst A	5-4
5-3	Effect of Steam Addition on Methanoï Synthesis Reaction Commercial Catalyst A (16 x 20 Mesh)/Exxon Aromatic 150	5-5
5-4	Process Variable Scan Catalyst A/Witco 40 Mineral Oil/Gas Blend A	5-8
5-5	Process Variable Scan Catalyst A/Witco 40 Mineral Oil/Gas Blend B	5-9
5 - 6	Process Variable Scan Catalyst A/Witco 40 Mineral Oil/Gas Blend C	5-10
5-7	Modified Catalysts to Product Dimethyl Ether Autoclave Tests: Catalyst A/Witco 40 Mineral Dil/ Lungi Feed Gas	5-16
5-8	Modified Catalysts to Produce Higher Alcohols Autoclave Tests: Catalyst A/Witco 40 Mineral Oil/ Lungi Feed Gas	5-18
5-9	Flash Vaporization of Reactor Effluent Commercial Catalyst A (-16+20 Mesh)/Witco 40	
	Mineral Oil/Lurgi Feed Gas	5-24

Table		Page
5-10	Flash Vaporization of Reactor Effluent Commercial Catalyst A (-16+20 Mesh)/Exxon Aromatic 150/Lurgi Feed Gas	5-25
5-11	Effect of MeOH Removal by Flash Vaporization CCI-C-79 (-16+20 Mesh)/Witch 40 Mineral Dil/ Lungi Feed Gas	5-28
5-12a	Effect of MeOH Removal by Flash Vaporization Catalyst A (-16+20 Mesh)/Witco 40 Mineral Oil/ Lurgi Feed Gas	5-29
-12b	-	5-29
5-13	Effect of MeOH Removal by Flash Vaporization Catalyst A/Witco 40 Mineral Oil/Lurgi Feed Gas	5-31
5-14	Effect of MeOH Removal by Flash Vaporization Catalyst A/Witco 40 Mineral Oil/Lurgi Feed Gas	5-32
6 - 1	Discontinuous Catalyst Activity Maintenance Commercial Catalyst A (-16+20 Mesh)/Witco 40 Mineral Oil/Lurgi Feed Gas	6-2
6-2	Bench Scale Unit Continuous Run - Catalyst A (3/32" 9)/Witco 40 Mineral Oil/Lurgi Gas	65
6-3	Discontinuous Process Develorment Unit Variable Scan- Catalyst A (3/32" Ø)/Witco 40 Mineral Oil/Lurgi Gas	6-10
5-4	Process Development Unit Continuous Runs - Catalyst A (3/32" @)/Witco 40 Mineral Oil/Lungi Gas	6-11
6-5	Catalyst Inspections from the PDU Program and Bench Scale Life Test	6-14
6-6	Product Alcohol Composition - BSU vs PDU	6-18
6-7	Catalyst Weight Recovery - 30 Day Run - BSU vs PDU	6-20
6-8	Catalyst Dimensions - Fresh and Recovered From 30 Day Run - BSU vs PDU	6-20
6-9	Trace Metal Analysis in Recovered Catalyst From 30 Day Run - BSU vs PDU	6-22
7-1	Methanol Synthesis Catalyst Poisoning Studies Reactor A - Commercial Catalyst A (3/16" 9 x 3/16" Pellets)	7-5
7-2	Methanol Synthesis Catalyst Poisoning Studies Reactor B - Commercial Catalyst A (3/16" Ø x 3/16" Pellets)	7-13
8-1	Approach to Equilibrium for Low Feed Gas Rates Commercial Catalyst A (16 x 20 Mesh)/Witco 40/	Ω Α

Tabl	<u>e</u>	
8-2	Calculated Reaction Rate Constant	Page
8-3	Chromatographic Analysis of Methanol Samples Identification of Minor Components	8-5
8-4	Chromatographic Analysis of Methanol Samples - Effect of Feed Gas	8-22
8-5	Activity Coefficients: MeOH/Witco 40 Mineral Oil and MeOH/Exxon 150 Aromatic	8-24
8-6	Solubility of Witco 40 Mineral Oil in Methanol-Water Solutions	8-26
8~7	Component Henry's Law Constants in Witco 40 Mineral	8-28
	· •	8-30